

297. Stereospecific Synthesis of a 9, 11, 12, 13, 13a, 14-hexahydro-dibenzo(*f, h*)pyrrolo(1,2-*b*)isoquinoline-alkaloid

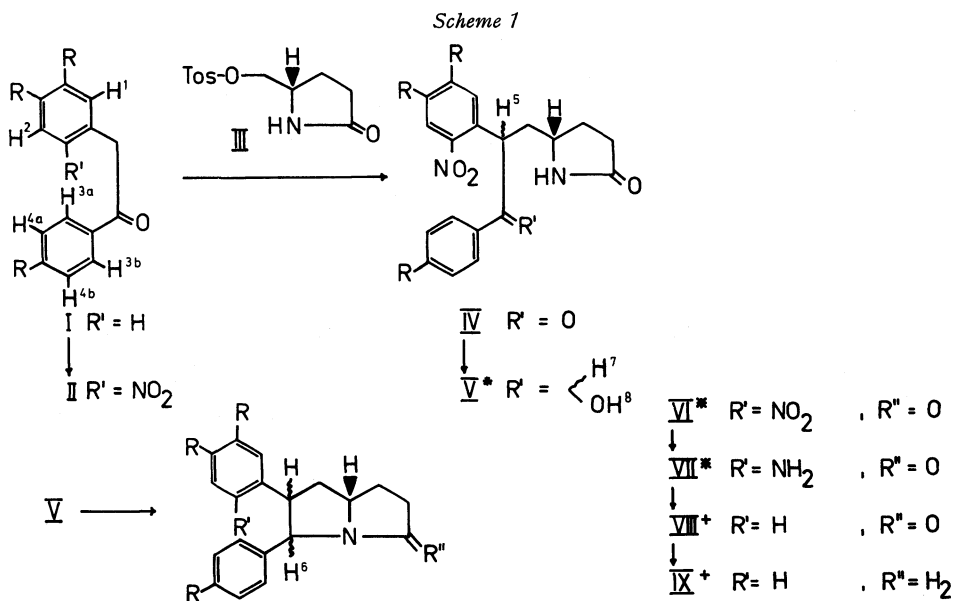
Preliminary communication

by Ludwig Faber and Wolfgang Wiegrebe

Pharmazeutisches Institut, Berne

(12. XI. 73)

Several syntheses of the title-type alkaloids starting from chiral proline-derivatives, are described in the literature [1-4]. All these routes produce racemates because they include intermediates with a carbonyl-group α to the asymmetrically substituted carbon-atom C(13a). We avoided racemisation by using the proline derivative III [5] and cyclising the phenanthrene-moiety at a late stage of our synthetic route, shown in the following schemes:

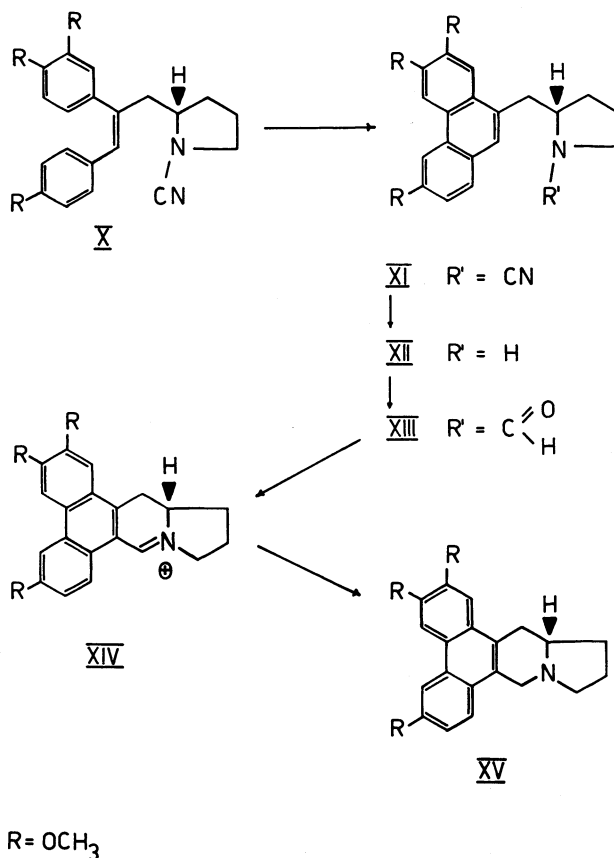


*) diastereomers were isolated +) mixture of diastereomers $R = \text{OCH}_3$

We obtained III by LiBH_4 -reduction of 5-oxo-proline methyl ester to 5-oxo-prolinol which was subsequently tosylated. The optical purity of 5-oxo-prolinol was only 51% of that of *Hardegger's* [5] product, who used LiAlH_4 . This disadvantage is compensated by a better yield of 5-oxo-prolinol, as compared with the LiAlH_4 -reduction.

Anisole + homoveratroylchloride \rightarrow I [6]. - I + 65% HNO_3 in glacial AcOH \rightarrow II, m.p. 197°; NMR. (chem. shift in δ): H_1 : 6.80, 1s; H_2 : 7.83, 1s; $\text{H}_{3a,b}$: 8.09, $2d \times d$, $J_o = 7$ Hz, $J_m = 1.7$ Hz; $\text{H}_{4a,b}$: 7.03, $2d \times d$, $J_o = 7$ Hz, $J_m = 1.7$ Hz. - II + III in acetone/ K_2CO_3 , 110°, 4-5 atm \rightarrow IV; yellow oil; M^+ : 428. NMR.: H_5 : 5.67, 1t, $J = 7.5$ Hz; $[\alpha]_D^{20} = +22^\circ$ (MeOH). - IV + NaBH_4 (70% EtOH) \rightarrow V. Va, low Rf., NMR.: H_7 : 4.74, 1d, $J = 7$ Hz; H_8 : 3.00, 1s. - Vb, high Rf, NMR.: H_7 : 4.77, 1d, $J = 7$ Hz; H_8 : 3.00, 1s. - V (a or b) + 1.25% HCl in glacial AcOH (20°) \rightarrow VI (a or b respectively). VIa (high Rf); m.p. 152°; phenylgroups *cis*; NMR.: H_6 : 5.08, 1d, $J = 8.5$ Hz; $[\alpha]_D^{20} = +89^\circ$ (CHCl_3). - VIb (low Rf); m.p. 194°; phenylgroups *trans*; NMR.: H_6 : 4.83, 1d, $J = 1.7$ Hz; $[\alpha]_D^{20} = +43.5^\circ$ (CHCl_3). - VI (a or b) + H_2 (*Raney-Ni*/EtOH) \rightarrow VII (a or b respectively). MS.: M^+ : 382, dominating peaks at 203 and 175. - VII (mixture of diastereomers) + NaNO_2 /2.5% aqueous HCl, followed by 30% H_3PO_2 \rightarrow VIII. MS. of both diastereomers: M^+ : 367, dominating peaks 203 and 175. - VIII (mixture of diastereomers) + LiAlH_4 /THF \rightarrow IX; diastereomers (colourless oils) were separated. IXa (low Rf), $[\alpha]_D^{25} = -214^\circ$ (MeOH); IXb (high

Scheme 2



Rf), $[\alpha]_D^{25} = -173^\circ$ (MeOH). MS. (both diastereomers): M^+ 353, base peak: 189. – IX (mixture of a and b) + BrCN in abs. $C_6H_6 \rightarrow X$. IR. ($CHCl_3$): 2200/cm. MS.: M^+ : 378, dominating peaks 283 and 95. – X (4.6×10^{-5} M in C_6H_{12} , saturated with air) was irradiated (254 nm) \rightarrow XI; m.p. $151-2^\circ$; UV.: λ_{max} (log ϵ): 257 (4.73), 285 (4.44), 311 (3.90), 342 (2.89), 359 nm (2.31); M^+ : 376; $[\alpha]_D^{25} = +74.7$ ($CHCl_3$). – XI + $LiAlH_4$ in THF \rightarrow XII (viscous oil). M^+ : 351, small intensity [7], dominating peaks 349, 282 and 70; $[\alpha]_D^{22} = +8.1^\circ$. – XII + 98% $HCOOH/180^\circ \rightarrow$ XIII; cyclisation (without further purification) with $POCl_3$ /toluene \rightarrow XIV; UV.: λ_{max} : 259; 268; 282; 320; 420 nm. XIV + $NaBH_4/MeOH \rightarrow$ XV, m.p. $206-11^\circ$. $[\alpha]_D^{22} = +66^\circ$ ($CHCl_3$); antofine: $[\alpha]_D^{22} = -131^\circ$ ($CHCl_3$). XV was identical with natural antofine [8] (TLC., UV., MS., IR.) but shows opposite optical rotation, 50% in magnitude for the reason stated. Because XV has S-configuration, antofine is of the R-configuration. This is in accord with our degradation of antofine to D-proline [9].

REFERENCES

- [1] B. Chauncy, E. Gellert & K. N. Trivedi, Austral. J. Chemistry, 22, 427 (1969).
- [2] R. B. Herbert & C. J. Moody, Chem. Commun. 1970, 121.
- [3] T. R. Govindachari, B. R. Pai, S. Prahahakar & T. S. Savitri, Tetrahedron 21, 2573 (1965).
- [4] B. Chauncy & E. Gellert, Austral. J. Chemistry, 23, 2503 (1970).
- [5] E. Hardegger & H. Ott, Helv. 38, 313 (1955).
- [6] A. Novelli & A. de Santis, Bol. Soc. Quim. Peru 30, (4) 155 (1964); Chem. Abstr. 64, 5069a.
- [7] W. Wiegrebe, L. Faber & H. Budzikiewicz, Liebigs Ann. Chem. 733, 125 (1970).
- [8] W. Wiegrebe, L. Faber, H. Brockmann jr., H. Budzikiewicz & U. Krüger, Liebigs Ann. Chem. 721, 154 (1969).
- [9] W. Wiegrebe, L. Faber & Th. Breyhan, Arch. Pharmaz. 304, 188 (1971).