# Synthesis and Pharmacological Characterization of Subtype-Selective Ligands, Including Radio- and <br> Fluorescence Labeled Ligands, for the Histamine $\mathrm{H}_{2}$ Receptor 

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## Chapter 1

## General Introduction

### 1.1 The Histamine $\mathrm{H}_{2}$ Receptor as a Prototypic Aminergic GPCR

The histamine $\mathrm{H}_{2}$ receptor $\left(\mathrm{H}_{2} \mathrm{R}\right)$ belongs to the superfamily of G -protein coupled receptors (GPCRs). ${ }^{1}$ GPCRs are integral membrane receptors and are characterized by seven hydrophobic transmembrane (TM) domains with an extracellular amino terminus and an intracellular carboxyl terminus. The extracellular regions combined with the transmembrane regions are important for ligand binding. ${ }^{2}$ The intracellular regions are substantially involved in signaling and feedback mechanisms. ${ }^{2}$ With around $30 \%$ of the most prominent approved drugs targeting these membrane receptors, GPCRs are the most important drug targets. ${ }^{3,4}$ GPCRs are mediated by numerous endogenous ligands e.g. biogenic amines (aminergic GPCRs), amino acids, peptides, proteins, purins and lipids, to name only a few. ${ }^{1,5,6}$

The $H_{2} R$ is one of currently four histamine receptor subtypes ( $H_{1} R, H_{3} R$ and $H_{4} R$ ), which are all activated by binding the endogenous ligand histamine and therefore are aminergic GPCRs. ${ }^{7-10}$ All histamine receptors belong to the rhodopsin family of GPCRs. ${ }^{1}$ The $\mathrm{H}_{2} \mathrm{R}$ is primarily located on parietal cells in the stomach, ${ }^{11}$ in mammalian brain, ${ }^{12,13}$ on neutrophiles and eosinophiles ${ }^{14}$ as well as on smooth muscle cells ${ }^{15}$ (e.g. in the heart, airways and uterus). An essential physiological function of the $\mathrm{H}_{2} \mathrm{R}$ is the control of the gastric acid secretion. ${ }^{8,11}$ Furthermore, activation of $\mathrm{H}_{2} R$ results in smooth muscle relaxation and positive inotropic and chronotropic effects. ${ }^{16}$

The $H_{2} R$ species isoforms (e.g. human $\left(h_{2} R\right)$, guinea pig $\left(g p H_{2} R\right)$, rat $\left(\mathrm{rH}_{2} R\right.$ ), mouse $\left(m H_{2} R\right)$ and dog $\left(\mathrm{cH}_{2} \mathrm{R}\right)$ ), like many GPCRs, interact similarily with their endogenous ligand, but quite differently with most synthetic ligands. ${ }^{17,18}$ The pharmacological differences between the $\mathrm{hH}_{2} \mathrm{R}$ and the $\mathrm{gpH}_{2} \mathrm{R}$ mainly concern agonists and not antagonists, which was very fortunate as the first potent antagonists for the treatment of gastroduodenal ulcers were developed relying on animal models. ${ }^{17}$ The $\mathrm{cH}_{2} \mathrm{R}$ exhibits an increased constitutive activity compared to $\mathrm{hH}_{2} \mathrm{R}$ and $\mathrm{rH}_{2} \mathrm{R} .{ }^{18}$ These findings show that for the development of highly potent and selective agonists it is crucial to study $\mathrm{hH}_{2} \mathrm{R}$ and not the orthologs.

### 1.2 G-Protein Activation and Signaling Pathways

In the classical model the active receptor conformation (either stabilized by agonist binding or constitutively active) is functioning as a guanosine nucleotide exchange factor (GEF) on the G $\alpha$ subunit of the heterotrimeric G-protein (Figure 1.1). ${ }^{5}$ The binding of the G-protein complex to the active receptor leads to conformational changes which result in the release of GDP from its binding site at the G $\alpha$ subunit and the formation of the ternary complex. ${ }^{19}$ Subsequently, GTP is bound and the ternary complex dissociates into the G $\alpha-G T P$ subunit, the $G \beta \gamma$ complex and the free receptor. ${ }^{19}$ Both subunits can interact with effector proteins resulting, through an increase or a decrease in the concentration of second messangers, in various cellular responses. ${ }^{19}$ After a certain period of time, the intrinsic GTPase activity of the G $\alpha$ subunit converts GTP to GDP and phosphate. ${ }^{19}$ The $G \alpha-G D P$ subunit re-associates with the $G \beta \gamma$ complex to the inactive heterotrimeric G-protein. ${ }^{19}$

Based on their structures and signaling pathways, G-proteins are grouped in four families according to their $G \alpha$ subunit: $G \alpha_{i / 0}, G \alpha_{s}, G \alpha_{q / 11}, G \alpha_{12 / 13} .^{20,21}$ The $H_{2} R$ predominantly couples to
$G \alpha_{s}$ proteins, resulting in an increase of the second messenger cAMP by stimulation of the isoforms of the effector protein adenylyl cyclase (Figure 1.1). ${ }^{22,23}$ By contrast, the $\mathrm{H}_{3} \mathrm{R}$ and the $\mathrm{H}_{4} \mathrm{R}$ signal mainly via $G \alpha_{i / 0}$ proteins, which inhibit the adenylyl cyclase. ${ }^{24}$ The $H_{1} R$ preferentially couples to $G \alpha_{q / 11}$ leading to the activation of phospholipase $C$ (PLC) and subsequent release of $\mathrm{IP}_{3}$ and DAG. ${ }^{24,25}$


Figure 1.1. Activation of the heterotrimeric G-protein by the agonist occupied receptor using the $H_{2} R$ as an example. $R$ represents the inactive receptor conformation and $R^{*}$ the active receptor conformation. The dissociated subunits ( $G \alpha_{s}$ and $G \beta \gamma$ complex) regulate effector proteins such adenylyl cyclase (AC), which is activated by $\mathrm{G} \alpha_{\mathrm{s}}$. Modified from Rasmussen et al. ${ }^{26}$

For analyzing GPCR-mediated guanine nucleotide exchange at G-proteins, a widely employed method is the $\left[{ }^{35} \mathrm{~S}\right]$ GTP $\gamma$ S binding assay. ${ }^{27}$ This assay utilizes, like the closely related steady-state GTPase assay, the intrinsic GTPase activity of the G $\alpha$ subunit. An advantage of the GTP $\gamma$ S binding assay (and GTPase assay) is that it assesses coupling at a proximal level, avoiding potential bias introduced by downstream events. ${ }^{27}$ For the $\mathrm{H}_{2} \mathrm{R}$ the usage of membranes of $\mathrm{Sf9}$ insect cells, which are expressing mammalian $\mathrm{H}_{2} \mathrm{R}-\mathrm{G}_{s \alpha}$ fusion proteins is well established. ${ }^{17,28,29}$ GPCR-G ${ }_{s \alpha}$ fusion proteins ensure a defined 1:1 stoichiometry of the signaling partners and efficient coupling. ${ }^{17,29,30}$ Therefore, the ternary complex formation is more efficient compared to the coexpression of $\mathrm{H}_{2} \mathrm{R}$ plus $\mathrm{G}_{\text {s }} \cdot{ }^{28}$ In our workgroup, the $\mathrm{H}_{2} \mathrm{R}-\mathrm{G}_{\mathrm{s} \alpha}$ fusion protein system is routinely employed for analyzing new ligands for the $\mathrm{H}_{2} \mathrm{R}$ in radioligand binding and functional studies (GTPYS binding assay and GTPase assay). ${ }^{31-36}$

### 1.3 G-Protein Independent Signaling, Ligand Classification and Functional Selectivity

Besides the signal transduction cascades mediated by G-proteins, GPCRs are reported to participate in numerous other protein-protein-interactions which initiate signaling pathways independent from G-protein activation. ${ }^{19,37,38}$ Most intriguing is the interaction with $\beta$-arrestins, which are mainly involved in receptor desensitization and internalization, but also act as alternative signal transducers. ${ }^{19,37} \beta$-Arrestin recruitment is initiated by phosphorylation of the active conformation of the GPCR by G-protein coupled receptor kinases (GRK). ${ }^{38,39}$ The $\beta$-arrestin binds to the cytosolic surface of the phosphorylated receptor and sterically hinders an interaction with the G-proteins. ${ }^{40}$ Furthermore, $\beta$-arrestins were reported to be involved in the degradation of second messengers. ${ }^{41,42}$ These two effects effectively lead to the deactivation of the G-protein mediated signal transduction. Beyond desensitization, the bound $\beta$-arrestin also mediates internalization via clathrin-coated pits. ${ }^{38}$

A classical "two state" model, which is often suitable for explaining the pharmacodynamic activity of ligands is the cubic ternary complex model. ${ }^{43-45}$ This model distinguishes between an active $\left(R^{*}\right)$ and inactive (R) receptor state, which are in equilibrium and are able to isomerize without agonist binding. This spontaneous activation of the receptor in the absence of an agonist is referred to as constitutive activity. ${ }^{46}$ The G-protein is able to bind to both states, albeit only the G-protein-active-receptor-complex $\left(\mathrm{R}^{*} \mathrm{G}\right)$ activates intra cellular signaling via a GDP-GTP exchange. Ligand binding can shift the equilibrium of the receptor state. Agonists bind with high affinitiy to $R^{*}$ and stabilize the active conformation. Inverse agonists prefer to bind to $R$ and stabilize the inactive conformation. Neutral antagonists bind with the same affinity to both conformations and therefore do not alter the equilibrium. With regard to $\beta$-arrestin mediated signaling, along site with other mechanism such as phosphorylisation, internalization and oligomerisation, there is growing evidence that there are multiple active and inactive receptor conformations. ${ }^{47,48}$ Structurally different ligands stabilize distinct receptor conformations leading to an activation of only a subset of cellular effectors ${ }^{48}$ This selective activation of only some of all possible signaling pathways has been referred to as 'functional selectivity', ${ }^{49}$ 'biased agonism' ${ }^{50}$ or 'differential receptor-linked effector actions ${ }^{\prime 51}$.

Recently, several monomeric and dimeric $\mathrm{H}_{2} \mathrm{R}$ ligands were investigated for biased agonism regarding G-protein activation and $\beta$-arrestin recruitment. ${ }^{52}$ The $\beta$-arrestin recruitment was measured by an enzyme fragment complementation assay using split luciferase fragments from P. termitilluminans, developed by Misawa et al. ${ }^{53}$ While all antagonists were unbiased, the investigated acyl- and carbamoyl guanidine agonists revealed varying degrees of G-Protein bias. ${ }^{52}$

## 1.4 $\mathrm{H}_{2} \mathrm{R}$ Antagonists

The classical $\mathrm{H}_{2} \mathrm{R}$ antagonists can be devided into two groups depicted in Figure 1.2: compounds comprising a flexible chain (group I) and compounds containing diaryl moiety (goup II). ${ }^{54}$ The antagonists consist of an aromatic system, which is linked to a polar, planar group (urea equivalent) by either a flexible chain (group I) or by a second aromatic system (group II). The
classification of the antagonists is made according to the aromatic system. Most $\mathrm{H}_{2} \mathrm{R}$ antagonists belong to one of four major structural classes: imidazole-, guanidinothiazole-, aminomethylfurane- and piperidinomethylphenoxy-containing compounds.


Compounds with diaryl structure


Examples:



mifentidine

Figure 1.2. Selected $H_{2} R$ antagonists and their classification into two groups: compounds with a flexible chain and compounds with a diaryl moiety. ${ }^{54}$

The aminopotentidine derivatives as well as fluorescent ligands (e.g. compound II) showed that within the piperidinomethylphenoxy-containing compounds additional substituents at the urea equivalent are well tolerated or provided additional $\mathrm{H}_{2} \mathrm{R}$ binding affinity (Figure 1.3). ${ }^{55-57}$ Up to date, iodoaminopotentidine, a piperidinomethylphenoxy-containing cyanoguanidine (Figure 1.3), which was also synthesized in a radiolabeled form ( $\left[{ }^{125} \mathrm{I}\right]$ iodoaminopotentidine), showed the highest affinity. ${ }^{13}$ This radioligand was used to map the $\mathrm{H}_{2} \mathrm{R}$ densities in human and mammalian brain. ${ }^{12,13}$ Recently, a series of piperidinomethylphenoxyalkylamine-containing ligands, coupled with various polar groups ("urea equivalents") such as cyanoguanidine, nitroethenediamine, amide or squaric amide moieties, and a terminal amino group, connected via a linker of different length, was developed by our group. ${ }^{35}$ The squaramides, which also tolerated propionylation at the terminal amino-group showed the highest affinities. UR-DE257, which showed a high affinity ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.55), was also synthesized in radiolabeled form ( $\left[{ }^{3} \mathrm{H}\right]$ UR-DE257) and is frequently used in competition binding experiments. ${ }^{35,36}$


lodopotentidine ${ }^{a}$ $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{i}}: 9.15 \mathrm{nM}$



Figure 1.3. Structures of exemplary piperidinomethylphenoxy-containing ligands. a: Hirschfeld et al and Ruat et al ${ }^{13,57}$; b: Malan et al ${ }^{58}$; c: Baumeister et al ${ }^{35}$.

Whereas in the late 1970s to 1980s, the $H_{2} \mathrm{R}$ antagonists ( $\mathrm{H}_{2}$ blockers) revolutionized the treatment of peptic ulcers, former blockbuster drugs like cimetidine, ranitidine or famotidine are outdated. They were largely superseded by the more effective proton pump inhibitors (e.g. omeprazole). Nevertheless, $\mathrm{H}_{2} \mathrm{R}$ antagonists are valuable molecular tools to study the $\mathrm{H}_{2} \mathrm{R}$, especially it's role in the brain, which is still far from being completely understood.

## 1.5 $\mathrm{H}_{2}$ R Agonists

The early agonists for the $\mathrm{H}_{2} \mathrm{R}$ were derived from histamine and consististed of an imidazole pharmakophore coupled to a guanidine by a flexible linker (e.g. impromidine and arpromidine, see Figure 1.4). ${ }^{59,60}$ Arpromidine and related compounds showed up to 400 times potency of histamine at the spontaneously beating guinea pig right atrium, but the strongly basic guanidine moiety led to poor oral bioavailability and CNS penetration. ${ }^{59}$ The bioisosteric replacement of the guanidine ( $\mathrm{p} K_{\mathrm{a}} \sim 13$ ) with an acylguanidine ( $\mathrm{p} K_{\mathrm{a}} \sim 8$ ) resulted in ligands with either retained or even increased agonistic potency (e.g. UR-PG80 and UR-AK24, see Figure 1.4). ${ }^{31,32}$ Modification of these $N^{G}$-acylated imidazolylpropylguanidines, which lacked subtype selectivity ( $H_{3} R$ and $H_{4} R$ ), to $N^{6}$-acylated aminothiazolylpropylguanidines led to highly potent and selective $\mathrm{H}_{2} \mathrm{R}$ agonists. ${ }^{34}$ Surprisingly, the $H_{2} R$ potency was increased up to 4000 -fold the potency of histamine by linking two acylguanidine moieties (e.g. UR-AK381, see Figure 1.4). ${ }^{33}$ The aminothiazole dimeric ligands are the most potent and selective $\mathrm{H}_{2} \mathrm{R}$ agonists known so far. Traditionally, dimeric (bivalent) ligands consist of two pharmacophoric moieties linked through a spacer and are designed to bridge two neighboring receptor protomers. ${ }^{61}$ Porthogese et al suggested a distance of about 22$27 \AA$ between the two orthosteric binding sites of a receptor dimer. ${ }^{62}$ Interestingly, the most active dimeric ligands have spacer of lengths insufficient to bridge the protomers of putative $\mathrm{H}_{2} \mathrm{R}$ dimers. ${ }^{33}$ The enormous gain in potency is speculated to result from an interaction with the orthosteric and an accessory binding site at the same protomer. ${ }^{33}$ Recently, it was shown that bioisosteric replacement of the acylguanidines with the more stable carbamoylguanidine led to dimeric ligands with retained potency and intrinsic activity (e.g. UR-NK22, see Figure 1.4). ${ }^{36}$ So far, there is no $H_{2} R$ agonist for therapeutic use on the market, but $H_{2} R$ agonists are valuable molecular tools to study the $\mathrm{H}_{2} \mathrm{R}$. Nonetheless, there are numerous possible indications e.g. as positive inotropic vasodilators for the treatment of congestive heart failure or as differentiationintroducing agents for treatment of acute myeloid leukemia (AML). For the later, the endogenous
agonist histamine is used as an orphan drug in combination with interleukin 2. Histamine promotes the activation of $T$ cells and natural killer cells by interleukin 2 , which results in the killing of cancer cells. ${ }^{63}$ Given the effect of histamine is mediated via $\mathrm{H}_{2} \mathrm{R}$, the application of highly selective $\mathrm{H}_{2} \mathrm{R}$ agonists might be beneficial in regard to potency and a reduction of adverse effects. Recently, investigation of the dimeric $\mathrm{N}^{6}$-carbamoylated aminothiazolylpropylguanidines on human monocytes revealed a high $\mathrm{H}_{2} \mathrm{R}$ agonist potency, suggesting that this class of compounds is a promising starting point for the development of $\mathrm{H}_{2} \mathrm{R}$ agonists for the treatment of AML. ${ }^{36}$




Arpromidine
$\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pEC}_{50}: 8.01^{\mathrm{a}}$
$\alpha: 1.00$
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC} \mathrm{C}_{50}: 6.72^{\mathrm{b}}$
$\alpha: 0.72$


UR-AK24 ${ }^{\text {b }}$
$\mathrm{hH}_{2} \mathrm{R}$ : $\mathrm{pEC} \mathrm{C}_{50}: 7.17$
$\alpha: 0.87$


UR-Bit24 ${ }^{\text {b }}$
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC}_{50}: 7.65$
$\alpha: 0.79$

$\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pEC} \mathrm{C}_{50}: 9.41$
$\alpha$ : 0.79
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC}_{50}: 8.11$
$\alpha: 0.53$


Figure 1.4. Structures of selected $\mathrm{H}_{2} \mathrm{R}$ agonists. Agonism measured on ${ }^{\text {a }}$ guinea pig right atrium ${ }^{32}$, ${ }^{\text {b }}$ steady-state GTPase assay ${ }^{31,33,34}$ or ${ }^{\mathrm{c}} \mathrm{GTP} \psi \mathrm{S}$ binding assay ${ }^{36}$.

### 1.6 Receptor Ligand Binding Assays and Labeled Molecular Tools FOR GPCRs

The initial step in every signaling cascade, that causes a receptor-mediated biological response, is the binding of a ligand to the receptor. There are multiple ways to utilize receptor-ligandinteractions in research, e.g. as a tool for determining receptor distribution, for identification of receptor subtypes and for screening of new compounds. ${ }^{64}$

The classical approach for the determination of ligand affinity is the radioligand binding assay, which has been unchallenged for a long time regarding sensity and reproducibility. ${ }^{64,65}$ Radioligand binding experiments can be divided in three basic types: Saturation binding experiments are used to determine the affinity of the radioligand and the number of specific binding sites. In kinetic experiments, the rate constants of association and dissoviation of a radioligand can be determined. Competition binding experiments are widely used to identify unlabeled compounds, which bind to the receptor in question by displacement of a radioligand. The major disadvantages include that radioligands are potentially hazardous to human health,
produce high costs in production and waste disposal, require special licences and laboratory equippment, and separation of bound from unbound ligand is necessary.

Today, new highly sensitive fluorescence and bioluminescence methods such as fluorescence polarization (FP), ${ }^{66}$ total internal reflection fluorescence (TIRF), ${ }^{67}$ fluorescence/bioluminescence resonance energy transfer (FRET/BRET), ${ }^{68}$ fluorescence recovery after photobleaching (FRAP), ${ }^{69}$ high content imaging ${ }^{70}$ and flow cytometry ${ }^{71}$ became promising alternatives. Like radioligands, fluorescent ligands can be used in the basic types of binding experiments. Several peptidic and non-peptidic fluorescent ligands were identified for GPCRs, including NPY, ${ }^{72,73}$ muscarinic ${ }^{74}$ and histamine ${ }^{55,56,75}$ receptors. In general, a fluorescent ligand comprises of a pharmacophore, a linker and the fluorophore. A major challenge in the development of small-molecule fluorescent ligands is to retain affinity, when a, compared to the ligand, bulky fluorophore is attached. In comparison, a radiolabel, especially tritium, does not alter the affinity of the ligand.

When selecting a radio- or fluorescent labeled ligand for binding experiments, several aspects have to be considered. ${ }^{65,76,77}$ Firstly, the ligand should be selective and bind with high affinity to the respective receptor. Secondly, high (radiochemical) purity and high specific activity (radioligand) or quantum yield (fluorescent ligand) is required. Thirdly, the labeled ligand should be chemically stable under assay conditions for at least the duration of the experiment performed. Furthermore, unspecific binding has to be considered, the choice of radio- or fluorescent label and, whether an agonist or an antagonist is desired as labeled ligand. Under unspecific binding all binding sites other than the receptor of interest are summarized. A competition binding assay, were only $50 \%$ of total radioligand binding is specific is considered adequate, $70 \%$ is good and $90 \%$ is excellent. ${ }^{76}$ Tritium is often considered as the radioisotope of choice. Compared to ${ }^{125}$ or the occasionally used ${ }^{32 / 33} \mathrm{P}$ or ${ }^{35} \mathrm{~S}$, tritium has a longer half-life (14-87 days vs. 12.3 years) and the tritiated compounds are more convenient in handeling with respect to safety precautions. ${ }^{76}$ When choosing a fluorophore, excitation and emission wavelengths, stoke shifts and quantum yields have to be considered. Generally, red-emitting ( $\lambda_{e m}:>600 \mathrm{~nm}$ ) fluorophores with a long stoke shift and a high quantum yield are preferred. ${ }^{77}$ Agonists label only an active conformation of the receptor and therefore, only a fraction of the total active receptor population. ${ }^{78}$ By contrast, antagonists bind to all receptor states with the same affinity according to the classical model described above. ${ }^{78}$

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## Chapter 2

## Scope and Objectives

The Histamine $\mathrm{H}_{2}$ receptor, an aminergic GPCR, is primarily known for its physiological role in the control of gastric acid secretion. ${ }^{1,2}$ Additionally, activation of $\mathrm{H}_{2} \mathrm{R}$ results in positive inotropic and chronotropic effects and smooth muscle relaxation. ${ }^{3}$ The $\mathrm{H}_{2} \mathrm{R}$ is primarily located on parietal cells in the stomach, ${ }^{2}$ in mammalian brain, ${ }^{4,5}$ on human neutrophiles and eosinophiles ${ }^{6}$ as well as on smooth muscle cells ${ }^{7}$. Antagonists, which were intensively studied as antiulcer therapeutics in the 1960s to 80s, play only a minor role today, but are still important. Although, there is no $\mathrm{H}_{2} \mathrm{R}$ agonist for therapeutic usage on the market, $\mathrm{H}_{2} \mathrm{R}$ agonists are valuable molecular tools to study the pharmacology of the $\mathrm{H}_{2} \mathrm{R}$. Nonetheless, there are numerous possible indications e.g. as positive inotropic vasodilators for the treatment of congestive heart failure or as differentiationintroducing agents for treatment of acute myeloid leukemia. The development of selective high affinity molecular tools for the $\mathrm{H}_{2} \mathrm{R}$, including agonists as well as labeled molecules like radioligands and fluorescent ligands, is very important to identifiy new ligands, investigate receptor distribution and further unravel its (patho-)physiological role.

The number of suitable high affinity radioligands for the $\mathrm{H}_{2} \mathrm{R}$ is very limited. [ ${ }^{3} \mathrm{H}$ ]Histamine, as well as several reported tritiated antagonists (e.g. [ $\left.{ }^{3} \mathrm{H}\right]$ cimetidine, ${ }^{8,9} \quad\left[{ }^{3} \mathrm{H}\right]$ ranitidine ${ }^{10}$ and $\left[{ }^{3} \mathrm{H}\right]$ tiotidine ${ }^{11}$ ) are less than ideal molecular tools to study the $\mathrm{H}_{2} \mathrm{R}$. As $\left[{ }^{3} \mathrm{H}\right]$ cimetidine is reported to label an imidazole recognition site rather than the $\mathrm{H}_{2} \mathrm{R}^{8}$ and ranitidine as well as histamine suffer from low $\mathrm{H}_{2} \mathrm{R}$ affinity and potency. ${ }^{5,11}\left[{ }^{3} \mathrm{H}\right]$ Tiotidine is frequently used as a radioligand, although it shows very high unspecific binding and is reported to address only a subpopulation of the $\mathrm{H}_{2} \mathrm{R} .{ }^{11}\left[{ }^{125} \mathrm{I}\right]$ lodoaminopotentidine suffers from a short half-life of only 60 days, but shows the highest affinity to the $\mathrm{H}_{2} \mathrm{R}$ reported so far $\left(\mathrm{gpH}_{2} \mathrm{R}: K_{d}\right.$ value: 0.34 nM$) .{ }^{5,12}$ An tritated alternative presents the recently published high-affinity tritium-labeled $\mathrm{H}_{2} \mathrm{R}$ antagonist [ ${ }^{3} \mathrm{H}$ ]UR-DE257 ${ }^{13}$ $\left(\mathrm{hH}_{2} \mathrm{R}: K_{d}\right.$ value: 31 nM ), which is structurally related with the squaramide BMY25368 ${ }^{14}$. This radioligand proved to be useful for the determination of $\mathrm{p} K_{\mathrm{i}}$ values, but turned out as an insurmountable antagonist in functional assays. ${ }^{13}$ One objective of this thesis was to design new ligands with a free terminal amino group and exploration of the applicability of them as precursors for the attachment of a radioactive (tritiated) moiety. Two reported classical antagonistic structures were used as scaffolds: the guanidinothiazoles and the aminopotentidines.

The guanidinothiazole-moiety is a privileged structure for $\mathrm{H}_{2} \mathrm{R}$ antagonism. Previous functional studies identified guanidinothiazole-containing ligands, e.g. famotidine and ICI127032, as surmountable $\mathrm{H}_{2} \mathrm{R}$ antagonists. ${ }^{15,16}$ In this thesis, two different guanidinothiazole-precursors derived from famotidine and ICI127032 had to be combined with an "urea" equivalent. As a strategy to enable radiolabeling, the introduction of diaminoalkyl-linkers varying in length at the "urea" equivalent was envisaged. The resulting terminal amino group could be propionylated by succinimidyl propionate, to obtain "cold forms" of the potential radioligands.

Aminopotentidine and its derivatives are reported as high affinity $\mathrm{H}_{2} \mathrm{R}$ antagonists. ${ }^{12}$ Interestingly, iodination in the 3 -position of the 4 -aminobenzoic acid amide moiety resulted in an enormous gain in affinity (iodoaminopotentidine). ${ }^{4,12}$ Aminopotentidine and its analogs containing different substituents (e.g. iodine, bromine, chlorine, trifluoromethyl) in position 3 were synthesized as precursors for radiolabeling. The derivatization of the anilinic amino group of these precursors was performed with various reagents (e.g. N-succinimidyl propionate, propionic acid chloride, methyl iodide or acetyl chloride), which are also commercially available in tritiated form. Anilinic
amines show a reduced nucleophilicity which resulted in a reduced reactivity in the (radio-)labeling reaction. To overcome this challenge, the synthesis and characterization of a series of aminopotentidine derivatives, containing a functionalized (propionylated, acetylated or methylated) aminomethyl substituent in position 4 of the aromatic ring was considered.

The availability of high affinity red-emitting $\mathrm{H}_{2} \mathrm{R}$ fluorescent ligands is very limited. Recently, a series of fluorescent ligands with a piperidinomethylphenoxypropylamino (potentidine) pharmacophore was reported. ${ }^{17}$ The most promising ligands within this series are the squaramide-type ligands UR-DE229 and UR-DE56. ${ }^{17}$ Both ligands consist of a BMY $2536^{14}$ pharmacophore, which is linked to a fluorescent label (pyridinium or cyanine) by a $n$-alkyl linker. Another objective of this thesis was the synthesis and biological characterisation of fluorescent high affinity $\mathrm{H}_{2} \mathrm{R}$ antagonists with improved optical and physicochemical properties to gain access to a wide range of potential applications, in particular to confocal microscopy and to high throughput or/and high content imaging. Therefore, the fluorescent labeled antagonists URDE229 and UR-DE56 were investigated in different assay/imaging systems. For the exploration of the impact of alkyl linker length and different net charges of fluorophores, a series of derivatives of UR-DE229 and UR-DE56 was considered.
$\mathrm{N}^{G}$-acylated amino(methyl)thiazolepropylguanidines represent a class of potent and selective histamine $H_{2} R$ agonists. ${ }^{18-20}$ As it was reported that $N^{G}$-acylguanidines undergo hydrolytic cleavage upon storage in aqueous solution, more stable analogues are needed. ${ }^{21,22}$ A bioisosteric approach, replacing the $N^{6}$-acylguanidine structure with a $N^{6}$-carbamoylguanidine, was planned in the final part of this thesis, aiming at $\mathrm{N}^{6}$-carbamoylated aminothiazole-containing compounds with high affinity and improved long term stability. Furthermore, structure-activity ( $\left.\mathrm{H}_{2} \mathrm{R}\right)$ and the structure-selectivity relationships ( $H_{2} R$ versus $H_{1} R, H_{3} R$ and $H_{4} R$ ) of this class of compounds were considerd.

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## Chapter 3

## Guanidinothiazoles: Towards the Squaramide- Type $\mathrm{H}_{2} \mathrm{R}$ Radioligand [ ${ }^{3} \mathrm{H}$ ]UR-SB69

Note: The synthesis and purification of $\left[{ }^{3} \mathrm{H}\right] 3.25$ was performed by Max Keller.

### 3.1 Introduction

Over the years, the endogenous agonist histamine and several $\mathrm{H}_{2} \mathrm{R}$ antagonists were synthesized in a radiolabeled form (Figure 3.1). Nevertheless, [ $\left.{ }^{3} \mathrm{H}\right]$ histamine, ${ }^{1}\left[{ }^{3} \mathrm{H}\right]$ cimetidine, ${ }^{2,3}\left[{ }^{3} \mathrm{H}\right]$ ranitidine ${ }^{4}$ and $\left[{ }^{3} \mathrm{H}\right]$ tiotidine ${ }^{5}$ are less than ideal molecular tools to study the $\mathrm{H}_{2} \mathrm{R}$. As $\left[{ }^{3} \mathrm{H}\right]$ cimetidine is reported to label an imidazole recognition site rather than the $\mathrm{H}_{2} \mathrm{R}^{2}$ and ranitidine as well as histamine suffer from low $\mathrm{H}_{2} \mathrm{R}$ affinity and potency, ${ }^{5,6}\left[{ }^{3} \mathrm{H}\right.$ ]tiotidine is frequently used as a radioligand, although it shows very high unspecific binding and is reported to address only a subpopulation of the $\mathrm{H}_{2} \mathrm{R}$. ${ }^{5}$ By contrast, the labeling of aminopotentidine with ${ }^{125}$ iodine resulted in a high affinity radioligand ([ ${ }^{125}$ I]iodoaminopotentidine, $\mathrm{gpH}_{2} \mathrm{R}$ : $K_{\mathrm{d}}$ value: 0.34 nM ) which was used e.g. for autoradiography of the $\mathrm{H}_{2} \mathrm{R}$ in human and rodent brain and heart as well as for saturation and kinetic binding studies. ${ }^{6,7}$ Although ${ }^{125}$ iodine labeled ligands have, compared to tritium labeled ligands, the advantage of a higher specific activity, their preparation and usage require higher safety precautions and the ligands can only be used for $4-5$ weeks after preparation. ${ }^{8}$


Figure 3.1. Structures of the endogenous ligand histamine and selected standard $\mathrm{H}_{2} \mathrm{R}$ antagonists. ${ }^{a}$ Kelley et al ${ }^{5}$, ${ }^{\mathrm{b}}$ Ruat et al ${ }^{6},{ }^{\mathrm{C}}$ Yellin et al ${ }^{9}$, ${ }^{\mathrm{d}}$ Preuss et $\mathrm{al}^{10}$, ${ }^{\mathrm{e}}$ Baumeister et al ${ }^{11}$.

An alternative to $\left[{ }^{[125}\right]$ ]iodoaminopotentidine is the recently published high-affinity tritium-labeled $\mathrm{H}_{2} \mathrm{R}$ antagonist [ ${ }^{3} \mathrm{H}$ ]UR-DE257 ( $K_{\mathrm{d}}$ value: 31 nM$)^{11}$, which is structurally related to the squaramide BMY25368. ${ }^{11,12}$ Although UR-DE257 was an insurmountable antagonist in functional assays, this radioligand proved to be very useful for the determination of $\mathrm{p} K_{\mathrm{i}}$ values. ${ }^{11}$

Aiming at the development of high affinity tritium-labeled surmountable $H_{2} R$ antagonists, guanidinothiazole-containing amines were considerd promising precursors. In previous functional studies, the antagonists with guanidinothiazole sub-structure, namely famotidine and ICI127032 were identified as surmountable $\mathrm{H}_{2} \mathrm{R}$ antagonists. ${ }^{9,13}$

The 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole precursor derived from famotidine and the 2-guanidino-4-(3-aminophenyl)thiazole precursor derived from ICI127032 were combined with the squaramide moiety ("urea" equivalent) of BMY25368 (general structure: Figure 3.2). In
order to enable radiolabeling, a terminal amino group connected to the squaramide via $n$-alkyl linker of different length, was introduced following the same strategy as in case of UR-DE257. The terminal amino group was propionylated using N -succinimidyl propionate to obtain "cold forms" of the potential radioligands. Furthermore, the squaramide moiety was replaced by a cyanoguanidine moiety resulting in tiotidine-like compounds.


Figure 3.2. General structure of the guanidinothiazoles as potential new radioligands.
Interestingly, linking two amino(methyl)thiazolepropyl containing acyl or carbamoyl guanidines resulted in high affinity $\mathrm{H}_{2} \mathrm{R}$ agonists (e.g. UR-NK22). ${ }^{14,15}$ Therefore, this strategy was also adapted to the guanidinothiazoles: The replacement of the amino(methyl)thiazolepropyl moiety of URNK22 with either 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole or 2-guanidino-4-(3aminophenyl)thiazole resulted in two bivalent ligands (Figure 3.3).


Figure 3.3. General structure of the dimeric guanidinothiazoles. ${ }^{a}$ Kagermeier et al ${ }^{14}$.
The amine precursors and the "cold forms" of the potentional radioligands as well as the two bivalent ligands were characterized in binding and functional (GTP ${ }^{\text {S }}$ binding assay) studies on recombinant histamine receptors (preferentially $\mathrm{hH}_{2} \mathrm{R}$ and $\mathrm{hH}_{3} \mathrm{R}$ ). Radiolabeling is accessible by coupling of the commercially available N -succinimidyl $\left[2,3-{ }^{3} \mathrm{H}\right]$ propionate with the respective amine precursor.

### 3.2 Results and Discussion

### 3.2.1 Chemistry

The synthesis of the amine building block 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole (3.3) according to published procedures is outlined in Scheme 3.1. ${ }^{16}$ Firstly, 2-guanidino-4chloromethylthiazole hydrochloride (3.2) was synthesized from amidinothiourea and 1,3dichloroacetone. Secondly, $\mathbf{3 . 2}$ was coupled with 2 -aminoethanthiole in a substitution reaction in the presence of sodium ethanolate to obtain 3.3 in excellent yield.


Scheme 3.1. Synthesis of 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole 3.3. Reagents and conditions: i) acetone, RT, 24 h, $56 \%$; ii) EtONa, EtOH, $0^{\circ} \mathrm{C}$ to RT, $24 \mathrm{~h}, 84 \%$.

The conformationally constrained amine building block 3.7 was synthesized from 1-(3-aminophenyl)ethan-1-one and amidinothiourea according to literature procedures (Scheme 3.2). ${ }^{17}$ After phthalimide protection 3.4 was brominated in alpha position and subsequently treated with amidinothiourea in order to form the protected building block 3.6 in a good yield of $70 \%$ (over two steps). Deprotection of the phthalimide group in a mixture of HCl and acetic acid afforded the conformationally constrained amine building block 3.7.


Scheme 3.2. Synthesis of 2-guanidino-4-(3-aminophenyl)thiazole 3.7. Reagents and conditions: i) Acetic acid, reflux, 2.5 h, $96 \%$; ii) $\mathrm{Br}_{2}, \mathrm{HBr}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{CHCl}_{3}, \mathrm{RT}, 0.5 \mathrm{~h}$, no purification; iii) $\mathrm{CH}_{3} \mathrm{CN}, \mathrm{EtOH}$, reflux, $5 \mathrm{~h}, 70 \%$; iv) HCl , acetic acid, reflux, 24 h, 52\%.

The synthetic route leading to the $\mathrm{H}_{2} \mathrm{R}$ antagonists 3.8 and ICI127032 (3.10) is depicted in Scheme 3.3 as described in literature ${ }^{17}$ with minor modifications. The formation of the thiourea 3.8 resulted from treatment of building block 3.7 with methylisothiocyanate. ICI127032 (3.10) was synthesized in two steps: Diphenylcyanocarbonimidate was first coupled with 3.7 and the resulting intermediate 3.9 was than treated with methylamine.


Scheme 3.3. Synthesis of the $\mathrm{H}_{2} \mathrm{R}$ antagonists 3.8 and ICI 127032 (3.10). ${ }^{17}$ Reagents and conditions: i) methylisothiocyanate, acetone, RT, ON, 53\%; ii) 2-propanol, RT, ON, no purification; iii) aq. methylamine solution, RT, ON, 55\%.

The synthesis of the cyanoguanidine-type guanidinothiazole $\mathbf{3 . 1 3}$ followed a three step route (Scheme 3.4). The building block 3.3 was treated with the cyanoguanidinylating reagent diphenylcyanocarbonimidate ${ }^{18}$ by analogy to published protocols. ${ }^{711}$ The resulting intermediate 3.11 was treated with an excess of 1,8 -diaminooctane in order to preferably form the monovalent ligand 3.12. The propionamide $\mathbf{3 . 1 3}$ were synthesized from $\mathbf{3 . 1 2}$ and N -succinimidyl propionate.


Scheme 3.4. Synthesis of the cyanoguanidine-type guanidinothiazole 3.13. Reagents and conditions: i) TEA, MeOH, RT, ON , no purification; ii) MeCN, $50^{\circ} \mathrm{C}, 3 \mathrm{~h}, 62 \%$; iii) TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, DMF, RT, $3 \mathrm{~h}, 22 \%$.

The synthesis of the squaramide-type guanidinothiazoles 3.24-3.26 was adopted from known procedures (Scheme 3.5). ${ }^{11,16}$ The reaction of precursor 3.3 and 3,4-diethoxycyclobut-3-ene-1,2dione gave the squaric acid ester amide $\mathbf{3 . 1 4}$ as an intermediate, which was treated with the respective mono-Boc protected diamine 3.15-3.17 to yield the tert-butyl carbamates 3.18-3.20. The mono-Boc protected amines were synthesized by treating an excess of the diamine with Di-tert-butyl dicarbonate. Final cleavage of the protecting group with TFA afforded the amines 3.213.23. The propionylated ligands $\mathbf{3 . 2 4 - 3 . 2 6}$ were accessible through acylation of the respective amine precursor 3.21-3.23 with N -succinimidyl propionate.


Scheme 3.5 Synthesis of the squaramide-type guanidinothiazoles 3.24-3.26. Reagents and conditions: i) TEA, EtOH, RT, ON, $87 \%$; ii) Di-tert-butyl dicarbonate, $\mathrm{CHCl}_{3}, 0^{\circ} \mathrm{C}$ to RT , $\mathrm{ON}, 41-63 \%$; iii) TEA, EtOH, $60-70{ }^{\circ} \mathrm{C}$ to $\mathrm{RT}, 6 \mathrm{~h}-2$ days, $63-$ $88 \%$; iv) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, RT, 1.5 h, 57-97\%; v) TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, RT, ON, 33-72\%.

The conformationally constrained squaramide-type guanidinothiazole 3.30 was synthesized by following the same protocol as in case of the compounds 3.24-3.26 (Scheme 3.6). Starting from precursor 3.9, treatment with 3,4-diethoxycyclobut-3-ene-1,2-dione gave the squaric acid ester amide intermediate 3.27, which was reacted with tert-butyl N -(8-aminooctyl)carbamate (3.16) to obtain the tert-butyl carbamate $\mathbf{3 . 2 8}$. Deprotection led to the amine precursor $\mathbf{3 . 2 9}$ which was propionylated with N -succinimidyl propionate.


Scheme 3.6 Synthesis of the conformationally constrained squaramide-type guanidinothiazole 3.30. Reagents and conditions: i) EtOH, RT, ON, no purification; ii) TEA, EtOH, $80{ }^{\circ} \mathrm{C}, \mathrm{ON}, 34 \%$; iii) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, 5 \mathrm{~h}, 61 \%$; iv) TEA, DMF, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 12 \%$.

The synthetic route leading to the bivalent guanidinothiazole containing ligands 3.34 and 3.35 was adopted from published protocols ${ }^{14}$ and is depicted in Scheme 3.7. Thiourea was methylated with methyl iodine and subsequently mono-Boc protected, resulting in the well-established guanidinylating reagent $\mathbf{3 . 3 2}$. Treatment of $\mathbf{3 . 3 2}$ with 1,6 -diisocyanohexane afforded the bivalent guanidinylating reagent 3.33. The bivalent ligands $\mathbf{3 . 3 4}$ and $\mathbf{3 . 3 5}$ were prepared by reaction of 3.33 with the respective guanidinothiazole building block ( $\mathbf{3 . 3}$ or $\mathbf{3 . 7}$ ) in the presence of a base, followed by preparative HPLC purification of the Boc-protected intermediates. The protecting group remained stable over the course of the purification, and deprotection was achieved by storage in the TFA consisting eluate for several hours, followed by a second preparative HPLC
purification. The addition of $\mathrm{Hg}(\mathrm{II})$ ions in guanidinylation reaction, as described in many published protocols ${ }^{14,19}$, led to many by-products and only traces of the desired product.


Scheme 3.7. Synthesis of the bivalent guanidinothiazole containing ligands $\mathbf{3 . 3 4}$ and 3.35. Reagents and conditions: i) Methyliodide, MeOH, reflux, $1.5 \mathrm{~h}, 100 \%$; ii) Di-tert-butyl dicarbonate, TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}$ to $\mathrm{RT}, \mathrm{ON}, 73 \%$; iii) TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, Ar-atmosphere, RT, ON, 88\%; iv)/v) DIPEA, MeOH, reflux, 5-48 h, deprotection after preparative HPLC in eluate consisting of $\mathrm{H}_{2} \mathrm{O}, \mathrm{MeCN}$ and TFA, $3-11 \%$ over two steps.

### 3.2.2 Biological Evaluation

## $H_{2} R$ affinity, selectivity and antagonism

The monovalent guanidinothiazole containing ligands 3.8, 3.10, 3.12, 3.13, 3.21-3.26, 3.29, 3.30, the bivalent guanidinothiazole containing ligands 3.34, $\mathbf{3 . 3 5}$ and exemplary published $\mathrm{H}_{2} \mathrm{R}$ ligands were investigated in equilibrium competition binding experiments on membrane preparations from Sf 9 insect cells, expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein, using either the antagonist [ $\left.{ }^{3} \mathrm{H}\right]$ URDE257 ${ }^{11}$ or $\left[{ }^{3} \mathrm{H}\right]$ tiotidine as radioligands. The selectivity of these compounds for the $h \mathrm{H}_{2} \mathrm{R}$ compared to $\mathrm{hH}_{3} \mathrm{R}$ was investigated by competition binding experiments using Sf 9 insect cell membranes co-expressing the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\alpha \mathrm{ai2}}$ and $\mathrm{G}_{\beta 1 \gamma^{2}}$ proteins using $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}^{\alpha}$-methylhistamine as radioligand. The "cold" form of the radioligand 3.25 , its precursor 3.22 and the bivalent ligands 3.34 and 3.35 were additionally investigated on membrane preparations of Sf9 insect cells, coexpressing either the $\mathrm{hH}_{1} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ mepyramine) and RGS 4 or $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ and $\mathrm{G}_{\beta 1{ }^{2} 2}$ proteins (radioligand: $\left[{ }^{3} \mathrm{H}\right]$ histamine or $\left[{ }^{3} \mathrm{H}\right]$ UR-PI294 ${ }^{20}$ ) in order to determine a "selectivity profile" at all four $\mathrm{H}_{2} \mathrm{R}$ subtypes. Selected displacement curves are shown in Figure 3.4 and the results are summarized in Table 3.1.

The cyanoguanidine-type guanidinothiazole containing amine precursor $\mathbf{3 . 1 2}$ showed lower $\mathrm{hH}_{2} \mathrm{R}$ affinity with a $\mathrm{p} K_{\mathrm{i}}$ value of 6.5 than the structurally related $\mathrm{H}_{2} \mathrm{R}$ antagonist famotidine ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.26). Propionylation was tolerated without any decrease of affinity (ligand 3.13). The $\mathrm{hH}_{2} \mathrm{R}$ affinity of the squaramide-type guanidinothiazole containing amine precursors 3.21-3.23 increased with the chain length of the $n$-alkandiyl linker ( $\mathrm{p} K_{\mathrm{i}}$ values: 5.94-8.42). The propionylated squaramide-type ligands 3.24-3.26 showed a high affinity at the $\mathrm{hH}_{2} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ values: 7.07-7.65). Interestingly, the affinities of $\mathbf{3 . 2 5}$ and $\mathbf{3 . 2 6}$ decreased and the affinity of $\mathbf{3 . 2 4}$ increased compared to the respective amine precursor. The conformationally constrained squaramide-type guanidinothiazole containing ligand 3.30 showed a decreased $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ value: 6.8) compared to the closely related, more flexible ligand 3.25 ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.65) and the cyanoguanidine-type analog ICI127032 (3.10) ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.70). The amine precursor 3.29 showed compared to $\mathbf{3 . 3 0}$ a clearly increased $\mathrm{p} K_{\mathrm{i}}$ value of 7.31 . Within the series of propionylated ligands ("cold" forms of potential radioligands) 3.25 showed the highest $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.65) which was comparable to the affinity of UR-DE257 ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.55).

The ligands 3.8, 3.21-3.26, 3.29 and 3.30 showed a low affinity at the $\mathrm{hH}_{3} \mathrm{R}$ with $\mathrm{pK}_{\mathrm{i}}$ values of $4.28-5.5$ and 3.10 showed a very low affinity ( $\mathrm{p} K_{\mathrm{i}}$ value $<4$ ). The precursor 3.22 and the "cold" form of the radioligand $\mathbf{3 . 2 5}$ showed a low affinity to the $\mathrm{hH}_{4} \mathrm{R}$ with $\mathrm{p} K_{\mathrm{i}}$ values of 4.3 and 4.4. 3.22 bound with moderate affinity to the $\mathrm{hH}_{1} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ value: 6.12), whereas binding of the propionylated 3.25 was not detectable up to a concentration of $100 \mu \mathrm{M}$.

The replacement of the amino(methyl)thiazolyl propyl head group of the bivalent ligand UR-NK22 with either a 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole residue (3.34) or a 2-guanidino-4( 3 aminophenyl)thiazole residue (3.35) resulted in high affinity $\mathrm{hH}_{2} \mathrm{R}$ ligands with $\mathrm{p} K_{\mathrm{i}}$ values of 7.3 and 7.14. Compared to the bivalent agonist UR-NK22 the $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ value: 8.07) ${ }^{14}$ and subtype selectivity was decreased.


Figure 3.4. Displacement of the respective radioligand by amine precursor $\mathbf{3 . 2 2}(\mathrm{A})$ or ligand $\mathbf{3 . 2 5}(\mathrm{B})$ from membrane preparations of Sf9 insect cells co-expressing the $\mathrm{hH}_{1} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein and RGS4 (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ mepyramine, $\mathrm{c}=$ $5 \mathrm{nM}, K_{\mathrm{d}}=4.5 \mathrm{nM}$ ), expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ), coexpressing the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ plus $\mathrm{G}_{\beta 1 y_{2} 2}$ proteins (radioligand: $\left[^{3} \mathrm{H}\right] \mathrm{N}^{\alpha}$-methylhistamine, $\mathrm{c}=3 \mathrm{nM}, K_{d}=3 \mathrm{nM}$ ) or coexpressing the $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\alpha i 2}$ plus $\mathrm{G}_{\beta 1 \gamma^{2}}$ proteins (radioligand: $\left[^{3} \mathrm{H}\right]$ UR-PI294, $\mathrm{c}=5 \mathrm{nM}, K_{\mathrm{d}}=5.1 \mathrm{nM}$ ). Displacement of the respective radioligand from membrane preparations of Sf9 insect cells (C) expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ or [ $\left.{ }^{3} \mathrm{H}\right]$ tiotidine, $\mathrm{c}=10 \mathrm{nM}, K_{\mathrm{d}}=12.75 \mathrm{nM}$ ) or co-expressing the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\alpha i 2}$ plus $\mathrm{G}_{\beta 1 y 2}$ proteins (radioligand: $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}^{\alpha}$-methylhistamine, $\mathrm{c}=3 \mathrm{nM}, K_{\mathrm{d}}=3 \mathrm{nM}$ ) by exemplary guanidinothiazoles. Data represent mean values $\pm$ SEM of 2-3 experiments performed in triplicate.

Table 3.1. Affinities of standard $\mathrm{H}_{2} \mathrm{R}$ ligands, UR-DE257, UR-NK22, the monovalent guanidinothiazole containing ligands 3.8, $\mathbf{3 . 1 0}, \mathbf{3 . 1 2}, \mathbf{3 . 1 3}, \mathbf{3 . 2 1 - 3 . 2 6}, \mathbf{3 . 2 9}, \mathbf{3 . 3 0}$ and the bivalent guanidinothiazole containing ligands $3.34,3.35$ to $\mathrm{hH}_{1-4} R$, obtained from equilibrium competition binding studies on membrane preparations from Sf9 insect cells, expressing the respective histamine receptor subtype.

| No. | $h_{1} \mathrm{R}^{\text {a }}$ |  | $\mathrm{hH}_{2} \mathrm{R}^{\text {b }}$ |  | $\mathrm{hH}_{3} \mathrm{R}^{\text {d }}$ |  | $\mathrm{hH}_{4} \mathrm{R}^{\text {e }}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{p} K_{\text {i }}$ | N | $\mathrm{p} K_{\text {i }}$ | N | $\mathrm{p} K_{\text {i }}$ | N | $\mathrm{p} K_{i}$ | N |
| His | - | - | $6.53 \pm 0.04$ | 3 | $7.8 \pm 0.1$ | 3 | $7.65 \pm 0.03$ | 3 |
| UR-DE257 | > $5.0^{11}$ | - | $7.55{ }^{11}$ | - | $5.42{ }^{11}$ | - | > $5.0^{11}$ | - |
| famotidine | - | - | $7.26 \pm 0.03$ | 2 | - | - | - | - |
| UR-NK22 | $6.06{ }^{14}$ | - | $8.07{ }^{14}$ | - | $5.94{ }^{14}$ |  | $5.69{ }^{14}$ | - |
| 3.8 | n.d. | - | $6.5 \pm 0.1$ | 3 | $4.28 \pm 0.02$ | 2 | n.d. | - |
| 3.10 | n.d. | - | $7.70 \pm 0.07$ | 3 | > 4.0 | 2 | n.d. | - |
| 3.12 | n.d. |  | $6.5 \pm 0.3$ | 3 | n.d. | - | n.d. | - |
| 3.13 | n.d. |  | $6.4 \pm 0.2$ | 3 | n.d. | - | n.d. | - |
| 3.21 | n.d. |  | $5.94 \pm 0.05$ | 3 | $4.95 \pm 0.03$ | 2 | n.d. | - |
| 3.22 | $6.12 \pm 0.08$ | 3 | $8.0 \pm 0.2$ | 3 | $4.87 \pm 0.09$ | 3 | $4.3 \pm 0.1^{f}$ | 3 |
| 3.23 | n.d. | - | $8.42 \pm 0.09^{\text {c }}$ | 4 | $5.5 \pm 0.1$ | 2 | n.d. | - |
| 3.24 | n.d. | - | $7.07 \pm 0.09$ | 3 | $5.2 \pm 0.1$ | 2 | n.d. | - |
| 3.25 | > 4.0 | 3 | $7.65 \pm 0.02$ | 3 | $5.3 \pm 0.1$ | 3 | $4.4 \pm 0.1{ }^{\text {f }}$ | 3 |
| 3.26 | n.d. | - | $7.43 \pm 0.02^{\text {c }}$ | 3 | $5.08 \pm 0.09$ | 3 | n.d. | - |
| 3.29 | n.d. | - | $7.31 \pm 0.07^{\text {c }}$ | 3 | $5.3 \pm 0.2$ | 2 | n.d. | - |
| 3.30 | n.d. | - | $6.8 \pm 0.1^{\text {c }}$ | 3 | $5.44 \pm 0.07$ | 3 | n.d. | - |
| 3.34 | n.d. | - | $7.3 \pm 0.1^{\text {c }}$ | 3 | $6.32 \pm 0.02$ | 4 | $6.21 \pm 0.04$ | 3 |
| 3.35 | n.d. | - | $7.14 \pm 0.05$ | 3 | $5.8 \pm 0.1$ | 3 | $6.09 \pm 0.06$ | 3 |

Competition binding assay on membrane preparations of $\mathrm{Sf9}$ insect cells: ${ }^{\text {a co-expression of the } \mathrm{hH}_{1} \mathrm{R}-\mathrm{G}_{\text {sas }} \text { fusion protein }}$ and RGS4 (radioligand: $\left.{ }^{3} \mathrm{H}\right]$ mepyramine, $\mathrm{c}=5 \mathrm{nM}, K_{d}=4.5 \mathrm{nM}$ ), ' expression of the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ or $\left.{ }^{\mathrm{c}}{ }^{3} \mathrm{H}\right]$ tiotidine, $\mathrm{c}=10 \mathrm{nM}, K_{\mathrm{d}}=12.75 \mathrm{nM}$ ), ${ }^{d}$ co-expression of the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\alpha \mathrm{ii}}$ and $\mathrm{G}_{\beta 1 \gamma_{2} 2}$ proteins (radioligand: $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}^{\alpha}$-methylhistamine, $\mathrm{c}=3 \mathrm{nM}, K_{\mathrm{d}}=3 \mathrm{nM}$ ) or ${ }^{\mathrm{e}}{ }^{\mathrm{co}}$-expression of the $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\text {di2 }}$ plus $\mathrm{G}_{\beta 1 \gamma_{2}}$ proteins (radioligand: $\left[^{3} \mathrm{H}\right]$ histamine $\mathrm{c}=10 \mathrm{nM}, K_{\mathrm{d}}=15.9 \mathrm{nM}$ or $\left.{ }^{\mathrm{f}}{ }^{3} \mathrm{H}\right]$ UR-PI294, $\mathrm{c}=5 \mathrm{nM}, K_{d}=$ 5.1 nM ). The incubation period was 60 min . Data were analyzed by nonlinear regression and were best fitted to fourparameter sigmoidal concentration-response curves. Data shown are means $\pm$ SEM of $N$ independent experiments, each performed in triplicate. The appreviation n.d. stands for not determined.

The monovalent guanidinothiazole containing ligands 3.8, 3.10, 3.12, 3.13, 3.21-3.26, 3.29, 3.30 and the bivalent guanidinothiazole containing ligands $3.34,3.35$ were investigated for $h H_{2} R$ agonism in the GTPץS binding assay on membrane preparations from Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. Ligands which exhibited no agonism were also investigated in the antagonistic mode versus histamine as agonist. Selected curves are shown in Figure 3.5 and the results are summarized in Table 3.2.

The investigated monovalent and bivalent guanidinothiazole containing ligands were identified as antagonists or inverse agonists in the GTP $\gamma$ S assay. All ligands were able to completely displace histamine, but only the $\mathrm{p} K_{\mathrm{b}}$ values of the ligands $\mathbf{3 . 8}, \mathbf{3 . 3 0}$ and 3.35 were in good agreement with the $\mathrm{p} K_{\mathrm{i}}$ values. The $\mathrm{p} K_{\mathrm{b}}$ values of the ligands 3.10, 3.22-3.26, 3.29 and 3.34 were considerably lower compared to the corresponding $\mathrm{p} K_{\mathrm{i}}$ values. Interestingly, propionylation of the amine precursors 3.22 and 3.29 ( $\mathrm{p} K_{\mathrm{b}}$ value: 6.95 and 6.68 ) was tolerated with nearly no change of the $\mathrm{p} K_{\mathrm{b}}$ values. Nonetheless, ligand 3.25 also showed the highest $\mathrm{p} K_{\mathrm{b}}$ value (7.04) within the series of propionylated ligands ("cold" forms of potential radioligands), which was only slightly decreased compared to UR-DE257 ( $\mathrm{p} K_{\mathrm{b}}$ value: 7.42 ) ${ }^{11}$.


Figure 3.5. Antagonism of the guanidinothiazole containing ligands (A) 3.22, 3.29, 3.34, 3.35 and (B) 3.24-3.26, 3.30 on $\mathrm{hH}_{2} \mathrm{R}$ determined in a GTP $\gamma S$ assay (antagonistic mode) on membrane preparations of Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {s } \alpha \text { s }}$ fusion protein. Histamine ( $1 \mu \mathrm{M}$ ) was used for stimulation. Data represent mean values $\pm$ SEM of 2-4 experiments performed in triplicate.

Table 3.2. $\mathrm{hH}_{2} \mathrm{R}$ antagonism and the calculated $\mathrm{p} K_{\mathrm{b}}$ values of UR-DE257, the monovalent guanidinothiazole containing ligands 3.8, 3.10, 3.12, 3.13, 3.21-3.26, 3.29, 3.30 and the bivalent guanidinothiazole containing ligands 3.34, 3.35 determined by a GTP $\gamma$ S assay.

| $\mathrm{hH}_{2} \mathrm{R}$ (GTPYS) |  |  |  |  | $\mathrm{hH}_{2} \mathrm{R}$ (GTP $\gamma$ S) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| No. | $\mathrm{pEC} C_{50}\left(\mathrm{p} K_{\mathrm{b}}\right)$ | N | $\alpha$ | No. | $\mathrm{p} E C_{50}\left(\mathrm{p} K_{\mathrm{b}}\right)$ | N | $\alpha$ |
| His | $5.80 \pm 0.06$ | 9 | 1.0 | 3.24 | $(6.6 \pm 0.2)$ | 3 | -0.16 |
| UR-NK22 | $8.03{ }^{14}$ | - | $0.92{ }^{14}$ | 3.25 | $(7.06 \pm 0.03)$ | 3 | -0.30 |
| UR-DE257 | $(7.42)^{11}$ | - | $0.08{ }^{11}$ | 3.26 | $(6.5 \pm 0.1)$ | 3 | -0.11 |
| 3.8 | $(6.9 \pm 0.1)$ | 4 | -0.23 | 3.29 | $(6.68 \pm 0.03)$ | 2 | -0.25 |
| 3.10 | $(7.0 \pm 0.2)$ | 3 | -0.33 | 3.30 | $(6.9 \pm 0.2)$ | 2 | -0.18 |
| 3.22 | $(6.95 \pm 0.03)$ | 3 | -0.06 | 3.34 | $(7.0 \pm 0.5)$ | 2 | -0.53 |
| 3.23 | $(7.5 \pm 0.2)$ | 4 | -0.07 | 3.35 | $(7.09 \pm 0.08)$ | 3 | -0.21 |

$\left[{ }^{35}\right.$ S]GTP $\gamma$ S assay performed with membrane preparations of $\overline{\mathrm{Sf} 9 \text { insect cells expressing the } \mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{s \alpha \mathrm{~s}} \text { fusion protein. }}$ The intrinsic activity ( $\alpha$ ) of histamine was set to 1.00 , and $\alpha$ values of investigated compounds were referred to this value. The $\mathrm{p} K_{\mathrm{B}}$ values of neutral antagonists were determined in the antagonist mode versus histamine ( $\mathrm{c}=1 \mu \mathrm{M}$ ) as agonist. Data represent mean values $\pm$ SEM of $N$ experiments, each performed in triplicate.

Discrepancies between $\mathrm{p} K_{\mathrm{i}}$ and $\mathrm{p} K_{\mathrm{b}}$ value of antagonists were observed by several groups. ${ }^{7,21}$ Possible explanation are different experimental setups which led to varying access to the $\mathrm{H}_{2} \mathrm{R}$ (e.g. guinea pig striatal membranes vs. intact isolated guinea pig heart) ${ }^{7}$ or, when a very similar setup was used (e.g. membrane preparations in both binding and functional studies), the use of different competitors (histamine vs. radiolabeled antagonist). ${ }^{21}$ Agonists and antagonists may stabilize different receptor conformations that exhibit different affinities for the investigated agonists/antagonists/inverse agonists. ${ }^{21}$ The antagonistic radioligand [ ${ }^{3} \mathrm{H}$ ]tiotidine was reported to bind only to a fraction of the functionally active $\mathrm{H}_{2} \mathrm{Rs}$. ${ }^{5}$

### 3.2.3 Chemical stability of $\mathbf{3 . 2 5}$

The ligand 3.25 showed the highest $\mathrm{hH}_{2} \mathrm{R}$ affinity and subtype selectivity and was a good candidate for radiolabeling. Therefore the chemical stability of the "cold" radioligand 3.25 was investigated under radioligand storage conditions ( $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O} ; 80: 20 ; \mathrm{v} / \mathrm{v}$ ) at a concentration of 15 $\mu \mathrm{M}$ at room temperature in the dark (Figure 3.6). After 7 days RP-HPLC analysis showed no decomposition.


Figure 3.6. RP-HPLC analysis ( $\lambda: 273 \mathrm{~nm}$ ) of 3.25 stock solution ( $t_{\mathrm{R}}=15.0 \mathrm{~min}$ ) in EtOH/ $\mathrm{H}_{2} \mathrm{O}(80: 20, \mathrm{v} / \mathrm{v})$ after different times of incubation.

### 3.2.4 Radiosynthesis

Based on the pharmacological and chemical stability data presented, squaramide-type guanidinothiazole 3.25 was also synthesized with a tritium label ( $\left[^{3} \mathrm{H}\right] 3.25 /\left[{ }^{3} \mathrm{H}\right]$ UR-SB69) (Scheme 3.8). For this purpuse, an excess of the amine precursor 3.22 was acylated with the commercially available N -succinimidyl $\left[2,3-{ }^{3} \mathrm{H}\right]$ propionate in the presence of DIPEA. After purification by HPLC, the radioligand $\left[{ }^{3} \mathrm{H}\right] 3.25$ was obtained in a radiochemical purity of $87 \%$ (Figure 3.7) with one impurity present. As the second peak ( $t_{\mathrm{R}}=14.3 \mathrm{~min}$, Figure 3.7 B ) is amounting to ca. $13 \%$ of the total peak area, determination of the specific activity of $\left.{ }^{3} \mathrm{H}\right] 3.25$ was not feasible. Therefore, the specific activity of $\left[{ }^{3} \mathrm{H}\right] 3.25$ was estimated based on the specific activity ( $2.41 \mathrm{TBq} / \mathrm{mmol}, 65.03$ $\mathrm{Ci} / \mathrm{mmol}$ ) of $\left[{ }^{3} \mathrm{H}\right] \cup \mathrm{R}-\mathrm{MK} 300^{19}$ prepared on the same day from the same lot of N -succinimidyl [ $2,3-$ ${ }^{3} \mathrm{H}$ ]propionate. In order to prevent oxidation of the radioligand, the antioxidant ascorbinic acid was added to the radioligand stock solution (final concentration: $6.69 \mu \mathrm{~mol} / \mathrm{L}\left[{ }^{3} \mathrm{H}\right] 3.25$ and 76.9 $\mu \mathrm{mol} / \mathrm{L}$ ascorbinic acid in EtOH/ $\mathrm{H}_{2} \mathrm{O}$ 80:20).


Scheme 3.8. Synthesis of the tritium-labeled squaramide-type guanidinothiazole $\left[{ }^{3} \mathrm{H}\right] \mathbf{3 . 2 5}$ ( $\left[{ }^{3} \mathrm{H}\right] \mathrm{UR}-\mathrm{SB} 69$ ). Reagents and conditions: i) DIPEA, anhydrous DMF, RT, 80 min , radiochemical yield: 64\%.


Figure 3.7. RP-HPLC analysis of the radioligand $\left[{ }^{3} \mathrm{H}\right] 3.25(\mathrm{~A})$ before and (B) after purification by RP-HPLC (conditions, see Experimental Section). UV chromatogram of [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ at 220 nm (black dashed line) and radiochromatogram of [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ (red line).

### 3.2.5 Biological Evaluation of $\left[{ }^{3} \mathrm{H}\right] 3.25$

## Determination of binding constants of $\left[{ }^{3} \mathrm{H}\right] 3.25$

The radioligand $\left[{ }^{3} \mathrm{H}\right] 3.25$ was characterized by saturation binding experiments on membrane preparations from Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein as well as on HEK293T$\mathrm{hH}_{2} \mathrm{R}-\mathrm{qs} 5$ and HEK293T-hH2R- $\beta$ Arr2 cells in suspension, both stably expressing the $\mathrm{hH}_{2} \mathrm{R}$. Additionally, kinetic binding experiments were performed on membrane preparations from $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. The results are summarized in Table 3.3. [ ${ }^{3} \mathrm{H}$ ]UR-SB69 bound in a saturable manner to both membranes and HEK cells (Figure 3.8).


Figure 3.8. Representative saturation isotherms (red line) of specific $\mathrm{hH}_{2} \mathrm{R}$ binding of $[3 \mathrm{H}] 3.25$ on (A) membrane preparations from Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein, (B) cell suspension of HEK293T-hH ${ }_{2} \mathrm{R}$-qs5 cells and (C) cell suspension of HEK293T-hH ${ }_{2} \mathrm{R}$ - $\beta$ Arr2 cells. Unspecific binding was determined in the presence of a 300 fold excess of famotidine. Specific binding was analyzed by a one-site binding equation. Error bars of specific binding and error bars of the Scatchard plot represent propagated errors calculated according to the Gaussian law. Error bars of total and unspecific binding represent the SEM. Experiments were performed in triplicate.

Unspecific binding was low when either membrane preparations or intact HEK293T-hH2R-qs5 cells were used ( $<17 \%$ at the $K_{d}$ value). Saturation binding studies performed with intact HEK293T-hH $\mathrm{H}_{2} \mathrm{R}$ - $\beta$ Arr2 cells resulted in considerably higher unspecific binding (<42\% at the $K_{\mathrm{d}}$
value). The specific binding versus [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ concentration was best fitted by nonlinear regression to a one-site binding model and the unspecific binding to a linear curve. The determined $K_{\mathrm{d}}$ values ( $15-22 \mathrm{nM}$ ) were similar to the $K_{\mathrm{i}}$ value of 23 nM determined with [ ${ }^{3} \mathrm{H}$ ]UR-DE257 on membrane preparations of Sf9 cells.

Table 3.3. $\mathrm{hH}_{2} \mathrm{R}$ binding characteristics of $\left[{ }^{3} \mathrm{H}\right] 3.25$ determined on membrane preparations from $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein, HEK293T-hH R -qs5 or HEK293T-hH $\mathrm{H}_{2} \mathrm{R}-\beta$ Arr2 cells at $25^{\circ} \mathrm{C}$.

| Receptor | [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ ( $\left.{ }^{3} \mathrm{H}\right]$ UR-SB69) |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | $K_{\text {i }}{ }^{\text {d }}$ | $K_{\text {d (sat) }}{ }^{\text {e }}$ | $K_{\mathrm{d}(\mathrm{kin})}{ }^{\text {f }}$ | $k_{\text {on }}{ }^{\text {g }}$ | $k_{\text {off }}{ }^{\text {n }}$ |
|  | [ nM ] | [ nM ] | [ nM ] | $\left[\mathrm{min}^{-1} \cdot \mathrm{nM}^{-1}\right]$ | $\left[\mathrm{min}^{-1}\right]$ |
| $\mathrm{hH}_{2} \mathrm{R}$ | $23 \pm 1^{\text {a }}$ | $15 \pm 1^{\text {a }} / 22 \pm 4^{\text {b }} /$ | $26.0 \pm 0.3^{\text {a }}$ | $0.00108 \pm 0.00001^{\text {a }}$ | $0.028 \pm 0.002^{\text {a }}$ |
|  |  | $19 \pm 4^{\text {c }}$ |  |  |  |

Radioligand binding assay determined on ${ }^{\text {a }}$ membrane preparations from Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein or cell suspension of ${ }^{\mathrm{b}} \mathrm{HEK} 293 \mathrm{~T}-\mathrm{hH} \mathrm{H}_{2} \mathrm{R}-\mathrm{qs} 5$ or ${ }^{\mathrm{C}} \mathrm{HEK} 293 \mathrm{~T}-\mathrm{hH}_{2} \mathrm{R}-\beta$ Arr2 cells. ${ }^{\mathrm{d}}$ Derived from competition binding with [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ (cf. Table 3.1). ${ }^{\mathrm{e}}$ Equilibrium dissociation constant determined by saturation binding. ${ }^{\mathrm{f}}$ Kinetically determined dissociation constant $K_{\mathrm{d}(\mathrm{kin})}=\mathrm{k}_{\mathrm{off}} / \mathrm{k}_{\mathrm{on}}$. ${ }^{\mathrm{g}}$ Association rate constant derived from nonlinear regression; calculated from $k_{\text {obs }}, k_{\text {off }}$ and the radioligand concentration ( $c=20 \mathrm{nM}$ ). ${ }^{h}$ Dissociation rate constant derived from nonlinear regression. Data represent means $\pm$ SEM from two to four independent experiments each performed in duplicate (association) or triplicate.

The association and dissociation curves of $\left[{ }^{3} \mathrm{H}\right] 3.25$ determined on membrane preparations of Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein are depicted in Figure 3.9. The association of the radioligand to the $\mathrm{hH}_{2} \mathrm{R}$ was complete after 80 min and could be described by a monophasic fit with a $k_{\text {on }}$ value of $0.00108 \mathrm{~min}^{-1} \cdot \mathrm{nM}^{-1}$. Also linearization of the association curve revealed a straight line ( $k_{\mathrm{ob}}$ value of $0.055 \mathrm{~min}^{-1}$ ). The dissociation of $\left.{ }^{3} \mathrm{H}\right] 3.25$ ( $\mathrm{c}=20 \mathrm{nM}, 90 \mathrm{~min}$ preincubation) in the presence of famotidine was incomplete, reaching a plateau at $23 \%$ of the initially bound radioligand. These data suggested a (pseudo)irreversible binding. ${ }^{22,23}$

Several GPCR ligands were reported to show a similar behavior in kinetic and functional experiments ${ }^{22-25}$ and several explanations were provided such as irreversible (covalent) binding to the receptor, ${ }^{26}$ a slow rate of dissociation from the receptor, ${ }^{22}$ a slow rate of interconversion between inactive and active receptor conformations, ${ }^{27}$ stabilization of an inactive ligand-specific receptor conformation, ${ }^{28,29}$ binding to a site distinct from the agonist binding site, ${ }^{30}$ internalization of the ligand-receptor-complex ${ }^{25}$. Nevertheless, the equilibrium dissociation constant $K_{\mathrm{d}(\mathrm{kin})}=26.0 \mathrm{nM}$, calculated from kinetics (nonlinear regression, $K_{\mathrm{d}}=k_{\text {off }} / k_{\mathrm{on}}$ ), were consistent with the $K_{\mathrm{d}}$ value obtained from saturation binding experiments (Table 3.3) and the pseudo-irreversible binding to the $\mathrm{hH}_{2} \mathrm{R}$ was far less pronounced compared to the radioligand [ ${ }^{3}$ H]UR-DE257 (plateau at $60-70 \%$ ). ${ }^{11}$


Figure 3.9. Association (A) and dissociation (B) kinetics of [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ determined at membrane preparations from Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein at room temperature. Association ( $\mathrm{c}=20 \mathrm{nM}$ ) to the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time (nonlinear regression: $k_{\text {obs }}=0.049 \mathrm{~min}^{-1}$ ). Inset: $\ln \left[B_{\text {eq }} /\left(B_{\text {eq }}-B_{t}\right)\right]$ versus time, $k_{\text {obs }}=$ slope $=0.055 \mathrm{~min}^{-1}$. Dissociation (preincubation: $90 \mathrm{~min}, \mathrm{c}=20 \mathrm{nM}$ ) in the presence of famotidine ( $\mathrm{c}=3 \mu \mathrm{M}$ ) from the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{\text {off }}=0.028 \mathrm{~min}^{-1}, t_{1 / 2}=25 \mathrm{~min}$, plateau $=23 \%$ ), Inset: $\ln \left[\left(B_{t}-B_{\text {plateau }}\right) /\left(B_{0}-B_{\text {plateau }}\right)\right]$ versus time, slope• $(-1)=k_{\text {off }}=0.027 \mathrm{~min}^{-1}$. Data represent means $\pm$ SEM from two independent experiments each performed either in duplicate (association) or triplicate (dissociation).

## Equilibrium competition binding experiments of $\left[{ }^{3} \mathrm{H}\right] 3.25$

$\left[{ }^{3} \mathrm{H}\right] 3.25$ was used in equilibrium competition binding experiments with membrane preparations of Sf9 cells (Table 3.4). [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ was completely displaceable by histamine and the standard $\mathrm{H}_{2} \mathrm{R}$ antagonists famotidine and ICI127032 (Figure 3.10). Interestingly, lamtidine could only displace $75 \%$ of $\left[{ }^{3} \mathrm{H}\right] 3.25$, but completely displace the radioligand $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257. The $\mathrm{p} K_{\mathrm{i}}$ values were consistently lower but the order remained the same compared to the ones determined with [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257.


Figure 3.10. Displacement of the radioligand [ $\left.{ }^{3} \mathrm{H}\right]$ UR-SB69 ( $\left[{ }^{3} \mathrm{H}\right] 3.25, \mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=15 \mathrm{nM}$ ) from membrane preparations of Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein by standard $\mathrm{H}_{2} \mathrm{R}$ ligands. Data represent mean values $\pm$ SEM of 3 experiments performed in triplicate.

Table 3.4. Affinities of the standard $\mathrm{H}_{2} \mathrm{R}$ ligands to $\mathrm{hH}_{2} \mathrm{R}$, obtained from equilibrium competition binding studies on membrane preparations from Sf9 insect cells, expressing the $\mathrm{hH}_{2} \mathrm{R}$ with the radioligands [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 and $\left[{ }^{3} \mathrm{H}\right] 3.25$.

## $\mathrm{p} K_{\mathrm{i}}$ values

|  | $\mathrm{p} K_{\mathrm{i}}$ values |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | [ ${ }^{3} \mathrm{H}$ ]UR-DE257 ${ }^{\text {a }}$ | N | [ $\left.{ }^{3} \mathrm{H}\right]$ UR-SB69 $\left({ }^{3} \mathrm{H} \text { ] } 3.25\right)^{\text {b }}$ | N |
| Histamine | $6.53 \pm 0.04$ | 3 | $5.77 \pm 0.08$ | 3 |
| Famotidine | $6.87{ }^{11}$ | - | $6.36 \pm 0.03$ | 3 |
| Lamtidine | $6.8 \pm 0.2$ | 3 | $6.22 \pm 0.07$ | 3 |
| IC1127032 (3.10) | $7.70 \pm 0.07$ | 3 | $7.26 \pm 0.03$ | 3 |

Competition binding assay performed on membrane preparations of Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. Radioligand: $\left.{ }^{\mathrm{a}}{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ or $\left.{ }^{\mathrm{b}}\left[{ }^{3} \mathrm{H}\right] 3.25, \mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=15 \mathrm{nM}\right)$. The incubation period was 60 min . Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means $\pm$ SEM of $N$ independent experiments, each performed in triplicate.

### 3.2.6 Chemical stability of $\left[{ }^{3} \mathrm{H}\right] 3.25$

The long-term stability of $\left[{ }^{3} \mathrm{H}\right] 3.25$ in stock solution was investigated by RP-HPLC (conditions, see experimental section). The radiochemical purity of $\left[{ }^{3} \mathrm{H}\right] 3.25$ after 15 month still amounted to $45 \%$ (Figure 3.11) with one major impurity present ( $t_{\mathrm{R}}=12.6 \mathrm{~min}$ ).


Figure 3.11. RP-HPLC analysis of the (A) radioligand stock solution $\left[{ }^{3} \mathrm{H}\right] 3.2515$ month after purification (UV chromatogram at 273 nm (black dashed line) and radiochromatogram (red line)) and (B) radioligand stock solution $\left.{ }^{3} \mathrm{H}\right] 3.2515$ month after purification (black dashed line) and stock solution of "cold" $\mathbf{3 . 2 5}$ in DMSO (blue line) (UVchromatograms at 273 nm ).

### 3.3 Experimental section

### 3.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 ( $0.040-0.063 \mathrm{~mm}$ ). Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution ( 0.8 g ninhydrin, 200 mL n-butanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA). Nuclear Magnetic Resonance ( ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR) spectra were recorded on a Bruker Avance-300 (7.05 T, $\left.{ }^{1} \mathrm{H}: 300 \mathrm{MHz},{ }^{13} \mathrm{C}: 75.5 \mathrm{MHz}\right)$, Avance-400 (9.40 T, ${ }^{1} \mathrm{H}$ : $400 \mathrm{MHz},{ }^{13} \mathrm{C}: 100.6 \mathrm{MHz}$ ), or Avance-600 ( $14.1 \mathrm{~T} ;{ }^{1} \mathrm{H}: 600 \mathrm{MHz},{ }^{13} \mathrm{C}: 150.9 \mathrm{MHz}$; cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in $\delta$ (ppm) relative to external standards. Multiplicities are specified with the following abbreviations: $s$ (singlet), d (doublet), t (triplet), q (quartet), qui (quintet), m (multiplet), br s (broad signal), as well as combinations thereof. In certain cases 2D-NMR techniques (COSY, HSQC, HMBC and NOESY) were used to assign ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ chemical shifts. High-resolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass Q-TOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K-2001 detector. A Nucleodur 100-5 C18 ( $250 \times 21 \mathrm{~mm}, 5 \mu \mathrm{~m}$, Macherey-Nagel, Dueren, Germany), a Kinetex XB-C18 100A ( $250 \times 21.2 \mathrm{~mm}, 5 \mu \mathrm{~m}$, Phenomenex, Aschaffenburg, Germany) and a YMC Triart C18 ( $150 \times 20$ $\mathrm{mm}, 5 \mu \mathrm{~m}, \mathrm{YMC}$ Europe GmbH , Dinslacken, Germany) served as RP-columns at a flow rate of 15 $\mathrm{ml} / \mathrm{min}$ at room temperature. In case of the cyanoguanidines 3.10, 3.12 and 3.13 the mobile phase consisted of mixtures of $\mathrm{CH}_{3} \mathrm{CN}$ and $0.1 \%$ aq. $\mathrm{NH}_{3}$. Mixtures of $\mathrm{CH}_{3} \mathrm{CN}$ and $0.05-0.1 \%$ aq. TFA were used as mobile phase for compounds 3.8, 3.21-3.26, 3.29, 3.30, 3.34, 3.35. A detection wavelength of 220 nm was used throughout. $\mathrm{CH}_{3} \mathrm{CN}$ was removed from the eluates under reduced pressure (final pressure: 80 mbar) at $45{ }^{\circ} \mathrm{C}$ prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a vacuubrand RZ 6 rotary vane vacuum pump). Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A ( $250 \times 4.6 \mathrm{~mm}, 5 \mu \mathrm{~m}$, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of $0.05 \%$ TFA in $\mathrm{CH}_{3} \mathrm{CN}(\mathrm{A})$ and $0.05 \%$ aq. TFA (B) were used as mobile phase. Helium degassing, room temperature, a flow rate of $0.8 \mathrm{~mL} / \mathrm{min}$ and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration: 100-500 $\mu \mathrm{M}$ ) were either prepared from stock solution ( 10 mM in DMSO) in a mixture of $\mathrm{CH}_{3} \mathrm{CN}$ and $\mathrm{H}_{2} \mathrm{O}$ corresponding to the initial eluent composition, or as a one to one mixture of the eluate (preparative HPLC) with Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, $30-$
$32 \mathrm{~min}: 80: 20-95: 5,32-42 \mathrm{~min}: 95: 5$ or gradient 3: 0-30 min: A/B 15:85-90:10, 30-35 min: 90:10. Microanalysis was performed on a Vario micro cube (Elementar, Langenselbold, Germany).

### 3.3.2 Experimental protocols and analytical data

The synthesis of amidinothiourea ${ }^{31}$ (3.1) and $N$-succinimidyl propionate ${ }^{32}$ were described elsewhere.

## 2-Guanidino-4-chloromethylthiazole hydrochloride (3.2) ${ }^{33}$

1,3-Dichloroacetone ( $1.08 \mathrm{~g}, 8.46 \mathrm{mmol}, 1 \mathrm{eq}$ ) dissolved in acetone ( 4 mL ) was added to a suspension of $3.1(1.0 \mathrm{~g}, 8.46 \mathrm{mmol}, 1 \mathrm{eq})$ in acetone ( 5 mL ). After approximately 10 min most of the solid dissolved and the supernatant turned yellow. The reaction mixture was stirred over night at room temperature. The product precipitated as the HCl salt. Separation by filtration and washing with acetone ( 5 mL ) afforded the product as a yellow solid ( $1.07 \mathrm{~g}, 56 \%$ ). Mp: 176.9$181.7^{\circ} \mathrm{C}$ decomposition (Lit. ${ }^{33} \mathrm{mp}$ : 191-193 ${ }^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.58\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{NH}_{3}\right.$ in MeOH 9:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}(300$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 4.74(\mathrm{~s}, 2 \mathrm{H}), 7.41(\mathrm{~s}, 1 \mathrm{H}), 8.39(\mathrm{~s}, 4 \mathrm{H}), 12.83(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(75.5$ $\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta(\mathrm{ppm}) 40.7,113.4,147.2,154.2,160.3$. HRMS: (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{ClN}_{4} \mathrm{~S}^{+}: 191.0153$, found: 191.0151. $\mathrm{C}_{5} \mathrm{H}_{7} \mathrm{ClN}_{4} \mathrm{~S} \cdot \mathrm{HCl}(190.65+36.46)$.

## 2-Guanidino-4-[(2-aminoethyl)thiomethyl]thiazole (3.3) ${ }^{33}$

2-Aminoethanthiole hydrochloride ( $500 \mathrm{mg}, 4.41 \mathrm{mmol}, 2 \mathrm{eq}$ ) dissolved in $\mathrm{EtOH}(5 \mathrm{~mL})$ were added dropwise to a sodium ethanolate solution (prepared from 250 mg Na in 8 mLEEOH ) at 0 ${ }^{\circ} \mathrm{C}$. The reaction mixture was stirred for 1.5 h at $0^{\circ} \mathrm{C} .3 .2(500 \mathrm{mg}, 2.20 \mathrm{mmol}, 1 \mathrm{eq})$ was added portion wise over 15 min at $0^{\circ} \mathrm{C}$. After stirring for overnight at room temperature, the solvent was evaporated under reduced pressure and the crude product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 90: 10$ isocratic). Removal of the solvent in vacuo afforded the product as yellow oil ( $430 \mathrm{mg}, 84 \%$ ). $R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{~N} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 9: 1\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 2.63-2.78(\mathrm{~m}, 4 \mathrm{H}), 3.56(\mathrm{~s}, 2 \mathrm{H}), 6.47(\mathrm{~s}, 1 \mathrm{H}), 6.85(\mathrm{br} \mathrm{s}$, $4 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}\right): \delta(\mathrm{ppm}) 32.6,35.2,41.3,106.8,149.6,159.0,175.8 . \mathrm{HRMS}:$ (ESI) $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{7} \mathrm{H}_{14} \mathrm{~N}_{5} \mathrm{~S}_{2}^{+}$: 232.0685, found: 232.0689. $\mathrm{C}_{7} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{~S}_{2}$ (231.34).

## 2-(3-Acetylphenyl)isoindoline-1,3-dione (3.4) ${ }^{17}$

1-(3-Aminophenyl)ethan-1-one ( $1 \mathrm{~g}, 7.40 \mathrm{mmol}, 1 \mathrm{eq}$ ) and phthalic anhydride ( $1.2 \mathrm{~g}, 8.14 \mathrm{mmol}$, $1.1 \mathrm{eq})$ were suspended in acetic acid ( 20 mL ). The reaction mixture was stirred under reflux for 2.5 h and part of the solvent was evaporated under reduced pressure. Water ( 10 mL ) was added and the resulting precipitate was filtered through a Buchner funnel. Removal of residual solvent in vacuo afforded the product as beige fine crystals ( $1.9 \mathrm{~g}, 96 \%$ ). Mp: $180-184.5^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.24$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 2.62(\mathrm{~s}, 3 \mathrm{H}), 7.58-7.68(\mathrm{~m}, 2 \mathrm{H}), 7.77-7.83(\mathrm{~m}$,
$2 \mathrm{H}), 7.92-8.00(\mathrm{~m}, 3 \mathrm{H}), 8.03-8.05(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 26.8,124.0,126.6$, 127.8, 129.5, 131.1, 131.6, 132.3, 134.7, 138.1, 167.1, 197.1. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{NO}_{3}{ }^{+}: 266.0812$, found: 266.0814. $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{NO}_{3}$ (265.27).

## 2-Guanidino-4-(3-phthalimidophenyl)thiazole hydrobromide (3.6) ${ }^{17}$

To a solution of 3.4 ( $4.36 \mathrm{~g}, 16.44 \mathrm{mmol}$, 1 eq ) in $\mathrm{CHCl}_{3}(50 \mathrm{~mL})$ was added HBr solution in acetic acid ( $45 \% \mathrm{w} / \mathrm{v}, 110 \mu \mathrm{~L}$ ) under stirring. Bromine ( $5.52 \mathrm{~g}, 34.52 \mathrm{mmol}, 2.1 \mathrm{eq}$ ) was added drop wise. The reaction mixture was stirred for 30 min at room temperature. Removal of the solvent in vacuo afforded the 3.5 as a white solid, which was applied to the next step without further purification. The crude 3.5 was dissolved in hot $\mathrm{CH}_{3} \mathrm{CN}(100 \mathrm{~mL})$ and poured in a hot solution of $3.1(1.94 \mathrm{~g}, 16.44 \mathrm{mmol}, 1 \mathrm{eq})$ in $\mathrm{EtOH}(100 \mathrm{~mL})$. The reaction mixture was stirred under reflux for 5 h . Removal of the solvent in vacuo afforded a beige solid, which was suspended in EtOAc and filtered through a Buchner funnel. 3.6 was afforded as a beige solid ( $5.12 \mathrm{~g}, 70 \%$ ). Mp: 311.9$315.4{ }^{\circ} \mathrm{C}$ decomposition (Lit. ${ }^{17} \mathrm{mp}:>300{ }^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.7\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{~N} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 6: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(400$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 7.43-7.45(\mathrm{~m}, 1 \mathrm{H}), 7.60(\mathrm{t}, 1 \mathrm{H}, J 7.83 \mathrm{~Hz}), 7.84(\mathrm{~s}, 1 \mathrm{H}), 7.92-7.95(\mathrm{~m}$, $2 \mathrm{H}), 7.97-8.00(\mathrm{~m}, 2 \mathrm{H}), 8.03-8.05(\mathrm{~m}, 2 \mathrm{H}), 8.25(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 12.04(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$, [D6]DMSO): $\delta(\mathrm{ppm}) 109.3,123.5,125.2,125.7,127.5,129.4,131.5,132.5,133.8,134.8,148.9$, 153.8, 160.0, 167.0. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S}^{+}: 364.0863$, found: 364.0866. $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}_{2} \mathrm{~S} \cdot \mathrm{HBr}(363.40+80.91)$.

## 2-Guanidino-4-(3-aminophenyl)thiazole (3.7) ${ }^{17}$

3.6 ( $5.09 \mathrm{~g}, 11.46 \mathrm{mmol}, 1 \mathrm{eq}$ ) was suspended in a mixture of concentrated hydrochloric acid ( 75 mL ) and acetic acid ( 75 mL ) and the reaction mixture was stirred under reflux overnight. The solvent was removed under reduced pressure and the residue was suspended in aqueous NaOH solution ( $2.5 \% \mathrm{w} / \mathrm{w}, 200 \mathrm{~mL}$ ). The resulting precipitate was filtered through a Buchner funnel and washed with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$. Recrystallisation from $\mathrm{CH}_{3} \mathrm{CN}$ and removal of residual solvent in vacuo afforded the product as a brown solid ( $1.40 \mathrm{~g}, 52 \%$ ). Mp: 219.3-223.1 ${ }^{\circ} \mathrm{C}$ decomposition (Lit. ${ }^{17} \mathrm{mp}$ : $\left.223-224{ }^{\circ} \mathrm{C}\right) . R_{\mathrm{f}}=0.4\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 5.04$ $(\mathrm{s}, 2 \mathrm{H}), 6.47-6.50(\mathrm{~m}, 1 \mathrm{H}), 6.92(\mathrm{~s}, 1 \mathrm{H}), 6.95-6.97(\mathrm{~m}, 1 \mathrm{H}), 7.00-7.02(\mathrm{~m}, 1 \mathrm{H}), 7.03-7.04(\mathrm{~m}, 1 \mathrm{H})$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 102.1,111.2,113.1,113.4,128.8,135.4,148.6,150.0$, 156.9, 174.7. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{~N}_{5} \mathrm{~S}^{+}$: 234.0808, found: 234.0807. $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{~N}_{5} \mathrm{~S}$ (233.29).

## 2-Guanidino-4-[3-(3-methylthioureido)phenyl]thiazole hydrotrifluoracetate (3.8) ${ }^{17}$

Methylisothiocyanate ( $32 \mathrm{mg}, 0.43 \mathrm{mmol}, 1 \mathrm{eq}$ ) was added to a stirring solution of 3.7 ( 100 mg , $0.43 \mathrm{mmol}, 1 \mathrm{eq}$ ) in acetone ( 2 mL ). The reaction mixture was stirred overnight at room temperature. The solvent was removed in vacuo and the residue was purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05\% aq. TFA 10:90-55:45, $t_{\mathrm{R}}=14.93 \mathrm{~min}$ ). The TFA-salt was obtained as a white solid ( 70 mg ; 53\%). Mp: $204.2{ }^{\circ} \mathrm{C}$ decomposition (Lit. ${ }^{17}$
oxalate mp: $219-222^{\circ} \mathrm{C}$, decomposition). $R_{\mathrm{f}}=0.25\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right)$. RP-HPLC (gradient 2, 220 nm ) (TFA-Salz): $95.8 \%\left(t_{\mathrm{R}}=13.67 \mathrm{~min}, k=3.7\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta$ (ppm) 2.92-2.93 (m, 3H), 7.32-7.33 (m, 1H), 7.36-7.39 (m, 1H), 7.65-7.66 (m, 1H), 7.69 (s, 1H), 7.81 (br s, 1H), 7.98 (s, 1H), 8.36 (br s, 4H), 9.65 (br s, 1H), 12.30 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, [D. ${ }_{6}$ ]DMSO): $\delta(p p m) 31.3,108.3,120.6,121.7,123.1,129.1,133.5,139.7,149.5,154.2,158.6$ (q, J $34 \mathrm{~Hz}, \mathrm{TFA})$, 160.6, 181.3. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{~N}_{6} \mathrm{~S}_{2}{ }^{+}: 307.0794$, found: 307.0797. $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{~N}_{6} \mathrm{~S}_{2} \cdot \mathrm{C}_{2} \mathrm{HF}_{3} \mathrm{O}_{2}$ (306.41+114.02).

## 4-[3-(2-Cyano-3-methylguanidino)phenyl]thiazole / ICI127032 (3.10) ${ }^{17}$

Diphenylcyanocarbonimidate ( $112 \mathrm{mg}, 0.47 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) was dissolved in 2-propanol ( 10 mL ). 3.7 ( $100 \mathrm{mg}, 0.43 \mathrm{mmol}, 1 \mathrm{eq}$ ) was added and the reaction mixture was stirred overnight at room temperature. The solvent was removed under reduced pressure, diethylether was added and solvent was again removed under reduced pressure. The solid (crude 3.9) was dissolved in aqueous methylamine solution ( $40 \% \mathrm{w} / \mathrm{w}, 5 \mathrm{~mL}$ ) and the reaction mixture was stirred overnight at room temperature. The solvent was removed in vacuo and the residue was purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : MeCN/0.1\% aq. $\mathrm{NH}_{3}$ 10:90-40:60, $t_{\mathrm{R}}=20.51 \mathrm{~min}$ ). $\mathbf{3 . 1 0}$ was obtained as a white solid ( 74.5 mg ; $55 \%$ ). Mp : $167.4-192.4^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.25$ $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} \mathrm{N}_{3}\right.$ in MeOH 9:1). RP-HPLC (gradient 3, 220 nm : $96.7 \%\left(t_{\mathrm{R}}=10.32 \mathrm{~min}, k=3.1\right) .{ }^{1} \mathrm{H}-$ NMR (300 MHz, [D ${ }_{6}$ ]DMSO): $\delta(\mathrm{ppm}) 2.78(\mathrm{~d}, 3 \mathrm{H}, J 4.48 \mathrm{~Hz}), 6.93(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 7.16-7.21(\mathrm{~m}, 3 \mathrm{H})$, 7.35 (t, 1H, J 7.97 Hz ), 7.60-7.63 (m, 1H), $7.71(\mathrm{~s}, 1 \mathrm{H}), 8.94(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, [D ${ }_{6}$ ]DMSO): $\delta$ (ppm) 28.8, 103.6, 117.5, 121.1, $122.0122 .8,129.0,135.6,138.0,148.8,157.1$, 158.9, 175.2. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{~N}_{8} \mathrm{~S}^{+}$: 315.1135, found: 315.1139. $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{~N}_{8} \mathrm{~S}$ (314.37).

## 1-(8-Aminooctyl)-2-cyano-3-(2-[2-(diaminomethyleneamino)thiazol-4ylmethylthio]ethyl)guanidine (3.12)

Triethylamine ( $200 \mathrm{mg}, 1.97 \mathrm{mmol}, 4 \mathrm{eq}$ ) was added to a suspension of 3.3 as hydrochloride ( 150 $\mathrm{mg}, 0.49 \mathrm{mmol}, 1 \mathrm{eq}$ ) in methanol ( 45 mL ). Diphenyl- N -cyanocarbonimidate ( $118 \mathrm{mg}, 0.49 \mathrm{mmol}$, $1 \mathrm{eq})$ was added and the reaction mixture was stirred overnight at room temperature. The solvent was partially removed under reduced pressure and $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ was added. The product was extracted with EtOAc ( $3 \times 15 \mathrm{~mL}$ ), the organic layers were combined and dried over sodium sulfate. Removal of the solvent in vacuo afforded 3.11 as yellow oil which was directly used for the next step. The crude 3.11 and octan-1,8-diamine ( $285 \mathrm{mg}, 1.97 \mathrm{mmol}, 3 \mathrm{eq}$ ) were suspended in $\mathrm{MeCN}(1 \mathrm{~mL})$ and the reaction mixture was stirred for 3 h at $50^{\circ} \mathrm{C}$. The solvent was removed in vacuo and the residue was purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 $\min : \mathrm{MeCN} / 1 \%$ aq. $\mathrm{NH}_{3} 15: 85-65: 35, t_{\mathrm{R}}=16.7 \mathrm{~min}$ ). 3.12 was obtained as white, highly hygroscopic solid (130 mg, 62.0\%). $R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 6: 1\right)$. RP-HPLC (gradient 2, $220 \mathrm{~nm}): 95.7 \%\left(t_{\mathrm{R}}=12.74 \mathrm{~min}, k=4.1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 1.35(\mathrm{brs}, 8 \mathrm{H}), 1.51-$ $1.57(\mathrm{~m}, 4 \mathrm{H}), 2.66-2.74(\mathrm{~m}, 4 \mathrm{H}) 3.18(\mathrm{t}, 2 \mathrm{H}, J 7.1 \mathrm{~Hz}), 3.39(\mathrm{t}, 2 \mathrm{H}, J 6.97 \mathrm{~Hz}), 3.68(\mathrm{~s}, 2 \mathrm{H}), 6.54(\mathrm{~s}$, $1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 22.1,27.6,27.7,30.2,30.28,30.33,31.8,32.7,41.9$,
42.0, 42.7, 107.0, 120.1, 149.5, 159.1, 161.1, 175.9. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{32} \mathrm{~N}_{9} \mathrm{~S}_{2}^{+}: 426.2217$, found: $426.2215 . \mathrm{C}_{17} \mathrm{H}_{31} \mathrm{~N}_{9} \mathrm{~S}_{2}(425.21)$.

## N-(8-[2-Cyano-3-(2-[2-(diaminomethylenamino)thiazol-4-ylmethylthio]ethyl)guanidine]octyl)propionamide (3.13)

Triethylamine ( $79 \mathrm{mg}, 0.78 \mathrm{mmol}, 3 \mathrm{eq}$ ) and N -succinimidyl propionate ( $34 \mathrm{mg}, 0.29 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) were added to a suspension of 3.12 ( $111 \mathrm{mg}, 0.26 \mathrm{mmol}, 1 \mathrm{eq}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 1 mL ) and DMF (1 mL ). The reaction mixture was stirred for 3 h at room temperature. The solvent was partially removed in vacuo and the residue was purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: $\mathrm{MeCN} / 1 \%$ aq. $\mathrm{NH}_{3} 15: 85-80: 20, t_{\mathrm{R}}=19.0 \mathrm{~min}$ ). 3.13 was obtained as white, hygroscopic solid ( $28 \mathrm{mg}, 22.3 \%$ ). $R_{\mathrm{f}}=0.4\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 6: 1\right)$. IR ( KBr ): 3355, 3330, 3180, 2930, 2850, 2160, 1635, 1585, 1555, 1460, 1355, 1250, 1170, 1000, 850, 640, $605 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): $95.9 \%\left(t_{\mathrm{R}}=16.82 \mathrm{~min}, k=5.7\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta$ (ppm) $1.12(\mathrm{t}, 3 \mathrm{H}, J 7.6 \mathrm{~Hz}), 1.34(\mathrm{br} \mathrm{s}, 8 \mathrm{H}), 1.47-1.57(\mathrm{~m}, 4 \mathrm{H}), 2.15-2.21(\mathrm{q}, 2 \mathrm{H}, J 7.7 \mathrm{~Hz}), 2.68(\mathrm{t}$, $2 \mathrm{H}, \mathrm{J} 6.9 \mathrm{~Hz}$ ), 3.13-3.19 (m, 4H), 3.38 (t, 2H, J 6.8 Hz ), 3.68 ( $\mathrm{s}, 2 \mathrm{H}), 6.54(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, [ $D_{6}$ ]DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 10.6,27.7,27.9,30.2,30.26$ (2C), 30.29, 30.4, 31.8, 32.7, 40.4, 42.0, 42.8, 107.0, 120.1, 149.5, 159.1, 161.1, 175.9, 177.0. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{20} \mathrm{H}_{36} \mathrm{~N}_{9} \mathrm{~S}_{2} \mathrm{O}^{+}$: 482.2479, found: 482.2484. Anal. calcd. for $\mathrm{C}_{20} \mathrm{H}_{35} \mathrm{~N}_{9} \mathrm{~S}_{2} \mathrm{O}$ : C 49.87, H 7.32, N 26.17, S 13.31, found: C 49.56, H 7.19, N 26.24, S 12.88. $\mathrm{C}_{20} \mathrm{H}_{35} \mathrm{~N}_{9} \mathrm{~S}_{2} \mathrm{O}$ (481.58).

## 2-(4-[2-(2-Ethoxy-3,4-dioxocyclobut-1-ene-1-ylamino)ethylthiomethyl]thiazol-2-yl)guanidine $(3.14)^{34}$

3,4-Diethoxycyclobut-3-ene-1,2-dione ( $246 \mathrm{mg}, 1.45 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) was dissolved in EtOH ( 15 mL ). 3.3 as hydrochloride ( $400 \mathrm{mg}, 1.32 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $533 \mathrm{mg}, 5.26 \mathrm{mmol}, 4 \mathrm{eq}$ ) dissolved in $\mathrm{EtOH}(15 \mathrm{~mL})$ were added drop wise under stirring. The reaction mixture was stirred over night at room temperature. The precipitated product was filtered off and washed with EtOH. Some product was still contained in the supernatant. EtOH was removed in vacuo and the residue was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3$ in $\mathrm{MeOH} 95: 5$ isocratic). The precipitated product and the column purified product were combined and the solvent was removed in vacuo. 3.14 was obtained as beige solid ( $408 \mathrm{mg}, 87 \%$ ). Mp: 187.5-189.9 ${ }^{\circ} \mathrm{C}$ decomposition (Lit. ${ }^{34} \mathrm{mp}>135{ }^{\circ} \mathrm{C}$ decomposition). $R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 6: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm})$ 1.34-1.39 (m, 3H), 2.63-2.68 (m, 2H), 3.46$3.48(\mathrm{~m}, 1 \mathrm{H}), 3.61(\mathrm{~s}, 2 \mathrm{H}), 3.67-3.68(\mathrm{~m}, 1 \mathrm{H}), 4.63-4.68(\mathrm{q}, 2 \mathrm{H}, \mathrm{J} 7.0 \mathrm{~Hz}), 6.47(\mathrm{~s}, 1 \mathrm{H}), 6.83(\mathrm{br} \mathrm{s}$, $3.7 \mathrm{H}), 8.66-8.85(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J} 75.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 15.5\left(-\mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 31.0(\mathrm{~d}$, $J 20 \mathrm{~Hz}$, thiazolyl-CH2), 31.4 (d, J $34 \mathrm{~Hz},-\mathrm{SCH}_{2} \mathrm{CH}_{2} \mathrm{NH}-$ ), 43.0 (d, J $68 \mathrm{~Hz},-\mathrm{SCH}_{2} \mathrm{CH}_{2} \mathrm{NH}-$ ), 68.8 (d, J 9 $\mathrm{Hz},-\mathrm{OCH}_{2} \mathrm{CH}_{3}$ ), 104.4 (d, J $10 \mathrm{~Hz}, \mathrm{C}^{5}$ thiazolyl), 147.4 (d, J $11 \mathrm{~Hz}, \mathrm{C}^{4}$ thiazolyl), 156.8 (( $\left.\left.\mathrm{NH}_{2}\right)_{2} \mathrm{C}=\mathrm{N}-\right)$, 172.4 (d, J 32 Hz , cyclobutendion), 175.4 ( $C^{2}$ thiazolyl), 176.7 (d, J 21 Hz , cyclobutendion), 182.0 (d, J $19 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}$ cyclobutendion), 189.2 (d, J $31 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}$ cyclobutendion). HRMS: (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{~N}_{5} \mathrm{~S}_{2} \mathrm{O}_{3}^{+}$: 356.0846, found: 356.0847. $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{~S}_{2} \mathrm{O}_{3}$ (355.43).

## General procedure for the synthesis of the Mono-Boc-protected diamines (3.15-3.17)

The corresponding alkane diamine (2 eq) was dissolved in chloroform ( 30 mL ). Di-tert-butyl dicarbonate ( 1 eq ) was dissolved in chloroform ( 25 mL ) and added drop wise over a period of 3 h under ice-cooling. The reaction mixture was allowed to warm up to room temperature while stirring overnight. Potentially precipitated alkane diamine was filtered off. The organic layer was washed with alkaline saturated NaCl solution ( 45 mL sat. aq. NaCl and 1 mL 1 M aq. NaOH ), saturated NaCl solution ( 45 mL ) and $\mathrm{H}_{2} \mathrm{O}(45 \mathrm{~mL})$. The organic layer was dried over sodium sulphate and concentrated by evaporation under reduced pressure. The residue was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 97.5: 2.5-90: 10$ ).

## tert-Butyl N -(2-aminoethyl)carbamate (3.15) ${ }^{35-38}$

Ethylendiamine ( $1.5 \mathrm{~g}, 25.0 \mathrm{mmol}, 2 \mathrm{eq}$ ) was treated with di-tert-butyldicarbonate ( $2.72 \mathrm{~g}, 12.5$ $\mathrm{mmol}, 1 \mathrm{eq}$ ) according to the general procedure. Removal of the solvent in vacuo afforded the product as slightly yellow oil ( $1.25 \mathrm{~g}, 62.5 \%$ ). $R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl} / 2 / 3 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 5: 1$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}(400$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.34(\mathrm{~s}, 2 \mathrm{H}), 1.38(\mathrm{~s}, 9 \mathrm{H}), 2.73(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.6 \mathrm{~Hz}), 3.10(\mathrm{~m}, 2 \mathrm{H}), 5.11(\mathrm{br} \mathrm{s}, 1 \mathrm{H}$ ) $)$. HRMS (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{7} \mathrm{H}_{17} \mathrm{~N}_{2} \mathrm{O}_{2}{ }^{+}$: 161.1285, found: 161.1284. $\mathrm{C}_{7} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}(160.22)$.

## tert-Butyl $N$-(8-aminooctyl)carbamate (3.16) ${ }^{38,39}$

Octane-1,8-diamine ( $2.0 \mathrm{~g}, 13.9 \mathrm{mmol}, 2 \mathrm{eq}$ ) was treated with di-tert-butyldicarbonate ( $1.5 \mathrm{~g}, 7.0$ $\mathrm{mmol}, 1 \mathrm{eq}$ ) according to the general procedure. Removal of the solvent in vacuo afforded the product as colorless oil ( $700 \mathrm{mg}, 41.2 \%$ ). $R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 9: 1$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.26(\mathrm{br} \mathrm{s}, 8 \mathrm{H}), 1.30(\mathrm{brs}, 2 \mathrm{H}), 1.38-1.41(\mathrm{~m}, 13 \mathrm{H}), 2.62-2.66(\mathrm{t}, 2 \mathrm{H} ; J 7.0 \mathrm{~Hz})$, 3.04-3.09 (m, 2H), 4.58 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 26.8,26.9,28.5,29.3,29.5$, 30.1, 33.9, 40.7, 42.3, 79.0, 156.1. HRMS (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{13} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{2}{ }^{+}: 245.2224$, found: 245.2225. $\mathrm{C}_{13} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{2}$ (244.38).

## tert-Butyl $N$-(10-aminodecyl)carbamate (3.17) ${ }^{38}$

Decane-1,10-diamine ( $2.0 \mathrm{~g}, 11.6 \mathrm{mmol}, 2 \mathrm{eq}$ ) was treated with di-tert-butyldicarbonate ( 1.3 g , $5.8 \mathrm{mmol}, 1 \mathrm{eq})$ according to the general procedure. Removal of the solvent in vacuo afforded the product as colorless oil ( $820 \mathrm{mg}, 51.2 \%$ ). $R_{\mathrm{f}}=0.6\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 7: 1$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.26(\mathrm{br} \mathrm{s}, 12 \mathrm{H}), 1.42-1.46(\mathrm{~m}, 13 \mathrm{H}), 1.67(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 2.67(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 7.1$ Hz ), 3.06-3.10 (m, 2H), 4.53 (br s, 1 H ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 26.9,27.0,28.6,29.4$, 29.55, 29.60, 29.62, 30.2, 33.7, 40.8, 42.3, 79.1, 156.1. HRMS (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{33} \mathrm{~N}_{2} \mathrm{O}_{2}{ }^{+}: 273.2537$, found: 273.2537. $\mathrm{C}_{15} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{2}$ (272.43).

## tert-Butyl [2-(2-[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1ylamino)ethyl]carbamate (3.18)

Compound 3.14 ( $150 \mathrm{mg}, 0.42 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in $\mathrm{EtOH}(10 \mathrm{~mL})$ and a solution of 3.15 $(74 \mathrm{mg}, 0.46 \mathrm{mmol}, 1.1 \mathrm{eq})$ in $\mathrm{EtOH}(10 \mathrm{~mL})$ was added under stirring. The reaction mixture was stirred over night at room temperature. Due to incomplete conversion triethylamine ( 214 mg , $2.11 \mathrm{mmol}, 5 \mathrm{eq}$ ) was added and the reaction mixture was stirred at $70^{\circ} \mathrm{C}$ for 6 h . The solution was allowed to cool to room temperature and stirred at this temperature over night. Removal of the solvent in vacuo and purification by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3$ in $\mathrm{MeOH} 95: 5-80: 20$ ) afforded the product as white solid ( $158 \mathrm{mg}, 79.2 \%$ ). Mp: 187.7-195.6 ${ }^{\circ} \mathrm{C}$ decomposition. $R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right): \delta(\mathrm{ppm})$ $1.36(\mathrm{~s}, 9 \mathrm{H}), 2.64-2.67(\mathrm{t}, 2 \mathrm{H}, J 6.7 \mathrm{~Hz}), 3.07-3.11(\mathrm{q}, 2 \mathrm{H}, J 5.9 \mathrm{~Hz}), 3.50(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.66-3.70(\mathrm{~m}$, $4 \mathrm{H}), 6.62(\mathrm{~s}, 1 \mathrm{H}), 6.88-7.09(\mathrm{br} \mathrm{s}, 5 \mathrm{H}), 7.58$ (br s, 2H). HRMS: (ESI): m/z [M+H] ${ }^{+}$, calcd. for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}_{2}{ }^{+}: 470.1639$, found: $470.1643 . \mathrm{C}_{18} \mathrm{H}_{27} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}_{2}$ (469.58).

## tert-Butyl [8-(2-[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1ylamino)octyl]carbamate (3.19)

3.14 ( $163 \mathrm{mg}, 0.46 \mathrm{mmol}, 1 \mathrm{eq}$ ), 3.16 ( $124 \mathrm{mg}, 0.50 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and triethylamine ( 232 mg , $2.29 \mathrm{mmol}, 5 \mathrm{eq}$ ) were dissolved in EtOH ( 20 mL ). The reaction mixture was stirred for 6 h at $70^{\circ} \mathrm{C}$. Removal of the solvent in vacuo and purification by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 95: 5-90: 10$ ) afforded the product as white solid ( $224 \mathrm{mg}, 88.1 \%$ ). Mp : $174.6-177.9{ }^{\circ} \mathrm{C}$ decomposition. $R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 9: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\right.$ DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 1.23-1.26(\mathrm{~m}, 8 \mathrm{H}), 1.33-1.36(\mathrm{~m}, 11 \mathrm{H}), 1.48-1.52(\mathrm{~m}$, $2 \mathrm{H}), 2.65-2.68(\mathrm{t}, 2 \mathrm{H}, J 6.6 \mathrm{~Hz}), 2.85-2.90(\mathrm{q}, 2 \mathrm{H}, J 6.6 \mathrm{~Hz}), 3.48(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.63(\mathrm{~s}, 2 \mathrm{H}), 3.69-3.71$ (m, 2H), $6.50(\mathrm{~s}, 1 \mathrm{H}), 6.72-6.75(\mathrm{~m}, 1 \mathrm{H}), 6.84(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 7.45(\mathrm{br} \mathrm{s}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\right.$ DMSO, COSY, HSQC, HMBC, NOESY): $\delta$ (ppm) 25.8, 26.2, 28.3, 28.6, 28.7, 29.4, 30.7, 31.1, 32.3, $40.2,42.6,43.3,77.3,104.5,147.5,155.6,156.9,167.5,167.8,175.3,182.3,182.5$. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}_{2}{ }^{+}$: 554.2578, found: 554.2584. $\mathrm{C}_{24} \mathrm{H}_{39} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}_{2}$ (553.74).
tert-Butyl [10-(2-[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1ylamino)decyl]carbamate (3.20)
3.14 ( $155 \mathrm{mg}, 0.41 \mathrm{mmol}, 1 \mathrm{eq}$ ), 3.17 ( $124 \mathrm{mg}, 0.45 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and triethylamine ( 209 mg , $2.06 \mathrm{mmol}, 5 \mathrm{eq})$ were suspended in EtOH ( 20 mL ). The reaction mixture was stirred overnight at $60{ }^{\circ} \mathrm{C}$. Removal of the solvent in vacuo and purification by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 95: 5$ isocratic) afforded the product as white solid ( $150 \mathrm{mg}, 63 \%$ ). Mp : 178.7-182.1 ${ }^{\circ} \mathrm{C}$ decomposition. $R_{\mathrm{f}}=0.45\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 6: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(300 \mathrm{MHz}$, [ $\mathrm{D}_{6}$ ]DMSO): $\delta(\mathrm{ppm}) 1.23-1.25(\mathrm{~m}, 14 \mathrm{H}), 1.36(\mathrm{~s}, 9 \mathrm{H}), 1.48-1.52(\mathrm{~m}, 2 \mathrm{H}), 2.65(\mathrm{t}, 2 \mathrm{H}, J 6.26 \mathrm{~Hz})$, 2.84-2.91 (m, 2H), $3.48(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.63(\mathrm{~s}, 2 \mathrm{H}), 3.69-3.71(\mathrm{~m}, 2 \mathrm{H}), 6.49(\mathrm{~s}, 1 \mathrm{H}), 6.74-6.84(\mathrm{~m}, 5 \mathrm{H})$, 7.47 (br s, 2H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(300 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 25.9,26.3,28.3,28.67,28.74,28.96$, 29.00, 29.5, 30.8, 31.1 32.3, 1C under solvent peak (38.7-40.3), 42.6, 43.3, 77.3, 104.5, 147.5,
155.6, 156.9, 167.5, 167.9, 175.5, 182.3, 182.5. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{26} \mathrm{H}_{44} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}_{2}{ }^{+}$: 582.2891, found: 582.2895. $\mathrm{C}_{26} \mathrm{H}_{43} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}_{2}$ (581.80).

## General procedure for the Boc-deprotection of 3.18-3.20 to the free amines 3.21-3.23

The Boc-protected amine $\mathbf{3 . 1 8 , 3 . 1 9}$ or $\mathbf{3 . 2 0}$ ( 1 eq ) was dissolved in a mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ and TFA ( 0.3 mL ) and stirred for 1.5 h at room temperature. The solvent was removed in vacuo, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added and then again removed. This process was repeated several times in order to remove residual TFA. Part of the product was directly used in the next step of the synthesis and a part was purified by preparative HPLC.

## 1-[4-(2-[2-(2-Aminoethylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]-ethylthiomethyl)thiazol-2-yl]guanidine-tri(hydrotrifluoracetate) (3.21)

3.18 ( $150 \mathrm{mg}, 0.32 \mathrm{mmol}, 1 \mathrm{eq}$ ) was treated according to the general procedure. The crude product was obtained as white hygroscopic solid ( 212 mg ). Further purification of 100 mg by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-30:70, $t_{\mathrm{R}}=$ 10.6 min ) afforded the product as white hygroscopic solid ( $54 \mathrm{mg}, 50.4 \%$ ). $R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M}\right.$ $\mathrm{NH}_{3}$ in MeOH 7:1). RP-HPLC (gradient 2, 220 nm ): $91.0 \% ~\left(t_{\mathrm{R}}=8.67 \mathrm{~min}, k=2.4\right.$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (400 $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 2.67$ (t, 2H, J 6.8 Hz ), 3.02 (br s, 2H), 3.69-3.71 (m, $4 \mathrm{H}), 3.79(\mathrm{~s}, 2 \mathrm{H}), 7.12(\mathrm{~s}, 1 \mathrm{H}), 7.78-7.90(\mathrm{br} \mathrm{s}, 5 \mathrm{H}), 8.40(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 12.42$ (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}(150$ $\mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 30.0,32.1,39.6,40.9,42.6,110.0,148.5,154.3$, 158.7 (q, J 32 Hz, TFA), 160.7, 167.9, 168.2, 182.4, 182.9. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{13} \mathrm{H}_{20} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}: 370.1114$, found: 370.1114. $\mathrm{C}_{13} \mathrm{H}_{19} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2} \cdot \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{9} \mathrm{O}_{6}(369.46+342.07)$.

## 1-[4-(2-[2-(8-Aminooctylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]ethylthiomethyl)thiazol-2-yl]guanidine-tri(hydrotrifluoracetate) (3.22)

3.19 ( $190 \mathrm{mg}, 0.34 \mathrm{mmol}, 1 \mathrm{eq}$ ) was treated according to the general procedure. The crude product was obtained as white hygroscopic solid ( 255 mg ). Further purification of 148 mg by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85-40:60, $t_{\mathrm{R}}=$ 11.0 min ) afforded the product as white hygroscopic solid ( $91 \mathrm{mg}, 57.3 \%$ ). $R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M}\right.$ $\mathrm{NH}_{3}$ in MeOH 7:1). RP-HPLC (gradient 1, 220 nm ): $95.4 \%\left(t_{\mathrm{R}}=13.22 \mathrm{~min}, k=4.3\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm})$ 1.23-1.27 (br s, 8H), 1.49-1.52 (m, 4H), 2.66-2.68 (t, 2H, J 6.8 Hz ), 2.74-2.79 (m, 2H), $3.50(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.70(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.79(\mathrm{~s}, 2 \mathrm{H}), 7.14(\mathrm{~s}, 1 \mathrm{H}), 7.69$ (br s, 5H), 8.46 (br s, 4H), 12.44 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta$ (ppm) 25.7, 26.9, 27.0, 28.37, 28.41, 30.0, 30.7, 32.2, 38.7, 42.5, 43.2, 110.2, 116.7 (q, J 297 Hz , TFA), 148.6, 154.2, 158.6 (q, J 33 Hz, TFA), 160.2, 167.6, 167.7, 182.3, 182.4. HRMS: (ESI): $\mathrm{m} / \mathrm{z}$ $[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{19} \mathrm{H}_{32} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}$: 454.2053, found: 454.2049. $\mathrm{C}_{19} \mathrm{H}_{31} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2} \cdot \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{9} \mathrm{O}_{6}(453.62+$ 342.07).

## 1-[4-(2-[2-(10-Aminodecylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]ethylthiomethyl)thiazol-2-yl]guanidine-tri(hydrotrifluoracetate) (3.23)

3.20 ( $50 \mathrm{mg}, 0.086 \mathrm{mmol}, 1 \mathrm{eq}$ ) was treated according to the general procedure. Purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85-70:30, $t_{\mathrm{R}}=$ 11.5 min ) afforded the product as white solid ( $60.4 \mathrm{mg}, 85.4 \%$ ). Mp : $98.1-100.8{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.2$ $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 6:1). RP-HPLC (gradient 2, 220 nm ): $96.8 \%\left(t_{\mathrm{R}}=13.82 \mathrm{~min}, k=4.5\right) .{ }^{1} \mathrm{H}-$ NMR (300 MHz, [D $\left.\mathrm{D}_{6}\right]$-DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 1.24$ (br s, 12H), 1.46-1.53 (m, 4H), 2.64-2.68 (m, 2H), 2.72-2.79 (m, 2H), 3.48 (br s, 2H), 3.69-3.71 (m, 2H), $3.79(\mathrm{~s}, 2 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H})$, 7.76 (br s, 5H), $8.59(\mathrm{~s}, 4 \mathrm{H}), 12.73$ (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]\right.$-DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 25.77,25.84,27.0,28.5,28.6,28.77,28.83,30.0,30.7,32.2,38.8,42.5,43.3,110.1$, 116.4 (q, J 296Hz, TFA), 148.5, 154.4, 159.1 (q, J 34 Hz, TFA), 160.3, 167.6, 167.9, 182.3, 182.5. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{21} \mathrm{H}_{36} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}$: 482.2366, found: 482.2364. Anal. calcd. for $\mathrm{C}_{21} \mathrm{H}_{35} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2} \cdot \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{9} \mathrm{O}_{6}$ : C 39.37, H 4.68, N 11.90, S 7.78, found: C 39.16, H 4.86, N 11.86, S 7.76. $\mathrm{C}_{21} \mathrm{H}_{35} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}_{2} \cdot \mathrm{C}_{6} \mathrm{H}_{3} \mathrm{~F}_{9} \mathrm{O}_{6}$ (481.68+342.07).

## N-[2-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1ylamino)ethyl]propionamide bis(hydrotrifluoracetate) (3.24)

3.21 ( $100 \mathrm{mg}, 0.17 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $68 \mathrm{mg}, 0.67 \mathrm{mmol}, 4 \mathrm{eq}$ ) were suspended in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 2.5 mL ). After stirring for 5 min N -succinimidyl propionate ( $32 \mathrm{mg}, 0.18 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) was added and the reaction mixture was stirred over night at room temperature. The conversion was incomplete due to the poor solubility of the educts in the solvent. DMF ( 2 mL ) was added and the mixture was stirred for 10 min at $50^{\circ} \mathrm{C}$ to dissolve the educts. The reaction mixture was allowed to cool to room temperature while stirring for additional 2 h . Removal of the solvent in vacuo and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-45:55, $t_{\mathrm{R}}=9.5 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $48 \mathrm{mg}, 52.8 \%$ ). $R_{\mathrm{f}}=0.3$ $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm ): $97.3 \%\left(t_{\mathrm{R}}=10.80 \mathrm{~min}, k=3.3\right) .{ }^{1} \mathrm{H}-$ NMR ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO): $\delta(\mathrm{ppm}) 0.97(\mathrm{t}, 3 \mathrm{H}, J 7.5 \mathrm{~Hz}), 2.03-2.09(\mathrm{q}, 2 \mathrm{H}, J 7.6 \mathrm{~Hz}), 2.67(\mathrm{t}, 2 \mathrm{H}$, $J 6.7 \mathrm{~Hz}), 3.19-3.21(\mathrm{~m}, 2 \mathrm{H}), 3.52(\mathrm{br} \mathrm{s}, 2 \mathrm{H}$, interfering with the water signal), 3.69-3.70(m,2H, interfering with the water signal), $3.79(\mathrm{~s}, 2 \mathrm{H}), 7.14(\mathrm{~s}, 1 \mathrm{H}), 7.59(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 7.92(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.38(\mathrm{br}$ $\mathrm{s}, 4 \mathrm{H}$ ), 12.29 (br s, 1 H ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]\right.$-DMSO): $\delta(\mathrm{ppm}) 9.8,28.4,30.0,32.1,39.7,42.5$, 42.9, 110.2, 116.8 (q, J 298 Hz, TFA), 148.6, 154.1, 158.8(q, J $33 \mathrm{~Hz}, \mathrm{TFA}$ ), 160.2, 167.6, 168.1, 173.3, 182.4, 182.6. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}_{2}{ }^{+}$: 426.1377, found: 426.1377. $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}_{2} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(425.53+228.04)$.

## N-[8-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1ylamino)octyl]propionamide bis(hydrotrifluoracetate) (3.25)

3.22 ( $100 \mathrm{mg}, 0.15 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $59 \mathrm{mg}, 0.59 \mathrm{mmol}, 4 \mathrm{eq}$ ) were suspended in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.5 \mathrm{~mL})$. After stirring for 5 min N -succinimidyl propionate ( $19 \mathrm{mg}, 0.16 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) was added and the reaction mixture was stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min:

MeCN/0.1\% aq. TFA 15:85-40:60, $t_{R}=14.6 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $66 \mathrm{mg}, 72 \%$ ). $R_{\mathrm{f}}=0.6\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 7: 1$ ). RP-HPLC (gradient $\left.2,220 \mathrm{~nm}\right): 98.8 \%\left(t_{\mathrm{R}}=\right.$ $15.21 \mathrm{~min}, k=5.1) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right): \delta(\mathrm{ppm}) 0.97(\mathrm{t}, 3 \mathrm{H}, J 7.6 \mathrm{~Hz}), 1.19-1.31(\mathrm{~m}$, $8 \mathrm{H}), 1.32-1.39(\mathrm{~m}, 2 \mathrm{H}), 1.45-1.55(\mathrm{~m}, 2 \mathrm{H}), 2.04(\mathrm{q}, 2 \mathrm{H}, J 7.6 \mathrm{~Hz}), 2.67(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 6.7 \mathrm{~Hz}), 2.97-3.02$ $(\mathrm{m}, 2 \mathrm{H}), 3.49(\mathrm{br} \mathrm{s}, 2 \mathrm{H}$, interfering with the water signal), $3.70(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.79(\mathrm{~s}, 2 \mathrm{H}), 7.14(\mathrm{br} \mathrm{s}$, 1 H ), 7.53 (br s, 2H), 7.68 (br s, 1H), 8.36 (br s, 4 H ), 12.24 (br s, 1 H ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\right.$ DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm})$ 10.0, 25.8, 26.4, 28.5, 28.6, 28.7, 29.1, 30.0, 30.7, 32.2, $38.4,42.5,43.3,110.3,116.6$ (q, J 296 Hz, TFA), 148.6, 154.1, 158.7 (q, J 33 Hz, TFA), 160.2, 167.5, 167.9, 172.6, 182.3, 182.4. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{22} \mathrm{H}_{36} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}_{2}{ }^{+}$: 510.2316, found: 510.2322. $\mathrm{C}_{22} \mathrm{H}_{35} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}_{2} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(509.69+228.04)$.

## N-[10-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)decyl]propionamide-bis(hydrotrifluoracetate) (3.26)

3.23 ( $144 \mathrm{mg}, 0.20 \mathrm{mmol}, 1 \mathrm{eq}$ ), N -succinimidyl propionate ( $26 \mathrm{mg}, 0.22 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and triethylamine ( $82 \mathrm{mg}, 0.81 \mathrm{mmol}, 4 \mathrm{eq}$ ) were dissolved in a mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3.5 \mathrm{~mL})$ and DMF (2 mL ). The reaction mixture was stirred for 2 h at room temperature. The conversion was incomplete due to the poor solubility of the educts in the solvent. The mixture was stirred for 1 h at $60^{\circ} \mathrm{C}$ to dissolve the educts. The reaction mixture was allowed to cool to room temperature while stirring for additional 48 h . Removal of the solvent in vacuo and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1\% aq. TFA 25:75-75:25, $t_{R}=12.08 \mathrm{~min}$ ) afforded the product as white fluffy solid ( $43 \mathrm{mg}, 32.5 \%$ ). Mp: 149.2-153.0 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.52\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7\right.$ $\mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 6: 1$ ). RP-HPLC (gradient $2,220 \mathrm{~nm}$ ): $98.8 \%\left(t_{\mathrm{R}}=18.05 \mathrm{~min}, k=6.2\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(400$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right): \delta(\mathrm{ppm}) 0.96(\mathrm{t}, 3 \mathrm{H}, \mathrm{J} 7.60 \mathrm{~Hz}), 1.22-1.25(\mathrm{~m}, 12 \mathrm{H}), 1.33-1.36(\mathrm{~m}, 2 \mathrm{H}), 1.48-1.49$ $(\mathrm{m}, 2 \mathrm{H}), 2.02(\mathrm{q}, 2 \mathrm{H}, J 7.60 \mathrm{~Hz}), 2.65-2.67(\mathrm{~m}, 2 \mathrm{H}), 2.97-3.00(\mathrm{~m}, 2 \mathrm{H}), 3.48(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.69(\mathrm{br} \mathrm{s}$, $2 \mathrm{H}), 3.78(\mathrm{~s}, 2 \mathrm{H}), 7.11(\mathrm{~s}, 1 \mathrm{H}), 7.59-7.67(\mathrm{~m}, 3 \mathrm{H}), 8.39(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 12.42$ (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}(150$ $\mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO): $\delta(\mathrm{ppm}) 10.0,25.8,26.4,28.5,28.6,28.7,28.90,28.91,29.1,30.0,30.7,32.2$, $38.4,42.5,43.2,110.0,116.9$ (q, J $298 \mathrm{~Hz}, \mathrm{TFA}$ ), 148.5, $154.2,159.2$ ( $q, J 32 \mathrm{~Hz}, \mathrm{TFA}$ ), 160.6, 167.5, 167.9, 172.6, 182.3, 182.4. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{24} \mathrm{H}_{40} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}_{2}{ }^{+}$: 538.2629, found: 538.2636. $\mathrm{C}_{24} \mathrm{H}_{39} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}_{2} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(537.74+228.04)$.

## tert-Butyl [8-(2-[3-(2-guanidinothiazol-4-yl)phenylamino]-3,4-dioxocyclobut-1-ene-1ylamino)octyl]carbamate (3.28)

3.7 ( $200 \mathrm{mg}, 0.86 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in $\mathrm{EtOH}(7 \mathrm{~mL})$ and poured into a solution of 3,4-diethoxy-3-cyclobutene-1,2-dione in $\mathrm{EtOH}(7 \mathrm{~mL})$. The reaction mixture was stirred over night at room temperature. The precipitated 3.27 was filtered off, washed with EtOH ( 2 mL ) and dried under vacuo.
3.27 ( $160 \mathrm{mg}, 0.45 \mathrm{mmol}, 1 \mathrm{eq}$ ), 3.16 ( $120 \mathrm{mg}, 0.49 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and triethylamine ( 227 mg , $2.24 \mathrm{mmol}, 5 \mathrm{eq})$ were suspended in $\mathrm{EtOH}(15 \mathrm{~mL})$. The reaction mixture was stirred over night at $80^{\circ} \mathrm{C}$ and subsequently cooled to room temperature. The precipitated product was filtered off and residual solvent was removed in vacuo. 3.28 was afforded as a beige solid ( $160 \mathrm{mg}, 33.6$ \%
over two steps). Mp: $262{ }^{\circ} \mathrm{C}$ decomposition. $R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 6: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(300$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right): \delta(\mathrm{ppm})$ 1.25-1.57 (m, 19H), 1.55-1.57 (m, 2H), 2.85-2.91 (m, 2H), 3.60-3.62 (m, $2 \mathrm{H}), ~ 6.76-6.79(\mathrm{~m}, 1 \mathrm{H}), 6.9(\mathrm{br} \mathrm{s}, 3.5 \mathrm{H}), 7.08-7.15(\mathrm{~m}, 2 \mathrm{H}), 7.32(\mathrm{t}, 1 \mathrm{H}, \mathrm{J} 7.84 \mathrm{~Hz}), 7.46-7.48(\mathrm{~m}$, $1 \mathrm{H}), 7.70(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.30(\mathrm{~s}, 1 \mathrm{H}), 9.67(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right): \delta(\mathrm{ppm}) 25.8,26.2$, 28.3, 28.6, 28.7, 29.5, 30.6, 30.7, 43.7, 77.3, 103.5, 115.2, 116.3, 119.2, 129.4, 136.1, 139.5, 148.9, 155.6, 157.1, 163.4, 169.3, 174.9, 180.1, 183.9. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{27} \mathrm{H}_{38} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}^{+}: 556.2700$, found: 556.2706. $\mathrm{C}_{27} \mathrm{H}_{37} \mathrm{~N}_{7} \mathrm{O}_{4} \mathrm{~S}(555.70)$.

## 1-[4-(3-[2-(8-Aminooctylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]phenyl)thiazol-2-yl]guanidine- hydrotrifluoracetate (3.29)

3.28 ( $140 \mathrm{mg}, 0.25 \mathrm{mmol}, 1 \mathrm{eq}$ ) was suspended in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 20 mL ). TFA ( $689 \mathrm{mg}, 6.05 \mathrm{mmol}, 24 \mathrm{eq}$ ) was added and the reaction mixture was stirred for 5 h at room temperature. Removal of the solvent in vacuo afforded the crude product as an off-white solid ( $140 \mathrm{mg}, 99 \%$ ). 50 mg were further purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05\% aq. TFA 15:85-65:35, $t_{R}=12.5 \mathrm{~min}$ ). $\mathbf{3 . 2 9}$ was obtained as a white fluffy solid ( $31 \mathrm{mg}, 61.3 \%$ ). Mp: 210.0-217.6 ${ }^{\circ} \mathrm{C}$ decomposition. $R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 6:1). RP-HPLC (gradient 2, 220 $\mathrm{nm}): 97.6 \%\left(t_{\mathrm{R}}=14.76 \mathrm{~min}, k=4.9\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm})$ 1.23-1.32 (m, 8H), 1.48-1.52 (m, 2H), 1.55-1.60 (m, 2H), 2.74-2.77 (m, 2H), 3.59-3.63 (m, 2H), 7.20-7.21 (m, 1H), 7.37 (t, 1H, J 7.86 Hz$), 7.54-7.55(\mathrm{~m}, 1 \mathrm{H}), 7.68(\mathrm{br} \mathrm{s}, 3 \mathrm{H}), 7.74(\mathrm{~s}, 1 \mathrm{H}), 8.12(\mathrm{br} \mathrm{s}$, 1 H ), 8.46 (br s, 4 H ), $8.55(\mathrm{~s}, 1 \mathrm{H}), 10.08(\mathrm{~s}, 1 \mathrm{H}), 12.53(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right.$, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 25.7,27.0,28.40,28.41,30.5,30.7,38.8,43.7,108.4,115.5,116.9$ ( $q, J 298 \mathrm{~Hz}, \mathrm{TFA}$ ), 117.2, 119.4, 129.8, 134.2, 139.9, 149.4, 154.2, 158.7 (q, J $32 \mathrm{~Hz}, \mathrm{TFA}$ ),160.2, 163.2, 169.6, 180.2, 183.7. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{22} \mathrm{H}_{30} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S}^{+}: 456.2176$, found: 456.2175. $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{~N}_{7} \mathrm{O}_{2} \mathrm{~S} \cdot \mathrm{C}_{2} \mathrm{HF}_{3} \mathrm{O}_{2}$ (455.58 + 114.02).

## N-[8-(2-[3-(2-Guanidinothiazol-4-yl)phenylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]propionamide-hydrotrifluoracetate (3.30)

3.29 ( $110 \mathrm{mg}, 0.16 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in a mixture of triethylamine ( $65 \mathrm{mg}, 0.64 \mathrm{mmol}, 4$ eq), DMF ( 1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$. N -Succinimidyl propionate ( $21 \mathrm{mg}, 0.18 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) was added and the reaction mixture was stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: $\mathrm{MeCN} / 0.05 \%$ aq. TFA 25:75-70:30, $t_{\mathrm{R}}=13.1 \mathrm{~min}$ ) afforded the product as white fluffy solid (12 $\mathrm{mg}, 11.6 \%$ ). Mp: 224-236.4 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.67\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 6:1). RP-HPLC (gradient 2, 220 $\mathrm{nm}): 98.5 \%\left(t_{\mathrm{R}}=18.85 \mathrm{~min}, k=6.5\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]\right.$-DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm})$ $0.96(\mathrm{t}, 3 \mathrm{H}, J 7.62 \mathrm{~Hz}), 1.22-1.58(\mathrm{~m}, 10 \mathrm{H}), 1.55-1.58(\mathrm{~m}, 2 \mathrm{H}), 2.02(\mathrm{q}, 2 \mathrm{H}, J 7.62 \mathrm{~Hz}), 2.97-3.01(\mathrm{~m}$, $2 \mathrm{H}), 3.60-3.61(\mathrm{~m}, ~ 2 \mathrm{H}), 7.13-7.15(\mathrm{~m}, 1 \mathrm{H}), 7.37(\mathrm{t}, 1 \mathrm{H}, J 7.92 \mathrm{~Hz}), 7.54-7.55(\mathrm{~m}, 1 \mathrm{H}), 7.67-7.71(\mathrm{~m}$, 2H), 7.91 (br s, 1H), 8.29 (br s, 4H), $8.56(\mathrm{~s}, 1 \mathrm{H}), 9.87(\mathrm{~s}, 1 \mathrm{H}), 12.26$ (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, [ $D_{6}$ ]-DMSO, COSY, HSQC, HMBC): $\delta$ (ppm) 10.0, 25.8, 26.3, 28.48, 28.54, 28.6, 29.1, 30.5, 38.3, 43.7, 108.3, 115.4, 117.1, 119.4, 129.8, 134.4, 139.8, 149.3, 154.2, 162.8, 163.1.169.5, 172.6, 180.2, 183.8. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{25} \mathrm{H}_{34} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S}^{+}$: 512.2438, found: 512.2443. $\mathrm{C}_{25} \mathrm{H}_{33} \mathrm{~N}_{7} \mathrm{O}_{3} \mathrm{~S} \cdot \mathrm{C}_{2} \mathrm{HF}_{3} \mathrm{O}_{2}(511.65+114.02)$.

## S-Methylthiouronium iodide (3.31) ${ }^{40}$

Methyliodide ( $9.30 \mathrm{~g}, 65.50 \mathrm{mmol}, 1 \mathrm{eq}$ ) was added to a stirring solution of thiourea ( $5 \mathrm{~g}, 65.50$ $\mathrm{mmol}, 1 \mathrm{eq}$ ) in $\mathrm{MeOH}(50 \mathrm{~mL})$. The reaction mixture was stirred under reflux for 1.5 h . The solvent was removed in vacuo and the residue was washed two times with diethyl ether ( 50 mL ). 3.31 was afforded as white solid ( $14.30 \mathrm{~g}, 99.9 \%$ ). Mp: 117.2-118.1 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.1$ (PE/EtOAc $3: 1$ ). ${ }^{1} \mathrm{H}-$ NMR (300 MHz, [D6]-DMSO): $\delta(\mathrm{ppm}) 2.56(\mathrm{~s}, 3 \mathrm{H}), 8.89(\mathrm{br} \mathrm{s}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right)$ : $\delta(\mathrm{ppm})$ 13.2, 171.0. HRMS: (ESI): $m / z[M+]^{+}$, calcd. for $\mathrm{C}_{2} \mathrm{H}_{7} \mathrm{~N}_{2} \mathrm{~S}^{+}$: 91.0324, found: 91.0327. $\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{~S} \cdot \mathrm{HI}(90.14+127.91)$.

## N-tert-Butoxycarbonyl-S-methylisothiourea (3.32) ${ }^{40}$

3.31 ( $13.65 \mathrm{~g}, 62.45 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $6.32 \mathrm{~g}, 62.45 \mathrm{mmol}, 1 \mathrm{eq}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(110 \mathrm{~mL})$ and cooled to $0{ }^{\circ} \mathrm{C}$. Di-tert-butyl dicarbonate ( $13.63 \mathrm{~g}, 62.45 \mathrm{mmol}, 1 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$ was added drop wise at $0{ }^{\circ} \mathrm{C}$. The reaction mixture was allowed to warm to room temperature and was stirred over night. The organic layer was washed two times with water $(200 \mathrm{~mL})$. Product which passed in the aqueous layer was extracted two times with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 100 mL ). The combined organic layers were dried over sodium sulfate. Removal of the solvent in vacuo and purification by column chromatography (eluent: PE/EtOAc 6:1-3:1) afforded the product as white solid ( $8.64 \mathrm{~g}, 72.6 \%$ ). Mp: 88.4-90.3 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.45$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.44(\mathrm{~s}, 9 \mathrm{H}), 2.38(\mathrm{~s}, 3 \mathrm{H}), 7.50(\mathrm{br} \mathrm{s}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ (ppm) 13.4, 28.1, 79.9, 172.9. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{7} \mathrm{H}_{15} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}^{+}$: 191.0849 , found: 191.0847. $\mathrm{C}_{7} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}$ (190.26).

## $N^{1}, N^{6}$-Bis([(tert-butoxycarbonylamino)(methylsulfanyl)methylene]aminocarbonyl)hexane-1,6diamine (3.33) ${ }^{14}$

3.32 ( $4.03 \mathrm{~g}, 21.1 \mathrm{mmol}, 2.2 \mathrm{eq}$ ) and triethylamine ( $0.97 \mathrm{~g}, 9.62 \mathrm{mmol}$, 1) were dissolved in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 17 mL ). 1,6-Diisocyanohexane ( $1.62 \mathrm{~g}, 9.6 \mathrm{mmol}, 1 \mathrm{eq}$ ) was added under Aratmosphere. The reaction mixture was stirred over night at room temperature. Removal of the solvent in vacuo and purification by column chromatography (eluent: PE/EtOAc 3:1-1:3) afforded the product as white solid ( $4.65 \mathrm{~g}, 88.0 \%$ ). Mp: $128.8-133.6^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.14$ (PE/EtOAc $3: 1$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.36-1.40(\mathrm{~m}, 4 \mathrm{H}), 1.48(\mathrm{~s}, 18 \mathrm{H}), 1.51-1.57(\mathrm{~m}, 4 \mathrm{H}), 2.30(\mathrm{~s}, 6 \mathrm{H}), 3.19-$ $3.24(\mathrm{~m}, 4 \mathrm{H}), 5.53(\mathrm{~m}, 2 \mathrm{H}), 12.30(\mathrm{~s}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.4,26.7,28.1$, 29.7, 40.1, 82.7, 151.2, 162.1, 167.4. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{22} \mathrm{H}_{41} \mathrm{~N}_{6} \mathrm{O}_{6} \mathrm{~S}_{2}{ }^{+}: 549.2524$, found: 549.2526. $\mathrm{C}_{22} \mathrm{H}_{40} \mathrm{~N}_{6} \mathrm{O}_{6} \mathrm{~S}_{2}$ (548.72).

## 1-(Amino[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]methylene)-3-(6-[3-(amino[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]methylene)ureido]hexyl)urea tetra(hydrotrifluoracetate) (3.34)

The HCl salt of 3.3 ( $244 \mathrm{mg}, 0.80 \mathrm{mmol}, 2.2 \mathrm{eq}$ ), 3.33 ( $200 \mathrm{mg}, 0.36 \mathrm{mmol}, 1 \mathrm{eq}$ ) and diisopropylethylamine ( $233 \mathrm{mg}, 1.80 \mathrm{mmol}, 5 \mathrm{eq}$ ) were suspended in $\mathrm{MeOH}(7 \mathrm{~mL})$. The reaction mixture was stirred under reflux for 5 h . The solvent was removed under reduced pressure and the residue was purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05\% aq. TFA 15:85-80:20, $t_{R}=15.2 \mathrm{~min}$ ). Removal of the MeCN in vacuo and lyophilisation afforded the Boc-deprotected 3.34 as a white solid. The residue was again purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05\% aq. TFA 10:90-65:35, $t_{\mathrm{R}}=12.7 \mathrm{~min}$ ). 3.34 was obtained as a white solid ( $45 \mathrm{mg}, 10.7 \%$ ). Mp: 59.1-61.9 ${ }^{\circ} \mathrm{C}$. RP-HPLC (gradient 2, 220 nm ): $96.2 \%\left(t_{\mathrm{R}}=12.95 \mathrm{~min}, k=4.2\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}\right.$, $\mathrm{HMBC}): \delta(\mathrm{ppm}) 1.25-1.26(\mathrm{~m}, 4 \mathrm{H}), 1.41-1.43(\mathrm{~m}, 4 \mathrm{H}), 2.66(\mathrm{t}, 4 \mathrm{H}, J 6.78 \mathrm{~Hz}), 3.06-3.09(\mathrm{~m}, 4 \mathrm{H})$, 3.44-3.47 ( $\mathrm{m}, 4 \mathrm{H}$, interfering with the water signal), $3.79(\mathrm{~s}, 4 \mathrm{H}), 7.13(\mathrm{~s}, 2 \mathrm{H}), 7.56(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 8.49$ (br s, 2H), 9.13 (br s, 2H), 10.34 (br s, 2H), 12.54 (br s, 2H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}, \mathrm{COSY}\right.$, HSQC, HMBC): $\delta(\mathrm{ppm}) 25.9,28.8,29.4,30.0,40.2,53.5,110.1,116.9$ (q, J $299 \mathrm{~Hz}, \mathrm{TFA}$ ), 148.5, 153.7, 153.8, 154.3, 159.2 ( $q, J 32 \mathrm{~Hz}, \mathrm{TFA}$ ), 160.5. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{24} \mathrm{H}_{43} \mathrm{~N}_{16} \mathrm{O}_{2} \mathrm{~S}_{4}{ }^{+}$: 715.2632, found: 715.2627. $\mathrm{C}_{24} \mathrm{H}_{42} \mathrm{~N}_{16} \mathrm{O}_{2} \mathrm{~S}_{4} \cdot \mathrm{C}_{8} \mathrm{H}_{4} \mathrm{~F}_{12} \mathrm{O}_{8}$ (714.95 + 456.09).

## 1-(Amino[3-(2-guanidinothiazol-4-yl)phenylamino]methylene)-3-(6-[3-(amino[3-(2-guanidino-thiazol-4-yl)phenylamino]methylene)ureido]hexyl)urea tetra(hydrotrifluoracetate) (3.35)

3.7 ( $280 \mathrm{mg}, 1.21 \mathrm{mmol}, 2.2 \mathrm{eq}$ ) and $3.33(300 \mathrm{mg}, 0.55 \mathrm{mmol}, 1 \mathrm{eq})$ were suspended in MeOH ( 15 mL ). The reaction mixture was stirred under reflux for 48 h . The solvent was removed under reduced pressure and the residue was purified by preparative HPLC (column: Nucleodur, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.05 \%$ aq. TFA $15: 85-80: 20, t_{\mathrm{R}}=19.1 \mathrm{~min}$ ). Removal of the MeCN in vacuo and lyophilisation afforded the Boc-deprotected 3.35 as a white solid. The residue was again purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05\% aq. TFA 15:85-45:55, $t_{R}=15.3 \mathrm{~min}$ ). 3.35 was obtained as a white solid ( $22 \mathrm{mg}, 3.4 \%$ ). Mp: $201.1^{\circ} \mathrm{C}$. RP-HPLC (gradient 2, 220 nm ): 98.5\% ( $t_{\mathrm{R}}=14.54 \mathrm{~min}, k=4.8$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\mathrm{DMSO}\right.$, COSY, HSQC, HMBC): $\delta(\mathrm{ppm})$ 1.26-1.29 (m, 4H), 1.42-1.45 (m, 4H), 3.10-3.13 (m, 4H), 7.30-7.32 (m, 2H), 7.53-7.55 (t, 2H, J 7.85 Hz ), $7.65(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 7.82(\mathrm{~s}, 2 \mathrm{H}), 7.92(\mathrm{~s}, 2 \mathrm{H}), 7.96-7.97(\mathrm{~m}, 2 \mathrm{H})$, 8.47-8.93 (m, 12H), 10.36 (br s, 2H), 10.81 (br s, 2H), 12.63 (br s, 2H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]-\right.$ DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 25.9,28.8,1 \mathrm{C}$ under solvent peak (38.7-40.3), 109.1, 115.9, 116.9 (q, J 297 Hz, TFA), 123.4, 125.3, 125.7, 130.3, 134.3, 134.8, 148.9, 153.6, 154.2, 159.3 (q, J 31 Hz, TFA), 161.0. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{30} \mathrm{H}_{39} \mathrm{~N}_{16} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}$: 719.2878, found: 719.2880. $\mathrm{C}_{30} \mathrm{H}_{38} \mathrm{~N}_{16} \mathrm{O}_{2} \mathrm{~S}_{2} \cdot \mathrm{C}_{8} \mathrm{H}_{4} \mathrm{~F}_{12} \mathrm{O}_{8}$ (718.86 + 456.09).

## N-[8-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]-[2,3- $\left.{ }^{3} \mathrm{H}\right]$ propionamide-hydrotrifluoracetate $\left(\left[{ }^{3} \mathrm{H}\right] 3.25\right)$

The radioligand $\left[{ }^{3} \mathrm{H}\right] 3.25$ was prepared by $\left[{ }^{3} \mathrm{H}\right]$ propionylation of the amine precursor 3.22 using reported protocols with minor modifications. ${ }^{11,19,24}$ A solution of N -succinimidyl [2,3${ }^{3} \mathrm{H}$ ]propionate (specific activity: $88 \mathrm{Ci} / \mathrm{mmol}, 1.5 \mathrm{mCi}, 3.0 \mu \mathrm{~g}, 17.05 \mathrm{nmol}$, from American Radiolabeled Chemicals Inc., St. Louis, MO via Biotrend, Köln, Germany) in hexane/EtOAc 9:1 was transferred into a 1.5 mL reaction vessel with a screw cap. The solvent was evaporated in a vacuum concentrator at approx. $30^{\circ} \mathrm{C}$ in 30 min . A solution of $3.22(0.35 \mathrm{mg}, 0.38 \mu \mathrm{~mol}, 22 \mathrm{eq})$ in a mixture of anhydrous DMF ( $50 \mu \mathrm{~L}$ ) and DIPEA ( $0.8 \mu \mathrm{~L}$ ) was added and the mixture was shaken at room temperature for 80 min . The analysis of the reaction mixture was performed with an RPHPLC system (cf. Figure 3.7) (Waters, Eschborn, Germany) consisting of two 510 pumps, a pump control module, a 486 UV/vis detector and a Flow-one Beta series A-500 radiodetector (Packard, Meriden; CT). A Luna C18 ( $3 \mu \mathrm{~m}, 150 \mathrm{~mm} \times 4.6 \mathrm{~mm}$ ) was used RP-column at a flow rate of 0.8 $\mathrm{mL} / \mathrm{min}$. Mixtures of $0.04 \%$ TFA in $\mathrm{CH}_{3} \mathrm{CN}(\mathrm{A})$ and $0.05 \%$ aq. TFA (B) were used as mobile phase. The following linear gradient was applied: 0-20 min, A/B 14:86-32:68; 20-22min, 32:68-95:5; 2232min, 95:5 isocratic), (UV detection: 220 nm ).

For the isolation of $\left[{ }^{3} \mathrm{H}\right] 3.25$ (peak at $t_{\mathrm{R}}=18.8 \mathrm{~min}$, $c f$. Figure 3.7) by analytical RP-HPLC (conditions as for the analysis of the reaction mixture, no radiometric detection), a $\mathrm{H}_{2} \mathrm{O} / \mathrm{TFA}$ ( $96: 4, \mathrm{v} / \mathrm{v}, 11.5 \mu \mathrm{~L}$ ) was added to the reaction mixture. The eluates (three injections), containing $\left.{ }^{3} \mathrm{H}\right] 3.25$, were combined in a 2 mL reaction vessel with a screw cap and the volume was reduced in a vacuum concentrator to $200 \mu \mathrm{~L}$. EtOH ( $800 \mu \mathrm{~L}$ ) was added and the solution was transferred into a 3 mL borosilicate glass vial with conical bottom (Wheaton NextGen $3 \mathrm{~mL} V$-vials). The 2 mL reaction vessel was rinsed twice with EtOH/H2O 80:20 (v/v, $2 \times 200 \mu \mathrm{~L}$ ) and the wash solutions were transferred to the 3 mL glass vials yielding the tentative stock solution with a volume of $1200 \mu \mathrm{~L}$.

As analysis of the tentative stock solution by radio-HPLC ( $4 \mu \mathrm{~L}$ dissolved in $126 \mu \mathrm{~L}$ of $0.05 \%$ aq TFA/MeCN 90:10; same conditions as for the isolation) revealed a second peak ( $t_{\mathrm{R}}=14.3 \mathrm{~min}, c f$. Figure 3.7B), amounting to ca. 13\% of the total peak area, determination of the specific activity of $\left[{ }^{3} \mathrm{H}\right] 3.25$ was not feasible. Therefore, the specific activity of $\left[{ }^{3} \mathrm{H}\right] 3.25$ was estimated based on the specific activity ( $2.41 \mathrm{TBq} / \mathrm{mmol}, 65.03 \mathrm{Ci} / \mathrm{mmol}$ ) of [ $\left.{ }^{3} \mathrm{H}\right] \mathrm{UR}-\mathrm{MK} 300^{19}$ prepared on the same day from the same lot of N -succinimidyl $\left[2,3-{ }^{3} \mathrm{H}\right.$ ]propionate. The molarity of the tentative stock solution was calculated from the specific activity of $\left[{ }^{3} \mathrm{H}\right]$ UR-MK300 ( $2.41 \mathrm{TBq} / \mathrm{mmol}$ ) and the activity concentration (determined with a LS 6500 liquid scintillation counter (Beckmann-Coulter, Munich, Germany)) of the tentative stock solution with respect to $\left[{ }^{3} \mathrm{H}\right] 3.25\left(2.47 \cdot 10^{-5} \mathrm{TBq} / \mathrm{mL}\right.$, $87 \%$ of the total activity concentration). The final total activity concentration was adjusted to 18.5 $\mathrm{MBq} / \mathrm{mL}(0.5 \mathrm{mCi} / \mathrm{mL})$ by the addition of $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(80: 20 \mathrm{v} / \mathrm{v})$ containing 0.22 mM ascorbinic acid ( $625 \mu \mathrm{~L}$ ) resulting in the final stock solution with $6.69 \mu \mathrm{~mol} / \mathrm{L}\left[{ }^{3} \mathrm{H}\right] 3.25$ and 76.9 $\mu \mathrm{mol} / \mathrm{L}$ ascorbinic acid (radiochemical yield for $\left[{ }^{3} \mathrm{H}\right] \mathbf{3 . 2 5}$ : 28.9 MBq, 64.3\%).

## Investigation of the chemical stability of 3.25

The chemical stability of the "cold form" of the radioligand (3.25) was investigated in a mixture of $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(80: 20, \mathrm{v} / \mathrm{v})$ at room temperature. Initially, a solution of $3.25(100 \mu \mathrm{M})$ in $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}$ ( $80: 20, \mathrm{v} / \mathrm{v}$ ) was prepared. The incubation was started by dilution of this $100 \mu \mathrm{M}$ solution of 3.25 with a mixture of $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}(80 / 20, \mathrm{v} / \mathrm{v})$ resulting in a $15 \mu \mathrm{M}$ solution of $\mathbf{3 . 2 5}$. After $0 \mathrm{~h}, 6 \mathrm{~h}, 12 \mathrm{~h}$, $24 \mathrm{~h}, 48 \mathrm{~h}, 72 \mathrm{~h}$ and 7 days a $8 \mu \mathrm{~L}$ aliquot was added to $192 \mu \mathrm{~L}$ of a mixture of $\mathrm{CH}_{3} \mathrm{CN} / 0.05 \%$ aq. TFA (10:90, v/v). A sample ( $80 \mu \mathrm{~L}$ ) was immediately analyzed by analytical HPLC (Gradient 2 , detection wavelength of 273 nm ).

### 3.3.3 Pharmacological Methods

## Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of Sf 9 insect cells, expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein or coexpressing the $h H_{1} R+R G S 4$, the $h H_{3} R+G_{i \alpha 2}+\beta_{1} \nu_{2}$ or $h H_{4} R+G_{i \alpha 2}+\beta_{1} \nu_{2}$ proteins was described elsewhere. ${ }^{41}$

Radioligand competition binding assays were performed as described previously with minor modifications, using the following radioligands: [ $\left.{ }^{3} \mathrm{H}\right] m e p y r a m i n e ~(H a r t m a n n ~ A n a l y t i c, ~$ Braunschweig, Germany; $\mathrm{hH}_{1} \mathrm{R}$, specific activity $\left.=20 \mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{d}}=4.5 \mathrm{nM}, \mathrm{c}_{\text {final }}=5 \mathrm{nM}\right),\left[{ }^{3} \mathrm{H}\right]$ URDE257 ${ }^{11}$ batch $04 / 2015\left(\mathrm{hH}_{2} \mathrm{R}\right.$, specific activity $\left.=32.89 \mathrm{Ci} / \mathrm{mmol}, K_{d}=12.2 \mathrm{nM}, \mathrm{c}_{\text {final }}=20 \mathrm{nM}\right)$, $\left[{ }^{3} \mathrm{H}\right]$ tiotidine (Hartmann Analytic, Braunschweig, Germany; $\mathrm{hH}_{2} \mathrm{R}$, specific activity $=78.42$ $\mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{d}}=12.75 \mathrm{nM}, \mathrm{c}_{\text {final }}=10 \mathrm{nM}$ ), $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}^{\alpha}$-methylhistamine (Hartmann Analytic, Braunschweig, Germany; $\mathrm{hH}_{3} \mathrm{R}$, specific activity $=80 \mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{d}}=3 \mathrm{nM}, \mathrm{c}_{\text {final }}=3 \mathrm{nM}$ ), [ ${ }^{3} \mathrm{H}$ ]histamine (Hartmann Analytic, Braunschweig, Germany; $\mathrm{hH}_{4} \mathrm{R}$, specific activity $=25 \mathrm{Ci} / \mathrm{mmol}$, $\left.K_{\mathrm{d}}=15.9 \mathrm{nM}, \mathrm{C}_{\text {final }}=10 \mathrm{nM}\right)$.and $\left[{ }^{3} \mathrm{H}\right]$ UR-PI294 ${ }^{20}\left(\mathrm{hH}_{4} \mathrm{R}\right.$, specific activity $=93.3 \mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{d}}=5.1$ $\left.n M, c_{\text {final }}=5 n M\right)$.

On the day of the experiment, Sf 9 membranes were thawed and sedimented by centrifugation at $13,000 \mathrm{rpm}$ at $4^{\circ} \mathrm{C}$ for 10 min . The membranes were resuspended in ice cold binding buffer ( 12.5 $\mathrm{mM} \mathrm{MgCl} 2,1 \mathrm{mM}$ EDTA and 75 mM Tris $/ \mathrm{HCl}, \mathrm{pH} 7.4$; in the following referred to as BB ) and adjusted to a protein concentration of $2-4 \mu \mathrm{~g} / \mu \mathrm{L} .80 \mu \mathrm{~L}$ BB containing $0.2 \% \mathrm{BSA}$ and the respective radioligand, followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ ), were added to every well of a 96 -well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension ( $10 \mu \mathrm{~L}$ ). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with $0.3 \%$ polyethylenimine, using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with $200 \mu \mathrm{~L}$ scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h . The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

## Functional GTP $\mathbf{~ S}$ assay on Sf9 insect cell membranes

GTPYS assays were performed as described previously ${ }^{14}$ with minor modifications. [ ${ }^{35}$ S]GTPץS (specific activity $=1000 \mathrm{Ci} / \mathrm{mmol}$ ) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to $0.5-1.5 \mu \mathrm{~g} / \mu \mathrm{L}$.

Agonist mode: $80 \mu \mathrm{~L}$ of BB containing BSA ( $0.05 \%$ final), GDP ( $1 \mu \mathrm{M}$ final) and $\left[{ }^{35} \mathrm{~S}\right] \mathrm{GTP} \psi \mathrm{S}(20 \mathrm{nCi}$ final), followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ ) were added to every well of a 96 -well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension (10 $\mu \mathrm{L}$ ). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTP $\gamma$ S assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine ( $1 \mu \mathrm{M}$ final).

## Radioligand binding assays with [ $\left.{ }^{3} \mathrm{H}\right]$ UR-SB69 on Sf9 insect cell membranes

The radioligand binding experiments were performed as described for competition binding assays on Sf9 insect cell membranes with minor adjustments.

Saturation binding: For the determination of total binding $10 \mu \mathrm{~L}$ of $\mathrm{H}_{2} \mathrm{O}$, followed by $70 \mu \mathrm{LB}$ containing $0.2 \%$ BSA, were added to every well of a 96 -well plate (Primaria clear flat bottom microplates, Corning, New York, USA). For the determination of unspecific binding $10 \mu \mathrm{~L}$ famotidine ( 300 fold excess compared to the radioligand concentration, $3-600 \mu \mathrm{M}$ ) in $\mathrm{H}_{2} \mathrm{O}$ was added. $10 \mu \mathrm{~L}$ of $\left[{ }^{3} \mathrm{H}\right] 3.25$ in concentrations of $1-200 \mathrm{nM}$ in $\mathrm{H}_{2} \mathrm{O}$ (for economic reasons the radioligand was mixed 1:4 with the cold form) were added to every well and the incubation was started by addition of the membrane suspension ( $10 \mu \mathrm{~L}$ ). The plates were shaken for 90 min at room temperature in the dark.

Competition binding: $\left[{ }^{3} \mathrm{H}\right] 3.25$ (specific activity $=65.03 \mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{D}}=15 \mathrm{nM}$ ) was used in a final concentration of 20 nM . The incubation time was 90 min .

Association: $70 \mu \mathrm{LB}$ containing $0.2 \% \mathrm{BSA}$, followed by $10 \mu \mathrm{~L}$ of membrane suspension, were added to every well of a 96 -well plate (Primaria clear flat bottom microplates, Corning, New York, USA). $10 \mu \mathrm{~L}$ [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ (final concentration: 20 nM ) in $\mathrm{H}_{2} \mathrm{O}$ and either $10 \mathrm{LL}_{2} \mathrm{O}$ (total binding) or $10 \mu \mathrm{~L}$ famotidine ( 300 -fold excess, unspecific binding) in $\mathrm{H}_{2} \mathrm{O}$ were added at different time points $(0-180 \mathrm{~min})$ at $25^{\circ} \mathrm{C}$. After the last addition the bound radioligand was separated from the free radioligand. The last time point (last addition of the radioligand) represented the shortest incubation time ( 0 min ) and the first time point the longest incubation time ( 180 min ).

Dissociation: $70 \mu \mathrm{LBB}$ containing $0.2 \% \mathrm{BSA}$, followed by $10 \mu \mathrm{~L}$ of membrane suspension, were added to every well of a 96 -well plate (Primaria clear flat bottom microplates, Corning, New York,

USA). $10 \mu \mathrm{~L}\left[{ }^{3} \mathrm{H}\right] 3.25$ (final concentration: 20 nM ) in $\mathrm{H}_{2} \mathrm{O}$ and either $10 \mu \mathrm{~L} \mathrm{H}_{2} \mathrm{O}$ (total binding) or $10 \mu$ L famotidine ( 300 -fold excess, unspecific binding) in $\mathrm{H}_{2} \mathrm{O}$ were added at different time points ( $0-180 \mathrm{~min}$ ). Every time point was preincubated for 60 min at $25^{\circ} \mathrm{C}$. After the incubation time was over $100 \mu \mathrm{~L}$ famotidine (final concentration: $3 \mu \mathrm{M}$ ) in BB was added. After the last addition the bound radioligand was separated from the free radioligand. The last time point (last addition of the radioligand) represented the shortest dissociation time ( 0 min ) and the first time point the longest dissociation time ( 180 min ).

## Cell culture

The preparation of stably transfected HEK cells (HEK293T-hH2R-qs5 ${ }^{42}$ and HEK293T-hH2R$\beta$ Arr2 ${ }^{43,44}$ ) was described elsewhere.

Cells were cultivated at $37{ }^{\circ} \mathrm{C}$ in a water saturated atmosphere containing $5 \% \mathrm{CO}_{2}$. Dulbecco's Modified Eagle Medium, containing $4.5 \mathrm{~g} / \mathrm{L}$ glucose, $3.7 \mathrm{~g} / \mathrm{L} \mathrm{NaHCO}_{3}, 110 \mathrm{mg} / \mathrm{L}$ sodium pyruvate (DMEM, Sigma-Aldrich Munich, Germany) and supplemented with $0.584 \mathrm{~g} / \mathrm{L}$ L-glutamine (Lglutamine solution, Sigma-Aldrich Munich, Germany), 1\% (v/v) Penicillin-Streptomycin (P/S, $10,000 \mathrm{U} / \mathrm{mL}$, Sigma-Aldrich Munich, Germany), $10 \%$ (v/v) fetal calf serum (FCS, Biochrom GmbH, Merck, Berlin, Germany) were used as a culture medium. Additionally, $100 \mu \mathrm{~g} / \mathrm{mL}$ hygromycin B (A.G. Scientific, Inc., San Diego, CA) and $400 \mu \mathrm{~g} / \mathrm{mL}$ G418 (Biochrom GmbH, Merck, Berlin, Germany) were added to the culture medium of HEK293T-hH2R-qs5 cells, and $400 \mu \mathrm{~g} / \mathrm{mL}$ zeocin (InvivoGen, San Diego,USA) and $600 \mu \mathrm{~g} / \mathrm{mL}$ G418 were added to the culture medium of HEK293T$h H_{2} R-\beta A r r 2$ cells.

## Radioligand saturation binding assays with [ $\left.{ }^{3} \mathrm{H}\right]$ UR-SB69 on intact HEK293T-hH ${ }_{2} \mathrm{R}$-qs5 and HEK293T-hH2R- $\beta$ Arr 2 cells

HEK293T cells were seeded in a $175-\mathrm{cm}^{2}$ culture flask 5-7 days prior to the experiment. On the day of the experiment, cells were trypsinized and detached with fresh culture medium ( 5 mL ). After centrifugation ( $250 \mathrm{~g}, 10 \mathrm{~min}$ ) the cell pellet was resuspended in Leibovitz's L-15 culture medium (L-15 medium, Gibco/Life Technologies, Carlsbad, USA) and the concentration was adjusted to $0.25-0.5 \cdot 10^{6}$ cells $/ \mathrm{mL} .80 \mu \mathrm{~L}$ cell suspension was added to every well of a 96 -well plate (Primaria clear flat bottom microplates, Corning, New York, USA). $10 \mu \mathrm{~L}$ of $\left[{ }^{3} \mathrm{H}\right] 3.25$ in concentrations of 1-200 nM in $\mathrm{H}_{2} \mathrm{O}$ (for economic reasons the radioligand was mixed 1:4 with the cold form) and either $10 \mu \mathrm{~L} \mathrm{H}_{2} \mathrm{O}$ (total binding) or $10 \mu \mathrm{~L}$ famotidine ( 300 fold excess, unspecific binding) in $\mathrm{H}_{2} \mathrm{O}$ were added to every well. The plates were shaken for 90 min at room temperature in the dark. The separation of bound from free radioligand was performed as described for membranes.

### 3.3.4 Data analysis

Retention factors $k$ were calculated according to $k=\left(t_{R^{-}} t_{0}\right) / t_{0}\left(t_{0}=\right.$ dead time). Corrected counts per minute (ccpm) from the GTPץS assay (agonist mode) were plotted against the $\log ($ concentration of the test compound), and data were analyzed by a four parameter logistic equation (GraphPad Prism Software 5.0, GraphPad Software, San Diego, CA), followed by normalization ( $0 \%=$ water value (basal activity), $100 \%=$ "top" histamine equation) and analysis by four-parameter logistic equation (log(agonist) vs. response - variable slope, GraphPad Prism). Data of the GTPYS assay (antagonist mode) were analysed by a four parameter logistic equation (GraphPad Prism), followed by normalization ( $100 \%$ = "top" of the four-parameter logistic fit, $0 \%$ $=$ unspecifically bound radioligand (ccpm) determined in the presence of famotidine at $100 \mu \mathrm{M}$ ) and analysis by four-parameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism). p/ $C_{50}$ values were converted into $\mathrm{p} K_{B}$ values according to the Cheng-Prusoff equation ${ }^{45}$. Specific binding data from saturation binding experiments were plotted against the total radio labeled ligand concentration (approximately equivalent to the "free" ligand concentration) and analyzed by a two-parameter equation describing hyperbolic binding (one site - specific binding, GraphPad Prism) and unspecific binding data was analyzed by linear regression. Specific binding data from association binding experiments were analyzed by a two parameter equation describing an exponential rise to a maximum (one-phase association, GraphPad Prism) to obtain the observed association constant $k_{\text {obs. }}$. Specific binding data from dissociation binding experiments were analyzed by a three parameter equation (one phase decay, GraphPad Prism) to obtain the dissociation rate constant $k_{\text {off. }}$ Kinetic dissociation constants $K_{\mathrm{d}(\text { kin })}$ were calculated from $k_{\text {on }}$ and $k_{\text {off }}\left(k_{\text {on }}=\left(k_{\text {obs }}-k_{\text {off }} /[\mathrm{L}] ; K_{\mathrm{d}(\text { kin })}=k_{\text {off }} / k_{\text {on }}\right)\right.$. Specific binding data from association and dissociation binding experiments were normalized ( $100 \%=$ $Y_{\text {max }}$ (association) or $Y_{0}$ (dissociation)). Total binding data from radioligand competition binding experiments were plotted against $\log (c o n c e n t r a t i o n ~ c o m p e t i t o r) ~ a n d ~ a n a l y z e d ~ b y ~ a ~ f o u r-~$ parameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism), followed by normalization ( $100 \%=$ "top" of the four-parameter logistic fit, $0 \%=$ unspecifically bound radioligand ligand determined in the presence of famotidine at $100 \mu \mathrm{M}$ ). Normalized data from competition binding experiments were again analyzed by a four-parameter logistic equation and obtained $\mathrm{p} C_{50}$ values were converted into $\mathrm{p} K_{\mathrm{i}}$ values according to the Cheng-Prusoff equation ${ }^{45}$.

### 3.4 Summary and Conclusion

The combination of the 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole structure derived from famotidine or the guanidino-4-(3-aminophenyl)thiazole structure derived from ICI127032 with the derivatized squaramide moiety of BMY25368 led to propionylated $\mathrm{H}_{2} \mathrm{R}$ high affinity antagonists (3.24-3.26 and 3.30, $\mathrm{p} K_{\mathrm{i}}$ values: 6.8-7.65). The linking of two guanidinothiazole pharmacophores by a carbamoyl guanidine linker resulted in the high affinity bivalent antagonists 3.34 and $\mathbf{3 . 3 5}$ ( $\mathrm{p} K_{\mathrm{i}}$ values: 7.3 and 7.14). The ligand $\mathbf{3 . 2 5}$ showed a high affinity to the $\mathrm{hH}_{2} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.65) and selectivity over the other subtypes (no affinity to the $\mathrm{hH}_{1} \mathrm{R}, \mathrm{hH}_{3} \mathrm{R}: \mathrm{p} K_{\mathrm{i}}$ value of 5.3 and $\mathrm{hH}_{4} \mathrm{R}$ : $\mathrm{p} K_{\mathrm{i}}$ value of 4.4). The radiolabeled form [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ (radiochemical purity of 87\%) bound in a saturable manner to membrane preparations of Sf9 cells and intact HEK293T cells, both expressing recombinant $\mathrm{hH}_{2}$ Rs and the specific binding was best fitted by nonlinear regression to a one-site binding model. The determined $K_{d}$ values ( $15-22 \mathrm{nM}$ ) were similar to the $K_{\mathrm{i}}$ value of 23 nM determined with [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 on membrane preparations of Sf9 cells. Although a part of $\left.{ }^{3} \mathrm{H}\right] \mathbf{3 . 2 5}$ bound in (pseudo)irreversible manner (plateau at $23 \%$ ), the kinetic $K_{\mathrm{d}}$ value of 26 nM was comparable to the equilibrium one and the radioligand was completely displacable by histamine, famotidine and ICI127032. Lamtidine, by contrast, could only displace $75 \%$ of $\left[{ }^{3} \mathrm{H}\right] \mathbf{3 . 2 5}$.

However, the results of the biological evaluation of $\left[{ }^{3} \mathrm{H}\right] 3.25$ should be looked at critically, as the radiochemical puritiy of the radioligand was only moderate to begin with (87\%) and stability studies showed that the radiochemical purity was further decreasing over a period of 15 month to $45 \%$. Furthermore, only one major impurity, with unknown $\mathrm{H}_{2} \mathrm{R}$ affinity and potency, was formed. This was all very surprising, as the squaramide radioligand [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 was stable in EtOH for at least 24 month. ${ }^{11}$ As such one can well assume that the 2-guanidino-4-[(2aminoethyl)thiomethyl]thiazole structure is the part vulnerable for decomposition. Most propably the impurity was formed by oxidation of the sulfide linker to either the sulfoxide or sulfone. For unequivocal identification of the impurity, a solution of the radioligand containing the impurity should be spiked with either the sulfoxide or the sulfone compound and be analyzed by RP-HPLC. For this purpose both compounds have to be specifically synthesized. Depending on the structure of the impurity, this results may also question the stability of previously described $\mathrm{H}_{2} \mathrm{R}$ radioligands like $\left[{ }^{3} \mathrm{H}\right]$ ranitidine, $\left[{ }^{3} \mathrm{H}\right]$ cimetidine and $\left[{ }^{3} \mathrm{H}\right]$ tiotidine which contain the same sulfide linker.

Nevertheless, $\left[{ }^{3} \mathrm{H}\right] 3.25$ could be a valuable molecular tool provided that purity and stability under storage conditions are improvable. Alternatively, it should also be considered to further optimize the derivatives of ICI127032 (e.g. 3.30) in order to obtain more stable ligands suitable for radiolabelling with high affinity.

### 3.5 References

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## Chapter 4

## Aminopotentidine Derivatives as Highly Potent $\mathrm{H}_{2}$ R Antagonists: Synthesis and Pharmacological Characterization of Amine Precursors and "Cold" Forms of Potential Radioligands

Note: Prior to the submission of this thesis, parts of this chapter (the synthesis of 4.1, 4.2 and 4.3) were published in cooperation with partners:

Baumeister, P.; Erdmann, D.; Biselli, S.; Kagermeier, N.; Elz, S.; Bernhardt, G.; Buschauer, A. [ ${ }^{3} \mathrm{H}$ ]UR-DE257: Development of a Tritium-Labeled Squaramide-Type Selective Histamine $\mathrm{H}_{2}$ Receptor Antagonist. ChemMedChem 2015, 10, 83-93.

### 4.1 Introduction

The piperidinomethylphenoxypropylamine (potentidine) moiety is a privileged structure for $\mathrm{H}_{2} \mathrm{R}$ antagonism and part of many high affinity ligands. Some of the most important representatives are aminopotentidine and its derivatives (Figure 4.1). ${ }^{1,2}$ It was reported that aminopotentidine shows a high antagonistic activity ( $\mathrm{p} K_{\mathrm{b}}$ value: 7.28 ) and a second or even a third substituent at the aromatic ring of the residue is well tolerated (e.g. iodoaminopotentidine: $\mathrm{p} K_{\mathrm{b}}$ value: 7.52 and diiodoaminopotentidine: $\mathrm{p} K_{\mathrm{b}}$ value: 7.79). ${ }^{1}$ Also compounds with only one substituent in 3position (e.g. Compound I) or even compounds with an aromatic heterocycle as aromatic ring (e.g. Compound II) showed a similar or slightly lower activity. ${ }^{1}$ Interestingly, the ${ }^{125}$ lodine labeled iodoaminopotentidine was a very high affinity ligand at the $\mathrm{hH}_{2} \mathrm{R}$ ( $K_{\mathrm{d}}$ value of 0.32 nM ) and the $\mathrm{p} K_{\mathrm{i}}$ values of aminopotentidine, iodoaminopotentidine and iodoazidopotentidine obtained with the afore mentioned radioligand were considerably higher than the corresponding $\mathrm{p} K_{\mathrm{b}}$ values. ${ }^{1,3}$ $\left[{ }^{125}\right.$ I]Iodoaminopotentidine was used to map the $\mathrm{H}_{2} \mathrm{R}$ densities in human and mammalian brain. ${ }^{3,4}$



Diiodoaminopotentidine $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{p} K_{\mathrm{b}}: 7.79^{\mathrm{a}}$

lodoazidopotentidine $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{b}}: 6.98^{\mathrm{a}}$ $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{i}}: 7.92^{\mathrm{a}}$


Aminopotentidine
$\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{b}}: 7.28^{\mathrm{a}}$ $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{i}}: 8.01^{\mathrm{a}}$


Compound I $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{b}}: 7.57^{\mathrm{a}}$

lodoaminopotentidine $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{b}}: 7.52^{\mathrm{a}}$ $\mathrm{gpH} \mathrm{H}_{2} \mathrm{R}: p K_{\mathrm{i}}: 9.15^{\mathrm{a}}$ $h H_{2} R: K_{d}: 0.32$ nM $^{b}$


Compound II $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{p}_{\mathrm{b}}: 6.89^{\mathrm{a}}$

$\mathrm{hH}_{2} \mathrm{R}$ : $\mathrm{p} K_{\mathrm{b}}: 7.22^{\mathrm{c}}$

Figure 4.1. Structures of aminopotentidine, selected aminopotentidine derivatives, tiotidine and UR-DE257. ${ }^{\text {a }}$ Hirschfeld et al. ${ }^{1 \mathrm{~b}}$ Traiffort et al. ${ }^{4}{ }^{4}$ Kelley et al. ${ }^{5}$ d Baumeister et al. ${ }^{6}$

Although ${ }^{125}$ iodine labeled ligands have, compared to tritium labeled ligands, the advantage of a higher specific activity, their preparation and usage require higher safety precautions and the ligands can only be used for $4-5$ weeks after preparation. ${ }^{7}$ In our workgroup tritiated compounds are highly preferred due to their longer half-life. The published tritiated radioligands showed a 100 -fold lower $\mathrm{H}_{2} \mathrm{R}$ affinity (e.g. $\left.{ }^{3} \mathrm{H}\right]$ tiotidine and $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257) compared to [ ${ }^{125}$ ]]iodoaminopotentidine. ${ }^{5,6}$

Here, the synthesis and pharmacological characterization of aminopotentidine and its analogs with different substituents (e.g. iodine, bromine, chlorine, trifluoromethyl) in the 3-position as precursors for potential tritiated ligands and synthesis of the "cold" form of potential radioligands are reported. The derivatisation of the terminal anilinic amino group of these
precursors was performed with different acylating reagents e.g. N-hydroxy succinimidyl propionate or propionic acid chloride, which are also available in tritiated form. Furthermore, a series of aminopotentidine derivatives containing a functionalized (propionylated, acetylated or methylated) aminomethyl substituent in 4-position of the aromatic ring was synthesized and characterized. In order to enable radiolabeling, the substituents of the amino-group were chosen with regard to commercially available labeling reagents such as tritiated N -hydroxy succinimidyl propionate, propionic acid chloride, acetyl chloride or methyl iodide.

### 4.2 Results And Discussion

### 4.2.1 Chemistry

The intermediate 1-(2-aminoethyl)-2-cyano-3-(3-[3-(1-piperidinylmethyl)phenoxy]propyl)guanidine 4.5 was prepared in a five step synthesis as described before ${ }^{1,2,8}$ with minor modifications (Scheme 4.1). Starting from 3-hydroxybenzaldehyde a reductive amination with piperidine in the presence of formic acid (Leuckart-Wallach reaction) led to 4.1. Addition of acrylonitrile to intermediate 4.1 led to 4.2 (variation of the Michael-addition reaction). The resulting cyanine 4.2 was reduced to the amine 4.3 by $\mathrm{LiAlH}_{4}$. Coupling of diphenyl- N cyanocarbonimidate with 4.3 and subsequently with ethylene diamine resulted in the cyanoguanidine 4.5.


Scheme 4.1. Synthesis of 1-(2-Aminoethyl)-2-cyano-3-(3-[3-(1-piperidinylmethyl)phenoxy]propyl)guanidine 4.5. Reagents and conditions: i) formic acid, $110^{\circ} \mathrm{C}, 3 \mathrm{~h}, 87 \%$; ii) benzyltrimethylammonium hydroxide, reflux, $24 \mathrm{~h}, 55 \%$; iii) $\mathrm{LiAlH}_{4}$, anhydrous diethyl ether, RT, $4 \mathrm{~h}, 99.6 \%$; iv) diphenyl- N -cyanocarbonimidate, 2-propanol, RT, 3 h , no purification; v) ethylene diamine, $\mathrm{CH}_{3} \mathrm{CN}, 150^{\circ} \mathrm{C}, 15 \mathrm{~min}$ under microwave radiation, $60 \%$ over two steps.

The initial step of the synthesis of the aminopotentidine derivatives was the amide coupling of 4.5 and the respective benzoic acid derivatives, resulting in either the amine-precursors (4.304.34), amine-protected precursors (4.39, 4.45) or directly in the final products (4.42-4.44, 4.49 and 4.50).

The benzoic acid derivatives 4.15-4.17 were prepared in a four step synthesis (Scheme 4.2). Starting from 4-methyl benzoic acid derivatives, substituted with trifluoromethyl, chlorine, or bromine in 3-position, a side chain bromination using N -bromosuccinimide and AIBN were performed. The 4-bromomethyl benzoic acid derivatives $4.6-4.8$ were transferred to the corresponding ethyl esters 4.9-4.11 using thionyl chloride and ethanol. A substitution reaction with sodium azide converted the ethyl esters 4.9-4.11 into the corresponding azides 4.12-4.14. The 4-aminomethyl benzoic acid ethylesters 4.15-4.17 were synthesized from 4.12-4.14 by a Staudinger Reaction using triphenylphosphine.

The phthalimide protected (4-Aminomethyl)benzoic acid (4.19) was synthesized from 4.18 and phthalic anhydride in the presence of acetic acid according to published protocols (Scheme 4.2). ${ }^{9,10} 4$-(Aminomethyl)-3-bromobenzoic acid (4.21) was synthesized from 4.14 by first
saponification of the ester and second Staudinger reaction of the azide with triphenylphosphine to the amine (Scheme 4.2). The phthalimide protection of 4.21 under the same conditions as of 4.19 failed. This could be due to steric hindrance of the substituent in 3-position.


Scheme 4.2. Synthesis of benzoic acid derivatives 4.15-4.17 and 4.19-4.21. Reagents and conditions: i) N Bromosuccinimide, $\mathrm{AIBN}, \mathrm{CHCl}_{3}, 75^{\circ} \mathrm{C}, 3-5 \mathrm{~h}$, no purification; ii) thionyl chloride, EtOH, $65^{\circ} \mathrm{C}, 2-24 \mathrm{~h}, 43-52 \%$ over two steps; iii) $\mathrm{NaN}_{3}$, anhydrous DMF, RT, $24 \mathrm{~h}, 49-91 \%$; iv) $\mathrm{PPh}_{3}$, THF/H2 $\mathrm{O}(1: 2, \mathrm{v} / \mathrm{v}), \mathrm{RT}, 24 \mathrm{~h}, 66-80 \%$; v) phthalic anhydride, AcOH, reflux, $3.5 \mathrm{~h}, 84 \%$; vi) NaOH, THF, RT, $24 \mathrm{~h}, 93 \%$; vii) $\mathrm{PPh}_{3}$, THF and $\mathrm{H}_{2} \mathrm{O}, \mathrm{RT}, 24 \mathrm{~h}, 27 \%$.

The 4-(Propionamidomethyl)benzoic acids 4.25-4.27 were synthesized from 4.15-4.17 by amide coupling with succinimidyl propionate followed by saponification of the ethyl ester (Scheme 4.3).


Scheme 4.3. Synthesis of benzoic acid derivatives 4.25-4.27. Reagents and conditions: i) TEA, succinimidyl propionate, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}$, ON, no purification; ii) Aq. $\mathrm{NaOH}, \mathrm{THF}, \mathrm{RT}, \mathrm{ON}, 49-84 \%$ over two steps.

4-[(Dimethylamino)methyl]benzoic acid (4.28) and 3-bromo-4-[(dimethylamino)methyl]benzoic acid (4.29) were synthesized by Eschweiler-Clarke-Methylation of the corresponding amino acid 4.18 or 4.21 with formaldehyde in the presence of formic acid (Scheme 4.4). ${ }^{11}$


Scheme 4.4. Synthesis of benzoic acid derivatives 4.28 and 4.29. Reagents and conditions: i) Aqueous formaldehyde, formic acid, reflux, ON, 94-100\%.

In scheme 4.5 the synthesis of the propionylated aminopotentidine derivatives 4.35-4.38 is depicted. Aminopotentidine ${ }^{1}$ (4.30), iodoaminopotentidine ${ }^{1,3}$ (4.34) and its derivatives 4.31-4.33 were synthesized from 4.5 by amide coupling with the corresponding 4-amino benzoic acid using TBTU as coupling reagent and DIPEA as base. The anilinic amino group of 4.30-4.34 showed a reduced nucleophilicity which made the labeling by amide coupling a challenge. The classical
reaction with succinimidyl propionate in the presence of triethylamine (see synthesis of 4.224.24, Scheme 4.3) failed. The propionylation by an excess of propionyl chloride in the presence of DMAP and TEA resulted in the desired products 4.35-4.38 in very low yields 11-23\%.


Scheme 4.5. Synthesis of the aminopotentidine derivatives 4.35-4.38. Reagents and conditions: i) TBTU, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, RT, ON, 34-68\%; ii) propionyl chloride, DMAP, TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 11-23 \%$.

The aminopotentidine derivative 4.41 was synthesized in three steps starting by amide coupling of 4.5 and 4.19 using same conditions by analogy with 4.30-4.34 (Scheme 4.6). The resulting 4.39 was first phthalimide deprotected by hydrazine and then propionylated using succinimidyl propionate.



Scheme 4.6. Synthesis of the aminopotentidine derivative 4.41. Reagents and conditions: i) TBTU, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}$, ON, $74 \%$; ii) Hydrazinium hydroxide, EtOH, RT, $4 \mathrm{~h}, 68 \%$; iii) TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 81 \%$.

In case of the aminopotentidine derivatives 4.42-4.44 the final step was the amide coupling of the amine precursor 4.5 with the respective benzoic acids 4.25-4.27 using the same conditions as for the preparation of 4.30-4.34 (Scheme 4.7).


Scheme 4.7. Synthesis of the aminopotentidine derivatives 4.42-4.44. Reagents and conditions: i) TBTU, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, RT, ON, 6-44\%.

The synthesis of the acetylated aminopotentidine derivatives 4.47 and 4.48 is depicted in Scheme 4.8. The amide coupling of 4.5 and 4.20 , followed by a Staudinger Reaction and
acetylation of the resulting amine 4.46 with acetyl chloride resulted in the product 4.48 . The acetylation of 4.40 under similar conditions led to 4.48.


Scheme 4.8. Synthesis of the aminopotentidine derivatives 4.47 and 4.48. Reagents and conditions: i) HBTU, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 44 \%$; ii) $\mathrm{PPh}_{3}, \mathrm{THF} / \mathrm{H}_{2} \mathrm{O}(1: 2, \mathrm{v} / \mathrm{v}), \mathrm{RT}, \mathrm{ON}, 19 \%$; iii) acetyl chloride, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 20 \%$; iv) acetyl chloride, DMAP, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, 2$ days, $25 \%$.

The dimethylated aminopotentidine derivatives 4.49 and 4.50 were synthesized by coupling of 4.5 with the respective benzoic acid using HBTU as coupling reagent and DIPEA as base (Scheme 4.9).


Scheme 4.9. Synthesis of the aminopotentidine derivatives 4.49 and 4.50. Reagents and conditions: i) HBTU, DIPEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 28-32 \%$.

## Adaptation of the synthetic route to radiosynthetic requirements

Radiosynthesis makes special demands on synthesis planning and reaction conditions. The labeling reaction is the key-step of the synthesis and should be the last step. In general, for radiosynthesis an excess of precursor compared to radioactive labeling reagent is used. This facilitates handling with respect of safety precautions and reduces the costs of the overall synthesis. Within the series of the "cold" forms of potential radioligands the propionylated ligands 4.37 and 4.38 showed the highest $\mathrm{hH}_{2} \mathrm{R}$ affinity (see biological evaluation) and were therefore considered for radiolabeling.

However, the reaction conditions for the synthesis of 4.37 and 4.38 were not suitable for radiosynthesis due to the necessary high excess of the "cold" labeling reagent propionic chloride and the low yields. In order to adjust the reaction conditions to radiosynthesis standards, a test reaction was carried out in small scale and with an excess of 4.33 ( $6.98 \mu \mathrm{~g}, 10 \mathrm{eq}$ ). HRMS analysis of the reaction mixture showed no 4.37 was formed. This preliminary test showed that the
radiosynthesis of 4.37 (and 4.38 ) will be a challenge. An alternative could be the synthesis of radiolabeled 4.50. Comound 4.50 showed high $\mathrm{hH}_{2} \mathrm{R}$ affinity in the range of UR-DE257 and tiotidine (see Biological evaluation). Dimethylation of 4.46 with methyliodide should be easily adaptable to radiosynthetic requirements. The main challenge will be to minimize formation of monomethylated and/or trimethylated by-product and the separation of $\mathbf{4 . 5 0}$ from the amine precursor and these by-products.

### 4.2.2 Biological Evaluation

## $H_{2} R$ affinity, selectivity compared to $H_{3} R$ and antagonism

The aminopotentidine derivatives 4.30-4.38, 4.40-4.44 and 4.46-4.50 were investigated in equilibrium competition binding experiments on membrane preparations from $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein using the antagonist [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 ${ }^{6}$ as radioligand. The selectivity of several compounds for the $h H_{2} R$ over the $h H_{3} R$ was investigated by competition binding experiments using membranes of $\mathrm{Sf9}$ insect cells co-expressing the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ and $\mathrm{G}_{\beta 1 \gamma 2}$ proteins using $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}^{\alpha}$-methylhistamine as radioligand. Additionally, representative compounds were examined for $\mathrm{hH}_{2} \mathrm{R}$ agonism in the GTP $\gamma \mathrm{S}$ binding assay on membrane preparations from $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. Ligands which exhibited no agonism were also investigated in antagonist mode versus histamine as agonist. Selected curves are depicted in Figure 4.1 and the results are summarized in Table 4.1.

Within the aminopotentidine derivatives iodoaminopotentidine (4.34) showed the highest $\mathrm{hH}_{2} \mathrm{R}$ affinity. The $\mathrm{p} K_{\mathrm{i}}$ value of 9.51 was in good agreement with the literature $\left(\mathrm{gpH}_{2} \mathrm{R}\right.$ : $\mathrm{p} K_{\mathrm{i}}$ value of $9.15^{1}$ and $\mathrm{hH}_{2} \mathrm{R}: K_{\mathrm{d}}$ value of $0.32 \mathrm{nM}^{4}$ ). Also the 3 -brominated aminopotentidine 4.33 showed a high $\mathrm{hH}_{2} \mathrm{R}$ affinity with a $\mathrm{p} K_{\mathrm{i}}$ value of 8.58 . While the introduction of a bromine or iodine substituent in 3 -position of the benzoic acid moiety (4.33 or 4.34 ) led to an increase in affinity compared to aminopotentidine (4.30, $\mathrm{p} K_{\mathrm{i}}$ value of 7.7 ), a trifluoromethyl or chlorine substituent (4.31 or 4.32) retained the $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ value of 7.57-7.7). Likewise, the exchange of the anilinic amino group in 4-position with an aminomethyl group (4.40) was tolerated with almost no change in affinity ( $\mathrm{p} K_{\mathrm{i}}$ value of 7.6 ) compared to 4.30. The 4-(propionamidomethyl)-3-bromobenzamide containing ligand 4.46 showed with a $\mathrm{p} K_{\mathrm{i}}$ value of 7.71 a decreased affinity compared to the 3 brominated aminopotentidine 4.33 .

Within the target compounds ("cold" forms of potential radioligands) the propionylated iodoaminopotentidine 4.38 and the propionylated 3-bromo aminopotentidine 4.37 showed the highest $\mathrm{hH}_{2} \mathrm{R}$ affinities with $\mathrm{p} K_{\mathrm{i}}$ values of 8.18 and 8.5. The propionylation of the 3 -halogenated aminopotentidines $4.35,4.36$ and 4.38 resulted in a decrease of $\mathrm{hH}_{2} \mathrm{R}$ affinity by one order of magnitude. By contrast, in case of 4.37, the propionylated analog of 4.33, the $\mathrm{hH}_{2} \mathrm{R}$ affinity was retained ( $\mathrm{p} K_{\mathrm{i}}$ value of 8.5). The 4-(propionamidomethyl)benzamide containing ligands 4.41-4.44 showed only moderate affinity to $\mathrm{hH}_{2} \mathrm{R}$ with $\mathrm{p} K_{\mathrm{i}}$ values of 6.58-7.2. The acetylation (4.48: $\mathrm{p} K_{\mathrm{i}}$ value of 6.4 and 4.47: $\mathrm{p} K_{\mathrm{i}}$ value of 7.09 ) as well as the propionylation (4.41: $\mathrm{p} K_{\mathrm{i}}$ value of 6.58 and 4.44: $\mathrm{p} K_{\mathrm{i}}$ value of 6.9) of the 4-aminomethyl benzamide-containing amine precursors 4.40 and 4.46 resulted in a decrease of $\mathrm{hH}_{2} \mathrm{R}$ affinity. The dimethylated analog of 4.40 (4.49) showed a low $\mathrm{hH}_{2} \mathrm{R}$ affinity with a $\mathrm{p} K_{\mathrm{i}}$ value of 5.61 . In contrast, 4.50, the dimethylated 4.46 , showed a high $\mathrm{hH}_{2} \mathrm{R}$ affinity with a $\mathrm{p} K_{\mathrm{i}}$ value of 7.54 .

The ligands $4.30,4.31,4.33-4.38,4.40,4.41,4.43$ and 4.44 showed a low $\mathrm{hH}_{3} \mathrm{R}$ affinity with $\mathrm{p} K_{\mathrm{i}}$ values of 4.5-5.22. Iodoaminopotentidine (4.34) showed with a 32000-fold higher affinity for the $\mathrm{hH}_{2} \mathrm{R}$ compared to $\mathrm{hH}_{3} \mathrm{R}$ the highest selectivity. Also the propionylated ligands 4.37 and 4.38 showed a lower but still excellent selectivity (6900- or 2500-fold).

All investigated ligands (4.30, 4.31, 4.33-4.35, 4.37, 4.38, 4.40, 4.41 and 4.44) were antagonists or inverse agonists in the GTP S binding assay. Except for the $\mathrm{p} K_{\mathrm{b}}$ values of 4.40 and 4.44 ( $\mathrm{p} K_{\mathrm{b}}$ value:
7.5 and 6.5 ), the calculated $\mathrm{p} K_{\mathrm{b}}$ values, obtained in the antagonistic mode, were considerably lower compared to the $\mathrm{p} K_{\mathrm{i}}$ values. Especially bromoaminopotentidine (4.33) and iodoaminopotentidine (4.34), which were high affinity $\mathrm{hH}_{2} \mathrm{R}$ antagonists in binding experiments ( $\mathrm{p} K_{\mathrm{i}}$ values of 8.58 and 9.51 ), showed only low activities in the GTP $\gamma$ S binding assay ( $\mathrm{p} K_{\mathrm{b}}$ values of 6.5 and 6.7). With a $\mathrm{p} K_{\mathrm{b}}$ value of 7.6 , the propionylated 3-bromo aminopotentidine 4.37 showed the highest activity among the investigated ligands.

Antagonism of aminopotentidine (4.30) and iodoaminopotentidine (4.34) ( $\mathrm{p} K_{\mathrm{b}}$ values of 6.0 and 6.7) measured with a GTPyS binding assay on membrane preparations from $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein were lower than described in literature ( $\mathrm{p} K_{\mathrm{b}}$ values of 6.63 and 7.46 , obtained by steady-state GTPase assay). ${ }^{5,12}$ Discrepancies between $\mathrm{p} K_{\mathrm{i}}$ values of aminopotentidine (4.30) and iodoaminopotentidine (4.34) from radioligand binding and $\mathrm{p} K_{\mathrm{b}}$ values from functional experiments were also described for the $\mathrm{gpH}_{2} \mathrm{R}$ by Hirschfeld et al $\left(\mathrm{gpH}_{2} \mathrm{R}\right.$ : $\mathrm{p} K_{\mathrm{i}}$ values of 8.01 and 9.15 vs. $\mathrm{p} K_{\mathrm{b}}$ values of 7.28 and 7.52$)^{1}$. Hirschfeld et al gave as possible explanation the different experimental setups which led to varying access to the $H_{2} R$ (e.g. guinea pig striatal membranes vs. intact isolated guinea pig right atrium). When a very similar experimental setup was used (in our study membrane preparations in both binding and functional studies), the use of different competitors (histamine vs. radiolabeled antagonist) could lead to the different results ${ }^{13}$. Agonists and antagonists may stabilize different receptor conformations that exhibit different affinities for the investigated agonists/antagonists/inverse agonists. ${ }^{13}$ For the antagonistic radioligand $\left[{ }^{3} \mathrm{H}\right]$ tiotidine it was already shown that it binds only to a fraction of the functionally active $\mathrm{H}_{2}$ Rs. ${ }^{5}$


Figure 4.1. Displacement of the radioligand [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 ( $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ) by (A) compounds 3.35-3.38 and (B) compounds 4.44, 4.46, 4.47, 4.50 and antagonism of (C) compounds 4.35, 4.37, 4.38 and (D) compounds 4.40, 4.41,4.47 on $\mathrm{hH}_{2} \mathrm{R}$ determined in a GTP $\gamma$ S assay (antagonistic mode). Histamine ( $1 \mu \mathrm{M}$ ) was used for stimulation. Both determined on membrane preparations of $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. Data represent mean values $\pm$ SEM of 2-4 experiments performed in triplicate.

Table 4.1. Affinities of the aminopotentidine derivatives 4.30-4.38, 4.40-4.44 and 4.46-4.50 to $\mathrm{hH}_{2,3} \mathrm{R}$, obtained from equilibrium competition binding studies and $\mathrm{hH}_{2} \mathrm{R}$ antagonism expressed in the calculated $\mathrm{p} K_{\mathrm{b}}$ values determined by a GTP $ү$ S assay.

| Compound | $\mathrm{hH}_{2} \mathrm{R}$ |  |  |  | $\mathrm{hH}_{3} \mathrm{R}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Binding ${ }^{\text {a }}$ $\mathrm{p} K_{i}$ | N | $\begin{aligned} & \text { GTPyS }^{\mathrm{c}} \\ & \mathrm{pEC} C_{50}\left(\mathrm{p} K_{\mathrm{B}}\right) \end{aligned}$ | N | Binding ${ }^{\text {d }}$ <br> $\mathrm{p} K_{\mathrm{i}}$ | N | $\begin{aligned} & \text { Selectivity } \\ & \text { hH }_{2} \mathrm{R}: \mathrm{hH}_{3} \mathrm{R} \end{aligned}$ |
| His | $6.53 \pm 0.04$ | 3 | $5.80 \pm 0.06$ | 9 | $7.8 \pm 0.1$ | 3 | 19:1 |
| 4.30 (APT) | $\begin{aligned} & 7.7 \pm 0.1 / \\ & \mathrm{gpH}_{2} \mathrm{R}: 8.01^{1} \end{aligned}$ | 3 | $\begin{aligned} & (6.0 \pm 0.4) / \\ & (6.63)^{5} \end{aligned}$ | 3 | $4.92 \pm 0.08$ | 3 | 1:600 |
| 4.31 | $7.7 \pm 0.2^{\text {b }}$ | 3 | $(6.6 \pm 0.4)$ | 4 | $5.03 \pm 0.07$ | 3 | 1:470 |
| 4.32 | $7.57 \pm 0.01$ | 3 | n.d. | - | n.d. | - | n.d. |
| 4.33 | $8.58 \pm 0.05$ | 2 | $(6.5 \pm 0.4)$ | 4 | $4.8 \pm 0.1$ | 3 | 1:6000 |
| 4.34 (IAPT) | $\begin{aligned} & 9.51 \pm 0.06 / \\ & \mathrm{gpH}_{2} \mathrm{R}: 9.15^{1} \end{aligned}$ | 3 | $\begin{aligned} & (6.7 \pm 0.1) / \\ & (7.46)^{12} \end{aligned}$ | 3 | $5.0 \pm 0.1$ | 4 | 1:32000 |
| 4.35 | $6.60 \pm 0.08$ | 3 | $(5.9 \pm 0.2)$ | 2 | $4.50 \pm 0.09$ | 4 | 1:130 |
| 4.36 | $6.7 \pm 0.1$ | 3 | n.d. | - | $4.7 \pm 0.1$ | 3 | 1:100 |
| 4.37 | $8.5 \pm 0.3$ | 3 | $(7.6 \pm 0.1)$ | 3 | $4.66 \pm 0.07$ | 4 | 1:6900 |
| 4.38 | $8.18 \pm 0.07$ | 3 | $(6.6 \pm 0.4)$ | 3 | $4.82 \pm 0.09$ | 4 | 1:2500 |
| 4.40 | $7.6 \pm 0.2^{\text {b }}$ | 2 | $(7.5 \pm 0.2)$ | 4 | $5.22 \pm 0.09$ | 2 | 1:240 |
| 4.41 | $6.58 \pm 0.04$ | 3 | $(5.4 \pm 0.3)$ | 3 | $4.75 \pm 0.06$ | 3 | 1:68 |
| 4.42 | $6.98 \pm 0.08$ | 3 | n.d. | - | n.d. | - | n.d. |
| 4.43 | $7.2 \pm 0.1$ | 3 | n.d. | - | $4.96 \pm 0.03$ | 2 | 1:170 |
| 4.44 | $6.9 \pm 0.1$ | 3 | $(6.5 \pm 0.1)$ | 4 | $5.04 \pm 0.08$ | 2 | 1:72 |
| 4.46 | $7.71 \pm 0.07$ | 3 | n.d. | - | n.d. | - | n.d. |
| 4.47 | $7.09 \pm 0.09$ | 3 | n.d. | - | n.d. | - | n.d. |
| 4.48 | $6.4 \pm 0.1$ | 4 | n.d. | - | n.d. | - | n.d. |
| 4.49 | $5.61 \pm 0.07$ | 3 | n.d. | - | n.d. | - | n.d. |
| 4.50 | $7.54 \pm 0.09$ | 3 | n.d. | - | n.d. | - | n.d. |

Competition binding assay on membrane preparations of Sf9 insect cells: ${ }^{\text {a }}$ expression of the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ or $\left.{ }^{\mathrm{b}}{ }^{3} \mathrm{H}\right]$ tiotidine, $\mathrm{c}=10 \mathrm{nM}, K_{\mathrm{d}}=12.75 \mathrm{nM}$ ), ${ }^{d}$ co-expression of the
 determined on membrane preparations of Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. The intrinsic activity ( $\alpha$ ) of histamine was set to 1.00 , and $\alpha$ values of investigated compounds were referred to this value. The $p K_{B}$ values were determined in the antagonist mode versus histamine $(c=1 \mu \mathrm{M})$ as agonist. ${ }^{\mathrm{e}}$ Selectivity represents the ratio of the corresponding $K_{\mathrm{i}}$ values. Data represent mean values $\pm$ SEM of N experiments performed in triplicate. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means $\pm$ SEM of $N$ independent experiments, each performed in triplicate.

### 4.3 EXPERIMENTAL SECTION

### 4.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Fluka (Buchs, Swiss), Alfa Aesar (Karlsruhe, Germany), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 ( $0.040-0.063 \mathrm{~mm}$ ). Microwave assisted synthesis was performed with an Initiator 2.0 (Biotage, Uppsala, Sweden) using microwave reaction vials (Biotage, Uppsala, Sweden) combined with caps and septa. Automated flash chromatography was performed with a 971-FP flash-purification system (Agilent Technologies, Santa Clara, CA). Pre-packed columns (SuperFlash SF10-4 g, SF12-8 g, SF 15-12 g und SF15-24 g, Agilent Technologies, Santa Clara, CA) were used throughout. Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution ( 0.8 g ninhydrin, 200 mL n-buthanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA). Nuclear Magnetic Resonance ( ${ }^{1} \mathrm{H} N M R$, ${ }^{13} \mathrm{C}$ NMR and ${ }^{19} \mathrm{~F}$ NMR) spectra were recorded on a Bruker Avance-300 (7.05 T, ${ }^{1} \mathrm{H}: 300 \mathrm{MHz},{ }^{13} \mathrm{C}$ : $75.5 \mathrm{MHz},{ }^{19} \mathrm{~F}: 282 \mathrm{MHz}$ ), Avance-400 ( $9.40 \mathrm{~T},{ }^{1} \mathrm{H}: 400 \mathrm{MHz},{ }^{13} \mathrm{C}: 100.6 \mathrm{MHz}$ ), or Avance-600 (14.1 $\mathrm{T} ;{ }^{1} \mathrm{H}: 600 \mathrm{MHz},{ }^{13} \mathrm{C}: 150.9 \mathrm{MHz}$ cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in $\delta(\mathrm{ppm})$ relative to external standards. Multiplicities are specified with the following abbreviations: $s$ (singlet), d (doublet), t (triplet), q (quartet), qui (quintet), m (multiplet), br s (broad signal), as well as combinations thereof. In certain cases 2DNMR techniques (COSY, HSQC, HMBC and NOESY) were used to assign ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ chemical shifts. Low-resolution mass spectrometry (MS) was performed on a Finnigan ThermoQuest TSQ 7000 instrument using an electrospray ionization (ESI) source or on a Finnigan SSQ 710A instrument (EI-MS, 70 eV ). High-resolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass Q-TOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K-2001 detector. A YMC Triart C18 ( $150 \times 20 \mathrm{~mm}, 5 \mu \mathrm{~m}$, YMC Europe GmbH , Dinslacken, Germany) served as RP-column at a flow rate of $15 \mathrm{~mL} / \mathrm{min}$ at room temperature. A detection wavelength of 220 nm and mixtures of $\mathrm{CH}_{3} \mathrm{CN}$ and $0.1 \%$ aq. $\mathrm{NH}_{3}$ as mobile phase were used throughout. $\mathrm{CH}_{3} \mathrm{CN}$ was removed from the eluates under reduced pressure (final pressure: 80 mbar) at $45{ }^{\circ} \mathrm{C}$ prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a Vacuubrand RZ 6 rotary vane vacuum pump). Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A ( $250 \times 4.6 \mathrm{~mm}, 5 \mu \mathrm{~m}, t_{0}=2.9 \mathrm{~min}$, Phenomenex, Aschaffenburg, Germany) served as RP-column for acidic runs (flow rate of $0.8 \mathrm{~mL} / \mathrm{min}$ ) and mixtures of $0.05 \%$ TFA in $\mathrm{CH}_{3} \mathrm{CN}(\mathrm{A})$ and $0.05 \%$ aq. TFA (B) were used as mobile phase. A YMC Triart C18 (150 x $2 \mathrm{~mm}, 5 \mu \mathrm{~m}, t_{0}=1.63 \mathrm{~min}$; YMC, Japan) served as RP-column for basic runs (flow rate of $0.35 \mathrm{~mL} / \mathrm{min}$ ) and mixtures of $0.1 \% \mathrm{NH}_{3}$ in
$\mathrm{CH}_{3} \mathrm{CN}(\mathrm{A})$ and $0.1 \%$ aq. $\mathrm{NH}_{3}(\mathrm{~B})$ were used as mobile phase. Helium degassing, room temperature and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration: 100-500 $\mu \mathrm{M}$ ) were either prepared from stock solution ( 10 mM in DMSO) in a mixture of $\mathrm{CH}_{3} \mathrm{CN}$ and $\mathrm{H}_{2} \mathrm{O}$ corresponding to the initial eluent composition, or as a one to one mixture of the eluate (preparative HPLC) with Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:2095:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, 30-32 min: 80:20-95:5, 32-42 min : $95: 5$ or gradient 3 (basic conditions, YMC Triat): 0-30 min: A/B 10:90-90:10, 30-32 min: 90:10-95:5, 32-42 min: 95:5. Microanalysis was performed on a Vario micro cube (Elementar, Langenselbold, Germany).

### 4.3.2 Experimental protocols and analytical data

## 3-(Piperidin-1-ylmethyl)phenol (4.1) 2,6

Piperidine ( $4.18 \mathrm{~g}, 49.1 \mathrm{mmol}, 2 \mathrm{eq}$ ) and formic acid ( $98 \%, 3,3 \mathrm{~g}, 71,3 \mathrm{mmol}, 2,6 \mathrm{eq}$ ) were added under ice cooling to 3-hydroxybenzaldehyde ( $3.00 \mathrm{~g}, 24.6 \mathrm{mmol}, 1 \mathrm{eq}$ ) and the reaction mixture was stirred at $110{ }^{\circ} \mathrm{C}$ for 3 h . After cooling to room temperature, the reaction mixture was poured in $\mathrm{H}_{2} \mathrm{O}(15 \mathrm{~mL})$. The aqueous solution was alkalized using ammonia solution ( $25 \%, \mathrm{w} / \mathrm{w}$ ). The precipitated white solid product was filtered off and dried in vacuo ( $4.06 \mathrm{~g}, 87 \%$ ). Mp: 134$138{ }^{\circ} \mathrm{C}$ (Lit. ${ }^{2}$ mp. $136-137{ }^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.2$ (EtOAc/PE $\left.85: 15\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}\right): ~ \delta$ (ppm) 1.37-1.40 (m, 2H), 1.45-1.50 (m, 4H), 2.29 (br s, 4H), $3.31(\mathrm{~s}, 2 \mathrm{H}), ~ 6.59-6.62(\mathrm{~m}, 1 \mathrm{H}), 6.67-$ $6.68(\mathrm{~d}, 1 \mathrm{H}, J 7.4 \mathrm{~Hz}), 6.70-6.71(\mathrm{~m}, 1 \mathrm{H}), 7.05-7.09(\mathrm{t}, 1 \mathrm{H}, J 7.7 \mathrm{~Hz}), 9.23(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}\right): \delta(\mathrm{ppm}) 24.0,25.6,53.9,62.9,113.7,115.4,119.3,128.9,140.1,157.2$. $\mathrm{MS}\left(\mathrm{LC}-\mathrm{MS}, \mathrm{Cl}, \mathrm{NH}_{3}, t_{\mathrm{R}}=11.7 \mathrm{~min}\right): \mathrm{m} / \mathrm{z}(\%) 192.1(100)[M+\mathrm{H}]^{+} . \mathrm{C}_{12} \mathrm{H}_{17} \mathrm{NO}$ (191.13).

## 3-[3-(Piperidin-1-ylmethyl)phenoxy]propanenitrile hydrochloride (4.2) ${ }^{2,6}$

4.1 ( $4.00 \mathrm{~g}, 20.9 \mathrm{mmol}, 1 \mathrm{eq}$ ) was suspended in acrylonitrile ( $14 \mathrm{~mL}, 210 \mathrm{mmol}, 10 \mathrm{eq}$ ) and a catalytic amount of benzyltrimethylammonium hydroxide ( $40 \%$ in $\mathrm{MeOH}, 0.2 \mathrm{~mL}$ ) was added. The reaction mixture was stirred over night under reflux. The excess of acrylonitrile was removed under reduced pressure and the oily residue was dissolved in diethyl ether ( 25 mL ). The organic layer was washed with sodium hydroxide solution ( $5 \%, \mathrm{w} / \mathrm{w}, 20 \mathrm{~mL}$ ) and $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$. The combined aqueous layers were extracted with diethyl ether ( 20 mL ). The organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The product was precipitated as HCl salt with HCl in 2-propanol (5-6 M). Removal of the solvent in vacuo afforded the product as white solid ( $3.25 \mathrm{~g}, 55 \%$ ). Mp: $163.5-165.7^{\circ} \mathrm{C}$ (Lit. ${ }^{2} \mathrm{Mp}: 161-162^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.2$ (EtOAc/PE 85:15). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right.$, HSQC, NOESY): $\delta(\mathrm{ppm}) 1.28-1.39(\mathrm{~m}, 1 \mathrm{H}), 1.66-1.88(\mathrm{~m}, 5 \mathrm{H}), 2.76-2.85(\mathrm{~m}, 2 \mathrm{H}), 3.03-3.06(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}$ 5.9 Hz ), 3.22-3.25 (d, $2 \mathrm{H}, \mathrm{J} 11.8 \mathrm{~Hz}$ ), 4.20-4.24 (m, 4H), 7.02-7.05 (dd, $1 \mathrm{H},{ }^{4}{ }^{4} 2.0 \mathrm{~Hz},{ }^{3} \mathrm{~J} 8.3 \mathrm{~Hz}$ ), 7.18$7.20(\mathrm{~d}, 1 \mathrm{H}, J 7.7 \mathrm{~Hz}), 7.35-7.39(\mathrm{~m}, 2 \mathrm{H}), 10.92(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{HSQC}\right.$, NOESY): $\delta(\mathrm{ppm}) 17.9,21.4,22.0,51.5,58.6,62.8,115.6,117.3,118.8,124.1,129.9,131.4,157.7$. MS (LC-MS, CI, $\left.\mathrm{NH}_{3}, t_{\mathrm{R}}=13.5 \mathrm{~min}\right): \mathrm{m} / \mathrm{z}(\%) 245.2(100)[\mathrm{M}+\mathrm{H}]^{+} . \mathrm{C}_{15} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O} \cdot \mathrm{HCl}(244.34+36.46)$.

## 3-[3-(Piperidin-1-ylmethyl)phenoxy]propan-1-amine (4.3) ${ }^{2,6}$

4.2 ( $3.00 \mathrm{~g}, 10.71 \mathrm{mmol}, 1 \mathrm{eq}$ ) was slowly added to a suspension of lithium aluminium hydride ( $609 \mathrm{mg}, 16.90 \mathrm{mmol}, 1.6 \mathrm{eq}$ ) in anhydrous diethyl ether ( 30 mL ) and the reaction mixture was stirred for 4 h at room temperature. After the remaining lithium aluminium hydride was hydrolyzed with sodium hydroxide solution ( $5 \%, \mathrm{w} / \mathrm{w}, 35 \mathrm{~mL}$ ), diethyl ether ( 50 mL ) was added and the insoluble material was filtered off. The two layers were separated and the organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Removal of the solvent in vacuo afforded the product as colorless oil ( $2.59 \mathrm{~g}, 99.6 \%$ ). $R_{\mathrm{f}}=0.1$ (EtOAc/MeOH + 1-3 drops TEA $80: 20$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm})$ $1.36(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.39-1.45(\mathrm{~m}, 2 \mathrm{H}), 1.54-1.59(\mathrm{~m}, 4 \mathrm{H}), 1.88-1.95$ (qui, $2 \mathrm{H}, J 6.5 \mathrm{~Hz}), 2.36(\mathrm{~m}, 4 \mathrm{H})$, $2.90(\mathrm{t}, 2 \mathrm{H}, J 6.7 \mathrm{~Hz}), 3.43(\mathrm{~s}, 2 \mathrm{H}), 4.04(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 6.2 \mathrm{~Hz}), 6.76-6.78(\mathrm{~m}, 1 \mathrm{H}), 6.87-6.89(\mathrm{~m}, 2 \mathrm{H}), 7.19$ $(\mathrm{m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 24.5,26.1,33.3,39.4,54.6,63.9,65.9,113.0,115.3$, 121.6, 129.1, 140.5, 159.1. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{25} \mathrm{~N}_{2} \mathrm{O}^{+}: 249.1961$, found: 249.1961. $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}$ (248.37).

## Phenyl- $N^{\prime}$-cyano-N-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)carbamimidate (4.4) ${ }^{8}$

4.3 ( $600 \mathrm{mg}, 2.42 \mathrm{mmol}, 1 \mathrm{eq}$ ) was added to a solution of diphenyl- N -cyanocarbonimidate ( 633 $\mathrm{mg}, 2,66 \mathrm{mmol}, 1,1 \mathrm{eq}$ ) in 2-propanol ( 90 mL ) and the reaction mixture was stirred for 3 h at room temperature. The solvent was removed under reduced pressure and the product was recristallized in diethylether. 4.4 obtained as a white solid ( $877 \mathrm{mg}, 93 \%$ ) which still contained the by-product phenol and was directly used for the next step. $R_{f}=0.9\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{~N} \mathrm{NH} 3\right.$ in MeOH 90:10). MS: (LC-MS, ESI, $\left.t_{R}=5.2 \mathrm{~min}\right): m / z(\%) 393,1(100)[M+H]^{+} . \mathrm{C}_{23} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{2}$ (392.22).

## 1-(2-Aminoethyl)-2-cyano-3-(3-[3-(1-piperidinylmethyl)phenoxy]propyl)guanidine (4.5) ${ }^{1}$

Ethylendiamine ( $307 \mathrm{mg}, 5.10 \mathrm{mmol}, 20 \mathrm{eq}$ ) was added to a stirring solution of 4.4 ( $100 \mathrm{mg}, 0.26$ $\mathrm{mmol}, 1 \mathrm{eq}$ ) in $\mathrm{CH}_{3} \mathrm{CN}(5 \mathrm{~mL})$. The reaction mixture was stirred for 15 min at $150{ }^{\circ} \mathrm{C}$ under microwave radiation. The solvent was removed under reduced pressure and the residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(7 \mathrm{~mL})$. The organic layer was washed three times with $\mathrm{H}_{2} \mathrm{O}(7 \mathrm{~mL})$ and three times with aqueous sodium hydroxide solution ( $5 \mathrm{M}, 7 \mathrm{~mL}$ ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Removal of the solvent in vacuo afforded the product as colorless oil ( $55 \mathrm{mg}, 60 \%$ ). $R_{\mathrm{f}}=$ $0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 90: 10\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm})$ 1.39-1.41 (m, 2H), 1.50-1.56 (m, 4H), 1.97-2.03 (qui, $2 \mathrm{H}, \mathrm{J} 6.2 \mathrm{~Hz}$ ), $2.33(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 2.80(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.2$ $\mathrm{Hz}), 3.18(\mathrm{t}, 2 \mathrm{H}, J 5.2 \mathrm{~Hz}), 3.39(\mathrm{~m}, 4 \mathrm{H}), 4.00(\mathrm{t}, 2 \mathrm{H}, J 5.8 \mathrm{~Hz}), 6.20(\mathrm{br} \mathrm{s}, 0.8 \mathrm{H}), 6.74-6.76(\mathrm{~m}, 1 \mathrm{H})$, 6.85-6.88 (m, 2H), $7.17(\mathrm{t}, 1 \mathrm{H}, J 7.7 \mathrm{~Hz}) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 1.46-1.48(\mathrm{~m}, 2 \mathrm{H})$, 1.57-1.63 (m, 4H), 2.02-2.08 (m, 2H), 2.42 (br s, 4H), 2.72-2.75 (t, $2 \mathrm{H}, J 6.3 \mathrm{~Hz}), 3.23-3.26(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}$ $6.3 \mathrm{~Hz}), 3.41-3.44(\mathrm{t}, 2 \mathrm{H}, J 6.7 \mathrm{~Hz}), 3.47(\mathrm{~s}, 2 \mathrm{H}), 4.05-4.08(\mathrm{t}, 2 \mathrm{H}, J 5.8 \mathrm{~Hz}), 6.86-6.90(\mathrm{~m}, 2 \mathrm{H}), 6.95-$ $6.96(\mathrm{~m}, 1 \mathrm{H}), 7.20-7.25(\mathrm{t}, 1 \mathrm{H}, J 7.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm})$ 24.4, 26.0, 28.9, 39.1, 41.8, 45.3, 54.6, 63.8, 65.4, 112.9, 115.1, 119.2, 122.0, 129.2, 140.5, 158.6, 161.3. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{19} \mathrm{H}_{31} \mathrm{~N}_{6} \mathrm{O}^{+}$: 359.2554, found: 359.2552. $\mathrm{C}_{19} \mathrm{H}_{30} \mathrm{~N}_{6} \mathrm{O}$ (358.49).

## General procedure for the synthesis of the ester protected benzoic acid derivatives 4.9-4.11

In order to remove residual water and the stabilisator from the $\mathrm{CHCl}_{3}$, the solvent was filtered through extra dry $\mathrm{AlO}_{3}(50 \mathrm{~g}$ for 100 mL ). The 4-methyl benzoic acid derivative ( 1 eq ) and AIBN ( 0.1 eq ) were suspended in $\mathrm{CHCl}_{3}(15-20 \mathrm{~mL})$. N -bromosuccinimide ( 1.5 eq ) was added portion wise and under stirring to the hot $\left(75^{\circ} \mathrm{C}\right)$ suspension. The reaction mixture was stirred for $3-5 \mathrm{~h}$ at $75{ }^{\circ} \mathrm{C}$. Removal of the solvent under reduced pressure afforded the 4 -(bromomethyl)benzoic acid derivatives 4.6-4.8 which were used for the next step without further purification. The crude 4.6, 4.7 or 4.8 was suspended in $\mathrm{EtOH}(10 \mathrm{~mL})$ and thionyl chloride ( $2-3 \mathrm{eq}$ ) was added drop wise. The reaction mixture was stirred for $2-24 \mathrm{~h}$ at $65^{\circ} \mathrm{C}$. The product was purified by automated flash chromatography. Varying amounts of the respective ethyl 4-(chloromethyl) benzoate derivative were formed as by-product and couldn't be separated by flash chromatography.

## Ethyl 4-(bromomethyl)-3-(trifluoromethyl)benzoate (4.9) ${ }^{14}$

4.9 was prepared from 4-methyl 3-(trifluoromethyl) benzoic acid ( $500 \mathrm{mg}, 2.45 \mathrm{mmol}, 1 \mathrm{eq}$ ), AIBN ( $40 \mathrm{mg}, 0.24 \mathrm{mmol}, 0.1 \mathrm{eq}$ ), N -bromosuccinimide ( $654 \mathrm{mg}, 3.67 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) in $\mathrm{CHCl}_{3}(20 \mathrm{~mL}$ ) and ester formation with thionyl chloride ( $583 \mathrm{mg}, 4.90 \mathrm{mmol}, 2 \mathrm{eq}$ ) in $\mathrm{EtOH}(10 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-97:3 in 20 min ) and removal of the solvent in vacuo afforded the product as colourless oil ( $280 \mathrm{mg}, 43 \%$ ). $R_{\mathrm{f}}=0.45$ (PE/EtOAc 95:5). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.41(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}$ $7.14 \mathrm{~Hz}), 4.38-4.45(\mathrm{q}, 2 \mathrm{H}, J 7.14 \mathrm{~Hz}), 4.78(\mathrm{~s}, 2 \mathrm{H}), 7.74-7.77(\mathrm{~m}, 1 \mathrm{H}), 8.22-8.25(\mathrm{~m}, 1 \mathrm{H}), 8.33(\mathrm{~m}$, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 14.3,41.4,61.7,123.6$ (q, 1C, $J_{\mathrm{C}, \mathrm{F}}$ 275.25 Hz ), $127.35\left(\mathrm{q}, 1 \mathrm{C}, J_{C, F} 5.85 \mathrm{~Hz}\right), 128.5\left(\mathrm{q}, 1 \mathrm{C}, J_{C, F} 31.52 \mathrm{~Hz}\right), 130.8,131.9,133.2,140.13(\mathrm{q}$, $1 \mathrm{C}, \mathrm{J}_{\mathrm{C}, \mathrm{F}} 1.40 \mathrm{~Hz}$ ), 164.9. ${ }^{19} \mathrm{~F}-\mathrm{NMR}\left(282 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm})-60.24 . \mathrm{HRMS}:(\mathrm{ESI}): m / z[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{BrF}_{3} \mathrm{O}_{2}^{+}$: 310.9889, found: 310.9892. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{ClF}_{3} \mathrm{O}_{2}^{+}$: 267.0394, found: 267.0395. $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{BrF}_{3} \mathrm{O}_{2}$ (311.10).

## Ethyl 4-(bromomethyl)-3-chlorobenzoate (4.10) ${ }^{15}$

4.10 was prepared from 3-chloro 4-methyl benzoic acid ( $380 \mathrm{mg}, 2.23 \mathrm{mmol}, 1 \mathrm{eq}$ ), AIBN ( 37 mg , $0.22 \mathrm{mmol}, 0.1 \mathrm{eq}$ ), N -bromosuccinimide ( $595 \mathrm{mg}, 3.34 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) in $\mathrm{CHCl}_{3}(15 \mathrm{~mL})$ and ester formation with thionyl chloride ( $691 \mathrm{mg}, 6.69 \mathrm{mmol}, 3 \mathrm{eq}$ ) in $\mathrm{EtOH}(10 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-97.5:2.5 in 25 min ) and removal of the solvent in vacuo afforded the product as colourless oil ( $320 \mathrm{mg}, 52 \%$ ). $R_{\mathrm{f}}=0.74(\mathrm{PE} / \mathrm{EtOAc} 6: 1) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.40(\mathrm{t}, 3 \mathrm{H}, \mathrm{J} 7.14 \mathrm{~Hz}), 4.35-4.42(\mathrm{q}$, $2 \mathrm{H}, J 7.14 \mathrm{~Hz}), 4.72(\mathrm{~s}, 2 \mathrm{H}), 7.56(\mathrm{~d}, 1 \mathrm{H}, J 8.14 \mathrm{~Hz}), 7.92-7.95(\mathrm{~m}, 1 \mathrm{H}), 8.06-8.07(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 14.3,43.0,61.6,128.2$ 130.6, 130.8, 132.1, 134.1, 139.5, 165.0. HRMS: ( ABCl ): m/z $[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{BrClO}_{2}{ }^{+}$: 276.9625, found: 276.9626. HRMS: (ABCI): m/z $[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{Cl}_{2} \mathrm{O}_{2}^{+}$: 233.0131, found: 233.0131. $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{BrClO}_{2}$ (277.54).

## Ethyl 4-(bromomethyl)-3-bromobenzoate (4.11) ${ }^{16,17}$

4.11 was prepared from 3-bromo 4-methyl benzoic acid ( $2.00 \mathrm{~g}, 9.30 \mathrm{mmol}, 1 \mathrm{eq}$ ), AIBN ( 153 mg , $0.93 \mathrm{mmol}, 0.1 \mathrm{eq})$, N -bromosuccinimide ( $2.48 \mathrm{~g}, 13.95 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) in $\mathrm{CHCl}_{3}(150 \mathrm{~mL})$ and ester formation with thionyl chloride ( $2.21 \mathrm{~g}, 18.60 \mathrm{mmol}, 2 \mathrm{eq}$ ) in $\mathrm{EtOH}(50 \mathrm{~mL})$ according to the general procedure. The crude product was purified first by column chromatography (PE/EtOAc 100:0-90:10) and then by automated flash chromatography (PE/EtOAc 100:0-85:15 in 40 min ). Removal of the solvent in vacuo afforded the product as a slightly yellow oil ( $1.39 \mathrm{~g}, 46 \%$ ). $R_{\mathrm{f}}=$ 0.60 (PE/EtOAc 5:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.40(\mathrm{t}, 3 \mathrm{H}, J 7.19 \mathrm{~Hz}), 4.36-4.42(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}$ $7.19 \mathrm{~Hz}), 4.72(\mathrm{~s}, 2 \mathrm{H}), 7.57(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J} 8.01 \mathrm{~Hz}), 7.98-8.00(\mathrm{~m}, 1 \mathrm{H}), 8.25(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 14.3,45.5,61.5,123.8,128.9,130.6,132.1,134.1,141.1,164.8$. HRMS: (ESI): $\mathrm{m} / \mathrm{z}$ $[M+H]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{Br}_{2} \mathrm{O}_{2}{ }^{+}$: 320.9120, found: 320.9114. HRMS: (ESI): $m / z[M+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{BrClO}_{2}{ }^{+}: 276.9626$, found: 276.9624. $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{Br}_{2} \mathrm{O}_{2}$ (322.00).

## General procedure for the synthesis of the azide derivatives 4.12-4.14

Sodium azide (4 eq) was added to a solution of 4.9-4.11 (1 eq) in anhydrous DMF (5-10mL). The reaction mixture was stirred for 24 h at room temperature. $\mathrm{H}_{2} \mathrm{O}(80 \mathrm{~mL})$ was added and the product was extracted three times with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{~mL})$. The organic layers were combined and the solvent was removed under reduced pressure. The residue was dissolved in EtOAc ( 80 mL ) and the organic layer was washed three times saturated $\mathrm{CaCl}_{2}$ solution ( 50 mL ), one time with brine ( 50 mL ) and was then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Removal of the solvent afforded the desired product.

## Ethyl 4-(azidomethyl)-3-(trifluoromethyl)benzoate (4.12)

4.12 was prepared from $4.9(490 \mathrm{mg}, 1.84 \mathrm{mmol}, 1 \mathrm{eq})$ and sodium azide ( $1478 \mathrm{mg}, 7.35 \mathrm{mmol}, 4$ eq) according to general procedure. Removal of the solvent in vacuo afforded the product as a slightly yellow oil ( $430 \mathrm{mg}, 83 \%$ ). $R_{\mathrm{f}}=0.5$ (PE/EtOAc 95:5). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm})$ $1.34(\mathrm{t}, 3 \mathrm{H}, J 7.10 \mathrm{~Hz}), 4.32-4.39(\mathrm{q}, 2 \mathrm{H}, J 7.10 \mathrm{~Hz}), 4.76(\mathrm{~s}, 2 \mathrm{H}), 7.85-7.88(\mathrm{~m}, 1 \mathrm{H}), 8.20(\mathrm{~m}, 1 \mathrm{H})$, 8.26-8.29 (m, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 13.9,49.86\left(\mathrm{q}, 1 \mathrm{C}, \mathrm{J}_{\mathrm{C}, \mathrm{F}} 2.08 \mathrm{~Hz}\right), 61.4$, $123.4\left(q, 1 C, J_{C, F} 274.04 \mathrm{~Hz}\right.$ ), $126.35\left(\mathrm{q}, 1 \mathrm{C}, J_{C, F} 5.81 \mathrm{~Hz}\right), 127.2\left(\mathrm{q}, 1 \mathrm{C}, J_{C, F} 30.85 \mathrm{~Hz}\right), 130.1,131.7$, 133.3, 138.70 ( $\mathrm{q}, 1 \mathrm{C}, J_{\mathrm{C}, \mathrm{F}} 1.46 \mathrm{~Hz}$ ), 164.1. ${ }^{19} \mathrm{~F}-\mathrm{NMR}\left(282 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm})-58.39 . \mathrm{HRMS}:$ (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{2}{ }^{+}$: 274.0798, found: 274.0797. $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{2}$ (273.22).

## Ethyl 4-(azidomethyl)-3-chlorobenzoate (4.13)

4.13 was prepared from $4.10(260 \mathrm{mg}, 1.12 \mathrm{mmol}, 1 \mathrm{eq})$ and sodium azide ( $290 \mathrm{mg}, 4.46 \mathrm{mmol}, 4$ eq) according to general procedure. Removal of the solvent in vacuo afforded the product as colorless oil ( $130 \mathrm{mg}, 49 \%$ ). $R_{\mathrm{f}}=0.44$ (PE/EtOAc 95:5). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.40(\mathrm{t}$, $3 \mathrm{H}, J 7.12 \mathrm{~Hz}$ ), 4.35-4.42 ( $\mathrm{q}, 2 \mathrm{H}, J 7.13 \mathrm{~Hz}$ ), $4.68(\mathrm{~s}, 2 \mathrm{H}), 7.54-7.57(\mathrm{~d}, 1 \mathrm{H}, J 8.00 \mathrm{~Hz}), 8.00-8.03(\mathrm{~m}$, $1 \mathrm{H}), 8.50-8.51(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.4,50.5,61.7,98.87,129.91,129.95$,
132.02, 140.9, 144.4, 164.7. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{ClN}_{3} \mathrm{O}_{2}^{+}$: 240.0534 , found: 240.0538. $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{ClN}_{3} \mathrm{O}_{2}$ (239.66).

## Ethyl 4-(azidomethyl)-3-bromobenzoate (4.14)

4.14 was prepared from 4.11 ( $290 \mathrm{mg}, 1.05 \mathrm{mmol}, 1 \mathrm{eq}$ ) and sodium azide ( $272 \mathrm{mg}, 4.18 \mathrm{mmol}, 4$ eq) according to general procedure. Removal of the solvent in vacuo afforded the product as yellow oil ( $270 \mathrm{mg}, 91 \%$ ). $R_{\mathrm{f}}=0.33$ (PE/EtOAc 95:5). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.40(\mathrm{t}$, $3 \mathrm{H}, J 7.12 \mathrm{~Hz}$ ), 4.36-4.42 (q, 2H, J 7.12 Hz ), $4.55(\mathrm{~s}, 2 \mathrm{H}), 7.48-7.50(\mathrm{~d}, 1 \mathrm{H}, J 7.96 \mathrm{~Hz}), 8.00-8.02(\mathrm{~m}$, $1 \mathrm{H}), 8.26(\mathrm{~m}, 1 \mathrm{H})$. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{BrN}_{3} \mathrm{O}_{2}{ }^{+}$: 284.0029, found: 284.0032. $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{BrN}_{3} \mathrm{O}_{2}$ (284.11).

## General procedure for the synthesis of the amine derivatives 4.15-4.17

4.12, 4.13 or 4.14 ( 1 eq ) and triphenylphosphine ( 1.12 eq ) were dissolved in a mixture of THF and $\mathrm{H}_{2} \mathrm{O}(5 / 1, \mathrm{v} / \mathrm{v}, 4 \mathrm{~mL})$. The reaction mixture was stirred over night at room temperature. The THF was removed under reduced pressure and the residue was diluted with $\mathrm{H}_{2} \mathrm{O}(1-7 \mathrm{~mL})$. The pH value was adjusted to two by adding aqueous HCl solution ( 0.5 M ). The aqueous layer was washed three times with EtOAc ( $10-20 \mathrm{~mL}$ ). Removal of the $\mathrm{H}_{2} \mathrm{O}$ afforded the desired product as HCl salt.

## Ethyl 4-(aminomethyl)-3-(trifluoromethyl)benzoate (4.15)

4.15 was prepared from $4.12(170 \mathrm{mg}, 0.62 \mathrm{mmol}, 1 \mathrm{eq})$ and triphenylphosphine ( $183 \mathrm{mg}, 0.70$ mmol, 1.12 eq ) dissolved in a mixture of THF and $\mathrm{H}_{2} \mathrm{O}(5 / 1, \mathrm{v} / \mathrm{v}, 4 \mathrm{~mL})$ according to general procedure. Removal of the solvent in vacuo afforded the product as yellow solid ( $140 \mathrm{mg}, 80 \%$ ). $R_{\mathrm{f}}=0.36\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 95: 5\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 1.34(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}$ 7.11 Hz ), 4.24 (br s, 2H), 4.33-4.40 (q, $2 \mathrm{H}, J 7.12 \mathrm{~Hz}$ ), 8.02-8.05 (d, 1H, J 8.17 Hz ), $8.21(\mathrm{~m}, 1 \mathrm{H})$, 8.29-8.32 (m, 1H), 8.99 (br s, 3H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 13.9,38.3\left(\mathrm{q}, 1 \mathrm{C}, \mathrm{J}_{\mathrm{C}-\mathrm{F}}\right.$ 3.40 Hz ), $61.4,123.4\left(q, 1 \mathrm{C}, J_{C-F} 274.35 \mathrm{~Hz}\right.$ ), $126.06\left(q, 1 \mathrm{C}, J_{C-F} 5.56 \mathrm{~Hz}\right), 127.3\left(q, 1 \mathrm{C}, J_{C-F} 30.57 \mathrm{~Hz}\right.$ ), 130.1, 131.0, 133.0, 137.1 (q, 1C, J $J_{C-F} 1.41 \mathrm{~Hz}$ ), 164.0. ${ }^{19} \mathrm{~F}-\mathrm{NMR}\left(282 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm})-$ 58.24. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{~F}_{3} \mathrm{NO}_{2}{ }^{+}$: 248.0893, found: 248.0895. $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~F}_{3} \mathrm{NO}_{2}$ (247.22).

## Ethyl 4-(aminomethyl)-3-chlorobenzoate (4.16)

4.16 was prepared from 4.13 ( $920 \mathrm{mg}, 3.84 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triphenylphosphine ( $1128 \mathrm{mg}, 4.30$ mmol, 1.12 eq$)$ dissolved in a mixture of THF and $\mathrm{H}_{2} \mathrm{O}(5 / 1, \mathrm{v} / \mathrm{v}, 20 \mathrm{~mL})$ according to general procedure. Removal of the solvent in vacuo afforded the product as white solid ( $630 \mathrm{mg}, 66 \%$ ). $R_{\mathrm{f}}$ $=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 95: 5\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 1.33(\mathrm{t}, 3 \mathrm{H}, \mathrm{J} 7.08$ $\mathrm{Hz}), 4.19(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 4.31-4.37(\mathrm{q}, 2 \mathrm{H}, J 7.10 \mathrm{~Hz}), 7.81-7.83(\mathrm{~m}, 1 \mathrm{H}), 7.96-7.98(\mathrm{~m}, 2 \mathrm{H}), 8.85(\mathrm{br} \mathrm{s}$,

3H). ${ }^{13} \mathrm{C}$-NMR ( $100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta(\mathrm{ppm}) 14.0,39.7,61.3,127.7,129.4,130.6,131.4,133.0$, 136.7, 164.1. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{CINO}_{2}{ }^{+}: 214.0629$, found: 214.0631 . $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{ClNO}_{2}$ (213.66).

## Ethyl 4-(aminomethyl)-3-bromobenzoate (4.17)

4.17 was prepared from $4.14(730 \mathrm{mg}, 2.51 \mathrm{mmol}, 1 \mathrm{eq})$ and triphenylphosphine ( $755 \mathrm{mg}, 2.88$ $\mathrm{mmol}, 1.12 \mathrm{eq})$ dissolved in a mixture of THF and $\mathrm{H}_{2} \mathrm{O}(5 / 1, \mathrm{v} / \mathrm{v}, 15 \mathrm{~mL})$ according to general procedure. Removal of the solvent in vacuo afforded the product as white solid ( $550 \mathrm{mg}, 73 \%$ ). $\mathrm{Mp}: 212-213{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 95: 5\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta$ (ppm) $1.34(\mathrm{t}, 3 \mathrm{H}, \mathrm{J} 7.12 \mathrm{~Hz}), 4.18(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 4.31-4.37(\mathrm{q}, 2 \mathrm{H}, \mathrm{J} 7.11 \mathrm{~Hz}), 7.77-7.79(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J} 8.11$ $\mathrm{Hz}), 8.01-8.03(\mathrm{~m}, 1 \mathrm{H}), 8.15(\mathrm{~m}, 1 \mathrm{H}), 8.78$ (br s, 3H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 14.0$, 41.8, 61.3, 123.1, 128.2, 130.3, 131.4, 132.6, 138.3, 164.0. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{BrNO}_{2}{ }^{+}: 258.0124$, found: 258.0127. $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{BrNO}_{2}$ (258.12).

## 4-((1,3-Dioxoisoindolin-2-yl)methyl)benzoic acid (4.19) ${ }^{9,10}$

(4-Aminomethyl)benzoic acid (4.18) ( $200 \mathrm{mg}, 1.32 \mathrm{mmol}, 1 \mathrm{eq}$ ) and phthalic anhydride ( 212 mg , $1.46 \mathrm{mmol}, 1.1 \mathrm{eq})$ were suspended in acetic acid ( 1 mL ) and the reaction mixture was stirred under reflux for 3.5 h . The solvent was removed under reduced pressure and the residue was suspended in $\mathrm{H}_{2} \mathrm{O}(2 \mathrm{~mL})$. The product was filtered off and dried in vacuo. 4.19 was obtained as white solid ( $312 \mathrm{mg}, 84 \%$ ). Mp: 262-266 ${ }^{\circ} \mathrm{C}$ (Lit. $\left.\mathrm{mp}^{9}: 267-270^{\circ} \mathrm{C}\right) . R_{\mathrm{f}}=0.8\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}+2\right.$ drops TFA 95:5). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): ~ \delta(\mathrm{ppm}) 4.84(\mathrm{~s}, 2 \mathrm{H}), 7.42(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J} 8.22 \mathrm{~Hz}), 7.85-$ $7.94(\mathrm{~m}, 6 \mathrm{H}), 12.95(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): ~ \delta(\mathrm{ppm}) 40.6,123.3,127.4,129.6$, 129.9, 131.6, 134.6, 141.5, 167.0, 167.7. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{NO}_{4}{ }^{+}: 282.0761$, found: 282.0764. $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{NO}_{4}$ (281.27).

## 4-(Azidomethyl)-3-bromobenzoic acid (4.20)

4.14 ( $1.29 \mathrm{~g}, 4.54 \mathrm{mmol}, 1 \mathrm{eq}$ ) was suspended in THF ( 50 mL ). Aqueous NaOH solution ( $1 \mathrm{~mol} / \mathrm{L}$, $23 \mathrm{~mL}, 5 \mathrm{eq}$ ) was added and the reaction mixture was stirred over night at room temperature. The pH value was adjusted with aqueous HCl solution ( $0.5 \mathrm{~mol} / \mathrm{L}$ ) to two. The THF was removed under reduced pressure and the product was extracted with EtOAc ( $3 \times 40 \mathrm{~mL}$ ). The organic layers were combined, washed with brine ( 40 mL ) and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Removal of the solvent in vacuo afforded the product as beige solid ( $1.08 \mathrm{~g}, 93 \%$ ). Mp: $124{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.1$ (PE / EtOAc 95:5). ${ }^{1} \mathrm{H}$-NMR ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right.$ ]DMSO): $\delta(\mathrm{ppm}) 4.64$ (s, 2H), 7.64 (d, 1H, J 7.92 Hz ), 7.95-7.98 (m, 1H), 8.12 (m, 1H). ${ }^{13} \mathrm{C}$-NMR ( $100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta(\mathrm{ppm}) 53.8,123.6,129.3,131.2,133.1,133.7$, 140.0, 166.2. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{8} \mathrm{H}_{7} \mathrm{BrN}_{3} \mathrm{O}_{2}{ }^{+}$: 255.9716 , found: 255.9719. $\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{BrN}_{3} \mathrm{O}_{2}$ (256.06).

## 4-(Aminomethyl)-3-bromobenzoic acid (4.21) ${ }^{18}$

4.20 ( $280 \mathrm{mg}, 1.09 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triphenylphosphine ( $321 \mathrm{mg}, 1.22 \mathrm{mmol}, 1.12 \mathrm{eq}$ ) were dissolved in a mixture of THF and $\mathrm{H}_{2} \mathrm{O}(5 / 1, \mathrm{v} / \mathrm{v}, 6 \mathrm{~mL})$. The reaction mixture was stirred over night at room temperature. The yellow precipitate was filtered off and washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (1 mL ). Removal of residual solvent in vacuo afforded the product as yellow solid ( $67 \mathrm{mg}, 27 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}+3$ drops of TFA): $\delta(\mathrm{ppm}) 4.20-4.22(\mathrm{~m}, 2 \mathrm{H}), 7.66(\mathrm{~d}, 1 \mathrm{H}, J 8.06 \mathrm{~Hz})$, 8.00-8.03 (m, 1H), $8.15(\mathrm{~m}, 1 \mathrm{H}), 8.44(\mathrm{br} \mathrm{s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}+3\right.$ drops of TFA): $\delta$ (ppm) 42.6, 123.6, 129.1, 130.5, 133.2, 133.6, 138.3, 166.1. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{8} \mathrm{H}_{9} \mathrm{BrNO}_{2}{ }^{+}$: 229.9811, found: 229.9811. $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{BrNO}_{2}$ (230.06).

## General procedure for the synthesis of the benzoic acid derivatives 4.25-4.27

4.15, 4.16 or 4.17 (1 eq) and TEA (3 eq) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. Succinimidyl propionate (1.5 eq) was added and the reaction mixture was stirred over night at room temperature. The mixture was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and the organic layer was washed three times with aqueous HCl solution ( $0.5 \mathrm{~mol} / \mathrm{L}, 10 \mathrm{~mL}$ ), two times with $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ and with brine ( 10 mL ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed under reduced pressure to afford the intermediates 4.22-4.24. These intermediates were ester deprotected subsequently. 4.22, 4.23 or 4.24 was suspended in THF ( $3-5 \mathrm{~mL}$ ). Aqueous NaOH solution ( $1 \mathrm{~mol} / \mathrm{L}, 5 \mathrm{eq}$ ) was added and the reaction mixture was stirred over night at room temperature. The pH value was adjusted with aqueous HCl solution ( $0.5 \mathrm{~mol} / \mathrm{L}$ ) to $2-7$. The THF was removed under reduced pressure and the product was extracted with EtOAc ( $3 \times 5-20 \mathrm{~mL}$ ). The organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Removal of the solvent under reduced pressure afforded the desired product.

## 4-(Propionamidomethyl)-3-(trifluoromethyl)benzoic acid (4.25)

4.25 was prepared from 4.15 ( $80 \mathrm{mg}, 0.28 \mathrm{mmol}, 1 \mathrm{eq}$ ), TEA ( $86 \mathrm{mg}, 0.84 \mathrm{mmol}, 3 \mathrm{eq}$ ) and succinimidyl propionate ( $72 \mathrm{mg}, 0.42 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$ according to general procedure. 4.22 was obtained as a yellow oily solid ( 90 mg , 99.9\%). Removal of the ester protecting group by aqueous NaOH solution ( $1 \mathrm{~mol} / \mathrm{L}, 1.4 \mathrm{~mL}$ ) afforded 4.25 as slightly pink solid ( $50 \mathrm{mg}, 66 \%$ ). $R_{\mathrm{f}}=0.2$ (PE / EtOAc +2 drops TFA 1:2). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 1.10$ ( $\mathrm{t}, 3 \mathrm{H}, J 7.59 \mathrm{~Hz}$ ), $2.27(\mathrm{q}, 2 \mathrm{H}, J 7.58 \mathrm{~Hz}), 4.53-4.54(\mathrm{~m}, 2 \mathrm{H}), 7.66(\mathrm{~d}, 1 \mathrm{H}, J 8.28 \mathrm{~Hz}), 8.21-8.24(\mathrm{~m}$, $2 \mathrm{H}), 8.49(\mathrm{t}, 1 \mathrm{H}, J 5.70 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 9.7,28.3,38.5\left(\mathrm{q}, 1 \mathrm{C}, J_{\mathrm{C}-\mathrm{F}} 2.89\right.$ Hz ), 123.9 ( $\mathrm{q}, 1 \mathrm{C}, J_{\mathrm{C}-\mathrm{F}} 273.79 \mathrm{~Hz}$ ), 126.1 ( $\mathrm{q} .1 \mathrm{C}, J_{\mathrm{C}-\mathrm{F}} 5.68 \mathrm{~Hz}$ ), 126.3 (q, 1C, J $J_{C-F} 30.78 \mathrm{~Hz}$ ), 128.9, 130.1, 133.1, 142.6 (q,1C, $J_{C-F} 1.46 \mathrm{~Hz}$ ), 165.9, 173.2. ${ }^{19} \mathrm{~F}-\mathrm{NMR}\left(282 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm})-$ 59.5. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~F}_{3} \mathrm{NO}_{3}{ }^{+}$: 276.0842, found: 276.0843. $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~F}_{3} \mathrm{NO}_{3}$ (275.23).

## 3-Chloro-4-(propionamidomethyl)benzoic acid (4.26)

4.26 was prepared from $4.16(200 \mathrm{mg}, 0.80 \mathrm{mmol}, 1 \mathrm{eq})$, TEA ( $243 \mathrm{mg}, 2.40 \mathrm{mmol}, 3 \mathrm{eq}$ ) and succinimidyl propionate ( $205 \mathrm{mg}, 1.20 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(8 \mathrm{~mL})$ according to general procedure. 4.23 was obtained as a white oily solid ( 230 mg ). Removal of the ester protecting group by aqueous NaOH solution ( $1 \mathrm{~mol} / \mathrm{L}, 4.1 \mathrm{~mL}$ ) afforded 4.26 as white solid ( 150 $\mathrm{mg}, 84 \%) . R_{\mathrm{f}}=0.2$ (PE / EtOAc +2 drops TFA 1:2). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 1.04(\mathrm{t}$, $3 \mathrm{H}, J 7.58 \mathrm{~Hz}), 2.21(\mathrm{q}, 2 \mathrm{H}, J 7.58 \mathrm{~Hz}), 4.35-4.37(\mathrm{~m}, 2 \mathrm{H}), 7.42(\mathrm{~d}, 1 \mathrm{H}, J 7.85 \mathrm{~Hz}), 7.86-7.89(\mathrm{~m}, 2 \mathrm{H})$, $8.40(\mathrm{t}, 1 \mathrm{H}, \mathrm{J} 5.72 \mathrm{~Hz}), 13.26(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 9.8,28.2,39.9,127.9$, 128.6, 129.5, 131.0, 132.0, 141.4, 165.9, 173.2. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{ClNO}_{3}{ }^{+}$: 242.0579, found: 242.0581. $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{ClNO}_{3}$ (241.67).

## 3-Bromo-4-(propionamidomethyl)benzoic acid (4.27)

4.27 was prepared from 4.17 ( $260 \mathrm{mg}, 1.01 \mathrm{mmol}, 1 \mathrm{eq}$ ), TEA ( $204 \mathrm{mg}, 2.02 \mathrm{mmol}, 2 \mathrm{eq}$ ) and succinimidyl propionate ( $258 \mathrm{mg}, 1.51 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to general procedure. 4.24 was obtained as a white oily solid ( $220 \mathrm{mg}, 70 \%$ ). Removal of the ester protecting group by aqueous NaOH solution ( $1 \mathrm{~mol} / \mathrm{L}, 3.2 \mathrm{~mL}$ ) afforded 4.27 as white solid ( 90 $\mathrm{mg}, 49 \%) . R_{\mathrm{f}}=0.2$ (PE / EtOAc +2 drops TFA 1:2). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 1.05(\mathrm{t}$, $3 \mathrm{H}, J 7.58 \mathrm{~Hz}), 2.22(\mathrm{q}, 2 \mathrm{H}, J 7.58 \mathrm{~Hz}), 4.33(\mathrm{~d}, 2 \mathrm{H}, J 5.87 \mathrm{~Hz}), 7.39-7.41(\mathrm{~m}, 1 \mathrm{H}), 7.90-7.93(\mathrm{~m}, 1 \mathrm{H})$, 8.06-8.07 (m, 1H), $8.39(\mathrm{t}, 1 \mathrm{H}, \mathrm{J} 5.86 \mathrm{~Hz}), 13.23(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm})$ $9.8,28.3,42.4,122.0,128.4,128.5,131.2,132.8,143.0,165.8,173.2$. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{BrNO}_{3}^{+}: 288.0052$, found: 288.0055. $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{BrNO}_{3}$ (286.13).

## General procedure for the methylation of 4.19 and 4.21 (synthesis of 4.28 and 4.29)

4.19 or 4.21 was dissolved in formic acid ( 1 mL ) under heating. Aqueous formaldehyde solution ( $37 \%, 1 \mathrm{~mL}$ ) was added and the reaction mixture was stirred over night under reflux. The mixture was cooled to room temperature and aqueous HCl solution $(20 \%, 1 \mathrm{~mL})$ was added. Removal of the solvent in vacuo afforded the desired product.

## 4-[(Dimethylamino)methyl]benzoic acid hydrochloride(4.28) ${ }^{11}$

4.28 was prepared from 4.19 ( $400 \mathrm{mg}, 2.65 \mathrm{mmol}, 1 \mathrm{eq}$ ) according to general procedure. The product was obtained as white solid ( $570 \mathrm{mg}, 100 \%$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}, \mathrm{MeOD}+3$ drops of TFA): $\delta(\mathrm{ppm}) 2.88(\mathrm{~s}, 6 \mathrm{H}), 4.42(\mathrm{~s}, 2 \mathrm{H}), 7.65-7.67(\mathrm{~m}, 2 \mathrm{H}), 8.11-8.13(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}, \mathrm{MeOD}$ +3 drops of TFA): $\delta(\mathrm{ppm}) 41.8,60.1,130.2,130.8,132.3,134.3,167.4$. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{NO}_{2}^{+}: 180.1019$, found: 180.1022. $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{NO}_{2} \cdot \mathrm{HCl}(179.22+36.46)$.

## 3-Bromo-4-[(dimethylamino)methyl]benzoic acid hydrochloride (4.29)

4.29 was prepared from $4.21(50 \mathrm{mg}, 0.22 \mathrm{mmol}, 1 \mathrm{eq})$ according to general procedure. The product was obtained as yellow hygroscopic solid ( $60 \mathrm{mg}, 94 \%$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}, \mathrm{MeOD}+3$ drops of TFA): $\delta(\mathrm{ppm}) 2.96(\mathrm{~s}, 6 \mathrm{H}), 4.60(\mathrm{~s}, 2 \mathrm{H}), 7.81(\mathrm{~d}, 1 \mathrm{H}, J 7.99 \mathrm{~Hz}), 8.07-8.09(\mathrm{~m}, 1 \mathrm{H}), 8.30-$ $8.31(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}, \mathrm{MeOD}+3$ drops of TFA): $\delta(\mathrm{ppm}) 42.4,59.8,125.3,129.1$, 133.1, 133.8, 134.28, 134.30, 165.9. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{BrNO}_{2}{ }^{+}: 258.0124$, found: 258.0145. $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{BrNO}_{2} \cdot \mathrm{HCl}(258.12+36.46)$.

## General procedure for the synthesis of the aminopotentidine derivatives 4.30-4.34

4-Aminobenzoic acid, 4-amino-3-(trifluoromethyl)benzoic acid, 4-amino-3-chlorobenzoic acid, 4-amino-3-bromobenzoic acid or 4-amino-3-iodobenzoic acid (1-1.5 eq) and DIPEA (3 eq) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2} \quad(5-100 \mathrm{~mL})$. $O$-(Benzotriazol-1-yl)- $N, N, N^{\prime}, N^{\prime}$-tetramethyluroniumtetrafluoroborate (TBTU, 1.2-1.5 eq) was added and the mixture was stirred for 10-30 min. Subsequently, 4.5 (1eq) was added and the reaction mixture was stirred over night at room temperature. The organic layer was further diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10-100 \mathrm{~mL})$ and washed three times with $\mathrm{H}_{2} \mathrm{O}(10-100 \mathrm{~mL})$ and three times with aqueous NaOH solution ( $5 \%, \mathrm{w} / \mathrm{w}, 10-100 \mathrm{~mL}$ ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed in vacuo. The resulting crude product was either directly used in the next synthesis step or purified by preparative HPLC.

## 4-Amino-N-(2-(2-cyano-3-(3-(3-(piperidin-1-

ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Aminopotentidine, 4.30) ${ }^{1,3}$
4.30 was prepared from 4-aminobenzoic acid ( $29 \mathrm{mg}, 0.21 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), DIPEA ( $54 \mathrm{mg}, 0.42$ mmol, 3 eq ), TBTU ( $67 \mathrm{mg}, 0.21 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) and $4.5(50 \mathrm{mg}, 0.14 \mathrm{mmol}, 1 \mathrm{eq}$ ) according to general procedure. Purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: MeCN/0.1\% aq. $\mathrm{NH}_{3} 33: 67-70: 30, t_{\mathrm{R}}=10.6 \mathrm{~min}$ ) afforded the product as white solid ( $23 \mathrm{mg}, 34 \%$ ). Mp: 66-80 ${ }^{\circ} \mathrm{C}$ (Lit. ${ }^{1,3} \mathrm{Mp}: 92-95{ }^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 7:2). RP-HPLC (gradient 2, $220 \mathrm{~nm}): 98 \%\left(t_{\mathrm{R}}=13.61 \mathrm{~min}, k=3.7\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.47-1.49(\mathrm{~m}, 2 \mathrm{H})$, 1.63-1.71 (m, 4H), 2.02-2.09 (qui, $2 \mathrm{H}, \mathrm{J} 6.1 \mathrm{~Hz}$ ), $2.59(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.38-3.55(\mathrm{~m}, 6 \mathrm{H}), 3.63(\mathrm{br} \mathrm{s}, 2 \mathrm{H})$, 4.03-4.07 (m, 4H), 6.59-6.64 (d, 2H, J 8.7 Hz), 6.84-6.92 (m, 2H), 7.01 (br s, 1H), $7.07(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 7.22 (t, 1H, J 7.9 Hz ), 7.57-7.60 (d, 2H, J 8.6 Hz ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right):$ $\delta(\mathrm{ppm}) 1.50(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.62-1.66(\mathrm{~m}, 4 \mathrm{H}), 2.02-2.06$ (qui, $2 \mathrm{H}, \mathrm{J} 6.2 \mathrm{~Hz}), 2.61(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.37-3.39$ $(\mathrm{m}, 2 \mathrm{H}), 3.42-3.47(\mathrm{~m}, 4 \mathrm{H}), 3.65(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 4.06(\mathrm{t}, 2 \mathrm{H}, J 5.8 \mathrm{~Hz}), 6.63-6.65(\mathrm{~d}, 2 \mathrm{H}, J 8.6 \mathrm{~Hz}), 6.90-$ $6.93(\mathrm{~m}, 2 \mathrm{H}), 6.98(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.25(\mathrm{t}, 1 \mathrm{H}, J 7.9 \mathrm{~Hz}), 7.56-7.59(\mathrm{~d}, 2 \mathrm{H}, J 8.9 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, $\left.\mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 24.5,25.8,29.9,40.4,40.5,42.7,55.0,63.9,66.8,114.7$, $115.3,117.5,119.9,122.7,123.7,130.0,130.6,137.6,153.4,160.4,161.5,171.1$. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{26} \mathrm{H}_{36} \mathrm{~N}_{7} \mathrm{O}_{2}^{+}: 478.2925$, found: 478.2927. $\mathrm{C}_{26} \mathrm{H}_{35} \mathrm{~N}_{7} \mathrm{O}_{2}$ (477.61).

## 4-Amino-3-trifluoromethyl-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Trifluoromethylaminopotentidine, 4.31)

4.31 was prepared from 4-amino-3-trifluoromethylbenzoic acid ( $172 \mathrm{mg}, 0.84 \mathrm{mmol}, 1 \mathrm{eq}$ ), DIPEA ( $325 \mathrm{mg}, 2.51 \mathrm{mmol}, 3 \mathrm{eq}$ ), TBTU ( $323 \mathrm{mg}, 1.00 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) and $4.5(300 \mathrm{mg}, 0.84 \mathrm{mmol}, 1 \mathrm{eq})$ according to general procedure. The solvent was removed under reduced pressure and the residue was purified by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 97.5: 2.5-95: 5\right)$. Removal of the solvent in vacuo afforded the product as yellow solid ( $267 \mathrm{mg}, 68 \%$ ). 70 mg was further purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: MeCN/0.1\% aq. $\mathrm{NH}_{3} 30: 70-60: 40, t_{\mathrm{R}}=14.4 \mathrm{~min}$ ). 4.31 was obtained as white solid ( $49 \mathrm{mg}, 66 \%$ ). Mp: $62-88^{\circ} \mathrm{C} . R_{\mathrm{f}}=$ $0.4\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 97:3). RP-HPLC (gradient 2, 220 nm ): $96 \%\left(t_{\mathrm{R}}=17.83 \mathrm{~min}, k=5.2\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm})$ 1.43-1.44 (m, 2H), 1.54-1.60(m, 4H), 2.00-2.06(m, 2H), 2.39 (br s, 4H), 3.38-3.48 (m, 8H), 4.04 (t, 2H, J5.87 Hz), 6.81-6.88 (m, 3H), $6.92(\mathrm{~m}, 1 \mathrm{H}), 7.19(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}$ $7.88 \mathrm{~Hz}), 7.72-7.74(\mathrm{~m}, 1 \mathrm{H}), 7.93-7.94(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 25.2,26.5$, $30.0,40.4,40.6,42.6,55.4,64.7,66.7,112.7$ (q, 1C, $J_{C-F} 30.35 \mathrm{~Hz}$ ), 114.7, 117.0, 117.4, 120.0, $122.1,123.5,126.4\left(q, 1 C, J_{C-F} 271.34 \mathrm{~Hz}\right), 127.6\left(q, 1 C, J_{C-F} 5.39 \mathrm{~Hz}\right), 130.3,133.0,139.8,150.5(q$, $1 \mathrm{C}, J_{C-F} 1.63 \mathrm{~Hz}$ ), 160.3, 161.6, 169.7. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{27} \mathrm{H}_{35} \mathrm{~F}_{3} \mathrm{~N}_{7} \mathrm{O}_{2}^{+}: 546.2799$, found: 546.2802. $\mathrm{C}_{27} \mathrm{H}_{34} \mathrm{~F}_{3} \mathrm{~N}_{7} \mathrm{O}_{2}$ (545.61).

## 4-Amino-3-chloro-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Chloroaminopotentidine, 4.32)

4.32 was prepared from 4-amino-3-chlorobenzoic acid ( $96 \mathrm{mg}, 0.56 \mathrm{mmol}, 1 \mathrm{eq}$ ), DIPEA ( 216 mg , $1.67 \mathrm{mmol}, 3 \mathrm{eq}$ ), TBTU ( $215 \mathrm{mg}, 0.67 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) and 4.5 ( $200 \mathrm{mg}, 0.56 \mathrm{mmol}, 1 \mathrm{eq}$ ) according to general procedure. The solvent was removed under reduced pressure and the residue was purified by automated flash chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 0-75: 25 \mathrm{in} 53 \mathrm{~min}\right)$. Removal of the solvent in vacuo afforded the product as white solid ( $140 \mathrm{mg}, 49 \%$ ). 90 mg was further purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3}$ 30:70-70:30, $t_{\mathrm{R}}=20.6 \mathrm{~min}$ ). 4.32 was obtained as white solid ( $60 \mathrm{mg}, 33 \%$ ). Mp: $100-103{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=$ $0.4\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.7 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 90:10). RP-HPLC (gradient 1, 220 nm ): $97.2 \%\left(t_{\mathrm{R}}=17.74 \mathrm{~min}, k=\right.$ 5.1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.35$ (br s, 2H$), 1.45-1.47$ (m, $4 \mathrm{H})$, 1.89-1.93 (m, 2 H ), $2.27(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.23-3.35(\mathrm{~m}, 8 \mathrm{H}$, interfering with the water signal), $3.95(\mathrm{t}$, $2 \mathrm{H}, J 6.13 \mathrm{~Hz}), 5.88(\mathrm{~s}, 2 \mathrm{H}), 6.76-6.77(\mathrm{~m}, 2 \mathrm{H}), 6.82-6.83(\mathrm{~m}, 2 \mathrm{H}), 7.07-7.19(\mathrm{~m}, 3 \mathrm{H}), 7.53-7.55(\mathrm{~m}$, $1 \mathrm{H}), 7.73(\mathrm{~m}, 1 \mathrm{H}), 8.30(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm})$ $24.0,25.5,28.6,38.4,38.8,41.0,53.9,62.8,65.0,112.7,114.1,114.6,116.1,118.0,120.9,122.2$, 127.2, 128.4, 129.0, 140.3, 147.4, 158.5, 159.5, 165.7. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{26} \mathrm{H}_{35} \mathrm{ClN}_{7} \mathrm{O}_{2}{ }^{+}: 512.2535$, found: 512.2544 . $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{ClN}_{7} \mathrm{O}_{2}$ (512.06).

## 4-Amino-3-bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Bromoaminopotentidine, 4.33)

4.33 was prepared from 4-amino-3-bromobenzoic acid ( $121 \mathrm{mg}, 0.56 \mathrm{mmol}, 1 \mathrm{eq}$ ), DIPEA ( 216 $\mathrm{mg}, 1.67 \mathrm{mmol}, 3 \mathrm{eq})$, TBTU ( $215 \mathrm{mg}, 0.67 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) and $4.5(200 \mathrm{mg}, 0.57 \mathrm{mmol}, 1 \mathrm{eq}$ )
according to general procedure. The crude product was obtained as sticky yellow solid ( 315 mg ). 215 mg was purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 40: 60-80: 20, t_{\mathrm{R}}=13.1 \mathrm{~min}$ ). 4.33 was obtained as white solid ( $115 \mathrm{mg}, 54 \%$ ). Mp: $94{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 90: 10\right)$. IR (KBr): 3325, 2935, 2165, 1585, 1500, $1300,1255,1160,755 \mathrm{~cm}^{-1}$. RP-HPLC (gradient $2,220 \mathrm{~nm}$ ): $98.7 \%\left(t_{\mathrm{R}}=16.76 \mathrm{~min}, k=4.8\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm})$ 1.42-1.43 (m, 2 H ), 1.54-1.60(m, 4 H$), 2.00-2.07$ (qui, $2 \mathrm{H}, J 6.1 \mathrm{~Hz}$ ), 2.41 (br s, 4H), 3.38-3.52 (m, 8H), $4.02(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.7 \mathrm{~Hz}), 4.53(\mathrm{~s}, 2 \mathrm{H}), 6.35(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.64(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.68$ (d, 1H, J 8.4 Hz ), 6.79-6.82 (dd, 1H, J $8.1 \mathrm{~Hz}, J 2.1 \mathrm{~Hz}$ ), 6.88 (d, 1H, J 7.6 Hz ), 6.92 (br s, 1H), 7.19 (t, $1 \mathrm{H}, J 7.8 \mathrm{~Hz}$ ), 7.44 (br s, 0.9 H ), $7.51-7.54(\mathrm{dd}, 1 \mathrm{H}, J 8.4 \mathrm{~Hz}, J 2.0 \mathrm{~Hz}), 7.91(\mathrm{~d}, 1 \mathrm{H}, J 2.0 \mathrm{~Hz}) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta(\mathrm{ppm}) 1.42-1.44(\mathrm{~m}, 2 \mathrm{H}), 1.53-1.59(\mathrm{~m}, 4 \mathrm{H}), 1.99-2.05(q u i, 2 \mathrm{H}, J 6.4 \mathrm{~Hz})$, 2.37 (br s, 4H), 3.35-3.46 (m, 8H), 4.03 (t, $2 \mathrm{H}, J 5.8 \mathrm{~Hz}), 6.78(\mathrm{~d}, 1 \mathrm{H}, J 8.6 \mathrm{~Hz}), 6.81-6.87(\mathrm{~m} \mathrm{2H})$, 6.90-6.91 (m, 1H), $7.19(\mathrm{t}, 1 \mathrm{H}, J 7.9 \mathrm{~Hz}), 7.54-7.56(\mathrm{dd}, 1 \mathrm{H}, J 8.5 \mathrm{~Hz}, J 2.1 \mathrm{~Hz}), 7.89(\mathrm{~d}, 1 \mathrm{H}, J 2.1 \mathrm{~Hz})$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 25.1,26.4,30.0,40.3,40.5,42.6,55.4,64.7,66.6,108.3$, 114.6, 115.3, 116.9, 119.9, 123.4, 124.1, 128.8, 130.2, 133.3, 139.8, 150.3, 160.2, 161.5, 169.5. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{26} \mathrm{H}_{35} \mathrm{BrN}_{7} \mathrm{O}_{2}{ }^{+}: 556.2030$, found: 556.2032. $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{BrN}_{7} \mathrm{O}_{2}$ (556.51).

## 4-Amino-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-3iodobenzamide (lodoaminopotentidine, 4.34) ${ }^{1}$

4.34 was prepared from 4-amino-3-iodobenzoic acid ( $147 \mathrm{mg}, 0.56 \mathrm{mmol}, 1 \mathrm{eq}$ ), DIPEA ( 216 mg , $1.67 \mathrm{mmol}, 3 \mathrm{eq}$ ), TBTU ( $215 \mathrm{mg}, 0.67 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) and 4.5 ( $200 \mathrm{mg}, 0.57 \mathrm{mmol}, 1 \mathrm{eq}$ ) according to general procedure. The crude product was obtained as yellow oil ( 419 mg ). 319 mg was purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : MeCN/0.1\% aq. $\mathrm{NH}_{3}$ 40:60-80:20, $t_{\mathrm{R}}=14.0 \mathrm{~min}$ ). 4.34 was obtained as white solid ( $166 \mathrm{mg}, 65 \%$ ). $\mathrm{Mp}: 94{ }^{\circ} \mathrm{C}$ (Lit. ${ }^{1} \mathrm{Mp}$ : $114-117{ }^{\circ} \mathrm{C}$ decomposition). $R_{\mathrm{f}}=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 90: 10\right)$. IR (KBr) 3320, 2935, 2165, 1590, 1490, 1300, 1260, $1150 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): $98.3 \% ~\left(t_{\mathrm{R}}=17.46 \mathrm{~min}, k=\right.$ 5.0). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 1.43-1.44(\mathrm{~m}, 2 \mathrm{H}), 1.54-1.60(\mathrm{~m}, 4 \mathrm{H}), 2.00-2.06$ (qui, 2 H , $J 6.3 \mathrm{~Hz}), 2.39(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.35-3.46(\mathrm{~m}, 8 \mathrm{H}), 4.03(\mathrm{t}, 2 \mathrm{H}, J 5.8 \mathrm{~Hz}), 6.74(\mathrm{~d}, 1 \mathrm{H}, J 8.5 \mathrm{~Hz}), 6.82-6.88$ $(\mathrm{m}, 2 \mathrm{H}), 6.91(\mathrm{~m}, 1 \mathrm{H}), 7.20(\mathrm{t}, 1 \mathrm{H}, J 7.9 \mathrm{~Hz}), 7.56-7.59(\mathrm{dd}, 1 \mathrm{H}, J 8.5 \mathrm{~Hz}, J 2.1 \mathrm{~Hz}), 8.11(\mathrm{~d}, 1 \mathrm{H}, J 2.1$ $\mathrm{Hz}){ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 25.2,26.5,30.1,40.4,40.6,42.6,55.4,64.8,66.7,82.3$, 114.2, 114.7, 117.0, 120.0, 123.5, 124.7, 129.8, 130.3, 139.8, 139.9, 153.0, 160.3, 161.6, 169.4. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{26} \mathrm{H}_{35} I \mathrm{~N}_{7} \mathrm{O}_{2}{ }^{+}$: 604.1891, found: 604.1896. Anal. calcd. for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{IN}_{7} \mathrm{O}_{2}$ : C 51.75, H 5.68, N 16.25, found: C 51.25, H 5.66, N 16.19. $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{IN}_{7} \mathrm{O}_{2}$ (603.51).

## General procedure for the propionylation of the aminopotentidine derivatives

The respective aminopotentidine derivative 4.30-4.34, 4-(dimetylamino) pyridine (DMAP, 0.1-1.1 eq) and triethylamine ( $2-5 \mathrm{eq}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2-3 \mathrm{~mL})$. The mixture was stirred for several minutes and propionyl chloride ( 3 eq ) was added. The reaction mixture was stirred over night at room temperature. The organic layer was further diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$, washed three times with aqueous NaOH solution ( $5 \% \mathrm{w} / \mathrm{w}, 5 \mathrm{~mL}$ ) and then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The product was purified by preparative HPLC.

## N-(2-[2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino]ethyl)-4(propionamido)benzamide (4.35)

4.35 was prepared from 4.30 ( $100 \mathrm{mg}, 0.21 \mathrm{mmol}, 1 \mathrm{eq}$ ), 4-(dimethylamino) pyridine ( $28 \mathrm{mg}, 0.23$ $\mathrm{mmol}, 1.1 \mathrm{eq}$ ), triethylamine ( $106 \mathrm{mg}, 1.05 \mathrm{mmol}, 5 \mathrm{eq}$ ) and propionyl chloride ( $58 \mathrm{mg}, 0.63$ $\mathrm{mmol}, 3 \mathrm{eq})$ according to general procedure. Due to incomplete conversion additional propionyl chloride ( $116 \mathrm{mg}, 1.26 \mathrm{mmol}, 6 \mathrm{eq}$ ) and triethylamine ( $212 \mathrm{mg}, 2.1 \mathrm{mmol}, 10 \mathrm{eq}$ ) were added. The reaction mixture was stirred for 17 h at room temperature. Purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 40: 60-70: 30, t_{\mathrm{R}}=11.3 \mathrm{~min}$ ) afforded the product as white solid ( $26 \mathrm{mg}, 23 \%$ ). Mp : 171-176 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}\right.$ in MeOH 95:5). IR (KBr): 3295, 2935, 2160, 1665, 1590, 1525, 1440, 1375, 1345, 1310, 1260, 1205, 845, 770, 690, $580 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): $97.9 \%\left(t_{\mathrm{R}}=15,71 \mathrm{~min}, k=4.4\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta(\mathrm{ppm}) 1.18-1.22(\mathrm{t}, 3 \mathrm{H}, J 7.6 \mathrm{~Hz}), 1.45-1.46(\mathrm{~m}, 2 \mathrm{H}), 1.56-1.61(\mathrm{~m}, 4 \mathrm{H}), 2.01-$ 2.07 (qui, $2 \mathrm{H}, J 6.2 \mathrm{~Hz}$ ), 2.38-2.44 (m, 6H), 3.40-3.44 (m, 4H), 3.48-3.51 (m, 4H), $4.04(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.9$ $\mathrm{Hz}), 6.84-6.90(\mathrm{~m}, 2 \mathrm{H}), 6.93(\mathrm{br} s, 1 \mathrm{H}), 7.21(\mathrm{t}, 1 \mathrm{H}, J 7.8 \mathrm{~Hz}), 7.65(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J} 8.7 \mathrm{~Hz}), 7.77(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J} 8.7$ $\mathrm{Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 10.1,25.1,26.4,30.0,31.2,40.4,40.7,42.5,55.3,64.6$, 66.7, 114.8, 117.1, 120.3, 123.5, 129.3, 130.2, 130.3, 139.4, 139.5, 143.5, 160.3, 161.6, 170.3, 175.6. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{29} \mathrm{H}_{40} \mathrm{~N}_{7} \mathrm{O}_{3}{ }^{+}$: 534.3187, found: 534.3189. $\mathrm{C}_{29} \mathrm{H}_{39} \mathrm{~N}_{7} \mathrm{O}_{3}$ (533.68).

N-(2-[2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino]ethyl)-3-(trifluoromethyl)-4-(propionamido)benzamide (4.36)
4.36 was prepared from 4.31 ( $183 \mathrm{mg}, 0.34 \mathrm{mmol}, 1 \mathrm{eq}$ ), 4-(dimethylamino) pyridine ( $45 \mathrm{mg}, 0.37$ mmol, 1.1 eq ), triethylamine ( $172 \mathrm{mg}, 1.70 \mathrm{mmol}, 5 \mathrm{eq}$ ) and propionyl chloride ( $94 \mathrm{mg}, 1.02$ $\mathrm{mmol}, 3 \mathrm{eq})$ according to general procedure. Due to incomplete conversion additional propionyl chloride ( $94 \mathrm{mg}, 1.02 \mathrm{mmol}, 3 \mathrm{eq}$ ) were added after stirring over night. The reaction mixture was stirred overnight at room temperature. Purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: MeCN/0.1\% aq. $\mathrm{NH}_{3} 35: 65-55: 45, t_{\mathrm{R}}=24.2 \mathrm{~min}$ ) afforded the product as white solid ( $31 \mathrm{mg}, 15 \%$ ). Mp: 137-138 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 97:3). RP-HPLC (gradient 2, $220 \mathrm{~nm}): 96.4 \%\left(t_{\mathrm{R}}=17.86 \mathrm{~min}, k=5.2\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, ~ N O E S Y\right):$ $\delta(\mathrm{ppm}) 1.22(\mathrm{t}, 3 \mathrm{H}, J 7.57 \mathrm{~Hz}), 1.45(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.56-1.60(\mathrm{~m}, 4 \mathrm{H}), 2.02-2.06(\mathrm{~m}, 2 \mathrm{H}), 2.41-2.48(\mathrm{~m}$, $6 \mathrm{H}), 3.41-3.45(\mathrm{~m}, 6 \mathrm{H}), 3.51-3.53(\mathrm{~m}, 2 \mathrm{H}), 4.04-4.06(\mathrm{~m}, 2 \mathrm{H}), 6.84-6.88(\mathrm{~m}, 2 \mathrm{H}), 6.94(\mathrm{~s}, 1 \mathrm{H}), 7.20$ $(\mathrm{t}, 1 \mathrm{H}, J 7.84 \mathrm{~Hz}), 7.72-7.73(\mathrm{~m}, 1 \mathrm{H}), 8.03-8.05(\mathrm{~m}, 1 \mathrm{H}), 8.18(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right.$, COSY, HSQC, HMBC, NOESY): $\delta$ (ppm) 10.0, 25.1, 26.4, 30.0, 30.4, 40.3, 40.7,42.3,55.3, 64.7, 66.6, $114.7,116.9,119.8,123.4,124.8\left(q, 1 C, J_{C-F} 272.45 \mathrm{~Hz}\right.$ ), 126.0 ( $q, 1 \mathrm{C}, J_{C-F} 30.42 \mathrm{~Hz}$ ), 126.8 (q, 1C, $J_{C-F} 5.27 \mathrm{~Hz}$ ), 130.2, 130.4, 132.5, 133.4, 139.4, 139.8, 160.2, 161.5, 168.5, 176.3. HRMS: (ESI): m/z $[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{30} \mathrm{H}_{39} \mathrm{~F}_{3} \mathrm{~N}_{7} \mathrm{O}_{3}{ }^{+}$: 602.3061, found: 602.3065. Anal. calcd. for $\mathrm{C}_{30} \mathrm{H}_{38} \mathrm{~F}_{3} \mathrm{~N}_{7} \mathrm{O}_{3} \cdot \mathrm{H}_{2} \mathrm{O}: \mathrm{C}$ 58.15, H 6.51, N 15.82, found: C 57.82, H 6.21, N 15.30. $\mathrm{C}_{30} \mathrm{H}_{38} \mathrm{~F}_{3} \mathrm{~N}_{7} \mathrm{O}_{3}$ (601.68).

## N-(2-[2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino]ethyl)-3-bromo-4(propionamido)benzamide (4.37)

4.37 was prepared from 4.33 ( $100 \mathrm{mg}, 0.18 \mathrm{mmol}, 1 \mathrm{eq}$ ), 4-(dimethylamino) pyridine ( $2.2 \mathrm{mg}, 0.02$ mmol, 0.1 eq ), triethylamine ( $36 \mathrm{mg}, 0.36 \mathrm{mmol}, 2 \mathrm{eq}$ ) and propionyl chloride ( $25 \mathrm{mg}, 0.27 \mathrm{mmol}$, 1.5 eq ) according to general procedure. Due to incomplete conversion additional propionyl chloride ( $25 \mathrm{mg}, 0.27 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), 4 -(dimethylamino)-pyridine ( $2.2 \mathrm{mg}, 0.018 \mathrm{mmol}, 0.1 \mathrm{eq}$ ) and triethylamine ( $36 \mathrm{mg}, 0.36 \mathrm{mmol}, 2 \mathrm{eq}$ ) were added after 3 h of stirring. The reaction mixture was stirred over night at room temperature. Purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 40: 60-60: 40, t_{\mathrm{R}}=17.6 \mathrm{~min}$ ) afforded the product as white solid ( $12 \mathrm{mg}, 11 \%$ ). Mp: $155-159{ }^{\circ} \mathrm{C}$ decomposition. $R_{\mathrm{f}}=0.6\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in MeOH 95:5). IR (KBr): 3310, 2935, 2165, 1600, 1540, 1510, 1465, 1275, 1195, 700, $565 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): $95.1 \%\left(t_{\mathrm{R}}=17.34 \mathrm{~min}, k=5.0\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta$ (ppm) $1.22(\mathrm{t}, 3 \mathrm{H}, J 7.6 \mathrm{~Hz}$ ), $1.44(\mathrm{~m}, 2 \mathrm{H}), 1.55-1.59$ (qui, $4 \mathrm{H}, J 5.6 \mathrm{~Hz}), 2.01-2.05(q u i, 2 \mathrm{H}, J 6.2 \mathrm{~Hz}$ ), $2.40(b r s, 4 H), 2.47-2.51(q, 2 H, J 7.6 \mathrm{~Hz}), 3.40-3.42(\mathrm{~m}, 4 \mathrm{H}), 3.45(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.47-3.49(\mathrm{~m}, 2 \mathrm{H})$, 4.03-4.05 (t, 2H, J 5.9 Hz ), 6.83-6.85 (dd, 1H, J $8.1 \mathrm{~Hz}, J 2.0 \mathrm{~Hz}), 6.87(\mathrm{~d}, 1 \mathrm{H}, J 7.6 \mathrm{~Hz}), 6.92(\mathrm{~m}, 1 \mathrm{H})$, 7.19 (t, 1H, J 7.8 Hz ), 7.75-7.77 (dd, 1H, J $8.5 \mathrm{~Hz}, J 2.0 \mathrm{~Hz}$ ), 7.90 (d, 1H, J 8.5 Hz ), 8.08 (d, 1H, J 2.0 $\mathrm{Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 10.0,25.1,26.4,30.0,30.7,40.3,40.7,42.3,55.3,64.7$, $66.6,114.7,116.9,117.5,119.8,123.4,126.2,128.0,130.2,133.1,133.2,139.7,140.4,160.2$, 161.5, 168.6, 175.5. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{29} \mathrm{H}_{39} \mathrm{BrN}_{7} \mathrm{O}_{3}{ }^{+}$: 612.2292, found: 612.2295. $\mathrm{C}_{29} \mathrm{H}_{38} \mathrm{BrN}_{7} \mathrm{O}_{3}$ (612.57).

## N-[2-(2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino)ethyl]-3-iodo-4(propionamido)benzamide (4.38)

4.38 was prepared from 4.34 ( $100 \mathrm{mg}, 0.17 \mathrm{mmol}, 1 \mathrm{eq}$ ), 4-(dimethylamino)pyridine ( $22 \mathrm{mg}, 0.18$ mmol, 1.1 eq ), triethylamine ( $84 \mathrm{mg}, 0.83 \mathrm{mmol}, 5 \mathrm{eq}$ ) and propionyl chloride ( $46 \mathrm{mg}, 0.5 \mathrm{mmol}$, $3 \mathrm{eq})$ according to general procedure. Due to incomplete conversion additional propionyl chloride ( $46 \mathrm{mg}, 0.50 \mathrm{mmol}, 3 \mathrm{eq}$ ) was added after 5 hours of stirring. The reaction mixture was stirred over night at room temperature. Purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: $\mathrm{MeCN} / 1 \%$ aq. $\mathrm{NH}_{3} 40: 60-80: 20, t_{\mathrm{R}}=13.7 \mathrm{~min}$ ) afforded the product as white solid ( $24 \mathrm{mg}, 22 \%$ ). Mp : $148-150{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 95: 5\right)$. IR ( KBr ): 3310, 2935, 2160, 1645, 1595, 1540, 1510, 1315, 1275, 1200, 1155, 1120, 785, 695, $570 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): $97.2 \%\left(t_{\mathrm{R}}=17.50 \mathrm{~min}, k=5.0\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 1.25(\mathrm{t}$, $3 \mathrm{H}, J 7.6 \mathrm{~Hz}), 1.44-1.45(\mathrm{~m}, 2 \mathrm{H}), 1.55-1.60(\mathrm{~m}, 4 \mathrm{H}), 2.01-2.07(q u i, 2 \mathrm{H}, J 6.2 \mathrm{~Hz}), 2.40(\mathrm{br} \mathrm{s}, 4 \mathrm{H})$, 2.45-2.50 (q, $2 \mathrm{H}, J 7.6 \mathrm{~Hz}$ ), 3.39-3.44 (m, 6H), 3.47-3.50 (m, 2H), 4.05 (t, $2 \mathrm{H}, J 5.9 \mathrm{~Hz}), 6.83-6.88$ (m, 2H), $6.93(\mathrm{~m}, 1 \mathrm{H}), 7.20(\mathrm{t}, 1 \mathrm{H}, J 7.8 \mathrm{~Hz}), 7.68(\mathrm{~d}, 1 \mathrm{H}, J 8.4 \mathrm{~Hz}), 7.78-7.81$ (dd, 1H, J $8.4 \mathrm{~Hz}, J 2.0$ $\mathrm{Hz}), 8.33(\mathrm{~d}, 1 \mathrm{H}, J 2.0 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm}) 10.2,25.2,26.5,30.0,30.8,40.4$, 40.7, 42.4, 55.4, 64.7, 66.7, 114.7, 117.0, 119.9, 123.5, 126.7, 128.9, 130.3, 132.3, 134.0, 139.7, 139.9, 143.5, 160.3, 161.6, 168.5, 175.5. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{29} \mathrm{H}_{39} \mathrm{IN}_{7} \mathrm{O}_{3}{ }^{+}$: 660.2154, found: 660.2160. $\mathrm{C}_{29} \mathrm{H}_{38} \mathrm{IN}_{7} \mathrm{O}_{3}$ (659.57).

## N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-((1,3-dioxoisoindolin-2-yl)methyl)benzamide (4.39)

4.19 ( $390 \mathrm{mg}, 1.39 \mathrm{mmol}, 1 \mathrm{eq}$ ) and DIPEA ( $538 \mathrm{mg}, 4.16 \mathrm{mmol}, 3 \mathrm{eq}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(120 \mathrm{~mL})$ and TBTU ( $534 \mathrm{mg}, 1.66,1.2 \mathrm{eq}$ ) was added. The mixture was stirred for 10 min at room temperature. 4.5 ( $497 \mathrm{mg}, 1.39 \mathrm{mmol}, 1 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ was added and the reaction mixture was stirred over night at room temperature. The organic layer was washed two times with $\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$, aqueous NaOH solution ( $5 \%, \mathrm{w} / \mathrm{w}, 100 \mathrm{~mL}$ ) and brine ( 100 mL ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solvent was removed under reduced pressure and the residue was purified by column chromatography ( $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH} 3$ in $\mathrm{MeOH} 97.5: 2.5-95: 5$ ). Removal of the solvent in vacuo afforded the product as light yellow solid ( $640 \mathrm{mg}, 74 \%$ ). Mp: 70$72{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 90: 10\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm}) 1.35-$ $1.37(\mathrm{~m}, 2 \mathrm{H}), 1.46-1.47(\mathrm{~m}, 4 \mathrm{H}), 1.87-1.94(\mathrm{~m}, 2 \mathrm{H}), 2.28(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.26-3.36(\mathrm{~m}, 8 \mathrm{H}), 3.94-3.97$ $(\mathrm{m}, 2 \mathrm{H}), 4.82(\mathrm{~s}, 2 \mathrm{H}), 6.77-6.84(\mathrm{~m}, 3 \mathrm{H}), 7.08-7.20(\mathrm{~m}, 3 \mathrm{H}), 7.38(\mathrm{~d}, 2 \mathrm{H}, J 8.27 \mathrm{~Hz}), 7.79(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}$ $8.30 \mathrm{~Hz}), 7.86-7.92(\mathrm{~m}, 2 \mathrm{H}), 8.57(\mathrm{t}, 1 \mathrm{H}, \mathrm{J} 5.36 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 19.6,24.3$, $25.8,28.9,39.85,39.94,41.3,54.5,63.7,65.7,113.3,115.3,119.0,122.1,123.6,127.8,128.6$, 129.3, 132.1, 133.2, 134.3, 139.8, 140.1, 158.6, 160.3, 168.0, 168.6. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{35} \mathrm{H}_{40} \mathrm{~N}_{7} \mathrm{O}_{4}{ }^{+}$: 622.3136, found: 622.3149. $\mathrm{C}_{35} \mathrm{H}_{39} \mathrm{~N}_{7} \mathrm{O}_{4}$ (621.74).

## 4-(Aminomethyl)-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.40)

4.39 ( $360 \mathrm{mg}, 0.58 \mathrm{mmol}, 1 \mathrm{eq}$ ) and hydrazinium hydroxide ( $145 \mathrm{mg}, 2.9 \mathrm{mmol}, 5 \mathrm{eq}$ ) were dissolved in EtOH ( 20 mL ). The reaction mixture was stirred for 4 h at room temperature and subsequently cooled down in the freezer for 30 min . The precipitated by-product 2,3-dihydrophthalazine-1,4-dione was filtered off and washed with EtOH ( 5 mL ). The organic layers were combined and the solvent was removed under reduced pressure. Purification by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 95: 5$ isocratic) afforded the product as white solid ( $193 \mathrm{mg}, 68 \%$ ). 70 mg of 4.40 were further purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: MeCN/0.1\% aq. $\mathrm{NH}_{3} 30: 70-60: 40, t_{\mathrm{R}}=14.4 \mathrm{~min}$ ). The product was obtained as white solid (49 mg, 47\%). Mp: 62-88 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.75\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 6 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 80: 20\right)$. RP-HPLC (gradient 2, 220 nm ): 98.9\% ( $t_{\mathrm{R}}=11.84 \mathrm{~min}, k=3.1$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}\right.$, $\mathrm{HMBC}): \delta(\mathrm{ppm}) 1.35$ (br s, 2H), 1.44-1.48 (m, 4H), 1.91 (qui, $2 \mathrm{H}, J 6.49 \mathrm{~Hz}$ ), 2.28 (br s, 4 H ), 3.25$3.29(\mathrm{~m}, 4 \mathrm{H}$, interfering with the water signal), 3.34-3.37 (m, 4 H , interfering with the water signal), $3.76(\mathrm{~s}, 2 \mathrm{H}), 3.95(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 6.20 \mathrm{~Hz}), 6.77-6.78(\mathrm{~m}, 1 \mathrm{H}), 6.83-6.84(\mathrm{~m}, 2 \mathrm{H}), 7.08-7.12(\mathrm{~m}$, $2 \mathrm{H}), 7.18(\mathrm{t}, 1 \mathrm{H}, J 7.98 \mathrm{~Hz}), 7.39(\mathrm{~d}, 2 \mathrm{H}, J 8.16 \mathrm{~Hz}), 7.77(\mathrm{~d}, 2 \mathrm{H}, J 8.18 \mathrm{~Hz}), 8.52(\mathrm{t}, 1 \mathrm{H}, J 5.45 \mathrm{~Hz})$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 24.0,25.6,28.6,38.4,38.9,40.8$, 45.1, 53.9, 62.8, 65.0, 112.7, 114.6, 118.0, 120.9, 126.8, 127.1, 129.0, 132.3, 140.3, 147.0, 158.5, 159.5, 166.7. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{27} \mathrm{H}_{38} \mathrm{~N}_{7} \mathrm{O}_{2}{ }^{+}$: 492.3081, found: 492.3085. $\mathrm{C}_{27} \mathrm{H}_{37} \mathrm{~N}_{7} \mathrm{O}_{2}$ (491.64).

## N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4(propionamidomethyl)benzamide (4.41)

4.40 ( $118 \mathrm{mg}, 0.24 \mathrm{mmol}, 1 \mathrm{eq}$ ), triethylamine ( $49 \mathrm{mg}, 0.48 \mathrm{mmol}, 2 \mathrm{eq}$ ) and succinimidyl propionate ( $62 \mathrm{mg}, 0.36 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) were dissolved $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$. The reaction mixture was stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 35: 65-70: 30$, $t_{\mathrm{R}}=11.8 \mathrm{~min}$ ) afforded the product as white solid ( $106 \mathrm{mg}, 81 \%$ ). $\mathrm{Mp}: 166-170{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.5$ ( $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 90: 10$ ). RP-HPLC (gradient 2, 220 nm ): $96.6 \%\left(t_{\mathrm{R}}=14.87 \mathrm{~min}, k=4.1\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.02(\mathrm{t}, 3 \mathrm{H}, J 7.59 \mathrm{~Hz}), 1.36(\mathrm{br} \mathrm{s}$, 2 H ), 1.46-1.47 (m, 4H), 1.91 (qui, $2 \mathrm{H}, J 6.47 \mathrm{~Hz}$ ), 2.14 ( $\mathrm{q}, 2 \mathrm{H}, J 7.60 \mathrm{~Hz}$ ), 2.29 (br s, 4H), 3.25-3.29 ( $\mathrm{m}, 2 \mathrm{H}$, interfering with the water signal), 3.33-3.36 ( $\mathrm{m}, 4 \mathrm{H}$, interfering with the water signal), $3.95(\mathrm{t}, 2 \mathrm{H}, J 6.15 \mathrm{~Hz}), 4.28(\mathrm{~d}, 2 \mathrm{H}, J 5.94 \mathrm{~Hz}), 6.78-6.79(\mathrm{~m}, 1 \mathrm{H}), 6.83-6.84(\mathrm{~m}, 2 \mathrm{H}), 7.07-7.12(\mathrm{~m}$, $2 \mathrm{H}), 7.17-7.20(\mathrm{~m}, 1 \mathrm{H}), 7.29(\mathrm{~d}, 2 \mathrm{H}, J 8.07 \mathrm{~Hz}), 7.77(\mathrm{~d}, 2 \mathrm{H}, J 8.22 \mathrm{~Hz}), 8.30(\mathrm{t}, 1 \mathrm{H}, J 5.88 \mathrm{~Hz}), 8.53$ (t, 1H, J 5.46 Hz ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]\right.$ DMSO, COSY, HSQC, HMBC): $\delta$ (ppm) 10.0, 24.0, 25.5, $28.5,28.6,38.4,38.9,40.8,41.7,53.9,62.7,65.0,112.7,114.7,118.0,121.0,126.9,127.2,129.1$, 132.7, 140.3, 143.1, 158.5, 159.5, 166.6, 173.0. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{30} \mathrm{H}_{42} \mathrm{~N}_{7} \mathrm{O}_{3}{ }^{+}$: 548.3344, found: 548.3349. Anal. calcd. for $\mathrm{C}_{30} \mathrm{H}_{41} \mathrm{~N}_{7} \mathrm{O}_{3}$ : C 65.79, H 7.55, N 17.90, found: C 65.50, H 7.43, N 17.84. $\mathrm{C}_{30} \mathrm{H}_{41} \mathrm{~N}_{7} \mathrm{O}_{3}$ (547.70).

## General procedure for the synthesis of the aminopotentidine derivatives 4.42-4.44

4.25, 4.26 or 4.27 ( 1 eq ) and DIPEA ( 3 eq ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( $15-35 \mathrm{~mL}$ ). O-(Benzotriazol-1-$\mathrm{yl})-\mathrm{N}, \mathrm{N}, \mathrm{N}^{\prime}, \mathrm{N}^{\prime}$-tetramethyluronium-tetrafluoroborate (TBTU, 1.2 eq ) was added and the mixture was stirred for 10-15 min. Subsequently, 4.5 ( 1 eq ) was added and the reaction mixture was stirred over night at room temperature. The organic layer was washed three times with $\mathrm{H}_{2} \mathrm{O}(30$ mL ), two times with aqueous NaOH solution ( $5 \%, \mathrm{w} / \mathrm{w}, 15-30 \mathrm{~mL}$ ) and 1-2 times with brine (15-30 mL ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed in vacuo. The resulting crude product was purified by preparative HPLC.

## N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-(propionamidomethyl)-3-(trifluoromethyl)benzamide (4.42)

4.42 was prepared from $4.25(80 \mathrm{mg}, 0.29 \mathrm{mmol}, 1 \mathrm{eq})$, DIPEA ( $109 \mathrm{mg}, 0.84 \mathrm{mmol}, 3 \mathrm{eq}$ ), TBTU $(105 \mathrm{mg}, 0.34 \mathrm{mmol}, 1.2 \mathrm{eq})$ and $4.5(104 \mathrm{mg}, 0.29 \mathrm{mmol}, 1 \mathrm{eq})$ according to general procedure. Removal of the solvent in vacuo and purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 25: 75-65: 35, t_{\mathrm{R}}=24.2 \mathrm{~min}$ ) afforded the product as white solid (10 mg, 6\%). Mp: 94-96 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 90: 10\right)$. RP-HPLC (gradient 2, 220 nm ): $95.0 \%\left(t_{\mathrm{R}}=17.77 \mathrm{~min}, k=5.1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.03$ (t, 3H, J 7.45 Hz ), 1.37 (br s, 2H), 1.48 (br s, 4H), 1.88-1.92 (m, 2H), $2.20(\mathrm{q}, 2 \mathrm{H}, J 7.59 \mathrm{~Hz}), 2.33(\mathrm{br}$ $\mathrm{s}, 3 \mathrm{H}), 2.88(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 3.24-3.44(8 \mathrm{H}$, interfering with the water signal), $3.95(\mathrm{t}, 2 \mathrm{H}, J 6.20 \mathrm{~Hz}), 4.45$ (d, 2H, J 5.72 Hz$), 6.80-6.85(\mathrm{~m}, 3 \mathrm{H}), 7.09-7.10(\mathrm{~m}, 2 \mathrm{H}), 7.18-7.21(\mathrm{~m}, 1 \mathrm{H}), 7.55(\mathrm{~d}, 1 \mathrm{H}, J 8.17 \mathrm{~Hz})$, $8.07(\mathrm{~d}, 1 \mathrm{H}, J 8.19 \mathrm{~Hz}), 8.14(\mathrm{~s}, 1 \mathrm{H}), 8.40(\mathrm{t}, 1 \mathrm{H} ; J 5.87 \mathrm{~Hz}), 8.80(\mathrm{t}, 1 \mathrm{H}, J 5.59 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150$
$\mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 9.8,23.8,25.4,28.3,28.5,38.4,38.5,38.9,40.6$, $53.7,62.6,65.0,112.9,114.8,118.0,121.0,124.1\left(q, 1 C, J_{C-F} 274.40 \mathrm{~Hz}\right), 124.6\left(q, 1 C, J_{C-F} 5.95 \mathrm{~Hz}\right)$, $125.0,126.1$ ( $q, 1 C, J_{C-F} 30.83 \mathrm{~Hz}$ ), 128.7, 129.1, 131.2, 133.2, 141.0, 158.5, 159.4, 165.1, 173.3. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{31} \mathrm{H}_{41} \mathrm{~F}_{3} \mathrm{~N}_{7} \mathrm{O}_{3}{ }^{+}$: 616.3217, found: 616.3223. $\mathrm{C}_{31} \mathrm{H}_{40} \mathrm{~F}_{3} \mathrm{~N}_{7} \mathrm{O}_{3}$ (615.70).

## 3-Chloro-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4(propionamidomethyl)benzamide (4.43)

4.43 was prepared from 4.26 ( $120 \mathrm{mg}, 0.50 \mathrm{mmol}, 1 \mathrm{eq}$ ), DIPEA ( $193 \mathrm{mg}, 1.49 \mathrm{mmol}, 3 \mathrm{eq}$ ), TBTU ( $191 \mathrm{mg}, 0.60 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) and $4.5(178 \mathrm{mg}, 0.50 \mathrm{mmol}, 1 \mathrm{eq})$ according to general procedure. Removal of the solvent in vacuo and purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 25: 75-70: 30, t_{\mathrm{R}}=20.9 \mathrm{~min}$ ) afforded the product as white solid (138 mg, 48\%). $R_{\mathrm{f}}=0.7\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 90: 10$ ). RP-HPLC (gradient 2, 220 nm ): $99.3 \%\left(t_{\mathrm{R}}=16.61 \mathrm{~min}, k=4.7\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.02$ (t, 3H, J 7.61 Hz ), 1.36 (br s, 2H), 1.48 (br s, 4 H ), 1.91 (qui, $2 \mathrm{H}, J 6.55 \mathrm{~Hz}$ ), $2.23(\mathrm{q}, 2 \mathrm{H}, J 7.56 \mathrm{~Hz}$ ), $2.32(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.24-3.39(\mathrm{~m}, 8 \mathrm{H}$, interfering with the water signal), $3.95(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 6.12 \mathrm{~Hz}), 4.33$ (d, $2 \mathrm{H}, J 5.82 \mathrm{~Hz}), 6.78-6.80(\mathrm{~m}, 1 \mathrm{H}), 6.84-6.85(\mathrm{~m}, 2 \mathrm{H}), 7.07-7.11(\mathrm{~m}, 2 \mathrm{H}), 7.20(\mathrm{t}, 1 \mathrm{H}, J 7.75 \mathrm{~Hz}), 7.37$ (d, 1H, J 7.99 Hz ), 7.75-7.76 (m, 1H), $7.87(\mathrm{~m}, 1 \mathrm{H}), 8.33(\mathrm{t}, 1 \mathrm{H} ; J 5.94 \mathrm{~Hz}), 8.65(\mathrm{t}, 1 \mathrm{H}, J 5.53 \mathrm{~Hz})$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 9.9,23.8,25.4,28.4,28.6,30.7$, 38.4, 39.9 (1C under solvent peak (38.7-40.3)), 40.6, 53.7, 62.6, 65.0, 112.8, 114.8, 118.0, 121.1, 126.0, 127.7, 128.5, 129.1, 131.9, 134.6, 139.7, 140.2, 158.5, 159.4, 165.1, 173.2. HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$, calcd. for $\mathrm{C}_{30} \mathrm{H}_{41} \mathrm{ClN}_{7} \mathrm{O}_{3}{ }^{+}$: 582.2954, found: 582.2960. $\mathrm{C}_{30} \mathrm{H}_{40} \mathrm{ClN}_{7} \mathrm{O}_{3}$ (582.15).

## 3-Bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4(propionamidomethyl)benzamide (4.44)

4.44 was prepared from 4.27 ( $62 \mathrm{mg}, 0.22 \mathrm{mmol}, 1 \mathrm{eq}$ ), DIPEA ( $84 \mathrm{mg}, 0.65 \mathrm{mmol}, 3 \mathrm{eq}$ ), TBTU ( $84 \mathrm{mg}, 0.26 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) and $4.5(78 \mathrm{mg}, 0.22 \mathrm{mmol}, 1 \mathrm{eq})$ according to general procedure. Removal of the solvent in vacuo and purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 35: 65-70: 30, t_{R}=16.2 \mathrm{~min}$ ) afforded the product as white solid ( $60 \mathrm{mg}, 44 \%$ ). $\mathrm{Mp}: 165{ }^{\circ} \mathrm{C}$ decomposition. $R_{\mathrm{f}}=0.7\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 90: 10\right)$. RP-HPLC (gradient 2, 220 nm ): 99.5\% ( $t_{\mathrm{R}}=16.51 \mathrm{~min}, k=4.7$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right.$, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 1.03(\mathrm{t}, 3 \mathrm{H}, J 7.63 \mathrm{~Hz}$ ), 1.35 (br s, 2H), 1.44-1.48 (m, 4H), 1.90 (qui, $2 \mathrm{H}, J 6.48 \mathrm{~Hz}$ ), $2.20(\mathrm{q}, 2 \mathrm{H}, J 7.63 \mathrm{~Hz}), 2.27(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.24-3.29(\mathrm{~m}, 4 \mathrm{H}), 3.33-3.36(\mathrm{~m}, 4 \mathrm{H}$, interfering with the water signal), $3.95(\mathrm{t}, 2 \mathrm{H}, J 6.11 \mathrm{~Hz}), 4.30(\mathrm{~d}, 2 \mathrm{H}, J 5.82 \mathrm{~Hz}), 6.76-6.78(\mathrm{~m}, 1 \mathrm{H})$, 6.82-6.83 (m, 2H), 7.07-7.11 (m, 2H), 7.17-7.20 (m, 1H), $7.34(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J} 8.11 \mathrm{~Hz}), 7.79-7.81(\mathrm{~m}, 1 \mathrm{H})$, 8.04-8.05 (m, 1H), $8.34(\mathrm{t}, 1 \mathrm{H} ; J 5.89 \mathrm{~Hz}), 8.65(\mathrm{t}, 1 \mathrm{H}, J 5.56 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right.$, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 9.9,24.0,25.5,28.4,28.6,38.4,38.9,40.6,42.4,53.9,62.8,65.0$, $112.7,114.6,118.0,120.9,122.0,126.5,128.3,129.0,130.9,134.7,140.3,141.2,158.4,159.4$, 165.0, 173.2. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{30} \mathrm{H}_{41} \mathrm{BrN}_{7} \mathrm{O}_{3}{ }^{+}$: 626.2449, found: 626.2454. $\mathrm{C}_{30} \mathrm{H}_{40} \mathrm{BrN}_{7} \mathrm{O}_{3}$ (626.60).

## 4-(Azidomethyl)-3-bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.45)

4.20 ( $400 \mathrm{mg}, 1.56 \mathrm{mmol}, 1.4 \mathrm{eq}$ ) and DIPEA ( $433 \mathrm{mg}, 3.35 \mathrm{mmol}, 3 \mathrm{eq}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 100 mL ). O-(Benzotriazol-1-yl)-N,N, $\mathrm{N}^{\prime}, \mathrm{N}^{\prime}$-tetramethyluronium-hexafluorophosphate (HBTU, 508 $\mathrm{mg}, 1.56 \mathrm{mmol}, 1.4 \mathrm{eq}$ ) was added and the mixture was stirred for 30 min . Subsequently, 4.5 (400 $\mathrm{mg}, 1.12 \mathrm{mmol}, 1 \mathrm{eq}$ ) was added and the reaction mixture was stirred over night at room temperature. The organic layer was washed two times with saturated aqueous $\mathrm{NaHCO}_{3}$ solution $(100 \mathrm{~mL})$ and with brine ( 100 mL ). Purification by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5-\right.$ $90: 10)$ and removal of the solvent in vacuo afforded the product as yellow oil ( $290 \mathrm{mg}, 44 \%$ ). $R_{f}=$ $0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 90: 10\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm})$ 1.57-1.59 (m, 2H$), 1.72-$ $1.78(\mathrm{~m}, 4 \mathrm{H}), 2.01-2.07(\mathrm{~m}, 2 \mathrm{H}), 2.94(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.40-3.45(\mathrm{~m}, 4 \mathrm{H}), 3.49-3.52(\mathrm{~m}, 2 \mathrm{H}), 3.99(\mathrm{~s}, 2 \mathrm{H})$, $4.07(\mathrm{t}, 2 \mathrm{H}, J 5.78 \mathrm{~Hz}), 4.54(\mathrm{~s}, 2 \mathrm{H}), 6.96-7.00(\mathrm{~m}, 2 \mathrm{H}), 7.09-7.10(\mathrm{~m}, 1 \mathrm{H}), 7.30(\mathrm{t}, 1 \mathrm{H}, J 7.94 \mathrm{~Hz})$, $7.51(\mathrm{~d}, 1 \mathrm{H}, J 7.94 \mathrm{~Hz}), 7.80-7.82(\mathrm{~m}, 1 \mathrm{H}), 8.07-8.08(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta(\mathrm{ppm})$ $22.1,23.4,28.4,39.2,39.3,40.9,53.0,53.7,61.2,65.7,114.9,116.7,122.8,123.3,126.4,128.3$, 129.7, 130.0, 131.6, 132.8, 135.6, 138.7, 159.1, 160.1, 167.2. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{27} \mathrm{H}_{35} \mathrm{BrN}_{9} \mathrm{O}_{2}{ }^{+}: 596.2092$, found: 596.2114. $\mathrm{C}_{27} \mathrm{H}_{34} \mathrm{BrN}_{9} \mathrm{O}_{2}$ (596.53).

## 4-(Aminomethyl)-3-bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.46)

4.45 ( $390 \mathrm{mg}, 0.65 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in a mixture of THF and $\mathrm{H}_{2} \mathrm{O}(1 / 2, \mathrm{v} / \mathrm{v}, 7.5 \mathrm{~mL})$. Triphenylphosphine ( $427 \mathrm{mg}, 1.63 \mathrm{mmol}, 2.5 \mathrm{eq}$ ) dissolved in mixture of THF and $\mathrm{H}_{2} \mathrm{O}(5 / 2, \mathrm{v} / \mathrm{v}, 7$ mL ) was added dropwise and the reaction mixture was stirred over night at room temperature. The solvent was partially removed under reduced pressure and the residue was purified by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 90: 10-\mathrm{CH}_{2} \mathrm{Cl}_{2} / 0.1 \% \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 90: 10\right)$. The product was obtained as yellow oil ( $200 \mathrm{mg}, 54 \%$ ). Further purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 1 \%$ aq. $\mathrm{NH}_{3} 10: 90-91: 9, t_{\mathrm{R}}=19.9 \mathrm{~min}$ ) afforded the product as white fluffy solid ( $70 \mathrm{mg}, 19 \%$ ). $\mathrm{Mp}: 73-77{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.4\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 0.2 \% \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 95: 5$ ). RP-HPLC (gradient 3, 220 nm ): 98.4\% ( $t_{\mathrm{R}}=28.19 \mathrm{~min}, k=16.3$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}\right.$, HSQC, HMBC): $\delta(\mathrm{ppm}) 1.46(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.57-1.61(\mathrm{~m}, 4 \mathrm{H}), 2.02-2.06(\mathrm{~m}, 2 \mathrm{H}), 2.44(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.41-$ $3.43(\mathrm{~m}, 4 \mathrm{H}), 3.48-3.50(\mathrm{~m}, 4 \mathrm{H}), 3.90(\mathrm{~s}, 2 \mathrm{H}), 4.05(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.82 \mathrm{~Hz}), 6.85-6.89(\mathrm{~m}, 2 \mathrm{H}), 6.94(\mathrm{~s}, 1 \mathrm{H})$, $7.21(\mathrm{t}, 1 \mathrm{H}, J 7.92 \mathrm{~Hz}), 7.53(\mathrm{~d}, 1 \mathrm{H}, J 7.99 \mathrm{~Hz}), 7.78-7.80(\mathrm{~m}, 1 \mathrm{H}), 8.04(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, $\left.\mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 25.0,26.3,30.0,40.4,40.7,42.3,46.7,55.3,64.6,66.6$, $114.7,117.0,119.8,123.5,124.3,127.7,130.1,130.3,132.7,135.8,139.5,146.1,160.3,161.5$, 168.9. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{27} \mathrm{H}_{37} \mathrm{BrN}_{7} \mathrm{O}_{2}{ }^{+}$: 570.2187, found: 570.2183. $\mathrm{C}_{27} \mathrm{H}_{36} \mathrm{BrN}_{7} \mathrm{O}_{2}$ (570.54).

## 4-(Acetamidomethyl)-3-bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.47)

4.46 ( $70 \mathrm{mg}, 0.12 \mathrm{mmol}, 1 \mathrm{eq}$ ) and DIPEA ( $32 \mathrm{mg}, 0.25 \mathrm{mmol}, 2 \mathrm{eq}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 mL ). Acetyl chloride ( $14 \mathrm{mg}, 0.18 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ was added drop wise
over a period of 2 h and the reaction mixture was stirred over night at room temperature. The organic layer was further diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ and washed two times with aqueous NaOH solution ( $5 \%, \mathrm{w} / \mathrm{w}, 20 \mathrm{~mL}$ ) and with brine ( 20 mL ). Removal of the solvent in vacuo and purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}: \mathrm{MeCN} / 0.1 \% \mathrm{aq} . \mathrm{NH}_{3}$ 20:80-91:9, $t_{\mathrm{R}}=16.05 \mathrm{~min}$ ) afforded the product as white fluffy solid ( $30 \mathrm{mg}, 40 \%$ ). The purity determined by analytical HPLC was under $95 \%$ and therefore the product was again purified by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: MeCN/0.1\% aq. $\mathrm{NH}_{3}$ 10:90-70:30, $t_{R}=23.7 \mathrm{~min}$ ). The product was obtained as white fluffy solid ( $15 \mathrm{mg}, 20 \%$ ). Mp: 89-95 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.3$ ( $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 0.2 \% \mathrm{NH}_{3}$ in MeOH 90:10). RP-HPLC (gradient 3, 220 nm ): 99.0\% ( $t_{\mathrm{R}}=28.01 \mathrm{~min}, k=16.2$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.45(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.56-1.60(\mathrm{~m}, 4 \mathrm{H}), 2.01-$ $2.05(\mathrm{~m}, 5 \mathrm{H}), 2.43(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.40-3.42(\mathrm{~m}, 4 \mathrm{H}), 3.47-3.49(\mathrm{~m}, 4 \mathrm{H}), 4.04(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.81 \mathrm{~Hz}), 4.44(\mathrm{~s}$, $2 \mathrm{H}), 6.84-6.88(\mathrm{~m}, 2 \mathrm{H}), 6.93(\mathrm{~m}, 1 \mathrm{H}), 7.20(\mathrm{t}, 1 \mathrm{H}, J 7.90 \mathrm{~Hz}), 7.39(\mathrm{~d}, 1 \mathrm{H}, J 8.04 \mathrm{~Hz}), 7.74-7.76(\mathrm{~m}$, $1 \mathrm{H}), 8.03(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \operatorname{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 22.5,25.0,26.3$, $30.0,40.3,40.7,42.3,44.5,55.3,64.6,66.6,114.8,117.0,119.8,123.5,124.1,127.5,130.1$, 130.3, 132.8, 136.1, 142.4 (2C), 160.3, 161.5, 168.8, 173.4. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{29} \mathrm{H}_{39} \mathrm{BrN}_{7} \mathrm{O}_{3}{ }^{+}: 612.2292$, found: 612.2298. $\mathrm{C}_{29} \mathrm{H}_{38} \mathrm{BrN}_{7} \mathrm{O}_{3}$ (612.57).

## 4-(Acetamidomethyl)-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.48)

4.40 ( $70 \mathrm{mg}, 0.14 \mathrm{mmol}, 1 \mathrm{eq}$ ), 4-(dimethylamino)-pyridine ( $17 \mathrm{mg}, 0.14 \mathrm{mmol}, 1 \mathrm{eq}$ ) and DIPEA $(28 \mathrm{mg}, 0.21 \mathrm{mmol}, 1.5 \mathrm{eq})$ were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$. Acetyl chloride ( $12 \mathrm{mg}, 0.16 \mathrm{mmol}$, $1.1 \mathrm{eq})$ dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ was added drop wise and the reaction mixture was stirred over night at room temperature. Additional acetyl chloride ( $12 \mathrm{mg}, 0.16 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and DIPEA ( 28 $\mathrm{mg}, 0.21 \mathrm{mmol}, 1.5 \mathrm{eq})$ were added and mixture was again stirred over night at room temperature. The organic layer was further diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$ and washed with $\mathrm{H}_{2} \mathrm{O}(50$ mL ), aqueous NaOH solution ( $1 \mathrm{M}, 50 \mathrm{~mL}$ ) and brine ( 50 mL ). The solvent was removed under reduced pressure and the residue was purified by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 0-\right.$ $90: 10$ ). The product was obtained as colourless oil ( $40 \mathrm{mg}, 53 \%$ ). Further purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: MeCN/0.1\% aq. $\mathrm{NH}_{3}$ 20:80-91:9, $t_{\mathrm{R}}$ $=14.02 \mathrm{~min}$ ) afforded the product as white fluffy solid ( $19 \mathrm{mg}, 25 \%$ ). Mp: $75-85{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.2$ ( $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.5 \mathrm{~N} \mathrm{NH}_{3}$ in MeOH 90:10). RP-HPLC (gradient 3, 220 nm ): 99.8\% ( $t_{\mathrm{R}}=25.91 \mathrm{~min}, k=14.9$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.45(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.56-1.60(\mathrm{~m}, 4 \mathrm{H}), 2.00$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 2.02-2.06 (qui, $2 \mathrm{H}, J 6.22 \mathrm{~Hz}$ ), 2.41 (br s, 4H), 3.41-3.43 (m, 4 H$), 3.46(\mathrm{~s}, 2 \mathrm{H}), 3.49-3.51(\mathrm{~m}$, $2 \mathrm{H}), 4.05(\mathrm{t}, 2 \mathrm{H}, J 5.87 \mathrm{~Hz}), 4.40(\mathrm{~s}, 2 \mathrm{H}), 6.84-6.89(\mathrm{~m}, 2 \mathrm{H}), 6.94(\mathrm{~m}, 1 \mathrm{H}), 7.21(\mathrm{t}, 1 \mathrm{H}, \mathrm{J} 7.87 \mathrm{~Hz})$, 7.35 (d, 2H, J 8.40 Hz ), 7.77 (m, 2H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \operatorname{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm})$ 22.5, 25.1, 26.4, 30.0, 40.3, 40.7, 42.4, 43.8, 55.3, 64.7, 66.6, 114.7, 116.9, 119.9, 123.4, 128.58, 128.60, 130.2, 134.2, 139.8, 144.1, 160.3, 161.6, 170.5, 173.2. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{29} \mathrm{H}_{40} \mathrm{~N}_{7} \mathrm{O}_{3}{ }^{+}: 534.3187$, found: 534.3191. $\mathrm{C}_{29} \mathrm{H}_{39} \mathrm{~N}_{7} \mathrm{O}_{3}$ (533.68).

## N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4((dimethylamino)methyl)benzamide (4.49)

4.28 ( $33 \mathrm{mg}, 0.15 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and DIPEA ( $90 \mathrm{mg}, 0.70 \mathrm{mmol}, 5 \mathrm{eq}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10$ mL ). O -(Benzotriazol-1-yl)-N, $\mathrm{N}, \mathrm{N}^{\prime}, \mathrm{N}^{\prime}$-tetramethyluronium-hexafluorophosphate (HBTU, 74 mg , $0.20 \mathrm{mmol}, 1.4 \mathrm{eq})$ was added and the mixture was stirred for 30 min . Subsequently, 4.5 ( 50 mg , $0.14 \mathrm{mmol}, 1 \mathrm{eq})$ was added and the reaction mixture was stirred over night at room temperature. The organic layer was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$ and washed with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$, aqueous NaOH solution ( $1 \mathrm{M}, 50 \mathrm{~mL}$ ) and brine ( 50 mL ). Purification by automated flash chromatography ( $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ 100:0-80:20 in 20 min ) and removal of the solvent in vacuo afforded the product as colourless oil ( $40 \mathrm{mg}, 56 \%$ ). Further purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 10: 90-91: 9, t_{\mathrm{R}}=21.50 \mathrm{~min}$ ) afforded the product as white fluffy solid ( 20 mg , 28\%). Mp: 62-68 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH}\right.$ in MeOH 95:5). RP-HPLC (gradient 3, 220 nm ): $99.0 \%\left(t_{\mathrm{R}}=30.29 \mathrm{~min}, k=17.6\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, $\left.\mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 1.45(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 1.57-1.61(\mathrm{~m}, 4 \mathrm{H}), 2.02-2.06$ (qui, $2 \mathrm{H}, \mathrm{J} 6.20$ $\mathrm{Hz}), 2.24(\mathrm{~s}, 6 \mathrm{H}), 2.43(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.41-3.43(\mathrm{~m}, 4 \mathrm{H}), 3.48-3.51(\mathrm{~m}, 6 \mathrm{H}), 4.05(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.87 \mathrm{~Hz}), 6.85-$ $6.89(\mathrm{~m}, 2 \mathrm{H}), 6.94-6.95(\mathrm{~m}, 1 \mathrm{H}), 7.21(\mathrm{t}, 1 \mathrm{H}, J 7.85 \mathrm{~Hz}), 7.40(\mathrm{~d}, 2 \mathrm{H}, J 8.22 \mathrm{~Hz}), 7.78-7.79(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 25.1,26.4,30.0,40.3,40.6,42.4,45.3$, $55.3,64.5,64.6,66.6,114.7,117.0,119.9,123.4,128.4,130.3,130.7,134.5,139.6,142.9,160.3$, 161.6, 170.5. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{29} \mathrm{H}_{42} \mathrm{~N}_{7} \mathrm{O}_{2}{ }^{+}$: 520.3395, found: 520.3394. $\mathrm{C}_{29} \mathrm{H}_{41} \mathrm{~N}_{7} \mathrm{O}_{2}$ (519.69).

## 3-Bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4((dimethylamino)methyl)benzamide (4.50)

4.29 ( $60 \mathrm{mg}, 0.20 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) and DIPEA ( $120 \mathrm{mg}, 0.93 \mathrm{mmol}, 5 \mathrm{eq}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(10 \mathrm{~mL}) . O-(B e n z o t r i a z o l-1-y \mathrm{l})-\mathrm{N}, \mathrm{N}, \mathrm{N}^{\prime}, \mathrm{N}^{\prime}$-tetramethyluronium-hexafluorophosphate ( $98 \mathrm{mg}, 0.26$ mmol, 1.4 eq ) was added and the mixture was stirred for 30 min . Subsequently, 4.5 ( $66 \mathrm{mg}, 0.19$ $\mathrm{mmol}, 1 \mathrm{eq}$ ) was added and the reaction mixture was stirred over night at room temperature. The organic layer was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$ and washed with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$, aqueous NaOH solution ( $1 \mathrm{M}, 50 \mathrm{~mL}$ ) and brine ( 50 mL ). Purification by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} 100 \%\right.$ $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 90: 10$ ) and removal of the solvent in vacuo afforded the product as colourless oil ( $50 \mathrm{mg}, 45 \%$ ). Further purification by preparative HPLC (column: YMC Triart $\mathrm{C}_{18}$, gradient: 0-30 min: $\mathrm{MeCN} / 0.1 \%$ aq. $\mathrm{NH}_{3} 10: 90-91: 9, t_{R}=22.38 \mathrm{~min}$ ) afforded the product as white fluffy solid ( $36 \mathrm{mg}, 32 \%$ ). Mp: 105-115 ${ }^{\circ} \mathrm{C}$. $R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.5 \mathrm{M} \mathrm{NH}_{3}\right.$ in MeOH 90:10). RP-HPLC (gradient 3, 220 nm ): 99.6\% ( $t_{\mathrm{R}}=33.11 \mathrm{~min}, k=19.3$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}\right.$, HMBC): $\delta(\mathrm{ppm}) 2.06(\mathrm{br} s, 2 \mathrm{H}), 1.56-1.60(\mathrm{~m}, 4 \mathrm{H}), 2.02-2.06$ (qui, $2 \mathrm{H}, \mathrm{J} 6.21 \mathrm{~Hz}$ ), $2.29(\mathrm{~s}, 6 \mathrm{H}), 2.41$ (br s, 4H), 3.41-3.45 (m, 6H), 3.49-3.51 (m, 2H), 3.61 ( $\mathrm{s}, 2 \mathrm{H}$ ), $4.05(\mathrm{t}, 2 \mathrm{H}, J 5.87 \mathrm{~Hz}), 6.84-6.89(\mathrm{~m}$, $2 \mathrm{H}), 6.93(\mathrm{~m}, 1 \mathrm{H}), 7.20(\mathrm{t}, 1 \mathrm{H}, J 7.85 \mathrm{~Hz}), 7.53(\mathrm{~d}, 1 \mathrm{H}, J 8.04 \mathrm{~Hz}), 7.76-7.78(\mathrm{~m}, 1 \mathrm{H}), 8.05(\mathrm{~d}, 1 \mathrm{H}, J$ 1.80 Hz ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 25.1,26.4,30.0,40.3,40.7$, $42.3,45.6,55.3,63.6,64.7,66.6,114.7,116.9,119.8,123.4,125.9,127.3,130.2,132.4,132.9$, 136.1, 139.8, 142.4, 160.2, 161.5, 168.9. HRMS: (ESI): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{29} \mathrm{H}_{41} \mathrm{BrN}_{7} \mathrm{O}_{2}{ }^{+}$: 598.2500, found: 598.2502. $\mathrm{C}_{29} \mathrm{H}_{40} \mathrm{BrN}_{7} \mathrm{O}_{2}$ (598.59).

### 4.3.3 Pharmacological Methods

## Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of Sf9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{s \alpha s}$ fusion protein or coexpressing the $h H_{3} R+G_{i \alpha 2}+\beta_{1} \psi_{2}$ proteins was described elsewhere. ${ }^{19}$

Radioligand competition binding assays were performed as described previously with minor adjustments using the following radioligands: $\left[{ }^{3} \mathrm{H}\right] \mathrm{UR}-\mathrm{DE} 257^{6}\left(\mathrm{hH} \mathrm{H}_{2} \mathrm{R}\right.$ : specific activity $=32.89$ $\mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{d}}=12.2 \mathrm{nM}, \mathrm{c}_{\text {final }}=20 \mathrm{nM}$ ) or $\left[{ }^{3} \mathrm{H}\right] N^{\alpha}$-methylhistamine (Hartmann Analytic, Braunschweig, Germany; $\mathrm{hH}_{3} \mathrm{R}$ : specific activity $=80 \mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{d}}=3 \mathrm{nM}, \mathrm{c}_{\text {final }}=3 \mathrm{nM}$ ).

On the day of the experiment $\mathrm{Sf9}$ membranes were thawed and sedimented by centrifugation at $13,000 \mathrm{rpm}$ at $4^{\circ} \mathrm{C}$ for 10 min . The membranes were resuspended in ice cold binding buffer ( 75 mM Tris/ $\mathrm{HCl}, \mathrm{pH} 7.4$ containing $12.5 \mathrm{mM} \mathrm{MgCl}_{2}, 1 \mathrm{mM}$ EDTA and; in the following referred to as BB) and adjusted to a protein concentration of $2-4 \mu \mathrm{~g} / \mu \mathrm{L} .80 \mu \mathrm{~L}$ BB containing $0.2 \% \mathrm{BSA}$ and the respective radioligand, followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ or 10 mM HCl , prepared less than 10 min prior), were added to every well of a 96-well plate (Primaria clear flat bottom microplates, Corning, New York, USA). Incubation was started by addition of the membrane suspension $(10 \mu \mathrm{~L})$. The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with $0.3 \%$ polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with $200 \mu \mathrm{~L}$ scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h . The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

## Functional GTP $\boldsymbol{\gamma}$ S assay on Sf9 insect cell membranes

GTP $\gamma$ S assays were performed as described previously ${ }^{20}$ with minor modifications. $\left[{ }^{35}\right.$ S]GTP $\gamma$ S (specific activity $=1000 \mathrm{Ci} / \mathrm{mmol}$ ) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to 0.5-1.5 $\mu \mathrm{g} / \mu \mathrm{L}$.

Agonist mode: $80 \mu \mathrm{~L}$ of BB containing BSA ( $0.05 \%$ final), GDP ( $1 \mu \mathrm{M}$ final) and [ ${ }^{35} \mathrm{~S}$ ]GTP $\gamma \mathrm{S}$ ( 20 nCi final), followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ ) were added to every well of a 96 -well plate (Primaria clear flat bottom microplates, Corning, New York, USA). Incubation was started by addition of the membrane suspension ( $10 \mu \mathrm{~L}$ ). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTP $\gamma$ S assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine ( $1 \mu \mathrm{M}$ final).

### 4.3.4 Data analysis

Retention factors $k$ were calculated according to $k=\left(t_{\mathrm{R}}-t_{0}\right) / t_{0}\left(t_{0}=\right.$ dead time; $t_{\mathrm{R}}=$ retention time). Corrected counts per minute (ccpm) from the GTPүS assay (agonist mode) were plotted against the $\log$ (concentration of the test compound), and data were analyzed by a four parameter logistic equation (GraphPad Prism Software 5.0, GraphPad Software, San Diego, CA), followed by normalization ( $0 \%=$ water value (basal activity), $100 \%$ = "top" histamine equation) and analysis by four-parameter logistic equation (log(agonist) vs. response - variable slope, GraphPad Prism). Data of the GTP $\gamma$ S assay (antagonist mode) were analysed by a four parameter logistic equation (GraphPad Prism), followed by normalization ( $100 \%=$ "top" of the fourparameter logistic fit, $0 \%=$ unspecifically bound radioligand (ccpm) determined in the presence of famotidine at $100 \mu \mathrm{M}$ ) and analysis by four-parameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism). $\mathrm{p} / C_{50}$ values were converted into $\mathrm{p} K_{\mathrm{B}}$ values according to the Cheng-Prusoff equation ${ }^{21}$. Total binding data from radioligand competition binding experiments were plotted against log(concentration competitor) and analyzed by a fourparameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism), followed by normalization ( $100 \%=$ "top" of the four-parameter logistic fit, $0 \%=$ unspecifically bound radioligand ligand determined in the presence of famotidine at $100 \mu \mathrm{M}$ ). Normalized data from competition binding experiments were again analyzed by a four-parameter logistic equation and obtained $\mathrm{p} / C_{50}$ values were converted into $\mathrm{p} K_{\mathrm{i}}$ values according to the Cheng-Prusoff equation ${ }^{21}$.

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## Chapter 5

## Fluorescence Labeled $\mathrm{H}_{2} \mathrm{R}$ Ligands with BMY25368 Core Structure: Synthesis, Characterization and Application in Flow Cytometry, Confocal Microscopy and High Content Imaging

Note: Prior to the submission of this thesis, parts of this chapter (the synthesis of 5.2, 5.7 and 5.10) were published in cooperation with partners:

Baumeister, P.; Erdmann, D.; Biselli, S.; Kagermeier, N.; Elz, S.; Bernhardt, G.; Buschauer, A. [ ${ }^{3} \mathrm{H}$ ]UR-DE257: Development of a Tritium-Labeled Squaramide-Type Selective Histamine $\mathrm{H}_{2}$ Receptor Antagonist. ChemMedChem 2015, 10, 83-93.

The synthesis of 5.3, 5.12, 5.13 and Py-5 were performed by Mengya Chen during her Master Thesis 2015.

### 5.1 Introduction

Fluorescence labeled GPCR ligands have become an attractive alternative to radiotracers for the investigation of ligand-receptor interactions. Besides advantages with respect to safety issues and waste disposal, fluorescent ligands are a prerequisite for the application of a plethora of optical techniques (confocal microscopy, FRET, ${ }^{1}$ FRAP, ${ }^{2}$ TIRF, ${ }^{3}$ high content imaging, ${ }^{4}$ fluorescence polarization ${ }^{5}$ ). In general, a fluorescent ligand consists of a pharmacophore, a linker and the fluorophore. A major challenge in the development of fluorescent ligands for aminergic GPCRs is to retain affinity, when a bulky fluorophore is attached to a relatively small ligand. It is important to consider that the attachment site, the type and length of the linker as well as the nature of the fluorophore (size, net charge and lipophilicity) might affect receptor affinity as well as selectivity and can lead to unfavorable physicochemical properties (high unspecific binding, aggregation, internalization, etc.). ${ }^{6}$ Nonetheless, various fluorescent ligands for aminergic GPCRs have been reported, for example for muscarinic, ${ }^{7-9} \alpha$ and $\beta$ adrenergic, ${ }^{10-12}$ dopamine, ${ }^{13}$ histamine $H_{1}{ }^{14,15}$ and $\mathrm{H}_{3}{ }^{16-18}$ receptors. In the $\mathrm{H}_{2} \mathrm{R}$ field, most of the reported fluorescent ligands are emitting at wavelengths below 550 nm (common fluorophores: fluorescein, acridine, 5-(dimethylamino)naphthalin-1-sulfonic acid amide (dansyl), N -methylanthranilic acid amide and 1-cyanoisoindol-2-yl). ${ }^{19,20}$ The majority of the fluorescent ligands consist of a piperidinomethylphenoxypropylamino (potentidine) pharmacophore, derived either from roxatidine or iodoaminopotentidine, which is linked to the fluorophore by an alkyl chain (Figure 5.1).

$\mathbf{R}=$


Roxatidine acetate ${ }^{\mathbf{a}}$
$\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pA}_{2}$ : 7.41


Compound IV ${ }^{\text {c }}$
(7-Nitrobenzo[c][1,2,5]oxadiazol-4-yl labeled) $\mathrm{rH}_{2} \mathrm{R}: \mathrm{p}_{\mathrm{i}}: 7.00$

lodoaminopotentidine ${ }^{\text {b }}$
$\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pK}_{\mathrm{i}}: 9.15 \mathrm{nM}$
$K_{\mathrm{d}}: 0.34 \mathrm{nM}$


Compound $\mathrm{V}^{\mathrm{a}}$
(7-Nitrobenzo[c][1,2,5]oxadiazol-4-yl labeled) $\mathrm{gpH}_{2} \mathrm{R}: \mathrm{pA}_{2}: 7.96$

Figure 5.1. Structures of the standard $\mathrm{H}_{2} \mathrm{R}$ antagonists roxatidine acetate and iodoaminopotentidine ${ }^{21}$ as well as fluorescently labeled derivatives ${ }^{19,20} \cdot \mathrm{pA}_{2}$ values were ${ }^{\text {a }}$ determined on the isolated guinea pig right atrium ${ }^{19}$ and $\mathrm{p} K_{\mathrm{i}}$ values and the $\mathrm{p} K_{\mathrm{d}}$ value were ${ }^{\mathrm{b}}$ determined on membranes of guinea pig striatum ${ }^{21,22}$ and ${ }^{\mathrm{c}}$ determined on homogenates of COS-7 cells transiently expressing the $\mathrm{rH}_{2} \mathrm{R}^{20}$.

Labeling with relatively small chromophores such as the N -methylanthranilic acid amide or a 7-nitrobenzo[c][1,2,5]oxadiazol-4-yl moiety resulted in fluorescent ligands (Compound III-V) with high $\mathrm{H}_{2} \mathrm{R}$ affinity ( $\mathrm{p} A_{2}$ or $\mathrm{p} K_{\mathrm{i}}$ values: >7.0) (Figure 5.1). ${ }^{19,20}$ However, the reported ligands were inapplicable for cell-based methods like confocal microscopy and flow cytometry due to the high cellular autofluorescence at the emission wavelength which resulted in low signal-to-noise ratios. In order to expand the range of applications and avoid the high cellular autofluorescence,
fluorescent ligands labeled with red-emitting fluorophores (emission wavelength > 600 nm ) are required.

Recently, with the squaramide-type radioligand $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 our group developed a highaffinity and highly subtype selective $\mathrm{hH}_{2} \mathrm{R}$ antagonist ( $K_{\mathrm{d}}$ value: 31 nM ) ${ }^{23}$ (Figure 5.2) consisting of the pharmacophore of BMY $2536^{24}$, which is linked to the tritium labeled propionic acid amide by a hexyl linker. Replacing the radiolabeled propionic acid amide by red-emitting fluorophores was the starting point for the development of high affinity $\mathrm{hH}_{2} \mathrm{R}$ fluorescent ligands. ${ }^{25}$ The most promising results were achieved by the pyridinium (Py-5) labeled ligand UR-DE229 and the cyanine labeled ligand (S0536) UR-DE56 (Figure 5.2). ${ }^{25,26}$ Both ligands were antagonists in the GTPase assay with $\mathrm{p} K_{\mathrm{b}}$ values of $7.06-7.66$ and showed a low unspecific binding in flow cytometric binding assays and confocal microscopy.


Figure 5.2. Chemical structures of the $\mathrm{H}_{2} \mathrm{R}$ radioligand $\left[{ }^{3} \mathrm{H}\right]$ UR-DE257, the parent compound $B M Y 2536$, as well as of the pyridinium labeled $\mathrm{H}_{2} \mathrm{R}$ antagonist UR-DE229 and the cyanine labeled $\mathrm{H}_{2} \mathrm{R}$ antagonist UR-DE56. Binding affinities (radioligand binding assay, $K_{\mathrm{d}}$ value) ${ }^{23}$ and antagonism (steady-state GTPase assay, $\mathrm{p} K_{\mathrm{b}}$ value) ${ }^{23,25}$ were determined on membranes of Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {saS }}$ fusion protein.

The present study is aiming at fluorescent high affinity $\mathrm{H}_{2} \mathrm{R}$ antagonists with improved optical and physicochemical properties to gain access to a wide range of potential applications, in particular to confocal microscopy and to high throughput or/and high content imaging systems. Therefore, the fluorescent labeled antagonists UR-DE229 and UR-DE56 were investigated in different assay systems. Furthermore, a small library of fluorescent ligands were synthesized for the exploration of the impact of length of the alkyl linker and the variation of the net charge of the fluorophores by coupling the positively charged pyrilium dye (Py-5) or differently charged cyanine dyes (positive: S2197, neutral: S0536 or negative: S0586, succinimidyl esters) with various amine precursors (Figure 5.3).


Figure 5.3. Chemical structures of the fluorescent dyes used for the preparation of the fluorescent $\mathrm{H}_{2} \mathrm{R}$ ligands.

### 5.2 Results and Discussion

### 5.2.1 Chemistry

Starting from 3-[3-(piperidin-1-ylmethyl)phenoxy]propan-1-amine (4.3) the amine precursors were prepared in a three step synthesis (Scheme 5.1 ) as described before. ${ }^{27} 4.3$ was treated with 3,4-diethoxycyclobut-3-ene-1,2-dione (5.1) to form the corresponding squaric acid ester amide 5.2, before a second amidation with mono-boc protected alkanediamines (5.3-5.5) was performed. The intermediates 5.3-5.5 were accessible by reaction of di-tert-butyl dicarbonate ( $\mathrm{Boc}_{2} \mathrm{O}$ ) with an excess of the alkanediamines. The resulting tert-butyl carbamates 5.6-5.8 were cleaved with either TFA or HCl and the products were purified by preparative HPLC to afford the amine precursors 5.9-5.11 as TFA-salts.


Scheme 5.1. Synthesis of the amine precursors 5.9, 5.10, 5.11. Reagents and conditions: i) EtOH, RT, $2 \mathrm{~h}, 78 \%$; ii) $\mathrm{Boc}_{2} \mathrm{O}$, $\mathrm{CHCl}_{3}, 0^{\circ} \mathrm{C}$ to RT, ON, 44-92\%; iii) EtOH, RT, ON, 42-83\%; iv) TFA or $\mathrm{HCl}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 65-93 \%$.

BMY 25368 was synthesized from 5.2 by treatment with ammonia in MeOH (Scheme 5.2) according to published protocols. ${ }^{24,28}$

The pyridinium labeled fluorescent ligands 5.12-5.14 were synthesized by direct coupling of the pyrilium dye Py-5 (chameleon label) with the respective amine precursor 5.9-5.11 under basic conditions (Scheme 5.3). ${ }^{29}$ The reaction progresses rapidly accompanied by a change in color from blue to red. The cyanine labeled fluorescent ligands $5.15-5.20$ were derived from the amine precursors 5.9-5.11 by amide coupling with the succinimidyl ester of the fluorescent dyes (S2197, S0536, S0586).


Scheme 5.2. Synthesis of the parent compound BMY 25368. Reagents and conditions: i) EtOH, RT, ON, 74\%.


Scheme 5.3. Synthesis of the fluorescent compounds 5.12-5.20. Reagents and conditions: i) DMF, TEA or DIPEA, RT, 90120 min, $24-32 \%$; ii) DMF, DIPEA, RT, 45-90 min, 18-44\%.

### 5.2.2 Fluorescence properties of the labeled ligands

The fluorescence properties of representative compounds (5.13, 5.14, 5.16, 5.18, 5.20) were determined in PBS at pH 7.4 and PBS containing $1 \%(\mathrm{w} / \mathrm{v})$ of BSA. The excitation and corrected emission spectra of the pyridinium labeled compound $\mathbf{5 . 1 4}$ and the cyanine labeled compounds (5.16, 5.18, 5.20) are depicted in Figure 5.4. The Stoke's shift was much more pronounced for the Py-5 labeled compounds (Figure 5.4, Table 5.1). These compounds (5.12-5.14) can be excited with the argon laser at 488 nm . The cyanine labeled compounds showed a considerably lower Stoke's shift, allowing excitation by the red diode laser at 635 nm .


Figure 5.4. Excitation and corrected emission spectra of A: Py-5 (5.14), B: S0223 (5.16), C: S0436 (5.18) and D: S0387 (5.20) labeled compounds dissolved in PBS containing 1\% BSA. Ligand concentration: 1.5-6 $\mu \mathrm{M}$. Excitation and emission spectra were recorded with an excitation slit of 10 nm and an emission slit of 10 nm . The excitation wavelength was chosen as close to the absorption maximum as possible (5.14) or at an inflection point (5.16, 5.18 and 5.20 ). The emission wavelength was chosen close to the emission maximum (5.14) or at an inflection point (5.16, 5.18 and 5.20).

The quantum yields (Table 5.1) were determined in PBS and in PBS containing 1\% (w/v) of BSA. For the pyridinium labeled compounds 5.13 and 5.14 the quantum yields were very low in PBS alone (less than $4 \%$ ), whereas the addition of BSA increased the quantum yield up to $24 \%$ ( 7 fold). In case of the investigated cyanine labeled compounds 5.16, 5.18, 5.20 the quantum yields were significantly higher in PBS ( $19-27 \%$ ) and increased ( $38-45 \%$ ) in the presence of BSA. This effect was described for several cyanine and pyridinium labeld ligands primarily resulting from intermolecular hydrophobic and electrostatic interactions of the fluorophores with the protein. ${ }^{30,31}$

Table 5.1. Fluorescence properties of compounds $\mathbf{5 . 1 3}, \mathbf{5 . 1 4}, \mathbf{5 . 1 6}$, $\mathbf{5 . 1 8}$ and $\mathbf{5 . 2 0}$ in PBS (pH 7.4) and PBS containing $1 \%$ BSA: excitation/emission maxima and fluorescent quantum yields $\phi$ (reference: cresyl violet perchlorate).

| Compound | Dye ${ }^{\text {a }}$ | PBS |  | PBS + 1\% BSA |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | $\lambda_{\text {ex }} / \lambda_{\text {em }}$ | $\phi$ (\%) | $\lambda_{\text {ex }} / \lambda_{\text {em }}$ | $\boldsymbol{\phi}$ (\%) |
| 5.13 | Py-5 | 457/706 | 3.2 | 481/646 | 22.9 |
| 5.14 | Py-5 | 458/705 | 3.7 | 481/646 | 24.4 |
| 5.16 | S0223 | 646/663 | 26.5 | 663/672 | 44.6 |
| 5.18 | S0436 | 648/669 | 19.3 | 667/676 | 41.7 |
| 5.20 | S0387 | 649/669 | 24.2 | 656/670 | 38.3 |

${ }^{\text {a }}$ Fluorescent dye used for the preparation of the fluorescent ligand.

### 5.2.3 Biological Evaluation

## $H_{2} R$ affinity, selectivity and antagonism

The fluorescent $\mathrm{H}_{2} \mathrm{R}$ antagonists $5.12-5.20$ as well as the parent compound BMY2536 and the amine precursors 5.9-5.11 were investigated in equilibrium competition binding experiments on membrane preparations from Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein using the antagonist $\left[{ }^{3} H\right] U R-D E 257^{23}$ as radioligand. The selectivity of the compounds for the $h H_{2} R$ compared to $\mathrm{hH}_{3} \mathrm{R}$ and, for selected ligands, to $\mathrm{hH}_{4} \mathrm{R}$ was investigated by competition binding experiments using Sf9 insect cell membranes co-expressing either the $\mathrm{hH}_{3} \mathrm{R}$ or the $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ and $\mathrm{G}_{\beta 1 \gamma^{2}}$ proteins with $\left[{ }^{3} \mathrm{H}\right]$ histamine as radioligand. Representative radioligand binding curves are depicted in Figure 5.5 and the results are shown in Table 5.2.

Radioligand competition binding experiments revealed that most of the fluorescent labels were tolerated with no or a slight decrease in affinity (Table 5.2). Exceptions were the cyanine labeled ligands 5.19 and 5.20 in which the introduction of the $S 0387$ fluorophore with a negative net charge resulted in a decrease in $\mathrm{hH}_{2} \mathrm{R}$ affinity (5.19: $\mathrm{p} K_{\mathrm{i}}$ value: 5.69, 5.20: $\mathrm{p} K_{\mathrm{i}}$ value: 5.88) compared to the corresponding amine-precursors (5.9: $\mathrm{p} K_{\mathrm{i}}$ value: 6.52, 5.10: $\mathrm{p} K_{\mathrm{i}}$ value: 7.87). The pyridinium labeled ligands 5.12-5.14 showed, independent of linker length, high $\mathrm{hH}_{2} \mathrm{R}$ affinities ( $\mathrm{p} K_{\mathrm{i}}$ values: 7.71-7.76) in the same range as the parent compound BMY 25368 ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.80 ). In case of 5.12, the $\mathrm{hH}_{2} \mathrm{R}$ affinity even increased with the labeling (amine precursor 5.9: $\mathrm{p} K_{\mathrm{i}}$ value: 6.52 compared with 5.12: $\mathrm{p} K_{\mathrm{i}}$ value: 7.75 ). In the cyanine series, ligand 5.16 , labeled with fluorophore S0223 (positive net charge), and ligand 5.18, labeled with fluorophore S0436 (neutral net charge), showed the highest $\mathrm{hH}_{2} \mathrm{R}$ affinity (5.16: $\mathrm{p} K_{\mathrm{i}}$ value: 7.67 and 5.18: $\mathrm{p} K_{\mathrm{i}}$ value: 7.11).


Figure 5.5. Displacement of the radioligand [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 ( $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ) by the fluorescent ligands 5.14, 5.16, 5.18 and $\mathbf{5 . 2 0}$ determined on membrane preparations of $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (A). Displacement of the respective radioligand from membrane preparations of Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}$ $\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: $\left[{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ), co-expressing the $\mathrm{hH}_{3} R$ and $\mathrm{G}_{\alpha i 2}$ plus $\mathrm{G}_{\beta 1 \gamma 2}$ proteins (radioligand: $\left[{ }^{3} \mathrm{H}\right]$ histamine $\mathrm{c}=15 \mathrm{nM}, K_{\mathrm{d}}=12.1 \mathrm{nM}$ ) or co-expressing the $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\alpha i 2}$ plus $\mathrm{G}_{\beta 1 y 2}$ proteins (radioligand: [ ${ }^{3} \mathrm{H}$ ]histamine $\mathrm{c}=10 \mathrm{nM}, K_{\mathrm{d}}=15.9 \mathrm{nM}$ ) by the fluorescent ligand 5.14 (B). Data represent mean values $\pm$ SEM of 2-3 experiments performed in triplicate.

Interestingly, labeling with Py-5, S0223 and S0436 led to an increase in $\mathrm{hH}_{3} \mathrm{R}$ receptor affinity up to two orders of magnitude compared to the corresponding amine precursors (Table 5.2). The pyridinium labeled ligands 5.12-5.14 and the cyanine labeled ligands 5.16 and 5.18 showed slight selectivity for the $\mathrm{hH}_{2} \mathrm{R}$ over $\mathrm{hH}_{3} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ values: 6.4-7.18). In case of the cyanine labeled ligands 5.15 and 5.17, the selectivity even changed in favour of the $\mathrm{hH}_{3} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.0 and 6.8).

Table 5.2. Affinities of the parent compound BMY2536, the amine precursors 5.9-5.11 and fluorescent ligands 5.125.20 to $\mathrm{hH}_{2-4} \mathrm{R}$, obtained from equilibrium competition binding studies on membrane preparations from Sf 9 insect cells, expressing the respective histamine receptor subtype.

| Compound | Dye ${ }^{\text {a }}$ | $\mathrm{n}^{\text {b }}$ | $\mathrm{hH}_{2} \mathrm{R}^{\text {c }}$ |  | $\mathrm{hH}_{3} \mathrm{R}^{\text {d }}$ |  | $\mathrm{hH}_{4} \mathrm{R}^{\text {e }}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $\mathrm{p} K_{\mathrm{i}}$ | N | $\mathrm{p} K_{\text {i }}$ | N | $\mathrm{p} K_{\text {i }}$ | N |
| His | - | - | $6.53 \pm 0.04$ | 3 | $7.8 \pm 0.1$ | 3 | $7.65 \pm 0.03$ | 3 |
| BMY 25368 | - | - | $7.80 \pm 0.01^{23}$ | 3 | $4.66 \pm 0.01$ | 3 | n.d. | - |
| 5.9 | - | 4 | $6.52 \pm 0.04$ | 4 | $4.96 \pm 0.07$ | 3 | n.d. | - |
| 5.10 | - | 6 | $7.87 \pm 0.02$ | 3 | $5.08 \pm 0.03$ | 3 | n.d. | - |
| 5.11 | - | 7 | $7.86 \pm 0.02$ | 3 | $5.12 \pm 0.05$ | 3 | n.d. | - |
| 5.12 | Py-5 | 4 | $7.75 \pm 0.02$ | 3 | $6.4 \pm 0.1$ | 3 | n.d. | - |
| 5.13 | Py-5 | 6 | $7.71 \pm 0.04$ | 3 | $7.11 \pm 0.08$ | 4 | $5.64 \pm 0.06$ | 3 |
| 5.14 | Py-5 | 7 | $7.763 \pm 0.008$ | 3 | $7.01 \pm 0.04$ | 3 | $5.57 \pm 0.07$ | 3 |
| 5.15 | S0223 | 4 | $6.57 \pm 0.02$ | 3 | $7.0 \pm 0.1$ | 3 | n.d. | - |
| 5.16 | S0223 | 6 | $7.67 \pm 0.07$ | 3 | $7.18 \pm 0.03$ | 2 | n.d. | - |
| 5.17 | S0436 | 4 | $6.49 \pm 0.04$ | 3 | $6.8 \pm 0.1$ | 3 | n.d. | - |
| 5.18 | S0436 | 6 | $7.105 \pm 0.003$ | 3 | $6.59 \pm 0.03$ | 2 | n.d. | - |
| 5.19 | S0387 | 4 | $5.69 \pm 0.08$ | 3 | $5.45 \pm 0.04$ | 3 | n.d. | - |
| 5.20 | S0387 | 6 | $5.88 \pm 0.09$ | 3 | n.d. | - | n.d. | - |

${ }^{\text {a }}$ Dye used for the preparation of the fluorescent ligand. ${ }^{\text {b }}$ Length of the linker between the squaric acid amide and the fluorescent dye given as the number of carbon atoms. Competition binding assay on membrane preparations of Sf9 insect cells: ${ }^{c}$ expression of the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ), ${ }^{\mathrm{d}}$ coexpression of the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\alpha \mathrm{i} 2}$ and $\mathrm{G}_{\beta 1 \gamma 2}$ proteins (radioligand: [ ${ }^{3} \mathrm{H}$ ]histamine $\mathrm{c}=15 \mathrm{nM}, K_{\mathrm{d}}=12.1 \mathrm{nM}$ ) or ${ }^{\mathrm{e}}$ coexpresson of the $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ plus $\mathrm{G}_{\beta 1 \gamma_{2}}$ proteins (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ histamine $\mathrm{c}=10 \mathrm{nM}, K_{\mathrm{d}}=15.9 \mathrm{nM}$ ). The incubation period was 60 min . Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means $\pm$ SEM of $N$ independent experiments, each performed in triplicate.

Compounds 5.13 and 5.14 were also investigated for $\mathrm{hH}_{4} \mathrm{R}$ affinity. These ligands showed a high preference for the $h H_{2} R$ over $h H_{4} R$. The impaired $h H_{2} R$ selectivity with respect to the $h H_{3} R$ limited the application of the fluorescent ligands to recombinant systems in which the $h H_{2} R$ was expressed.

The amine precursors 5.9, 5.10, 5.11 as well as representative fluorescent ligands (5.13, 5.14, $5.16,5.18$ ) were investigated for $\mathrm{hH}_{2} \mathrm{R}$ antagonism in a GTP $\gamma \mathrm{S}$ assay on membrane preparations of Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. ${ }^{32}$ The results are summarized in Table 5.3 and representative concentration-response curves derived from GTP $\gamma$ S assays performed in antagonist mode are depicted in Figure 5.6A.

The parent compound BMY 25368, the amine precursors 5.9-5.11 as well as the investigated fluorescent ligands $5.13,5.14,5.16$ and 5.18 were antagonists at $\mathrm{hH}_{2} \mathrm{R}$ in a GTP S S assay. In general, the $\mathrm{p} K_{\mathrm{b}}$ values were in good agreement with the $\mathrm{p} K_{\mathrm{i}}$ values from radioligand competition binding. Only the amine intermediates 5.9 and 5.10 as well as the cyanine labeled ligand 5.18 showed significally lower $\mathrm{p} K_{\mathrm{b}}$ values compared to the $\mathrm{p} K_{\mathrm{i}}$ values (difference up to one order of magnitude). Furthermore, representative fluorescent ligands (5.13, 5.14, 5.16 and 5.19) were investigated for $\mathrm{hH}_{2} \mathrm{R}$ agonism in a $\beta$-arrestin2 recruitment assay (split luciferase complementation) on HEK293T-hH2R- $\beta$ Arr2 cells. ${ }^{33}$ All investigated ligands were antagonists in the $\beta$-arrestin2 recruitment assay, indicating that no $\beta$-arrestin2 mediated internalization of the receptor-ligand-complex took place (Figure 5.6B).


Figure 5.6. Antagonism of the fluorescent ligands 5.13 , $5.14,5.16$ and 5.18 on $\mathrm{hH}_{2} R$ determined in a GTP S assay (antagonistic mode) on membrane preparations of $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (A). In the antagonist mode histamine ( $1 \mu \mathrm{M}$ ) was used for stimulation. Concentration-response curves of histamine, the fluorescent ligands $5.13,5.14,5.16$ and 5.19 on $\mathrm{hH}_{2} \mathrm{R}$ determined by a luciferase complementation assay measuring $\beta$ arrestin2 recruitment on HEK293T-hH2R- 3 Arr2 cells (B). Data represent mean values $\pm$ SEM of 2-3 experiments performed in duplicate ( $\beta$-arrestin2) or triplicate (GTP $\gamma \mathrm{S}$ ).

Table 5.3. $\mathrm{hH}_{2} \mathrm{R}$ antagonism and the calculated $\mathrm{p} K_{\mathrm{b}}$ values of histamine, BMY 25368, the amine-precursors 5.9-5.11 and the fluorescent ligands 5.13, 5.14, $\mathbf{5 . 1 6}$ and $\mathbf{5 . 1 8}$ determined by a GTP $\mathcal{S}$ assay.

| No. | Dye ${ }^{\text {a }}$ | $\mathrm{n}^{\text {b }}$ | $\mathrm{hH}_{2} \mathrm{R}(\mathrm{GTPYS})^{\text {c }}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $\mathrm{p} K_{\mathrm{b}}\left(\mathrm{pEC} \mathrm{C}_{50}\right)$ | N | $\boldsymbol{\alpha}$ |
| His | - | - | $(5.80 \pm 0.06)$ | 9 | 1.0 |
| BMY 25368 | - | - | $7.03 \pm 0.02$ | 2 | $-0.011 \pm 0.007$ |
| 5.9 | - | 4 | $5.8 \pm 0.2$ | 2 | $-0.040 \pm 0.003$ |
| 5.10 | - | 6 | $6.73 \pm 0.08$ | 3 | $-0.025 \pm 0.005$ |
| 5.11 | - | 7 | $7.1 \pm 0.1$ | 3 | $-0.011 \pm 0.006$ |
| 5.13 | Py-5 | 6 | $7.21 \pm 0.04$ | 3 | $-0.016 \pm 0.002$ |
| 5.14 | Py-5 | 7 | $7.9 \pm 0.1$ | 3 | $-0.038 \pm 0.006$ |
| 5.16 | S0223 | 6 | $7.73 \pm 0.04$ | 3 | $-0.026 \pm 0.003$ |
| 5.18 | S0436 | 6 | $6.49 \pm 0.03$ | 3 | $-0.015 \pm 0.005$ |

[^0]
## Confocal microscopy

The $\mathrm{hH}_{2} \mathrm{R}$ binding of the fluorescently labeled ligands $5.14,5.16$ and 5.18 was also investigated by confocal microscopy (Figure 5.7). All investigated ligands were localized at the cell membrane of HEK293T-hH2R-qs5 cells. The ligands 5.14 and 5.18 showed a low unspecific binding at a concentration of 100 nM in the presence of a 300 -fold excess of famotidine, whereas 5.16 showed a higher unspecific binding.


Figure 5.7. Localization of the fluorescent $\mathrm{H}_{2} \mathrm{R}$ ligands 5.14 ( 100 nM ), 5.16 ( 100 nM ) and 5.18 ( 100 nM ) at the membrane of HEK293T-hH $\mathrm{H}_{2} \mathrm{R}-\mathrm{qs} 5$ cells determined by confocal microscopy after 20 min of incubation ( $25{ }^{\circ} \mathrm{C}$ ). Unspecific binding was determined in the presence of famotidine ( 300 -fold excess). All images were acquired with a Zeiss Axiovert 200 M microscope equipped with the LSM 510 Laser scanner. A $63 x / 1.40$ oil immersion objective was used.

Flow cytometric $h_{2}$ R binding studies on HEK293T-hH $\mathbf{H}_{2}$ R-qs5 cells
Fluorescent ligands 5.12-5.18, which showed moderate to high $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ values $>6.0$ ) were used for binding studies by flow cytometry on HEK293T-hH2R-qs5 cells ${ }^{26}$. Representative saturation binding curves are depicted in Figure 5.8 and the results are summarized in Table 5.4. All investigated ligands afforded $K_{\mathrm{d}}$ values which were in good agreement with the $K_{\mathrm{i}}$ values obtained from competition binding experiments with [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 on membranes of Sf9 insect cells. Within the pyridinium labeled ligands 5.12-5.14 the carbon linker length ( $n=4-7$ ) had no significant influence on affinity ( $K_{d}$ value: $14.9-27.9 n M$ ), and unspecific binding was low (under $10 \%$ relative to total binding around the $K_{d}$ ).

The cyanine labeled ligand 5.16 (positive charge of the fluorophore) showed the highest affinity ( $K_{d}$ value: 13.9 nM ) within the cyanine series, but a higher unspecific binding ( $20 \%$ relative to
total binding around the $K_{d}$ value). The introduction of a sulfonic acid group into the cyanine moiety and the associated change of fluorophore charge to neutrality (compound 5.18) resulted in a slight decrease in affinity ( $K_{d}$ value: 48.2 nM ), but had a positive effect regarding unspecific binding (around $10 \%$ relative to total binding around the $K_{d}$ value).


Figure 5.8. Representative flow cytometric saturation binding experiments on HEK293T-hH2R-qs5 cells with the fluorescent ligands (A) 5.12, (B) 5.14, (C) 5.16 and (D) 5.18 . Unspecific binding was determined in the presence of famotidine ( 300 -fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 90 min. Error bars of specific binding represent propagated errors calculated according to the Gaussian law. Error bars of total and unspecific binding represent the SEM. Experiments were performed in dublicate.

Table 5.3. $\mathrm{hH}_{2} \mathrm{R}$ saturation binding data of fluorescent ligands 5.12-5.18 determined by flow cytometry on HEK293$\mathrm{hH}_{2} \mathrm{R}$-qs5 cells in comparison with competition binding data determined on membrane preparations from $\mathrm{Sf9}$ insect cells, expressing the $\mathrm{hH}_{2} \mathrm{R}$.

| No | Dye ${ }^{\text {a }}$ | $\mathrm{n}^{\text {b }}$ | $K_{\text {d }}(\mathrm{nM})^{\text {c }}$ | $K_{\mathrm{i}}(\mathrm{nM})^{\text {d }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 5.12 | Py-5 | 4 | $27.9 \pm 2.0$ | $18 \pm 1$ |
| 5.13 | Py-5 | 6 | $14.9 \pm 2.3$ | $20 \pm 2$ |
| 5.14 | Py-5 | 7 | $19.7 \pm 4.4$ | $23.6 \pm 0.4$ |
| 5.15 | S0223 | 4 | $289 \pm 29$ | $270 \pm 14$ |
| 5.16 | S0223 | 6 | $13.9 \pm 2.2$ | $22 \pm 4$ |
| 5.17 | S0436 | 4 | $684 \pm 95$ | $328 \pm 28$ |
| 5.18 | S0436 | 6 | $48.2 \pm 2.3$ | $78.5 \pm 0.5$ |

${ }^{\text {a }}$ Dye used for the preparation of the fluorescent ligand. ${ }^{\text {b }}$ Length of the linker between the squaric acid amide and fluorescent dye given as the number of carbon atoms. ${ }^{\text {c }}$ Flow cytometric saturation binding experiments on HEK293T$\mathrm{hH}_{2} \mathrm{R}$-qs5 cells. Unspecific binding was determined in the presence of famotidine ( 300 -fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 90 min . Data represent mean $\pm$ SEM from three independent experiments (each performed in duplicate). Specific binding data were analyzed by an equation describing one-site (monophasic) binding. ${ }^{\text {d Competition binding assay on membrane preparations of Sf9 insect cells }}$ $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: [ $\left.{ }^{3} \mathrm{H}\right] \mathrm{UR}-\mathrm{DE} 257, \mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ), cf. results from Table 5.2 here shown as $K_{\mathrm{i}}$ values to facilitate comparison.

The association and dissociation kinetics of $5.14,5.16$ and 5.18 were determined on HEK293T$\mathrm{hH} \mathrm{H}_{2} \mathrm{R}$-qs5 cells at $37^{\circ} \mathrm{C}$ using flow cytometry (cf. Figure 5.9). The Py-5 labeled ligand 5.14 ( $c=50$ nM ) showed a fast association i. e. the plateau was reached after approx. 20 min . The dissociation of 5.14 ( $c=50 \mathrm{nM}, 90 \mathrm{~min}$ pre-incubation) in the presence of famotidine was incomplete. After 150 min , the residual specific binding of the fluorescent ligand amounted to approximately $60 \%$ (curve plateau at 57\%). The investigated cyanine dyes 5.16 and 5.18 (c = 15 nM and 25 nM ) showed much slower association, i. e. the plateau was reached after 140 min . The association is strongly concentration dependent which resulted in the differences in association rate between $5.14(\mathrm{c}=50 \mathrm{nM}$ ) and $\mathbf{5 . 1 6}(\mathrm{c}=15 \mathrm{nM})$ or 5.18 ( $\mathrm{c}=25 \mathrm{nM}$ ). A comparison of the association rate constants ( $k_{\text {on }}$ ) which are concentration independent, revealed that Py-5 ligand 5.14 ( $k_{\text {on }}$ value: $0.0043 \mathrm{~min}^{-1} \mathrm{nM}^{-1}$ ) associated two times faster than 5.16 ( $k_{\text {on }}$ value: 0.002 $\min ^{-1} \mathrm{nM}^{-1}$ ) and 4 times faster 5.18 ( $k_{\text {on }}$ value: $0.00093 \mathrm{~min}^{-1} \mathrm{nM}^{-1}$ ) (Table 5.5). The dissociation of the ligands 5.16 and 5.18 ( 90 min pre-incubation) was incomplete after 150 min , reaching a plateau at $22 \%$ (5.16) and $67 \%$ (5.18) of initially bound ligand. These data suggested a (pseudo)irreversible binding ${ }^{34,35}$ of 5.14, 5.16 and 5.18.


Figure 5.9. Association and dissociation kinetics of $\mathbf{5 . 1 4}(\mathrm{A}, \mathrm{B}), \mathbf{5 . 1 6}(\mathrm{C}, \mathrm{D})$ and $\mathbf{5 . 1 8}(\mathrm{E}, \mathrm{F})$ determined at intact HEK293T$h H_{2} \mathrm{R}$-qs5 cells at $37^{\circ} \mathrm{C}$ by flow cytometry. (A) Association of 5.14 ( $\mathrm{c}=50 \mathrm{nM}$ ) to the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time (nonlinear regression: $k_{\text {obs }}=0.24 \mathrm{~min}^{-1}$ ). Inset: $\ln \left[B_{\mathrm{eq}} /\left(\mathrm{B}_{\mathrm{eq}}-\mathrm{B}_{\mathrm{t}}\right)\right]$ versus time, $k_{\text {obs }}=$ slope $=0.23 \mathrm{~min}^{-1}$. (B) Dissociation of 5.14 (preincubation: $90 \mathrm{~min}, \mathrm{c}=50 \mathrm{nM}$ ) in the presence of famotidine ( $\mathrm{c}=15 \mu \mathrm{M}$ ) from the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{\text {off }}=0.027 \mathrm{~min}^{-1}, t_{1 / 2}=26 \mathrm{~min}$, plateau $=57 \%$ ), Inset: $\ln \left[\left(\mathrm{B}_{\mathrm{t}}-\mathrm{B}_{\text {plateau }}\right) /\left(\mathrm{B}_{0}-\mathrm{B}_{\text {plateau }}\right)\right]$ versus time, slope $\cdot(-1)=k_{\text {off }}=0.023 \mathrm{~min}^{-1}$. (C) Association of 5.16 ( $\left.\mathrm{c}=15 \mathrm{nM}\right)$ to the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time (nonlinear regression: $\left.k_{\text {obs }}=0,061 \mathrm{~min}^{-1}\right)$. Inset: $\ln \left[\mathrm{B}_{\mathrm{eq}} /\left(\mathrm{B}_{\text {eq }}-\mathrm{B}_{\mathrm{t}}\right)\right]$ versus time, $k_{\text {obs }}=$ slope $=0.045 \mathrm{~min}^{-1}$. (D) Dissociation of 5.16 (preincubation: $90 \mathrm{~min}, c=15 \mathrm{nM}$ ) in the presence of famotidine ( $c=4.5 \mu \mathrm{M}$ ) from the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{\text {off }}=0.011 \mathrm{~min}^{-1}, t_{1 / 2}=64 \mathrm{~min}$, plateau $=22 \%)$, Inset: $\ln \left[\left(B_{t}-B_{\text {plateau }}\right) /\left(B_{0}-B_{\text {plateau }}\right)\right]$ versus time, slope $\cdot(-1)=k_{\text {off }}==0.012 \mathrm{~min}^{-1}$. (E) Association of 5.18 ( $C=$ 25 nM ) to the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time (nonlinear regression: $\left.k_{\text {obs }}=0,038 \mathrm{~min}^{-1}\right)$. Inset: $\ln \left[B_{\mathrm{eq}} /\left(\mathrm{B}_{\mathrm{eq}}-B_{t}\right)\right]$ versus time, $k_{\text {obs }}=$ slope $=0,035 \mathrm{~min}^{-1}$. (D) Dissociation of 5.18 (preincubation: $90 \mathrm{~min}, \mathrm{c}=25 \mathrm{nM}$ ) in the presence of famotidine ( $\mathrm{c}=$ $7.5 \mu \mathrm{M}$ ) from the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{\text {off }}=$ $0.015 \mathrm{~min}^{-1}, t_{1 / 2}=75 \mathrm{~min}$, plateau $\left.=67 \%\right)$, Inset: $\operatorname{In}\left[\left(B_{t}-B_{\text {plateau }}\right) /\left(B_{0}-B_{\text {plateau }}\right)\right]$ versus time, slope $\cdot(-1)=k_{\text {off }}=0.012 \mathrm{~min}^{-1}$. Data represent means $\pm$ SEM from two to three independent experiments (each performed in duplicate).

Nevertheless, the equilibrium dissociation constants of 5.14 ( $K_{\mathrm{d}(\mathrm{kin})}$ value: 6.7 nM$)$, 5.16 ( $K_{\mathrm{d}(\mathrm{kin})}$ value: 6.3 nM ) and 5.18 ( $K_{\mathrm{d}(\mathrm{kin})}$ value: 16.7 nM ), calculated from kinetics (nonlinear regression, $\left.K_{\mathrm{d}(\mathrm{kin})}=k_{\text {off }} / k_{\text {on }}\right)$, were consistent with the $K_{\mathrm{d}}$ values obtained from saturation binding experiments (Table 5.5).

Pseudo-irreversible binding to the human, rat and mouse $\mathrm{H}_{2} \mathrm{R}$ was also observed in case of the closely related radioligand [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 using either Sf9 insect cell membranes or HEK293T$\mathrm{hH}_{2} \mathrm{R}$-CreLuc cells. ${ }^{23}$ Squaramides, such as BMY25368 and the amine precursor 5.10, were described as insurmountable $\mathrm{H}_{2} \mathrm{R}$ antagonists. ${ }^{23}$ Unlike standard the antagonist famotidine, both compounds caused a concentration-dependent depression of the maximal agonist response of the guinea pig right atrium.

Several GPCR ligands were reported to show a similar behavior in kinetic and functional experiments ${ }^{34-37}$ and several explanations were provided such as irreversible (covalent) binding to the receptor, ${ }^{38}$ a slow rate of dissociation from the receptor, ${ }^{34}$ a slow rate of interconversion between inactive and active receptor conformations, ${ }^{39}$ stabilization of an inactive ligand-specific receptor conformation, ${ }^{40,41}$ binding to a site distinct from the agonist binding site ${ }^{42}$ or internalization of the ligand-receptor-complex ${ }^{37}$.

Table 5.4. $\mathrm{hH}_{2} \mathrm{R}$ binding characteristics of $\mathbf{5 . 1 4}, \mathbf{5 . 1 6}$ and $\mathbf{5 . 1 8}$ determined by flow cytometry on HEK293T-hH $\mathrm{H}_{2} \mathrm{R}$-qs5 cells at $37^{\circ} \mathrm{C}$.

|  | Binding Kinetics |  |  | Saturation Binding |
| :---: | :---: | :---: | :---: | :---: |
| No | $k_{\text {off }}{ }^{\text {a }}$ <br> $\left(\min ^{-1}\right)$ | $\begin{aligned} & k_{\mathrm{on}}{ }^{\mathrm{b}} \\ & \left(\min ^{-1} \mathrm{nM}^{-1}\right) \end{aligned}$ | $\begin{aligned} & k_{\text {off }} / k_{\text {on }}{ }^{c} \\ & K_{\mathrm{d}(\text { kin. })}(\mathrm{nM}) \end{aligned}$ | $\begin{aligned} & K_{\mathrm{d}(\mathrm{sat})}{ }^{\mathrm{d}} \\ & (\mathrm{nM}) \end{aligned}$ |
| 5.14 | $0.027 \pm 0.002$ | $0.0043 \pm 0.001$ | $6.7 \pm 1.7$ | $19.7 \pm 4.4$ |
| 5.16 | $0.011 \pm 0.003$ | $0.0020 \pm 0.0009$ | $6.3 \pm 2.7$ | $13.9 \pm 2.2$ |
| 5.18 | $0.015 \pm 0.013$ | $0.00093 \pm 0.00021$ | $16.70 \pm 3.3$ | $48.2 \pm 2.3$ |

${ }^{a}$ Dissociation rate constant derived from nonlinear regression. ${ }^{b}$ Association rate constant derived from nonlinear regression; calculated from $k_{\text {obs }}, k_{\text {off }}$ and the fluorescent ligand concentration (cf. Figure 5.9 and experimental section). ${ }^{c}$ Kinetically determined dissociation constant. ${ }^{\text {d Equilibrium dissociation constant determined by saturation binding. }}$ Data represent means $\pm$ SEM from two to three independent experiments (each performed in duplicate).

Regardless of the slow dissociation, the pyridinium labeled ligand 5.14 and the cyanine labeled ligand 5.18 were selected for equilibrium competition binding experiments with HEK293T-hH $\mathrm{h}_{2}$ Rqs5 cells. The results are summarized in Table 5.6 and the competition binding curves are depicted in Figure 5.10. Both ligands were completely displaceable by histamine and standard $\mathrm{H}_{2} \mathrm{R}$ antagonists like famotidine and ICI127032.


Figure 5.10. Displacement of $(A)$ the fluorescent ligand $5.14\left(c=50 n M, K_{d}=19.7 n M\right.$ ) or $[B)$ the fluorescent ligand 5.18 ( $c=25 \mathrm{nM}, K_{\mathrm{d}}=48.2 \mathrm{nM}$ ) by histamine, famotidine and ICl 127032 determined on HEK293T-hH $\mathrm{H}_{2} \mathrm{R}-\mathrm{qs} 5$ cell by flow cytometry. Cells were incubated with the fluorescent ligands and the test compounds at RT in the dark for 90 min. Data represent means $\pm$ SEM from three independent experiments (each performed in duplicate).

The $\mathrm{p} K_{\mathrm{i}}$ values of the standard $\mathrm{H}_{2} \mathrm{R}$ antagonists famotidine and ICl 127032 determined by flow cytometric equilibrium competition binding with the fluorescently labeled ligand 5.14 or 5.18 were in good agreement with data derived from radioligand equilibrium binding experiments with [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 at sf9 insect cell membranes. However, the $\mathrm{p} K_{\mathrm{i}}$ value of the endogenous ligand histamine determined by flow cytometry with either 5.14 or 5.18 was considerably lower compared to radioligand binding at sf9 insect cell membranes. This could be explained by the different receptor-G-protein-complexes used. In the $\mathrm{Sf9}$-system a $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein was utilized, which was reported to have a high affinity ( $\mathrm{p} K_{\text {ih }}$ value: 7.10) and a low affinity binding site ( $\mathrm{p} k_{\mathrm{il}}$ value: 5.08 ) for histamine when $\left[{ }^{3} \mathrm{H}\right]$ tiotidine was used as a radioligand (shallow biphasic competition binding curve). ${ }^{43}$ In case [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 was used as radioligand, the resulting doseresponse curve of histamine was best fitted by a four-parameter logistic fit (shallow monophasic competition binding curve). The corresponding $\mathrm{p} K_{\mathrm{i}}$ value of 6.53 was between the ones of the high affinity and low affinity binding site, which indicates, together with the shallow curve slope, that the high affinity and low affinity binding site are also available but not resolved in this experimental set up. In the HEK293T-hH2R-qs5 cells the $\mathrm{hH}_{2} \mathrm{R}$ was stably co-expressed with the chimeric $\mathrm{G}_{\alpha}$ protein qs5-HA. It was reported, that these cells showed no high affinity binding site for histamine in radioligand equilibrium competition binding experiments with [ ${ }^{3} \mathrm{H}$ ]tiotidine which resulted in monophasic competition binding curves and a $\mathrm{p} K_{\mathrm{i}}$ value of $3.95{ }^{26}$ This $\mathrm{p} K_{\mathrm{i}}$ value was in good agreement with the results determined by flow cytometry using fluorescent ligand 5.14 or 5.18 as competitor and the same cell type (Table 5.5).

Table 5.5. $\mathrm{hH}_{2} \mathrm{R}$ binding ( $\mathrm{p} K_{\mathrm{i}}$ values) of histamine, famotidine and ICl 127032 determined by radioligand competition binding ( $\left[^{3} \mathrm{H}\right]$ UR-DE257 ${ }^{\mathrm{a}}$ or $\left[^{3} \mathrm{H}\right]$ tiotidine ${ }^{\mathrm{b}}$ ) and by displacement of fluorescent ligands $5.14^{\mathrm{c}}$ and $\mathbf{5 . 1 8}{ }^{\mathrm{d}}$, respectively.

|  | Radioligand Binding | Flow Cytometry (5.14) | Flow Cytometry (5.18) $^{\text {d }}$ |
| :--- | :--- | :--- | :--- |
| No | $p K_{\mathrm{i}}$ | $\boldsymbol{p} K_{\mathrm{i}}$ | $p K_{\mathrm{i}}$ |
| Histamine | $6.53 \pm 0.04^{\mathrm{a}}\left(3.93^{\mathrm{b}}\right)^{26}$ | $4.30 \pm 0.04$ | $4.61 \pm 0.05$ |
| Famotidine | $7.25 \pm 0.03^{\mathrm{a}}$ | $7.90 \pm 0.02$ | $7.58 \pm 0.04$ |
| ICI127032 | $7.70 \pm 0.07^{\mathrm{a}}$ | $7.73 \pm 0.02$ | $7.73 \pm 0.03$ |

${ }^{\mathrm{a}, \mathrm{b}}$ Determined by competition binding with ${ }^{\mathrm{a}}{ }^{3} \mathrm{H} \mathrm{H}$ UR-DE257 ( $\mathrm{c}=20 \mathrm{nM}$ ) at membrane preparations of sf9 insect cells expressing $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ or with $\left.{ }^{\mathrm{b}}{ }^{3} \mathrm{H}\right]$ tiotidine $(\mathrm{c}=5 \mathrm{nM})$ at intact HEK293T-hH ${ }_{2} \mathrm{R}-\mathrm{qs} 5$ cells. ${ }^{\mathrm{c}, \mathrm{d}}$ Determined by competition binding with ${ }^{\mathrm{C}} 5.14(\mathrm{c}=50 \mathrm{nM})$ or ${ }^{d} 5.18(\mathrm{c}=25 \mathrm{nM})$ at intact HEK293T-hH R -qs5 cells. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means $\pm$ SEM of 3 independent experiments, each performed in ${ }^{\text {a,b }}$ triplicate or ${ }^{\text {c,d }}$ duplicate.

## High content imaging on adherent and suspended HEK293T-hH2R-qs5 cells

The fluorescent ligands $5.12,5.14-5.18$ and 5.20 were also applied to high content imaging $\mathrm{hH}_{2} \mathrm{R}$ binding assays enabling measurement on live and adherent HEK293T-hH ${ }_{2}$ R-qs5 cells in the 96well plate format. The fluorescent ligands were incubated with the cells for 60 min at room temperature, washed with PBS, and images were obtained by an IN Cell Analyzer 2000 plate reader. Figure 5.11 shows representative images after incubation with the fluorescent ligands $5.14(250 \mathrm{nM}), 5.16(75 \mathrm{nM}), 5.18(75 \mathrm{nM})$ and $\mathbf{5 . 2 0}(500 \mathrm{nM})$ for 60 min , followed by a washing step. All fluorescent ligands were localized at the cell membrane and there was a clear difference between total and unspecific binding.

 ( 500 nM ) at the membrane of HEK293T-hH2R-qs5 cells determined by high content imaging after 60 min of incubation (RT). Hoechst 33342 was used as a nuclear stain. Unspecific binding of 5.14 (E), 5.16 (F), 5.18 (G) and 5.20 (H) were determined in the presence of famotidine (300-fold excess). All images were acquired with a INCell Analyzer 2000. Scale bar: $20 \mu \mathrm{~m}$.

The acquired images were transformed into saturation binding curves (representative curves are shown in Figure 5.12). The $K_{d}$ values determined by high content imaging at adherent HEK cells (Table 5.7) were generally in good agreement with the ones obtained by flow cytometry
(Table 5.4). The pyridinium labeled ligands 5.12 and 5.14 showed a slightly lower affinity for the $\mathrm{hH}_{2} \mathrm{R}$ ( $K_{\mathrm{d}}$ values: 74.6 nM and 90.9 nM ) compared to flow cytometry and radioligand competition binding. In contrast the cyanine labeled ligands 5.15-5.18 and $\mathbf{5 . 2 0}$ showed a similar or slightly higher affinity for the $\mathrm{hH}_{2} \mathrm{R}$. Within the series the ligands $\mathbf{5 . 1 6}$ and 5.18 showed the highest affinity with $K_{\mathrm{d}}$ values of 16.2 nM and 17.9 nM .


Figure 5.12. High content imaging: saturation binding experiments on adherent HEK293T-hH $\mathrm{H}_{2} \mathrm{R}-\mathrm{qs} 5$ cells with fluorescent ligands (A) 5.14, (B) 5.16, (C) 5.18 and (D) $\mathbf{5 . 2 0}$. Unspecific binding was determined in the presence of famotidine ( 300 -fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min. Error bars represent mean $\pm$ SEM from at least two independent experiments (each performed in duplicate).

Table 5.6. $\mathrm{hH}_{2} \mathrm{R}$ saturation binding data of fluorescent ligands 5.12, 5.14-5.18 and 5.20 determined by high content imaging on adherent HEK293-hH ${ }_{2} \mathrm{R}$-qs5 cells.

|  |  | Adherent Cells |  |  |
| :--- | :--- | :--- | :--- | :--- |
| No | Dye $^{\mathrm{a}}$ | $\mathrm{n}^{\mathrm{b}}$ | $\boldsymbol{K}_{\mathrm{d}}(\mathrm{nM})$ | $\mathbf{N}$ |
| $\mathbf{5 . 1 2}$ | Py-5 | 4 | $74.6 \pm 0.5$ | 2 |
| $\mathbf{5 . 1 4}$ | Py-5 | 7 | $90.9 \pm 8.0$ | 3 |
| $\mathbf{5 . 1 5}$ | S0223 | 4 | $105 \pm 21$ | 5 |
| $\mathbf{5 . 1 6}$ | S0223 | 6 | $16.2 \pm 4.1$ | 2 |
| $\mathbf{5 . 1 7}$ | S0436 | 4 | $179 \pm 44$ | 3 |
| $\mathbf{5 . 1 8}$ | S0436 | 6 | $17.9 \pm 5.6$ | 3 |
| $\mathbf{5 . 2 0}$ | $S 0387$ | 6 | $446 \pm 41$ | 2 |

${ }^{\text {a }}$ Dye used for the preparation of the fluorescent ligand. ${ }^{\text {b }}$ Length of the linker between the squaric acid amide and the fluorescent dye given as the number of carbon atoms. 'Saturation binding experiments on adherent cells determined by high content imaging. Unspecific binding was determined in the presence of famotidine ( 300 -fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min . Error bars represent mean $\pm$ SEM from N independent experiments (each performed in duplicate). Specific binding data were analyzed by an equation describing one-site (monophasic) binding.

Additionally, the association and dissociation kinetics of 5.18 were determined on adherent HEK293T-hH2R-qs5 cells at $37^{\circ} \mathrm{C}$ using high content imaging. The fluorescent ligand ( 50 nM ) was added to the cells and the images of the same well were acquired at different time points (no washing step necessary). Figure 5.13 (A-D) shows the binding of 5.18 after increasing perids of incubation. The dissociation of 5.18 ( 50 nM ) was determined in the presence of famotidine (300fold excess) after 60 min pre-incubation with the fluorescent ligand. Sample images at different time points are shown in Figure 5.13 ( $\mathrm{E}-\mathrm{H}$ ). The resulting association and dissociation binding curves are depicted in Figure 5.14. Association and dissociation rate of 5.18 ( $k_{\text {on }}$ value: 0.0098 $\mathrm{min}^{-1} \mathrm{nM}^{-1}$ and $k_{\text {off }}$ value: $0.091 \mathrm{~min}^{-1}, c f$. Table 5.7 ) were ten times faster compared to flow cytometry. In high content imaging 5.18 ( $c=50 \mathrm{nM}$ ) showed a fast association which was completed after approximately 10 min . The dissociation of 5.18 ( $c=50 \mathrm{nM}$ ) in the presence of famotidine was incomplete. After 25 min , the residual specific binding of the fluorescent ligand reached a plateau at approximately $78 \%$, suggesting (pseudo)irreversible binding, which was in good agreement with the results determined by flow cytometry. The images of the dissociation showed that the residual fluorescent ligand was preferentially located at the cell membrane and there was only low fluorescence in the cytoplasm. It is a matter of speculation why the association and dissociation kinetics were ten times faster compared to the kinetics determined by flow cytometry. Presumably, this was attributable to the different experimental conditions (adherent cells in monolayer vs cellsuspension, cell concentration: $3 \cdot 10^{9}$ cells $/ \mathrm{mL}$ vs $0.5-1 \cdot 10^{6}$ cells/mL in assay, fluorescent ligand concentration: 50 nM vs 25 nM ).


Figure 5.13.Time-dependent binding of fluorescent ligand $5.18(50 \mathrm{nM})$ at HEK293T-hH2R-qs5 cells measured after 0 $\min (A), 5 \mathrm{~min}(B), 10 \mathrm{~min}(C)$ and $20 \mathrm{~min}(D)$ and replacement of $5.18(50 \mathrm{nM})$ with famotidine ( 300 -fold excess) after pre-incubation for 60 min measured after $0 \mathrm{~min}(E), 5 \mathrm{~min}(F), 10 \mathrm{~min}(G)$ and $50 \mathrm{~min}(H)$. Cells were incubated with fluorescent ligand at $37{ }^{\circ} \mathrm{C}$ and images were acquired with high content imaging (INCell Analyzer 2000), scale bar: 20 $\mu \mathrm{m}$.


Figure 5.14. Association (A) and dissociation (B) kinetics of 5.18 at adherent HEK293T-hH2R-qs5 cells at $37{ }^{\circ} \mathrm{C}$ determined by high content imaging. (A) Association of 5.18 ( $c=50 \mathrm{nM}$ ) to the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time (nonlinear regression: $k_{\text {obs }}=0.40 \mathrm{~min}^{-1}$. Inset: $\ln \left[\mathrm{B}_{\mathrm{eq}} /\left(\mathrm{B}_{\mathrm{eq}}-\mathrm{B}\right)\right]$ versus time, $k_{\mathrm{obs}}=$ slope $=0,41 \mathrm{~min}^{-1}$. (B) Dissociation of 18 (preincubation: $60 \mathrm{~min}, \mathrm{c}=50 \mathrm{nM}$ ) in the presence of famotidine ( $\mathrm{c}=15 \mu \mathrm{M}$ ) from the $\mathrm{hH}_{2} \mathrm{R}$ as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{\text {off }}=0.091 \mathrm{~min}^{-1}, t_{1 / 2}=13 \mathrm{~min}$, plateau $=78 \%$ ), inset left: magnification of the first 60 min , inset right: $\ln \left[\left(B_{t}-B_{\text {plateau }}\right) /\left(B_{0}-B_{\text {plateau }}\right)\right]$ versus time, slope $\cdot(-1)=k_{\text {off }}=0.099 \mathrm{~min}^{-1}$. Data represent means $\pm$ SEM from two independent experiments (each performed in duplicate).

Nevertheless, the dissociation constant of 5.18 ( $K_{\mathrm{d}(\mathrm{kin})}$ value: 9.3 nM ), calculated from kinetics (nonlinear regression, $K_{d}=k_{\text {off }} / k_{\text {on }}$ ), was consistent with the $K_{d}$ value obtained from saturation binding experiments (Table 5.8).

Table 5.7. $\mathrm{hH}_{2} \mathrm{R}$ binding characteristics of 5.18 determined by high content imaging on adherent HEK293T-hH2 R -qs5 cells $\left(37^{\circ} \mathrm{C}\right)$

| No | $k_{\text {off }}{ }^{a}$ <br> $\left(\min ^{-1}\right)$ | $\begin{aligned} & k_{\mathrm{on}}^{\mathrm{b}} \\ & \left(\min ^{-1} \mathrm{nM}^{-1}\right) \end{aligned}$ | $\begin{aligned} & k_{\text {offf }} / k_{\text {on }}{ }^{c} \\ & K_{\mathrm{d}(\text { kin. })}(\mathrm{nM}) \end{aligned}$ | $\begin{aligned} & K_{\mathrm{d}(\mathrm{sat})}{ }^{\mathrm{d}} \\ & (\mathrm{nM}) \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: |
| 5.18 (IN Cell) | $0.091 \pm 0.084$ | $0.0098 \pm 0.0004$ | $9.3 \pm 0.4$ | $17.9 \pm 5.6$ |

${ }^{\text {a }}$ Dissociation rate constant derived from nonlinear regression. ${ }^{\text {b }}$ Association rate constant derived from nonlinear regression; calculated from $k_{\text {obs }}, k_{\text {off }}$ and the fluorescent ligand concentration (cf. Figure 5. and experimental section). ${ }^{c}$ Kinetically determined dissociation constant. ${ }^{\text {d Equilibrium dissociation constant determined by saturation binding. }}$ Data represent means $\pm$ SEM from two independent experiments (each performed in duplicate).

By high content imaging ( 96 well format) it was also possible to perform competition binding experiments with the cyanine labeled ligand $\mathbf{5 . 1 8}$ on $\mathrm{HEK} 293 \mathrm{~T}-\mathrm{hH}_{2} \mathrm{R}$-qs5 cells. The results are summarized in Table 5.9 and the competition binding curves are depicted in Figure 5.15. Ligand 5.18 was completely displaceable by histamine and the standard $\mathrm{H}_{2} \mathrm{R}$ antagonists lamtidine and ICI 127032 . The $\mathrm{p} K_{\mathrm{i}}$ values of lamtidine and ICI127032 were in good agreement with the results from radioligand competition binding. The $\mathrm{p} K_{\mathrm{i}}$ value of the endogenous ligand histamine determined by high content imaging with $\mathbf{5 . 1 8}$ was considerably lower compared to radioligand binding at sf9 insect cell membranes, but in good agreement with the results from flow cytometry, when using HEK293T-hH 2 R -qs5 cells.


Figure 5.15. Displacement of the fluorescent ligand 5.18 ( $c=50 \mathrm{nM}, K_{d}=17.9 \mathrm{nM}$ ) by histamine, lamtidine and ICl 127032 determined on adherent HEK293T-hH2R-qs5 cells by high content imagig. Cells were incubated with the fluorescent ligands and the test compounds at RT in the dark for 60 min . Data represent means $\pm$ SEM from three independent experiments (each performed in duplicate or triplicate).

Table 5.8. $\mathrm{hH}_{2} \mathrm{R}$ binding ( $\mathrm{p} K_{\mathrm{i}}$ values) of histamine, lamtidine and ICl 127032 determined by radioligand binding ([ $\left.{ }^{3} \mathrm{H}\right] \mathrm{UR}$ DE257) and by fluorescent ligand binding (5.18).

|  | Radioligand Binding ${ }^{\mathrm{a}}$ | Fluorescent Ligand $\mathbf{5 . 1 8}{ }^{\mathrm{b}}$ |
| :--- | :--- | :--- |
| No | $\mathrm{p} K_{\mathrm{i}}$ | $\mathrm{p} K_{\mathrm{i}}$ |
| Histamine | $6.53 \pm 0.04$ | $4.54 \pm 0.26$ |
| Lamtidine | $6.8 \pm 0.2$ | $7.23 \pm 0.03$ |
| ICI127032 | $7.70 \pm 0.07$ | $7.59 \pm 0.09$ |

${ }^{\text {a }}$ Determined by competition binding with $\left[{ }^{3} \mathrm{H}\right]$ UR-DE257 ( $\mathrm{c}=20 \mathrm{nM}$ ) at membrane preparations of sf9 insect cells expressing $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$. ${ }^{\mathrm{b}}$ Determined by competition binding with $5.18(\mathrm{c}=50 \mathrm{nM})$ at adherent HEK293T-hH ${ }_{2} \mathrm{R}$-qs5 cells. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data represent means $\pm$ SEM of 3 independent experiments, each performed in triplicate or duplicate.

The pyridinium labeled ligands 5.12 and 5.14 were also applied to binding studies on suspended HEK293T-hH2R-qs5 cells using an imaging flow cytometer. This cytometer acquires multiple highresolution images of every cell in the flow. The fluorescent ligands were incubated with the cells for 60 min at room temperature. The fluorescent ligands 5.12 and 5.14 were mainly localized at the cell membrane. Exemplary images of cells incubated with 5.12 ( 500 nM ) are shown in Figure 5.16A. Only focused single cells were included in the data analysis and masks of the cell membrane and the whole cell were superimposed, enabling the determination of the fluorescent intensity in the region of interest.(Figure 5.16B and 5.16C).


Figure 5.16. Typical images of focused single cells after 60 min of incubation with pyridinium labeled ligand 5.12 ( $\mathrm{c}=$ 500 nM ). Ch01: bright field, Ch05: emission at 702/85 nm, Ch01/Ch05: merged channels. Light blue coloring: (B) cell membrane mask and (C) cell mask.

The resulting saturation binding curves are shown in Figure 5.17 and the results are summarized in Table 5.10. The $K_{d}$ values of $5.12(88.3 \mathrm{nM})$ and $5.14(64.9 \mathrm{nM})$ were by approx. a factor of 3 higher than the $K_{d}$ values obtained by flow cytometry (Table 5.4). By contrast, the respective
dissociation constants were consistent with those ( $K_{d}$ values: 74.6 nM and 90.9 nM ) determined by high content imaging (Table 6), irrespective of the data processing procedure.


Figure 5.17. Imaging flow cytometric saturation binding experiments on adherent HEK293T-hH ${ }_{2} \mathrm{R}-\mathrm{qs} 5$ cells with pyridinium labeled ligands (A) 5.12 and $(B)$ 5.14. For the evaluation of the cell images the cell membrane mask was used. Unspecific binding was determined in the presence of famotidine ( 300 -fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min . Error bars represent mean $\pm$ SEM from at least two independent experiments (each performed in duplicate).

Table 5.9. $\mathrm{hH}_{2} \mathrm{R}$ saturation binding data of fluorescent ligands $\mathbf{5 . 1 2}$ and $\mathbf{5 . 1 4}$ determined by imaging flow cytometry on HEK293T-hH $\mathrm{H}_{2}$ R-qs 5 cells.

## Imaging Flow Cytometry

| No | $\boldsymbol{K}_{\mathrm{d}}(\mathrm{nM}){\text { whole cell } \text { mask }^{\mathrm{a}}} \boldsymbol{K}_{\mathrm{d}}(\mathrm{nM})$ cell membrane mask ${ }^{\mathrm{b}}$ |  |
| :--- | :--- | :--- |
| $\mathbf{5 . 1 2}$ | $87.9 \pm 2.0$ | $88.3 \pm 5.5$ |
| $\mathbf{5 . 1 4}$ | $62.1 \pm 2.0$ | $64.9 \pm 4.6$ |

[^1]Additionally, histamine and ICI127032 were investigated by equilibrium competition binding experiments with fluorescent ligand $\mathbf{5 . 1 4}$ on HEK293T-hH2R-qs5 cells using the imaging flow cytometer. The resulting $\mathrm{p} K_{\mathrm{i}}$ values were in good agreement with the results from flow cytometry and high content image analysis (Table 5.11).

Table 5.10. $\mathrm{hH}_{2} \mathrm{R}$ binding ( $\mathrm{p} K_{\mathrm{i}}$ values) of histamine and ICl127032 determined fluorimetrically on HEK293T-hH $\mathrm{H}_{2} \mathrm{R}$-qs5 cells using different systems: flow cytometry (5.14), high content imaging on adherent cells (5.18) and imaging flow cytometry (5.14).

|  | Flow Cytometry | IN Cell Analyzer | ImageStream X |
| :--- | :--- | :--- | :--- |
|  | $(5.14)^{\mathrm{a}}$ | $(5.18)^{\mathrm{b}}$ | $(5.14)^{\mathrm{c}}$ |
| No | $\mathrm{p} K_{\mathrm{i}}$ | $\mathrm{p} K_{\mathrm{i}}$ | $\mathrm{p} K_{\mathrm{i}}$ |
| Histamine | $4.30 \pm 0.04$ | $4.54 \pm 0.26$ | $4.73 \pm 0.08$ |
| ICl127032 | $7.73 \pm 0.02$ | $7.59 \pm 0.09$ | $7.81 \pm 0.05$ |

[^2]
### 5.3 EXPERIMENTAL SECTION

### 5.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). The active esters of the cyanine dyes (S0586, S0536, S2197) were obtained from FEW Chemicals (Bitterfeld-Wolfen, Germany). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 ( $0.040-0.063 \mathrm{~mm}$ ). Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution ( 0.8 g ninhydrin, 200 mL n-buthanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA). Nuclear Magnetic Resonance ( ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR) spectra were recorded on a Bruker Avance-300 ( $7.05 \mathrm{~T},{ }^{1} \mathrm{H}: 300 \mathrm{MHz},{ }^{13} \mathrm{C}: 75.5$ MHz ), Avance-400 (9.40 T, ${ }^{1} \mathrm{H}: 400 \mathrm{MHz},{ }^{13} \mathrm{C}: 100.6 \mathrm{MHz}$ ), or Avance-600 (14.1 T; ${ }^{1} \mathrm{H}: 600 \mathrm{MHz},{ }^{13} \mathrm{C}$ : 150.9 MHz; cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in $\delta(\mathrm{ppm})$ relative to external standards. Multiplicities are specified with the following abbreviations: $s$ (singlet), $d$ (doublet), $t$ (triplet), $q$ (quartet), qui (quintet), $m$ (multiplet), br s (broad signal), as well as combinations thereof. In certain cases 2D-NMR techniques (COSY, HSQC and HMBC ) were used to assign ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ chemical shifts. Low-resolution mass spectrometry (MS) was performed on a Finnigan ThermoQuest TSQ 7000 instrument using an electrospray ionization (ESI) source or on a Finnigan SSQ 710A instrument (EI-MS, 70 eV ). Highresolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass QTOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K-2001 detector. A Nucleodur 100-5 C18 (250 x21 mm, $5 \mu \mathrm{~m}$, Macherey-Nagel, Dueren, Germany) and a Kinetex XB-C18 100A ( $250 \times 21.2 \mathrm{~mm}, 5 \mu \mathrm{~m}$, Phenomenex, Aschaffenburg, Germany) served as RP-columns at a flow rate of $15 \mathrm{ml} / \mathrm{min}$ at room temperature. Mixtures of $\mathrm{CH}_{3} \mathrm{CN}$ and $0.1 \%$ aq. TFA were used as mobile phase and a detection wavelength of 220 nm was used throughout. $\mathrm{CH}_{3} \mathrm{CN}$ was removed from the eluates containing BMY 25368, $\mathbf{5 . 9}$ and $\mathbf{5 . 1 0}$ under reduced pressure (final pressure: 80 mbar ) at $45^{\circ} \mathrm{C}$ prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a vacuubrand RZ 6 rotary vane vacuum pump). In case of the fluorescently labeled ligands, the eluates were directly used for lyophilisation. Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A ( $250 \times 4.6 \mathrm{~mm}, 5 \mu \mathrm{~m}$, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of $0.05 \%$ TFA in $\mathrm{CH}_{3} \mathrm{CN}(\mathrm{A})$ and $0.05 \%$ aq. TFA (B) were use as mobile phase. Helium degassing, room temperature, a flow rate of $0.8 \mathrm{~mL} / \mathrm{min}$ and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration: 100-500 $\mu \mathrm{M}$ ) were either prepared from stock solution ( $5-10 \mathrm{mM}$ in DMSO or DMSO/20 mM HCL 50:50) in a mixture of A and B corresponding to the initial eluent composition of the run, or as a one to one dilution of the eluate with

Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5.

### 5.3.2 Experimental protocols and analytical data

The synthesis of the compound $4.3^{23,27}$ was described in chapter 4 and the tetrafluoroborate salt of Py-5 ${ }^{29,44}$ was synthesized by Mengya Chen during her master thesis.

## 3-Ethoxy-4-(3-[3-(piperidin-1-ylmethyl)phenoxy]propylamino)cyclobut-3-ene-1,2-dione (5.2) ${ }^{23}$

3,4-Diethoxycyclobut-3-ene-1,2-dione ( $226 \mathrm{mg}, 1.33 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) was dissolved in EtOH ( 10 mL ) and added to a solution of compound $4.3(300 \mathrm{mg}, 1.21 \mathrm{mmol}, 1 \mathrm{eq})$ in $\mathrm{EtOH}(10 \mathrm{~mL})$. The reaction mixture was stirred for 2 h at room temperature. After removal of the solvent under reduced pressure, the residue was dissolved in EtOAc ( 30 mL ). The organic layer was washed with $\mathrm{H}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$ and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solvent was evaporated under reduced pressure and the crude product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}$ in MeOH 100:0-95:5). Removal of the solvent in vacuo afforded the product as yellow oil ( $350 \mathrm{mg}, 78 \%$ ). $R_{\mathrm{f}}$ $=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 95: 5\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm})$ 1.38-1.42 (m, 5H), 1.52-1.58 (m, 4H), 2.06-2.12 (m, 2H), $2.36(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.42(\mathrm{~s}, 2 \mathrm{H}), 3.66(\mathrm{br} \mathrm{s}$, 1.5 H ), 3.86 (br s, 0.6H), $4.00(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 4.66-4.71(\mathrm{q}, 2 \mathrm{H}, J 7.3 \mathrm{~Hz}), 6.50(\mathrm{br}, 0.2 \mathrm{H}), 6.72-6.75(\mathrm{~m}$, $1 \mathrm{H})$, 6.86-6.89 (m, 2H), 7.16-7.20 (t, $1 \mathrm{H}, \mathrm{J} 7.7 \mathrm{~Hz}$ ), $7.27(\mathrm{br}, 0.7 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 15.8,24.3$, 25.9, 30.1, 42.4, 54.5, 63.7, 64.8, 69.7, 112.9, 115.1, 122.0, 129.1, 140.3, 158.5, 172.6, 177.5, 182.7, 189.6. HRMS: (ESI) $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{21} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{4}^{+}: 373.2122$, found: 373.2146. $\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{4}$ (372.20).

## 3-Amino-4-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclo-but-3-ene-1,2-dionehydrotrifluoracetate (BMY 25368) ${ }^{24}$

Compound 5.2 ( $180 \mathrm{mg}, 0.48 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in EtOH ( 20 mL ) and a solution of $\mathrm{NH}_{3}$ in $\mathrm{MeOH}(7 \mathrm{~N}, 10 \mathrm{~mL})$ was added. The reaction mixture was stirred over night at room temperature. The solvent was evaporated under reduced pressure and the residue was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 95: 5-90: 10$ ). Removal of the solvent in vacuo afforded the product as light yellow solid ( $123 \mathrm{mg}, 74.1 \%$ ). Mp: $199^{\circ} \mathrm{C}$ decomposition (Lit. ${ }^{24} \mathrm{mp}$ HCl-salt: 254-257 ${ }^{\circ} \mathrm{C}$ ). IR (KBr): 3295, 3135, 2930, 1805, 1645, 1570, 1530, 1260, $695 \mathrm{~cm}^{-1} .50 \mathrm{mg}$ of the product were further purified by preparative HPLC (column: Nucleodur, gradient: 0-30 $\min : M e C N / 0.1 \%$ aq. TFA 15:85-75:25, $t_{R}=8.8 \mathrm{~min}$ ). Removal of the solvent from eluate by evaporation and lyophilisation afforded the product as a hygroscopic white solid ( 53 mg ). $R_{\mathrm{f}}=0.4$ $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{~N} \mathrm{NH}_{3}\right.$ in MeOH 7:1). RP-HPLC (gradient 2, 220 nm ) (TFA-Salz): $99 \%\left(t_{\mathrm{R}}=11.38 \mathrm{~min}, k=\right.$ 2.9). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right)($ TFA-Salz): $\delta(\mathrm{ppm})$ 1.33-1.39 (m, 1H), 1.57-1.70 (m, 3H), 1.80-1.83 (m, 2H), 1.98-2.01 (m, 2H), 2.82-2.89 (m, 2H), 3.29-3.32 (m, 2H), $3.67(b r s, 2 H$, interfering with the water signal), 4.04-4.07 (t, $2 \mathrm{H}, J 6.0 \mathrm{~Hz}$ ), 4.22-4.24 (d, 2H, J 4.9 Hz), 7.02-7.09
(m, 3H), 7.35-7.39 (t, 1H, J 7.9 Hz), 7.55 (br s, 1.8H), $7.68(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 9.48(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{1} \mathrm{H}-\mathrm{NMR}(400$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right)($ TFA-Salz): $\delta(\mathrm{ppm})$ 1.46-1.56 (m, 1H), 1.70-1.83 (m, 3H), 1.91$1.95(\mathrm{~m}, 2 \mathrm{H}), 2.07-2.17$ (qui, $2 \mathrm{H}, \mathrm{J} 6.2 \mathrm{~Hz}$ ), 2.90-2.97 (m, 2H), 3.42-3.45 (m, 2H, J 12.6 Hz ), $3.84(\mathrm{br}$ $\mathrm{s}, 2 \mathrm{H}), 4.11-4.14(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 5.9 \mathrm{~Hz}), 4.23(\mathrm{~s}, 2 \mathrm{H}), 7.01-7.04(\mathrm{~m}, 3 \mathrm{H}), 7.34-7.38(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100$ $\mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}$ (TFA-Salz): $\delta(\mathrm{ppm}) 22.7,24.1,31.6,42.5,54.1,61.7,66.3,117.1$, $118.3,124.4,131.4,131.7,160.7,170.6,170.8,184.1,184.4$. HRMS (ESI) (TFA-Salz): $m / z[M+H]^{+}$, calcd. for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3}{ }^{+}: 344.1969$, found: $344.1973 . \mathrm{C}_{19} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \cdot \mathrm{C}_{2} \mathrm{HF}_{3} \mathrm{O}_{2}(343.19+114.02)$.

## General procedure for the synthesis of the Mono-Boc-protected diamines

The corresponding alkane diamine ( 2 eq or 6 eq in case of butane-1,4-diamine) was dissolved in chloroform ( $30-35 \mathrm{~mL}$ ). Di-tert-butyl dicarbonate ( 1 eq ) was dissolved in chloroform ( $25-60 \mathrm{~mL}$ ) and added dropwise over a period of 3 h under ice-cooling. The reaction mixture was allowed to warm up to room temperature while stirring overnight. Potentially precipitated alkane diamine was filtered off. The organic layer was washed with alkaline saturated NaCl solution ( 45 mL sat. aq. NaCl and 1 mL 1 M aq. NaOH ), saturated NaCl solution $(45 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(45 \mathrm{~mL})$. The organic layer was dried over sodium or magnesium sulphate and concentrated by evaporation under reduced pressure. The residue was purified by column chromatography on silica gel.

## tert-Butyl $\mathbf{N}$-(4-aminobutyl)carbamate (5.3) ${ }^{45}$ Mengya Chen master thesis

Butane-1,4-diamine ( $2.42 \mathrm{~g}, 27.4 \mathrm{mmol}, 6 \mathrm{eq}$ ) was treated with di-tert-butyl dicarbonate ( 1 g , $4.57 \mathrm{mmol}, 1 \mathrm{eq})$ according to the general procedure. The product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1 \%$ aq. $\mathrm{NH}_{3}$ in MeOH 25:1-7:1). Removal of the solvent in vacuo afforded the product as beige hygroscopic solid (790 mg, 92\%). $R_{f}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{NH}_{3}\right.$ in MeOH $10: 1)^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.42(\mathrm{~s}, 9 \mathrm{H}), 1.52(\mathrm{~m}, 4 \mathrm{H}), 2.76(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.11(\mathrm{br} \mathrm{s}, 2 \mathrm{H})$, 4.77 (br s, 0.8 H ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 27.30,28.44,29.22,40.28,41.06,79.08$, 156.14. MS (LC-MS, ESI): $m / z$ (\%) 189 (100) $[\mathrm{M}+\mathrm{H}]^{+} . \mathrm{C}_{9} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ (188.15).

## tert-Butyl $N$-(6-aminohexyl)carbamate (5.4) ${ }^{45}$

Hexane-1,6-diamine ( $1.5 \mathrm{~g}, 12.9 \mathrm{mmol}, 2 \mathrm{eq}$ ) was treated with di-tert-butyldicarbonate ( $1.4 \mathrm{~g}, 6.5$ mmol, 1 eq) according to the general procedure. The product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH}_{3}$ in MeOH 97.5:2.5-90:10). Removal of the solvent in vacuo afforded the product as colorless oil ( $620 \mathrm{mg}, 44.3 \%$ ). $R_{\mathrm{f}}=0.8\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in MeOH $5: 1) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.29-1.33(\mathrm{~m}, 4 \mathrm{H}), 1.40-1.48(\mathrm{~m}, 13 \mathrm{H}), 1.53(\mathrm{br} \mathrm{s}, 2 \mathrm{H})$, $2.67(\mathrm{t}, 2 \mathrm{H}, J 6.9 \mathrm{~Hz}), 3.07-3.11(\mathrm{t}, 2 \mathrm{H}, J 6.5 \mathrm{~Hz}), 4.55(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) . \mathrm{MS}\left(\mathrm{LC}-\mathrm{MS}, \mathrm{ESI}, t_{\mathrm{R}}=1.2 \mathrm{~min}\right): \mathrm{m} / \mathrm{z}$ (\%) 217.2 (100) $[M+H]^{+}, 161.1$ (28) $\left[M-\mathrm{C}_{4} \mathrm{H}_{8}\right]^{+}$. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{11} \mathrm{H}_{25} \mathrm{~N}_{2} \mathrm{O}_{2}{ }^{+}$: 217.1911, found: 217.1931. $\mathrm{C}_{11} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2}$ (216.33).

## tert-Butyl $N$-(7-aminoheptyl)carbamate (5.5) ${ }^{45}$

Heptan-1,7-diamine ( $2.0 \mathrm{~g}, 15.4 \mathrm{mmol}, 2 \mathrm{eq}$ ) was treated with di-tert-butyldicarbonate ( $1.7 \mathrm{~g}, 7.7$ mmol, 1 eq ) according to the general procedure. The product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH} 3$ in MeOH 97.5:2.5-90:10) Removal of the solvent in vacuo afforded the product as colorless, oily solid ( $950 \mathrm{mg}, 53.7 \%$ ). $R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH} 3\right.$ in $\mathrm{MeOH} 5: 1) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.25(\mathrm{brs}, 6 \mathrm{H}), 1.31(\mathrm{brs}, 2 \mathrm{H}), 1.37-1.42(\mathrm{~m}, 13 \mathrm{H})$, $2.61(\mathrm{t}, 2 \mathrm{H}, J 6.9 \mathrm{~Hz}), 3.00-3.05(\mathrm{~m}, 2 \mathrm{H}), 4.67(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) . \mathrm{MS}\left(L C-M S, E S I, t_{\mathrm{R}}=1.4 \mathrm{~min}\right): \mathrm{m} / \mathrm{z}(\%)$ 231.2 (100), $[M+]^{+}, 175.1$ (23) $\left[M-\mathrm{C}_{4} \mathrm{H}_{8}\right]^{+}$. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{12} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{2}^{+}$: 231.1067, found: 231.2066. $\mathrm{C}_{12} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{2}$ (230.35).

## General procedure for the synthesis of Boc-protected squaramides with alkylamine spacers

Compound 5.2 ( 1 eq ) was dissolved in EtOH ( $15-25 \mathrm{~mL}$ ) and added to the respective mono-Bocprotected diamine ( 1.1 eq ) in EtOH ( $15-25 \mathrm{~mL}$ ). The reaction mixture was stirred over night at room temperature. The solvent was evaporated under reduced pressure and the product was crystallized from EtOAc ( 2 mL ), $\mathrm{NH}_{3}$ in $\mathrm{MeOH}(7 \mathrm{M}, 50 \mu \mathrm{~L}$ ) and n-hexane ( 3 mL ). The precipitate was purified by column chromatography on silica gel.

## tert-Butyl [4-([3,4-dioxo-2-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclobut-1-en-1yl]amino)butyl]carbamate (5.6)

Compound 5.6 was synthesized from $5.2(2.8 \mathrm{~g}, 7.63 \mathrm{mmol}, 1 \mathrm{eq})$ and $\mathbf{3}(1.58 \mathrm{~g}, 8.39 \mathrm{mmol}, 1.1$ eq) according to the general procedure. The product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1 \%$ aq. $\mathrm{NH}_{3}$ in $\mathrm{MeOH} 40: 1-20: 1$ ). Removal of the solvent in vacuo afforded the product as light yellow hygroscopic solid ( $1.66 \mathrm{~g}, 42 \%$ ). $R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 2 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 5: 1\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.41(\mathrm{~s}, 9 \mathrm{H}), 1.49-1.73(\mathrm{~m}, 6 \mathrm{H}), 1.88(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 2.00-2.19(\mathrm{~m}$, $2 \mathrm{H}), 2.95(\mathrm{br} s, 4 \mathrm{H}), 3.09-3.15(\mathrm{~m}, 2 \mathrm{H}), 3.65(\mathrm{q}, J 6.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.80(\mathrm{q}, J 6.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.94(\mathrm{~s}, 2 \mathrm{H})$, $4.15(\mathrm{t}, \mathrm{J} 6.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.87(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.85-6.91(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.99(\mathrm{br} \mathrm{s}, 1.8 \mathrm{H}) .{ }^{13} \mathrm{C}-$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 22.6,23.5,26.9,28.5,30.4,40.1,41.1,44.1,53.4,53.8,62.2$, 65.4, 79.1, 116.2, 116.7, 122.9, 129.9, 141.8, 156.2, 159.3, 168.0, 168.6, 182.4, 182.5. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{28} \mathrm{H}_{43} \mathrm{~N}_{4} \mathrm{O}_{5}^{+}$: 515.3228, found: 515.3230. $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{~N}_{4} \mathrm{O}_{5}$ (514.32).

## tert-Butyl (6-[(3,4-dioxo-2-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclobut-1-en-1yl )amino]hexyl)carbamate (5.7) ${ }^{23}$

Compound 5.7 was synthesized from 5.2 ( $160 \mathrm{mg}, 0.43 \mathrm{mmol}, 1 \mathrm{eq}$ ) and 5.4 ( $102 \mathrm{mg}, 0.47 \mathrm{mmol}$, $1.1 \mathrm{eq})$ according to the general procedure. The product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 97.5: 2.5-95: 5$ ). Removal of the solvent in vacuo afforded the product as light yellow solid (193 mg, 83\%). Mp: 115.9-118.7 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.8\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 90: 10) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.30-1.47(\mathrm{~m}, 17 \mathrm{H}), 1.59-1.62(\mathrm{~m}, 6 \mathrm{H}), 2.08-$ 2.14 (qui, $2 \mathrm{H}, J 6.3 \mathrm{~Hz}$ ), $2.46(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.03-3.08(\mathrm{~m}, 2 \mathrm{H}), 3.48(\mathrm{~s}, 2 \mathrm{H}), 3.60-3.65(\mathrm{~m}, 2 \mathrm{H}), 3.79-$
$3.84(\mathrm{~m}, 2 \mathrm{H}),, 4.02-4.05(\mathrm{t}, 2 \mathrm{H}, J 6.0 \mathrm{~Hz}), 5.13(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.76-6.78(\mathrm{dd}, 1 \mathrm{H}, J 8.1 \mathrm{~Hz}, J 1.9 \mathrm{~Hz}), 6.84-$ $6.89(\mathrm{~m}, 2 \mathrm{H}), 7.16-7.20(\mathrm{t}, 1 \mathrm{H}, J 7.8 \mathrm{~Hz}), 7.36(\mathrm{br} \mathrm{s}, 1.5 \mathrm{H})$. HRMS (ESI): $m / z[M+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{30} \mathrm{H}_{47} \mathrm{~N}_{4} \mathrm{O}_{5}^{+}: 543.3541$, found: 543.3562. $\mathrm{C}_{30} \mathrm{H}_{46} \mathrm{~N}_{4} \mathrm{O}_{5}$ (542.72).
tert-Butyl (7-[(3,4-dioxo-2-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclobut-1-en-1$\mathrm{yl})$ amino]heptyl)carbamate (5.8)

Compound 5.8 was synthesized from $5.2(260 \mathrm{mg}, 0.70 \mathrm{mmol}, 1 \mathrm{eq})$ und $5.5(177 \mathrm{mg}, 0.77 \mathrm{mmol}$, $1.1 \mathrm{eq})$ according to the general procedure. The product was purified by column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 3.5 \mathrm{M} \mathrm{NH}_{3}$ in $\mathrm{MeOH} 97.5: 2.5-95: 5$ ). Removal of the solvent in vacuo afforded the product as light yellow solid (204 mg, 48\%). Mp: 109.8-112.9 ${ }^{\circ} \mathrm{C}$. $R_{\mathrm{f}}=0.9\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 7: 1) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm})$ 1.25-1.43 (m, 19H), 1.52-1.59 (m, 6H), 2.08-2.14 (qui, $2 \mathrm{H}, \mathrm{J} 6.3 \mathrm{~Hz}$ ), 2.37 (br s, 4H), 3.01-3.06 (m 2H), 3.40 (s, 2H), 3.60-3.64 (m, 2H), 3.82-3.84 (m, $2 \mathrm{H}), 4.01-4.04(\mathrm{t}, \mathrm{J} 6.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.15(\mathrm{br} \mathrm{s}, 0.7 \mathrm{H}), 6.73-6.75(\mathrm{~m}, 1 \mathrm{H}), 6.85-6.87(\mathrm{~m}, 2 \mathrm{H}), 7.16(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}$ 7.8 Hz ), 7.42 (br s, 0.7 H ), $7.50(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 24.4,26.0,26.5$, $26.6,28.6,28.8,30.0,30.9,31.0,40.5,41.7,44.8,54.7,64.0,64.9,79.3,113.4,115.3,122.1$, 129.2, 140.1, 156.6, 158.8, 168.4, 182.6, 182.7. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{31} \mathrm{H}_{49} \mathrm{~N}_{4} \mathrm{O}_{5}{ }^{+}$: 557.3697, found: 557.3713. $\mathrm{C}_{31} \mathrm{H}_{48} \mathrm{~N}_{4} \mathrm{O}_{5}$ (556.75).

## 3-((4-Aminobutyl)amino)-4-((3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)amino)cyclobut-3-ene-1,2-dione bis(hydrotrifluoracetate) (5.9) ${ }^{23}$

Compound 5.6 ( $100 \mathrm{mg}, 0.11 \mathrm{mmol}, 1 \mathrm{eq}$ ) was stirred for 24 h at room temperature in a mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25-30 \mathrm{~mL})$ and TFA ( 1 mL ). Removal of the solvent in vacuo afforded the product as a yellow hygroscopic solid ( $118 \mathrm{mg}, 94 \%$ ). For pharmacological characterization 57 mg of 9 were further purified by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 5:95-60:40, $t_{\mathrm{R}}=12.8 \mathrm{~min}$ ). The TFA-salt was obtained as white sticky solid ( $40 \mathrm{mg} ; 65 \%$ ). $R_{\mathrm{f}}=0.1$ $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.8 \mathrm{M} \mathrm{NH} 3\right.$ in $\left.\mathrm{MeOH} 90: 10\right)$. RP-HPLC (gradient 2, 220 nm ): $98 \%\left(t_{\mathrm{R}}=10.66 \mathrm{~min}, k=2.7\right)$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}\right): \delta(\mathrm{ppm}) 1.47-1.54(1 \mathrm{H}, \mathrm{m}), 1.70-1.83(7 \mathrm{H}, \mathrm{m}), 1,92-1.94(2 \mathrm{H}, \mathrm{m})$, $2.10(2 \mathrm{H}$, qui, J 6.3 Hz$), 2.91-2.97(4 \mathrm{H}, \mathrm{m}), 3.42-3.44(2 \mathrm{H}, \mathrm{m}), 3.62(2 \mathrm{H}, \mathrm{br} \mathrm{s}), 3.83(2 \mathrm{H}, \mathrm{br} \mathrm{s}), 4.12$ $(2 \mathrm{H}, \mathrm{t}, J 6.0 \mathrm{~Hz}), 4.22(2 \mathrm{H}, \mathrm{s}), 7.02-7.04(2 \mathrm{H}, \mathrm{m}), 7.08(1 \mathrm{H}, \mathrm{br} \mathrm{s}), 7.36(1 \mathrm{H}, \mathrm{t}, J 7.9 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150$ $\left.\mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}\right)($ TFA-Salz): $\delta(\mathrm{ppm}) 22.7,24.1,25.3,29.1,31.7,40.2,42.4,44.3,54.1,61.7,66.2$, 118.4, 124.5, 131.4, 131.7, 160.8, 162.5 (q, J 40 Hz, TFA), 169.5, 169.7, 183.6 (2C). HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{23} \mathrm{H}_{35} \mathrm{~N}_{4} \mathrm{O}_{3}{ }^{+}$: 415.2709, found: 415.2709. $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{~N}_{4} \mathrm{O}_{3} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(414.26+$ 228.05).

## General procedure for the Boc-deprotection of the amine precursors

The corresponding Boc-protected squaramide derivative ( 5.7 or 5.8 ) was stirred for $4-24 \mathrm{~h}$ at room temperature in a mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25-30 \mathrm{~mL})$ and HCl in 2-propanol ( $5-6 \mathrm{M}, 13-15 \mathrm{~mL}$ ). Removal of the solvent in vacuo afforded the product as HCl salt. Part of the product was further purified by preparative HPLC.

## 3-((6-Aminohexyl)amino)-4-((3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)amino)cyclobut-3-ene-1,2-dione bis(hydrotrifluoracetate) (5.10) ${ }^{23}$

Compound 5.10 was synthesized from 5.7 ( $185 \mathrm{mg}, 0.34 \mathrm{mmol}, 1 \mathrm{eq}$ ) according to the general procedure. Removal of the solvent in vacuo afforded the product as yellow hygroscopic solid (188 mg ). For pharmacological characterization 40 mg of 5.10 were purified by preparative HPLC (column: Nucleodur, gradient: $0-30 \mathrm{~min}$ : $\mathrm{MeCN} / 0.1 \%$ aq. TFA $15: 85-75: 25, t_{\mathrm{R}}=8.4 \mathrm{~min}$ ). The TFAsalt was obtained as highly hygroscopic sticky solid ( 45 mg ; 93\%). $R_{\mathrm{f}}$ (TFA-Salz) $=0.1\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M}\right.$ $\mathrm{NH}_{3}$ in MeOH 7:1). RP-HPLC (gradient 2, 220 nm ) (TFA-Salz): $98 \%\left(t_{\mathrm{R}}=12.17 \mathrm{~min}, k=3.2\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}$ ) (TFA-Salz): $\delta(\mathrm{ppm})$ 1.42-1.44 (m, 4H), 1.48-1.57 (m, 1H), 1.62-1.68 (m, 4H), 1.70-1.84 (m, 3H), 1.91-1.96 (m, 2H), 2.07-2.14 (qui, $2 \mathrm{H}, \mathrm{J} 6.3 \mathrm{~Hz}$ ), 2.90-2.98 (m, 4H), 3.43-3.46 (m, 2 H ), $3.60(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.83(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 4.13(\mathrm{t}, 2 \mathrm{H}, J 5.9 \mathrm{~Hz}), 4.23(\mathrm{~s}, 2 \mathrm{H}), 7.02-7.07(\mathrm{~m}, 3 \mathrm{H}), 7.37(\mathrm{t}$, $1 \mathrm{H}, J 7.9 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}\right)($ TFA-Salz): $\delta(\mathrm{ppm}) 22.7,24.1,26.77,26.85,28.4$, $31.6,32.0,40.6,42.4,45.0,54.1,61.7,66.3,117.2,118.3,124.5,131.4,131.7,160.8,162.2$ (q, J 41 Hz, TFA), 169.7 (2C), 183.6, 183.9. HRMS (TFA-Salz): (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{25} \mathrm{H}_{39} \mathrm{~N}_{4} \mathrm{O}_{3}{ }^{+}$: 443.3017, found: 443.3019. $\mathrm{C}_{25} \mathrm{H}_{38} \mathrm{~N}_{4} \mathrm{O}_{3} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(442.60+228.05)$.

## 3-((7-Aminoheptyl)amino)-4-((3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)amino)cyclobut-3-ene-1,2-dione bis(hydrotrifluoracetate) (5.11)

Compound 5.11 was synthesized from 5.8 ( $187 \mathrm{mg}, 0.33 \mathrm{mmol}, 1 \mathrm{eq}$ ) according to the general procedure. Removal of the solvent in vacuo afforded the product as yellow hygroscopic solid (197 mg ). For pharmacological characterization 85 mg of 5.11 were further purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85-75:25, $t_{\mathrm{R}}=9.1 \mathrm{~min}$ ). The TFA-salt was obtained as white hygroscopic solid ( $84 \mathrm{mg}, 76 \%$ ). $R_{\mathrm{f}}$ (TFA-Salz) $=0.5\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 7 \mathrm{M} \mathrm{NH} 3\right.$ in MeOH 5:1). RP-HPLC (gradient 2, 220 nm ) (TFA-Salz): $98 \% ~\left(t_{\mathrm{R}}=12.83 \mathrm{~min}, k=3.4\right.$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ) (TFA-Salz): $\delta(\mathrm{ppm}) 1.28-1.39(\mathrm{~m}, 7 \mathrm{H}), 1.51(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 1.61-1.68(\mathrm{~m}, 3 \mathrm{H})$, 1.79-1.82 (m, 2H), 1.97-2.03 (qui, $2 \mathrm{H}, J 6.3 \mathrm{~Hz}$ ), 2.72-2.89 (m, 4H), 3.29-3.32 (m, 2H), $3.48(\mathrm{br} \mathrm{s}$, $2 \mathrm{H}), 3.67-3.68(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 4.05(\mathrm{t}, 2 \mathrm{H}, J 6.2 \mathrm{~Hz}), 4.23(\mathrm{~d}, 2 \mathrm{H}, J 5.0 \mathrm{~Hz}), 7.00-7.05(\mathrm{~m}, 2 \mathrm{H}), 7.09(\mathrm{br} \mathrm{s}$, $1 \mathrm{H}), 7.36(\mathrm{t}, 1 \mathrm{H}, J 7.9 \mathrm{~Hz}), 7.76-7.82(\mathrm{br} \mathrm{s}, 5 \mathrm{H}), 9.68(\mathrm{br} \mathrm{s}, 0.9 \mathrm{H}) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}\right.$, COSY, HSQC, HMBC) (TFA-Salz): $\delta(\mathrm{ppm}) 1.39(\mathrm{br} \mathrm{s}, 6 \mathrm{H}), 1.48-1.67(\mathrm{~m}, 5 \mathrm{H}), 1.71-1.82(\mathrm{~m}, 3 \mathrm{H})$, 1.90-1.94 (m, 2H), 2.07-2.13 (qui, $2 \mathrm{H}, \mathrm{J} 6.2 \mathrm{~Hz}$ ), 2.89-2.98 (m, 4 H ), 3.42-3.45 (m, 2 H ), $3.59(\mathrm{br} \mathrm{s}$, $2 \mathrm{H}), 3.83(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 4.12(\mathrm{t}, 2 \mathrm{H}, J 5.9 \mathrm{~Hz}), 4.23(\mathrm{~s}, 2 \mathrm{H}), 7.01-7.07(\mathrm{~m}, 3 \mathrm{H}), 7.35(\mathrm{t}, 1 \mathrm{H}, J 7.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-$ NMR (100 MHz, [ $\left.\mathrm{D}_{4}\right] \mathrm{MeOH}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}$ ) (TFA-Salz): $\delta(\mathrm{ppm}) 22.7,24.0,27.0,27.2,28.4$, 29.6, 31.6, 32.0, 40.6, 42.4, 45.1, 54.0, 61.7, 66.2, 117.3, 117.8 (q, J 291 Hz, TFA), 118.2, 124.5, 131.3, 131.7, 160.7, 162.2 (q, J $37 \mathrm{~Hz}, \mathrm{TFA}$ ), 169.5 (2C), 183.3, 183.5. HRMS (TFA-Salz): (ESI): $m / z$ $[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{26} \mathrm{H}_{41} \mathrm{~N}_{4} \mathrm{O}_{3}{ }^{+}: 457.3173$, found.: 457.3178. $\mathrm{C}_{26} \mathrm{H}_{40} \mathrm{~N}_{4} \mathrm{O}_{3} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(456.63+$ 228.05).

## General procedure for the synthesis of pyridinium-labeled fluorescent ligands

The respective amine-precursor 5.9, 5.10 or 5.11 ( 1 eq ) was dissolved in anhydrous DMF (200$400 \mu \mathrm{~L}$ ) and DIPEA (10 eq) or TEA (2-3 drops). The pyrylium dye Py-5 (3-8 eq) in anhydrous DMF
(180-200 $\mu \mathrm{l}$ ) was added portion wise ( $50 \mu \mathrm{~L}$ ) every 20-30 min. Subsequent to the last addition, the reaction mixture was incubated for additional 30 min at RT in the dark. The reaction was stopped by addition of TFA ( $10 \%$ in $\mathrm{H}_{2} \mathrm{O}_{\text {millio, }} 50-200 \mu \mathrm{~L}$ ). After dilution with a mixture of $\mathrm{H}_{2} \mathrm{O}_{\text {millia }}$ with $5 \% \mathrm{MeCN}$ and $0.1 \%$ TFA ( 2 mL ) the respective fluorescently labeled ligand was isolated as TFA-salt by preparative HPLC. After lyophilisation of the eluate the product was obtained as a red highly hygroscopic solid.

## Compound 5.9 labeled with Py-5 (5.12) ${ }^{25}$ Mengya Chen master thesis

5.9 ( $1.95 \mathrm{mg}, 4 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ), Py-5 ( $4.41 \mathrm{mg}, 12 \mu \mathrm{~mol}, 3 \mathrm{eq}$ ) and TEA (2-3 drops) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: $0-30 \mathrm{~min}: \mathrm{MeCN} / 0.1 \% \mathrm{aq}$. TFA 5:95-90:10, $t_{\mathrm{R}}=15.5 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.12 as a highly hygroscopic, red solid ( $1.17 \mathrm{mg}, 32 \%$ ). RP-HPLC (gradient 2, 220 nm ): 96.9\% ( $t_{\mathrm{R}}=16.4 \mathrm{~min}, k=4.7$ ). HRMS (ESI): $\mathrm{m} / \mathrm{z}[M]^{+}$calcd. for $\mathrm{C}_{42} \mathrm{H}_{54} \mathrm{~N}_{5} \mathrm{O}_{3}{ }^{+}: 676.4221$, found: 676.4227. $\mathrm{C}_{42} \mathrm{H}_{54} \mathrm{~N}_{5} \mathrm{O}_{3}{ }^{+} \cdot \mathrm{C}_{4} \mathrm{HF}_{6} \mathrm{O}_{4}(676.42+227.04)$.

## Compound 5.10 labeled with Py-5 (5.13) Mengya Chen master thesis

5.10 ( $2.21 \mathrm{mg}, 5.00 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ), Py-5 ( $11.02 \mathrm{mg}, 30.00 \mu \mathrm{~mol}, 6 \mathrm{eq}$ ) and TEA ( $2-3 \mathrm{drops}$ ) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-90:10, $t_{R}=16.4 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.13 as a highly hygroscopic, red solid ( $1.46 \mathrm{mg}, 31 \%$ ). RP-HPLC (gradient $2,220 \mathrm{~nm}): 95.1 \%\left(t_{\mathrm{R}}=18.1 \mathrm{~min}, k=5.2\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}\right): \delta(\mathrm{ppm}) 1.48-1.59(\mathrm{~m}$, 5 H ), 1.65-1.70 (m, 2H), 1.72-1.80 (m, 2H), 1.82-1.88 (m, 3H), 1.93-1.95 (m, 2H), 2.08-2.13 (qui, $2 \mathrm{H}, J 6.3 \mathrm{~Hz}), 2.80(\mathrm{~s}, 6 \mathrm{H}), 2.92-2.97(\mathrm{~m}, 2 \mathrm{H}), 3.03(\mathrm{~s}, 6 \mathrm{H}), 3.42-3.45(\mathrm{~m}, 2 \mathrm{H}), 3.62(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.83$ (br s, 2H), $4.13(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 6.0 \mathrm{~Hz}), 4.23(\mathrm{~s}, 2 \mathrm{H}), 4.36-4.38(\mathrm{~m}, 2 \mathrm{H}), 6.58(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J} 15.3 \mathrm{~Hz}), 6.77(\mathrm{~d}, 2 \mathrm{H}$, $J 8.8 \mathrm{~Hz})$, 6.93-6.97 (m, 1H), 7.00-7.07 (m, 4H), $7.37(\mathrm{t}, 1 \mathrm{H}, J 8.2 \mathrm{~Hz}), 7.44(\mathrm{~d}, 2 \mathrm{H}, J 8.9 \mathrm{~Hz}), 7.63-$ $7.68(\mathrm{~m}, 1 \mathrm{H}), 7.71(\mathrm{~s}, 2 \mathrm{H})$. HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}]^{+}$calcd. for $\mathrm{C}_{44} \mathrm{H}_{58} \mathrm{~N}_{5} \mathrm{O}_{3}{ }^{+}: 704.4534$, found: 704.4542. $\mathrm{C}_{44} \mathrm{H}_{58} \mathrm{~N}_{5} \mathrm{O}_{3}{ }^{+} \cdot \mathrm{C}_{4} \mathrm{HF}_{6} \mathrm{O}_{4}(704.45+227.04)$.

## Compound 5.11 labeled with Py-5 (5.14)

5.11 ( $5 \mathrm{mg}, 7.30 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ), Py-5 ( $21.4 \mathrm{mg}, 58.42 \mu \mathrm{~mol}, 8 \mathrm{eq}$ ) and DIPEA ( $12.8 \mu \mathrm{~L}, 73.03 \mu \mathrm{~mol}, 10$ eq) were applied according to general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 5:95-45:55, $t_{\mathrm{R}}=28.9 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.14 as a highly hygroscopic, red solid ( $1.59 \mathrm{mg}, 24 \%$ ). RP-HPLC (gradient 2, 220 nm ): $96 \%\left(t_{\mathrm{R}}=18.4 \mathrm{~min}, k=5.4\right.$ ). $\mathrm{HRMS}(E S I): m / z[M]^{+}$calcd. for $\mathrm{C}_{45} \mathrm{H}_{60} \mathrm{~N}_{5} \mathrm{O}_{3}{ }^{+}$: 718.4691, found: 718.4698. $\mathrm{C}_{45} \mathrm{H}_{60} \mathrm{~N}_{5} \mathrm{O}_{3}{ }^{+} \cdot \mathrm{C}_{4} \mathrm{HF}_{6} \mathrm{O}_{4}(718.47+227.04)$.

## General procedure for the synthesis of cyanine-labeled fluorescent ligands

A solution of the activated fluorescent dye ( 1 eq ) in anhydrous DMF (30-100 $\mu \mathrm{l}$ ) was added to the respective amine-precursor 5.9 or $5.10(2.0-3.3 \mathrm{eq})$ dissolved in anhydrous DMF ( $100 \mu \mathrm{~L}$ ) and DIPEA (8-10 eq). The reaction mixture was incubated for $45-90 \mathrm{~min}$ at RT in the dark. The reaction was stopped by addition of TFA ( $10 \%$ in $\mathrm{H}_{2} \mathrm{O}_{\text {millio, }} 45-60 \mu \mathrm{~L}$ ). After dilution with a mixture of $\mathrm{H}_{2} \mathrm{O}_{\text {milliQ }}$ with $5 \% \mathrm{MeCN}$ and $0.1 \%$ TFA (150-250 $\mu \mathrm{L}$ ) the respective fluorescently labeled ligand was isolated as TFA-salt by preparative HPLC. After lyophilisation of the eluate the product was obtained as a blue fluffy solid.

## Compound 5.9 labeled with SO223 (5.15)

5.9 ( $8.88 \mathrm{mg}, 13.12 \mu \mathrm{~mol}, 2.5 \mathrm{eq}$ ), S2197 ( $3.65 \mathrm{mg}, 4.84 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ) and DIPEA ( $7.7 \mu \mathrm{~L}, 44.20$ $\mu \mathrm{mol}, 8 \mathrm{eq})$ were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: $0-30 \mathrm{~min}: M e C N / 0.1 \%$ aq. TFA 15:85-70:30, $t_{\mathrm{R}}=24.1 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.15 as a hygroscopic, blue fluffy solid ( 1.13 mg , 18\%). RP-HPLC ( gradient 1, 220 nm ): $96 \%\left(t_{\mathrm{R}}=25.3 \mathrm{~min}, k=7.7\right)$. HRMS (ESI): $m / z[M]^{+}$calcd. for $\mathrm{C}_{55} \mathrm{H}_{71} \mathrm{~N}_{6} \mathrm{O}_{4}^{+}$: 879.5531, found: 879.5533. $\mathrm{C}_{55} \mathrm{H}_{71} \mathrm{~N}_{6} \mathrm{O}_{4}^{+} \cdot \mathrm{C}_{4} \mathrm{HF}_{6} \mathrm{O}_{4}{ }^{-}$(880.21+227.04).

## Compound 5.10 labeled with SO223 (5.16)

5.10 ( $11.0 \mathrm{mg}, 17.03 \mu \mathrm{~mol}, 3 \mathrm{eq}$ ), S2197 ( $3.75 \mathrm{mg}, 5.68 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ) and DIPEA ( $9.9 \mu \mathrm{~L}, 56.76$ $\mu \mathrm{mol}, 10 \mathrm{eq}$ ) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: $0-30 \mathrm{~min}: \mathrm{MeCN} / 0.1 \% \mathrm{aq}$. TFA 15:85-70:30, $t_{\mathrm{R}}=24.9 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.16 as a hygroscopic, blue fluffy solid ( 2.49 mg , $39 \%$ ). RP-HPLC (gradient 2, 220 nm ): $99 \%\left(t_{\mathrm{R}}=25.6 \mathrm{~min}, k=7.8\right.$ ). HRMS (ESI): $m / z[M]^{+}$calcd. for $\mathrm{C}_{57} \mathrm{H}_{75} \mathrm{~N}_{6} \mathrm{O}_{4}^{+}$: 907.5844, found: 907.5839. $\mathrm{C}_{57} \mathrm{H}_{75} \mathrm{~N}_{6} \mathrm{O}_{4}^{+} \cdot \mathrm{C}_{4} \mathrm{HF}_{6} \mathrm{O}_{4}^{-}$(908.26+227.04).

## Compound 5.9 labeled with SO436(5.17) ${ }^{25}$

5.9 ( $7.78 \mathrm{mg}, 12.11 \mu \mathrm{~mol}, 2.5 \mathrm{eq}$ ), S 0536 ( $3.40 \mathrm{mg}, 4.84 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ) and DIPEA ( $6.8 \mu \mathrm{~L}, 38.75$ $\mu \mathrm{mol}, 8 \mathrm{eq})$ were applied according to general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-65:35, $t_{\mathrm{R}}=22.6 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.17 as a hygroscopic, blue fluffy solid ( 1.49 mg , $25 \%$ ). RP-HPLC (gradient 1, 220 nm ): $97 \%\left(t_{\mathrm{R}}=23.8 \mathrm{~min}, k=7.2\right.$ ). HRMS (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{58} \mathrm{H}_{77} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{~S}^{+}: 1001.5569$, found: 1001.5573. $\mathrm{C}_{58} \mathrm{H}_{76} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{~S} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(1001.34+228.05)$.

## Compound 5.10 labeled with SO436 (5.18) ${ }^{25}$

5.10 ( $8.0 \mathrm{mg}, 11.93 \mu \mathrm{~mol}, 3.3 \mathrm{eq}$ ), S0536 ( $2.55 \mathrm{mg}, 3.63 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ) and DIPEA ( $5.1 \mu \mathrm{~L}, 29.01$ $\mu \mathrm{mol}, 8 \mathrm{eq})$ were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 5:95-65:35, $t_{\mathrm{R}}=27.0 \mathrm{~min}$ ) and removal
of the solvent by lyophilisation afforded 5.18 as a hygroscopic, blue fluffy solid ( $1.53 \mathrm{mg}, 34 \%$ ). RP-HPLC (gradient 2, 220 nm ): 97\% ( $t_{\mathrm{R}}=23.1 \mathrm{~min}, k=7.0$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{4}\right] \mathrm{MeOH}\right): \delta$ (ppm) 1.33-1.36 (5H, m), 1.43-1.52 (5H, m), 1.58-1.61 ( $2 \mathrm{H}, \mathrm{m}$ ), 1.67-1.84 (18H, m), 1.90-1.94 ( 4 H , m), 1.96-2.02 (2H, m), 2.04-2.08 (2H, m), 2.20(2H, t, J 7.1 Hz), 2.87-2.95 (4H, m), 3.14 (2H, t, J 6.8 $\mathrm{Hz}), 3.40-3.42(2 \mathrm{H}, \mathrm{m}), 3.56(2 \mathrm{H}, \mathrm{br}$ s), $3.78(2 \mathrm{H}, \mathrm{br} \mathrm{s}), 4.08-4.14(6 \mathrm{H}, \mathrm{m}), 4.20(2 \mathrm{H}, \mathrm{s}), 6.26(1 \mathrm{H}, \mathrm{d}, J$ $13.7 \mathrm{~Hz}), 6.32(1 \mathrm{H}, \mathrm{d}, J 13.7 \mathrm{~Hz}), 6.60(1 \mathrm{H}, J 12.7 \mathrm{~Hz}), 6.98-7.01(2 \mathrm{H}, \mathrm{m}), 7.08(1 \mathrm{H}, \mathrm{s}), 7.23-7.34$ $(5 \mathrm{H}, \mathrm{m}), 7.40(2 \mathrm{H}, \mathrm{t}, J 7.7 \mathrm{~Hz}), 7.47(2 \mathrm{H}, \mathrm{t}, J 7.0 \mathrm{~Hz}), 8.22(2 \mathrm{H}, \mathrm{t}, \mathrm{J} 13.1 \mathrm{~Hz}) . \mathrm{HRMS}(\mathrm{ESI}): m / z[\mathrm{M}+\mathrm{H}]^{+}$ calcd. for $\mathrm{C}_{60} \mathrm{H}_{81} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{~S}^{+}: 1029.5882$, found: 1029.5883. $\mathrm{C}_{60} \mathrm{H}_{80} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{~S} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(1029.40+228.05)$.

## Compound 5.9 labeled with S0387 (5.19).

5.9 ( $5.1 \mathrm{mg}, 7.93 \mu \mathrm{~mol}, 2.5 \mathrm{eq}$ ), S0586 ( $2.6 \mathrm{mg}, 3.17 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ) and DIPEA ( $4.4 \mu \mathrm{~L}, 25.38 \mu \mathrm{~mol}, 8$ eq) were applied according to general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 5:95-55:45, $t_{\mathrm{R}}=22.6 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.19 as a hygroscopic, blue fluffy solid ( $1.82 \mathrm{mg}, 44 \%$ ). RP-HPLC (gradient 1, 220 nm ): 98\% ( $t_{\mathrm{R}}=18.6 \mathrm{~min}, k=5.4$ ). HRMS (ESI): $m / z[M+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{58} \mathrm{H}_{77} \mathrm{~N}_{6} \mathrm{O}_{10} \mathrm{~S}_{2}^{+}: 1081.5137$, found: 1081.5140. $\mathrm{C}_{58} \mathrm{H}_{76} \mathrm{~N}_{6} \mathrm{O}_{10} \mathrm{~S}_{2} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(1081.40+228.05)$.

## Compound 5.10 labeled with SO387 (5.20)

5.10 ( $4.76 \mathrm{mg}, 7.09 \mu \mathrm{~mol}, 2 \mathrm{eq}$ ), S0586 ( $2.85 \mathrm{mg}, 3.54 \mu \mathrm{~mol}, 1 \mathrm{eq}$ ) and DIPEA ( $4.8 \mu \mathrm{~L}, 27.61 \mu \mathrm{~mol}$, 8 eq) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 5:95-55:45, $t_{\mathrm{R}}=23.5 \mathrm{~min}$ ) and removal of the solvent by lyophilisation afforded 5.20 as a hygroscopic, blue fluffy solid ( $0.91 \mathrm{mg}, 19 \%$ ). RP-HPLC (gradient 2, 220 nm ): 99\% ( $t_{\mathrm{R}}=17.6 \mathrm{~min}, k=5.1$ ). HRMS (ESI): $m / z[M+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{60} \mathrm{H}_{81} \mathrm{~N}_{6} \mathrm{O}_{10} \mathrm{~S}_{2}^{+}$: 1109.5450 , found: 1109.5447. $\mathrm{C}_{60} \mathrm{H}_{80} \mathrm{~N}_{6} \mathrm{O}_{10} \mathrm{~S}_{2} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}{ }^{-}(1109.45+228.05)$.

## Fluorescence spectroscopy and determination of quantum yields ${ }^{30}$

The recording of fluorescence spectra and the determination of quantum yields were performed with a Cary Eclipse spectrofluorometer (Varian Inc., Mulgrave, Victoria, Australia). The photomultiplier voltage of the spectrofluorimeter was set to 400 V throughout. Depicted excitation and emission spectra were recorded with an excitation slit of 10 nm and an emission slit of 10 nm . Appropriate concentrations of the fluorescent ligands, with absorbances between 0.1 and 0.2 at the respective excitation wavelength, were determined with a Cary 100 UV/VIS (Varian Inc., Mulgrave, Victoria, Australia) photometer. Absorption spectra were recorded within a concentration range of $1.5-6 \mu \mathrm{M}$. The excitation wavelength was chosen as close to the absorption maximum as possible $(5.13,5.14)$ or at an inflection point (5.20, 5.18, 5.16 and cresyl violet perchlorate). For the determination of quantum yields, cresyl violet perchlorate (Acros Organics, Geel, Belgium), with a reported quantum yield of $54 \%$ in $\mathrm{EtOH}^{46}$, was used as a standard. All spectra were recorded in acryl cuvettes ( $10 \times 10 \mathrm{~mm}$, Ref. 67.755, Sarstedt, Nümbrecht, Germany).

Solutions of the fluorescent ligands in PBS and PBS containing 1\% BSA (both pH 7.4) were prepared from 5 mM stock solutions in DMSO. Spectra of the cresyl violet standard were recorded in EtOH. The pure solvents with the same DMSO content were used as reference. The absorption spectra were immediately recorded at $22{ }^{\circ} \mathrm{C}$. The emission spectra were recorded within 15-20 min after preparation of the solutions at a temperature of $22{ }^{\circ} \mathrm{C}$ using a 'medium scan rate'. The filter settings were 'auto' for the excitation and 'open' for the emission filter. Fluorescence spectra were recorded at two different slit adjustments (excitation/emission): $10 / 10 \mathrm{~nm}$ and $10 / 5 \mathrm{~nm}$. The emission starting point was set 15 nm above the excitation wavelength. From every emission spectrum the corresponding reference spectrum was subtracted and the resulting net spectrum was multiplied with the corresponding lamp correction spectrum. These corrected net spectra were integrated up to 850 nm . From every raw absorption spectrum the corresponding reference spectrum was subtracted to afford the net absorption spectra. The absorbance at the excitation wavelength was obtained from the net absorption spectra. The quantum yield was calculated according to the following equation:
$\phi_{F(X)}=\left(A_{s} / A_{x}\right)\left(F_{x} / F_{s}\right)\left(n_{x} / n_{s}\right)^{2} \phi_{F(S)}$
$A_{s}$ is the absorbance and $F_{s}$ the integral of the corrected net emission spectrum of the cresyl violet standard solution. $A_{x}$ and $F_{x}$ are the absorbance and the integral of the corrected net emission spectrum of the fluorescent ligand. The refractive indices of the solvents for the fluorescent ligands and the cresyl violet standard are denoted $n_{x}$ and $n_{s}$ (fluorescent ligands: $n_{P B S}$ $=1.33$; BSA content was neglected and cresyl violet: $\left.n_{E t O H}=1.36\right)$. The reported quantum yield of cresyl violet perchlorate (in this case $54 \%$ ) is referred to as $\phi_{F(X)}$.

### 5.4.3 Pharmacological Methods

## Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein or coexpressing the $h H_{3} R+G_{i \alpha 2}+\beta_{1} \gamma_{2}$ or $h H_{4} R+G_{i \alpha 2}+\beta_{1} \gamma_{2}$ proteins was described elsewhere. ${ }^{47}$

Radioligand competition binding assays were performed as described previously with minor adjustments using the following radioligands: $\left[{ }^{3} \mathrm{H}\right]$ UR-DE257 ${ }^{23}$ (specific activity $=32.89 \mathrm{Ci} / \mathrm{mmol}$, $\mathrm{hH}_{2} \mathrm{R}: K_{\mathrm{d}}=12.2 \mathrm{nM}, \mathrm{c}_{\text {final }}=20 \mathrm{nM}$ ) and $\left[{ }^{3} \mathrm{H}\right]$ histamine (Hartmann Analytic, Braunschweig, Germany; specific activity $=25 \mathrm{Ci} / \mathrm{mmol}^{\mathrm{h}} \mathrm{hH}_{3} \mathrm{R}: K_{\mathrm{d}}=12.1 \mathrm{nM}, \mathrm{c}_{\text {final }}=15 \mathrm{nM}, \mathrm{hH}_{4} \mathrm{R}: K_{\mathrm{d}}=15.9 \mathrm{nM}$, $\mathrm{c}_{\text {final }}=10 \mathrm{nM}$ ).

On the day of the experiment Sf 9 membranes were thawed and sedimented by centrifugation at $13,000 \mathrm{rpm}$ at $4{ }^{\circ} \mathrm{C}$ for 10 min . The membranes were resuspended in ice cold binding buffer ( 12.5 $\mathrm{mM} \mathrm{MgCl}_{2}, 1 \mathrm{mM}$ EDTA and 75 mM Tris/ $\mathrm{HCl}, \mathrm{pH} 7.4$; in the following referred to as BB ) and adjusted to a protein concentration of $2-4 \mu \mathrm{~g} / \mu \mathrm{L} .80 \mu \mathrm{~L}$ BB containing $0.2 \%$ BSA and the respective radioligand, followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ ), were added to every well of a 96 -well plate (in case of fluorescent ligands: Primaria clear flat bottom microplates, Corning, New York, USA; for other ligands: PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the
membrane suspension ( $10 \mu \mathrm{~L}$ ). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with 0.3\% polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with $200 \mu \mathrm{~L}$ scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h. The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

## Functional GTP $\boldsymbol{\gamma}$ S assay on Sf9 insect cell membranes

GTPץS assays were performed as described previously ${ }^{48}$ with minor modifications. $\left[{ }^{35}\right.$ S]GTP S S (specific activity $=1000 \mathrm{Ci} / \mathrm{mmol}$ ) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to 0.5-1.5 $\mu \mathrm{g} / \mu \mathrm{L}$.

Agonist mode: $80 \mu \mathrm{~L}$ of BB containing BSA ( $0.05 \%$ final), GDP ( $1 \mu \mathrm{M}$ final) and [ ${ }^{35} \mathrm{~S}$ ]GTP $\gamma \mathrm{S}$ ( 20 nCi final), followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ ) were added to every well of a 96-well plate (in case of fluorescent ligands: Primaria microplates; for other ligands: PP microplates). Incubation was started by addition of the membrane suspension ( $10 \mu \mathrm{~L}$ ). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTPץS assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine ( $1 \mu \mathrm{M}$ final).

## Cell culture

The preparation of stably transfected HEK cells (HEK293T-hH $\mathrm{H}_{2} \mathrm{R}-\mathrm{qs} 5^{26}$ and HEK293T-hH ${ }_{2} \mathrm{R}$ $\beta$ Arr2 ${ }^{33,49}$ ) was described elsewhere.

Cells were cultivated at $37{ }^{\circ} \mathrm{C}$ in a water saturated atmosphere containing $5 \% \mathrm{CO}_{2}$. Dulbecco's Modified Eagle Medium, containing $4.5 \mathrm{~g} / \mathrm{L}$ glucose, $3.7 \mathrm{~g} / \mathrm{L} \mathrm{NaHCO}_{3}, 110 \mathrm{mg} / \mathrm{L}$ sodium pyruvate (DMEM, Sigma-Aldrich Munich, Germany) and supplemented with $0.584 \mathrm{~g} / \mathrm{L}$ L-glutamine (Lglutamine solution, Sigma-Aldrich Munich, Germany), 1\% (v/v) Penicillin-Streptomycin (P/S, $10,000 \mathrm{U} / \mathrm{mL}$, Sigma-Aldrich Munich, Germany), $10 \%$ (v/v) fetal calf serum (FCS, Biochrom GmbH, Merck, Berlin, Germany) were used as a culture medium. Additionally, $100 \mu \mathrm{~g} / \mathrm{mL}$ hygromycin B (A.G. Scientific, Inc., San Diego, CA) and $400 \mu \mathrm{~g} / \mathrm{mL}$ G418 (Biochrom GmbH, Merck, Berlin, Germany) were added to the culture medium of HEK293T-hH ${ }_{2}$ R-qs5 cells, and $400 \mu \mathrm{~g} / \mathrm{mL}$ zeocin (InvivoGen, San Diego,USA) and $600 \mu \mathrm{~g} / \mathrm{mL}$ G418 were added to the culture medium of HEK293T$h H_{2} R-\beta A r r 2$ cells.

## Flow cytometric binding assays

All flow cytometric binding studies were performed with a FACS Calibur ${ }^{\text {TM }}$ flow cytomter (Becton Dickinson, Heidelberg, Germany), equipped with an argon laser ( 488 nm ) and a red diode laser $(635 \mathrm{~nm})$ according to general protocols ${ }^{26,50}$ with minor adjustments. The following instrument settings were used: FSC: E-1, SSC: $280 \mathrm{~V}, \mathrm{Fl}-3: 600 \mathrm{~V}$ and $\mathrm{Fl}-4: 420-550 \mathrm{~V}$. All samples were prepared in duplicate and recorded either in channel $\mathrm{Fl}-3$ (pyridinium dyes, excitation: 488 nm , emission filter: $>670 \mathrm{~nm}$ ) or in $\mathrm{Fl}-4$ (cyanine dyes, excitation: 635 nm , emission filter: $661 \pm 18 \mathrm{~nm}$ ). Sample measurement was complete after $30-45 \mathrm{~s}$ (this corresponds to approx. 20,000-90,000 gated events).

HEK293T-hH ${ }_{2} \mathrm{R}$-qs5 cells were seeded in a $175-\mathrm{cm}^{2}$ culture flask 5-7 days prior to the experiment. On the day of the experiment, cells were trypsinized and detached with fresh culture medium ( 5 mL ). After centrifugation ( $250 \mathrm{~g}, 10 \mathrm{~min}$ ) the cell pellet was resuspended in Leibovitz's L-15 culture medium ( $\mathrm{L}-15$ medium, Gibco/Life Technologies, Carlsbad, USA) and the concentration was adjusted to $0.5-1.0 \cdot 10^{6}$ cells $/ \mathrm{mL}$.

Saturation binding: $500 \mu \mathrm{~L}$ of the cell suspension were either added to $5 \mu \mathrm{~L}$ of DMSO/ $\mathrm{H}_{2} \mathrm{O}(30 / 70$, $\mathrm{v} / \mathrm{v}$, total binding) or to 5 L of famotidine in $\mathrm{DMSO} / \mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}$, unspecific binding, 300-fold excess to the fluorescent ligand). Incubation was started by the addition of $5 \mu \mathrm{~L}$ of fluorescent ligand in $\mathrm{DMSO} / \mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}, 100$-fold concentrated) in intervals of 1 min (measuring time per sample $=$ one concentration) starting with the lowest concentration of total binding. After 90 min of shaking in the dark at $25^{\circ} \mathrm{C}$, the samples were transferred to 5 mL polystyrol FACS tubes (Sarstedt, Nümbrecht,Germany) and immediately measured.

Competition binding: To $500 \mu \mathrm{~L}$ of cell suspension, $5 \mu \mathrm{~L}$ of competitor in DMSO/ $\mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v})$; 100 -fold concentrated) were added at increasing concentrations and $5 \mu \mathrm{~L}$ of fluorescent ligand (concentration in the assay: 50 nM (5.14) or $25 \mathrm{nM}(5.18)$ ) in DMSO/ $\mathrm{H}_{2} \mathrm{O}$ (30/70, v/v, 100-fold concentrated) in intervals of 1 min (measuring time per sample). The incubation time was 90 min at $25^{\circ} \mathrm{C}$.

Association: $500 \mu \mathrm{~L}$ of the cell suspension were either added to $5 \mu \mathrm{~L}$ of DMSO/ $\mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}$, total binding) or to $5 \mu \mathrm{~L}$ of famotidine in $\mathrm{DMSO} / \mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}$, unspecific binding, 300 -fold excess to the fluorescent ligand). Incubation started by addition of $5 \mu \mathrm{~L}$ of fluorescent ligand in DMSO $/ \mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}, 100$-fold concentrated, final concentration: $50 \mathrm{nM}(5.14), 15 \mathrm{nM}(5.16)$ or $25 \mathrm{nM}(5.18)$ ). The incubation at $37^{\circ} \mathrm{C}$ was stopped after different periods of time ( $0-120 \mathrm{~min}$ ) by measuring the samples.

Dissociation: $500 \mu \mathrm{~L}$ of the cell suspension were either added to $5 \mu \mathrm{~L}$ DMSO/ $\mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}$, total binding) or to $5 \mu \mathrm{~L}$ famotidine in $\mathrm{DMSO} / \mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}$, unspecific binding, 300 -fold excess to the fluorescent ligand). $5 \mu \mathrm{~L}$ of fluorescent ligand in $\mathrm{DMSO} / \mathrm{H}_{2} \mathrm{O}(30 / 70, \mathrm{v} / \mathrm{v}, 100$-fold concentrated, final concentration of the fluorescent ligands: $50 \mathrm{nM}(\mathbf{5 . 1 4}), 15 \mathrm{nM}(\mathbf{5 . 1 6})$ or 25 nM (5.18)) were added to every vessel and the samples were incubated at $25^{\circ} \mathrm{C}$ for 90 min . The samples were centrifuged ( $250 \mathrm{~g}, 3.5 \mathrm{~min}$,) and the supernatant, containing excess fluorescent ligand was aspirated. $500 \mu \mathrm{~L} \mathrm{pf}$ - 15 medium containing famotidine ( 300 -fold excess to the final fluorescent ligand concentration, $15 \mu \mathrm{M}, 4.5 \mu \mathrm{M}$ or $7.5 \mu \mathrm{M}$ ) were added to the cell pellet, before
the cells were resuspended. The incubation at $37^{\circ} \mathrm{C}$ was stopped after different periods of time (0-150 min ) by measuring the samples.

For data analysis the software FlowJo V10 (FlowJo, LLC, Ashland, USA) was used throughout. The geometrical mean values of $\mathrm{Fl}-3$ or $\mathrm{Fl}-4$ were obtained for a subpopulation of the gated cells which exhibited a high receptor density.

## High content Imaging

Fluorescent ligand binding experiments with adherent HEK293T-hH ${ }_{2}$ R-qs5 cells (IN Cell Analyzer)

For high content imaging of adherent cells, a wide-field cell imaging system, the IN Cell Analyzer 2000 (GE Healthcare, Little Chalfont, UK), was used. An objective with a 20-fold magnification and a numerical aperture of 0.45 , combined with a polychroic mirror (QUAD1), was used throughout. For imaging, two channels with different excitation/emission filters were applied: Cy5 channel (for cyanine dyes, excitation filter: 645/30 nm, emission filter: 705/72 nm, exposure time: 1,000 ms ) or Cy3 channel (pyridinium dyes, excitation filter: 543/22 nm, emission filter: 605/64 nm, exposure time: 1,000 ms) and DAPI channel (H33342, excitation filter: 350/50 nm, emission filter: 455/50 nm, exposure time: 90 ms ). 2.5-D images (imaging modality, the system uses the camera's CCD/sCMOS chip to integrate the signal over the specified $Z$ section and then deconvolves the result for a pseudo 3-D projection) were obtained throughout. In some cases a brightfield channel (excitation filter: 473/10 nm, emission filter: 455/50 nm, exposure time: 50 ms , imaging modality: 2-D images, the system acquires a standard two-dimensional image) was additionally applied. In every channel, 4 images were obtained per well. Binding experiments were performed in duplicate and were repeated at least twice.

One day prior to the experiment, HEK293T-hH2R-qs5 cells were trypsinized and detached with DMEM medium (high glucose without phenol red (Gibco/Life Technologies, Carlsbad, USA) containing $1 \%(v / v)$ Penicillin-Streptomycin (P/S, 10,000 U/mL, Gibco/Life Technologies, Carlsbad, USA) and $10 \%(\mathrm{v} / \mathrm{v})$ FCS (in the following referred to as DMEM w/o medium). The cell suspension was adjusted to $0.6-0.75 \cdot 10^{6}$ cells $/ \mathrm{mL}$ and $200 \mu \mathrm{l}$ (120,000-150,000 cells/well) were seeded in every well of a $\mu$-slide-96-well plate (Ibidi, Martinsried, Germany) using an automated reagent dispenser (Multidrop, Thermo Fisher Scientific, Waltham, USA). The cells were cultivated at $37{ }^{\circ} \mathrm{C}$ overnight in a water saturated atmosphere containing $5 \% \mathrm{CO}_{2}$.

On the day of the experiment, the medium was removed and cells (concentration approximately $3 \cdot 10^{8}$ cells/well) were covered with $80 \mu$ fresh DMEM w/o medium additionally containing $0.1 \%$ BSA (Albumin bovine fraction V, SERVA, Heidelberg, Germany) and H33342 (1 $\mu \mathrm{g} / \mathrm{mL}$ in $\mathrm{H}_{2} \mathrm{O}$, Sigma Aldrich, Munich, Germany).

Saturation binding: For determination of total binding, DMEM w/o medium (10 $\mu \mathrm{L}$ ) and for unspecific binding DMEM w/o medium ( $10 \mu \mathrm{~L}$ ) and famotidine as competitor ( 300 -fold referring to the fluorescent ligand) were added. Incubation was started by addition of DMEM w/o medium $(10 \mu \mathrm{~L})$ containing the respective concentrations of the fluorescent ligand (10-fold concentrated) to every well.

Competition binding: DMEM w/o medium ( $10 \mu \mathrm{~L}$ ) containing the investigated ligands at various concentrations ( 10 -fold concentrated) was added. For determination of unspecific binding famotidine ( $100 \mu \mathrm{M}$ final) was used. Incubation time started by addition of DMEM w/o medium $(10 \mu \mathrm{~L})$ containing the fluorescent ligand $\mathbf{1 8}$ (10-fold concentrated, 50 nM final) to every well.

After incubation at room temperature in the dark for 60 min , the medium was removed and the cells were washed with PBS ( $100 \mu \mathrm{~L}$ ) and covered with DMEM w/o medium ( $100 \mu \mathrm{~L}$ ) followed by immediate acquisition of images at $37^{\circ} \mathrm{C}$.

For kinetic studies the cells were seeded in a 96-well plate one day prior to the experiment as described before. For one association or dissociation experiment only four wells of the plate (two wells for total binding and unspecific binding respectively) were required. The remaining wells were used for other experiments like saturation or competition binding experiments. On the day of the experiment the medium was removed and cells were incubated with $100 \mu$ DMEM w/o medium containing $0.1 \%$ BSA and H33342 ( $1 \mu \mathrm{~g} / \mathrm{mL}$, Sigma Aldrich, Munich, Germany) for 1 h . The medium was carefully aspirated, the cells were washed with PBS ( $100 \mu \mathrm{~L}$ ) and covered with $100 \mu$ I DMEM w/o medium. Either $10 \mu \mathrm{~L}$ (total binding) or $20 \mu \mathrm{~L}$ (unspecific binding) medium were removed from the wells. DMEM w/o medium ( $10 \mu \mathrm{~L}$ ) containing famotidine ( 10 -fold concentrated, $15 \mu \mathrm{M}$ final) was added to the two wells of unspecific binding.

Association: DMEM w/o medium (10 $\mu \mathrm{L}$ ) containing the fluorescent ligand 18 (10-fold concentrated, 50 nM final) was added to every well, and the plate was immediately transferred to the IN Cell Analyzer 2000. Images at different time points between 0 h and 1 h were acquired at $37^{\circ} \mathrm{C}$.

Dissociation: DMEM w/o medium ( $10 \mu \mathrm{~L}$ ) containing the fluorescent ligand 18 (10-fold concentrated, 50 nM final) was added to every well, and the plate was incubated in the dark for 1 $h$. The medium was aspirated and DMEM w/o medium ( $100 \mu \mathrm{~L}$ ) containing famotidine ( $15 \mu \mathrm{M}$ ) was added. Immediately, images at different time points between 0 h and 1 h were acquired at $37^{\circ} \mathrm{C}$.

For data analysis, the software Developer Toolbox 1.9.2 (IN Cell Investigator, GE Healthcare, Little Chalfont, UK) was used. For counting the nuclei per image a segmentation algorithm was applied to the images of the DAPI-channel (nuclei staining) to define the targets (nuclei) and the measure "count" (output: number of targets contained within the region of interest) was applied to the images. For the quantification of bound ligand an object segmentation algorithm was applied to the images of Cy3/Cy5 channel to define the targets (areas where the fluorescent ligand is bound). In case of the pyridinium labeled ligands, the density measure "density level" (measures the gray level intensity within the targets; uncalibrated intensity unit for IN Cell images, the higher the value, the brighter the pixel) was applied to the targets. As statistical function "sum" (output: sum of the density levels of all identified targets within one image) was chosen and the result was divided through the nuclei count. In case of the cyanine labeled ligands the density measure "Density x Area" (mean density within the target outline multiplied by its area, i.e. total density within the target outline) was applied to the targets. As statistical function "sum" (output: sum of the Density $x$ Area of all identified targets within one image) was chosen and the result was divided through the nuclei count.

## Imaging Flow Cytometry (ImageStreamX)

Imaging flow cytometry was performed with an ImageStreamX Imaging Cytometer (Amnis/Merck Millipore, Darmstadt, Germany), equipped with an objective with 40-fold magnification and a numerical aperture of 0.75 . For imaging, an argon laser ( 488 nm , excitation wavelength) and a diode laser ( 785 nm , side scatter) were used, instrument settings were 488 nm laser: $75-85 \mathrm{~mW}$, 785 nm laser: 13 mW and velocity $60 \mathrm{~mm} / \mathrm{s}$. Images were obtained in channel CH5 (emission filter: 702/85 nm). Measurement was completed after counting 1,000 to 1,500 cells within the defined area limits (bright field, upper limit: 250 and lower limit: 100, measuring time 1.5 to 2.0 min per sample). Binding experiments were performed in duplicate and were repeated at least twice.

HEK293T-hH2R-qs5 cells were detached with DMEM w/o medium. After centrifugation ( $300 \mathrm{~g}, 5$ min ) the cell pellet was resuspended in fresh DMEM w/o medium. The cell suspension was adjusted to $2 \cdot 10^{6}$ cells $/ \mathrm{mL}$.

Saturation binding: For determination of total binding DMEM w/o medium ( $20 \mu \mathrm{~L}$ ) containing $1 \%$ BSA (Albumin bovine fraction V, SERVA, Heidelberg, Germany) and for unspecific binding DMEM w/o medium ( $20 \mu \mathrm{~L}$ ) containing $1 \%$ BSA and famotidine as competitor ( 300 -fold referring to the fluorescent ligand) were added to 1.5 ml reaction vessels (Sarstedt, Nümbrecht, Germany). The cell suspension ( $160 \mu \mathrm{~L}$ ) was transferred to the vessels and DMEM w/o medium ( $20 \mu \mathrm{~L}$ ) containing $1 \%$ BSA and the fluorescent ligand in different concentrations (10-fold concentrated) was added. Samples were prepared at intervals of 2 min (measuring time per sample) starting with the highest concentration. After incubation for 60 min at room temperature in the dark, samples were filtered through a $70 \mu \mathrm{~m}$ nylon cell strainer (Falcon/Corning Inc., New York, USA) and images of the suspended cells were acquired with an ImageStreamX Imaging cytometer.

Competition binding: DMEM w/o medium ( $210 \mu \mathrm{~L}$ ) containing the investigated ligands in various concentrations ( 10 -fold concentrated) was added to 1.5 ml reaction vessels. For determination of unspecific binding, famotidine ( $100 \mu \mathrm{M}$ final) was used. The cell suspension ( $160 \mu \mathrm{~L}$ ) was transferred to the vessels and DMEM w/o medium ( $10 \mu \mathrm{~L}$ ) containing the fluorescent ligand 14 (10-fold concentrated, 70 nM final) was added. All following steps were carried out as described for saturation binding.

For data analysis, the software IDEAS 6.0 (Amnis/Merck Millipore, Darmstadt, Germany) was used. From the gated cells only focused single cells were included in the data analysis. Furthermore, cells with high fluorescence intensity in the cytoplasm (dead or dying cells) were excluded. Then a mask for the cell membrane was created, allowing analysis of the fluorescent intensity in the area of the cell membrane. Additionally, the fluorescent intensity of the whole cell was analyzed.

## Confocal Microscopy

Confocal microscopy was performed with a Zeiss Axiovert 200 M microscope equipped with the LSM 510 Laser scanner. A 63x/1.40 oil immersion objective was used.

Two days prior to the experiment, HEK293T-hH2R-qs5 cells were trypsinized and seeded in a ibiTreat $\mu$-slide 8-well chambered coverslip (Ibidi, Planegg, Germany) in DMEM ( $0.6 \cdot 10^{6}$ cells $/ \mathrm{mL}$, $250 \mu \mathrm{~L}$ per well) containing $10 \%$ FCS and $1 \%$ P/S ( $10,000 \mathrm{U} / \mathrm{mL}$, Sigma-Aldrich Munich, Germany). On the day of the experiment, the confluency of the cells was approximately 80-90\%. The culture medium was replaced with L-15 containing $5 \%$ FCS and $1 \% \mathrm{P} / \mathrm{S}(120 \mu \mathrm{~L})$. For the determination of total binding, blank $\mathrm{L}-15$ medium ( $40 \mu \mathrm{~L}$ ) and L-15 medium containing the respective fluorescent ligand ( $40 \mu \mathrm{~L}, 5$-fold concentrated, 100 nM final) were added. Unspecific binding was determined by analogy; with the exception that blank medium was replaced with L-15 medium containing famotidine ( $40 \mu \mathrm{~L}, 5$-fold concentrated, $30 \mu \mathrm{M}$ final). Images of total and unspecific binding were acquired after an incubation period of 20 min at room temperature. Table 5.12 shows the settings for the detection of the investigated fluorescent ligands.

Table 5.11. Settings of the confocal microscope for the detection of the fluorescent ligands 5.14, 5.16 and 5.18.

| Compounds | Excitation (laser intensity) | Filter | Pinhole ( $\mu \mathrm{m})$ |
| :--- | :--- | :--- | :--- |
| $\mathbf{5 . 1 4}$ | $488 \mathrm{~nm}(10 \%)$ | LP 560 | 106 |
| $\mathbf{5 . 1 6}$ | $633 \mathrm{~nm}(10 \%)$ | LP 650 | 122 |
| $\mathbf{5 . 1 8}$ | $633 \mathrm{~nm}(10 \%)$ | LP 650 | 122 |

## Beta-Arrestin2 recruitment assay

The $\beta$-Arrestin2 recruitment assays were performed as described previously for the $H_{1} R$ using HEK293T-hH ${ }_{2} R$ - $\beta$ Arr2 cells, stably expressing the $h_{2} R$-ElucC and $\beta$ Arr2-ElucNfusion constructs. ${ }^{49}$

One day prior to the experiment, HEK293T-hH2R- $\beta$ Arr2 cells were trypsinized and detached with DMEM medium (high glucose without phenol red (Sigma Aldrich, Munich, Germany) containing $1 \%(\mathrm{v} / \mathrm{v}) \mathrm{P} / \mathrm{S}$ and $5 \%(\mathrm{v} / \mathrm{v})$ FCS. The cell suspension was adjusted to $1.1 \cdot 10^{6} \mathrm{cells} / \mathrm{mL}$ and $90 \mu \mathrm{l}$ (100,000 cells/well) were seeded in every well of a sterile, luciferase assay compatible, F-bottom 96 -well plate (Cellstar ${ }^{\circledR}$, Greiner Bio-One, Kremsmünster, Österreich). The cells were cultivated at $37{ }^{\circ} \mathrm{C}$ overnight in a water saturated atmosphere containing $5 \% \mathrm{CO}_{2}$. The investigated ligands were added at increasing concentrations ( $10 \mu \mathrm{~L}$ ), and the plate was incubated at $25^{\circ} \mathrm{C}$ for 60 min under shaking. $50 \mu \mathrm{~L}$ of the medium were removed, and $50 \mu \mathrm{~L}$ of Bright-Glo reagent (Promega, Madison, USA) were added. Bioluminescence was immediately measured for 1 s per well using a GENios Pro microplate reader (Tecan, Salzburg, Austria).

### 5.3.4 Data analysis

Retention factors $k$ were calculated according to $k=\left(t_{R^{-}} t_{0}\right) / t_{0}\left(t_{0}=\right.$ dead time; $t_{\mathrm{R}}=$ retention time). Corrected counts per minute (ccpm) from the GTP $\gamma S$ assay (agonist mode) were plotted against the $\log (c o n c e n t r a t i o n ~ o f ~ t h e ~ t e s t ~ c o m p o u n d), ~ a n d ~ d a t a ~ w e r e ~ a n a l y z e d ~ b y ~ a ~ f o u r ~$ parameter logistic equation (GraphPad Prism Software 5.0, GraphPad Software, San Diego, CA), followed by normalization ( $0 \%=$ water value (basal activity), $100 \%=$ "top" histamine equation) and analysis by four-parameter logistic equation (log(agonist) vs. response - variable slope, GraphPad Prism). Data of the GTPYS assay (antagonist mode) were analysed by a four parameter logistic equation (GraphPad Prism), followed by normalization (100\% = "top" of the fourparameter logistic fit, $0 \%=$ unspecifically bound radioligand (ccpm) determined in the presence of famotidine at $100 \mu \mathrm{M}$ ) and analysis by four-parameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism). $\mathrm{p} / C_{50}$ values were converted into $\mathrm{p} K_{B}$ values according to the Cheng-Prusoff equation ${ }^{51}$. The luminescence (RLU) from the $\beta$ Arrestin2 recruitment assay (agonist mode) were plotted against log(concentration of the test compound) and analyzed by a four parameter logistic equation (GraphPad Prism) followed by normalization $(0 \%=$ water value (basal activity), $100 \%=$ "top" histamine equation) and analysis by fourparameter logistic equation (log(agonist) vs. response - variable slope, GraphPad Prism). Specific binding data from saturation binding experiments were plotted against the "free" fluorescently labeled ligand concentration and analyzed by a two-parameter equation describing hyperbolic binding (one site - specific binding, GraphPad Prism). Specific binding data from association binding experiments were analyzed by a two parameter equation describing an exponential rise to a maximum (one-phase association, GraphPad Prism) to obtain the observed association constant $k_{\text {obs }}$. Specific binding data from dissociation binding experiments were analyzed by a three parameter equation (one phase decay, GraphPad Prism) to obtain the dissociation rate constant $k_{\text {off. }}$. Kinetic dissociation constants $K_{\mathrm{d}(\mathrm{kin})}$ were calculated from $k_{\text {on }}$ and $k_{\text {off }}\left(k_{\text {on }}=\left(k_{\text {obs }}-\right.\right.$ $\left.\left.k_{\text {off }}\right) /[L] ; K_{\text {d (kin) }}=k_{\text {off }} / k_{\text {on }}\right)$. Specific binding data from association and dissociation binding experiments were normalized $\left(100 \%=Y_{\max }\right.$ (association) or $Y_{0}$ (dissociation)). Total binding data from radioligand and fluorescent ligand competition binding experiments were plotted against $\log$ (concentration competitor) and analyzed by a four-parameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism), followed by normalization ( $100 \%=$ "top" of the four-parameter logistic fit, $0 \%=$ unspecifically bound radioligand/ fluorescent ligand determined in the presence of famotidine at $100 \mu \mathrm{M}$ ). Normalized data from competition binding experiments was analyzed by a four-parameter logistic equation (log(inhibitor) vs response variable slope, GraphPad Prism) and obtained $\mathrm{p} / C_{50}$ values were converted into $\mathrm{p} K_{\mathrm{i}}$ values according to the Cheng-Prusoff equation. ${ }^{51}$

### 5.4 Summary and Conclusion

The introduction of different fluorophores by derivatization of amino-functionalized precursors, structurally related to $B M Y 2536$, led to fluorescently labeled $H_{2} R$ antagonists. The highest affinities on the $\mathrm{hH}_{2} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ values $>7.0$ ) in radioligand competition assays were obtained by the pyridinium labeled ligands 5.12-5.14 and the cyanine labeled ligands 5.16 (positively charged fluorophore, net charge: $2^{+}$) and $\mathbf{5 . 1 8}$ (electroneutral fluorophore, net charge: $1^{+}$). Interestingly, labeling with S0387, the cyanine dye with the negative net charge, led to a decrease in $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ values < 6.0). While the linker length (4-7 carbon atoms) had no significant influence on the $\mathrm{hH}_{2} \mathrm{R}$ affinity within the pyridinium ligands, the cyanine ligands with the hexyl linker ( $\mathbf{5 . 1 6}$ and 5.18) showed an increased $\mathrm{hH}_{2} \mathrm{R}$ affinity compared to the butyl linker derivatives. Even though the low selectivity towards the $\mathrm{hH}_{3} \mathrm{R}$ limited the application to recombinant systems, the investigated fluorescent ligands proved to be useful tools for binding studies using different techniques (flow cytometry and high content imaging). The ligands 5.12-5.18 and 5.20 bound in a saturable manner to the $\mathrm{hH}_{2} \mathrm{R}$ (flow cytometry) and the determined $K_{\mathrm{d}}$ values (best results: 5.12: $27.9 \mathrm{nM}, 5.13: 14.9 \mathrm{nM}, 5.14: 19.7 \mathrm{nM}, 5.16: 13.9 \mathrm{nM}$ and 5.18: 48.2 nM ) were in good agreement with the corresponding $K_{\mathrm{i}}$ values. The $K_{\mathrm{d}(\text { kin })}$ values of 5.14, $\mathbf{5 . 1 6}$ and 5.18, calculated from kinetic experiments (nonlinear regression, flow cytometry) were consistent with the $K_{\mathrm{d}}$ values determined in saturation binding experiments. This indicates that the investigated fluorescent ligands follow in part the law of mass action even though they showed an incomplete dissociation (insurmountable antagonism). A similar behavior has been reported for several closely related ligands like the radioligand [ ${ }^{3} \mathrm{H}$ ]UR-DE257. Its amine precursor 5.10 as well as related squaramide type derivatives were reported as insurmountable antagonists, which caused a concentration-dependent depression of the maximal response to the agonist relative to investigated standard ligands (guinea pig right atrium) ${ }^{23}$. Investigation of the association and dissociation kinetics of the cyanine labeled ligand 5.18 with high content imaging (INcCell Analyzer) revealed also incomplete dissociation and showed that the residual bound ligand is still located in the cell membrane. A possible explanation for the (pseudo-)irreversible binding of the fluorescent ligands $\mathbf{5 . 1 4 , 5 . 1 6}$ and 5.18 is a slow rate of dissociation from the receptor as also suggested for the radioligand [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257. ${ }^{23}$ Nonetheless, the fluorescent ligands $\mathbf{5 . 1 4}$ and 5.18 can be used for the determination of binding affinities of unlabeled ligands in competition binding assays (shown for flow cytometry and two high content imaging systems: IN Cell Analyzer and imaging flow cytometry).

The high affinity fluorescent $\mathrm{H}_{2} \mathrm{R}$ ligands $\mathbf{5 . 1 2 - 5 . 1 4 ,} \mathbf{5 . 1 6}$ and $\mathbf{5 . 1 8}$ are an attractive nonradioactive alternative to the structurally related radioligand $\left[{ }^{3} \mathrm{H}\right]$ UR-DE $257^{23}$ with similar pharmacological properties. Additionally these fluorescent ligands can be versatile molecular tools giving access to a plethora of optical techniques such as confocal microscopy, FRET, FRAP, TIRF, high content imaging and fluorescence polarization.

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# 6. Carbamoylguanidine- Type $\mathrm{H}_{2}$ R Ligands: Exploration of Stability and Selectivity Compared to the Acylguanidine- Analogues 

Note: the synthesis of the intermediates 6.13-6.18 and the cabamoylguanidines 6.47-6.52 as well as the investigation of the chemical stability of 6.49, 6.50, 6.52, UR-Bit22, UR-Bit23 and UR-Bit29 were performed by Claudia Honisch during her Master Thesis 2015.

The radioligand binding experiments at the dopamine receptors were performed by Lisa Forster during her doctoral thesis (ongoing).

### 6.1 Introduction

$\mathrm{N}^{\mathrm{G}}$-acylated hetarylpropylguanidines represent a class of potent histamine $\mathrm{H}_{2} \mathrm{R}$ agonists. ${ }^{1-3}$ The first generation $\mathrm{N}^{\mathrm{G}}$-acylguanidine ligands with an imidazole mojety lacked subtype selectivity, especially over the $\mathrm{H}_{3} \mathrm{R}$ and $\mathrm{H}_{4} \mathrm{R}$ (e.g. UR-AK24, Figure 6.1). ${ }^{3}$ The bioisosteric replacement of the imidazole with a amino(methyl)thiazole moiety resulted in ligands with improved selectivity for the $\mathrm{H}_{2} \mathrm{R}$ and retained $\mathrm{H}_{2} \mathrm{R}$ potency (Figure 6.1 and Figure 6.2). ${ }^{1,4}$ Investigation of optically active acylguanidine-type compounds (e.g. UR-AK24 and UR-PG267, Figure 6.1) in the GTPase assay revealed eudismic ratios from 1.1 to 3.2 , indicating that stereochemistry plays only a minor role, if any. ${ }^{4}$


UR-AK24 (racemate)
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC} 50: 7.17, \alpha: 0.87$
$\mathrm{hH}_{3} \mathrm{R}: \mathrm{p} K_{\mathrm{b}}: 7.71$
$\mathrm{hH}_{4} \mathrm{R}: \mathrm{pEC} C_{50}: 7.82, \alpha: 0.89$
eudismic ratio 1.5

## (R)-UR-AK24

$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC} 50: 6.92, \alpha: 1.01$
(S)-UR-AK24
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pE} C_{50}$ : 7.10, $\alpha: 1.07$


UR-PG267 (racemate)
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC}_{50}: 7.30, \alpha: 0.82 \quad$ (R)-UR-PG267
$\mathrm{hH}_{3} \mathrm{R}: \mathrm{p} K_{\mathrm{B}}:<5$
$\mathrm{hH}_{4} \mathrm{R}: \mathrm{p} K_{\mathrm{B}}:<5$
eudismic ratio 1.9
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC} \mathrm{C}_{50}: 7.27, \alpha: 0.65$
(S)-UR-PG267
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC}_{50}$ : 7.55, $\alpha: 0.87$

Figure 6.1. Monovalent $N^{G}$-acylated 3-(imidazol-4-yl)propylguanidine UR-AK24 and the corresponding $N^{G}$-acylated 3-(2-amino-4-methylthiazol-5-yl)propylguanidine UR-PG267. Potencies were determined in a steady-state GTPase assay on membrane preparations of $\mathrm{Sf9}$ insect cells expressing the respective receptor.

Interestingly, a broad variety of aliphatic and aromatic hydrocarbon residues was well tolerated in this class of $\mathrm{H}_{2} \mathrm{R}$ agonists (Figure 6.2). An increase in $\mathrm{H}_{2} \mathrm{R}$ potency was achieved by linking two acylguanidine pharmacophores together via alkyl spacer. ${ }^{2}$ Bivalent ligands with an n-octyl linker (e.g. UR-AK381, Figure 6.2), showed the highest potency ( $\mathrm{pEC} C_{50}: 8.11, \alpha: 0.53$ ), although spacer length is too short to bridge the two orthosteric binding pockets of the individuals protomers of a putative receptor dimer. ${ }^{2}$ The increased potency compared to that of the corresponding monovalent ligands seems to result from binding to an additional (allosteric) binding site of the same receptor. ${ }^{2}$


$\mathrm{R}=$


UR-Bit22 $\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC} 50: 5.83$ $\alpha: 0.56$


UR-Bit23
$\mathrm{hH}_{2} \mathrm{R}$ : pEC 50 : 7.02
$\alpha: 0.68$


UR-Bit29
$\mathrm{hH}_{2} \mathrm{R}$ : $\mathrm{pEC} \mathrm{C}_{50}: 7.30$ $\alpha: 0.71$


UR-AK421
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC}_{50}: 7.61$ $\alpha: 0.42$


UR-AK381
$\mathrm{hH}_{2} \mathrm{R}$ : pEC 50 : 8.11
$\alpha$ : 0.53


UR-Bit24
$\mathrm{hH}_{2} \mathrm{R}: \mathrm{pEC} \mathrm{C}_{50}: 7.65$
$\alpha: 0.79$

Figure 6.2. Representative $\mathrm{H}_{2} \mathrm{R}$ agonists: monomeric and bivalent $\mathrm{N}^{\mathrm{G}}$-acylated 3-(2-amino-4-methylthiazol-5yl)propylguanidines and the bivalent carbamoylguainidine UR-NK22. ${ }^{1,2,5}$ Potencies were determined in a steady-state GTPase assay on membrane preparations of Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{s \alpha \mathrm{~s}}$ fusion protein.

As it was reported that $N^{G}$-acylguanidines undergo hydrolytic cleavage upon storage in aqueous solution, more stable analogues are needed. ${ }^{5,6}$ A bioisosteric approach replacing the $N^{G}$ acylguanidine structure with a carbamoylguanidine has proven useful. ${ }^{5,7,8}$ This was successfully
applied in case of the bivalent $\mathrm{N}^{\mathrm{G}}$-acylated hetarylpropylguanidine $\mathrm{H}_{2} \mathrm{R}$ agonists leading to highly potent ligands with improved long term stability (e.g. UR-NK22 (Figure 6.2), no decomposition after 7 days in PBS, PH 7.4 at RT). ${ }^{5}$

The aminothiazole moiety is a privileged structure for dopamine receptors. The most prominent example is the nonselective $D_{2-4} R$ agonist pramipexole (Figure 6.3). But also bulkier pramipexole derivatives such as $\mathrm{CJ}-1639$ with high $\mathrm{D}_{2-4} R$ affinity were reported. ${ }^{9}$ In preliminary binding studies on the $h D_{\text {2long }} R$ and $h D_{3} R$ variants, UR-NK22 and other bivalent amino(methyl)thiazole containing ligands showed in part moderate to high affinity towards these receptors (ongoing Dissertation Lisa Forster).



Pramipexole


Figure 6.3. $\mathrm{D}_{2-4} \mathrm{R}$ ligands pramipexole and $\mathrm{CJ}-1639$.

For exploration of the structure-activity relationship $\left(\mathrm{H}_{2} \mathrm{R}\right)$ and the structure-selectivity relationships ( $H_{2} R$ versus $H_{1} R, H_{3} R$ and $H_{4} R$ ) of this class of compounds, in addition to bivalent ligands, a series of carbamoylguanidines with various aminothiazole-based substructures, i.e., the 3-(2-amino-4-methylthiazol-5-yl)propyl moiety, a conformationally constrained (aminothiazolyl)phenyl and a 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl portion were synthesized (Figure 6.4) in this doctoral project. Additionally, two homobivalent ligands were prepared by replacement of the amino(methyl)thiazolepropyl moiety in UR-NK22 with either a (aminothiazolyl)phenyl or a pyrazolylpropyl moiety.

The synthesized monovalent and bivalent ligands were investigated in competition binding and functional assays ( $G T P \gamma S$ binding and $\beta$-arrestin2 recruitment assay). In addition, selected compounds were investigated for $D_{2 / 3} R$ binding affinity on homogenates of HEK cells stably expressing the $h D_{\text {2long }} R$ or $h D_{3} R$.



Figure 6.4. Bioisosteric replacement of the $\mathrm{N}^{\mathrm{G}}$-acylguanidine in $\mathrm{H}_{2} \mathrm{R}$ agonists by a carbamoylguanidine and conformal restriction by replacement of the 3-(2-amino-4-methylthiazol-5-yl)propyl moiety by either an (aminothiazolyl)phenyl or an 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety.

### 6.2 Results and DIscussion

### 6.2.1 Chemistry

The synthesis of the Boc-protected amine building block 3-(2-amino-4-methylthiazol-5-yl) propylamine (6.5) is outlined in Scheme 6.1 according to published procedures. ${ }^{1}$ Starting from 6-chlorohexan-2-one, a Gabriel reaction led to the phthalimide 6.1. Subsequent bromination at room temperature resulted in the thermodynamically more stable intermediate 6.2. After evaporation of the solvent, the residue was treated with thiourea to give the 2 -amino- 4 methylthiazole derivative 6.3 in a substitution/ring closure reaction. Protection of the free amino group by a tert-butoxycarbonyl function and subsequent hydrazinolysis led to the amine intermediate 3-(2-amino-4-methylthiazol-5-yl) propylamine (6.5).


Scheme 6.1. Synthesis of the Boc-protected aminothiazole 6.5. Reagents and conditions: i) phthalimide, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{DMF}$, $80{ }^{\circ} \mathrm{C}, 24 \mathrm{~h}, 70 \%$; ii) $\mathrm{Br}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 1,6$-dioxane, $\mathrm{RT}, 1 \mathrm{~h}$, no purification; iii) DMF, $100{ }^{\circ} \mathrm{C}$ to $\mathrm{RT}, \mathrm{ON}, 86 \%$; iv) di-tertbutyldicarbonate, triethylamine, 4-(dimethylamino)-pyridine, $\mathrm{CHCl}_{3}, \mathrm{RT}, \mathrm{ON}, 20 \%$; v) hydrazine-monohydrate, EtOH, RT, ON, 83\%.

The synthesis of the second amine building block 6.7 was carried out by analogy with the synthesis of 2-guanidino-4-(3-phthalimidophenyl)thiazole hydrobromide (3.6) and is outlined in Scheme 6.2. ${ }^{10}$ The ketone 3.4 was treated with bromine and subsequently with thiourea in order to form the protected aminothiazole building block 6.6 in a good yield of $78 \%$ over two steps. Deprotection of the phthalimide group in a mixture of HCl and acetic acid afforded the conformationally constrained amine building block 6.7.



Scheme 6.2. Synthesis of 2-amino-4-(3-aminophenyl)thiazole (6.7). Reagents and conditions: i) $\mathrm{Br}_{2}, \mathrm{HBr}$ in $\mathrm{AcOH}, \mathrm{CHCl}_{3}$, RT, 1h, no purification; ii) EtOH, MeCN, reflux, $3 \mathrm{~h}, 78 \%$; iii) $\mathrm{HCl}, \mathrm{AcOH}$, reflux, $\mathrm{ON}, 60 \%$.

The synthetic route of the third amine building block was adopted from the synthesis of pramipexole (Scheme 6.3). ${ }^{11}$ 4-Amino cyclohexanol was amino protected using
$N$-(ethoxycarbonyl)phthalimide; then the hydroxy group was oxidized by pyridinium chlorochromate. The resulting ketone 6.9 was brominated in alpha position and subsequently treated with thiourea to afford the phthalimide protected aminothiazole building block 6.11. 4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (6.12) was obtained by refluxing 6.11 in a mixture of hydrochloric acid and acetic acid.


Scheme 6.3. Synthesis of the 4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (6.12). Reagents and conditions: i) $\mathrm{K}_{2} \mathrm{CO}_{3}$, $\mathrm{H}_{2} \mathrm{O}, \mathrm{RT}, 30 \mathrm{~min}, 98 \%$; ii) pyridinium chlorochromate, anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, 3.5 \mathrm{~h}, 71 \%$; iii) $\mathrm{Br}_{2}, 1,6$-dioxane, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}$, 1.5 h , no purification; iv) DMF, $100^{\circ} \mathrm{C}, 2 \mathrm{~h}, 92 \%$; v) $\mathrm{HCl}, \mathrm{AcOH}$, reflux, $\mathrm{ON}, 73 \%$.

The $N^{G}$-carbamoylated guanidines were synthesized by guanidinylation of the amine building blocks 6.5, 6.7 and 6.12. S-methylcarbamoyl thiourea derivatives were used as guanidinylating reagents. These S-methylcarbamoyl thiourea derivatives (6.13-6.25) were synthesized by treatment of $N$-tert-butoxycarbonyl-S-methylisothiourea (3.32, synthesis in chapter 3 ) with the respective isocyanate (Scheme 6.4) in the presence of TEA at room temperature. The isocyanates were either commercially available (synthesis of 6.13-6.18) or were prepared by Curtius rearrangement from the carboxylic acid (see synthesis of 6.19-6.25). The branched carboxylic acids, which were previously synthesized in our group, ${ }^{1,12}$ were treated with oxalyl chloride and the resulting acid chloride was converted into acyl azide by treatment with sodium azide. Thermal decomposition of the acyl azide led to the isocyanate, which was used for the synthesis of the corresponding S-methylcarbamoyl thiourea derivative (Scheme 6.4). The $\mathrm{N}^{\mathrm{G}}$ carbamoylated guanidines 6.47-6.67 were prepared by treating the amine building blocks 6.5, 6.7 or 6.12 with the respective S-methylcarbamoyl thiourea derivative 6.13-6.25 in the presence of $\mathrm{HgCl}_{2}$ and base. The resulting protected carbamoylguanidine-type intermediates 6.26-6.46 were treated with TFA to obtain the $N^{G}$-carbamoylated guanidines 6.47-6.67 in a yield of 16-75\%. after purification by preparative HPLC.



c,

Scheme 6.4. Synthesis of the S-methylcarbamoyl thiourea derivatives 6.13-6.25 and the $\mathrm{N}^{\mathrm{G}}$-carbamoylated guanidines 6.47-6.67. Reagents and conditions: i) TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}, \mathrm{ON}, 57-85 \%$; ii) Oxalylchloride, $\mathrm{DMF}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{O}^{\circ} \mathrm{C}-\mathrm{RT}, 25$ min, no purification; iii) Sodium azide in $\mathrm{H}_{2} \mathrm{O}$, acetone, ice bath, 30 min , no purification; iv) $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, reflux, 30 min , no purification; v) TEA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}$, ON, $38-77 \%$ over four steps; vi) $\mathrm{HgCl}_{2}$, TEA, anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, Ar-atmosphere, RT, ON, 35-97\%; vii) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, RT, ON, $16-75 \%$

The bivalent carbamoylguanidine-type compounds 6.70 and 6.71 were synthesized using established protocols (Scheme 6.5). ${ }^{5,13}$ The amine precursor 6.7 or 3-(1-trityl-1H-pyrazol-4yl )prop-1-yl-amine was treated with guanidinylating reagent 3.33 in the presence of $\mathrm{HgCl}_{2}$ and base. Subsequent treatment of the protected intermediates 6.68 and 6.69 with TFA afforded the bivalent $N^{G}$-carbamoylated guanidines $\mathbf{6 . 7 0}$ and 6.71.


Scheme 6.5. Synthesis of the bivalent $N^{G}$-carbamoylated guanidines (6.70, 6.71). Reagents and conditions: i) $\mathrm{HgCl}_{2}$, TEA, anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, Ar-atmosphere, RT, ON, 81-91\%; ii) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT}$, ON, 52-67\%.

### 6.2.2 Chemical stability of monovalent carbamoylguanidines compared to acylguanidines

$N^{G}$-Carbamoylguanidines are reported to show higher stability in basic solution compared to the respective $\mathrm{N}^{\mathrm{G}}$-acylguanidines. ${ }^{6}$ Previous investigations on stability (assay conditions: PBS pH 7.4) of bivalent aminothiazole-containing carbamoylguanidines compared to their acylguanidine counterparts showed that after 7 days $55 \%$ of the acylated guanidine decomposed while the carbamoylated guanidine remained intact. ${ }^{5}$

In order to compare the chemical stability of the monovalent thiazole containing carbamoylated guanidines with the corresponding acylated guanidines, and to investigate the influence of various hydrocarbon residues, the compounds $6.49,6.50$ and 6.52 as well as the acylguanidines UR-Bit22, UR-Bit23 and UR-Bit29 were dissolved in PBS (pH 7.4), incubated at RT for 7 days and analysed by analytical HPLC (conditions see experimental section). Whereas the carbamoylated guanidines remained stable over this period of time (Figure 6.5), decomposition of the acylated guanidines was highly dependent on the nature of hydrocarbon residue (Figure 6.6). After 7 days approximately $62 \%$ of the acylguanidine UR-Bit23 was decomposed. However, only $51 \%$ of URBit29 and only $33 \%$ of UR-Bit22 was decomposed after the same time. The formation of a decomposition product ( $\mathrm{t}_{\mathrm{R}}: 1.97 \mathrm{~min}$ ) could be observed over a period 7 days (Figure 6.6).


B


C


Figure 6.5. HPLC chromatograms ( $\lambda: 220 \mathrm{~nm}$ ) of the $N^{G}$-carbamoylated guanidines after different time of incubation in PBS (pH 7.4): (A) 6.49, $\mathrm{t}_{\mathrm{R}}: 9.14 \mathrm{~min}$; (B) 6.50, $\mathrm{t}_{\mathrm{R}}: 8.23 \mathrm{~min}$; and (C) 6.52, $\mathrm{t}_{\mathrm{R}}: 9.52 \mathrm{~min}$.


Figure 6.6. HPLC chromatograms ( $\lambda$ : 220 nm ) of the $N^{G}$-acylated guanidines after different time of incubation in PBS ( pH 7.4 ): (A) UR-Bit22 ( $\mathrm{t}_{\mathrm{R}}: 6.78 \mathrm{~min}$ ), Inset: reduced scaling, decomposition product ( $\mathrm{t}_{\mathrm{R}}: 1.97$ ); (B) UR-Bit23, $\mathrm{t}_{\mathrm{R}}: 7.71 \mathrm{~min}$, Inset: reduced scaling, decomposition product ( $\mathrm{t}_{\mathrm{R}}$ : 1.96); and (C) UR-Bit29, $\mathrm{t}_{\mathrm{R}}: 10.38 \mathrm{~min}$, Inset: reduced scaling, decomposition product ( $t_{R}$ : 1.97).

### 6.2.3 Biological Evaluation

## $\mathrm{hH}_{2} \mathrm{R}$ affinities and subtype selectivities

The aminothiazole containing ligands 6.47-6.67, the bivalent ligands 6.70 and 6.71 , histamine, UR-NK22, UR-Bit23 and pramipexole were investigated in equilibrium competition binding experiments on membrane preparations from $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein using the antagonist [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257 ${ }^{14}$ as radioligand. Selected displacement curves are shown in Figure 6.7 and the results are summarized in Table 6.1. The selectivity of representative compounds for the $h H_{2} R$ compared to $h H_{1} R, h H_{3} R$ and $h H_{4} R$ was investigated by competition binding experiments using membrane preparations from Sf 9 insect cells co-expressing either the $\mathrm{hH}_{1} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein and RGS4 ( $\left[{ }^{3} \mathrm{H}\right]$ Mepyramine as radioligand) or the $\mathrm{hH}_{3 / 4} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ and $\mathrm{G}_{\beta 1 \gamma^{2}}$ proteins ( $\left[{ }^{3} \mathrm{H}\right]$ histamine or $\left[{ }^{3} \mathrm{H}\right]$ UR-PI294 as radioligand).

The monovalent $\mathrm{N}^{6}$-carbamoylated amino(methyl)thiazolyl propylguanidines 6.47-6.59 showed moderate to high $\mathrm{hH}_{2} \mathrm{R}$ affinity (Table 6.1). In general, aliphatic as well as aromatic residues were well tolerated, only ligand 6.50 which contains a phenyl residue showed a low affinity with a $\mathrm{p} K_{\mathrm{i}}$ value of 5.50. The highest $\mathrm{hH}_{2} \mathrm{R}$ affinities showed the compounds 6.55 ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.40 ) and 6.48 ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.54) which contain a branched cyclic or a linear aliphatic residue, respectively. In a similar manner as the bivalent $\mathrm{N}^{G}$-acylated amino(methyl)thiazolylpropylguanidines, bivalent $\mathrm{N}^{\mathrm{G}}$ carbamoylated amino(methyl)thiazolyl propylguanidines (e.g. UR-NK22) ${ }^{5}$ showed a higher affinity to the $\mathrm{hH}_{2} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ value: 8.02) compared to the monovalent ligands (6.47-6.59). In comparison, ligand 6.51 showed a considerably higher $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ value: 7.16 ) than the corresponding acylguanidine UR-Bit23 ( $\mathrm{p} K_{\mathrm{i}}$ value: 6.3). Incorporation of an aminothiazolylphenyl (compounds 6.60-6.65) or a 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl (compounds 6.66 and 6.67 ) moiety resulted in a decrease in $\mathrm{hH}_{2} \mathrm{R}$ affinity by one to two order(s) of magnitude. In comparison with the $D_{2 / 3} R$ ligand pramipexole ( $p K_{i}$ value: 4.86), which also contains a 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl head group, the ligands 6.66 and 6.67 showed increased affinity for the $\mathrm{hH}_{2} \mathrm{R}$ with $\mathrm{p} K_{\mathrm{i}}$ values of 5.95 and 6.29 . The replacement of the amino(methyl)thiazolyl propyl head group of the bivalent ligand UR-NK22 with either an aminothiazolylphenyl residue (6.70) or a pyrazolylpropyl residue (6.71) resulted in a strong decrease of $\mathrm{hH}_{2} \mathrm{R}$ affinity ( $\mathrm{p} K_{\mathrm{i}}$ value: 5.98 and 6.75).

The monovalent $\mathrm{N}^{6}$-carbamoylated amino(methyl)thiazolyl propylguanidines 6.47-6.59 showed a clear preference for the $\mathrm{hH}_{2} \mathrm{R}$ over the other subtypes. In case of the aminothiazolylphenylguanidines 6.60-6.65 and the bivalent ligands (6.70 and 6.71) the affinity for the $\mathrm{hH}_{4} \mathrm{R}$ was low ( $\mathrm{p} K_{\mathrm{i}}$ values 4-5) and, except for 6.63, a slight preference for the $h H_{2} \mathrm{R}$ was obtained. While 2 -amino-4,5,6,7-tetrahydrobenzothiazol-6-yl containing ligand 6.66 showed a similar affinity to $\mathrm{hH}_{1} \mathrm{R}, \mathrm{hH}_{2} \mathrm{R}, \mathrm{hH}_{3} \mathrm{R}$ with $\mathrm{p} K_{\mathrm{i}}$ values of 5.71-5.95 and a lower affinity to the $\mathrm{hH}_{4} \mathrm{R}$, ligand 6.67 showed a preference for $\mathrm{hH}_{2} \mathrm{R}$ over the other subtypes of around one order of magnitude.


Figure 6.7. Displacement of the respective radioligand from membrane preparations of Sf 9 insect cells (A) coexpressing the $\mathrm{hH}_{1} \mathrm{R}-\mathrm{G}_{\text {saS }}$ fusion protein and RGS4 (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ mepyramine, $\mathrm{c}=5 \mathrm{nM}, K_{\mathrm{d}}=4.5 \mathrm{nM}$ ), (B) expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ UR-DE257, $\mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ), (C) co-expressing the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ plus $\mathrm{G}_{\beta 1 \gamma 2}$ proteins (radioligand: $\left.{ }^{3} \mathrm{H}\right] \mathrm{UR}-\mathrm{PI} 294 \mathrm{c}=2 \mathrm{nM}, K_{d}=1.1 \mathrm{nM}$ ) or (D) co-expressing the $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\alpha i 2}$ plus $\mathrm{G}_{\beta 1 \gamma 2}$ proteins (radioligand: [ ${ }^{3} \mathrm{H}$ ]histamine $\mathrm{c}=10 \mathrm{nM}, K_{d}=15.9 \mathrm{nM}$ ) by exemplary monovalent carbamoylated guanidines. Data represent mean values $\pm$ SEM of 2-3 experiments performed in triplicate.

Table 6.1. Affinities of histamine, pramipexole, UR-NK22, UR-Bit23, the monovalent carbamoylated guanidines 6.476.67 and the bivalent carbamoylated guanidines 6.70-6.71 to $\mathrm{hH}_{1-4} \mathrm{R}$, obtained from equilibrium competition binding studies on membrane preparations from Sf 9 insect cells, expressing the respective histamine receptor subtype.

| Compound | $\mathrm{hH}_{1} \mathrm{R}^{\text {a }}$ |  | $\mathrm{hH}_{2} \mathrm{R}^{\text {b }}$ |  | $\mathrm{hH}_{3} \mathrm{R}^{\text {c }}$ |  | $\mathrm{hH}_{4} \mathrm{R}^{\text {e }}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{p} K_{i}$ | N | $\mathrm{p} K_{\text {i }}$ | N | $\mathrm{p} K_{\text {i }}$ | N | $\mathrm{p} K_{\text {i }}$ | N |
| Histamine | n.d. | - | $6.53 \pm 0.04$ | 3 | $7.8 \pm 0.1$ | 3 | $7.65 \pm 0.03$ | 3 |
| Pramipexole | n.d. | - | $4.86 \pm 0.07$ | 5 | n.d. | - | n.d. | - |
| UR-NK22 | $6.06 \pm 0.05^{5}$ | 2 | $8.07 \pm 0.05^{5}$ | 3 | $5.94 \pm 0.16^{5}$ | 4 | $5.69 \pm 0.07^{5}$ | 3 |
| UR-Bit23 | n.d. | - | $6.3 \pm 0.4$ | 2 | n.d. | - | n.d. | - |
| 6.47 | $4.54 \pm 0.02$ | 3 | $6.98 \pm 0.11$ | 3 | $4.35 \pm 0.01^{\text {d }}$ | 3 | $4.06 \pm 0.06$ | 2 |
| 6.48 | $5.11 \pm 0.03$ | 3 | $7.54 \pm 0.07$ | 4 | $5.25 \pm 0.02^{\text {d }}$ | 3 | $5.09 \pm 0.02$ | 2 |
| 6.49 | $4.61 \pm 0.09$ | 3 | $6.77 \pm 0.27$ | 3 | n.d. | - | $4.50 \pm 0.06$ | 2 |
| 6.50 | n.d. | - | $5.50 \pm 0.08$ | 3 | n.d. | - | $4.65 \pm 0.05$ | 2 |
| 6.51 | $5.21 \pm 0.02$ | 3 | $7.16 \pm 0.05$ | 3 | n.d. | - | $4.72 \pm 0.09$ | 2 |
| 6.52 | n.d. | - | $6.8 \pm 0.2$ | 4 | n.d. | - | $4.83 \pm 0.06$ | 2 |
| 6.53 | $5.8 \pm 0.1$ | 3 | $7.14 \pm 0.08$ | 2 | $5.49 \pm 0.01^{\text {d }}$ | 3 | $5.44 \pm 0.02^{f}$ | 3 |
| 6.54 | $5.87 \pm 0.09$ | 3 | $7.20 \pm 0.04$ | 3 | $5.04 \pm 0.04$ | 3 | $5.49 \pm 0.08^{f}$ | 3 |
| 6.55 | $5.63 \pm 0.06$ | 3 | $7.40 \pm 0.01$ | 2 | $5.00 \pm 0.08$ | 3 | $5.72 \pm 0.05^{f}$ | 3 |
| 6.56 | $5.31 \pm 0.02$ | 3 | $6.83 \pm 0.08$ | 3 | $5.10 \pm 0.02$ | 3 | $5.58 \pm 0.01$ | 3 |
| 6.57 | $5.23 \pm 0.03$ | 3 | $6.99 \pm 0.05$ | 3 | $4.93 \pm 0.03$ | 3 | $5.23 \pm 0.03^{f}$ | 3 |
| 6.58 | $5.10 \pm 0.08$ | 3 | $7.11 \pm 0.03$ | 3 | $4.78 \pm 0.03$ | 3 | $5.17 \pm 0.02^{f}$ | 3 |
| 6.59 | $5.42 \pm 0.07$ | 3 | $7.15 \pm 0.02$ | 4 | $5.13 \pm 0.01$ | 3 | $5.43 \pm 0.01^{\dagger}$ | 3 |
| 6.60 | n.d. | - | $5.28 \pm 0.06$ | 4 | n.d. | - | <4.0 | 2 |
| 6.61 | n.d. | - | $6.16 \pm 0.07$ | 4 | $4.96 \pm 0.07^{\text {d }}$ | 3 | $5.02 \pm 0.06$ | 2 |
| 6.62 | n.d. | - | $5.35 \pm 0.02$ | 3 | n.d. | - | $4.87 \pm 0.03$ | 2 |
| 6.63 | n.d. | - | $3.43 \pm 0.07$ | 3 | n.d. | - | $4.2 \pm 0.1$ | 2 |
| 6.64 | n.d. | - | $5.7 \pm 0.1$ | 3 | n.d. | - | $4.46 \pm 0.01$ | 2 |
| 6.65 | n.d. | - | $5.40 \pm 0.05$ | 3 | n.d. | - | $4.5 \pm 0.1$ | 2 |
| 6.66 | $5.71 \pm 0.02$ | 3 | $5.95 \pm 0.06$ | 4 | $5.82 \pm 0.03^{\text {d }}$ | 3 | $4.78 \pm 0.05$ | 2 |
| 6.67 | $5.92 \pm 0.02$ | 3 | $6.29 \pm 0.08$ | 3 | $5.48 \pm 0.06^{\text {d }}$ | 3 | $4.52 \pm 0.01$ | 2 |
| 6.70 | n.d. | - | $5.98 \pm 0.06$ | 3 | n.d. | - | $5.0 \pm 0.2$ | 2 |
| 6.71 | n.d. | - | $6.75 \pm 0.04$ | 3 | n.d. | - | $5.02 \pm 0.07$ | 3 |

Competition binding assay on membrane preparations of Sf9 insect cells ${ }^{\text {a }}$ co-expression of the $\mathrm{hH}_{1} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein and RGS4 (radioligand: [ ${ }^{3}$ H]mepyramine, $\mathrm{c}=5 \mathrm{nM}, K_{\mathrm{d}}=4.5 \mathrm{nM}$ ), ${ }^{\mathrm{b}} \mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein (radioligand: $\left.{ }^{3} \mathrm{H}\right] \mathrm{UR}-\mathrm{DE} 257, \mathrm{c}=20 \mathrm{nM}, K_{\mathrm{d}}=12.2 \mathrm{nM}$ ), ${ }^{c} \mathrm{co}-$ expression of the $\mathrm{hH}_{3} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ and $\mathrm{G}_{\beta 1 \gamma^{2}}$ proteins (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ histamine $\mathrm{C}=15 \mathrm{nM}, K_{d}=12.1 \mathrm{nM}$ or $\left.{ }^{d}{ }^{3} \mathrm{H}\right]$ UR-P1294 $\mathrm{c}=2 \mathrm{nM}, K_{d}=1.1 \mathrm{nM}$ ) or ${ }^{\mathrm{e}}$ co-expresson of the $\mathrm{hH}_{4} \mathrm{R}$ and $\mathrm{G}_{\text {ai2 }}$ plus $\mathrm{G}_{\beta 112}$ proteins (radioligand: [ $\left.{ }^{3} \mathrm{H}\right]$ histamine $\mathrm{c}=10 \mathrm{nM}, K_{d}=15.9 \mathrm{nM}$ or $\left.{ }_{[ }{ }^{3} \mathrm{H}\right]$ UR-PI294 $\mathrm{c}=5 \mathrm{nM}, K_{d}=5.1 \mathrm{nM}$ ). The incubation period was 60 min. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means $\pm$ SEM of N independent experiments, each performed in triplicate.

## $h_{2} \mathbf{R}$ agonism or antagonism in the GTPץS binding assay and $\beta$ Arrestin2 recruitment assay

The $N^{G}$-carbamoylated guanidines 6.47-6.71, histamine, UR-NK22, UR-AK421, UR-Bit22, UR-Bit23, UR-Bit24, UR-Bit29 and pramipexole were examined for $\mathrm{hH}_{2} \mathrm{R}$ agonism in the GTP $\gamma$ S binding assay on membrane preparations from Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. Ligands which exhibited no agonism were also investigated in the antagonistic mode versus histamine as agonist. All identified agonists and selected antagonists were additionally investigated for agonism in the $\beta$ Arrestin2 recruitment assay on HEK293T-hH2R- $\beta$ Arr2 cells, stably expressing the $h H_{2} R$-ElucC and $\beta$ Arr2-ElucNfusion constructs. ${ }^{15}$ Prior studies suggested only minor differences between $\beta$ Arrestin1 and $\beta$ Arrestin2 recruitment ${ }^{15}$ and therefore the $\beta$ Arrestin1 was not considered. Representative concentration response curves are depicted in Figure 6.8 (GTPүS binding assay) and Figure 6.9 ( $\beta$ Arrestin2 recruitment assay). The results from these experiments are summarized in Table 6.2.

The monovalent $\mathrm{N}^{\mathrm{G}}$-carbamoylated amino(methyl)thiazolylpropylguanidines 6.47-6.59 were partial to full agonists in the GTP $\gamma S$ binding assay and showed moderate to high $\mathrm{hH}_{2} \mathrm{R}$ potencies (up to 80 fold the potency of histamine; $\mathrm{pEC}_{50}$ values of 6.3-7.7) generally in good accordance to the acylguanidines (results from the GTPase assay) ${ }^{1,16}$ (Table 6.2). At high concentrations ( $\geq 10-$ $100 \mu \mathrm{M}$ ) the signal decreased again resulting in nearly bell-shaped concentration response curves (Figure 6.8). Compound 6.51 ( $\alpha$ : $0.91, \mathrm{pEC}_{50}$ : 7.5) showed the highest potency combined with full agonism, whereas the corresponding acylguanidine UR-Bit23 ( $\alpha: 0.68, \mathrm{pEC}_{50}$ : 6.59) was a partial agonist with only moderate potency.

Interestingly, the aminothiazolylphenyl containing monovalent ligands 6.60-6.65 and bivalent ligand 6.70 showed weak antagonism or inverse agonism at the $\mathrm{hH}_{2} \mathrm{R}$ with $\mathrm{p} K_{\mathrm{b}}$ values in good accordance to the $\mathrm{p} K_{\mathrm{i}}$ values.

Incorporation of the less flexible 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl (compounds 6.66 and 6.67) moiety resulted in partial agonism (6.67, $\alpha: 0.57, \mathrm{pEC}_{50}: 6.7$ ) or in weak partial agonism (6.66, $\alpha: 0.16, \mathrm{pEC}_{50}$ : 5.57) due to bell-shaped concentration response curves. In the antagonist mode of the GTP $\gamma S$ assay, 6.66 and 6.67 act as antagonists with $\mathrm{p} K_{\mathrm{b}}$ values of 5.57 and 4.33 . Also the structurally related dopamine receptor agonist pramipexole was a weak partial agonist at the $\mathrm{hH}_{2} \mathrm{R}\left(\alpha: 0.66, \mathrm{pEC}_{50}\right.$ : 5.07 ) with a sigmoidal curve. Although 6.67 was a racemic mixture, it was more potent than the enantiomerically pure pramipexole.

The pyrazole containing bivalent ligand 6.71 ( $\alpha$ : $0.38, \mathrm{pEC}_{50}$ : 7.0) was a partial agonist with low potency compared to the corresponding amino(methyl)thiazole containing full agonist UR-NK22 ( $\alpha: 0.92, \mathrm{pEC}_{50}: 8.03$ ).

All full or partial agonists identified in the GTP $\gamma S$ assay (6.47-6.49, 6.71) showed a lower potency and efficacy in the $\beta$ Arrestin2 recruitment assay. The agonistic $N^{6}$-carbamoylated guanidine-type ligands exhibited some functional bias towards G-protein activation. This is in agreement with the findings for acylguanidines and bivalent carbamoylguanidines. ${ }^{15}$ The bias was most pronounced in case of 6.53, which was a full agonist in the GTP $\gamma$ S assay ( $\alpha: 0.91$ ) and a weak partial agonist in the $\beta$ Arrestin2 recruitment assay ( $\alpha: 0.33$ ). Nearly similar bias was determined for $6.53,6.57$ and 6.58 which were partial agonists in the GTP $\gamma$ S assay ( $\alpha$ : 0.52-0.68) and very weak partial agonists in the $\beta$ Arrestin2 recruitment assay ( $\alpha: 0.10-0.16$ ). Also in the $\beta$ Arrestin2 recruitment assay of the
ligands 6.53-6.59 the signal decreased again at high concentrations ( $\geq 30-100 \mu \mathrm{M}$ ) resulting in nearly bell-shaped concentration response curves. Interestingly, the 2-amino-4,5,6,7tetrahydrobenzothiazole containing compounds 6.66 and 6.67 , which showed the most pronounced bell-shaped curves in the GTPyS assay, didn't show a decreased signal at high concentrations in the $\beta$ Arrestin2 recruitment assay. Ligand 6.66 and two selected antagonists ( 6.63 and 6.65 ) identified in the GTPyS exhibited no $\beta$ Arrestin2 recruitment, too.

Bell shaped (or u-shaped) concentration response curves were described at several GPCRs such as muscarinic receptors, ${ }^{17} \beta_{2}$-adrenergic receptors ${ }^{18}$ and serotonin receptors ${ }^{19}$. Some reasons for such a curve shape could be cytotoxicity, ${ }^{20}$ binding to multiple binding sites, ${ }^{21}$ multiple targets, receptor oligomers, ${ }^{22}$ agonist-induced desensitization ${ }^{18}$ or even due to physical properties like self-association of ligands into colloidal particals ${ }^{23}$.

In case of the GTPүS assay plausible explanations would be binding to multiple binding sites or direct interaction with the $G_{s a s}$ subunit of the fusion protein due to the use of membrane preparations from Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein instead of live cells. While in the GTP $\gamma$ S assay all agonists (6.47-6.59, 6.66 and 6.67 ) showed more or less pronounced bell-shape curves, in the $\beta$ Arrestin2 recruitment assay only the agonists 6.53-6.59 showed a similar behavior. Interestingly, the ligands $6.48,6.50,6.52,6.66$ and 6.67 with a distinguished bell-shaped curve in the GTP $\gamma S$ assay showed in the $\beta$ Arrestin2 recruitment assay a sigmoidal concentration response curve indicating that a different mechanism (e.g. cytotoxicity) is leading to the curve shape in the $\beta$ Arrestin2 recruitment assay. Moreover, it should be mentioned that all chiral compounds (6.53-6.59, 6.66 and 6.67 ) were racemic mixtures. Therefore, the bell shaped curve could also be due to opposite effects of the enantiomers. However, investigation of enantiomeric pure acylguanidine-type compounds in the GTPase assay revealed only low eudismic ratios (1.1-3.2), indicating the stereochemistry of the acyl residue plays only a minor role. ${ }^{4}$ However, according to the current state of knowledge this is all just speculation.


Figure 6.8. Concentration-response curves of representative monovalent carbamoylated guanidines on $h H_{2} R$ determined by $\left[{ }^{35} \mathrm{~S}\right] \mathrm{GTP} \gamma \mathrm{S}$ binding assay on membrane preparations of $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein. (A) Ligands 6.48, 6.51, 6.64, 6.66 and 6.67. (B) Ligands 6.47, 6.49 and 6.55 . (C) Ligands 6.50, 6.52-6.54. (D) Ligands 6.56-6.59. (E) and (F) Exemplary ligands measured in the antagonist mode; histamine (1 $\mu \mathrm{M}$ ) was used for stimulation. Data represent mean values $\pm$ SEM of 2-5 experiments performed in triplicate.


Figure 6.9. Concentration-response curves of representative monovalent carbamoylated guanidines on $\mathrm{hH}_{2} \mathrm{R}$ determined by a luciferase complementation assay measuring $\beta$-arrestin2 recruitment on HEK293T-hH2R- $\beta$ Arr2 cells. (A) and (B): Ligands with sigmoidal dose-response-curves. (C) and (D): Ligands with bell-shaped dose-response-curves. Data represent mean values $\pm$ SEM of 3 experiments performed in duplicate.

Table 6.2. $\mathrm{hH}_{2} \mathrm{R}$ agonism or antagonism and the calculated $\mathrm{p} E C_{50}$ or $\mathrm{p} K_{\mathrm{b}}$ values of histamine, pramipexole, UR-NK22, UR-AK421, UR-Bit22, UR-Bit23, UR-Bit24, UR-Bit29, the monovalent carbamoylated guanidines 6.47-6.67 and the bivalent carbamoylated guanidines 6.70-6.71 determined by a GTP $\gamma S$ assay and $\beta$ Arrestin2 recruitment assay.

| Compound | $\mathbf{h H}_{2} \mathrm{R}$ (GTPYS) ${ }^{\mathbf{a}}$ |  |  | $\mathbf{h H}_{2} \mathrm{R}$ ( $\beta$ Arrestin2 Recruitment) ${ }^{\text {b }}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{pEC} C_{50}\left(\mathrm{p} K_{\mathrm{b}}\right)$ | N | $\alpha$ | $\mathrm{p} E C_{50}$ | N | $\alpha$ |
| Histamine | $5.80 \pm 0.06$ | 9 | 1.0 | $5.38 \pm 0.04$ | 19 | 1.0 |
| Pramipexole | $5.07 \pm 0.06$ | 5 | $0.66 \pm 0.08$ | $4.4 \pm 0.1$ | 4 | $0.35 \pm 0.03$ |
| UR-NK22 | $8.03 \pm 0.02^{5}$ | 2 | $0.92 \pm 0.01^{5}$ | $7.19{ }^{15}$ | - | $0.30^{15}$ |
| UR-AK421 | $7.61{ }^{\text {c1 }}$ | - | $0.42^{\text {c } 1}$ | n.d. | - | n.d. |
| UR-Bit22 | $5.83{ }^{\text {c } 16}$ | - | $0.56{ }^{\text {c } 16}$ | n.d. | - | n.d. |
| UR-Bit23 | $\begin{aligned} & 6.59 \pm 0.06 / \\ & 7.02^{c 1} \end{aligned}$ | 3 - | $\begin{aligned} & 0.68 \pm 0.13 / \\ & 0.68^{c 1} \end{aligned}$ | n.d. | - | n.d. |
| UR-Bit24 | $7.65{ }^{\text {c } 1}$ | - | $0.79^{\text {c }}$ | $7.72{ }^{15}$ | - | $0.14{ }^{15}$ |
| UR-Bit29 | $7.30{ }^{\text {c } 1}$ | - | $0.71{ }^{\text {c } 1}$ | n.d. | - | n.d. |
| 6.47 | $6.75 \pm 0.06$ | 3 | $0.79 \pm 0.09$ | $6.77 \pm 0.04$ | 3 | $0.32 \pm 0.01$ |
| 6.48 | $7.41 \pm 0.07$ | 3 | $0.60 \pm 0.09$ | $7.07 \pm 0.02$ | 3 | $0.28 \pm 0.03$ |
| 6.49 | $6.8 \pm 0.2$ | 3 | $0.96 \pm 0.07$ | $6.41 \pm 0.03$ | 3 | $0.43 \pm 0.06$ |
| 6.50 | $6.3 \pm 0.2$ | 2 | $0.59 \pm 0.02$ | $5.47 \pm 0.06$ | 3 | $0.25 \pm 0.02$ |
| 6.51 | $7.5 \pm 0.1$ | 4 | $0.91 \pm 0.07$ | $7.00 \pm 0.08$ | 3 | $0.33 \pm 0.03$ |
| 6.52 | $7.4 \pm 0.2$ | 3 | $0.72 \pm 0.06$ | $6.78 \pm 0.03$ | 3 | $0.29 \pm 0.02$ |
| 6.53 | $7.2 \pm 0.1$ | 5 | $0.68 \pm 0.07$ | $6.9 \pm 0.1$ | 3 | $0.10 \pm 0.01$ |
| 6.54 | $7.53 \pm 0.06$ | 3 | $0.69 \pm 0.02$ | $6.45 \pm 0.06$ | 3 | $0.20 \pm 0.02$ |
| 6.55 | $7.66 \pm 0.08$ | 3 | $0.65 \pm 0.03$ | $6.65 \pm 0.09$ | 3 | $0.40 \pm 0.06$ |
| 6.56 | $7.46 \pm 0.03$ | 3 | $0.59 \pm 0.02$ | $6.55 \pm 0.02$ | 3 | $0.25 \pm 0.03$ |
| 6.57 | $7.51 \pm 0.06$ | 3 | $0.52 \pm 0.03$ | $6.8 \pm 0.1$ | 3 | $0.11 \pm 0.01$ |
| 6.58 | $7.5 \pm 0.1$ | 3 | $0.60 \pm 0.04$ | $6.57 \pm 0.08$ | 3 | $0.16 \pm 0.02$ |
| 6.59 | $7.76 \pm 0.04$ | 3 | $0.60 \pm 0.07$ | $6.52 \pm 0.07$ | 3 | $0.22 \pm 0.01$ |
| 6.60 | $(4.91 \pm 0.09)$ | 3 | -0.037 | n.d. | - | n.d. |
| 6.61 | $(6.14 \pm 0.03)$ | 3 | -0.54 | n.d. | - | n.d. |
| 6.62 | $(5.40 \pm 0.02)$ | 3 | -0.47 | n.d. | - | n.d. |
| 6.63 | $(4.8 \pm 0.1)$ | 3 | $-0.20 \pm 0.02$ | - | - | $0.018 \pm 0.009$ |
| 6.64 | $(5.44 \pm 0.02)$ | 3 | -0.62 | n.d. | - | n.d. |
| 6.65 | $(5.52 \pm 0.05)$ | 3 | $-0.28 \pm 0.04$ | - | - | $0.043 \pm 0.009$ |
| 6.66 | $\begin{aligned} & 6.83 \pm 0.06 / \\ & (5.57 \pm 0.04) \end{aligned}$ | 3 3 | $0.16 \pm 0.07$ | - | - | $0.045 \pm 0.005$ |

Table 6.2 continued.

| Compound | $\mathrm{hH}_{2} \mathrm{R}(\mathrm{GTP} \boldsymbol{\gamma} \mathrm{S})^{\mathbf{a}}$ |  |  | $\mathrm{hH}_{2} \mathrm{R}$ ( $\beta$ Arrestin2 Recruitment) ${ }^{\text {b }}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{pEC} \mathrm{C}_{50}\left(\mathrm{p} K_{\mathrm{b}}\right)$ | N | $\alpha$ | $\mathrm{p} E C_{50}\left(\mathrm{p} K_{\mathrm{b}}\right)$ | N | $\alpha$ |
|  | $6.7 \pm 0.3 /$ | 3 |  |  |  |  |
| 6.67 | $(4.33 \pm 0.09)$ | 3 | $0.53 \pm 0.05$ | $5.36 \pm 0.05$ | 3 | $0.31 \pm 0.03$ |
| 6.70 | $(5.98 \pm 0.06)$ | 3 | -0.46 | n.d. | - | n.d. |
| 6.71 | $7.0 \pm 0.8$ | 2 | $0.38 \pm 0.03$ | $5.75 \pm 0.07$ | 3 | $0.26 \pm 0.08$ |

$\left.{ }^{\mathrm{a}}{ }^{35} \mathrm{~S}\right]$ GTP $\gamma \mathrm{S}$ assay determined on membrane preparations of $\mathrm{Sf9}$ insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{s \alpha S}$ fusion protein.
${ }^{\mathrm{b}} \beta$ Arrestin2 recruitment assay determined on HEK293T-h $\mathrm{H}_{2} \mathrm{R}-\beta$ Arr2 cells, stably expressing the $\mathrm{hH}_{2} \mathrm{R}$-ElucC and $\beta$ Arr2ElucNfusion constructs. The incubation period was 60 min . The intrinsic activity ( $\alpha$ ) of histamine was set to 1.00 , and $\alpha$ values of investigated compounds were referred to this value. The $\mathrm{p} K_{\mathrm{B}}$ values of neutral antagonists were determined in the antagonist mode versus histamine ( $c=1 \mu \mathrm{M}$ ) as agonist. Data represent mean values $\pm$ SEM of $N$ independent experiments performed in triplicate (GTP $\gamma S$ assay) or duplicate ( $\beta$ Arrestin2 recruitment assay). ${ }^{\text {c }}$ Steady-state GTPase assay determined on membrane preparations of Sf 9 insect cells expressing the $\mathrm{hH}_{2} \mathrm{R}-\mathrm{G}_{\text {sas }}$ fusion protein.

## $\mathrm{hD}_{\text {2long }} \mathrm{R}$ and $\mathrm{hD}_{3} \mathrm{R}$ affinities of carbamoylated guanidines

Aminothiazole containing ligands such as pramipexole and its derivatives are described as high affintiy dopamine receptor ligands (preferred $D_{2} R, D_{3} R$ and $D_{4} R$ ). ${ }^{24}$ Therefore, haloperidol, selected $N^{6}$-carbamoylated guanidines containing an aminothiazole moiety (6.53, 6.55, 6.58, $6.59,6.61,6.66,6.67$ ) and the bivalent ligand 6.71 were investigated in equilibrium competition binding experiments on homogenates of HEK293T-CRE-Luc-hD 2long R and/or HEK293T-CRE-Luc$\mathrm{hD}_{3} \mathrm{R}$ cells using $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}$-methylspiperone as radioligand. The results are summarized in Table 6.3.

For the standard agonist pramipexole biphasic displacement curves were reported at $\mathrm{hD}_{2 \text { long }} \mathrm{R}$ and $\mathrm{hD}_{3} \mathrm{R}$ with $\mathrm{p} K_{\mathrm{i}}$ values for high ( $\mathrm{p} K_{\mathrm{H}}$ value) and low affinity binding ( $\mathrm{p} K_{\mathrm{L}}$ value). ${ }^{25}$ The standard antagonist haloperidol and the investigated ligands 6.53, 6.55, 6.58, 6.59, 6.61, 6.66, 6.67 and 6.71 showed monophasic displacement curves. The $\mathrm{p} K_{\mathrm{i}}$ values of haloperidol at the $\mathrm{hD}_{2 \text { long }} \mathrm{R}$ and $\mathrm{hD}_{3} \mathrm{R}$ were in good accordance with literature results. ${ }^{26}$

The amino(methyl)thiazolylpropyl (6.53, 6.55, 6.58 and 6.59 ), aminothiazolylphenyl (6.61) and 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl ( 6.66 and 6.67) containing ligand(s) showed a weak to moderate affinity for the $\mathrm{hD}_{2 \text { long }} \mathrm{R}\left(\mathrm{p} K_{\mathrm{i}}\right.$ value 5.6-6.6). The bivalent ligand (6.71) showed a moderate affinity with a $\mathrm{p} K_{\mathrm{i}}$ value of 6.97. The amino(methyl)thiazolylpropyl containing ligands $6.53,6.55,6.58$ and 6.59 clearly preferred the $\mathrm{hH}_{2} \mathrm{R}$ over the $\mathrm{hD}_{2 \text { long }} \mathrm{R}$. The 2 -amino-4,5,6,7-tetrahydrobenzothiazol-6-yl containing ligands ( $\mathbf{6 . 6 6}$ and 6.67) and the bivalent ligand 6.71 showed a comparable affinity to the $\mathrm{hH}_{2} \mathrm{R}$ and the $\mathrm{hD}_{2 \text { long }} \mathrm{R}$.

The ligands $6.53,6.55,6.58,6.59$ and 6.71 showed a moderate affinity for the $h D_{3} R$ with $p K_{i}$ values between 6.9 and 7.6. The ligands $\mathbf{6 . 6 1}, 6.66$ and $\mathbf{6 . 6 7}$ showed a low affinity for the $h D_{3} R$ with $\mathrm{p} K_{\mathrm{i}}$ values between 5.3 and 5.9. The $\mathrm{N}^{6}$-carbamoylated amino(methyl)thiazolylpropylguanidines $6.53,6.55,6.58$ and 6.66 bound non-selectively to both the $h H_{2} R$ and $h D_{3} R$ receptors. The 2 -amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety containing ligand 6.67 showed a preference towards $h H_{2} R$ and $h D_{2 l o n g} R$ over the $h D_{3} R$. In contrast, the bivalent ligand 6.71 showed a preference towards $\mathrm{hD}_{3} \mathrm{R}$ over the $\mathrm{hD}_{2 \text { long }} \mathrm{R}$ and $\mathrm{hH}_{2} \mathrm{R}$.


Figure 6.10. Displacement of $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}$-methylspiperone from homogenates of (A) HEK293T-CRE-Luc-hD ${ }_{2 l o n g} R$ cells ( $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}$-Methylspiperone: $K_{\mathrm{d}}=0.014 \mathrm{nM}, \mathrm{c}=0.05 \mathrm{nM}$ ) or (B) HEK293T-CRE-Luc-hD ${ }_{3} \mathrm{R}$ cells ( $\left.{ }^{3} \mathrm{H}\right] \mathrm{N}$-methylspiperone: $K_{\mathrm{d}}=$ $0.026 \mathrm{nM}, \mathrm{c}=0.05 \mathrm{nM}$ ) by selected monomeric carbamoylated guanidines. Data represent mean values $\pm$ SEM of 3 experiments performed in triplicate.

Table 6.3. Affinities of haloperidole, pramipexole and the monovalent carbamoylated guanidines 6.53, 6.55, 6.58, 6.59,
6.61, $6.66,6.67$ and 6.71 to the dopamine receptors $h D_{2 l o n g} R$ and $h D_{3} R$, obtained from equilibrium competition binding studies.

| Compound | $h D_{\text {2long }} \mathrm{R}^{\text {a }}$ |  | $\mathrm{hD}_{3} \mathrm{R}^{\mathrm{b}}$ |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{p} K_{i}$ | N | $\mathrm{p} K_{i}$ | N |
| Haloperidol | $9.60{ }^{\text {c } 26 /}$ | - | $8.64{ }^{\text {c } 26} /$ | - |
|  | $9.3 \pm 0.2$ | 3 | $8.7 \pm 0.1$ | 3 |
| Pramipexole | $\left(\mathrm{p} K_{\mathrm{H}}: 7.40 / \mathrm{p} K_{\mathrm{L}}: 5.44\right)^{\mathrm{d} 25}$ | - | $\left(\mathrm{p} K_{\mathrm{H}}: 9.06 / \mathrm{p} K_{\mathrm{L}}: 7.36\right)^{\text {d } 25}$ | - |
| 6.53 | $6.3 \pm 0.1$ | 3 | $7.07 \pm 0.06$ | 3 |
| 6.55 | $6.58 \pm 0.03$ | 3 | $7.36 \pm 0.04$ | 3 |
| 6.58 | $6.28 \pm 0.08$ | 3 | $7.19 \pm 0.06$ | 3 |
| 6.59 | $6.32 \pm 0.08$ | 3 | $6.88 \pm 0.04$ | 3 |
| 6.61 | $5.6 \pm 0.2$ | 3 | $5.89 \pm 0.05$ | 3 |
| 6.66 | $5.9 \pm 0.1$ | 3 | $5.9 \pm 0.2$ | 3 |
| 6.67 | $6.26 \pm$ | 1 | $5.3 \pm 0.1$ | 3 |
| 6.71 | $6.97 \pm 0.06$ | 3 | $7.6 \pm 0.2$ | 3 |

Determined by displacing [ $\left.{ }^{3} \mathrm{H}\right] \mathrm{N}$-methylspiperone ( ${ }^{\mathrm{a}} \mathrm{hD}$ 2long $\mathrm{R}: K_{\mathrm{d}}=0.0149 \mathrm{nM}, \mathrm{c}=0.05 \mathrm{nM}$ or ${ }^{\mathrm{b}} \mathrm{hD}{ }_{3} \mathrm{R}: K_{\mathrm{d}}=0.0258 \mathrm{nM}$, $\mathrm{c}=$ 0.05 nM ) by increasing concentrations of the respective ligand at homogenates of ${ }^{\mathrm{a}}$ HEK293T-CRE-Luc-hD $\mathrm{Dlong} R$ or ${ }^{b}$ HEK293T-CRE-Luc-hD $D_{3} R$ cells. ${ }^{c}$ determined on CHO cells stably expressing the $h D_{2 \text { short }} R$ or $h D_{3} R$. ${ }^{d}$ High/low affinity binding determined on CHO cells stably expressing the $h D_{2 \text { short }} R$ or $h D_{3} R$. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means $\pm$ SEM of $N$ independent experiments, each performed in triplicate.

### 6.3 Experimental Section

### 6.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Fluka (Buchs, Swiss), Alfa Aesar (Karlsruhe, Germany), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 ( $0.040-0.063 \mathrm{~mm}$ ). Automated flash chromatography was performed with a 971-FP flash-purification system (Agilent Technologies, Santa Clara, CA). Pre-packed columns (SuperFlash SF10-4 g, SF12-8 g, SF 15-12 g und SF15-24 g, Agilent Technologies, Santa Clara, CA) were used throughout. Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution ( 0.8 g ninhydrin, 200 mL n -buthanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA) or on a FTS 3000 MX spectrometer (Excalibur Series, Bio-Rad, Hercules, CA) equipped with an ATR unit (Specac Golden Gate Diamond Single Reflection ATR system). Nuclear Magnetic Resonance ( ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR) spectra were recorded on a Bruker Avance- 300 ( $7.05 \mathrm{~T},{ }^{1} \mathrm{H}: 300 \mathrm{MHz},{ }^{13} \mathrm{C}: 75.5 \mathrm{MHz}$ ), Avance- 400 ( $9.40 \mathrm{~T},{ }^{1} \mathrm{H}: 400 \mathrm{MHz},{ }^{13} \mathrm{C}: 100.6 \mathrm{MHz}$ ), or Avance-600 ( $14.1 \mathrm{~T} ;{ }^{1} \mathrm{H}: 600 \mathrm{MHz},{ }^{13} \mathrm{C}: 150.9 \mathrm{MHz}$; cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in $\delta(\mathrm{ppm})$ relative to external standards. Multiplicities are specified with the following abbreviations: $s$ (singlet), $d$ (doublet), $t$ (triplet), $q$ (quartet), qui (quintet), $m$ (multiplet), br $s$ (broad signal), as well as combinations thereof. In certain cases 2D-NMR techniques (COSY, HSQC, HMBC and NOESY) were used to assign ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ chemical shifts. High-resolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass Q-TOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K2001 detector. A Nucleodur 100-5 C18 ( $250 \times 21 \mathrm{~mm}, 5 \mu \mathrm{~m}$, Macherey-Nagel, Dueren, Germany), a Kinetex XB-C18 100A ( $250 \times 21.2 \mathrm{~mm}, 5 \mu \mathrm{~m}$, Phenomenex, Aschaffenburg, Germany) and a Interchim Puriflash PF15 C18 HQ ( $120 \mathrm{~g}, 15 \mu \mathrm{~m}$, Interchim S. A., Montluçon, France) served as RPcolumns at a flow rate of either $15 \mathrm{~mL} / \mathrm{min}$ (Kinetex and Phenomenex column) or $30 \mathrm{~mL} / \mathrm{min}$ (Interchim column) at room temperature. A detection wavelength of 220 nm and mixtures of $\mathrm{CH}_{3} \mathrm{CN}$ and $0.05-0.1 \%$ aq. TFA as mobile phase were used throughout. $\mathrm{CH}_{3} \mathrm{CN}$ was removed from the eluates under reduced pressure (final pressure: 80 mbar ) at $45{ }^{\circ} \mathrm{C}$ prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a vacuubrand RZ 6 rotary vane vacuum pump). Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A ( $250 \times 4.6 \mathrm{~mm}, 5 \mu \mathrm{~m}, t_{0}=2.9 \mathrm{~min}$, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of $0.05 \%$ TFA in $\mathrm{CH}_{3} \mathrm{CN}$ (A) and $0.05 \%$ aq. TFA (B) were used as mobile phase. Helium degassing, room temperature, a flow rate of $0.8 \mathrm{~mL} / \mathrm{min}$ and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration:
$100-500 \mu \mathrm{M}$ ) were either prepared from stock solution ( 10 mM in 20 mM aqueous HCl solution or 10 mM in a mixture of DMSO / in 20 mM aqueous HCl solution 1:1) in a mixture of $A$ and $B$ corresponding to the initial eluent composition of the run, or as a one to one dilution of the eluate (preparative HPLC) with Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 3: 0-30 min: A/B 15:85-90:10, 30-35 min: 90:10. Microanalysis was performed on a Vario micro cube (Elementar, Langenselbold, Germany).

### 6.3.2 Experimental protocols and analytical data

The branched carboxylic acids (3-methyl 4-(4-methylphenyl)butanoic acid, 3-methyl 5-phenyl pentanoic acid, 3-cyclohexyl butanoic acid, 3-(4-prop-2-yl phenyl) butanoic acid, 4-methyl 3phenyl pentanoic acid, 3-phenyl pentanoic acid, 4-cyclohexyl 3-methyl butanoic acid) were synthesized by Anja Kraus ${ }^{12}$. The amine precursor 3-(1-trityl-1H-pyrazol-4-yl)prop-1-yl-amine was synthesized according to published protocols ${ }^{13}$.

## 2-(5-Oxohexyl)isoindoline-1,3-dione (6.1) ${ }^{1,16}$

A solution of phthalimide ( $1.29 \mathrm{~g}, 8.74 \mathrm{mmol}, 1 \mathrm{eq}$ ), 6-chlorohexan-2-one ( $2 \mathrm{~g}, 14.86 \mathrm{mmol}, 1.7$ eq) and $\mathrm{K}_{2} \mathrm{CO}_{3}(1.39 \mathrm{~g}, 10.05 \mathrm{mmol}, 1.15 \mathrm{eq})$ in DMF ( 17 mL ) were stirred at $80{ }^{\circ} \mathrm{C}$ for 24 h to obtain a white suspension. Ice cold water ( 60 mL ) was added and the product was extracted with $\mathrm{CHCl}_{3}(3 \times 50 \mathrm{~mL})$. The organic layers were combined and washed with brine ( 50 mL ). The solvent was removed under reduced pressure. The crude product was dissolved in $\mathrm{CHCl}_{3}(100 \mathrm{~mL})$ and washed with water $(2 \times 50 \mathrm{~mL})$ and brine $(50 \mathrm{~mL})$. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed under reduced pressure and the residue was purified by column chromatography (eluent: PE/EtOAc 90:10-70:30). Removal of the solvent in vacuo afforded the product as a white solid ( $1.5 \mathrm{~g}, 70 \%$ ). Mp: $68.5^{\circ} \mathrm{C}$ (Lit. ${ }^{16} \mathrm{mp}: 73-75^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.3$ (PE/EtOAc $7: 3$ ). ${ }^{1} \mathrm{H}-$ NMR (400 MHz, CDCl ${ }_{3}$ ) : $\delta(\mathrm{ppm})$ 1.51-1.67 (m, 4H), $2.07(\mathrm{~s}, 3 \mathrm{H}), 2.43(\mathrm{t}, 2 \mathrm{H}, J 7.15 \mathrm{~Hz}), 3.63(\mathrm{t}, 2 \mathrm{H}$, $J 6.94 \mathrm{~Hz}), 7.63-7.67(\mathrm{~m}, 2 \mathrm{H}), 7.75-7.80(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 20.8,27.9$, 29.9, 37.5, 42.9, 123.2, 132.1, 133.9, 168.4, 208.3. HRMS (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{NO}_{3}{ }^{+}$: 246.1125, found: 246.1128. $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{3}$ (245.28).

## 2-[3-(2-Amino-4-methylthiazol-5-yl)propyl]-1,3-dihydro-2H-isoindol-1,3-dione hydrobromide $(6.3)^{1,16}$

6.1 ( $5.00 \mathrm{~g}, 20.38 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in dioxane ( 100 mL ). A solution of bromine ( 3.91 g , $24.46 \mathrm{mmol}, 1.2 \mathrm{eq})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{~mL})$ was added drop wise and the reaction mixture was stirred for 1 h at room temperature. Removal of the solvent in vacuo afforded 6.2 as a brown oil. The crude intermediate was dissolved in DMF ( 100 mL ) and thiourea ( $1.55 \mathrm{~g}, 20.38 \mathrm{mmol}, 1 \mathrm{eq}$ ) was added. The reaction mixture was stirred at $100{ }^{\circ} \mathrm{C}$ for 3 h and then over night at room temperature. The solvent was removed under reduced pressure and the residue was suspended
in EtOAc ( 100 mL ). Subsequently, the precipitate was filtered off and washed with EtOAc (100 ml ). Removal of residual solvent in vacuo afforded 6.3 as a beige solid ( $6.71 \mathrm{~g}, 86 \%$ ). Mp: $211.0^{\circ} \mathrm{C}$ (Lit. $\left.{ }^{16} \mathrm{mp}: 242{ }^{\circ} \mathrm{C}\right) . R_{\mathrm{f}}=0.30\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right) \delta$ (ppm): 1.78-1.91 (m, 2H), 2.08-2.14 (m, 3H), 2.61-2.73 (m, 2H), 3.59-3.63 (m, 2H), 7.81-7.88 (m, $4 \mathrm{H})$. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}^{+}$: 302.0958, found: 302.0963. $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}$. $\mathrm{HBr}(301.09+80.91)$.
tert-Butyl 4-methyl-5-[3-(1,3-dioxo-1,3-dihydro-2H-isoindol-2-yl)propyl]thiazol-2-ylcarbamate $(6.4)^{1,16}$
6.3 ( $3.58 \mathrm{~g}, 9.40 \mathrm{mmol}, 1 \mathrm{eq}$ ), di-tert-butyldicarbonate ( $2.46 \mathrm{~g}, 11.28 \mathrm{mmol}, 1.2 \mathrm{eq}$ ), triethylamine $(1.43 \mathrm{~g}, 14.09 \mathrm{mmol}, 1.5 \mathrm{eq})$ and 4-(dimethylamino)-pyridine ( $105 \mathrm{mg}, 0.94 \mathrm{mmol}, 0.1 \mathrm{eq}$ ) were dissolved in $\mathrm{CHCl}_{3}(40 \mathrm{ml})$. The mixture was stirred overnight at room temperature. The precipitate was filtered off. The product containing supernatant was washed with hydrochloric acid $(0.25 \mathrm{M}, 2 \times 100 \mathrm{~mL})$, brine ( 100 mL ) and $\mathrm{H}_{2} \mathrm{O}(100 \mathrm{~mL})$. The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed under reduced pressure. The residue was purified by automated flash chromatography (PE/EA 100:0-55:45 in 45 min ). Removal of the solvent under reduced pressure afforded the product as yellow foam ( $750 \mathrm{mg}, 20 \%$ ). Mp: $81.5^{\circ} \mathrm{C}$ (Lit. ${ }^{16} \mathrm{mp}: 70-$ $72{ }^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.3$ (PE/EA 60:40). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.51(\mathrm{~s}, 9 \mathrm{H}), 1.95-2.02(\mathrm{~m}, 2 \mathrm{H})$, $2.22(\mathrm{~s}, 3 \mathrm{H}), 2.71(\mathrm{t}, 2 \mathrm{H}, J 7.80 \mathrm{~Hz}), 3.74(\mathrm{t}, 2 \mathrm{H}, J 7.15 \mathrm{~Hz}), 7.69-7.72(\mathrm{~m}, 2 \mathrm{H}), 7.83-7.85(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.4,23.7,28.3,30.2,37.5,82.4,123.1,123.3,132.1,134.0$, 141.5, 152.5, 157.6, 168.3. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{~S}^{+}$: 402.1482, found: 402.1489. $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{~S}$ (401.48).

## tert-Butyl 5-(3-aminopropyl)-4-methylthiazol-2-ylcarbamate (6.5) ${ }^{1,16}$

To a suspension of 6.4 ( $710 \mathrm{mg}, 1.77 \mathrm{mmol}, 1 \mathrm{eq}$ ) in $\mathrm{EtOH}(7 \mathrm{ml})$ hydrazine-monohydrate ( 443 $\mathrm{mg}, 8.84 \mathrm{mmol}, 5 \mathrm{eq})$ was added. The reaction mixture was stirred over night at room temperature. The precipitated phthalhydrazide (by-product) was filtered off. The product which was dissolved in the supernatant was purified by automated flash chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75\right.$ $\mathrm{N} \mathrm{NH}_{3}$ in MeOH 100:0-90:10 in 35 min ). Removal of the solvent in vacuo afforded the product as yellow oil ( $400 \mathrm{~g}, 83 \%$ ). $R_{\mathrm{f}}=0.56\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 80: 20\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, [ $\mathrm{D}_{6}$ ]DMSO): $\delta(\mathrm{ppm}) 1.46(\mathrm{~s}, 9 \mathrm{H}), 1.54-1.61(\mathrm{~m}, 2 \mathrm{H}), 2.11(\mathrm{~s}, 3 \mathrm{H}), 2.55(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 6.83 \mathrm{~Hz}), 2.63(\mathrm{t}, 2 \mathrm{H}$, $J 7.44 \mathrm{~Hz}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.5,23.3,28.3,34.9,41.2,82.1,123.8,141.6$, 152.8, 157.5. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}^{+}$: 272.1427, found: 272.1433 . $\mathrm{C}_{12} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}$ (271.38).

## 2-Amino-4-(3-phthalimidophenyl)thiazole hydrobromide (6.6)

To a solution of $3.4(2 \mathrm{~g}, 7.54 \mathrm{mmol}, 1 \mathrm{eq})$ in $\mathrm{CHCl}_{3}(20 \mathrm{~mL})$ was added HBr solution in acetic acid ( $45 \% \mathrm{w} / \mathrm{v}, 1 \mathrm{~mL}$ ) under stirring. Bromine ( $1.2 \mathrm{~g}, 7.54 \mathrm{mmol}, 1 \mathrm{eq}$ ) in $\mathrm{CHCl}_{3}(10 \mathrm{~mL})$ was added drop wise. The reaction mixture was stirred for 1 h at room temperature. Removal of the solvent in vacuo afforded the 3.5 as a white solid, which was applied to the next step without further purification. The crude 3.5 was dissolved in hot $\mathrm{CH}_{3} \mathrm{CN}(30 \mathrm{~mL})$ and poured in a hot solution of
thiourea ( $574 \mathrm{mg}, 7.54 \mathrm{mmol}, 1 \mathrm{eq}$ ) in $\mathrm{EtOH}(30 \mathrm{~mL}$ ). The reaction mixture was stirred under reflux for 3 h . Removal of the solvent in vacuo afforded a beige solid, which was recristallized in EtOAc ( 100 mL ) and filtered through a Buchner funnel. 6.6 was afforded as a white solid ( 2.35 g , $78 \%$ ). Mp: 269-279 ${ }^{\circ} \mathrm{C}$, decomposition. $R_{f}=0.7\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(400$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 7.26(\mathrm{~s}, 1 \mathrm{H}), 7.51-7.54(\mathrm{~m}, 1 \mathrm{H}), 7.66(\mathrm{t}, 1 \mathrm{H}, \mathrm{J} 7.90$ $\mathrm{Hz}), 7.81-7.85(\mathrm{~m}, 2 \mathrm{H}), 7.93-7.96(\mathrm{~m}, 2 \mathrm{H}), 7.98-8.01(\mathrm{~m}, 2 \mathrm{H}), 8.94(\mathrm{br} \mathrm{s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$, [D6]DMSO): $\delta(\mathrm{ppm}) 104.3,124.0,125.3,125.9,128.7,130.1,130.6,131.9,133.0,135.4,139.7$, 167.3, 170.6. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{12} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}^{+}$: 322.0645, found: 322.0650. $\mathrm{C}_{17} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S} \cdot \mathrm{HBr}(321.35+80.91)$.

## 2-Amino-4-(3-aminophenyl)thiazole (6.7) ${ }^{27}$

6.6 ( $1.69 \mathrm{~g}, 4.20 \mathrm{mmol}, 1 \mathrm{eq}$ ) was suspended in a mixture of concentrated hydrochloric acid ( 30 $\mathrm{mL})$ and acetic acid ( 30 mL ) and the reaction mixture was stirred under reflux overnight. The solvent was removed under reduced pressure and the residue was suspended in aqueous NaOH solution ( $0.03 \mathrm{M}, 30 \mathrm{~mL}$ ). The resulting precipitate (by-product: phthalic acid) was filtered through a Buchner funnel and washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$. Aqueous layers were combined and part of the solvent was removed under reduced pressure. Aqueous $\mathrm{NH}_{3}$ solution ( $25 \%, 5 \mathrm{~mL}$ ) was added and the resulting precipitate was filtered off. The solid was washed with $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$ and the residual solvent was removed under reduced pressure. 6.7 was afforded as a yellow solid ( $560 \mathrm{mg}, 60 \%$ ). Mp: $177-178{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.6\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(300 \mathrm{MHz}$, [ $\mathrm{D}_{6}$ ]DMSO): $\delta(\mathrm{ppm}) 5.05(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 6.43-6.47(\mathrm{~m}, 1 \mathrm{H}), 6.77(\mathrm{~s}, 1 \mathrm{H}), 6.91-7.03(\mathrm{~m}, 5 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ (100 MHz, [D $D_{6}$ DMSO): $\delta(\mathrm{ppm}) 101.1,111.9,113.5,114.0,129.3,136.0,149.1,151.2,168.3$. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{~N}_{3} \mathrm{~S}^{+}: 192.0590$, found: 192.0590. $\mathrm{C}_{9} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{~S}$ (191.25).

## 2-((1r, 4r)-4-hydroxycyclohexyl)isoindoline-1,3-dione (6.8) ${ }^{28}$

$\mathrm{K}_{2} \mathrm{CO}_{3}(10.50 \mathrm{~g}, 75.97 \mathrm{mmol}, 1.75 \mathrm{eq})$ was added to a solution of trans-4-aminocyclohexanol ( 5.00 $\mathrm{g}, 43.41 \mathrm{mmol}, 1 \mathrm{eq})$ in $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$. Under stirring N -(ethoxycarbonyl)phthalimide ( 10.24 g , $49.92 \mathrm{mmol}, 1.15 \mathrm{eq}$ ) was added and the reaction mixture was stirred for 30 min at room temperature. The resulting precipitate was filtered off and washed with $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$. Removal of the residual solvent in vacuo afforded the product as a beige solid ( $4.94 \mathrm{mg}, 98 \%$ ). \%). Mp: 183$185{ }^{\circ} \mathrm{C}$ (Lit. ${ }^{28} \mathrm{mp}: 177-178{ }^{\circ} \mathrm{C}$ ). $R_{\mathrm{f}}=0.2$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta(\mathrm{ppm})$ 1.21-1.32 (m, 2H), 1.66-1.69 (m, 2H), 1.90-1.93 (m, 2H), 2.08-2.19 (m, 2H), 3.41-3.49 (m, 1H), 3.91-3.99 (m, 1H), 4.64 (d, 1H, J 4.26 Hz ), $7.80-7.85(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta$ (ppm) 27.8, 35.1, 49.9, 68.5, 123.4, 131.9, 134.8, 168.3. HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{NO}_{3}{ }^{+}: 246.1125$, found: $246.1126 . \mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{3}$ (245.28).

## 2-(4-oxocyclohexyl)isoindoline-1,3-dione (6.9) ${ }^{28,29}$

6.8 ( $2.6 \mathrm{~g}, 10.61 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 70 mL ). Under Ar-atmosphere first two spoons of Celite ${ }^{\circledR}$ were added followed by pyridinium chlorochromate ( $5.58 \mathrm{~g}, 25.88$
$\mathrm{mmol}, 2.44 \mathrm{eq}$ ). The reaction mixture was stirred for 3.5 h at room temperature. 2-Propanol (2 mL ) was added and after stirring for additional 30 min the solvent was removed under reduced pressure. The residue was purified by automated flash chromatography (EtOAc 100:0 isocratic for 10 min ). Removal of the solvent afforded the product as light beige solid ( $1.82 \mathrm{~g}, 71 \%$ ). Mp: 144$145{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.4$ (PE/EtOAc 1:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 2.02-2.10(\mathrm{~m}, 2 \mathrm{H}), 2.48-2.54$ $(\mathrm{m}, 4 \mathrm{H}), 2.65-2.79(\mathrm{~m}, 2 \mathrm{H}), 4.57-4.68(\mathrm{~m}, 1 \mathrm{H}), 7.69-7.75(\mathrm{~m}, 2 \mathrm{H}), 7.80-7.87(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(75$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 28.6,39.9,48.3,123.3,131.8,134.1,168.1,209.0$. HRMS: (ESI): $m / z[\mathrm{M}+\mathrm{H}]^{+}$ calcd. for $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{NO}_{3}{ }^{+}$: 244.0968, found: 244.0973. $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}_{3}$ (243.26).

## 2-(2-Amino-4,5,6,7-tetrahydrobenzo[d]thiazol-6-yl)isoindoline-1,3-dione (6.11) ${ }^{11,29}$

6.9 ( $1.93 \mathrm{~g}, 7.93 \mathrm{mmol}, 1 \mathrm{eq}$ ) was dissolved in a mixture of dioxane ( 22 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(14 \mathrm{~mL})$. Bromine ( $1.40 \mathrm{~g}, 8.73 \mathrm{mmol}, 1.1 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(24 \mathrm{~mL})$ was added drop wise and the reaction mixture was stirred for 1.5 h at room temperature. The solvent was removed under reduced pressure and the residue (crude 6.10) was dissolved in DMF ( 60 mL ). After the addition of thiourea ( $0.60 \mathrm{~g}, 7.93 \mathrm{mmol}, 1 \mathrm{eq}$ ) the reaction mixture was stirred for 2 h at $100^{\circ} \mathrm{C}$. The solvent was removed under reduced pressure and the residue was suspended in EtOAc ( 50 mL ). The precipitate was filtered off and the residual solvent was removed in vacuo to afford the product as a beige solid ( $2.77 \mathrm{~g}, 92 \%$ ). Mp : $>300{ }^{\circ} \mathrm{C}$ decomposition (Lit. ${ }^{11} \mathrm{mp}: 244-246{ }^{\circ} \mathrm{C}$ decomposition). $R_{\mathrm{f}}=0.90\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta$ (ppm) 1.99-2.05(m, 1H), 2.60-2.77 (m, 3H), 2.88-2.93 (m, 1H), 3.11-3.36 (m, 1H), 4.41-4.49 (m, $1 \mathrm{H})$, 7.84.7.90 ( $\mathrm{m}, 4 \mathrm{H}$ ). HRMS: (ESI): $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{H}]^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}^{+}: 300.0801$, found: 300.0817. $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}$ (299.35).

## 4,5,6,7-Tetrahydrobenzo[d]thiazole-2,6-diamine (6.12) ${ }^{29}$

6.11 ( $200 \mathrm{mg}, 0.53 \mathrm{mmol}, 1 \mathrm{eq}$ ) was suspended in a mixture of hydrochloric acid ( $37 \%, \mathrm{w} / \mathrm{v}, 5 \mathrm{~mL}$ ) and acetic acid $(5 \mathrm{~mL})$. The reaction mixture was stirred over night at $100{ }^{\circ} \mathrm{C}$. The solvent was removed under reduced pressure and the residue was dissolved in $\mathrm{H}_{2} \mathrm{O}(5 \mathrm{~mL})$. The pH value was adjusted to 6 by addition of NaOH solution ( $1 \mathrm{M}, 2 \mathrm{~mL}$ ). The aqueous layer was washed three times with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in order to remove by-products. Then additional NaOH solution ( 5 mL ) was added until a pH value of 9 was reached. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 10 mL ) and six times with EtOAc ( 10 mL ). In order to extract remaining product from the aqueous layer $\mathrm{NaCl}(1$ spoon) was dissolved in the mixture and the layer was again extracted three times with EtOAc ( 10 mL ). Water was partly removed under reduced pressure and the aqueous layer was again extracted five times with EtOAc ( 100 mL ). All organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Removal of the solvent in vacuo afforded the product as a yellow hygroscopic solid (65 $\mathrm{mg}, 73 \%)$. $R_{\mathrm{f}}=0.13\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): ~ \delta(\mathrm{ppm})$ 1.40-1.53 (m, 1H), 1.78-1.82 (m, 1H), 2.13-2.21 (m, 1H), 2.31-2.41 (m, 2H), 2.60-2.67 (m, 1H), 2.94-3.03 (m, 1H), 6.60 (br s, 2H). HRMS: (ESI): $m / z[M+H]^{+}$calcd. for $\mathrm{C}_{7} \mathrm{H}_{12} \mathrm{~N}_{3} \mathrm{~S}^{+}$: 170.0746, found: 170.0747. $\mathrm{C}_{7} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{~S}$ (169.25).

## General procedure for the synthesis of S-methylcarbamoyl thiourea derivatives (6.13-6.18) from isocyanates

The respective isocyanate ( 1 eq ) and triethylamine ( 2.25 eq ) were added to a solution of 3.32 (1.5 $\mathrm{eq})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(17-20 \mathrm{~mL})$. The reaction mixture was stirred overnight at room temperature. The organic layer was washed with three times with water ( 30 mL ) and subsequently with brine ( 30 mL ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed under reduced pressure. The crude product was purified by flash chromatography.

## S-Methyl-(N-(tertbutoxycarbonyl))-(N'-propylcarbamoyl)thiourea (6.13) Claudia Honisch master thesis

6.13 was prepared from propylisocyanate ( $149 \mathrm{mg}, 1.75 \mathrm{mmol}$ ), 3.32 ( $500 \mathrm{mg}, 2.63 \mathrm{mmol}$ ) and triethylamine ( $400 \mathrm{mg}, 3.94 \mathrm{mmol}$ ) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min ) and removal of the solvent in vacuo afforded the product as a white solid ( $0.31 \mathrm{~g}, 64 \%$ ). Mp: 56.1-60.2 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.72$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 0.94(\mathrm{t}, 3 \mathrm{H}, J 7.4 \mathrm{~Hz}), 1.47(\mathrm{~s}, 9 \mathrm{H}), 1.56(\mathrm{~m}, 2 \mathrm{H}), 2.31(\mathrm{~s}$, $3 \mathrm{H}), 3.18(\mathrm{q}, 2 \mathrm{H}, \mathrm{J} 6.9 \mathrm{~Hz}), 12.34(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 11.4,14.3,22.9,28.0$, 41.9, 82.6, 151.1, 161.9, 167.5. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{11} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 276.1376$, found: 276.1379. $\mathrm{C}_{11} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (275.37).

## S-Methyl-(N-(tertbutoxycarbonyl))-(N'-hexylcarbamoyl)thiourea (6.14) Claudia Honisch master thesis

6.14 was prepared from hexylisocyanate ( $223 \mathrm{mg}, 1.75 \mathrm{mmol}$ ), 3.32 ( $500 \mathrm{mg}, 2.63 \mathrm{mmol}$ ) and triethylamine ( $400 \mathrm{mg}, 3.94 \mathrm{mmol}$ ) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min ) and removal of the solvent in vacuo afforded the product as a white solid ( $0.45 \mathrm{~g}, 81 \%$ ). Mp: 56.4-59.2 ${ }^{\circ} \mathrm{C} . \mathrm{R}_{\mathrm{f}}=0.67$ (PE/EtOAc $3: 1$ ). ${ }^{1} \mathrm{H}-$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm})$ 0.85-0.90 (m, 3H), 1.26-1.37 (m, 6H), $1.47(\mathrm{~s}, 9 \mathrm{H}), 1.49-1.59(\mathrm{~m}$, $2 \mathrm{H}), 2.31(\mathrm{~s}, 3 \mathrm{H}), 3.19-3.24(\mathrm{~m}, 2 \mathrm{H}), 12.35(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.0$, 14.3, 22.6, 26.6, 28.0, 29.7, 31.5, 40.2, 82.6, 151.1, 161.8, 167.7. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{14} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 318.1846$, found: $318.1851 . \mathrm{C}_{14} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}(317.45)$.

## S-Methyl-(N-(tertbutoxycarbonyl))-(N'-cyclohexylcarbamoyl)thiourea (6.15) Claudia Honisch master thesis

6.15 was prepared from cyclohexylisocyanate ( $219 \mathrm{mg}, 1.75 \mathrm{mmol}$ ), 3.32 ( $500 \mathrm{mg}, 2.63 \mathrm{mmol}$ ) and triethylamine ( 2.25 eq., $400 \mathrm{mg}, 3.94 \mathrm{mmol}$ ) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min ) and removal of the solvent in vacuo afforded the product as a white solid ( $0.47 \mathrm{~g}, 85 \%$ ). Mp: 149.5-152.3 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.12-1.26(\mathrm{~m}, 3 \mathrm{H}), 1.31-1.42(\mathrm{~m}, 2 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H}), 1.59-1.66(\mathrm{~m}, 1 \mathrm{H}), 1.71-1.76(\mathrm{~m}$, $2 \mathrm{H}), 1.93-1.97(\mathrm{~m}, 2 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}), 3.53-3.63(\mathrm{~m}, 1 \mathrm{H}), 12.43(\mathrm{br}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ :
$\delta(\mathrm{ppm}) 14.4,24.9,25.5,28.0,33.1,49.1,82.7,151.0,160.9,167.6$. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 316.1689$, found: 316.1700. $\mathrm{C}_{14} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (315.43).

## S-Methyl-(N-(tert-butoxycarbonyl))-(N'-(phenylcarbamoyl)thiourea) (6.16) Claudia Honisch master thesis

6.16 was prepared from phenylisocyanate ( $210 \mathrm{mg}, 1.75 \mathrm{mmol}$ ), 3.32 ( $500 \mathrm{mg}, 2.63 \mathrm{mmol}$ ) and triethylamine ( $400 \mathrm{mg}, 3.9 \mathrm{mmol}$ ) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-90:10 in 25 min ) and removal of the solvent in vacuo afforded the product as a white solid ( $0.31 \mathrm{~g}, 57 \%$ ). Mp: 126.3-129.6 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.66$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.50(\mathrm{~s}, 9 \mathrm{H}), 2.40(\mathrm{~s}, 3 \mathrm{H}), 7.08-7.12(\mathrm{~m}, 1 \mathrm{H}), 7.31-7.35(\mathrm{~m}, 2 \mathrm{H})$, 7.42 (br s, 1H), 7.49-7.51 (m, 2H), 12.22 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.6,28.0$, 83.1, 119.4, 124.0, 129.1, 137.9, 151.0, 159.3, 169.2. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 310.1220$, found: 310.1228. $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (309.38)

## S-Methyl-(N-(tertbutoxycarbonyl))-(N'-benzylcarbamoyl)thiourea (6.17) Claudia Honisch master thesis

6.17 was prepared from benzylisocyanate ( $234 \mathrm{mg}, 1.75 \mathrm{mmol}$ ), 3.32 ( $500 \mathrm{mg}, 2.63 \mathrm{mmol}$ ) and triethylamine ( $400 \mathrm{mg}, 3.94 \mathrm{mmol}$ ) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min ) and removal of the solvent in vacuo afforded the product as a white solid ( $0.45 \mathrm{~g}, 80 \%$ ). Mp: 103.8-106.6 ${ }^{\circ} \mathrm{C} . \mathrm{R}_{\mathrm{f}}=0.77$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.49(\mathrm{~s}, 9 \mathrm{H}), 2.34(\mathrm{~s}, 3 \mathrm{H}), 4.43(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J} 6.09 \mathrm{~Hz}), 6.09(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 7.27-7.37 (m, 5H), 12.38 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.5,28.0,44.2,83.0$, 127.56, 127.64, 128.8, 138.2, 151.0, 155.3, 168.5. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}$: 324.1376, found: 324.1384. $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (323.41).

## S-Methyl-(N-(tertbutoxycarbonyl))-(N'-(2-phenylethyl)carbamoyl)thiourea <br> Claudia

 Honisch master thesis6.18 was prepared from phenethylisocyanate ( $258 \mathrm{mg}, 1.75 \mathrm{mmol}$ ), 3.32 ( $500 \mathrm{mg}, 2.63 \mathrm{mmol}$ ) and triethylamine ( $400 \mathrm{mg}, 3.94 \mathrm{mmol}$ ) according to general procedure. . Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min ) and removal of the solvent in vacuo afforded the product as a white solid ( $0.43 \mathrm{~g}, 73 \%$ ). Mp: $101.8-103.7^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.73$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.49(\mathrm{~s}, 9 \mathrm{H}), 2.29(\mathrm{~s}, 3 \mathrm{H}), 2.86(\mathrm{t}, 2 \mathrm{H}, J 7.18 \mathrm{~Hz}) 3.47-3.52$ $(\mathrm{m}, 2 \mathrm{H}), 7.17-7.34(\mathrm{~m}, 5 \mathrm{H}), 12.32(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.4,28.1,35.9$, 41.4, 82.8, 126.5, 128.6, 128.8, 138.8, 151.1, 161.7, 168.9. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 338.1533$, found: 338.1546. $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (337.44).

## General procedure for the synthesis of S-methylcarbamoyl thiourea derivatives (6.19-6.25) from carbonic acids

The respective carbon acid (1 eq) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.5-4.5 \mathrm{~mL})$ and cooled with an ice bath. DMF ( $25-45 \mu \mathrm{~L}$ ) and oxalyl chloride ( 1.5 eq ) were added under Ar-atmosphere. The reaction mixture was stirred for 10 min under cooling. The ice bath was removed and stirring was continued for another 15 min at room temperature. The solvent was carefully removed under reduced pressure (water bath temperature under $30^{\circ} \mathrm{C}$ ). The residue was dissolved in anhydrous acetone ( $2.5-4.5 \mathrm{~mL}$ ) and added drop wise under cooling to an ice cold solution of sodium azide $(2.4 \mathrm{eq})$ in $\mathrm{H}_{2} \mathrm{O}(1-3 \mathrm{~mL})$. The reaction mixture was stirred for 30 min under cooling. Brine (5-10 mL ) was added and the acyl azide was extracted three times with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 10 mL ). The organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Molecular sieve was added and the solvent was partially removed under reduced pressure. The resulting yellow solution was stirred for 30 min under reflux conditions to afford the isocyanate. The solution was cooled to room temperature and 3.32 ( 1 eq ) and triethylamine ( 5 eq ) were added. The reaction mixture was stirred over night at room temperature. The molecular sieve was filtered off and the organic layer was washed three times with $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ and three times with brine $(10 \mathrm{~mL})$. The organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The crude product was purified by either automated flash chromatography or column chromatography.

## S-Methyl-(N-(tertbutoxycarbonyl))-(N'-[2-methyl 3-(4-methylphenyl)propyl]carbamoyl)thiourea

 (6.19)6.19 was prepared from 3-methyl 4-(4-methylphenyl)butanoic acid ( $100 \mathrm{mg}, 0.52 \mathrm{mmol}, 1 \mathrm{eq}$ ), oxalyl chloride ( $99 \mathrm{mg}, 0.75 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), sodium azide ( $85 \mathrm{mg}, 1.30 \mathrm{mmol}, 2.4 \mathrm{eq}$ ), 3.32 ( 100 $\mathrm{mg}, 0.52 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $263 \mathrm{mg}, 2.60 \mathrm{mmol}, 5 \mathrm{eq}$ ) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 20 min ) and removal of the solvent in vacuo afforded the product as colourless oil ( $140 \mathrm{mg}, 71 \%$ ). $R_{\mathrm{f}}=0.80$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 0.92(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J} 6.5 \mathrm{~Hz}), 1.48(\mathrm{~s}, 9 \mathrm{H}), 1.92-2.04$ $(\mathrm{m}, 1 \mathrm{H}), 2.31-2.33(\mathrm{~m}, 6 \mathrm{H}), 2.39-2.44(\mathrm{~m}, 1 \mathrm{H}), 2.63-2.68(\mathrm{~m}, 1 \mathrm{H}), 3.08-3.22(\mathrm{~m}, 2 \mathrm{H}), 7.04-7.10(\mathrm{~m}$, 4 H ), 12.36 (br s, 1 H ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.3,17.7,21.0,28.0,35.7,40.8,46.0$, 82.6, 128.9, 129.0, 135.5, 137.1, 151.2, 162.0, 167.2. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{19} \mathrm{H}_{30} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 380.2002$, found: 380.2009. $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (379.52).

## S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-methyl 4-phenylbutyl)carbamoyl]thiourea (6.20)

6.20 was prepared from 3 -methyl 5-phenyl pentanoic acid ( $150 \mathrm{mg}, 0.78 \mathrm{mmol}, 1 \mathrm{eq}$ ), oxalyl chloride ( $149 \mathrm{mg}, 1.17 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), sodium azide ( $122 \mathrm{mg}, 1.87 \mathrm{mmol}, 2.4 \mathrm{eq}$ ), $3.32(148 \mathrm{mg}$, $0.78 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $395 \mathrm{mg}, 3.90 \mathrm{mmol}, 5 \mathrm{eq}$ ) according to general procedure. Purification by column chromatography ( $\mathrm{PE} / E t O A c 3: 1$ isocratic) and removal of the solvent in vacuo afforded the product as colourless oil ( $220 \mathrm{mg}, 74 \%$ ). $R_{\mathrm{f}}=0.66$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 1.01(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J} 6.57 \mathrm{~Hz}), 1.43-1.51(\mathrm{~m}, 10 \mathrm{H}), 1.62-1.78(\mathrm{~m}, 2 \mathrm{H}), 2.35(\mathrm{br} \mathrm{s}, 3 \mathrm{H})$, 2.55-2.77 (m, 2H), 3.05-3.25(m, 2H), 7.15-7.31 (m, 5H), $12.40(b r s, 1 H) .{ }^{13} \mathrm{C}-\mathrm{NMR}(100 \mathrm{MHz}$,
$\mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 14.4,17.7,28.2,33.3,33.4,36.3,46.1,82.7,125.9,128.46,128.49,142.5,151.3$, 162.2, 167.4. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{19} \mathrm{H}_{30} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}$: 380.2002, found: 380.2025. $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (379.52).

## S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-cyclohexyl propyl)carbamoyl]thiourea (6.21)

6.21 was prepared from 3-cyclohexyl butanoic acid ( $150 \mathrm{mg}, 0.88 \mathrm{mmol}, 1 \mathrm{eq}$ ), oxalyl chloride $(168 \mathrm{mg}, 1.32 \mathrm{mmol}, 1.5 \mathrm{eq})$, sodium azide ( $137 \mathrm{mg}, 2.11 \mathrm{mmol}, 2.4 \mathrm{eq}$ ), 3.32 ( $167 \mathrm{mg}, 0.88$ mmol, 1 eq ) and triethylamine ( $446 \mathrm{mg}, 4.41 \mathrm{mmol}, 5 \mathrm{eq}$ ) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent in vacuo afforded the product as colourless oil ( $210 \mathrm{mg}, 67 \%$ ). $R_{\mathrm{f}}=0.81$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 0.89(\mathrm{~d}, 3 \mathrm{H}, J 6.91 \mathrm{~Hz}), 0.97-1.30(\mathrm{~m}, 6 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H}), 1.62-1.76(\mathrm{~m}, 6 \mathrm{H})$, $2.37(\mathrm{br} \mathrm{s}, 3 \mathrm{H}), 2.99-3.09(\mathrm{~m}, 1 \mathrm{H}), 3.22-3.30(\mathrm{~m}, 1 \mathrm{H}), 12.43(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta(\mathrm{ppm}) 14.4,14.5,26.67,26.74,26.8,28.1,28.6,30.9,38.7,40.6,44.1,82.6,151.3,162.2,167.2$. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{32} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 358.2159$, found: 358.2199. $\mathrm{C}_{17} \mathrm{H}_{31} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (357.51).

## S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-(4-prop-2-yl phenyl) propyl)carbamoyl]thiourea (6.22)

6.22 was prepared from 3-(4-prop-2-yl phenyl) butanoic acid ( $150 \mathrm{mg}, 0.73 \mathrm{mmol}, 1 \mathrm{eq}$ ), oxalyl chloride ( $138 \mathrm{mg}, 1.09 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), sodium azide ( $114 \mathrm{mg}, 1.75 \mathrm{mmol}, 2.4 \mathrm{eq}$ ), 3.32 ( 138 mg , $0.73 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $368 \mathrm{mg}, 3.64 \mathrm{mmol}, 5 \mathrm{eq}$ ) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent in vacuo afforded the product as colourless oil ( $220 \mathrm{mg}, 77 \%$ ). $R_{\mathrm{f}}=0.74$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm})$ 1.24-1.30 (m, 9 H$), 1.48(\mathrm{~s}, 9 \mathrm{H}), 2.29(\mathrm{br} \mathrm{s}, 3 \mathrm{H}), 2.85-2.99(\mathrm{~m}, 2 \mathrm{H}), 3.24-3.33$ $(\mathrm{m}, 1 \mathrm{H}), 3.44-3.53(\mathrm{~m}, 1 \mathrm{H}), 7.12-7.21(\mathrm{~m}, 4 \mathrm{H}), 12.37(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ (ppm) 14.4, 19.4, 24.1, 28.15, 28.23, 33.8, 39.6, 47.1, 82.7, 126.8, 127.2, 141.5, 147.3, 151.2, 162.1, 167.5. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{20} \mathrm{H}_{32} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}$: 394.2159, found: 394.2159. $\mathrm{C}_{20} \mathrm{H}_{31} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (393.55).

## S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(3-methyl 2-phenyl butyl)carbamoyl]thiourea (6.23)

6.23 was prepared from 4-methyl 3 -phenyl pentanoic acid ( $150 \mathrm{mg}, 0.78 \mathrm{mmol}, 1 \mathrm{eq}$ ), oxalyl chloride ( $149 \mathrm{mg}, 1.17 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), sodium azide ( $122 \mathrm{mg}, 1.87 \mathrm{mmol}, 2.4 \mathrm{eq}$ ), 3.32 ( 148 mg , $0.78 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $395 \mathrm{mg}, 3.90 \mathrm{mmol}, 5 \mathrm{eq}$ ) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent in vacuo afforded the product as colourless oil ( $130 \mathrm{mg}, 44 \%$ ). $R_{\mathrm{f}}=0.81$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 0.75(\mathrm{~d}, 3 \mathrm{H}, J 6.71 \mathrm{~Hz}), 1.02(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J} 6.71 \mathrm{~Hz}), 1.48(\mathrm{~s}, 9 \mathrm{H}), 1.85-1.97(\mathrm{~m}$, $1 \mathrm{H}), 2.24(\mathrm{~s}, 3 \mathrm{H}), 2.50-2.58(\mathrm{~m}, 1 \mathrm{H}), 3.25-3.37(\mathrm{~m}, 1 \mathrm{H}), 3.79-3.88(\mathrm{~m}, 1 \mathrm{H}), 7.10-7.35(\mathrm{~m}, 5 \mathrm{H}), 12.36$ (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.4,20.5,20.9,28.0,31.5,43.1,52.6,82.8,126.7$, $128.48,128.54,141.6,151.0,161.4,167.7$. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{19} \mathrm{H}_{30} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}$: 380.2002, found: 380.2034. $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (379.52).

## S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-phenyl butyl)carbamoyl]thiourea (6.24)

6.24 was prepared from 3-phenyl pentanoic acid ( $150 \mathrm{mg}, 0.84 \mathrm{mmol}, 1 \mathrm{eq}$ ), oxalyl chloride ( 160 $\mathrm{mg}, 1.26 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), sodium azide ( $131 \mathrm{mg}, 2.02 \mathrm{mmol}, 2.4 \mathrm{eq}$ ), $3.32(160 \mathrm{mg}, 0.84 \mathrm{mmol}, 1$ eq) and triethylamine ( $426 \mathrm{mg}, 4.21 \mathrm{mmol}, 5 \mathrm{eq}$ ) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent in vacuo afforded the product as colourless oil ( $190 \mathrm{mg}, 62 \%$ ). $R_{\mathrm{f}}=0.71$ ( $\mathrm{PE} / \mathrm{EtOAc} 3: 1$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 0.82(\mathrm{t}, 3 \mathrm{H}, J 7.37 \mathrm{~Hz}), 1.48-1.85(\mathrm{~m}, 11 \mathrm{H}), 2.37(\mathrm{br}, 3 \mathrm{H}), 2.66-2.76(\mathrm{~m}, 1 \mathrm{H}), 3.25-$ $3.33(\mathrm{~m}, 1 \mathrm{H}), 3.56-3.65(\mathrm{~m}, 1 \mathrm{H}), 7.15-7.36(\mathrm{~m}, 5 \mathrm{H}), 12.53(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ (ppm) 12.0, 14.4, 26.8, 28.2, 45.7, 47.8, 82.7, 126.8, 128.0, 128.8, 142.6, 151.3, 162.0, 167.5. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 366.1846$, found: 366.1895. $\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (365.49).

## S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(3-cyclohexyl 2-methyl propyl)carbamoyl]thiourea (6.25)

6.25 was prepared from 4-cyclohexyl 3-methyl butanoic acid ( $150 \mathrm{mg}, 0.76 \mathrm{mmol}, 1 \mathrm{eq}$ ), oxalyl chloride ( $144 \mathrm{mg}, 1.13 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), sodium azide ( $118 \mathrm{mg}, 1.82 \mathrm{mmol}, 2.4 \mathrm{eq}$ ), 3.32 ( 144 mg , $0.76 \mathrm{mmol}, 1 \mathrm{eq}$ ) and triethylamine ( $383 \mathrm{mg}, 3.78 \mathrm{mmol}, 5 \mathrm{eq}$ ) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent in vacuo afforded the product as colourless oil ( $110 \mathrm{mg}, 38 \%$ ). $R_{\mathrm{f}}=0.72$ (PE/EtOAc 3:1). ${ }^{1} \mathrm{H}-\mathrm{NMR}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm})$ 0.79-0.95 (m, 5H), 0.99-1.06 (m, 1H), 1.14-1.34 (m, 5H), 1.49-1.83 (m, 16H), 2.68 (br s, 2H), 2.97-3.24 (m, 2H), 13.08 (br s, 1 H ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 14.6,17.9$, $26.3,26.4,26.7,28.0,30.3,32.9,34.2,34.8,42.4,46.6,83.6,150.9,162.0,168.2$. HRMS (ESI) $\mathrm{m} / \mathrm{z}$ $(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{18} \mathrm{H}_{34} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+}: 372.2315$, found: 372.2321. $\mathrm{C}_{18} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ (371.54).

## General procedure for the synthesis of the $\boldsymbol{N}^{G}$-carbamoylated guanidines

The respective S-methylcarbamoyl thiourea derivative 6.13-6.25 or 3.33 (1 eq) and the amine 6.5, 6.7, 6.12 (1-1.5 eq) or 3-(1-trityl-1H-pyrazol-4-yl)prop-1-yl-amine (3 eq) were dissolved in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5-25 \mathrm{~mL})$. Triethylamine (2.25-5 eq) and $\mathrm{HgCl}_{2}$ (2-2.2 eq) were added under Aratmosphere. The reaction mixture was stirred over night at room temperature. The resulting suspension was filtered through Celite ${ }^{\circledR}$ in order to remove the mercury salt and the crude product was purified by either automated flash chromatography or column chromatography. Removal of the solvent in vacuo afforded the Boc-protected products 6.26-6.46, 6.68 and 6.69. Subsequently, the deprotection was performed by stirring the intermediate in a mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and TFA over night at room temperature. The solvent was removed in vacuo and the product was purified by preparative HPLC.

## 4-Methyl-5-(3-(3-(propylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.47) Claudia Honisch master thesis

6.47 was prepared from $6.13(150 \mathrm{mg}, 0.54 \mathrm{mmol}, 1 \mathrm{eq}), 6.5(148 \mathrm{mg}, 0.54 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}$ ( $293 \mathrm{mg}, 1.08 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $164 \mathrm{mg}, 1.62 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25$
mL ) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min ) afforded the Boc-protected intermediate 6.26 as a glassy colourless solid ( $250 \mathrm{mg}, 93 \%$ ). $230 \mathrm{mg}(0.46 \mathrm{mmol})$ of 6.26 were dissolved in a mixture of TFA (1 $\mathrm{mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05\% aq. TFA 15:85-37:63, $t_{R}=12.6 \mathrm{~min}$ ) afforded the product as white fluffy solid ( $140 \mathrm{mg}, 58 \%$ ). Mp : $72.4-73.2{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.62\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right)$. IR (KBr): 602.13660 .12697 .18726 .78 $\begin{array}{llllllllllllllllll}761.84 & 797.25 & 844.40 & 1144.09 & 1197.67 & 1242.49 & 1269.25 & 1391.01 & 1439.56 & 1467.73 & 1545.20\end{array}$ $1697.452882 .672972 .533122 .823288 .70 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): 99.23\% $\left(t_{\mathrm{R}}=10.92\right.$ $\min , k=2.8) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): \delta(\mathrm{ppm}) 0.85(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}$ 7.36 Hz ), 1.42-1.48 (m, 2H), 1.72 (qui, $2 \mathrm{H}, J 7.23 \mathrm{~Hz}$ ), $2.07(\mathrm{~s}, 3 \mathrm{H}), 2.59(\mathrm{t}, 2 \mathrm{H}, J 7.47 \mathrm{~Hz}), 3.04-3.07$ $(\mathrm{m}, 2 \mathrm{H}), 3.22-3.26(\mathrm{~m}, 2 \mathrm{H}$, interfering with the water signal), $7.49(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.49(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 8.87$ (br s, 2H), 8.99 (br s, 1H), 10.22 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] D M S O, ~ C O S Y, ~ H S Q C, ~ H M B C\right.$, NOESY): $\delta(\mathrm{ppm}) 11.1,11.6,22.0,22.2,28.9,1 \mathrm{C}$ under solvent peak (38.7-40.3), 40.9, 116.3, 132.6, 153.7, 153.8, 167.6. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{12} \mathrm{H}_{23} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 299.1649, found: 299.1651. $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(298.16+228.05)$.

## 4-Methyl-5-(3-(3-(hexylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.48) Claudia Honisch master thesis

6.48 was prepared from 6.14 ( $150 \mathrm{mg}, 0.47 \mathrm{mmol}, 1 \mathrm{eq}$ ), 6.5 ( $128 \mathrm{mg}, 0.47 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), $\mathrm{HgCl}_{2}$ ( $257 \mathrm{mg}, 0.95 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $144 \mathrm{mg}, 1.42 \mathrm{mmol}, 2.25 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25$ mL ) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min ) afforded the Boc-protected intermediate 6.27 as a glassy colourless solid ( $90 \mathrm{mg}, 35 \%$ ). $85 \mathrm{mg}(0.16 \mathrm{mmol})$ of 6.27 were dissolved in a mixture of TFA (1 $\mathrm{mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05\% aq. TFA 24:76-46:54, $t_{R}=13.0 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $50 \mathrm{mg}, 56 \%$ ). $R_{\mathrm{f}}=0.56\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in MeOH 9:1). RP-HPLC (220 nm, gradient 2): 99.1\% ( $t_{\mathrm{R}}=16.70 \mathrm{~min}, k$ $=4.8) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): \delta(\mathrm{ppm}) 0.84-0.86(\mathrm{~m}, 3 \mathrm{H})$, 1.24-1.28 (m, 6H), 1.39-1.43 (m, 2H), 1.71 (qui, $2 \mathrm{H}, J 7.27 \mathrm{~Hz}$ ), $2.05(\mathrm{~s}, 3 \mathrm{H}), 2.58(\mathrm{t}, 2 \mathrm{H}, J 7.44 \mathrm{~Hz})$, 3.06-3.09 (m, 2H), 3.22-3.25 (m, 2 H , interfering with the water signal), $6.57(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.49(\mathrm{br} \mathrm{s}$, $1 \mathrm{H}), 8.50(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 8.73(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 9.00(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 10.39(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right.$, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 11.8,13.8,21.98,22.03,25.8,28.8,29.0,30.8,2 \mathrm{Cs}$ under solvent peak (38.7-40.3), 116.3, 133.5, 153.7, 153.8, 167.4. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{29} \mathrm{~N}_{6} \mathrm{OS}^{+}: 341.2118$, found: 341.2127. $\mathrm{C}_{15} \mathrm{H}_{28} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}$ (340.49+228.05).

## 4-Methyl-5-(3-(3-(cyclohexylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.49) Claudia Honisch master thesis

6.49 was prepared from 6.15 ( $150 \mathrm{mg}, 0.47 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.5(129 \mathrm{mg}, 0.47 \mathrm{mmol}, 1.5 \mathrm{eq}), \mathrm{HgCl}_{2}$ ( $258 \mathrm{mg}, 0.95 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $144 \mathrm{mg}, 1.43 \mathrm{mmol}, 2.25 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25$ mL ) according to the general procedure. Purification by automated flash chromatography
(PE/EtOAc 100:0-85:15 in 30 min ) afforded the Boc-protected intermediate 6.28 as a glassy colourless solid ( $0.14 \mathrm{~g}, 55 \%$ ). $140 \mathrm{mg}(0.26 \mathrm{mmol})$ of 6.28 were dissolved in a mixture of TFA (1 $\mathrm{mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05\% aq. TFA 19:81-42:58, $t_{\mathrm{R}}=13.9 \mathrm{~min}$ ) afforded the product as white fluffy solid ( $99 \mathrm{mg}, 67 \%$ ). Mp : $69.6-71.8^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.82\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right)$. FT-ATR: $723,798,839,895,1129,1178$, $1315,1435,1543,1654,2858,2937,3086,3280 \mathrm{~cm}^{-1}$. RP-HPLC (gradient $\left.2,220 \mathrm{~nm}\right): 99.10 \%\left(t_{\mathrm{R}}=\right.$ $14.55 \mathrm{~min}, k=4.0) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): \delta(\mathrm{ppm}) 1.11-$ $1.32(\mathrm{~m}, 5 \mathrm{H}), 1.51-1.54(\mathrm{~m}, 1 \mathrm{H}), 1.63-1.77(\mathrm{~m}, 6 \mathrm{H}), 2.08(\mathrm{~s}, 3 \mathrm{H}), 2.59(\mathrm{t}, 2 \mathrm{H}, J 7.51 \mathrm{~Hz}), 3.23(\mathrm{q}, 2 \mathrm{H}$, J 6.25 Hz ), 3.44-3.46 (m, 1H), 7.47 (br s, 1H), 8.47 (br s, 2H), 8.96 (br s, 1H), $9.12(b r s, 2 H), 10.06$ (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]\right.$ DMSO, COSY, HSQC, HMBC, NOESY): $\delta$ (ppm) 11.3, 21.9, 24.1, 24.9, 28.7, 32.0, 1C under solvent peak (38.7-40.3), 48.3, 116.3, 131.4, 152.8, 153.7, 167.9. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{27} \mathrm{~N}_{6} \mathrm{OS}^{+}: 339.1962$, found: 339.1970. $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}$ (338.47 + 228.05).

## 4-Methyl-5-(3-(3-(phenylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.50) Claudia Honisch master thesis

6.50 was prepared from $6.16(100 \mathrm{mg}, 0.32 \mathrm{mmol}, 1 \mathrm{eq}), 6.5(88 \mathrm{mg}, 0.32 \mathrm{mmol}, 1.5 \mathrm{eq}), \mathrm{HgCl}_{2}$ ( $175 \mathrm{mg}, 0.65 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $98 \mathrm{mg}, 0.97 \mathrm{mmol}, 2.25 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25$ mL ) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min ) afforded the Boc-protected intermediate 6.29 as a glassy colourless solid ( $110 \mathrm{mg}, 64 \%$ ). $100 \mathrm{mg}(0.19 \mathrm{mmol})$ of 6.29 were dissolved in a mixture of TFA (5 $\mathrm{mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05\% aq. TFA 19:81-42:58, $t_{R}=12.4 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $63 \mathrm{mg}, 60 \%$ ). $R_{\mathrm{f}}=0.80\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in MeOH 9:1). FT-ATR: 693, 719, 753, 798, 835, 1126, 1178, 1316, 1446, 1498, 1551, 1640, 1595, 2363, 3094, $3276 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): 98.98\% ( $t_{\mathrm{R}}=$ $14.75 \mathrm{~min}, k=4.09) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): \delta(\mathrm{ppm}) 1.75$ (qui, $2 \mathrm{H}, J 7.21 \mathrm{~Hz}$ ), $2.09(\mathrm{~s}, 3 \mathrm{H}), 2.61(\mathrm{t}, 2 \mathrm{H}, J 7.50 \mathrm{~Hz}), 3.28(\mathrm{q}, 2 \mathrm{H}, J 6.47 \mathrm{~Hz}), 7.08-7.12(\mathrm{~m}, 1 \mathrm{H})$, $7.32-7.36(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.45(\mathrm{~m}, 2 \mathrm{H}), 8.56(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 8.93(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 9.05(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 10.47(\mathrm{br} \mathrm{s}$, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): \delta(\mathrm{ppm}) 11.4,21.9,28.7,1 \mathrm{C}$ under solvent peak (38.7-40.3), 116.3, 119.6, 123.9, 129.0, 133.8, 137.5, 153.4 (2C), 167.8. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 333.1492 , found: 333.1496. $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}$ (332.43 + 228.05).

## 4-Methyl-5-(3-(3-((phenylmethyl)carbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.51) Claudia Honisch master thesis

6.51 was prepared from 6.17 ( $150 \mathrm{mg}, 0.464 \mathrm{mmol}, 1 \mathrm{eq}$ ), 6.5 ( $126 \mathrm{mg}, 0.464 \mathrm{mmol}, 1.5 \mathrm{eq}$ ), $\mathrm{HgCl}_{2}(252 \mathrm{mg}, 0.928 \mathrm{mmol}, 2 \mathrm{eq})$ and triethylamine ( $141 \mathrm{mg}, 1.392 \mathrm{mmol}, 2.25 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min ) afforded the Boc-protected intermediate 6.30
as a glassy colourless solid ( $170 \mathrm{mg}, 67 \%$ ). $90 \mathrm{mg}(0.16 \mathrm{mmol})$ of 6.30 were dissolved in a mixture of TFA ( 1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred for 6 h at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: $\mathrm{MeCN} / 0.05 \%$ aq. TFA 19:81-42:58, $t_{\mathrm{R}}=12.2 \mathrm{~min}$ ) afforded the product as white hygroscopic solid (49 mg, 52\%). $R_{\mathrm{f}}=0.73\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 9: 1$ ). FT-ATR: 723.1, 797.7, 834.9, 1129.4, 1177.8, 1252.4, 1431.3, 1543.1, 1640.0, $3090.0,3276.3 \mathrm{~cm}^{-1}$. RP-HPLC (gradient 2, 220 nm ): $97.4 \%\left(t_{\mathrm{R}}=14.89 \mathrm{~min}, k=4.1\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): ~ \delta$ (ppm) 1.72 (qui, $2 \mathrm{H}, J 7.22 \mathrm{~Hz}$ ), 2.07 (s, 3 H ), 2.58 (t, $2 \mathrm{H}, J 7.49 \mathrm{~Hz}$ ), 3.23 (q, $2 \mathrm{H}, J 6.66 \mathrm{~Hz}$ ), 4.30 (d, $2 \mathrm{H}, \mathrm{J} 5.83 \mathrm{~Hz}), 7.25-7.28(\mathrm{~m}, 3 \mathrm{H}), 7.32-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.98(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.52(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 9.01(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 9.10 (br s, 2H), 10.35 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] D M S O, ~ C O S Y, ~ H S Q C, ~ H M B C, ~ N O E S Y\right): ~ \delta$ (ppm) 11.3, 21.9, 28.7, 40.0, 42.7, 116.3, 127.1, 127.2, 128.4, 131.6, 138.6, 153.7, 153.8, 167.8. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 347.1649, found: 347.1653. $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}$ (346.45 + 228.05).

## 4-Methyl-5-(3-(3-((2-phenylethyl)carbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.52) Claudia Honisch master thesis

6.52 was prepared from 6.18 ( $150 \mathrm{mg}, 0.444 \mathrm{mmol}, 1 \mathrm{eq}$ ), 6.5 ( $120 \mathrm{mg}, 0.444 \mathrm{mmol}, 1 \mathrm{eq}$ ), $\mathrm{HgCl}_{2}$ ( $241 \mathrm{mg}, 0.888 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $135 \mathrm{mg}, 1.332 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25$ mL ) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min ) afforded the Boc-protected intermediate 6.31 as a glassy colourless solid ( $190 \mathrm{mg}, 76 \%$ ). $190 \mathrm{mg}(0.34 \mathrm{mmol})$ of 6.31 were dissolved in a mixture of TFA (1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred for 6 h at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05\% aq. TFA 19:81-46:54, $t_{\mathrm{R}}=13.2 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $11 \mathrm{mg}, 53 \%$ ). $R_{\mathrm{f}}=$ $0.89\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in MeOH 9:1). RP-HPLC (220 nm, gradient 2): 98.6\% ( $t_{\mathrm{R}}=15.0 \mathrm{~min}, k=$ 4.2) ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): ~ \delta(\mathrm{ppm}) 1.71$ (qui, $2 \mathrm{H}, J 7.19 \mathrm{~Hz}$ ), $2.08(\mathrm{~s}, 3 \mathrm{H}), 2.58-2.60(\mathrm{~m}, 2 \mathrm{H}), 2.75(\mathrm{t}, 2 \mathrm{H}, \mathrm{J} 7.24 \mathrm{~Hz}), 3.24(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.32-3.33(\mathrm{~m}, 2 \mathrm{H}), 7.19-7.22$ $(\mathrm{m}, 3 \mathrm{H}), 7.28-7.30(\mathrm{~m}, 2 \mathrm{H}), 7.60(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.52(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 9.04(\mathrm{br}, 3 \mathrm{H}), 10.29(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 11.5,21.9,28.8,35.0,40.0,40.7$, 116.3, 126.2, 128.4, 128.6, 131.7, 138.9, 153.6, 153.7, 167.7. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{~N}_{6} \mathrm{OS}^{+}: 361.1805$, found: 361.1812. $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(360.48+228.05)$.

## 4-Methyl-5-(3-[3-([2-methyl 3-(4-methylphenyl)propyl]carbamoyl)guanidino]propyl)-2aminothiazole bis(hydrotrifluoroacetate) (6.53)

6.53 was prepared from 6.19 ( $84 \mathrm{mg}, 0.22 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.5(60 \mathrm{mg}, 0.22 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(120$ $\mathrm{mg}, 0.44 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $67 \mathrm{mg}, 0.66 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1-1:1) afforded the Boc-protected intermediate 6.32 as a yellow oil ( $90 \mathrm{mg}, 68 \%$ ). 90 mg of 6.32 were dissolved in a mixture of TFA ( 1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 0:100-65:35, $t_{\mathrm{R}}=16.9 \mathrm{~min}$ ) afforded the product
as white hygroscopic solid ( 30.1 mg , 32\%). $R_{\mathrm{f}}=0.73\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 4: 1$ ). RP-HPLC (gradient 2, 220 nm ): 97.9\% ( $t_{\mathrm{R}}=20.21 \mathrm{~min}, k=5.97$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}\right.$, HMBC, NOESY): $\delta(\mathrm{ppm}) 0.78$ (d, $3 \mathrm{H}, J 6.69 \mathrm{~Hz}$ ), 1.72 (qui, $2 \mathrm{H}, J 7.20 \mathrm{~Hz}$ ), 1.82-1.88 (m, 1H), 2.07 $(\mathrm{s}, 3 \mathrm{H}), 2.25(\mathrm{~s}, 3 \mathrm{H}), 2.28-2.31(\mathrm{~m}, 1 \mathrm{H}), 2.55-2.60(\mathrm{~m}, 3 \mathrm{H}), 2.89-2.93(\mathrm{~m}, 1 \mathrm{H}), 3.05-3.09(\mathrm{~m}$, $1 \mathrm{H}), 3.23(\mathrm{q}, 2 \mathrm{H}, \mathrm{J} 6.50 \mathrm{~Hz}), 7.03-7.07(\mathrm{~m}, 4 \mathrm{H}), 7.53(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.51(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 8.99(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 9.14$ (br s, 2H), 10.49 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): ~ \delta(\mathrm{ppm})$ 11.3, 17.0, 20.6, 21.9, 28.8, 34.9, 2Cs under solvent peak (38.7-40.3), 44.6, 116.3, 128.76, 128.79, 131.6, 134.7, 137.0, 153.77, 153.78, 167.9. HRMS (ESI) $\mathrm{m} / \mathrm{z}(M+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{20} \mathrm{H}_{31} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 403.2275, found: 403.2277. $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(402.56+228.05)$.

## 4-Methyl-5-(3-[3-([2-methyl 4-phenylbutyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.54)

6.54 was prepared from 6.20 ( $90 \mathrm{mg}, 0.24 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.5(64 \mathrm{mg}, 0.24 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(129$ $\mathrm{mg}, 0.47 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $72 \mathrm{mg}, 0.71 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.33 as a colourless oil ( $90 \mathrm{mg}, 63 \%$ ). 90 mg ( 0.15 mmol ) of 6.33 were dissolved in a mixture of TFA ( 1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 mL ) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-70:30, $t_{R}=18.22 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $30.0 \mathrm{mg}, 32 \%$ ). Mp: 61-68 ${ }^{\circ} \mathrm{C}$. $R_{\mathrm{f}}=0.35\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1 \% \mathrm{NH}_{3}\right.$ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm ): $99.8 \%\left(t_{\mathrm{R}}=18.49 \mathrm{~min}, k=5.38\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, [D6]DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 0.89$ (d, 3H, J 6.45 Hz ), 1.33-1.39 (m, 1H), 1.56$1.63(\mathrm{~m}, 2 \mathrm{H}), 1.71$ (qui, $2 \mathrm{H}, \mathrm{J} 7.21 \mathrm{~Hz}$ ), $2.05(\mathrm{~s}, 3 \mathrm{H}), 2.51-2.65(\mathrm{~m}, 4 \mathrm{H}), 2.96-3.01(\mathrm{~m}, 1 \mathrm{H}), 3.05-3.09$ $(\mathrm{m}, 1 \mathrm{H}), 3.21-3.25(\mathrm{~m}, 2 \mathrm{H}$, interfering with the water signal), 7.14-7.19 (m,3H), 7.24-7.26 (m, 2H), 7.51(br s, 1H), 8.49 (br s, 2H), 8.82 (br s, 2H), 8.98(br s, 1H), 10.32 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, [D. ${ }_{6}$ ]DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 11.7,17.2,22.0,28.9,32.4,32.5,35.6,1 \mathrm{C}$ under solvent peak (38.7-40.3), 44.8, 116.3, 125.6, 128.20, 128.23, 132.9, 142.2, 153.7, 153.8, 167.5. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 403.2275, found: 403.2277. $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}$ (402.56 + 228.05).

## 4-Methyl-5-(3-[3-([2-cyclohexylpropyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.55)

6.55 was prepared from 6.21 ( $90 \mathrm{mg}, 0.25 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.5(68 \mathrm{mg}, 0.25 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(137$ $\mathrm{mg}, 0.50 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $76 \mathrm{mg}, 0.76 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.34 as a colourless oil ( $130 \mathrm{mg}, 89 \%$ ). 130 mg ( 0.22 mmol ) of 6.34 were dissolved in a mixture of TFA ( 1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 mL ) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-70:30, $t_{R}=18.8 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $80.0 \mathrm{mg}, 59 \%$ ). Mp: 51-57 ${ }^{\circ} \mathrm{C}$. $R_{\mathrm{f}}=0.34\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N}\right.$ $\mathrm{NH}_{3}$ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm ): 99.0\% ( $t_{\mathrm{R}}=18.61 \mathrm{~min}, k=5.42$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( 600
$\mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 0.78$ (d, $\left.3 \mathrm{H}, \mathrm{J} 6.8 \mathrm{~Hz}\right), 0.90-1.22(\mathrm{~m}, 6 \mathrm{H})$, 1.43-1.47 (m, 1H), 1.55-1.61 (m, 3H), 1.67-1.74 (m, 4H), 2.07 (s, 3H), 2.58-2.60 (m, 2H), 2.90-2.94 (m, 1H), 3.10-3.14 (m, 1H), 3.22-3.25 (m, 2H), 7.44 (br s, 1H), 8.49 (br s, 2H), 8.98-9.07 (m, 3H), 10.34 (br s, 1H), 13.55 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right.$, NOESY): $\delta$ (ppm) 11.4, 14.0, 22.0, 26.06, 26.16, 26.23, 27.9, 28.8, 30.3, 37.8, 2 Cs under solvent peak (38.740.3), 42.9, 116.3, 131.8, 153.70, 153.74, 167.8. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{18} \mathrm{H}_{33} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 381.2431, found: 381.2436. $\mathrm{C}_{18} \mathrm{H}_{32} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(380.56+228.05)$.

## 4-Methyl-5-(3-[3-([2-(4-prop-2-ylphenyl)propyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.56)

6.56 was prepared from 6.22 ( $90 \mathrm{mg}, 0.23 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.5(62 \mathrm{mg}, 0.23 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(124$ $\mathrm{mg}, 0.46 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $69 \mathrm{mg}, 0.69 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.35 as a colourless oil ( $120 \mathrm{mg}, 85 \%$ ). 120 mg ( 0.19 mmol ) of 6.35 were dissolved in a mixture of TFA $(1 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-70:30, $t_{R}=19.1 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( 80.0 mg , $64 \%$ ). Mp : 59-64 ${ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.34\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.5 \mathrm{~N} \mathrm{NH}_{3}\right.$ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm ): $96.3 \%\left(t_{\mathrm{R}}=19.33 \mathrm{~min}, k=5.67\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, [ $\mathrm{D}_{6}$ ]DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 1.16-1.18(\mathrm{~m}, 9 \mathrm{H}), 1.72$ (qui, $2 \mathrm{H}, \mathrm{J} 7.14 \mathrm{~Hz}$ ), 2.08 $(\mathrm{s}, 3 \mathrm{H}), 2.58-2.61(\mathrm{~m}, 2 \mathrm{H}), 2.82-2.89(\mathrm{~m}, 2 \mathrm{H}), 3.22-3.25(\mathrm{~m}, 4 \mathrm{H}), 7.14-7.18(\mathrm{~m}, 4 \mathrm{H}), 7.38(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 8.51 (br s, 2H), 9.01 (br s, 1H), 9.23 (br s, 2H), 10.36 (br s, 1 H$) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right.$, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 11.2,19.1,21.9,23.9,28.7,33.0,38.6,1 \mathrm{C}$ under solvent peak (38.7-40.3), 46.0, 116.3, 126.3, 127.0, 131.3, 141.5, 146.4, 153.68, 153.71, 168.0. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{21} \mathrm{H}_{33} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 417.2431, found: 417.2435. $\mathrm{C}_{21} \mathrm{H}_{32} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}$ (416.59 + 228.05).

## 4-Methyl-5-(3-[3-([3-methyl 2-phenylbutyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.57)

6.57 was prepared from $6.23(90 \mathrm{mg}, 0.24 \mathrm{mmol}, 1 \mathrm{eq}), 6.5(64 \mathrm{mg}, 0.24 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(129$ $\mathrm{mg}, 0.47 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $72 \mathrm{mg}, 0.71 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.36 as a colourless oil ( $110 \mathrm{mg}, 77 \%$ ). 110 mg ( 0.18 mmol ) of 6.36 were dissolved in a mixture of TFA $(1 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. The solvent was removed in vacuo and the residue was purified by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-70:30, $t_{\mathrm{R}}=$ $17.9 \mathrm{~min})$. The resulting product $(80 \mathrm{mg})$ showed an insufficient purity and was purified a second time by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 5:95-65:35, $t_{R}=20.2 \mathrm{~min}$ ). Lyophilisation afforded the product as white hygroscopic solid ( $40.0 \mathrm{mg}, 17 \%$ ). Mp : $59-64{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.34\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.5 \mathrm{~N} \mathrm{NH}_{3}\right.$ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm ): 98.6\% ( $t_{\mathrm{R}}=$ $18.31 \mathrm{~min}, k=5.31$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): \delta(\mathrm{ppm}) 0.66$ (d,
$3 \mathrm{H}, J 6.73 \mathrm{~Hz}), 0.91(\mathrm{~d}, 3 \mathrm{H}, J 6.72 \mathrm{~Hz}), 1.69(q u i, 2 H, J 7.17 \mathrm{~Hz}), 1.83-1.88(\mathrm{~m}, 1 \mathrm{H}), 2.06(\mathrm{~s}, 3 \mathrm{H})$, 2.52-2.57 (m, 3H), 3.19-3.20 (m, 2H), 3.31-3.36 (m, 1H), 3.56-3.60 (m, 1H), 7.11-7.16 (m, 3H), 7.19-7.22 (m, 1H), 7.28-7.30 (m, 2H), 8.45 (br s, 2H), 8.93 (br s, 1H), 9.16 (br s, 2H), 10.08 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, ~ C O S Y, ~ H S Q C, ~ H M B C, ~ N O E S Y\right): ~ \delta(p p m) ~ 11.3, ~ 20.0, ~ 20.7, ~ 21.9, ~$ 28.7, 30.4, 1 C under solvent peak (38.7-40.3), 42.0, 51.5, 116.3, 126.4, 128.1, 128.4, 131.3, 141.4, 153.51, 153.54, 167.9. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{20} \mathrm{H}_{31} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 403.2275, found: 403.2282. $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(402.56+228.05)$.

## 4-Methyl-5-(3-[3-([2-phenylbutyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.58)

6.58 was prepared from 6.24 ( $90 \mathrm{mg}, 0.25 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.5(67 \mathrm{mg}, 0.25 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(134$ $\mathrm{mg}, 0.50 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $75 \mathrm{mg}, 0.74 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.37 as a colourless oil ( $110 \mathrm{mg}, 76 \%$ ). 110 mg ( 0.19 mmol ) of 6.37 were dissolved in a mixture of TFA $(1 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: $\mathrm{MeCN} / 0.1 \% \mathrm{aq}$. TFA 10:90-70:30, $t_{\mathrm{R}}=16.1 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( 80.0 mg , $69 \%$ ). $\mathrm{Mp}: 55-60{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.34\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.5 \mathrm{~N} \mathrm{NH} 3\right.$ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm ): $97.5 \%\left(t_{\mathrm{R}}=16.71 \mathrm{~min}, k=4.76\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, [D6]DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm}) 0.71(\mathrm{t}, 3 \mathrm{H}, \mathrm{J} 7.32 \mathrm{~Hz}), 1.46-1.54(\mathrm{~m}, 1 \mathrm{H}), 1.64-$ $1.73(\mathrm{~m}, 3 \mathrm{H}), 2.06(\mathrm{~s}, 3 \mathrm{H}), 2.56-2.58(\mathrm{~m}, 2 \mathrm{H}), 2.62-2.67(\mathrm{~m}, 1 \mathrm{H}), 3.20-3.29(\mathrm{~m}, 3 \mathrm{H}), 3.34-3.38(\mathrm{~m}$, $1 \mathrm{H}), 7.18-7.21(\mathrm{~m}, 3 \mathrm{H}), 7.28-7.31(\mathrm{~m}, 3 \mathrm{H}), 8.50(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 8.98(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 9.08(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 10.37(\mathrm{br}$ $\mathrm{s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): \delta(\mathrm{ppm}) 11.4,11.7,22.0$, 25.8, 28.8, 1 C under solvent peak (38.7-40.3), 44.6, 46.7, 116.3, 126.4, 127.7, 128.4, 132.0, 142.5, 153.67, 153.69, 167.8. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 389.2118, found: 389.2118. $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(388.53+228.05)$.

## 4-Methyl-5-(3-[3-([3-cyclohexyl 2-methylpropyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.59)

6.59 was prepared from $6.25(80 \mathrm{mg}, 0.22 \mathrm{mmol}, 1 \mathrm{eq}), 6.5(59 \mathrm{mg}, 0.22 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(117$ $\mathrm{mg}, 0.43 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $66 \mathrm{mg}, 0.65 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.38 as a colourless oil ( $140 \mathrm{mg}, 97 \%$ ). 140 mg ( 0.23 mmol ) of 6.38 were dissolved in a mixture of TFA ( 1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90-70:30, $t_{R}=19.9 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( 110.0 mg , $75 \%$ ). Mp: 61-66 ${ }^{\circ} \mathrm{C}$. $R_{\mathrm{f}}=0.34\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1 \mathrm{~N} \mathrm{NH} 3\right.$ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm ): $99.7 \%\left(t_{\mathrm{R}}=20.10 \mathrm{~min}, k=5.93\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, [D6]DMSO, COSY, HSQC, HMBC, NOESY): $\delta(\mathrm{ppm})$ 0.73-0.88 (m, 5H), 0.93-0.97 (m, 1H), 1.07-1.23 $(\mathrm{m}, 4 \mathrm{H}), 1.26-1.31(\mathrm{~m}, 1 \mathrm{H}), 1.59-1.75(\mathrm{~m}, 8 \mathrm{H}), 2.08(\mathrm{~s}, 3 \mathrm{H}), 2.58-2.61(\mathrm{~m}, 2 \mathrm{H}), 2.84-2.89(\mathrm{~m}, 1 \mathrm{H})$, 3.02-3.06 (m, 1H), 3.23-3.26 (m, 2H), 7.47 (br s,1H), 8.51 (br s, 2H), 9.00-9.06 (m, 3H), $10.42(\mathrm{br} \mathrm{s}$,

1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}, \mathrm{NOESY}\right): ~ \delta(\mathrm{ppm}) 11.4,17.7,22.0,25.7$, 25.8, 26.2, 28.8, 29.6, 32.5, 33.5, 34.2, 1C under solvent peak (38.7-40.3), 41.7, 45.3, 116.3, 131.9, 153.77, 153.78, 167.8. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{19} \mathrm{H}_{35} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 395.2588 , found: 395.2590. $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(394.58+228.05)$.

## 4-(3-(3-(Propylcarbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.60)

6.60 was prepared from $6.13(80 \mathrm{mg}, 0.29 \mathrm{mmol}, 1 \mathrm{eq}), 6.7(56 \mathrm{mg}, 0.29 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(158$ $\mathrm{mg}, 0.58 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $88 \mathrm{mg}, 0.87 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-50:50 in 25 min ) afforded the Boc-protected intermediate 6.39 as a yellow oil ( 70 mg , $57 \%$ ). 70 mg ( 0.17 mmol ) of 6.39 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85-45:55, $t_{\mathrm{R}}=$ 10.3 min ) afforded the product as white hygroscopic solid ( $40.0 \mathrm{mg}, 44 \%$ ). $R_{\mathrm{f}}=0.10$ (PE/EtOAc 2:1). RP-HPLC (gradient 1, 220 nm ): 99.2\% ( $t_{\mathrm{R}}=13.87 \mathrm{~min}, k=3.78$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, [ $\mathrm{D}_{6}$ ]DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 0.86(\mathrm{t}, 3 \mathrm{H}, \mathrm{J} 7.50 \mathrm{~Hz}), 1.43-1.49(\mathrm{~m}, 2 \mathrm{H}), 3.06-3.10(\mathrm{~m}$, $2 \mathrm{H}), 7.15(\mathrm{~s}, 1 \mathrm{H}), 7.23-7.24(\mathrm{~m}, 1 \mathrm{H}), 7.49(\mathrm{t}, 1 \mathrm{H}, J 7.74 \mathrm{~Hz}), 7.60(\mathrm{t}, 1 \mathrm{H}, J 5.56 \mathrm{~Hz}), 7.72-7.73(\mathrm{~m}$, 1H), 7.78-7.80 (m, 1H), 8.62 (br s, 1H), 8.97 (br s, 1H), $10.15(\mathrm{~s}, 1 \mathrm{H}), 10.74(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 11.2,22.2,41.0,103.0,123.0,124.81,124.84$, 130.1, 134.0, 135.5, 147.0, 153.5, 153.6, 168.7. HRMS (ESI) $\mathrm{m} / \mathrm{z}(M+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 319.1336, found: 319.1346. $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(318.40+228.05)$.

## 4-(3-(3-(Hexylcarbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.61)

6.61 was prepared from $6.14(90 \mathrm{mg}, 0.28 \mathrm{mmol}, 1 \mathrm{eq}), 6.7(54 \mathrm{mg}, 0.28 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(154$ $\mathrm{mg}, 0.57 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $86 \mathrm{mg}, 0.85 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-55:45 in 25 min ) afforded the Boc-protected intermediate 6.40 as a yellow solid ( 110 mg , $84 \%$ ). $110 \mathrm{mg}(0.24 \mathrm{mmol})$ of 6.40 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: $0-30 \mathrm{~min}$ : MeCN/0.1\% aq. TFA 20:80-70:30, $t_{\mathrm{R}}=$ 13.4 min ) afforded the product as white hygroscopic solid ( $60.0 \mathrm{mg}, 43 \%$ ). $R_{\mathrm{f}}=0.10$ ( $\mathrm{PE} / \mathrm{EtOAc}$ 2:1). RP-HPLC (gradient 1, 220 nm ): 99.2\% ( $t_{\mathrm{R}}=19.25 \mathrm{~min}, k=5.64$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, [ $\mathrm{D}_{6}$ ]DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm})$ 0.84-0.87 (m, 3H), 1.22-1.30 (m, 6H), 1.41-1.44 (m, 2H), 3.09-3.13 (m, 2H), $7.15(\mathrm{~s}, 1 \mathrm{H}), 7.23-7.24(\mathrm{~m}, 1 \mathrm{H}), 7.49(\mathrm{t}, 1 \mathrm{H}, J 7.93 \mathrm{~Hz}), 7.59(\mathrm{t}, 1 \mathrm{H}, J 5.50 \mathrm{~Hz})$, $7.73(\mathrm{~m}, 1 \mathrm{H}), 7.79(\mathrm{~d}, 1 \mathrm{H}, 7.89 \mathrm{~Hz}), 8.65(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.97(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 10.26(\mathrm{~s}, 1 \mathrm{H}), 10.80(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-$ NMR ( $\left.150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 13.9,22.0,25.9,28.8,30.9,1 \mathrm{C}$ under solvent peak (38.7-40.3), 102.9, 122.9, 124.76, 124.80, 130.1, 134.0, 135.5, 147.0, 153.4, 153.6, 168.7. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{~N}_{6} \mathrm{OS}^{+}: 361.1805$, found: 361.1813. $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{6} \mathrm{OS}$. $\mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(360.48+228.05)$.

## 4-(3-(3-(Cyclohexylcarbamoyl)guanidino)phenyl)-2-aminothiazole (6.62)

bis(hydrotrifluoroacetate)
6.62 was prepared from 6.15 ( $90 \mathrm{mg}, 0.29 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.7(55 \mathrm{mg}, 0.29 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(155$ $\mathrm{mg}, 0.57 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $87 \mathrm{mg}, 0.86 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-50:50 in 25 min ) afforded the Boc-protected intermediate 6.41 as a yellow oil ( 90 mg , $69 \%) .90 \mathrm{mg}(0.20 \mathrm{mmol})$ of 6.41 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 20:80-50:50, $t_{\mathrm{R}}=$ 11.8 min ) afforded the product as white hygroscopic solid ( $70 \mathrm{mg}, 61 \%$ ). $\mathrm{Mp}: 80-84{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.05$ (PE /EtOAc 2:1). RP-HPLC (gradient 2, 220 nm ): $99.5 \%\left(t_{\mathrm{R}}=16.34 \mathrm{~min}, k=4.6\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}$, [ $\mathrm{D}_{6}$ ]DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm})$ 1.14-1.32 (m, 5H), 1.51-1.54 (m, 1H), 1.63-1.66 (m, 2H), 1.77-1.80 (m, 2H), 3.45-3.51 (m, 1H), $7.15(\mathrm{~s}, 1 \mathrm{H}), 7.23-7.25(\mathrm{~m}, 1 \mathrm{H}), 7.50(\mathrm{t}, 1 \mathrm{H}, J 7.91 \mathrm{~Hz}), 7.59$ (d, 1H, J 7.58 Hz ), $7.73(\mathrm{~m}, 1 \mathrm{H}), 7.79-7.81(\mathrm{~m}, 1 \mathrm{H}), 8.65(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.96(\mathrm{br} s, 1 \mathrm{H}), 9.93(\mathrm{~s}, 1 \mathrm{H})$, $10.74(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 24.6,25.4,32.5,48.9$, 103.5, 123.6, 125.3, 125.5, 130.6, 134.4, 136.0, 147.4, 153.1, 154.1, 169.2. HRMS (ESI) $\mathrm{m} / \mathrm{z}$ $(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{23} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 359.1649, found: 359.1649. $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(358.46+$ 228.05).

## 4-(3-(3-(Phenylcarbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.63)

6.63 was prepared from 6.16 ( $81 \mathrm{mg}, 0.26 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.7(50 \mathrm{mg}, 0.26 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(142$ $\mathrm{mg}, 0.52 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $79 \mathrm{mg}, 0.78 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-50:50 in 30 min ) afforded the Boc-protected intermediate 6.42 as a yellow oil ( 30 mg , $25 \%) .30 \mathrm{mg}(0.07 \mathrm{mmol})$ of 6.42 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85-70:30, $t_{\mathrm{R}}=$ 13.0 min ) afforded the product as white hygroscopic solid ( $27 \mathrm{mg}, 71 \%$ ). $R_{\mathrm{f}}=0.4\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N}\right.$ $\mathrm{NH}_{3}$ in MeOH 9:1). RP-HPLC (gradient 1, 220 nm ): $99.4 \%\left(t_{\mathrm{R}}=16.70 \mathrm{~min}, k=4.76\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600$ $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 7.10-7.14(\mathrm{~m}, 2 \mathrm{H}), 7.26-7.28(\mathrm{~m}, 1 \mathrm{H}), 7.34-7.37$ $(\mathrm{m}, 2 \mathrm{H}), 7.45-7.46(\mathrm{~m}, 2 \mathrm{H}), 7.51(\mathrm{t}, 1 \mathrm{H}, \mathrm{J} 7.84 \mathrm{~Hz}), 7.77-7.78(\mathrm{~m}, 1 \mathrm{H}), 7.81-7.83(\mathrm{~m}, 1 \mathrm{H}), 8.76-8.92$ (m, 3H), $9.92(\mathrm{~s}, 1 \mathrm{H}), 10.10(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 10.70(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]\right.$ DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 102.9,119.5,122.9,124.0,124.6,124.8,129.1,130.1,134.0,136.1,137.3,148.0$, 151.1, 153.3, 168.5. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 353.1179, found: 353.1180. $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(352.41+228.05)$.

## 4-(3-(3-((Phenylmethyl)carbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.64)

6.64 was prepared from 6.17 ( $85 \mathrm{mg}, 0.26 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.7(50 \mathrm{mg}, 0.26 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(142$ $\mathrm{mg}, 0.52 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $79 \mathrm{mg}, 0.78 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(8 \mathrm{~mL})$
according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-60:40 in 30 min ) afforded the Boc-protected intermediate 6.43 as a yellow oil ( 110 mg , $90 \%$ ). $100 \mathrm{mg}(0.21 \mathrm{mmol})$ of 6.43 were dissolved in a mixture of TFA ( 1 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred for 12 h at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85-70:30, $t_{\mathrm{R}}=$ 13.1 min ) afforded the product as white hygroscopic solid ( $90 \mathrm{mg}, 70 \%$ ). $R_{\mathrm{f}}=0.1$ ( $\mathrm{PE} / \mathrm{EtOAc} 2: 1$ ). RP-HPLC (gradient 1, 220 nm ): $99.1 \%\left(t_{\mathrm{R}}=16.57 \mathrm{~min}, k=4.71\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}, \mathrm{MeOD}, \mathrm{COSY}$, HSQC, HMBC): $\delta(\mathrm{ppm}) 4.41(\mathrm{~s}, 2 \mathrm{H}), 7.25-7.28(\mathrm{~m}, 1 \mathrm{H}), 7.32-7.35(\mathrm{~m}, 4 \mathrm{H}), 7.40-7.41(\mathrm{~m}, 1 \mathrm{H}), 7.60$ ( $\mathrm{t}, 1 \mathrm{H}, J 7.81 \mathrm{~Hz}$ ), 7.71-7.72 (m, 1H), 7.75-7.77 (m, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}, \mathrm{MeOD}, \mathrm{COSY}, \mathrm{HSQC}$, HMBC): $\delta(\mathrm{ppm}) 44.6,104.7,124.8,127.1,127.9,128.5,128.6,129.7,132.3,133.3,135.4,139.4$, 142.1, 155.3, 155.9, 172.6. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{~N}_{6} \mathrm{OS}^{+}: 367.1336$, found: 367.1341. $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(366.44+228.05)$.

## 4-(3-(3-((2-Phenylethyl)carbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.65)

6.65 was prepared from 6.18 ( $88 \mathrm{mg}, 0.26 \mathrm{mmol}, 1 \mathrm{eq}$ ), $6.7(50 \mathrm{mg}, 0.26 \mathrm{mmol}, 1 \mathrm{eq}), \mathrm{HgCl}_{2}(142$ $\mathrm{mg}, 0.52 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $79 \mathrm{mg}, 0.78 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-60:40 in 30 min ) afforded the Boc-protected intermediate 6.44 as a yellow solid ( 80 mg , $63 \%$ ). 80 mg ( 0.17 mmol ) of 6.44 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85-70:30, $t_{\mathrm{R}}=$ 13.9 min ) afforded the product as white hygroscopic solid ( $50 \mathrm{mg}, 50 \%$ ). $R_{\mathrm{f}}=0.3\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.75 \mathrm{~N}\right.$ $\mathrm{NH}_{3}$ in MeOH 9:1). RP-HPLC (gradient 1, 220 nm ): $99.8 \% ~\left(t_{\mathrm{R}}=17.32 \mathrm{~min}, k=4.97\right.$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (600 $\left.\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 2.77(\mathrm{t}, 2 \mathrm{H}, ~ J 7.16 \mathrm{~Hz}), 3.36-3.39(\mathrm{~m}, 2 \mathrm{H}), 7.13(\mathrm{~s}$, $1 \mathrm{H}), 7.20-7.23(\mathrm{~m}, 4 \mathrm{H}), 7.29-7.31(\mathrm{~m}, 2 \mathrm{H}), 7.48(\mathrm{t}, 1 \mathrm{H}, J 7.95 \mathrm{~Hz}), 7.58(\mathrm{t}, 1 \mathrm{H}, J 5.59 \mathrm{~Hz}), 7.72-7.73$ $(\mathrm{m}, 1 \mathrm{H}), 7.79-7.80(\mathrm{~m}, 1 \mathrm{H}), 8.62(\mathrm{br} s, 1 \mathrm{H}), 8.96(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 10.06(\mathrm{~s}, 1 \mathrm{H}), 10.67(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $\left.150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 34.9,40.7,102.9,122.9,124.7,124.8,126.3$, $128.4,128.7,130.1,133.9,135.9,138.8,147.6,153.40,153.44,168.5$. HRMS (ESI) $m / z(M+H)^{+}$ calcd. for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{~N}_{6} \mathrm{OS}^{+}: 381.1492$, found: 381.1493. $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(380.47+228.05)$.

## 6-(3-(Hexylcarbamoyl)guanidino)-2-amino-4,5,6,7-tetrahydrobenzo[d]thiazole bis(hydrotrifluoroacetate) (6.66)

6.66 was prepared from 6.14 ( $140 \mathrm{mg}, 0.44 \mathrm{mmol}, 1 \mathrm{eq}$ ), 6.12 ( $90 \mathrm{mg}, 0.53 \mathrm{mmol}, 1.2 \mathrm{eq}$ ), $\mathrm{HgCl}_{2}$ ( $239 \mathrm{mg}, 0.88 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $134 \mathrm{mg}, 1.32 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(11$ $\mathrm{mL})$ according to the general procedure. Purification by automated flash chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ $/ \mathrm{MeOH}$ 100:0-85:15 in 30 min ) afforded the Boc-protected intermediate 6.45 as a yellow solid ( $110 \mathrm{mg}, 57 \%$ ). $110 \mathrm{mg}(0.25 \mathrm{mmol})$ of 6.45 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 10:90$60: 40, t_{\mathrm{R}}=12.0 \mathrm{~min}$ ) afforded the product as white hygroscopic solid ( $46 \mathrm{mg}, 33 \%$ ). Mp: 121-
$122{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.2\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.7 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\left.\mathrm{MeOH} 9: 1\right)$. RP-HPLC (gradient 1, 220 nm ): 95.3\% ( $t_{\mathrm{R}}=17.10$ $\min , k=4.90) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): \delta(\mathrm{ppm}) 0.84-0.86(\mathrm{~m}, 3 \mathrm{H})$, 1.24-1.29(m, 6H), 1.39-1.42 (m, 2H), 1.83-1.97 (m, 2H), 2.51-2.60 (m, 3H), 2.86-2.89 (m, 1H), 3.06$3.09(\mathrm{~m}, 2 \mathrm{H}), 4.04-4.05(\mathrm{~m}, 1 \mathrm{H}), 7.50(\mathrm{~s}, 1 \mathrm{H}), 8.67(\mathrm{br} s, 4 \mathrm{H}), 9.07(\mathrm{~s}, 1 \mathrm{H}), 10.17(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 14.3,21.7,22.5,26.3,26.8,28.4,29.3,31.3$, 1 C under solvent peak (38.7-40.3), 46.9, 112.0, 136.5, 153.7, 154.2, 168.9. HRMS (ESI) $\mathrm{m} / \mathrm{z}(M+\mathrm{H})^{+}$ calcd. for $\mathrm{C}_{15} \mathrm{H}_{27} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 339.1962, found: 339.1962. $\mathrm{C}_{15} \mathrm{H}_{26} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}$ (338.47+228.05).

## 6-(3-((Phenylmethyl)carbamoyl)guanidino)-2-amino-4,5,6,7-tetrahydrobenzo[d]thiazole bis(hydrotrifluoroacetate) (6.67)

6.67 was prepared from 6.17 ( $192 \mathrm{mg}, 0.59 \mathrm{mmol}, 1 \mathrm{eq}), 6.12(110 \mathrm{mg}, 0.65 \mathrm{mmol}, 1.1 \mathrm{eq}), \mathrm{HgCl}_{2}$ ( $320 \mathrm{mg}, 1.18 \mathrm{mmol}, 2 \mathrm{eq}$ ) and triethylamine ( $180 \mathrm{mg}, 1.77 \mathrm{mmol}, 3 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(16$ mL ) according to the general procedure. Purification by automated flash chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right.$ 100:0-85:15 in 30 min$)$ afforded the Boc-protected intermediate 6.46 as a yellow solid ( $80 \mathrm{mg}, 34 \%$ ). $80 \mathrm{mg}(0.18 \mathrm{mmol})$ of 6.46 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1\% aq. TFA 15:85$60: 40, t_{\mathrm{R}}=11.2 \mathrm{~min}$ ) afforded the product as white solid ( $16 \mathrm{mg}, 16 \%$ ). $\mathrm{Mp}: 140-145{ }^{\circ} \mathrm{C} . R_{\mathrm{f}}=0.2$ ( $\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.7 \mathrm{~N} \mathrm{NH}_{3}$ in MeOH 9:1). RP-HPLC (gradient 1, 220 nm ): $97.0 \%\left(t_{\mathrm{R}}=14.40 \mathrm{~min}, k=3.97\right.$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 1.83-1.89(\mathrm{~m}, 1 \mathrm{H}), 1.95-1.97(\mathrm{~m}, 1 \mathrm{H})$, 2.53-2.60 (m, 3H), 2.86-2.89 (m, 1H), 4.03-4.05 (m, 1H), 4.30 (d, $2 \mathrm{H}, J 5.70 \mathrm{~Hz}), 7.24-7.28(\mathrm{~m}, 3 \mathrm{H})$, 7.32-7.34 (m, 2H), 8.02 (br s, 1H), 8.74 (br s, 4H), 9.07 (br s, 1H), 10.36 (br s, 1H). ${ }^{13} \mathrm{C}-\mathrm{NMR}(150$ $\mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}$ ): $\delta(\mathrm{ppm}) 21.2,26.3,27.9,42.7,46.4,111.5,127.1,127.2$, 128.4, 138.6, 140.2, 153.2, 153.8, 168.5. HRMS (ESI) $m / z(M+H)^{+}$calcd. for $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{~N}_{6} \mathrm{OS}^{+}$: 345.1492, found: 345.1485. $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{OS} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(344.44+228.05)$.

## 1-(Amino[3-(2-aminothiazol-4-yl)phenylamino]methylene)-3-(6-[3-(amino[3-(2-aminothiazol-4$\mathrm{yl}) \mathrm{phenylamino}$ ]methylene)ureido]hexyl)urea tetra(hydrotrifluoracetate) (6.70)

6.70 was prepared from $3.33(130 \mathrm{mg}, 0.24 \mathrm{mmol}, 1 \mathrm{eq}), 6.7(136 \mathrm{mg}, 0.71 \mathrm{mmol}, 3 \mathrm{eq}), \mathrm{HgCl}_{2}$ ( $143 \mathrm{mg}, 0.53 \mathrm{mmol}, 2.2 \mathrm{eq}$ ) and DIPEA ( $155 \mathrm{mg}, 1.20 \mathrm{mmol}, 5 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 mL ) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 80:20-20:80 in 30 min ) afforded the Boc-protected intermediate 6.68 as a yellow foam ( 180 mg , $91 \%) .160 \mathrm{mg}(0.19 \mathrm{mmol})$ of 6.68 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: $0-30 \mathrm{~min}$ : MeCN/0.1\% aq. TFA 15:85-45:55, $t_{\mathrm{R}}=$ 13.4 min ) afforded the product as white solid ( $140 \mathrm{mg}, 67 \%$ ). Mp: $128-133{ }^{\circ} \mathrm{C}$. $R_{f}=0.1$ (PE/EtOAc 2:1). RP-HPLC (gradient 1, 220 nm ): $97.3 \%\left(t_{\mathrm{R}}=14.99 \mathrm{~min}, k=4.17\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}(600 \mathrm{MHz}, \mathrm{MeOD}$, COSY, HSQC, HMBC): $\delta(\mathrm{ppm}) 1.38-1.41(\mathrm{~m}, 2 \mathrm{H}), 1.56-1.58(\mathrm{~m}, 2 \mathrm{H}), 3.24(\mathrm{t}, 2 \mathrm{H}, J 7.03 \mathrm{~Hz}), 7.35-$ $7.41(\mathrm{~m}, 1 \mathrm{H}), 7.60(\mathrm{t}, 1 \mathrm{H}, J 7.95 \mathrm{~Hz}), 7.71(\mathrm{~m}, 1 \mathrm{H}), 7.75-7.76(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}, \mathrm{MeOD}$, COSY, HSQC, HMBC): $\delta$ (ppm) 27.4, 30.3, 40.7, 104.6, 124.7, 127.0, 127.9, 132.1, 133.2, 135.5, 142.0, 155.3, 155.9, 172.6. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{28} \mathrm{H}_{35} \mathrm{~N}_{12} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+}$: 635.2442, found: 635.2443. $\mathrm{C}_{28} \mathrm{H}_{34} \mathrm{~N}_{12} \mathrm{O}_{2} \mathrm{~S}_{2} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(634.78+228.05)$.

## 1-(Amino[3-(1H-pyrazol-4-yl)propylamino]methylene)-3-(6-[3-(amino[3-(1H-pyrazol-4yl)propylamino]methylene)ureido]hexyl)urea tetra(hydrotrifluoracetate) (6.71)

6.71 was prepared from 3.33 ( $129 \mathrm{mg}, 0.24 \mathrm{mmol}, 1 \mathrm{eq}$ ), 3 -( 1 -trityl-1 H -pyrazol-4-yl)prop-1-ylamine ( $260 \mathrm{mg}, 0.71 \mathrm{mmol}, 3 \mathrm{eq}$ ), $\mathrm{HgCl}_{2}(143 \mathrm{mg}, 0.53 \mathrm{mmol}, 2.2 \mathrm{eq}$ ) and DIPEA ( $155 \mathrm{mg}, 1.20$ $\mathrm{mmol}, 5 \mathrm{eq}$ ) dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ according to the general procedure. Purification by automated flash chromatography ( $\mathrm{PE} / \mathrm{EtOAc}$ 100:0-70:30 in 20 min ) afforded the Boc-protected intermediate 6.69 as a white foam ( $230 \mathrm{mg}, 81 \%$ ). $190 \mathrm{mg}(0.16 \mathrm{mmol})$ of 6.69 were dissolved in a mixture of TFA ( 2 mL ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min : $\mathrm{MeCN} / 0.1 \%$ aq. TFA 10:90-55:45, $t_{\mathrm{R}}=13.8 \mathrm{~min}$ ) afforded the product as white solid ( 80 mg , $52 \%$ ). Mp: $84-88{ }^{\circ} \mathrm{C}$. $R_{\mathrm{f}}=0.1\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / 1.7 \mathrm{~N} \mathrm{NH}_{3}\right.$ in $\mathrm{MeOH} 9: 1$ ). RP-HPLC (gradient 2, 220 nm ): $97.3 \%\left(t_{R}=12.84 \mathrm{~min}, k=3.43\right) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}, \mathrm{COSY}, \mathrm{HSQC}, \mathrm{HMBC}\right): ~ \delta(\mathrm{ppm}) 1.26$ (br s, 4H), 1.41-1.43 (m, 4H), 1.76 (qui, $4 \mathrm{H}, \mathrm{J} 7.25 \mathrm{~Hz}$ ), $2.46(\mathrm{t}, 4 \mathrm{H}, \mathrm{J} 7.60 \mathrm{~Hz}$ ), 3.06-3.09 (m, 4H), 3.21-3.24 (m, 4H), $7.49(\mathrm{~s}, 6 \mathrm{H}), 8.47$ (br s, 4H), $9.00(b r s, 2 \mathrm{H}), 10.34(\mathrm{br} \mathrm{s}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(150 \mathrm{MHz}$, [D6]DMSO, COSY, HSQC, HMBC): $\delta(\mathrm{ppm})$ 20.5, 25.9, 28.9, 29.2, 1C under solvent peak (38.740.3), 40.4, 118.7, 132.2, 153.78, 153.82. HRMS (ESI) $\mathrm{m} / \mathrm{z}(\mathrm{M}+\mathrm{H})^{+}$calcd. for $\mathrm{C}_{22} \mathrm{H}_{39} \mathrm{~N}_{12} \mathrm{O}_{2}^{+}$: 503.3313, found: 503.3309. $\mathrm{C}_{22} \mathrm{H}_{38} \mathrm{~N}_{12} \mathrm{O}_{2} \cdot \mathrm{C}_{4} \mathrm{H}_{2} \mathrm{~F}_{6} \mathrm{O}_{4}(502.63+228.05)$.

## Investigation of the chemical stability Claudia Honisch Master Thesis

The chemical stability of the compounds $6.49,6.50$ and 6.52 as well as the acylguanidines URBit22, UR-Bit23 and UR-Bit29 was investigated in PBS ( $\mathrm{pH}=7.4$ ) at room temperature. The incubation was started by dilution of a 10 mM solution of the compound in aqueous $\mathrm{HCl}(20 \mathrm{mM})$ with PBS ( pH 7.4 ), resulting in a $100 \mu \mathrm{M}$ solution. After $0 \mathrm{~h}, 72 \mathrm{~h}$ and 7 days a $20 \mu \mathrm{~L}$ aliquot was added to $20 \mu \mathrm{~L}$ of a mixture of aqueous TFA solution ( $1 \%$ )/ $/ \mathrm{H}_{2} \mathrm{O}_{\text {milipore }} / \mathrm{MeCN}(6: 3: 1, \mathrm{v}: \mathrm{v}: \mathrm{v}$ ). A sample ( $20 \mu \mathrm{~L}$ ) was immediately analyzed by analytical HPLC. A system from Agilent Technologies, composed of a 1290 Infinity binary pump equipped with a degasser, a 1290 Infinity autosampler, a 1290 Infinity thermostated column compartment and a 1260 Infinity diode array was used. A Kinetex XB-C18 100A ( $100 \times 3 \mathrm{~mm}, 2.6 \mu \mathrm{~m}$, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of $\mathrm{CH}_{3} \mathrm{CN}(\mathrm{A})$ and $0.05 \%$ aq. TFA (B) were used as mobile phase. Helium degassing, $25^{\circ} \mathrm{C}$, a flow rate of $0.5 \mathrm{~mL} / \mathrm{min}$ and a detection wavelength of 220 nm were used throughout. The following linear gradient was applied for the HPLC analysis: $0-15 \mathrm{~min}: A / B$ 5:95-35:65, 15-18 min: 35:65-95:5, 18-23 min: 95:5 (isocratic).

### 6.3.3 Pharmacological Methods

## Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of Sf 9 insect cells expressing the $h \mathrm{H}_{2} \mathrm{R}-\mathrm{G}_{s \alpha s}$ fusion protein or coexpressing the $h H_{1} R+R G S 4$, the $h H_{3} R+G_{i \alpha 2}+\beta_{1} \psi_{2}$ or $h H_{4} R+G_{i \alpha 2}+\beta_{1} \psi_{2}$ proteins was described elsewhere. ${ }^{30}$

Radioligand competition binding assays were performed as described previously with minor adjustments using the following radioligands: $\left.{ }^{3} \mathrm{H}\right]$ mepyramine (Hartmann Analytic, Braunschweig, Germany; specific activity $\left.=20 \mathrm{Ci} / \mathrm{mmol}^{2} \mathrm{hH}_{1} \mathrm{R}: K_{\mathrm{d}}=4.5 \mathrm{nM}, \mathrm{c}_{\text {final }}=5 \mathrm{nM}\right),\left[{ }^{3} \mathrm{H}\right] \mathrm{UR}-$ DE257 ${ }^{14}\left(\mathrm{hH}_{2} \mathrm{R}\right.$ : specific activity $\left.=32.89 \mathrm{Ci} / \mathrm{mmol}, K_{\mathrm{d}}=12.2 \mathrm{nM}, \mathrm{c}_{\text {final }}=20 \mathrm{nM}\right),\left[{ }^{3} \mathrm{H}\right] \mathrm{UR}-\mathrm{PI} 294^{31}$ (specific activity $=93.3 \mathrm{Ci} / \mathrm{mmol} ; \mathrm{hH}_{3} \mathrm{R}: K_{\mathrm{d}}=1.1 \mathrm{nM}, \mathrm{c}_{\text {final }}=2 \mathrm{nM} ; \mathrm{hH}_{4} \mathrm{R}: K_{\mathrm{d}}=5.1 \mathrm{nM}, \mathrm{c}_{\text {final }}=5 \mathrm{nM}$ ) and $\left[{ }^{3} \mathrm{H}\right.$ ]histamine (Hartmann Analytic, Braunschweig, Germany; specific activity $=25 \mathrm{Ci} / \mathrm{mmol}$; $\mathrm{hH}_{3} \mathrm{R}: K_{\mathrm{d}}=12.1 \mathrm{nM}, \mathrm{c}_{\text {final }}=15 \mathrm{nM}$; $\left.\mathrm{hH}_{4} \mathrm{R}: K_{\mathrm{d}}=15.9 \mathrm{nM}, \mathrm{c}_{\text {final }}=10 \mathrm{nM}\right)$.

On the day of the experiment Sf9 membranes were thawed and sedimented by centrifugation at $13,000 \mathrm{rpm}$ at $4{ }^{\circ} \mathrm{C}$ for 10 min . The membranes were resuspended in ice cold binding buffer ( 12.5 $\mathrm{mM} \mathrm{MgCl} 2,1 \mathrm{mM}$ EDTA and 75 mM Tris/ $\mathrm{HCl}, \mathrm{pH} 7.4$; in the following referred to as BB ) and adjusted to a protein concentration of $2-4 \mu \mathrm{~g} / \mu \mathrm{L} .80 \mu \mathrm{~L}$ BB containing $0.2 \%$ BSA and the respective radioligand, followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ ), were added to every well of a 96 -well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension ( $10 \mu \mathrm{~L}$ ). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with $0.3 \%$ polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with $200 \mu \mathrm{~L}$ of scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h. The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

## Functional GTPүS assay on Sf9 insect cell membranes

GTP $\gamma$ S assays were performed as described previously ${ }^{5}$ with minor modifications. $\left[{ }^{35}\right.$ S]GTP $\gamma$ S (specific activity $=1000 \mathrm{Ci} / \mathrm{mmol}$ ) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to 0.5-1.5 $\mu \mathrm{g} / \mu \mathrm{L}$.

Agonist mode: $80 \mu \mathrm{~L}$ of BB containing BSA ( $0.05 \%$ final), GDP ( $1 \mu \mathrm{M}$ final) and $\left[{ }^{35} \mathrm{~S}\right] \mathrm{GTP} \gamma \mathrm{S}(20 \mathrm{nCi}$ final), followed by $10 \mu \mathrm{~L}$ of the investigated ligands at various concentrations (dissolved in $\mathrm{H}_{2} \mathrm{O}$ ) were added to every well of a 96 -well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension (10 $\mu \mathrm{L}$ ). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters
(Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTP $\gamma$ S assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine ( $1 \mu \mathrm{M}$ final).

## Cell culture

The preparation of stably transfected HEK293T-hH2R- $\beta$ Arr2 ${ }^{15,32}$ cells was described elsewhere.

Cells were cultivated at $37{ }^{\circ} \mathrm{C}$ in a water saturated atmosphere containing $5 \% \mathrm{CO}_{2}$. Dulbecco's Modified Eagle Medium, containing $4.5 \mathrm{~g} / \mathrm{L}$ glucose, $3.7 \mathrm{~g} / \mathrm{L} \mathrm{NaHCO}_{3}, 110 \mathrm{mg} / \mathrm{L}$ sodium pyruvate (DMEM, Sigma-Aldrich Munich, Germany) and supplemented with $0.584 \mathrm{~g} / \mathrm{L}$ L-glutamine (Lglutamine solution, Sigma-Aldrich Munich, Germany), 1\% (v/v) penicillin-streptomycin (P/S, $10,000 \mathrm{U} / \mathrm{mL}$, Sigma-Aldrich Munich, Germany), $10 \%$ ( $\mathrm{v} / \mathrm{v}$ ) fetal calf serum (FCS, Biochrom GmbH, Merck, Berlin, Germany) were used as a culture medium. Additionally, $400 \mu \mathrm{~g} / \mathrm{mL}$ zeocin (InvivoGen, San Diego,USA) and $600 \mu \mathrm{~g} / \mathrm{mL}$ G418 were added to the culture medium of HEK293T$h H_{2} R-\beta A r r 2$ cells.

## $\beta$-Arrestin2 recruitment assay

The $\beta$-Arrestin2 recruitment assays were performed as described previously for the $H_{1} R$ using HEK293T-h $\mathrm{H}_{2} \mathrm{R}-\beta$ Arr2 cells, stably expressing the $\mathrm{hH}_{2} \mathrm{R}$-ElucC and $\beta$ Arr2-ElucNfusion constructs ${ }^{32}$.

One day prior to the experiment, HEK293T-hH2R- 3 Arr2 cells were trypsinized and detached with DMEM medium (high glucose without phenol red (Sigma Aldrich, Munich, Germany) containing $1 \%(v / v) P / S$ and $5 \%(v / v)$ FCS. The cell suspension was adjusted to $1.1 \cdot 10^{6}$ cells $/ \mathrm{mL}$ and $90 \mu \mathrm{l}$ (100,000 cells/well) were seeded in every well of a sterile, luciferase assay compatible, F-bottom 96 -well plate (Cellstar ${ }^{\circledR}$, Greiner Bio-One, Kremsmünster, Österreich). The cells were cultivated at $37{ }^{\circ} \mathrm{C}$ overnight in a water saturated atmosphere containing $5 \% \mathrm{CO}_{2}$. The investigated ligands were added at increasing concentrations ( $10 \mu \mathrm{~L}$ ), and the plate was incubated at $25^{\circ} \mathrm{C}$ for 60 min under shaking. $50 \mu \mathrm{~L}$ of the medium were removed, and $50 \mu \mathrm{~L}$ of Bright-Glo reagent (Promega, Madison, USA) were added. Bioluminescence was immediately measured for 1 s per well using a GENios Pro microplate reader (Tecan, Salzburg, Austria).

## Radioligand competition binding assay on homogenates from HEK293T cells expressing the $\mathrm{hD}_{\text {2long }} \mathrm{R}$ or $\mathrm{hD}_{3} \mathrm{R}$

The preparation of the cells (HEK293T-CRE-Luc-hD 2long $R$ and HEK293T-CRE-Luc-hD ${ }_{3} R$ cells), their cultivation, the preparation of the homogenates and radioligand competition binding assays were performed by Lisa Foster as a part of her ongoing dissertation.

Radioligand competition binding assays were performed as described for Sf9 cell membranes with some adjustments using the radioligand $\left[{ }^{3} \mathrm{H}\right] \mathrm{N}$-methylspiperone (specific activity of 77
$\mathrm{Ci} / \mathrm{mmol}$, Novandi Chemistry AB, Södertälje, Sweden; hD 2long R : $K_{\mathrm{d}}=0.014 \mathrm{nM}, \mathrm{C}_{\text {final }}=0.05 \mathrm{nM}$; $\left.h D_{3} R: K_{\mathrm{d}}=0.026 \mathrm{nM}, \mathrm{C}_{\text {final }}=0.05 \mathrm{nM}\right)$.

Cell lines expressing the human $D_{2 l o n g} R$ and $D_{3} R$ were grown in 150 mm dishes to $80-90 \%$ confluency. Cells were rinsed with ice-cold PBS and scraped from the dishes using a cell scraper in the presence of harvest buffer ( 10 mM Tris $\mathrm{HCl}, 0.5 \mathrm{mM}$ EDTA, $5.5 \mathrm{mM} \mathrm{KCl}, 140 \mathrm{mM} \mathrm{NaCl}$; pH 7.4) supplemented with protease inhibitors. After centrifugation ( $500 \mathrm{~g}, 5 \mathrm{~min}$ ), the $\mathrm{D}_{2 \text { long }} R$ expressing cells were resuspended in homogenate buffer ( 50 mM Tris $\cdot \mathrm{HCl}, 5 \mathrm{mM}$ EDTA, 1.5 mM $\mathrm{CaCl}_{2}, 5 \mathrm{mM} \mathrm{MgCl} 2,5 \mathrm{mM} \mathrm{KCl}, 120 \mathrm{mM} \mathrm{NaCl} ; \mathrm{pH} 7.4$ ) and the $\mathrm{D}_{3} \mathrm{R}$ expressing cells were resuspended in Tris- $\mathrm{MgSO}_{4}$ buffer ( 10 mM Tris $\cdot \mathrm{HCl}, 5 \mathrm{mM} \mathrm{MgSO} \mathrm{M}_{4} ; \mathrm{pH} 7.4$ ) and stored at $-80^{\circ} \mathrm{C}$. After thawing, the cells were resuspended in homogenate buffer or Tris- $\mathrm{MgSO}_{4}$ buffer respectively and homogenized using an Ultraturrax ( 5 times for 5 s on ice). The homogenate was centrifuged ( $50000 \mathrm{~g}, 6^{\circ} \mathrm{C}, 15 \mathrm{~min}$ ), the pellet resuspended in binding buffer ( 50 mM Tris $\cdot \mathrm{HCl}$, $\mathrm{pH}, 7.4$, containing 1 mM EDTA, $5 \mathrm{mM} \mathrm{MgCl} 2,100 \mu \mathrm{~g} / \mathrm{mL}$ bacitracin;) and homogenized using a syringe and needle. The homogenate was stored in small aliquots at $-80^{\circ} \mathrm{C}$.

For radioligand binding, to $80 \mu \mathrm{~L}$ of the homogenate suspension, $10 \mu \mathrm{~L}$ of radioligand and $10 \mu \mathrm{~L}$ of the ligand at various concentrations (dissolved in binding buffer), were added to every well of a 96 -well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Unspecific binding was determined using (+)-butaclamol at a final concentration of $2 \mu \mathrm{M}$, instead of the ligand. The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with $0.3 \%$ polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96 -well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with $200 \mu \mathrm{~L}$ of scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h . The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

### 6.3.4 Data analysis

Retention factors $k$ were calculated according to $k=\left(t_{R^{-}} t_{0}\right) / t_{0}\left(t_{0}=\right.$ dead time $)$.
Total binding data from radioligand competition binding experiments were plotted against $\log$ (concentration competitor) and analyzed by a four-parameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism Software 5.0, San Diego, CA), followed by normalization $(100 \%=$ "top" of the four-parameter logistic fit, $0 \%=$ unspecifically bound radioligand/ fluorescent ligand determined in the presence of famotidine at $100 \mu \mathrm{M}$ ). Normalized data from competition binding experiments was again analyzed by a four-parameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism) and obtained $\mathrm{p} / C_{50}$ values were converted into $\mathrm{p} K_{\mathrm{i}}$ values according to the Cheng-Prusoff equation ${ }^{33}$.

Data of the GTP $\gamma$ S assay (agonist mode) were processed by plotting the corrected counts per minute (ccpm) against log(concentration). In most cases data analysis by bell-shaped fit was ambiguous due to lack of data points at high concentrations (>100 $\mu \mathrm{M}$ ). The data points at high concentrations ( $\geq 10-100 \mu \mathrm{M}$ ), where the signal decreased again, were excluded in the analysis by a four parameter logistic equation (GraphPad Software). The concentration response curve was normalized ( $0 \%=$ water value (basal activity), $100 \%=$ "top" histamine equation) and again analyzed by a four-parameter logistic equation (log(agonist) vs. response - variable slope, GraphPad Prism).

Data from the GTP S assay (antagonist mode) were processed by plotting the ccpm against $\log$ (concentration) and analysis by a four parameter logistic equation (GraphPad Prism), followed by normalization ( $100 \%=$ "top" of the four-parameter logistic fit, $0 \%=$ unspecifically bound $\left[{ }^{35} \mathrm{~S}\right]$ GTP $\gamma$ S (ccpm) determined in the presence of famotidine at $100 \mu \mathrm{M}$ ) and analysis by fourparameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism). p/C $C_{50}$ values were converted into $\mathrm{p} K_{\mathrm{B}}$ values according to the Cheng-Prusoff equation ${ }^{33}$.

Data of the beta-Arrestin2 recruitment assay (agonist mode) were processed by plotting the luminescence (RLU) against $\log$ (concentration). In all cases data analysis by bell-shaped fit was ambiguous due to lack of data points at high concentrations (>300 $\mu \mathrm{M}$ ). The data points at high concentrations ( $\geq 10-100 \mu \mathrm{M}$ ), where the signal decreased again, were excluded in the analysis by a four parameter logistic equation (GraphPad Prism). The concentration response curve was normalized ( $0 \%=$ water value (basal activity), $100 \%=$ "top" histamine equation) and again analyzed by a four-parameter logistic equation (log(agonist) vs. response - variable slope, GraphPad Prism).

### 6.4 Summary and Conclusion

The bioisosteric replacement of the acylguanidine moiety in the monomeric $\mathrm{N}^{6}$-acylated amino(methyl)thiazolylpropylguanidines by a carbamoylguanidine moiety resulted in ligands with retained or even improved $\mathrm{hH}_{2} \mathrm{R}$ potency with $\mathrm{pEC} C_{50}$ values of 6.3-7.76 (ligands 6.47-6.59). A variety of aliphatic and aromatic residues were well tolerated. Compounds containing an amino(methyl)thiazolyl propyl moiety achieved up to 80 fold the potency of histamine and partial to full agonistic activities with more or less pronounced bell-shaped concentration-response curves. Additionally, the ligands 6.53, 6.55, 6.58 and 6.59 were investigated for dopamine receptor affinity ( $h D_{\text {2long }} R$ and $h D_{3} R$ ). These ligands showed affinity to the $h D_{2 \text { long }} R\left(p K_{i}\right.$ : 6.28-6.58) and $\mathrm{hD}_{3} \mathrm{R}$ ( $\mathrm{p} \mathrm{K}_{\mathrm{i}}$ : 6.88-7.36). There was a clear preference for the $\mathrm{hH}_{2} \mathrm{R}$ over the $\mathrm{hD} \mathrm{D}_{\text {2long }}$ R. But only ligand 6.59 showed a clear preference towards the $h_{2} R$ over the $h D_{3} R$ and the ligands 6.53, 6.55 and 6.58 bound non-selective to both receptors. Incorporation of the less flexible 2 -amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety, derived from pramipexole, resulted in partial (6.67, $\mathrm{E}_{\text {max }}: 0.53, \mathrm{pEC} C_{50}: 6.7$ ) or in weak partial agonism ( $6.66, \mathrm{E}_{\text {max }}: 0.16, \mathrm{pEC} C_{50}: 6.83$ ) and bell-shaped concentration-response curves. Interestingly, these two ligands showed a decreased $\mathrm{hD}_{2 \text { long }} \mathrm{R}$ and $\mathrm{hD}_{3} \mathrm{R}$ affinity compared to pramipexole. Incorporation of an aminothiazolylphenyl moiety (compounds $6.60-6.65$ ) resulted in weak antagonism at the $\mathrm{hH}_{2} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{b}}: 4.8-6.14$ ). Replacement of the amino(methyl)thiazolepropyl moiety of UR-NK22 with either a (aminothiazolyl)phenyl (6.70) or a pyrazolylpropyl moiety (6.71) resulted in strongly decreased $\mathrm{hH}_{2} \mathrm{R}$ affinity and antagonistic or partial agonistic activity.

All full or partial agonists identified in the GTP $\gamma$ S assay showed a lower potency and efficacy in the $\beta$ Arrestin2 recruitment assay. The ligands 6.53-6.59 also showed in the $\beta$ Arrestin2 recruitment assay more or less pronounced bell-shaped concentration-response curves. While in case of the GTP $\gamma$ S assay plausible explanations would be binding through multiple binding sites or direct interaction with the $G_{\text {sas }}$ subunit of the fusion protein, in the $\beta$ Arrestin2 recruitment assay also cytotoxicity could be the reason.

High affinity ligands like $6.48,6.51$ and 6.55 were highly selective for $h H_{2} R$ over the other subtypes (Figure 6.11). The investigation of the binding affinity of aminothiazole containing ligands like 6.55 to the $h D_{2 \text { long }} R$ and the $h D_{3} R$ showed that achieving selectivity over these dopamine receptors might be a challenge. Notwithstanding, the monomeric $N^{G}$-carbamoylated amino(methyl)thiazolyl propylguanidines represent a good alternative to monomeric acylguanidines and dimeric ligands with high stability against hydrolytic cleavage. However, according to the current state of knowledge this is just speculation.


Figure 6.11. $N^{6}$-Carbamoylated 3-(2-amino-4-methylthiazol-5-yl)propylguanidines 6.48, $\mathbf{6 . 5 1}$ and $\mathbf{6 . 5 5}$ as high potency $\mathrm{H}_{2} \mathrm{R}$ agonists.

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Chapter 7

## SUMMARY

The $H_{2} R$, an aminergic GPCR, is one of four receptor subtypes $\left(H_{1} R, H_{2} R, H_{3} R, H_{4} R\right)$ which mediate the action of the biogenic amine histamine. Activation of $H_{2} R$ results e. g. in gastric acid secretion, ${ }^{1,2}$ positive inotropic and chronotropic effects ${ }^{3}$. In humans, the $H_{2} R$ is located on parietal cells in the stomach, ${ }^{2}$ in the brain, ${ }^{4,5}$ on neutrophiles and eosinophiles ${ }^{6}$ as well as on smooth muscle cells ${ }^{7}$. However, the (patho-) physical role of the $\mathrm{H}_{2} \mathrm{R}$, especially in the brain, is still far from being understood. Therefore this work aimed at the development of selective high affinity molecular tools for the $\mathrm{H}_{2} \mathrm{R}$, including agonists, antagonists and radiolabeled as well as fluorescent $\mathrm{H}_{2} \mathrm{R}$ ligands.

The number of high affinity tritiated radioligands for the $\mathrm{H}_{2} \mathrm{R}$ is very limited. Guanidinothiazole containing ligands such as famotidine or ICI127032 represent a class of surmountable $\mathrm{H}_{2} \mathrm{R}$ antagonists. ${ }^{8,9}$ The combination of the 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole structure derived from famotidine or the guanidino-4-(3-aminophenyl)thiazole structure derived from ICI127032 with a derivatized squaramide or a cyanoguanidine moiety ("urea equivalent") led to propionylated high affinity $\mathrm{H}_{2} \mathrm{R}$ antagonists. N -[8-(2-[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]propionamide (3.25) showed a $\mathrm{p} K_{\mathrm{i}}$ value of 7.65 at the $\mathrm{hH}_{2} \mathrm{R}$ and selectivity over the other subtypes (no affinity at the $\mathrm{hH}_{1} \mathrm{R}$, $\mathrm{hH} \mathrm{H}_{3} \mathrm{R}$ : $\mathrm{p} K_{\mathrm{i}}$ value of 5.3 and $\mathrm{hH}_{4} \mathrm{R}$ : $\mathrm{p} K_{\mathrm{i}}$ value of 4.4). The radiolabled form [ $\left.{ }^{3} \mathrm{H}\right] 3.25$ bound in a saturable manner ( $K_{\mathrm{d}}$ values of 15-22 nM) to membrane preparations of Sf9 cells and intact Hek cells both recombinantly expressing $\mathrm{hH}_{2}$ Rs. Although a part of $\left[{ }^{3} \mathrm{H}\right] 3.25$ bound in (pseudo)irreversible manner (plateau at 23\%), the kinetic $K_{d}$ value of 26 nM was comparable to that determined at equilibrium, and the radioligand $\left.{ }^{3} \mathrm{H}\right] 3.25$ was completely displacable by histamine, famotidine and ICI127032. However, the radioligand showed a radiochemical purity of only $87 \%$ and low stability in stock solution (radiochemical purity: $<45 \%$ after 15 month). Nevertheless, $\left[{ }^{3} \mathrm{H}\right] \mathbf{3 . 2 5}$ can be a valuable molecular tool provided that purity and stability under storage conditions are improvable.

Aminopotentidine and it's derivatives are reported as high affinity $\mathrm{H}_{2} \mathrm{R}$ antagonists. Iodination in 3 position of the 4-aminobenzoic acid amide moiety results in an enormous gain in affinity (iodoaminopotentidine). ${ }^{4,5}$ Aminopotentidine and its analogs with different substituents (e.g. iodine, bromine, chlorine, trifluoromethyl) in position 3, were prepared and propionylated. Within the series of propionylated derivatives the brominated ligand (4.37) and the iodinated ligand (4.38) showed the highest $\mathrm{hH}_{2} \mathrm{R}$ affinities ( $\mathrm{p} K_{\mathrm{i}}$ values of 8.5 and 8.18 ) along with excellent selectivities over the $\mathrm{hH}_{3} \mathrm{R}$ (6900- or 2500 -fold). In general, for radiosynthesis an excess of precursor compared to radioactive labeling reagent is used. However, an adjustment of the reaction conditions for radiosynthesis might be challenging due to the necessary high excess of the "cold" labeling reagent propionic chloride and the low yields in the "cold" reaction. A test reaction under radiosynthesis conditions failed. To overcome this problem, a series of aminopotentidine derivatives containing a functionalized (propionylated, acetylated or methylated) aminomethyl substituent in 4-position of the aromatic ring was prepared. The dimethylated 3 -bromo substituted ligand 4.50 showed the highest affinity within the series with a $\mathrm{p} K_{\mathrm{i}}$ value of 7.54 . The synthesis of radiolabeled 4.50 is accessible by dimethylation of 4.46 with [ $\left.{ }^{3} \mathrm{H}\right]$ methyl iodide.

Fluorescent ligands have become an attractive alternative to radioligands for the investigation of ligand-receptor interactions. Besides advantages with respect to safety issues and waste disposal,
fluorescent ligands are a prerequisite for the application of a plethora of optical techniques (confocal microscopy, FRET, ${ }^{10}$ FRAP, ${ }^{11}$ TIRF, ${ }^{12}$ high content imaging, ${ }^{13}$ fluorescence polarization ${ }^{14}$ ). In order to expand the range of applications and avoid the high cellular autofluorescence, fluorescent ligands labeled with red-emitting fluorophores (emission wavelength > 600 nm ) are required. Recently, a series of fluorescent ligands with a piperidinomethylphenoxypropylamino (potentidine) pharmacophore was reported. ${ }^{15}$ The most promising ligands within this series were the squaramide-type ligands UR-DE229 and UR-DE56 which contained pyridinium or a cyanine fluorophore. Aiming at fluorescent high affinity $\mathrm{H}_{2} \mathrm{R}$ antagonists with improved optical and physicochemical properties to gain access to a wide range of potential applications, the fluorescent labeled antagonists UR-DE229 (5.12) and UR-DE56 (5.18) were prepared and investigated in different assay systems (radioligand competition binding assay, GTP $\gamma \mathrm{S}$ binding assay, flow cytometric binding assay and high content imaging). Furthermore, a small library of fluorescent ligands was synthesized for the exploration of the impact of length of the alkyl linker and the net charge of the fluorophores by coupling the positively charged pyrilium dye (Py-5) or differently charged cyanine dyes with amine precursors by linkers, differing in length (number of atoms). The highest affinities to the $\mathrm{hH}_{2} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ values $>7.0$ ) in radioligand competition assays were obtained in case of the pyridinium labeled ligands 5.12-5.14 and the cyanine labeled ligands 5.16 (positively charged fluorophore, net charge: $2^{+}$) and 5.18 (electroneutral fluorophore, net charge: $1^{+}$). While the linker length (4-7 carbon atoms) had no significant influence on the $\mathrm{hH}_{2} \mathrm{R}$ affinity in case of the pyridinium ligands, the cyanine ligands with a hexyl linker ( 5.16 and 5.18 ) showed an increased $\mathrm{hH}_{2} \mathrm{R}$ affinity compared analogs, containing a butyl linker. Despite the low selectivity towards the $\mathrm{hH}_{3} \mathrm{R}$ the investigated fluorescent ligands proved to be useful tools for binding studies using different techniques (flow cytometry, high content imaging and confocal microscopy), when genetically engineered cells, expressing the $\mathrm{H}_{2} \mathrm{R}$ were used. Investigated ligands bound in a saturable manner to the $\mathrm{hH}_{2} \mathrm{R}$ (flow cytometry and high content imaging) and the determined $K_{d}$ values were in good agreement with the corresponding $K_{i}$ values from radiolligand binding experiments. $K_{d}(k i n)$ values calculated from kinetic experiments (flow cytometry or high content imaging) were consistent with the $K_{\mathrm{d}}$ values determined in saturation binding experiments even though they showed an incomplete dissociation (insurmountable antagonism). High content imaging and confocal microscopy showed that residual bound ligand was still located in the cell membrane. Nonetheless, the fluorescent ligands $\mathbf{5 . 1 4}$ and $\mathbf{5 . 1 8}$ also proved to be useful for the determination of binding affinities of unlabeled ligands in competition binding assays (flow cytometry and high content imaging).
$\mathrm{N}^{\mathrm{G}}$-acylated amino(methyl)thiazolepropylguanidines are reported as a class of potent and selective histamine $\mathrm{H}_{2} \mathrm{R}$ agonists, but the acylguanidine group is prone to hydrolytic cleavage upon storage in aqueous solution ${ }^{16,17}$. A bioisosteric approach, replacing the $\mathrm{N}^{G}$-acylguanidine structure with a $\mathrm{N}^{\mathrm{G}}$-carbamoylguanidine, led in many cases to more stable compounds. ${ }^{17-19}$ For exploration of the structure-activity $\left(\mathrm{H}_{2} \mathrm{R}\right)$ and the structure-selectivity relationships of this class of compounds, in addition to dimeric ligands, a series of carbamoylguanidines with various aminothiazole-based substructures, i.e., the 3-(2-amino-4-methylthiazol-5-yl)propyl moiety, a conformationally constrained (aminothiazolyl)phenyl and a 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl portion were prepared. Compounds containing the conformationally constrained aminothiazolylphenyl moiety were antagonists with only a weak affinity to the $\mathrm{hH}_{2} \mathrm{R}$ and compounds containing the less flexible 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety,
derived from pramipexole, showed partial (6.67, $\alpha$ : $0.53, \mathrm{pEC} 50$ value: 6.7) or weak partial agonism (6.66, $\alpha$ : $0.16, p E C_{50}$ value: 6.83) and bell-shaped concentration-response curves. Amino(methyl)thiazolyl propyl containing compounds achieved up to 80 -fold the potency of histamine ( $\mathrm{pEC} C_{50}$ values of 6.3-7.76) and partial to full agonistic activities with more or less pronounced bell-shaped concentration-response curves. A variety of aliphatic and aromatic residues was well tolerated and the agonistic $N^{G}$-carbamoylated guanidine-type ligands exhibited some functional bias towards G-protein activation. Additionally, representative ligands were investigated for dopamine receptor affinity ( $h D_{2 l o n g} R$ and $h D_{3} R$ ). These ligands showed affinity to the $\mathrm{hD}_{\text {2long }} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ value: 5.6-6.97) and $\mathrm{h} \mathrm{D}_{3} \mathrm{R}$ ( $\mathrm{p} K_{\mathrm{i}}$ value: 5.3-7.6).

In conclusion, this work afforded new radiolabeled and fluorescence labeled molecular tools for the $\mathrm{hH}_{2} \mathrm{R}$ and showed that $\mathrm{N}^{\mathrm{G}}$-carbamoylated amino(methyl)thiazolepropylguanidines are GProtein biased, high affinity $\mathrm{hH}_{2} \mathrm{R}$ agonists with good longterm stability.

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## ApPendix

## NMR SPECTRA


${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ) of 3.8

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ) of 3.8

${ }^{1} \mathrm{H}$-NMR spectrum (300 MHz, [D $\mathrm{D}_{6}$ ]DMSO) of $\mathbf{3 . 1 0}$

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO) of 3.10

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of $\mathbf{3 . 1 2}$





${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of $\mathbf{3 . 1 2}$

${ }^{1} \mathrm{H}$-NMR spectrum ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 3.13

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of $\mathbf{3 . 1 3}$

${ }^{1} \mathrm{H}$-NMR spectrum ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.21

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.21

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.22

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of $\mathbf{3 . 2 2}$

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $300 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.23

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.23

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.24

${ }^{13} \mathrm{C}$-NMR spectrum (150 MHz, [ $\left.\mathrm{D}_{6}\right]$-DMSO) of 3.24

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.25

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of $\mathbf{3 . 2 5}$

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.26

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.26

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.29

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.29

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.30


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${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of $\mathbf{3 . 3 0}$

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.34

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum (150 MHz, $\left[\mathrm{D}_{6}\right]$-DMSO) of 3.34

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.35

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 3.35

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.30

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.30

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.31

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.31

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ) of 4.32

${ }^{13} \mathrm{C}$-NMR spectrum ( 150 MHz , [ $\left.\mathrm{D}_{6}\right]$ DMSO) of 4.32

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.33

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.33

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.34

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.34

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.35

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.35

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.36

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.36

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.37

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.37

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.38

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.38

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO) of 4.40

${ }^{13}$ C-NMR spectrum (150 MHz, [D6 $]$ DMSO) of 4.40

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ) of 4.41

${ }^{13}$ C-NMR spectrum ( 150 MHz , [ $\left.\mathrm{D}_{6}\right]$ DMSO) of 4.41

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO) of 4.42

${ }^{13}$ C-NMR spectrum (150 MHz, [D6 $]$ DMSO) of 4.42

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO) of 4.43

${ }^{13}$ C-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ) of 4.43

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO) of 4.44

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$ DMSO) of 4.44

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.46

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.46

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.47

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum (150 MHz, CD $\mathrm{C}_{3} \mathrm{OD}$ ) of 4.47

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.48

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.48

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.49

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.49

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 4.50

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum (150 MHz, CD $\mathrm{C}_{3} \mathrm{OD}$ ) of $\mathbf{4 . 5 0}$

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of BMY 25368

${ }^{13} \mathrm{C}$-NMR spectrum ( 150 MHz , [D $\mathrm{D}_{6}$ ]-DMSO) of BMY 25368

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 5.9

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 5.9

${ }^{1} \mathrm{H}$-NMR spectrum ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 5.10

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of $\mathbf{5 . 1 0}$

${ }^{1} \mathrm{H}$-NMR spectrum ( $400 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ) of 5.11

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 5.11

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of $\mathbf{5 . 1 3}$


[^3]
${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.47

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.47

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.48

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.48

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.49





${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.49

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.50



[^4]
${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.51


${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.51

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.52

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.52

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.53

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.53

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.54

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.54

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.55

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.55

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.56

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.56

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.57

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.57

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.58

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.58

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.59

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.59

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.60

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.60

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.61

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.61

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.62

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.62

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.63

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.63

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 6.64

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 6.64

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.65

${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.65

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.66

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${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.66

${ }^{1} \mathrm{H}$-NMR spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.67


${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum (150 MHz, $\left[\mathrm{D}_{6}\right]$-DMSO) of 6.67

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 6.70

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of 6.70

${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.71

${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of 6.71

## RP-HPLC Chromatograms



RP-HPLC analysis (purity control) of 3.8, 3.10, 3.12, 3.13, 3.21-3.24


RP-HPLC analysis (purity control) of 3.25, 3.26, 3.29, 3.30, 3.34, 3.35, 4.30, 4.31


RP-HPLC analysis (purity control) of 4.32-4.38, 4.40


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RP-HPLC analysis (purity control) of 4.50, BMY2536, 5.9-5.14


RP-HPLC analysis (purity control) of 5.15-5.20, 6.47, 6.48


RP-HPLC analysis (purity control) of 6.49-6.56


RP-HPLC analysis (purity control) of 6.57-6.64


RP-HPLC analysis (purity control) of 6.65-6.67, 6.70, 6.71

## Abbreviations

| AC | adenylyl cyclase |
| :---: | :---: |
| AcOH | acetic acid |
| AIBN | azobisisobutyronitrile |
| AML | acute myeloid leukemia |
| aq. | aqueous |
| Ar | aryl |
| $\mathrm{B}_{\text {max }}$ | maximum number of binding sites |
| Boc | tert-butoxycarbonyl |
| Bq | becquerel |
| br s | broad signal |
| BRET | bioluminescence resonance energy transfer |
| BSA | bovine serum albumin |
| C | concentration |
| cAMP | cyclic adenosine monophosphate |
| COSY | correlated spectroscopy |
| CNS | central nervous system |
| cpm | counts per million |
| d | doublet |
| $\delta$ | chemical shift |
| DAG | diacylglycerol |
| DCM | dichloromethane |
| DIPEA | $N, N$-diisopropylethylamine |
| DMAP | 4-dimethylaminopyridine |
| DMEM | dulbecco's modified eagle medium, |
| DMF | dimethylformamide |
| DMSO | dimethylsulfoxide |
| $\mathrm{D}_{\mathrm{n}} \mathrm{R}$ | Dopamine receptor subtypes, $n=1,2,3,4,5$; human isoform: $h D_{n} R$ |
| DPM | desintegrations per minute |
| $\mathrm{EC}_{50}$ | agonist concentration which induces 50\% of the maximum response |
| EDTA | ethylenediaminetetraacetic acid |
| $\mathrm{E}_{\text {max }}$ | intrinsic activity |
| eq. | equivalent(s) |
| ESI | electrospray ionization |
| EtOAc | ethyl acetate |
| EtOH | ethanol |
| FCS | fetal calf serum |
| FP | fluorescence polarization |
| FRAP | fluorescence recovery after photobleaching |
| FRET | Förster resonance energy transfer |
| GDP | guanosine diphosphate |
| GEF | guanosine nucleotide exchange factor |
| GPCR | G-Protein coupled receptor |
| GRK | G-Protein coupled receptor kinase |
| GTP | guanosine triphosphate |
| $h \mathrm{D}_{2} \mathrm{R}_{\text {short }}$ | short splice variant of the human dopamine $\mathrm{D}_{2}$ receptor |


| $\mathrm{H}_{\mathrm{n}} \mathrm{R}$ | Histamine receptor subtypes, $n=1,2,3,4$; isoforms: $\operatorname{dog}\left(\mathrm{cH}_{n} \mathrm{R}\right)$, guinea pig $\left(\mathrm{gpH}_{n} \mathrm{R}\right)$, human $\left(\mathrm{hH}_{n} \mathrm{R}\right)$, mouse $\left(m H_{n} \mathrm{R}\right)$, rat $\left(\mathrm{rH}_{n} \mathrm{R}\right)$ |
| :---: | :---: |
| h | hour(s) |
| HBTU | 2-(1H-benzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate |
| HEK293T | human embryonic kidney cells |
| HMBC | heteronuclear multiple bond correlation |
| HPLC | high-performance liquid chromatography |
| HRMS | high resolution mass spectrometry |
| HSQC | heteronuclear single quantum coherence |
| $\mathrm{IC}_{50}$ | inhibitor concentration which suppresses $50 \%$ of an agonist induced effect or displaces $50 \%$ of a labeled ligand from the binding site |
| $\mathrm{IP}_{3}$ | inositol trisphosphate |
| IR | infrared spectroscopy |
| $k$ | retention factor |
| $K_{\text {b }}$ | dissociation constant derived from a functional assay |
| $K_{\text {d }}$ | dissociation constant derived from a saturation binding assay |
| $K_{\text {i }}$ | dissociation constant derived from a competition binding assay |
| $k_{\text {obs }}$ | observed association rate constant |
| $k_{\text {off }}$ | dissociation rate constant |
| $k_{\text {on }}$ | association rate constant |
| lit. | literature |
| m | multiplet |
| M | molar (mol/L) |
| MeCN | acetonitrile |
| MeOH | methanol |
| mol | mole(s) |
| mp | melting point |
| $N^{\text {G }}$ | guanidine nitrogen |
| NMR | nuclear magnetic resonance |
| NOESY | nuclear Overhauser effect spectroscopy |
| ON | over night |
| PBS | phosphate buffered saline |
| PE | petroleum ether |
| PLC | phospholipase C |
| q | quartet |
| qui | quintet |
| $R_{\text {f }}$ | relative to front |
| RP | reversed phase |
| RT | room temperature |
| S | (1) singlet or (2) second(s) |
| SEM | standard error of the mean |
| t | (1) triplet or (2) time |
| TBTU | 2-(1H-Benzotriazole-1-yl)-1,1,3,3-tetramethylaminium tetrafluoroborate |
| TEA | triethyl amine |
| TFA | trifluoroacetic acid |
| THF | tetrahydrofuran |


| TIRF | total internal reflection fluorescence |
| :--- | :--- |
| TLC | thin layer chromatography |
| TM | transmembrane |
| $t_{\text {R }}$ | retention time |
| Tris | tris(hydroxymethyl)aminomethane |
| UV | ultraviolet |

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Einige der experimentellen Arbeiten wurden in Zusammenarbeit mit anderen Personen durchgeführt. Entsprechende Vermerke finden sich in den entsprechenden Kapiteln. Eine detaillierte Auflistung aller Kooperationen enthält zudem der Abschnitt „Acknowledgements and declaration of collaborations".

Weitere Personen waren an der inhaltlich-materiellen Erstellung der vorliegenden Arbeit nicht beteiligt. Insbesondere habe ich hierfür nicht die entgeltliche Hilfe eines Promotionsberaters oder anderer Personen in Anspruch genommen. Niemand hat von mir, weder unmittelbar noch mittelbar, geldwerte Leistungen für Arbeiten erhalten, die im Zusammenhang mit dem Inhalt der vorgelegten Dissertation stehen.

Sabrina Biselli


[^0]:    ${ }^{\text {a }}$ Dye used for the preparation of the fluorescent ligand. ${ }^{\text {b }}$ Length of the linker between the squaric acid amide and fluorescent dye given as the number of carbon atoms. ${ }^{c}\left[{ }^{35}\right.$ S]GTP $\psi$ S assay determined on membrane preparations of Sf9 insect cells expressing the $h H_{2} R-G_{s \alpha s}$ fusion protein. The intrinsic activity ( $\alpha$ ) of histamine was set to 1.00 , and $\alpha$ values of investigated compounds were referred to this value. The $\mathrm{p} K_{\mathrm{B}}$ values of neutral antagonists were determined in the antagonist mode versus histamine ( $c=1 \mu \mathrm{M}$ ) as agonist. Data represent mean values $\pm$ SEM of 2-3 experiments performed in triplicate.

[^1]:    For the evaluation of cell images either (A) the whole cell mask or (B) the cell membrane mask was used. Unspecific binding was determined in the presence of famotidine ( 300 -fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min . Error bars represent mean $\pm$ SEM from three independent experiments (each performed in duplicate). Specific binding data were analyzed by an equation describing one-site (monophasic) binding.

[^2]:    ${ }^{2}$ Determined by displacement of $5.14\left(c=50 \mathrm{nM}\right.$ ) in the presence of increasing concentrations of the respective $\mathrm{H}_{2} R$ ligand at intact cells by flow cytometry. ${ }^{\text {b }}$ Determined by displacement of 5.18 ( $c=50 \mathrm{nM}$ ) in the presence of increasing concentrations of the respective $\mathrm{H}_{2} \mathrm{R}$ ligand at adherent cells. ${ }^{\mathrm{C}}$ Determined by competition binding of 5.14 ( $\mathrm{c}=70 \mathrm{nM}$ ) in the presence of increasing concentrations of the respective $\mathrm{H}_{2} \mathrm{R}$ ligand at intact cells by imaging flow cytometry, for the evaluation of cell images the cell membrane mask was used. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data represent means $\pm$ SEM of 3 independent experiments, each performed in triplicate or duplicate.

[^3]:    ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectrum ( $600 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) of $\mathbf{5 . 1 8}$

[^4]:    ${ }^{13} \mathrm{C}$-NMR spectrum ( $150 \mathrm{MHz},\left[\mathrm{D}_{6}\right]$-DMSO) of $\mathbf{6 . 5 0}$

[^5]:    RP-HPLC analysis (purity control) of 4.41-4.44, 4.46-4.49

