Synthesis and Pharmacological Characterization of Subtype-Selective Ligands, Including Radio- and Fluorescence Labeled Ligands, for the Histamine H₂ Receptor

Dissertation

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CHAPTER 1

GENERAL INTRODUCTION

1.1 The Histamine H_2 Receptor as a Prototypic Aminergic GPCR

The histamine H₂ receptor (H₂R) belongs to the superfamily of G-protein coupled receptors (GPCRs).¹ GPCRs are integral membrane receptors and are characterized by seven hydrophobic transmembrane (TM) domains with an extracellular amino terminus and an intracellular carboxyl terminus. The extracellular regions combined with the transmembrane regions are important for ligand binding.² The intracellular regions are substantially involved in signaling and feedback mechanisms.² With around 30% of the most prominent approved drugs targeting these membrane receptors, GPCRs are the most important drug targets.^{3,4} GPCRs are mediated by numerous endogenous ligands e.g. biogenic amines (aminergic GPCRs), amino acids, peptides, proteins, purins and lipids, to name only a few.^{1,5,6}

The H₂R is one of currently four histamine receptor subtypes (H₁R, H₃R and H₄R), which are all activated by binding the endogenous ligand histamine and therefore are aminergic GPCRs.⁷⁻¹⁰ All histamine receptors belong to the rhodopsin family of GPCRs.¹ The H₂R is primarily located on parietal cells in the stomach,¹¹ in mammalian brain,^{12,13} on neutrophiles and eosinophiles¹⁴ as well as on smooth muscle cells¹⁵ (e.g. in the heart, airways and uterus). An essential physiological function of the H₂R is the control of the gastric acid secretion.^{8,11} Furthermore, activation of H₂R results in smooth muscle relaxation and positive inotropic and chronotropic effects.¹⁶

The H₂R species isoforms (e.g. human (hH₂R), guinea pig (gpH₂R), rat (rH₂R), mouse (mH₂R) and dog (cH₂R)), like many GPCRs, interact similarily with their endogenous ligand, but quite differently with most synthetic ligands.^{17,18} The pharmacological differences between the hH₂R and the gpH₂R mainly concern agonists and not antagonists, which was very fortunate as the first potent antagonists for the treatment of gastroduodenal ulcers were developed relying on animal models.¹⁷ The cH₂R exhibits an increased constitutive activity compared to hH₂R and rH₂R.¹⁸ These findings show that for the development of highly potent and selective agonists it is crucial to study hH₂R and not the orthologs.

1.2 G-PROTEIN ACTIVATION AND SIGNALING PATHWAYS

In the classical model the active receptor conformation (either stabilized by agonist binding or constitutively active) is functioning as a guanosine nucleotide exchange factor (GEF) on the G α subunit of the heterotrimeric G-protein (Figure 1.1).⁵ The binding of the G-protein complex to the active receptor leads to conformational changes which result in the release of GDP from its binding site at the G α subunit and the formation of the ternary complex.¹⁹ Subsequently, GTP is bound and the ternary complex dissociates into the G α -GTP subunit, the G $\beta\gamma$ complex and the free receptor.¹⁹ Both subunits can interact with effector proteins resulting, through an increase or a decrease in the concentration of second messangers, in various cellular responses.¹⁹ After a certain period of time, the intrinsic GTPase activity of the G α subunit converts GTP to GDP and phosphate.¹⁹ The G α -GDP subunit re-associates with the G $\beta\gamma$ complex to the inactive heterotrimeric G-protein.¹⁹

Based on their structures and signaling pathways, G-proteins are grouped in four families according to their G α subunit: G $\alpha_{i/o}$, G α_s , G $\alpha_{q/11}$, G $\alpha_{12/13}$.^{20,21} The H₂R predominantly couples to

 $G\alpha_s$ proteins, resulting in an increase of the second messenger cAMP by stimulation of the isoforms of the effector protein adenylyl cyclase (Figure 1.1).^{22,23} By contrast, the H₃R and the H₄R signal mainly via $G\alpha_{i/o}$ proteins, which inhibit the adenylyl cyclase.²⁴ The H₁R preferentially couples to $G\alpha_{q/11}$ leading to the activation of phospholipase C (PLC) and subsequent release of IP₃ and DAG.^{24,25}

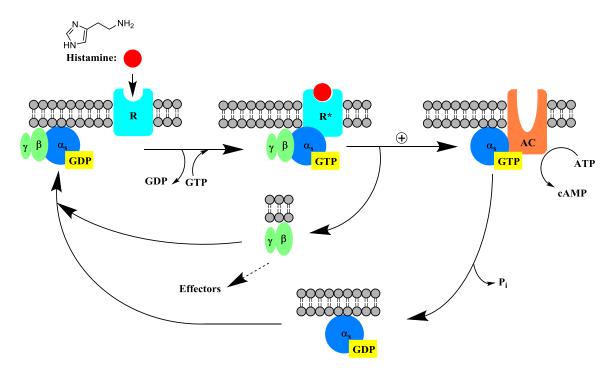


Figure 1.1. Activation of the heterotrimeric G-protein by the agonist occupied receptor using the H_2R as an example. R represents the inactive receptor conformation and R* the active receptor conformation. The dissociated subunits ($G\alpha_s$ and $G\beta\gamma$ complex) regulate effector proteins such adenylyl cyclase (AC), which is activated by $G\alpha_s$. Modified from Rasmussen et al.²⁶

For analyzing GPCR-mediated guanine nucleotide exchange at G-proteins, a widely employed method is the [35 S]GTP γ S binding assay.²⁷ This assay utilizes, like the closely related steady-state GTPase assay, the intrinsic GTPase activity of the G α subunit. An advantage of the GTP γ S binding assay (and GTPase assay) is that it assesses coupling at a proximal level, avoiding potential bias introduced by downstream events.²⁷ For the H₂R the usage of membranes of Sf9 insect cells, which are expressing mammalian H₂R-G_{s α} fusion proteins is well established.^{17,28,29} GPCR-G_{s α} fusion proteins ensure a defined 1:1 stoichiometry of the signaling partners and efficient coupling.^{17,29,30} Therefore, the ternary complex formation is more efficient compared to the coexpression of H₂R plus G_{s α}.²⁸ In our workgroup, the H₂R-G_{s α} fusion protein system is routinely employed for analyzing new ligands for the H₂R in radioligand binding and functional studies (GTP γ S binding assay and GTPase assay).³¹⁻³⁶

1.3 G-PROTEIN INDEPENDENT SIGNALING, LIGAND CLASSIFICATION AND FUNCTIONAL SELECTIVITY

Besides the signal transduction cascades mediated by G-proteins, GPCRs are reported to participate in numerous other protein-protein-interactions which initiate signaling pathways independent from G-protein activation.^{19,37,38} Most intriguing is the interaction with β -arrestins, which are mainly involved in receptor desensitization and internalization, but also act as alternative signal transducers.^{19,37} β -Arrestin recruitment is initiated by phosphorylation of the active conformation of the GPCR by G-protein coupled receptor kinases (GRK).^{38,39} The β -arrestin binds to the cytosolic surface of the phosphorylated receptor and sterically hinders an interaction with the G-proteins.⁴⁰ Furthermore, β -arrestins were reported to be involved in the degradation of second messengers.^{41,42} These two effects effectively lead to the deactivation of the G-protein mediated signal transduction. Beyond desensitization, the bound β -arrestin also mediates internalization via clathrin-coated pits.³⁸

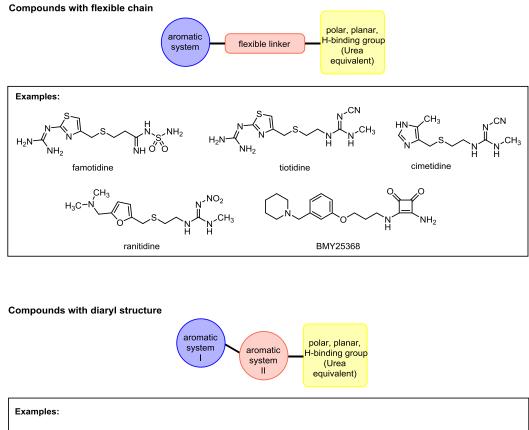
A classical "two state" model, which is often suitable for explaining the pharmacodynamic activity of ligands is the cubic ternary complex model.⁴³⁻⁴⁵ This model distinguishes between an active (R*) and inactive (R) receptor state, which are in equilibrium and are able to isomerize without agonist binding. This spontaneous activation of the receptor in the absence of an agonist is referred to as constitutive activity.⁴⁶ The G-protein is able to bind to both states, albeit only the G-protein-active-receptor-complex (R*G) activates intra cellular signaling via a GDP-GTP exchange. Ligand binding can shift the equilibrium of the receptor state. Agonists bind with high affinitiy to R* and stabilize the active conformation. Inverse agonists prefer to bind to R and stabilize the inactive conformation. Neutral antagonists bind with the same affinity to both conformations and therefore do not alter the equilibrium. With regard to β -arrestin mediated signaling, along site with other mechanism such as phosphorylisation, internalization and oligomerisation, there is growing evidence that there are multiple active and inactive receptor conformations.^{47,48} Structurally different ligands stabilize distinct receptor conformations leading to an activation of only a subset of cellular effectors ⁴⁸ This selective activation of only some of all possible signaling pathways has been referred to as 'functional selectivity',⁴⁹ 'biased agonism'⁵⁰ or 'differential receptor-linked effector actions'⁵¹.

Recently, several monomeric and dimeric H_2R ligands were investigated for biased agonism regarding G-protein activation and β -arrestin recruitment.⁵² The β -arrestin recruitment was measured by an enzyme fragment complementation assay using split luciferase fragments from *P. termitilluminans*, developed by Misawa *et al.*⁵³ While all antagonists were unbiased, the investigated acyl- and carbamoyl guanidine agonists revealed varying degrees of G-Protein bias.⁵²

1.4 H₂R ANTAGONISTS

The classical H_2R antagonists can be devided into two groups depicted in Figure 1.2: compounds comprising a flexible chain (group I) and compounds containing diaryl moiety (goup II).⁵⁴ The antagonists consist of an aromatic system, which is linked to a polar, planar group (urea equivalent) by either a flexible chain (group I) or by a second aromatic system (group II). The

classification of the antagonists is made according to the aromatic system. Most H_2R antagonists belong to one of four major structural classes: imidazole-, guanidinothiazole-, aminomethylfurane- and piperidinomethylphenoxy-containing compounds.



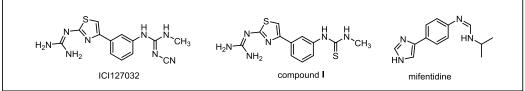


Figure 1.2. Selected H_2R antagonists and their classification into two groups: compounds with a flexible chain and compounds with a diaryl moiety.⁵⁴

The aminopotentidine derivatives as well as fluorescent ligands (e.g. compound **II**) showed that within the piperidinomethylphenoxy-containing compounds additional substituents at the urea equivalent are well tolerated or provided additional H₂R binding affinity (Figure 1.3).⁵⁵⁻⁵⁷ Up to date, iodoaminopotentidine, a piperidinomethylphenoxy-containing cyanoguanidine (Figure 1.3), which was also synthesized in a radiolabeled form ([¹²⁵I]iodoaminopotentidine), showed the highest affinity.¹³ This radioligand was used to map the H₂R densities in human and mammalian brain.^{12,13} Recently, a series of piperidinomethylphenoxyalkylamine-containing ligands, coupled with various polar groups ("urea equivalents") such as cyanoguanidine, nitroethenediamine, amide or squaric amide moieties, and a terminal amino group, connected via a linker of different length, was developed by our group.³⁵ The squaramides, which also tolerated propionylation at the terminal amino-group showed the highest affinities. UR-DE257, which showed a high affinity (pK_i value: 7.55), was also synthesized in radiolabeled form ([³H]UR-DE257) and is frequently used in competition binding experiments.^{35,36}

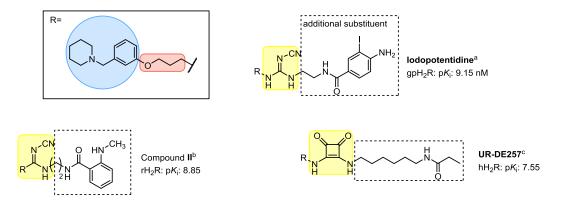


Figure 1.3. Structures of exemplary piperidinomethylphenoxy-containing ligands. a: Hirschfeld et al and Ruat et $al^{13,57}$; b: Malan et al^{58} ; c: Baumeister et al^{35} .

Whereas in the late 1970s to 1980s, the H_2R antagonists (H_2 blockers) revolutionized the treatment of peptic ulcers, former blockbuster drugs like cimetidine, ranitidine or famotidine are outdated. They were largely superseded by the more effective proton pump inhibitors (e.g. omeprazole). Nevertheless, H_2R antagonists are valuable molecular tools to study the H_2R , especially it's role in the brain, which is still far from being completely understood.

1.5 H₂R AGONISTS

The early agonists for the H_2R were derived from histamine and consististed of an imidazole pharmakophore coupled to a guanidine by a flexible linker (e.g. impromidine and arpromidine, see Figure 1.4).^{59,60} Arpromidine and related compounds showed up to 400 times potency of histamine at the spontaneously beating guinea pig right atrium, but the strongly basic guanidine moiety led to poor oral bioavailability and CNS penetration.⁵⁹ The bioisosteric replacement of the guanidine ($pK_a \sim 13$) with an acylguanidine ($pK_a \sim 8$) resulted in ligands with either retained or even increased agonistic potency (e.g. UR-PG80 and UR-AK24, see Figure 1.4).^{31,32} Modification of these N^G-acylated imidazolylpropylguanidines, which lacked subtype selectivity (H₃R and H₄R), to N^{G} -acylated aminothiazolylpropylguanidines led to highly potent and selective H₂R agonists.³⁴ Surprisingly, the H_2R potency was increased up to 4000-fold the potency of histamine by linking two acylguanidine moieties (e.g. UR-AK381, see Figure 1.4).³³ The aminothiazole dimeric ligands are the most potent and selective H₂R agonists known so far. Traditionally, dimeric (bivalent) ligands consist of two pharmacophoric moieties linked through a spacer and are designed to bridge two neighboring receptor protomers.⁶¹ Porthogese *et al* suggested a distance of about 22-27 Å between the two orthosteric binding sites of a receptor dimer.⁶² Interestingly, the most active dimeric ligands have spacer of lengths insufficient to bridge the protomers of putative H₂R dimers.³³ The enormous gain in potency is speculated to result from an interaction with the orthosteric and an accessory binding site at the same protomer.³³ Recently, it was shown that bioisosteric replacement of the acylguanidines with the more stable carbamoylguanidine led to dimeric ligands with retained potency and intrinsic activity (e.g. UR-NK22, see Figure 1.4).³⁶

So far, there is no H_2R agonist for therapeutic use on the market, but H_2R agonists are valuable molecular tools to study the H_2R . Nonetheless, there are numerous possible indications e.g. as positive inotropic vasodilators for the treatment of congestive heart failure or as differentiationintroducing agents for treatment of acute myeloid leukemia (AML). For the later, the endogenous agonist histamine is used as an orphan drug in combination with interleukin 2. Histamine promotes the activation of T cells and natural killer cells by interleukin 2, which results in the killing of cancer cells.⁶³ Given the effect of histamine is mediated via H_2R , the application of highly selective H_2R agonists might be beneficial in regard to potency and a reduction of adverse effects. Recently, investigation of the dimeric N^G -carbamoylated aminothiazolylpropylguanidines on human monocytes revealed a high H_2R agonist potency, suggesting that this class of compounds is a promising starting point for the development of H_2R agonists for the treatment of AML.³⁶

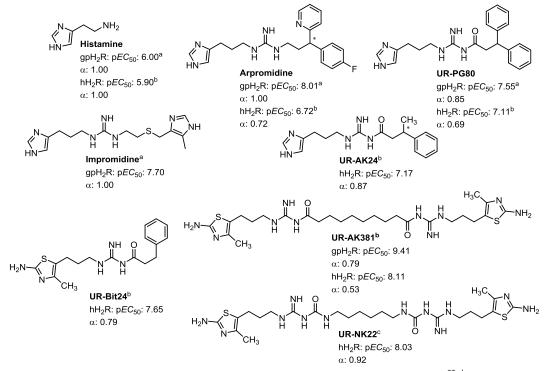


Figure 1.4. Structures of selected H_2R agonists. Agonism measured on ^aguinea pig right atrium³², ^bsteady-state GTPase assay^{31,33,34} or ^cGTPyS binding assay³⁶.

1.6 RECEPTOR LIGAND BINDING ASSAYS AND LABELED MOLECULAR TOOLS FOR GPCRS

The initial step in every signaling cascade, that causes a receptor-mediated biological response, is the binding of a ligand to the receptor. There are multiple ways to utilize receptor-ligand-interactions in research, e.g. as a tool for determining receptor distribution, for identification of receptor subtypes and for screening of new compounds.⁶⁴

The classical approach for the determination of ligand affinity is the radioligand binding assay, which has been unchallenged for a long time regarding sensity and reproducibility.^{64,65} Radioligand binding experiments can be divided in three basic types: Saturation binding experiments are used to determine the affinity of the radioligand and the number of specific binding sites. In kinetic experiments, the rate constants of association and dissoviation of a radioligand can be determined. Competition binding experiments are widely used to identify unlabeled compounds, which bind to the receptor in question by displacement of a radioligand. The major disadvantages include that radioligands are potentially hazardous to human health,

produce high costs in production and waste disposal, require special licences and laboratory equippment, and separation of bound from unbound ligand is necessary.

Today, new highly sensitive fluorescence and bioluminescence methods such as fluorescence polarization (FP),⁶⁶ total internal reflection fluorescence (TIRF),⁶⁷ fluorescence/bioluminescence resonance energy transfer (FRET/BRET),⁶⁸ fluorescence recovery after photobleaching (FRAP),⁶⁹ high content imaging⁷⁰ and flow cytometry⁷¹ became promising alternatives. Like radioligands, fluorescent ligands can be used in the basic types of binding experiments. Several peptidic and non-peptidic fluorescent ligands were identified for GPCRs, including NPY,^{72,73} muscarinic⁷⁴ and histamine^{55,56,75} receptors. In general, a fluorescent ligand comprises of a pharmacophore, a linker and the fluorophore. A major challenge in the development of small-molecule fluorescent ligands is to retain affinity, when a, compared to the ligand, bulky fluorophore is attached. In comparison, a radiolabel, especially tritium, does not alter the affinity of the ligand.

When selecting a radio- or fluorescent labeled ligand for binding experiments, several aspects have to be considered.^{65,76,77} Firstly, the ligand should be selective and bind with high affinity to the respective receptor. Secondly, high (radiochemical) purity and high specific activity (radioligand) or quantum yield (fluorescent ligand) is required. Thirdly, the labeled ligand should be chemically stable under assay conditions for at least the duration of the experiment performed. Furthermore, unspecific binding has to be considered, the choice of radio- or fluorescent label and, whether an agonist or an antagonist is desired as labeled ligand. Under unspecific binding all binding sites other than the receptor of interest are summarized. A competition binding assay, were only 50% of total radioligand binding is specific is considered adequate, 70% is good and 90% is excellent.⁷⁶ Tritium is often considered as the radioisotope of choice. Compared to ¹²⁵I or the occasionally used ^{32/33}P or ³⁵S, tritium has a longer half-life (14-87 days vs. 12.3 years) and the tritiated compounds are more convenient in handeling with respect to safety precautions.⁷⁶ When choosing a fluorophore, excitation and emission wavelengths, stoke shifts and quantum yields have to be considered. Generally, red-emitting (λ_{em} : > 600 nm) fluorophores with a long stoke shift and a high quantum yield are preferred. ⁷⁷ Agonists label only an active conformation of the receptor and therefore, only a fraction of the total active receptor population.⁷⁸ By contrast, antagonists bind to all receptor states with the same affinity according to the classical model described above.⁷⁸

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Chapter 2

SCOPE AND OBJECTIVES

The Histamine H₂ receptor, an aminergic GPCR, is primarily known for its physiological role in the control of gastric acid secretion.^{1,2} Additionally, activation of H₂R results in positive inotropic and chronotropic effects and smooth muscle relaxation.³ The H₂R is primarily located on parietal cells in the stomach,² in mammalian brain,^{4,5} on human neutrophiles and eosinophiles⁶ as well as on smooth muscle cells⁷. Antagonists, which were intensively studied as antiulcer therapeutics in the 1960s to 80s, play only a minor role today, but are still important. Although, there is no H₂R agonist for therapeutic usage on the market, H₂R agonists are valuable molecular tools to study the pharmacology of the H₂R. Nonetheless, there are numerous possible indications e.g. as positive inotropic vasodilators for the treatment of congestive heart failure or as differentiation-introducing agents for treatment of acute myeloid leukemia. The development of selective high affinity molecular tools for the H₂R, including agonists as well as labeled molecules like radioligands and fluorescent ligands, is very important to identify new ligands, investigate receptor distribution and further unravel its (patho-)physiological role.

The number of suitable high affinity radioligands for the H_2R is very limited. [³H]Histamine, as well as several reported tritiated antagonists (e.g. [³H]cimetidine,^{8,9} [³H]ranitidine¹⁰ and [³H]tiotidine¹¹) are less than ideal molecular tools to study the H₂R. As [³H]cimetidine is reported to label an imidazole recognition site rather than the H_2R^8 and ranitidine as well as histamine suffer from low H₂R affinity and potency.^{5,11} [³H]Tiotidine is frequently used as a radioligand, although it shows very high unspecific binding and is reported to address only a subpopulation of the H_2R .¹¹ [¹²⁵]lodoaminopotentidine suffers from a short half-life of only 60 days, but shows the highest affinity to the H₂R reported so far (gpH₂R: K_d value: 0.34 nM).^{5,12} An tritated alternative presents the recently published high-affinity tritium-labeled H₂R antagonist [³H]UR-DE257¹³ (hH₂R: K_d value: 31 nM), which is structurally related with the squaramide BMY25368¹⁴. This radioligand proved to be useful for the determination of pK_i values, but turned out as an insurmountable antagonist in functional assays.¹³ One objective of this thesis was to design new ligands with a free terminal amino group and exploration of the applicability of them as precursors for the attachment of a radioactive (tritiated) moiety. Two reported classical antagonistic structures were used as scaffolds: the guanidinothiazoles and the aminopotentidines.

The guanidinothiazole-moiety is a privileged structure for H_2R antagonism. Previous functional studies identified guanidinothiazole-containing ligands, e.g. famotidine and ICI127032, as surmountable H_2R antagonists.^{15,16} In this thesis, two different guanidinothiazole-precursors derived from famotidine and ICI127032 had to be combined with an "urea" equivalent. As a strategy to enable radiolabeling, the introduction of diaminoalkyl-linkers varying in length at the "urea" equivalent was envisaged. The resulting terminal amino group could be propionylated by succinimidyl propionate, to obtain "cold forms" of the potential radioligands.

Aminopotentidine and its derivatives are reported as high affinity H₂R antagonists.¹² Interestingly, iodination in the 3-position of the 4-aminobenzoic acid amide moiety resulted in an enormous gain in affinity (iodoaminopotentidine).^{4,12} Aminopotentidine and its analogs containing different substituents (e.g. iodine, bromine, chlorine, trifluoromethyl) in position 3 were synthesized as precursors for radiolabeling. The derivatization of the anilinic amino group of these precursors was performed with various reagents (e.g. N-succinimidyl propionate, propionic acid chloride, methyl iodide or acetyl chloride), which are also commercially available in tritiated form. Anilinic

amines show a reduced nucleophilicity which resulted in a reduced reactivity in the (radio-)labeling reaction. To overcome this challenge, the synthesis and characterization of a series of aminopotentidine derivatives, containing a functionalized (propionylated, acetylated or methylated) aminomethyl substituent in position 4 of the aromatic ring was considered.

The availability of high affinity red-emitting H₂R fluorescent ligands is very limited. Recently, a series of fluorescent ligands with a piperidinomethylphenoxypropylamino (potentidine) pharmacophore was reported.¹⁷ The most promising ligands within this series are the squaramide-type ligands UR-DE229 and UR-DE56.¹⁷ Both ligands consist of a BMY 2536¹⁴ pharmacophore, which is linked to a fluorescent label (pyridinium or cyanine) by a n-alkyl linker. Another objective of this thesis was the synthesis and biological characterisation of fluorescent high affinity H₂R antagonists with improved optical and physicochemical properties to gain access to a wide range of potential applications, in particular to confocal microscopy and to high throughput or/and high content imaging. Therefore, the fluorescent labeled antagonists UR-DE229 and UR-DE56 were investigated in different assay/imaging systems. For the exploration of the impact of alkyl linker length and different net charges of fluorophores, a series of derivatives of UR-DE229 and UR-DE56 was considered.

 N^{G} -acylated amino(methyl)thiazolepropylguanidines represent a class of potent and selective histamine H₂R agonists.¹⁸⁻²⁰ As it was reported that N^{G} -acylguanidines undergo hydrolytic cleavage upon storage in aqueous solution, more stable analogues are needed.^{21,22} A bioisosteric approach, replacing the N^{G} -acylguanidine structure with a N^{G} -carbamoylguanidine, was planned in the final part of this thesis, aiming at N^{G} -carbamoylated aminothiazole-containing compounds with high affinity and improved long term stability. Furthermore, structure-activity (H₂R) and the structure-selectivity relationships (H₂R versus H₁R, H₃R and H₄R) of this class of compounds were considerd.

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Chapter 3

Guanidinothiazoles: Towards the Squaramide- Type H₂R Radioligand [³H]UR-SB69

Note: The synthesis and purification of $[^{3}H]$ **3.25** was performed by Max Keller.

3.1 INTRODUCTION

Over the years, the endogenous agonist histamine and several H₂R antagonists were synthesized in a radiolabeled form (Figure 3.1). Nevertheless, [³H]histamine,¹ [³H]cimetidine,^{2,3} [³H]ranitidine⁴ and [³H]tiotidine⁵ are less than ideal molecular tools to study the H₂R. As [³H]cimetidine is reported to label an imidazole recognition site rather than the H₂R² and ranitidine as well as histamine suffer from low H₂R affinity and potency,^{5,6} [³H]tiotidine is frequently used as a radioligand, although it shows very high unspecific binding and is reported to address only a subpopulation of the H₂R.⁵ By contrast, the labeling of aminopotentidine with ¹²⁵iodine resulted in a high affinity radioligand ([¹²⁵I]iodoaminopotentidine, gpH₂R: K_d value: 0.34 nM) which was used e.g. for autoradiography of the H₂R in human and rodent brain and heart as well as for saturation and kinetic binding studies.^{6,7} Although ¹²⁵iodine labeled ligands have, compared to tritium labeled ligands, the advantage of a higher specific activity, their preparation and usage require higher safety precautions and the ligands can only be used for 4-5 weeks after preparation.⁸

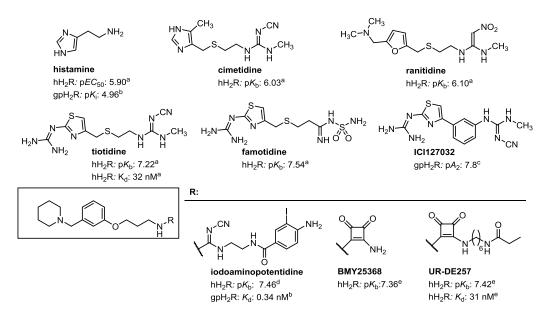


Figure 3.1. Structures of the endogenous ligand histamine and selected standard H_2R antagonists. ^aKelley et al⁵, ^bRuat et al⁶, ^cYellin et al⁹, ^dPreuss et al¹⁰, ^eBaumeister et al¹¹.

An alternative to $[^{125}I]$ iodoaminopotentidine is the recently published high-affinity tritium-labeled H₂R antagonist $[^{3}H]$ UR-DE257 (K_{d} value: 31 nM)¹¹, which is structurally related to the squaramide BMY25368.^{11,12} Although UR-DE257 was an insurmountable antagonist in functional assays, this radioligand proved to be very useful for the determination of p K_{i} values.¹¹

Aiming at the development of high affinity tritium-labeled surmountable H_2R antagonists, guanidinothiazole-containing amines were considerd promising precursors. In previous functional studies, the antagonists with guanidinothiazole sub-structure, namely famotidine and ICI127032 were identified as surmountable H_2R antagonists.^{9,13}

The 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole precursor derived from famotidine and the 2-guanidino-4-(3-aminophenyl)thiazole precursor derived from ICI127032 were combined with the squaramide moiety ("urea" equivalent) of BMY25368 (general structure: Figure 3.2). In

order to enable radiolabeling, a terminal amino group connected to the squaramide via *n*-alkyl linker of different length, was introduced following the same strategy as in case of UR-DE257. The terminal amino group was propionylated using N-succinimidyl propionate to obtain "cold forms" of the potential radioligands. Furthermore, the squaramide moiety was replaced by a cyanoguanidine moiety resulting in tiotidine-like compounds.

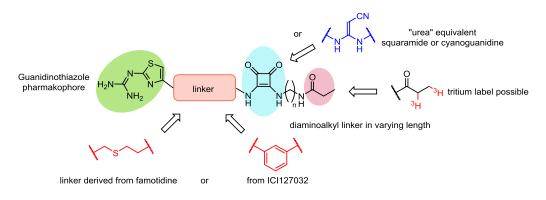


Figure 3.2. General structure of the guanidinothiazoles as potential new radioligands.

Interestingly, linking two amino(methyl)thiazolepropyl containing acyl or carbamoyl guanidines resulted in high affinity H₂R agonists (e.g. UR-NK22).^{14,15} Therefore, this strategy was also adapted to the guanidinothiazoles: The replacement of the amino(methyl)thiazolepropyl moiety of UR-NK22 with either 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole or 2-guanidino-4-(3-aminophenyl)thiazole resulted in two bivalent ligands (Figure 3.3).

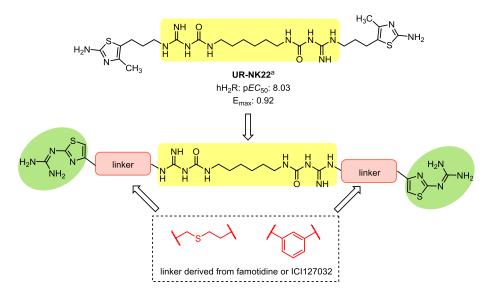


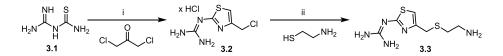
Figure 3.3. General structure of the dimeric guanidinothiazoles. ^aKagermeier et al¹⁴.

The amine precursors and the "cold forms" of the potentional radioligands as well as the two bivalent ligands were characterized in binding and functional (GTP γ S binding assay) studies on recombinant histamine receptors (preferentially hH₂R and hH₃R). Radiolabeling is accessible by coupling of the commercially available N-succinimidyl [2,3-³H]propionate with the respective amine precursor.

3.2 RESULTS AND DISCUSSION

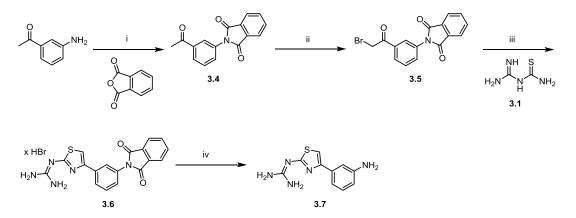
3.2.1 Chemistry

The synthesis of the amine building block 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole (**3.3**) according to published procedures is outlined in Scheme 3.1.¹⁶ Firstly, 2-guanidino-4-chloromethylthiazole hydrochloride (**3.2**) was synthesized from amidinothiourea and 1,3-dichloroacetone. Secondly, **3.2** was coupled with 2-aminoethanthiole in a substitution reaction in the presence of sodium ethanolate to obtain **3.3** in excellent yield.



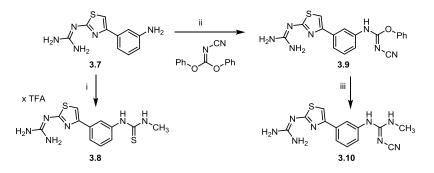
Scheme 3.1. Synthesis of 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole 3.3. Reagents and conditions: i) acetone, RT, 24 h, 56%; ii) EtONa, EtOH, 0 °C to RT, 24 h, 84%.

The conformationally constrained amine building block **3.7** was synthesized from 1-(3aminophenyl)ethan-1-one and amidinothiourea according to literature procedures (Scheme 3.2).¹⁷ After phthalimide protection **3.4** was brominated in alpha position and subsequently treated with amidinothiourea in order to form the protected building block **3.6** in a good yield of 70% (over two steps). Deprotection of the phthalimide group in a mixture of HCl and acetic acid afforded the conformationally constrained amine building block **3.7**.



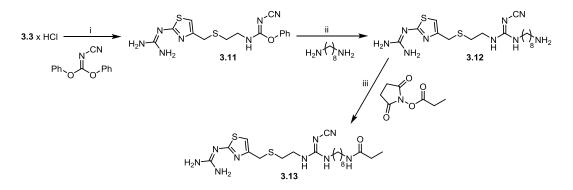
Scheme 3.2. Synthesis of 2-guanidino-4-(3-aminophenyl)thiazole 3.7. Reagents and conditions: i) Acetic acid, reflux, 2.5 h, 96%; ii) Br₂, HBr, CH₂Cl₂, CHCl₃, RT, 0.5 h, no purification; iii) CH₃CN, EtOH, reflux, 5 h, 70%; iv) HCl, acetic acid, reflux, 24 h, 52%.

The synthetic route leading to the H_2R antagonists **3.8** and ICI127032 (**3.10**) is depicted in Scheme 3.3 as described in literature¹⁷ with minor modifications. The formation of the thiourea **3.8** resulted from treatment of building block **3.7** with methylisothiocyanate. ICI127032 (**3.10**) was synthesized in two steps: Diphenylcyanocarbonimidate was first coupled with **3.7** and the resulting intermediate **3.9** was than treated with methylamine.



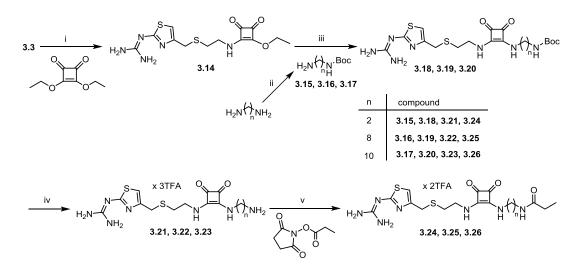
Scheme 3.3. Synthesis of the H_2R antagonists **3.8** and ICI127032 (**3.10**).¹⁷ Reagents and conditions: i) methylisothiocyanate, acetone, RT, ON, 53%; ii) 2-propanol, RT, ON, no purification; iii) aq. methylamine solution, RT, ON, 55%.

The synthesis of the cyanoguanidine-type guanidinothiazole **3.13** followed a three step route (Scheme 3.4). The building block **3.3** was treated with the cyanoguanidinylating reagent diphenylcyanocarbonimidate¹⁸ by analogy to published protocols.^{7,11} The resulting intermediate **3.11** was treated with an excess of 1,8-diaminooctane in order to preferably form the monovalent ligand **3.12**. The propionamide **3.13** were synthesized from **3.12** and N-succinimidyl propionate.



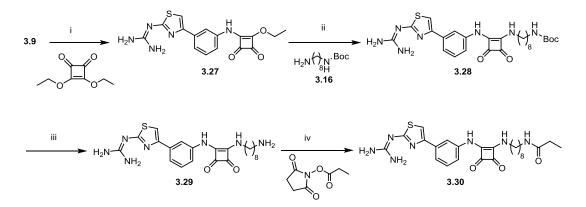
Scheme 3.4. Synthesis of the cyanoguanidine-type guanidinothiazole **3.13**. Reagents and conditions: i) TEA, MeOH, RT, ON, no purification; ii) MeCN, 50 °C, 3 h, 62%; iii) TEA, CH₂Cl₂, DMF, RT, 3 h, 22%.

The synthesis of the squaramide-type guanidinothiazoles **3.24-3.26** was adopted from known procedures (Scheme 3.5).^{11,16} The reaction of precursor **3.3** and 3,4-diethoxycyclobut-3-ene-1,2-dione gave the squaric acid ester amide **3.14** as an intermediate, which was treated with the respective mono-Boc protected diamine **3.15-3.17** to yield the *tert*-butyl carbamates **3.18-3.20**. The mono-Boc protected amines were synthesized by treating an excess of the diamine with Di*tert*-butyl dicarbonate. Final cleavage of the protecting group with TFA afforded the amines **3.21-3.23**. The propionylated ligands **3.24-3.26** were accessible through acylation of the respective amine precursor **3.21-3.23** with N-succinimidyl propionate.



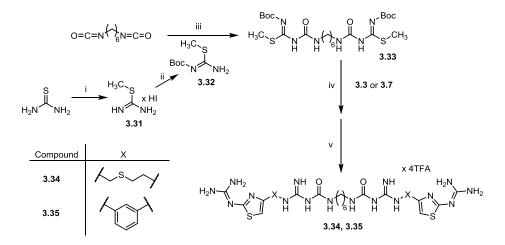
Scheme 3.5 Synthesis of the squaramide-type guanidinothiazoles **3.24-3.26**. Reagents and conditions: i) TEA, EtOH, RT, ON, 87%; ii) Di-*tert*-butyl dicarbonate, CHCl₃, 0 °C to RT, ON, 41-63%; iii) TEA, EtOH, 60-70 °C to RT, 6 h - 2 days, 63-88%; iv) TFA, CH₂Cl₂, RT, 1.5 h, 57-97%; v) TEA, CH₂Cl₂, RT, ON, 33-72%.

The conformationally constrained squaramide-type guanidinothiazole **3.30** was synthesized by following the same protocol as in case of the compounds **3.24-3.26** (Scheme 3.6). Starting from precursor **3.9**, treatment with 3,4-diethoxycyclobut-3-ene-1,2-dione gave the squaric acid ester amide intermediate **3.27**, which was reacted with *tert*-butyl N-(8-aminooctyl)carbamate (**3.16**) to obtain the *tert*-butyl carbamate **3.28**. Deprotection led to the amine precursor **3.29** which was propionylated with N-succinimidyl propionate.



Scheme 3.6 Synthesis of the conformationally constrained squaramide-type guanidinothiazole **3.30**. Reagents and conditions: i) EtOH, RT, ON, no purification; ii) TEA, EtOH, 80 °C, ON, 34%; iii) TFA, CH₂Cl₂, RT, 5 h, 61%; iv) TEA, DMF, CH₂Cl₂, RT, ON, 12%.

The synthetic route leading to the bivalent guanidinothiazole containing ligands **3.34** and **3.35** was adopted from published protocols¹⁴ and is depicted in Scheme 3.7. Thiourea was methylated with methyl iodine and subsequently mono-Boc protected, resulting in the well-established guanidinylating reagent **3.32**. Treatment of **3.32** with 1,6-diisocyanohexane afforded the bivalent guanidinylating reagent **3.33**. The bivalent ligands **3.34** and **3.35** were prepared by reaction of **3.33** with the respective guanidinothiazole building block (**3.3** or **3.7**) in the presence of a base, followed by preparative HPLC purification of the Boc-protected intermediates. The protecting group remained stable over the course of the purification, and deprotection was achieved by storage in the TFA consisting eluate for several hours, followed by a second preparative HPLC



purification. The addition of Hg(II) ions in guanidinylation reaction, as described in many published protocols^{14,19}, led to many by-products and only traces of the desired product.

Scheme 3.7. Synthesis of the bivalent guanidinothiazole containing ligands **3.34** and **3.35**. Reagents and conditions: i) Methyliodide, MeOH, reflux, 1.5 h, 100%; ii) Di-*tert*-butyl dicarbonate, TEA, CH_2Cl_2 , 0 °C to RT, ON, 73%; iii) TEA, CH_2Cl_2 , Ar-atmosphere, RT, ON, 88%; iv)/v) DIPEA, MeOH, reflux, 5-48 h, deprotection after preparative HPLC in eluate consisting of H_2O , MeCN and TFA, 3-11% over two steps.

3.2.2 Biological Evaluation

H₂R affinity, selectivity and antagonism

The monovalent guanidinothiazole containing ligands **3.8**, **3.10**, **3.12**, **3.13**, **3.21-3.26**, **3.29**, **3.30**, the bivalent guanidinothiazole containing ligands **3.34**, **3.35** and exemplary published H₂R ligands were investigated in equilibrium competition binding experiments on membrane preparations from Sf9 insect cells, expressing the hH₂R-G_{sαS} fusion protein, using either the antagonist [³H]UR-DE257¹¹ or [³H]tiotidine as radioligands. The selectivity of these compounds for the hH₂R compared to hH₃R was investigated by competition binding experiments using Sf9 insect cell membranes co-expressing the hH₃R and G_{αi2} and G_{β1γ2} proteins using [³H]N^α-methylhistamine as radioligand. The "cold" form of the radioligand **3.25**, its precursor **3.22** and the bivalent ligands **3.34** and **3.35** were additionally investigated on membrane preparations of Sf9 insect cells, co-expressing either the hH₁R-G_{sαS} fusion protein (radioligand: [³H]mepyramine) and RGS4 or hH₄R and G_{αi2} and G_{β1γ2} proteins (radioligand: [³H]UR-PI294²⁰) in order to determine a "selectivity profile" at all four H₂R subtypes. Selected displacement curves are shown in Figure 3.4 and the results are summarized in Table 3.1.

The cyanoguanidine-type guanidinothiazole containing amine precursor **3.12** showed lower hH_2R affinity with a p K_i value of 6.5 than the structurally related H_2R antagonist famotidine (p K_i value: 7.26). Propionylation was tolerated without any decrease of affinity (ligand **3.13**). The hH_2R affinity of the squaramide-type guanidinothiazole containing amine precursors **3.21-3.23** increased with the chain length of the n-alkandiyl linker (p K_i values: 5.94-8.42). The propionylated squaramide-type ligands **3.24-3.26** showed a high affinity at the hH_2R (p K_i values: 7.07-7.65). Interestingly, the affinities of **3.25** and **3.26** decreased and the affinity of **3.24** increased compared to the respective amine precursor. The conformationally constrained squaramide-type guanidinothiazole containing ligand **3.30** showed a decreased hH_2R affinity (p K_i value: 6.8) compared to the closely related, more flexible ligand **3.25** (p K_i value: 7.65) and the cyanoguanidine-type analog ICI127032 (**3.10**) (p K_i value: 7.70). The amine precursor **3.29** showed compared to **3.30** a clearly increased p K_i value of 7.31. Within the series of propionylated ligands ("cold" forms of potential radioligands) **3.25** showed the highest hH_2R affinity (p K_i value: 7.65) which was comparable to the affinity of UR-DE257 (p K_i value: 7.55).

The ligands **3.8**, **3.21-3.26**, **3.29** and **3.30** showed a low affinity at the hH₃R with pK_i values of 4.28-5.5 and **3.10** showed a very low affinity (pK_i value <4). The precursor **3.22** and the "cold" form of the radioligand **3.25** showed a low affinity to the hH₄R with pK_i values of 4.3 and 4.4. **3.22** bound with moderate affinity to the hH₁R (pK_i value: 6.12), whereas binding of the propionylated **3.25** was not detectable up to a concentration of 100 μ M.

The replacement of the amino(methyl)thiazolyl propyl head group of the bivalent ligand UR-NK22 with either a 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole residue (**3.34**) or a 2-guanidino-4-(3 aminophenyl)thiazole residue (**3.35**) resulted in high affinity hH_2R ligands with pK_i values of 7.3 and 7.14. Compared to the bivalent agonist UR-NK22 the hH_2R affinity (pK_i value: 8.07)¹⁴ and subtype selectivity was decreased.

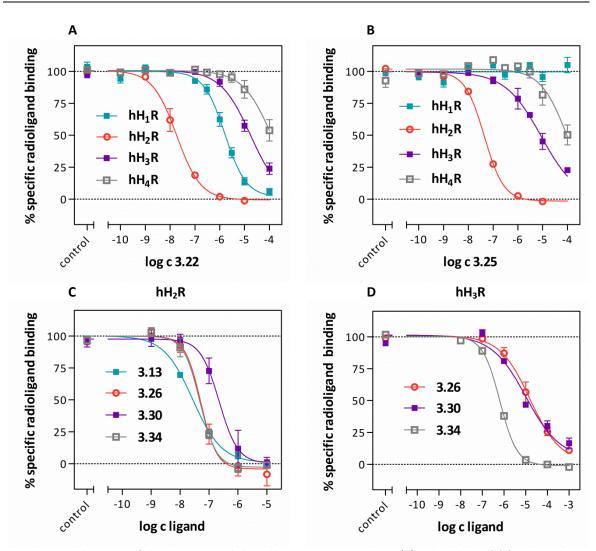


Figure 3.4. Displacement of the respective radioligand by amine precursor **3.22** (A) or ligand **3.25** (B) from membrane preparations of Sf9 insect cells co-expressing the hH₁R-G_{sαS} fusion protein and RGS4 (radioligand: [³H]mepyramine, c = 5 nM, $K_d = 4.5$ nM), expressing the hH₂R-G_{sαS} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, $K_d = 12.2$ nM), co-expressing the hH₃R and G_{αi2} plus G_{β1γ2} proteins (radioligand: [³H]UR-DE257, c = 3 nM, $K_d = 3$ nM) or co-expressing the hH₄R and G_{αi2} plus G_{β1γ2} proteins (radioligand: [³H]UR-PI294, c = 5 nM, $K_d = 5.1$ nM). Displacement of the respective radioligand from membrane preparations of Sf9 insect cells (C) expressing the hH₂R-G_{sαS} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, $K_d = 12.2$ nM or [³H]tiotidine, c = 10 nM, $K_d = 12.75$ nM) or co-expressing the hH₃R and G_{αi2} plus G_{β1γ2} proteins (radioligand: [³H]N^α-methylhistamine, c = 3 nM, $K_d = 3$ nM) by exemplary guanidinothiazoles. Data represent mean values ± SEM of 2-3 experiments performed in triplicate.

Table 3.1. Affinities of standard H₂R ligands, UR-DE257, UR-NK22, the monovalent guanidinothiazole containing ligands **3.8**, **3.10**, **3.12**, **3.13**, **3.21-3.26**, **3.29**, **3.30** and the bivalent guanidinothiazole containing ligands **3.34**, **3.35** to $hH_{1-4}R$, obtained from equilibrium competition binding studies on membrane preparations from Sf9 insect cells, expressing the respective histamine receptor subtype.

	hH₁Rª		hH₂R⁵		hH₃R ^d		hH₄R ^e	
No.	р <i>К</i> і	N	р <i>К</i> і	Ν	р <i>К</i> і	N	p <i>K</i> i	Ν
His	-	-	6.53 ± 0.04	3	7.8 ± 0.1	3	7.65 ± 0.03	3
UR-DE257	> 5.0 ¹¹	-	7.55 ¹¹	-	5.42 ¹¹	-	> 5.0 ¹¹	-
famotidine	-	-	7.26 ± 0.03	2	-	-	-	-
UR-NK22	6.06 ¹⁴	-	8.07 ¹⁴	-	5.94 ¹⁴		5.69 ¹⁴	-
3.8	n.d.	-	6.5 ± 0.1	3	4.28 ± 0.02	2	n.d.	-
3.10	n.d.	-	7.70 ± 0.07	3	> 4.0	2	n.d.	-
3.12	n.d.		6.5 ± 0.3	3	n.d.	-	n.d.	-
3.13	n.d.		6.4 ± 0.2	3	n.d.	-	n.d.	-
3.21	n.d.		5.94 ± 0.05	3	4.95 ± 0.03	2	n.d.	-
3.22	6.12 ± 0.08	3	8.0 ± 0.2	3	4.87 ± 0.09	3	4.3 ± 0.1^{f}	3
3.23	n.d.	-	$8.42 \pm 0.09^{\circ}$	4	5.5 ± 0.1	2	n.d.	-
3.24	n.d.	-	7.07 ± 0.09	3	5.2 ± 0.1	2	n.d.	-
3.25	> 4.0	3	7.65 ± 0.02	3	5.3 ± 0.1	3	4.4 ± 0.1^{f}	3
3.26	n.d.	-	$7.43 \pm 0.02^{\circ}$	3	5.08 ± 0.09	3	n.d.	-
3.29	n.d.	-	7.31 ± 0.07 ^c	3	5.3 ± 0.2	2	n.d.	-
3.30	n.d.	-	$6.8 \pm 0.1^{\circ}$	3	5.44 ± 0.07	3	n.d.	-
3.34	n.d.	-	7.3 ± 0.1 ^c	3	6.32 ± 0.02	4	6.21 ± 0.04	3
3.35	n.d.	-	7.14 ± 0.05	3	5.8 ± 0.1	3	6.09 ± 0.06	3

Competition binding assay on membrane preparations of Sf9 insect cells: ^aco-expression of the hH₁R-G_{sα5} fusion protein and RGS4 (radioligand: [³H]mepyramine, c = 5 nM, K_d = 4.5 nM), ^bexpression of the hH₂R-G_{sα5} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, K_d = 12.2 nM or ^c[³H]tiotidine, c = 10 nM, K_d = 12.75 nM), ^dco-expression of the hH₃R and G_{αi2} and G_{β1v2} proteins (radioligand: [³H]N^α-methylhistamine, c = 3 nM, K_d = 3 nM) or ^eco-expression of the hH₄R and G_{αi2} plus G_{β1v2} proteins (radioligand: [³H]N^α-methylhistamine c = 10 nM, K_d = 15.9 nM or ^f[³H]UR-PI294, c = 5 nM, K_d = 5.1 nM). The incubation period was 60 min. Data were analyzed by nonlinear regression and were best fitted to fourparameter sigmoidal concentration-response curves. Data shown are means ± SEM of N independent experiments, each performed in triplicate. The appreviation n.d. stands for not determined. The monovalent guanidinothiazole containing ligands **3.8**, **3.10**, **3.12**, **3.13**, **3.21-3.26**, **3.29**, **3.30** and the bivalent guanidinothiazole containing ligands **3.34**, **3.35** were investigated for hH_2R agonism in the GTP γ S binding assay on membrane preparations from Sf9 insect cells expressing the hH_2R - $G_{s\alpha S}$ fusion protein. Ligands which exhibited no agonism were also investigated in the antagonistic mode versus histamine as agonist. Selected curves are shown in Figure 3.5 and the results are summarized in Table 3.2.

The investigated monovalent and bivalent guanidinothiazole containing ligands were identified as antagonists or inverse agonists in the GTP γ S assay. All ligands were able to completely displace histamine, but only the p K_b values of the ligands **3.8**, **3.30** and **3.35** were in good agreement with the p K_i values. The p K_b values of the ligands **3.10**, **3.22-3.26**, **3.29** and **3.34** were considerably lower compared to the corresponding p K_i values. Interestingly, propionylation of the amine precursors **3.22** and **3.29** (p K_b value: 6.95 and 6.68) was tolerated with nearly no change of the p K_b values. Nonetheless, ligand **3.25** also showed the highest p K_b value (7.04) within the series of propionylated ligands ("cold" forms of potential radioligands), which was only slightly decreased compared to UR-DE257 (p K_b value: 7.42)¹¹.

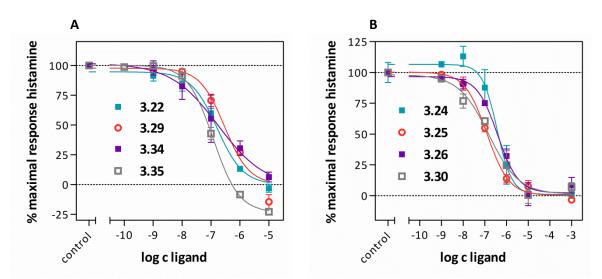


Figure 3.5. Antagonism of the guanidinothiazole containing ligands (A) **3.22**, **3.29**, **3.34**, **3.35** and (B) **3.24-3.26**, **3.30** on hH_2R determined in a GTP γ S assay (antagonistic mode) on membrane preparations of Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein. Histamine (1 μ M) was used for stimulation. Data represent mean values ± SEM of 2-4 experiments performed in triplicate.

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Table 3.2. hH₂R antagonism and the calculated pK_b values of UR-DE257, the monovalent guanidinothiazole containing ligands **3.8**, **3.10**, **3.12**, **3.13**, **3.21-3.26**, **3.29**, **3.30** and the bivalent guanidinothiazole containing ligands **3.34**, **3.35** determined by a GTP γ S assay.

hH₂R (GTPγS)					hH₂R (GTPγS)			
No.	р <i>ЕС</i> ₅₀ (р <i>К</i> _b)	Ν	α	No.	р <i>ЕС</i> ₅₀ (р <i>К</i> _b)	Ν	α	
His	5.80 ± 0.06	9	1.0	3.24	(6.6 ± 0.2)	3	-0.16	
UR-NK22	8.03 ¹⁴	-	0.92 ¹⁴	3.25	(7.06 ± 0.03)	3	-0.30	
UR-DE257	(7.42) ¹¹	-	0.08 ¹¹	3.26	(6.5 ± 0.1)	3	-0.11	
3.8	(6.9 ± 0.1)	4	-0.23	3.29	(6.68 ± 0.03)	2	-0.25	
3.10	(7.0 ± 0.2)	3	-0.33	3.30	(6.9 ± 0.2)	2	-0.18	
3.22	(6.95 ± 0.03)	3	-0.06	3.34	(7.0 ± 0.5)	2	-0.53	
3.23	(7.5 ± 0.2)	4	-0.07	3.35	(7.09 ± 0.08)	3	-0.21	

 $[^{35}S]$ GTP γ S assay performed with membrane preparations of Sf9 insect cells expressing the hH₂R-G_{saS} fusion protein. The intrinsic activity (α) of histamine was set to 1.00, and α values of investigated compounds were referred to this value. The p K_B values of neutral antagonists were determined in the antagonist mode versus histamine (c = 1 μ M) as agonist. Data represent mean values ± SEM of N experiments, each performed in triplicate.

Discrepancies between pK_i and pK_b value of antagonists were observed by several groups.^{7,21} Possible explanation are different experimental setups which led to varying access to the H₂R (e.g. guinea pig striatal membranes vs. intact isolated guinea pig heart)⁷ or, when a very similar setup was used (e.g. membrane preparations in both binding and functional studies), the use of different competitors (histamine vs. radiolabeled antagonist).²¹ Agonists and antagonists may stabilize different receptor conformations that exhibit different affinities for the investigated agonists/antagonists/inverse agonists.²¹ The antagonistic radioligand [³H]tiotidine was reported to bind only to a fraction of the functionally active H₂Rs.⁵

3.2.3 Chemical stability of 3.25

The ligand **3.25** showed the highest hH_2R affinity and subtype selectivity and was a good candidate for radiolabeling. Therefore the chemical stability of the "cold" radioligand **3.25** was investigated under radioligand storage conditions (EtOH/H₂O; 80:20; v/v) at a concentration of 15 μ M at room temperature in the dark (Figure 3.6). After 7 days RP-HPLC analysis showed no decomposition.

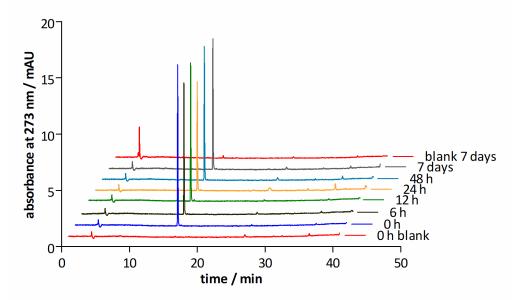
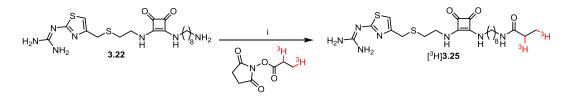


Figure 3.6. RP-HPLC analysis (λ : 273 nm) of **3.25** stock solution (t_R = 15.0 min) in EtOH/H₂O (80:20, v/v) after different times of incubation.

3.2.4 Radiosynthesis

Based on the pharmacological and chemical stability data presented, squaramide-type guanidinothiazole **3.25** was also synthesized with a tritium label ([³H]**3.25**/ [³H]UR-SB69) (Scheme 3.8). For this purpuse, an excess of the amine precursor **3.22** was acylated with the commercially available N-succinimidyl [2,3-³H]propionate in the presence of DIPEA. After purification by HPLC, the radioligand [³H]**3.25** was obtained in a radiochemical purity of 87% (Figure 3.7) with one impurity present. As the second peak ($t_R = 14.3 \text{ min}$, Figure 3.7B) is amounting to ca. 13% of the total peak area, determination of the specific activity of [³H]**3.25** was not feasible. Therefore, the specific activity of [³H]**3.25** was estimated based on the specific activity (2.41 TBq/mmol, 65.03 Ci/mmol) of [³H]UR-MK300¹⁹ prepared on the same day from the same lot of N-succinimidyl [2,3-³H]propionate. In order to prevent oxidation of the radioligand, the antioxidant ascorbinic acid was added to the radioligand stock solution (final concentration: 6.69 µmol/L [³H]**3.25** and 76.9 µmol/L ascorbinic acid in EtOH/ H₂O 80:20).



Scheme 3.8. Synthesis of the tritium-labeled squaramide-type guanidinothiazole [³H]**3.25** ([³H]UR-SB69). Reagents and conditions: i) DIPEA, anhydrous DMF, RT, 80 min, radiochemical yield: 64%.

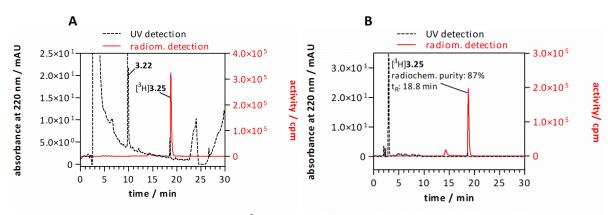


Figure 3.7. RP-HPLC analysis of the radioligand $[^{3}H]$ **3.25** (A) before and (B) after purification by RP-HPLC (conditions, see Experimental Section). UV chromatogram of $[^{3}H]$ **3.25** at 220 nm (black dashed line) and radiochromatogram of $[^{3}H]$ **3.25** (red line).

3.2.5 Biological Evaluation of [³H]3.25

Determination of binding constants of [³H]3.25

The radioligand [³H]**3.25** was characterized by saturation binding experiments on membrane preparations from Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein as well as on HEK293T- hH_2R -qs5 and HEK293T- hH_2R - β Arr2 cells in suspension, both stably expressing the hH_2R . Additionally, kinetic binding experiments were performed on membrane preparations from Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein. The results are summarized in Table 3.3. [³H]UR-SB69 bound in a saturable manner to both membranes and HEK cells (Figure 3.8).

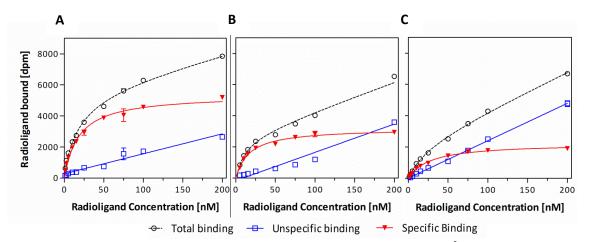


Figure 3.8. Representative saturation isotherms (red line) of specific hH_2R binding of $[{}^{3}H]$ **3.25** on (A) membrane preparations from Sf9 insect cells expressing the hH_2R - $G_{s\alpha S}$ fusion protein, (B) cell suspension of HEK293T- hH_2R -qs5 cells and (C) cell suspension of HEK293T- hH_2R - β Arr2 cells. Unspecific binding was determined in the presence of a 300 fold excess of famotidine. Specific binding was analyzed by a one-site binding equation. Error bars of specific binding and error bars of the Scatchard plot represent propagated errors calculated according to the Gaussian law. Error bars of total and unspecific binding represent the SEM. Experiments were performed in triplicate.

Unspecific binding was low when either membrane preparations or intact HEK293T-hH₂R-qs5 cells were used (<17% at the K_d value). Saturation binding studies performed with intact HEK293T-hH₂R- β Arr2 cells resulted in considerably higher unspecific binding (<42% at the K_d

value). The specific binding versus $[{}^{3}H]$ **3.25** concentration was best fitted by nonlinear regression to a one-site binding model and the unspecific binding to a linear curve. The determined K_{d} values (15-22 nM) were similar to the K_{i} value of 23 nM determined with $[{}^{3}H]$ UR-DE257 on membrane preparations of Sf9 cells.

Table 3.3. hH_2R binding characteristics of $[^3H]$ **3.25** determined on membrane preparations from Sf9 insect cells expressing the hH_2R - $G_{s\alpha S}$ fusion protein, HEK293T- hH_2R -qs5 or HEK293T- hH_2R - β Arr2 cells at 25 °C.

[³ H]3.25 ([³ H]UR-SB69)						
Receptor	<i>K</i> i ^d	K _{d (sat)} ^e	ر الاط (kin)	k on ^g	k _{off} ^h	
	[nM]	[nM]	[nM]	[min ⁻¹ ·nM ⁻¹]	[min ⁻¹]	
hH ₂ R	23 ± 1ª	$15 \pm 1^{a}/22 \pm 4^{b}/$ 19 ± 4 ^c	26.0 ± 0.3 ^a	0.00108 ± 0.00001 ^a	0.028 ± 0.002 ^a	

Radioligand binding assay determined on ^amembrane preparations from Sf9 insect cells expressing the hH₂R-G_{sas} fusion protein or cell suspension of ^bHEK293T-hH₂R-qs5 or ^cHEK293T-hH₂R-βArr2 cells. ^dDerived from competition binding with [³H]UR-DE257, c = 20 nM, K_d = 12.2 nM (cf. Table 3.1). ^eEquilibrium dissociation constant determined by saturation binding. ^fKinetically determined dissociation constant $K_{d (kin)}$ = k_{off}/k_{on}. ^gAssociation rate constant derived from nonlinear regression; calculated from k_{obs} , k_{off} and the radioligand concentration (c= 20 nM). ^hDissociation rate constant derived from nonlinear discociation. Data represent means ± SEM from two to four independent experiments each performed in duplicate (association) or triplicate.

The association and dissociation curves of $[{}^{3}H]$ **3.25** determined on membrane preparations of Sf9 insect cells expressing the hH₂R-G_{sαS} fusion protein are depicted in Figure 3.9. The association of the radioligand to the hH₂R was complete after 80 min and could be described by a monophasic fit with a k_{on} value of 0.00108 min⁻¹·nM⁻¹. Also linearization of the association curve revealed a straight line (k_{ob} value of 0.055 min⁻¹). The dissociation of $[{}^{3}H]$ **3.25** (c = 20 nM, 90 min pre-incubation) in the presence of famotidine was incomplete, reaching a plateau at 23% of the initially bound radioligand. These data suggested a (pseudo)irreversible binding.^{22,23}

Several GPCR ligands were reported to show a similar behavior in kinetic and functional experiments²²⁻²⁵ and several explanations were provided such as irreversible (covalent) binding to the receptor,²⁶ a slow rate of dissociation from the receptor,²² a slow rate of interconversion between inactive and active receptor conformations,²⁷ stabilization of an inactive ligand-specific receptor conformation,^{28,29} binding to a site distinct from the agonist binding site,³⁰ internalization of the ligand-receptor-complex²⁵. Nevertheless, the equilibrium dissociation constant $K_{d(kin)}$ = 26.0 nM, calculated from kinetics (nonlinear regression, $K_d = k_{off}/k_{on}$), were consistent with the K_d value obtained from saturation binding experiments (Table 3.3) and the pseudo-irreversible binding to the hH₂R was far less pronounced compared to the radioligand [³H]UR-DE257 (plateau at 60-70%).¹¹

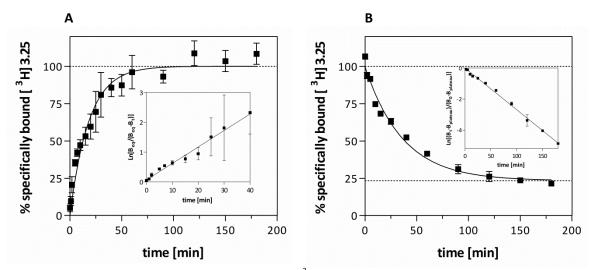


Figure 3.9. Association (A) and dissociation (B) kinetics of $[{}^{3}H]$ **3.25** determined at membrane preparations from Sf9 insect cells expressing the hH₂R-G_{sα5} fusion protein at room temperature. Association (c = 20 nM) to the hH₂R as a function of time (nonlinear regression: $k_{obs} = 0.049 \text{ min}^{-1}$). Inset: $\ln[B_{eq}/(B_{eq} - B_t)]$ versus time, $k_{obs} = \text{slope} = 0.055 \text{ min}^{-1}$. Dissociation (preincubation: 90 min, c = 20 nM) in the presence of famotidine (c = 3 μ M) from the hH₂R as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{off} = 0.028 \text{ min}^{-1}$, $t_{1/2} = 25 \text{ min}$, plateau = 23%), Inset: $\ln[(B_t-B_{plateau})/(B_0-B_{plateau})]$ versus time, slope-(-1) = $k_{off} = 0.027 \text{ min}^{-1}$. Data represent means ± SEM from two independent experiments each performed either in duplicate (association) or triplicate (dissociation).

Equilibrium competition binding experiments of [³H]3.25

 $[{}^{3}H]$ **3.25** was used in equilibrium competition binding experiments with membrane preparations of Sf9 cells (Table 3.4). $[{}^{3}H]$ **3.25** was completely displaceable by histamine and the standard H₂R antagonists famotidine and ICI127032 (Figure 3.10). Interestingly, lamtidine could only displace 75% of $[{}^{3}H]$ **3.25**, but completely displace the radioligand $[{}^{3}H]$ UR-DE257. The *pK*_i values were consistently lower but the order remained the same compared to the ones determined with $[{}^{3}H]$ UR-DE257.

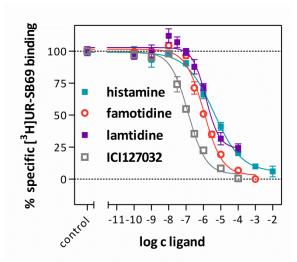


Figure 3.10. Displacement of the radioligand [³H]UR-SB69 ([³H]**3.25**, c = 20 nM, K_d = 15 nM) from membrane preparations of Sf9 insect cells expressing the hH₂R-G_{sαS} fusion protein by standard H₂R ligands. Data represent mean values ± SEM of 3 experiments performed in triplicate.

	p <i>K</i> _i values						
	[³ H]UR-DE257 ^a	Ν	[³ H]UR-SB69 ([³ H]3.25) ^b	Ν			
Histamine	6.53 ± 0.04	3	5.77 ± 0.08	3			
Famotidine	6.87 ¹¹	-	6.36 ± 0.03	3			
Lamtidine	6.8 ± 0.2	3	6.22 ± 0.07	3			
ICI127032 (3.10)	7.70 ± 0.07	3	7.26 ± 0.03	3			

Table 3.4. Affinities of the standard H_2R ligands to hH_2R , obtained from equilibrium competition binding studies on membrane preparations from Sf9 insect cells, expressing the hH_2R with the radioligands [³H]UR-DE257 and [³H]**3.25**.

Competition binding assay performed on membrane preparations of Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein. Radioligand: ${}^{a}[{}^{3}H]UR-DE257$, c = 20 nM, K_d = 12.2 nM or ${}^{b}[{}^{3}H]$ **3.25**, c = 20 nM, K_d = 15 nM). The incubation period was 60 min. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means ± SEM of N independent experiments, each performed in triplicate.

3.2.6 Chemical stability of [³H]3.25

The long-term stability of $[{}^{3}H]$ **3.25** in stock solution was investigated by RP-HPLC (conditions, see experimental section). The radiochemical purity of $[{}^{3}H]$ **3.25** after 15 month still amounted to 45% (Figure 3.11) with one major impurity present (t_{R} = 12.6 min).

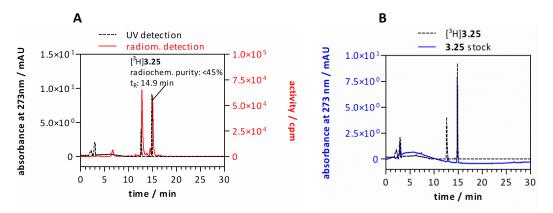


Figure 3.11. RP-HPLC analysis of the (A) radioligand stock solution [³H]**3.25** 15 month after purification (UV chromatogram at 273 nm (black dashed line) and radiochromatogram (red line)) and (B) radioligand stock solution [³H]**3.25** 15 month after purification (black dashed line) and stock solution of "cold" **3.25** in DMSO (blue line) (UV-chromatograms at 273 nm).

3.3 EXPERIMENTAL SECTION

3.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 (0.040-0.063 mm). Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution (0.8 g ninhydrin, 200 mL n-butanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA). Nuclear Magnetic Resonance (¹H NMR and ¹³C NMR) spectra were recorded on a Bruker Avance-300 (7.05 T, ¹H: 300 MHz, ¹³C: 75.5 MHz), Avance-400 (9.40 T, ¹H: 400 MHz, ¹³C: 100.6 MHz), or Avance-600 (14.1 T; ¹H: 600 MHz, ¹³C: 150.9 MHz; cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in δ (ppm) relative to external standards. Multiplicities are specified with the following abbreviations: s (singlet), d (doublet), t (triplet), q (quartet), qui (quintet), m (multiplet), br s (broad signal), as well as combinations thereof. In certain cases 2D-NMR techniques (COSY, HSQC, HMBC and NOESY) were used to assign ¹H and ¹³C chemical shifts. High-resolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass Q-TOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K-2001 detector. A Nucleodur 100-5 C18 (250 x 21 mm, 5 μm, Macherey-Nagel, Dueren, Germany), a Kinetex XB-C18 100A (250 x 21.2 mm, 5 µm, Phenomenex, Aschaffenburg, Germany) and a YMC Triart C18 (150 x 20 mm, 5 µm, YMC Europe GmbH, Dinslacken, Germany) served as RP-columns at a flow rate of 15 ml/min at room temperature. In case of the cyanoguanidines 3.10, 3.12 and 3.13 the mobile phase consisted of mixtures of CH₃CN and 0.1% aq. NH₃. Mixtures of CH₃CN and 0.05-0.1% aq. TFA were used as mobile phase for compounds 3.8, 3.21-3.26, 3.29, 3.30, 3.34, 3.35. A detection wavelength of 220 nm was used throughout. CH₃CN was removed from the eluates under reduced pressure (final pressure: 80 mbar) at 45 °C prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a vacuubrand RZ 6 rotary vane vacuum pump). Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A (250 x 4.6 mm, 5 µm, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of 0.05% TFA in CH₃CN (A) and 0.05% aq. TFA (B) were used as mobile phase. Helium degassing, room temperature, a flow rate of 0.8 mL/min and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration: $100-500 \mu$ M) were either prepared from stock solution (10 mM in DMSO) in a mixture of CH_3CN and H_2O corresponding to the initial eluent composition, or as a one to one mixture of the eluate (preparative HPLC) with Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, 3032 min: 80:20-95:5, 32-42 min: 95:5 or gradient 3: 0-30 min: A/B 15:85-90:10, 30-35 min: 90:10. Microanalysis was performed on a Vario micro cube (Elementar, Langenselbold, Germany).

3.3.2 Experimental protocols and analytical data

The synthesis of amidinothiourea³¹ (**3.1**) and N-succinimidyl propionate³² were described elsewhere.

2-Guanidino-4-chloromethylthiazole hydrochloride (3.2)³³

1,3-Dichloroacetone (1.08 g, 8.46 mmol, 1 eq) dissolved in acetone (4 mL) was added to a suspension of **3.1** (1.0 g, 8.46 mmol, 1 eq) in acetone (5 mL). After approximately 10 min most of the solid dissolved and the supernatant turned yellow. The reaction mixture was stirred over night at room temperature. The product precipitated as the HCl salt. Separation by filtration and washing with acetone (5 mL) afforded the product as a yellow solid (1.07 g, 56%). Mp: 176.9-181.7 °C decomposition (Lit.³³ mp: 191-193 °C). $R_f = 0.58$ (CH₂Cl₂/NH₃in MeOH 9:1). ¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 4.74 (s, 2H), 7.41 (s, 1H), 8.39 (s, 4H), 12.83 (s, 1H). ¹³C-NMR (75.5 MHz, [D₆]DMSO): δ (ppm) 40.7, 113.4, 147.2, 154.2, 160.3. HRMS: (ESI): m/z [M+H]⁺ calcd. for C₅H₈ClN₄S⁺: 191.0153, found: 191.0151. C₅H₇ClN₄S · HCl (190.65 + 36.46).

2-Guanidino-4-[(2-aminoethyl)thiomethyl]thiazole (3.3)³³

2-Aminoethanthiole hydrochloride (500 mg, 4.41 mmol, 2 eq) dissolved in EtOH (5 mL) were added dropwise to a sodium ethanolate solution (prepared from 250 mg Na in 8 mL EtOH) at 0 °C. The reaction mixture was stirred for 1.5 h at 0 °C. **3.2** (500 mg, 2.20 mmol, 1 eq) was added portion wise over 15 min at 0 °C. After stirring for overnight at room temperature, the solvent was evaporated under reduced pressure and the crude product was purified by column chromatography (eluent: $CH_2Cl_2/$ 3.5 M NH₃ in MeOH 90:10 isocratic). Removal of the solvent *in vacuo* afforded the product as yellow oil (430 mg, 84%). $R_f = 0.2$ ($CH_2Cl_2/$ 7 N NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 2.63-2.78 (m, 4H), 3.56 (s, 2H), 6.47 (s, 1H), 6.85 (br s, 4H). ¹³C-NMR (100 MHz, [D₄]MeOH): δ (ppm) 32.6, 35.2, 41.3, 106.8, 149.6, 159.0, 175.8. HRMS: (ESI) m/z [M+H]⁺ calcd. for $C_7H_{14}N_5S_2^+$: 232.0685, found: 232.0689. $C_7H_{13}N_5S_2$ (231.34).

2-(3-Acetylphenyl)isoindoline-1,3-dione (3.4)¹⁷

1-(3-Aminophenyl)ethan-1-one (1 g, 7.40 mmol, 1 eq) and phthalic anhydride (1.2 g, 8.14 mmol, 1.1 eq) were suspended in acetic acid (20 mL). The reaction mixture was stirred under reflux for 2.5 h and part of the solvent was evaporated under reduced pressure. Water (10 mL) was added and the resulting precipitate was filtered through a Buchner funnel. Removal of residual solvent *in vacuo* afforded the product as beige fine crystals (1.9 g, 96%). Mp: 180-184.5°C. $R_f = 0.24$ (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 2.62 (s, 3H), 7.58-7.68 (m, 2H), 7.77-7.83 (m,

2H), 7.92-8.00 (m, 3H), 8.03-8.05 (m, 1H). ¹³C-NMR (75.5 MHz, CDCl₃): δ (ppm) 26.8, 124.0, 126.6, 127.8, 129.5, 131.1, 131.6, 132.3, 134.7, 138.1, 167.1, 197.1. HRMS: (ESI): $m/z \ [M+H]^+$ calcd. for $C_{16}H_{12}NO_3^+$: 266.0812, found: 266.0814. $C_{16}H_{11}NO_3$ (265.27).

2-Guanidino-4-(3-phthalimidophenyl)thiazole hydrobromide (3.6)¹⁷

To a solution of **3.4** (4.36 g, 16.44 mmol, 1 eq) in CHCl₃ (50 mL) was added HBr solution in acetic acid (45 % w/v, 110 µL) under stirring. Bromine (5.52 g, 34.52 mmol, 2.1 eq) was added drop wise. The reaction mixture was stirred for 30 min at room temperature. Removal of the solvent *in vacuo* afforded the **3.5** as a white solid, which was applied to the next step without further purification. The crude **3.5** was dissolved in hot CH₃CN (100 mL) and poured in a hot solution of **3.1** (1.94 g, 16.44 mmol, 1 eq) in EtOH (100 mL). The reaction mixture was stirred under reflux for 5 h. Removal of the solvent *in vacuo* afforded a beige solid, which was suspended in EtOAc and filtered through a Buchner funnel. **3.6** was afforded as a beige solid (5.12 g, 70%). Mp: 311.9-315.4 °C decomposition (Lit.¹⁷ mp: >300 °C). $R_f = 0.7$ (CH₂Cl₂/ 7 N NH₃ in MeOH 6:1). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 7.43-7.45 (m, 1H), 7.60 (t, 1H, *J* 7.83 Hz), 7.84 (s, 1H), 7.92-7.95 (m, 2H), 7.97-8.00 (m, 2H), 8.03-8.05 (m, 2H), 8.25 (br s, 4H), 12.04 (br s, 1H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 109.3, 123.5, 125.2, 125.7, 127.5, 129.4, 131.5, 132.5, 133.8, 134.8, 148.9, 153.8, 160.0, 167.0. HRMS: (ESI): m/z [M+H]⁺ calcd. for C₁₈H₁₄N₅O₂S⁺: 364.0863, found: 364.0866. C₁₈H₁₃N₅O₂S · HBr (363.40 + 80.91).

2-Guanidino-4-(3-aminophenyl)thiazole (3.7)¹⁷

3.6 (5.09 g, 11.46 mmol, 1eq) was suspended in a mixture of concentrated hydrochloric acid (75 mL) and acetic acid (75mL) and the reaction mixture was stirred under reflux overnight. The solvent was removed under reduced pressure and the residue was suspended in aqueous NaOH solution (2.5 % w/w, 200 mL). The resulting precipitate was filtered through a Buchner funnel and washed with H₂O (50 mL). Recrystallisation from CH₃CN and removal of residual solvent *in vacuo* afforded the product as a brown solid (1.40 g, 52%). Mp: 219.3-223.1 °C decomposition (Lit.¹⁷ mp: 223-224 °C). *R*_f = 0.4 (CH₂Cl₂ / 7 N NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 5.04 (s, 2H), 6.47-6.50 (m, 1H), 6.92 (s, 1H), 6.95-6.97 (m, 1H), 7.00-7.02 (m, 1H), 7.03-7.04 (m, 1H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 102.1, 111.2, 113.1, 113.4, 128.8, 135.4, 148.6, 150.0, 156.9, 174.7. HRMS: (ESI): *m/z* [*M*+H]⁺ calcd. for C₁₀H₁₂N₅S⁺: 234.0808, found: 234.0807. C₁₀H₁₁N₅S (233.29).

2-Guanidino-4-[3-(3-methylthioureido)phenyl]thiazole hydrotrifluoracetate (3.8)¹⁷

Methylisothiocyanate (32 mg, 0.43 mmol, 1 eq) was added to a stirring solution of **3.7** (100 mg, 0.43 mmol, 1 eq) in acetone (2 mL). The reaction mixture was stirred overnight at room temperature. The solvent was removed *in vacuo* and the residue was purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05% aq. TFA 10:90-55:45, t_R = 14.93 min). The TFA-salt was obtained as a white solid (70 mg; 53%). Mp: 204.2 °C decomposition (Lit.¹⁷

oxalate mp: 219-222°C, decomposition). $R_f = 0.25$ (CH₂Cl₂/7M NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm) (TFA-Salz): 95.8% ($t_R = 13.67$ min, k = 3.7). ¹H-NMR (600 MHz, [D₆]DMSO): δ (ppm) 2.92-2.93 (m, 3H), 7.32-7.33 (m, 1H), 7.36-7.39 (m, 1H), 7.65-7.66 (m, 1H), 7.69 (s, 1H), 7.81 (br s, 1H), 7.98 (s, 1H), 8.36 (br s, 4H), 9.65 (br s, 1H), 12.30 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO): δ (ppm) 31.3, 108.3, 120.6, 121.7, 123.1, 129.1, 133.5, 139.7, 149.5, 154.2, 158.6 (q, J 34 Hz, TFA), 160.6, 181.3. HRMS: (ESI): m/z [M+H]⁺ calcd. for C₁₂H₁₅N₆S₂⁺: 307.0794, found: 307.0797. C₁₂H₁₄N₆S₂ · C₂HF₃O₂ (306.41 + 114.02).

4-[3-(2-Cyano-3-methylguanidino)phenyl]thiazole / ICI127032 (3.10)¹⁷

Diphenylcyanocarbonimidate (112 mg, 0.47 mmol, 1.1 eq) was dissolved in 2-propanol (10 mL). **3.7** (100 mg, 0.43 mmol, 1 eq) was added and the reaction mixture was stirred overnight at room temperature. The solvent was removed under reduced pressure, diethylether was added and solvent was again removed under reduced pressure. The solid (crude **3.9**) was dissolved in aqueous methylamine solution (40% w/w, 5mL) and the reaction mixture was stirred overnight at room temperature. The solvent was removed *in vacuo* and the residue was purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 10:90-40:60, $t_{\rm R}$ = 20.51 min). **3.10** was obtained as a white solid (74.5 mg; 55%). Mp: 167.4-192.4 °C. $R_{\rm f}$ = 0.25 (CH₂Cl₂/7M NH₃ in MeOH 9:1). RP-HPLC (gradient 3, 220 nm: 96.7% ($t_{\rm R}$ = 10.32 min, k = 3.1). ¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 2.78 (d, 3H, *J* 4.48 Hz), 6.93 (br s, 4H), 7.16-7.21 (m, 3H), 7.35 (t, 1H, *J* 7.97 Hz), 7.60-7.63 (m, 1H), 7.71 (s, 1H), 8.94 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO): δ (ppm) 28.8, 103.6, 117.5, 121.1, 122.0 122.8, 129.0, 135.6, 138.0, 148.8, 157.1, 158.9, 175.2. HRMS: (ESI): m/z [M+H]⁺ calcd. for C₁₃H₁₅N₈S⁺: 315.1135, found: 315.1139. C₁₃H₁₄N₈S (314.37).

1-(8-Aminooctyl)-2-cyano-3-(2-[2-(diaminomethyleneamino)thiazol-4ylmethylthio]ethyl)guanidine (3.12)

Triethylamine (200 mg, 1.97 mmol, 4 eq) was added to a suspension of **3.3** as hydrochloride (150 mg, 0.49 mmol, 1 eq) in methanol (45 mL). Diphenyl-*N*-cyanocarbonimidate (118 mg, 0.49 mmol, 1 eq) was added and the reaction mixture was stirred overnight at room temperature. The solvent was partially removed under reduced pressure and H₂O (10 mL) was added. The product was extracted with EtOAc (3 x 15 mL), the organic layers were combined and dried over sodium sulfate. Removal of the solvent *in vacuo* afforded **3.11** as yellow oil which was directly used for the next step. The crude **3.11** and octan-1,8-diamine (285 mg, 1.97 mmol, 3 eq) were suspended in MeCN (1 mL) and the reaction mixture was stirred for 3 h at 50 °C. The solvent was removed *in vacuo* and the residue was purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/1% aq. NH₃ 15:85-65:35, $t_R = 16.7$ min). **3.12** was obtained as white, highly hygroscopic solid (130 mg, 62.0%). $R_f = 0.3$ (CH₂Cl₂/7M NH₃ in MeOH 6:1). RP-HPLC (gradient 2, 220 nm): 95.7% ($t_R = 12.74$ min, k = 4.1). ¹H-NMR (600 MHz, CD₃OD): δ (ppm) 1.35 (br s, 8H), 1.51-1.57 (m, 4H), 2.66-2.74 (m, 4H) 3.18 (t, 2H, *J* 7.1 Hz), 3.39 (t, 2H, *J* 6.97 Hz), 3.68 (s, 2H), 6.54 (s, 1H). ¹³C-NMR (150 MHz, CD₃OD): δ (ppm) 22.1, 27.6, 27.7, 30.2, 30.28, 30.33, 31.8, 32.7, 41.9,

42.0, 42.7, 107.0, 120.1, 149.5, 159.1, 161.1, 175.9. HRMS: (ESI): $m/z [M+H]^+$ calcd. for $C_{17}H_{32}N_9S_2^+$: 426.2217, found: 426.2215. $C_{17}H_{31}N_9S_2$ (425.21).

N-(8-[2-Cyano-3-(2-[2-(diaminomethylenamino)thiazol-4-ylmethylthio]ethyl)guanidine]octyl)propionamide (3.13)

Triethylamine (79 mg, 0.78 mmol, 3 eq) and N-succinimidyl propionate (34 mg, 0.29 mmol, 1.1 eq) were added to a suspension of **3.12** (111 mg, 0.26 mmol, 1eq) in CH₂Cl₂ (1 mL) and DMF (1 mL).The reaction mixture was stirred for 3 h at room temperature. The solvent was partially removed *in vacuo* and the residue was purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/1% aq. NH₃ 15:85-80:20, $t_R = 19.0$ min). **3.13** was obtained as white, hygroscopic solid (28 mg, 22.3%). $R_f = 0.4$ (CH₂Cl₂/7M NH₃ in MeOH 6:1). IR (KBr): 3355, 3330, 3180, 2930, 2850, 2160, 1635, 1585, 1555, 1460, 1355, 1250, 1170, 1000, 850, 640, 605 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 95.9% ($t_R = 16.82$ min, k = 5.7). ¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.12 (t, 3H, *J* 7.6 Hz), 1.34 (br s, 8H), 1.47-1.57 (m, 4H), 2.15-2.21 (q, 2H, *J* 7.7 Hz), 2.68 (t, 2H, *J* 6.9 Hz), 3.13-3.19 (m, 4H), 3.38 (t, 2H, *J* 6.8 Hz), 3.68 (s, 2H), 6.54 (s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 10.6, 27.7, 27.9, 30.2, 30.26 (2C), 30.29, 30.4, 31.8, 32.7, 40.4, 42.0, 42.8, 107.0, 120.1, 149.5, 159.1, 161.1, 175.9, 177.0. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₀H₃₆N₉S₂O⁺: 482.2479, found: 482.2484. Anal. calcd. for C₂₀H₃₅N₉S₂O (2 49.87, H 7.32, N 26.17, S 13.31, found: C 49.56, H 7.19, N 26.24, S 12.88. C₂₀H₃₅N₉S₂O (481.58).

2-(4-[2-(2-Ethoxy-3,4-dioxocyclobut-1-ene-1-ylamino)ethylthiomethyl]thiazol-2-yl)guanidine (3.14)³⁴

3,4-Diethoxycyclobut-3-ene-1,2-dione (246 mg, 1.45 mmol, 1.1 eq) was dissolved in EtOH (15 mL). 3.3 as hydrochloride (400 mg, 1.32 mmol, 1 eq) and triethylamine (533 mg, 5.26 mmol, 4 eq) dissolved in EtOH (15 mL) were added drop wise under stirring. The reaction mixture was stirred over night at room temperature. The precipitated product was filtered off and washed with EtOH. Some product was still contained in the supernatant. EtOH was removed in vacuo and the residue was purified by column chromatography (eluent: $CH_2Cl_2/7$ M NH₃ in MeOH 95:5 isocratic). The precipitated product and the column purified product were combined and the solvent was removed in vacuo. 3.14 was obtained as beige solid (408 mg, 87%). Mp: 187.5-189.9 °C decomposition (Lit.³⁴ mp >135 °C decomposition). $R_{\rm f}$ = 0.5 (CH₂Cl₂/7M NH₃ in MeOH 6:1).¹H-NMR (400 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.34-1.39 (m, 3H), 2.63-2.68 (m, 2H), 3.46-3.48 (m, 1H), 3.61 (s, 2H), 3.67-3.68 (m, 1H), 4.63-4.68 (q, 2H, J 7.0 Hz), 6.47 (s, 1H), 6.83 (br s, 3.7H), 8.66-8.85 (d, 1H, J 75.8 Hz). ¹³C-NMR (100 MHz, CD₃OD): δ (ppm) 15.5 (-OCH₂CH₃), 31.0 (d, J 20 Hz, thiazolyl-CH2), 31.4 (d, J 34 Hz, -SCH2CH2NH-), 43.0 (d, J 68 Hz, -SCH2CH2NH-), 68.8 (d, J 9 Hz, -OCH₂CH₃), 104.4 (d, *J* 10 Hz, C⁵ thiazolyl), 147.4 (d, *J* 11 Hz, C⁴ thiazolyl), 156.8 ((NH₂)₂C=N-), 172.4 (d, J 32 Hz, cyclobutendion), 175.4 (**C**² thiazolyl), 176.7 (d, J 21 Hz, cyclobutendion), 182.0 (d, J 19 Hz, **C**=O cyclobutendion), 189.2 (d, J 31 Hz, **C**=O cyclobutendion). HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₁₃H₁₈N₅S₂O₃⁺: 356.0846, found: 356.0847. C₁₃H₁₇N₅S₂O₃ (355.43).

General procedure for the synthesis of the Mono-Boc-protected diamines (3.15-3.17)

The corresponding alkane diamine (2 eq) was dissolved in chloroform (30 mL). Di-*tert*-butyl dicarbonate (1 eq) was dissolved in chloroform (25 mL) and added drop wise over a period of 3 h under ice-cooling. The reaction mixture was allowed to warm up to room temperature while stirring overnight. Potentially precipitated alkane diamine was filtered off. The organic layer was washed with alkaline saturated NaCl solution (45 mL sat. aq. NaCl and 1 mL 1 M aq. NaOH), saturated NaCl solution (45 mL) and H₂O (45 mL). The organic layer was dried over sodium sulphate and concentrated by evaporation under reduced pressure. The residue was purified by column chromatography (eluent: $CH_2Cl_2/2$ M NH₃ in MeOH 97.5:2.5-90:10).

tert-Butyl N-(2-aminoethyl)carbamate (3.15)³⁵⁻³⁸

Ethylendiamine (1.5 g, 25.0 mmol, 2 eq) was treated with di-*tert*-butyldicarbonate (2.72 g, 12.5 mmol, 1 eq) according to the general procedure. Removal of the solvent *in vacuo* afforded the product as slightly yellow oil (1.25 g, 62.5%). $R_f = 0.5$ (CH₂Cl₂/3 M NH₃ in MeOH 5:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.34 (s, 2H), 1.38 (s, 9H), 2.73 (t, 2H, *J* 5.6 Hz), 3.10 (m, 2H), 5.11 (br s, 1H,). HRMS (ESI): *m/z* [*M*+H]⁺ calcd. for C₇H₁₇N₂O₂⁺: 161.1285, found: 161.1284. C₇H₁₆N₂O₂ (160.22).

tert-Butyl N-(8-aminooctyl)carbamate (3.16)^{38,39}

Octane-1,8-diamine (2.0 g, 13.9 mmol, 2 eq) was treated with di-*tert*-butyldicarbonate (1.5 g, 7.0 mmol, 1 eq) according to the general procedure. Removal of the solvent *in vacuo* afforded the product as colorless oil (700 mg, 41.2%). $R_f = 0.3$ (CH₂Cl₂/3.5 M NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.26 (br s, 8H), 1.30 (br s, 2H), 1.38-1.41 (m, 13H), 2.62-2.66 (t, 2H; *J* 7.0 Hz), 3.04-3.09 (m, 2H), 4.58 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 26.8, 26.9, 28.5, 29.3, 29.5, 30.1, 33.9, 40.7, 42.3, 79.0, 156.1. HRMS (ESI): $m/z [M+H]^+$ calcd. for C₁₃H₂₉N₂O₂⁺: 245.2224, found: 245.2225. C₁₃H₂₈N₂O₂ (244.38).

tert-Butyl N-(10-aminodecyl)carbamate (3.17)³⁸

Decane-1,10-diamine (2.0 g, 11.6 mmol, 2 eq) was treated with di-*tert*-butyldicarbonate (1.3 g, 5.8 mmol, 1 eq) according to the general procedure. Removal of the solvent *in vacuo* afforded the product as colorless oil (820 mg, 51.2%). $R_{\rm f}$ = 0.6 (CH₂Cl₂/7 M NH₃ in MeOH 7: 1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.26 (br s, 12H), 1.42-1.46 (m, 13H), 1.67 (br s, 2H), 2.67 (t, 2H, *J* 7.1 Hz), 3.06-3.10 (m, 2H), 4.53 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 26.9, 27.0, 28.6, 29.4, 29.55, 29.60, 29.62, 30.2, 33.7, 40.8, 42.3, 79.1, 156.1. HRMS (ESI): *m/z* [*M*+H]⁺ calcd. for C₁₅H₃₃N₂O₂⁺: 273.2537, found: 273.2537. C₁₅H₃₂N₂O₂ (272.43).

tert-Butyl [2-(2-[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)ethyl]carbamate (3.18)

Compound **3.14** (150 mg, 0.42 mmol, 1 eq) was dissolved in EtOH (10 mL) and a solution of **3.15** (74 mg, 0.46 mmol, 1.1 eq) in EtOH (10 mL) was added under stirring. The reaction mixture was stirred over night at room temperature. Due to incomplete conversion triethylamine (214 mg, 2.11 mmol, 5 eq) was added and the reaction mixture was stirred at 70 °C for 6 h. The solution was allowed to cool to room temperature and stirred at this temperature over night. Removal of the solvent *in vacuo* and purification by column chromatography (eluent: $CH_2Cl_2 / 7$ M NH₃ in MeOH 95:5- 80:20) afforded the product as white solid (158 mg, 79.2%). Mp: 187.7-195.6 °C decomposition. $R_f = 0.2$ ($CH_2Cl_2/7$ M NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, [D₆]-DMSO): δ (ppm) 1.36 (s, 9H), 2.64-2.67 (t, 2H, *J* 6.7 Hz), 3.07-3.11 (q, 2H, *J* 5.9 Hz), 3.50 (br s, 2H), 3.66-3.70 (m, 4H), 6.62 (s, 1H), 6.88-7.09 (br s, 5H), 7.58 (br s, 2H). HRMS: (ESI): m/z [M+H]⁺, calcd. for $C_{18}H_{28}N_7O_4S_2^+$: 470.1639, found: 470.1643. $C_{18}H_{27}N_7O_4S_2$ (469.58).

tert-Butyl [8-(2-[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]carbamate (3.19)

3.14 (163 mg, 0.46 mmol, 1 eq), **3.16** (124 mg, 0.50 mmol, 1.1 eq) and triethylamine (232 mg, 2.29 mmol, 5 eq) were dissolved in EtOH (20 mL). The reaction mixture was stirred for 6 h at 70 °C. Removal of the solvent *in vacuo* and purification by column chromatography (eluent: CH₂Cl₂ / 7 M NH₃ in MeOH 95:5- 90:10) afforded the product as white solid (224 mg, 88.1%). Mp: 174.6-177.9 °C decomposition. $R_f = 0.5$ (CH₂Cl₂/3 M NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, [D₆]-DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 1.23-1.26 (m, 8H), 1.33-1.36 (m, 11H), 1.48-1.52 (m, 2H), 2.65-2.68 (t, 2H, *J* 6.6 Hz), 2.85-2.90 (q, 2H, *J* 6.6 Hz), 3.48 (br s, 2H), 3.63 (s, 2H), 3.69-3.71 (m, 2H), 6.50 (s, 1H), 6.72-6.75 (m, 1H), 6.84 (br s, 4H), 7.45 (br s, 2H). ¹³C-NMR (100 MHz, [D₆]-DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 25.8, 26.2, 28.3, 28.6, 28.7, 29.4, 30.7, 31.1, 32.3, 40.2, 42.6, 43.3, 77.3, 104.5, 147.5, 155.6, 156.9, 167.5, 167.8, 175.3, 182.3, 182.5. HRMS: (ESI): $m/z [M+H]^+$, calcd. for C₂₄H₄₀N₇O₄S₂⁺: 554.2578, found: 554.2584. C₂₄H₃₉N₇O₄S₂ (553.74).

tert-Butyl [10-(2-[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)decyl]carbamate (3.20)

3.14 (155 mg, 0.41 mmol, 1 eq), **3.17** (124 mg, 0.45 mmol, 1.1 eq) and triethylamine (209 mg, 2.06 mmol, 5 eq) were suspended in EtOH (20 mL). The reaction mixture was stirred overnight at 60 °C. Removal of the solvent *in vacuo* and purification by column chromatography (eluent: CH₂Cl₂/3.5 M NH₃ in MeOH 95:5 isocratic) afforded the product as white solid (150 mg, 63%). Mp: 178.7-182.1 °C decomposition. $R_{\rm f}$ = 0.45 (CH₂Cl₂/7 M NH₃ in MeOH 6: 1). ¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 1.23-1.25 (m, 14H), 1.36 (s, 9H), 1.48-1.52 (m, 2H), 2.65 (t, 2H, *J* 6.26 Hz), 2.84-2.91 (m, 2H), 3.48 (br s, 2H), 3.63 (s, 2H), 3.69-3.71 (m, 2H), 6.49 (s, 1H), 6.74-6.84 (m, 5H), 7.47 (br s, 2H). ¹³C-NMR (300 MHz, [D₆]DMSO): δ (ppm) 25.9, 26.3, 28.3, 28.67, 28.74, 28.96, 29.00, 29.5, 30.8, 31.1 32.3, 1C under solvent peak (38.7-40.3), 42.6, 43.3, 77.3, 104.5, 147.5,

155.6, 156.9, 167.5, 167.9, 175.5, 182.3, 182.5. HRMS: (ESI): $m/z [M+H]^+$, calcd. for C₂₆H₄₄N₇O₄S₂⁺: 582.2891, found: 582.2895. C₂₆H₄₃N₇O₄S₂ (581.80).

General procedure for the Boc-deprotection of 3.18-3.20 to the free amines 3.21-3.23

The Boc-protected amine **3.18**, **3.19** or **3.20** (1 eq) was dissolved in a mixture of CH_2Cl_2 (2 mL) and TFA (0.3 mL) and stirred for 1.5 h at room temperature. The solvent was removed *in vacuo*, CH_2Cl_2 was added and then again removed. This process was repeated several times in order to remove residual TFA. Part of the product was directly used in the next step of the synthesis and a part was purified by preparative HPLC.

1-[4-(2-[2-(2-Aminoethylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]—ethylthiomethyl)thiazol-2-yl]guanidine-tri(hydrotrifluoracetate) (3.21)

3.18 (150 mg, 0.32 mmol, 1 eq) was treated according to the general procedure. The crude product was obtained as white hygroscopic solid (212 mg). Further purification of 100 mg by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-30:70, t_R = 10.6 min) afforded the product as white hygroscopic solid (54 mg, 50.4%). R_f = 0.3 (CH₂Cl₂/7 M NH₃ in MeOH 7:1). RP-HPLC (gradient 2, 220 nm): 91.0% (t_R = 8.67 min, k = 2.4). ¹H-NMR (400 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 2.67 (t, 2H, *J* 6.8 Hz), 3.02 (br s, 2H), 3.69-3.71 (m, 4H), 3.79 (s, 2H), 7.12 (s, 1H), 7.78-7.90 (br s, 5H), 8.40 (br s, 4H), 12.42 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 30.0, 32.1, 39.6, 40.9, 42.6, 110.0, 148.5, 154.3, 158.7 (q, *J* 32 Hz, TFA), 160.7, 167.9, 168.2, 182.4, 182.9. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₃H₂₀N₇O₂S₂⁺: 370.1114, found: 370.1114. C₁₃H₁₉N₇O₂S₂ · C₆H₃F₉O₆ (369.46 + 342.07).

1-[4-(2-[2-(8-Aminooctylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]ethylthiomethyl)thiazol-2yl]guanidine-tri(hydrotrifluoracetate) (3.22)

3.19 (190 mg, 0.34 mmol, 1 eq) was treated according to the general procedure. The crude product was obtained as white hygroscopic solid (255 mg). Further purification of 148 mg by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-40:60, t_R = 11.0 min) afforded the product as white hygroscopic solid (91 mg, 57.3%). R_f = 0.2 (CH₂Cl₂/7 M NH₃ in MeOH 7:1). RP-HPLC (gradient 1, 220 nm): 95.4% (t_R = 13.22 min, k = 4.3). ¹H-NMR (600 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 1.23-1.27 (br s, 8H), 1.49-1.52 (m, 4H), 2.66-2.68 (t, 2H, *J* 6.8 Hz), 2.74-2.79 (m, 2H), 3.50 (br s, 2H), 3.70 (br s, 2H), 3.79 (s, 2H), 7.14 (s, 1H), 7.69 (br s, 5H), 8.46 (br s, 4H), 12.44 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 25.7, 26.9, 27.0, 28.37, 28.41, 30.0, 30.7, 32.2, 38.7, 42.5, 43.2, 110.2, 116.7 (q, *J* 297 Hz, TFA), 148.6, 154.2, 158.6 (q, *J* 33 Hz, TFA), 160.2, 167.6, 167.7, 182.3, 182.4. HRMS: (ESI): m/z [*M*+H]⁺, calcd. for C₁₉H₃₂N₇O₂S₂⁺: 454.2053, found: 454.2049. C₁₉H₃₁N₇O₂S₂ · C₆H₃F₉O₆ (453.62 + 342.07).

1-[4-(2-[2-(10-Aminodecylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]ethylthiomethyl)thiazol-2-yl]guanidine-tri(hydrotrifluoracetate) (3.23)

3.20 (50 mg, 0.086 mmol, 1 eq) was treated according to the general procedure. Purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-70:30, t_R = 11.5 min) afforded the product as white solid (60.4 mg, 85.4%). Mp: 98.1-100.8 °C. R_f = 0.2 (CH₂Cl₂/7 M NH₃ in MeOH 6:1). RP-HPLC (gradient 2, 220 nm): 96.8% (t_R = 13.82 min, k = 4.5). ¹H-NMR (300 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 1.24 (br s, 12H), 1.46-1.53 (m, 4H), 2.64-2.68 (m, 2H), 2.72-2.79 (m, 2H), 3.48 (br s, 2H), 3.69-3.71 (m, 2H), 3.79 (s, 2H), 7.13 (s, 1H), 7.76 (br s, 5H), 8.59 (s, 4H), 12.73 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 25.77, 25.84, 27.0, 28.5, 28.6, 28.77, 28.83, 30.0, 30.7, 32.2, 38.8, 42.5, 43.3, 110.1, 116.4 (q, *J* 296Hz, TFA), 148.5, 154.4, 159.1 (q, *J* 34 Hz, TFA), 160.3, 167.6, 167.9, 182.3, 182.5. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂₁H₃₆N₇O₂S₂⁺: 482.2366, found: 482.2364. Anal. calcd. for C₂₁H₃₅N₇O₂S₂ · C₆H₃F₉O₆: C 39.37, H 4.68, N 11.90, S 7.78, found: C 39.16, H 4.86, N 11.86, S 7.76. C₂₁H₃₅N₇O₂S₂ · C₆H₃F₉O₆ (481.68+ 342.07).

N-[2-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)ethyl]propionamide bis(hydrotrifluoracetate) (3.24)

3.21 (100 mg, 0.17 mmol, 1 eq) and triethylamine (68 mg, 0.67 mmol, 4 eq) were suspended in CH₂Cl₂ (2.5 mL). After stirring for 5 min N-succinimidyl propionate (32 mg, 0.18 mmol, 1.1 eq) was added and the reaction mixture was stirred over night at room temperature. The conversion was incomplete due to the poor solubility of the educts in the solvent. DMF (2 mL) was added and the mixture was stirred for 10 min at 50°C to dissolve the educts. The reaction mixture was allowed to cool to room temperature while stirring for additional 2 h. Removal of the solvent in vacuo and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-45:55, $t_{\rm R}$ = 9.5 min) afforded the product as white hygroscopic solid (48 mg, 52.8%). $R_{\rm f}$ = 0.3 (CH₂Cl₂/7 M NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 97.3% (t_R = 10.80 min, k = 3.3). ¹H-NMR (400 MHz, [D₆]-DMSO): δ (ppm) 0.97 (t, 3H, J 7.5 Hz), 2.03-2.09 (q, 2H, J 7.6 Hz), 2.67 (t, 2H, J 6.7 Hz), 3.19-3.21 (m, 2H), 3.52 (br s, 2H, interfering with the water signal), 3.69-3.70 (m, 2H, interfering with the water signal), 3.79 (s, 2H), 7.14 (s, 1H), 7.59 (br s, 2H), 7.92 (br s, 1H), 8.38 (br s, 4H), 12.29 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO): δ (ppm) 9.8, 28.4, 30.0, 32.1, 39.7, 42.5, 42.9, 110.2, 116.8 (q, J 298 Hz, TFA), 148.6, 154.1, 158.8(q, J 33 Hz, TFA), 160.2, 167.6, 168.1, 173.3, 182.4, 182.6. HRMS: (ESI): $m/z [M+H]^+$, calcd. for $C_{16}H_{24}N_7O_3S_2^+$: 426.1377, found: 426.1377. $C_{16}H_{23}N_7O_3S_2 \cdot C_4H_2F_6O_4$ (425.53 + 228.04).

N-[8-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]propionamide bis(hydrotrifluoracetate) (3.25)

3.22 (100 mg, 0.15 mmol, 1 eq) and triethylamine (59 mg, 0.59 mmol, 4 eq) were suspended in CH_2Cl_2 (2.5 mL). After stirring for 5 min N-succinimidyl propionate (19 mg, 0.16 mmol, 1.1 eq) was added and the reaction mixture was stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min:

MeCN/0.1% aq. TFA 15:85-40:60, $t_{\rm R}$ = 14.6 min) afforded the product as white hygroscopic solid (66 mg, 72%). $R_{\rm f}$ = 0.6 (CH₂Cl₂/7 M NH₃ in MeOH 7:1). RP-HPLC (gradient 2, 220 nm): 98.8% ($t_{\rm R}$ = 15.21 min, k = 5.1). ¹H-NMR (400 MHz, [D₆]-DMSO): δ (ppm) 0.97 (t, 3H, J 7.6 Hz), 1.19-1.31 (m, 8H), 1.32-1.39 (m, 2H), 1.45-1.55 (m, 2H), 2.04 (q, 2H, J 7.6 Hz), 2.67 (t, 2H, J 6.7 Hz), 2.97-3.02 (m, 2H), 3.49 (br s, 2H, interfering with the water signal), 3.70 (br s, 2H), 3.79 (s, 2H), 7.14 (br s, 1H), 7.53 (br s, 2H), 7.68 (br s, 1H), 8.36 (br s, 4H), 12.24 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 10.0, 25.8, 26.4, 28.5, 28.6, 28.7, 29.1, 30.0, 30.7, 32.2, 38.4, 42.5, 43.3, 110.3, 116.6 (q, *J* 296 Hz, TFA), 148.6, 154.1, 158.7 (q, *J* 33 Hz, TFA), 160.2, 167.5, 167.9, 172.6, 182.3, 182.4. HRMS: (ESI): m/z [*M*+H]⁺, calcd. for C₂₂H₃₆N₇O₃S₂⁺: 510.2316, found: 510.2322. C₂₂H₃₅N₇O₃S₂ · C₄H₂F₆O₄ (509.69 + 228.04).

N-[10-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)decyl]propionamide-bis(hydrotrifluoracetate) (3.26)

3.23 (144 mg, 0.20 mmol, 1 eq), N-succinimidyl propionate (26 mg, 0.22 mmol, 1.1 eq) and triethylamine (82 mg, 0.81 mmol, 4 eq) were dissolved in a mixture of CH_2Cl_2 (3.5 mL) and DMF (2 mL). The reaction mixture was stirred for 2 h at room temperature. The conversion was incomplete due to the poor solubility of the educts in the solvent. The mixture was stirred for 1 h at 60°C to dissolve the educts. The reaction mixture was allowed to cool to room temperature while stirring for additional 48 h. Removal of the solvent in vacuo and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 25:75-75:25, $t_{\rm R}$ = 12.08 min) afforded the product as white fluffy solid (43 mg, 32.5%). Mp: 149.2-153.0 °C. R_f = 0.52 (CH₂Cl₂/7 M NH₃ in MeOH 6:1). RP-HPLC (gradient 2, 220 nm): 98.8% (t_R = 18.05 min, k = 6.2). ¹H-NMR (400 MHz, [D₆]-DMSO): δ (ppm) 0.96 (t, 3H, J 7.60 Hz), 1.22-1.25 (m, 12H), 1.33-1.36 (m, 2H), 1.48-1.49 (m, 2H), 2.02 (q, 2H, J 7.60 Hz), 2.65-2.67 (m, 2H), 2.97-3.00 (m, 2H), 3.48 (br s, 2H), 3.69 (br s, 2H), 3.78(s, 2H), 7.11 (s, 1H), 7.59-7.67 (m, 3H), 8.39 (br s, 4H), 12.42 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO): δ (ppm) 10.0, 25.8, 26.4, 28.5, 28.6, 28.7, 28.90, 28.91, 29.1, 30.0, 30.7, 32.2, 38.4, 42.5, 43.2, 110.0, 116.9 (q, J 298 Hz, TFA), 148.5, 154.2, 159.2 (q, J 32Hz, TFA), 160.6, 167.5, 167.9, 172.6, 182.3, 182.4. HRMS: (ESI): $m/z [M+H]^+$, calcd. for $C_{24}H_{40}N_7O_3S_2^+$: 538.2629, found: 538.2636. $C_{24}H_{39}N_7O_3S_2 \cdot C_4H_2F_6O_4$ (537.74 + 228.04).

tert-Butyl [8-(2-[3-(2-guanidinothiazol-4-yl)phenylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]carbamate (3.28)

3.7 (200 mg, 0.86 mmol, 1 eq) was dissolved in EtOH (7 mL) and poured into a solution of 3,4diethoxy-3-cyclobutene-1,2-dione in EtOH (7 mL). The reaction mixture was stirred over night at room temperature. The precipitated **3.27** was filtered off, washed with EtOH (2 mL) and dried *under vacuo*.

3.27 (160 mg, 0.45 mmol, 1 eq), **3.16** (120 mg, 0.49 mmol, 1.1 eq) and triethylamine (227 mg, 2.24 mmol, 5 eq) were suspended in EtOH (15 mL). The reaction mixture was stirred over night at 80 °C and subsequently cooled to room temperature. The precipitated product was filtered off and residual solvent was removed *in vacuo*. **3.28** was afforded as a beige solid (160 mg, 33.6 %

over two steps). Mp: 262 °C decomposition. $R_f = 0.5$ (CH₂Cl₂/7 M NH₃ in MeOH 6:1). ¹H-NMR (300 MHz, [D₆]-DMSO): δ (ppm) 1.25-1.57 (m, 19H), 1.55-1.57 (m, 2H), 2.85-2.91 (m, 2H), 3.60-3.62 (m, 2H), 6.76-6.79 (m, 1H), 6.9 (br s, 3.5H), 7.08-7.15 (m, 2H), 7.32 (t, 1H, *J* 7.84 Hz), 7.46-7.48 (m, 1H), 7.70 (br s, 1H), 8.30 (s, 1H), 9.67 (s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO): δ (ppm) 25.8, 26.2, 28.3, 28.6, 28.7, 29.5, 30.6, 30.7, 43.7, 77.3, 103.5, 115.2, 116.3, 119.2, 129.4, 136.1, 139.5, 148.9, 155.6, 157.1, 163.4, 169.3, 174.9, 180.1, 183.9. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₇H₃₈N₇O₄S⁺: 556.2700, found: 556.2706. C₂₇H₃₇N₇O₄S (555.70).

1-[4-(3-[2-(8-Aminooctylamino)-3,4-dioxocyclobut-1-ene-1-ylamino]phenyl)thiazol-2yl]guanidine- hydrotrifluoracetate (3.29)

3.28 (140 mg, 0.25 mmol, 1eq) was suspended in CH₂Cl₂ (20 mL). TFA (689 mg, 6.05 mmol, 24 eq) was added and the reaction mixture was stirred for 5 h at room temperature. Removal of the solvent *in vacuo* afforded the crude product as an off-white solid (140 mg, 99%). 50 mg were further purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05% aq. TFA 15:85-65:35, $t_{\rm R}$ = 12.5 min). **3.29** was obtained as a white fluffy solid (31 mg, 61.3%). Mp: 210.0-217.6 °C decomposition. $R_{\rm f}$ = 0.2 (CH₂Cl₂/7 M NH₃ in MeOH 6:1). RP-HPLC (gradient 2, 220 m): 97.6% ($t_{\rm R}$ = 14.76 min, k = 4.9). ¹H-NMR (400 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 1.23-1.32 (m, 8H), 1.48-1.52 (m, 2H), 1.55-1.60 (m, 2H), 2.74-2.77 (m, 2H), 3.59-3.63 (m, 2H), 7.20-7.21 (m, 1H), 7.37 (t, 1H, *J* 7.86 Hz), 7.54-7.55 (m, 1H), 7.68 (br s, 3H), 7.74 (s, 1H), 8.12 (br s, 1H), 8.46 (br s, 4H), 8.55 (s, 1H), 10.08 (s, 1H), 12.53 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 25.7, 27.0, 28.40, 28.41, 30.5, 30.7, 38.8, 43.7, 108.4, 115.5, 116.9 (q, *J* 298 Hz, TFA), 117.2, 119.4, 129.8, 134.2, 139.9, 149.4, 154.2, 158.7 (q, *J* 32 Hz, TFA),160.2, 163.2, 169.6, 180.2, 183.7. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₂H₃₀N₇O₂S⁺: 456.2176, found: 456.2175. C₂₂H₂₉N₇O₂S · C₂HF₃O₂ (455.58 + 114.02).

N-[8-(2-[3-(2-Guanidinothiazol-4-yl)phenylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]propionamide-hydrotrifluoracetate (3.30)

3.29 (110 mg, 0.16 mmol, 1 eq) was dissolved in a mixture of triethylamine (65 mg, 0.64 mmol, 4 eq), DMF (1 mL) and CH₂Cl₂ (1 mL). N-Succinimidyl propionate (21mg, 0.18 mmol, 1.1 eq) was added and the reaction mixture was stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05% aq. TFA 25:75-70:30, t_R = 13.1 min) afforded the product as white fluffy solid (12 mg, 11.6%). Mp: 224-236.4 °C. R_f = 0.67 (CH₂Cl₂/7 M NH₃ in MeOH 6:1). RP-HPLC (gradient 2, 220 nm): 98.5% (t_R = 18.85 min, k = 6.5). ¹H-NMR (400 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 0.96 (t, 3H, *J* 7.62 Hz), 1.22-1.58 (m, 10H), 1.55-1.58 (m, 2H), 2.02 (q, 2H, *J* 7.62 Hz), 2.97-3.01 (m, 2H), 3.60-3.61 (m, 2H), 7.13-7.15 (m, 1H), 7.37 (t, 1H, *J* 7.92 Hz), 7.54-7.55(m, 1H), 7.67-7.71(m, 2H), 7.91 (br s, 1H), 8.29 (br s, 4H), 8.56 (s, 1H), 9.87 (s, 1H), 12.26 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 10.0, 25.8, 26.3, 28.48, 28.54, 28.6, 29.1, 30.5, 38.3, 43.7, 108.3, 115.4, 117.1, 119.4, 129.8, 134.4, 139.8, 149.3, 154.2, 162.8, 163.1,169.5, 172.6, 180.2, 183.8. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂₅H₃₄N₇O₃S⁺: 512.2438, found: 512.2443. C₂₅H₃₃N₇O₃S · C₂HF₃O₂ (511.65 + 114.02).

S-Methylthiouronium iodide (3.31)⁴⁰

Methyliodide (9.30 g, 65.50 mmol, 1 eq) was added to a stirring solution of thiourea (5 g, 65.50 mmol, 1 eq) in MeOH (50 mL). The reaction mixture was stirred under reflux for 1.5 h. The solvent was removed *in vacuo* and the residue was washed two times with diethyl ether (50 mL). **3.31** was afforded as white solid (14.30 g, 99.9%). Mp: 117.2-118.1°C. $R_f = 0.1$ (PE/EtOAc 3:1). ¹H-NMR (300 MHz, [D₆]-DMSO): δ (ppm) 2.56 (s, 3H), 8.89 (br s, 4H). ¹³C-NMR (75 MHz, [D₆]-DMSO): δ (ppm) 13.2, 171.0. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂H₇N₂S⁺: 91.0324, found: 91.0327. C₂H₆N₂S · HI (90.14 + 127.91).

N-tert-Butoxycarbonyl-S-methylisothiourea (3.32)⁴⁰

3.31 (13.65 g, 62.45 mmol, 1 eq) and triethylamine (6.32 g, 62.45 mmol, 1 eq) were dissolved in CH₂Cl₂ (110 mL) and cooled to 0 °C. Di-*tert*-butyl dicarbonate (13.63 g, 62.45 mmol, 1 eq) dissolved in CH₂Cl₂ (30 mL) was added drop wise at 0 °C. The reaction mixture was allowed to warm to room temperature and was stirred over night. The organic layer was washed two times with water (200 mL). Product which passed in the aqueous layer was extracted two times with CH₂Cl₂ (100 mL). The combined organic layers were dried over sodium sulfate. Removal of the solvent *in vacuo* and purification by column chromatography (eluent: PE/EtOAc 6:1- 3:1) afforded the product as white solid (8.64 g, 72.6%). Mp: 88.4-90.3 °C. *R*_f = 0.45 (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 1.44 (s, 9H), 2.38 (s, 3H), 7.50 (br s, 2H). ¹³C-NMR (75 MHz, CDCl₃): δ (ppm) 13.4, 28.1, 79.9, 172.9. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₇H₁₅N₂O₂S⁺: 191.0849, found: 191.0847. C₇H₁₄N₂O₂S (190.26).

*N*¹,*N*⁶-Bis([(*tert*-butoxycarbonylamino)(methylsulfanyl)methylene]aminocarbonyl)hexane-1,6diamine (3.33)¹⁴

3.32 (4.03 g, 21.1 mmol, 2.2 eq) and triethylamine (0.97 g, 9.62 mmol, 1) were dissolved in anhydrous CH_2Cl_2 (17 mL). 1,6-Diisocyanohexane (1.62 g, 9.6 mmol, 1 eq) was added under Aratmosphere. The reaction mixture was stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by column chromatography (eluent: PE/EtOAc 3:1- 1:3) afforded the product as white solid (4.65 g, 88.0%). Mp: 128.8-133.6 °C. R_f = 0.14 (PE/EtOAc 3:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.36-1.40 (m, 4H), 1.48 (s, 18H), 1.51-1.57 (m, 4H), 2.30 (s, 6H), 3.19-3.24 (m, 4H), 5.53 (m, 2H), 12.30 (s, 2H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.4, 26.7, 28.1, 29.7, 40.1, 82.7, 151.2, 162.1, 167.4. HRMS: (ESI): m/z [M+H]⁺, calcd. for $C_{22}H_{41}N_6O_6S_2^+$: 549.2524, found: 549.2526. $C_{22}H_{40}N_6O_6S_2$ (548.72).

1-(Amino[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]methylene)-3-(6-[3-(amino[2-(2-guanidinothiazol-4-ylmethylthio)ethylamino]methylene)ureido]hexyl)urea tetra(hydrotrifluoracetate) (3.34)

The HCl salt of **3.3** (244 mg, 0.80 mmol, 2.2 eq), **3.33** (200mg, 0.36 mmol, 1 eq) and diisopropylethylamine (233 mg, 1.80 mmol, 5 eq) were suspended in MeOH (7 mL). The reaction mixture was stirred under reflux for 5 h. The solvent was removed under reduced pressure and the residue was purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05% aq. TFA 15:85-80:20, $t_R = 15.2$ min). Removal of the MeCN *in vacuo* and lyophilisation afforded the Boc-deprotected **3.34** as a white solid. The residue was again purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05% aq. TFA 10:90-65:35, $t_R = 12.7$ min). **3.34** was obtained as a white solid (45 mg, 10.7%). Mp: 59.1-61.9 °C. RP-HPLC (gradient 2, 220 nm): 96.2% ($t_R = 12.95$ min, k = 4.2). ¹H-NMR (600 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 1.25-1.26 (m, 4H), 1.41-1.43 (m, 4H), 2.66 (t, 4H, *J* 6.78 Hz), 3.06-3.09 (m, 4H), 3.44-3.47 (m, 4H, interfering with the water signal), 3.79 (s, 4H), 7.13 (s, 2H), 7.56 (br s, 2H), 8.49 (br s, 2H), 9.13 (br s, 2H), 10.34 (br s, 2H), 12.54 (br s, 2H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 25.9, 28.8, 29.4, 30.0, 40.2, 53.5, 110.1, 116.9 (q, *J* 299 Hz, TFA), 148.5, 153.7, 153.8, 154.3, 159.2 (q, *J* 32 Hz, TFA), 160.5. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₄H₄₃N₁₆O₂S₄⁺: 715.2632, found: 715.2627. C₂₄H₄₂N₁₆O₂S₄ · C₈H₄F₁₂O₈ (714.95 + 456.09).

1-(Amino[3-(2-guanidinothiazol-4-yl)phenylamino]methylene)-3-(6-[3-(amino[3-(2-guanidino-thiazol-4-yl)phenylamino]methylene)ureido]hexyl)urea tetra(hydrotrifluoracetate) (3.35)

3.7 (280 mg, 1.21 mmol,2.2 eq) and **3.33** (300 mg, 0.55 mmol, 1 eq) were suspended in MeOH (15 mL). The reaction mixture was stirred under reflux for 48h. The solvent was removed under reduced pressure and the residue was purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05% aq. TFA 15:85-80:20, $t_{\rm R}$ = 19.1 min). Removal of the MeCN *in vacuo* and lyophilisation afforded the Boc-deprotected **3.35** as a white solid. The residue was again purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.05% aq. TFA 15:85-45:55, $t_{\rm R}$ = 15.3 min). **3.35** was obtained as a white solid (22 mg, 3.4%). Mp: 201.1 °C. RP-HPLC (gradient 2, 220 nm): 98.5% ($t_{\rm R}$ = 14.54 min, k = 4.8). ¹H-NMR (600 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 1.26-1.29 (m, 4H), 1.42-1.45 (m, 4H), 3.10-3.13 (m, 4H), 7.30-7.32 (m, 2H), 7.53-7.55 (t, 2H, *J* 7.85 Hz), 7.65 (br s, 2H), 7.82 (s, 2H), 7.92 (s, 2H), 7.96-7.97 (m, 2H), 8.47-8.93 (m, 12H), 10.36 (br s, 2H), 10.81 (br s, 2H), 12.63 (br s, 2H). ¹³C-NMR (150 MHz, [D₆]-DMSO, COSY, HSQC, HMBC): δ (ppm) 25.9, 28.8, 1C under solvent peak (38.7-40.3), 109.1, 115.9, 116.9 (q, *J* 297 Hz, TFA), 123.4, 125.3, 125.7, 130.3, 134.3, 134.8, 148.9, 153.6, 154.2, 159.3 (q, *J* 31 Hz, TFA), 161.0. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₃₀H₃₉N₁₆O₂S₂⁺: 719.2878, found: 719.2880. C₃₀H₃₈N₁₆O₂S₂ · C₈H₄F₁₂O₈ (718.86 + 456.09).

N-[8-(2-[2-(2-Guanidinothiazol-4-ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]-[2,3-³H]propionamide-hydrotrifluoracetate ([³H]3.25)

The radioligand [³H]**3.25** was prepared by [³H]propionylation of the amine precursor **3.22** using reported protocols with minor modifications.^{11,19,24} A solution of N-succinimidyl [2,3-³H]propionate (specific activity: 88 Ci/mmol, 1.5 mCi, 3.0 µg, 17.05 nmol, from American Radiolabeled Chemicals Inc., St. Louis, MO via Biotrend, Köln, Germany) in hexane/EtOAc 9:1 was transferred into a 1.5 mL reaction vessel with a screw cap. The solvent was evaporated in a vacuum concentrator at approx. 30 °C in 30 min. A solution of **3.22** (0.35 mg, 0.38 µmol, 22 eq) in a mixture of anhydrous DMF (50 µL) and DIPEA (0.8 µL) was added and the mixture was shaken at room temperature for 80 min. The analysis of the reaction mixture was performed with an RP-HPLC system (*cf.* Figure 3.7) (Waters, Eschborn, Germany) consisting of two 510 pumps, a pump control module, a 486 UV/vis detector and a Flow-one Beta series A-500 radiodetector (Packard, Meriden; CT). A Luna C18 (3 µm, 150 mm x 4.6 mm) was used RP-column at a flow rate of 0.8 mL/min. Mixtures of 0.04% TFA in CH₃CN (A) and 0.05% aq. TFA (B) were used as mobile phase. The following linear gradient was applied: 0-20 min, A/B 14:86-32:68; 20-22min, 32:68-95:5; 22-32min, 95:5 isocratic), (UV detection: 220 nm).

For the isolation of $[{}^{3}$ H]**3.25** (peak at $t_{R} = 18.8 \text{ min}$, *cf.* Figure 3.7) by analytical RP-HPLC (conditions as for the analysis of the reaction mixture, no radiometric detection), a H₂O/TFA (96:4, v/v, 11.5 µL) was added to the reaction mixture. The eluates (three injections), containing $[{}^{3}$ H]**3.25**, were combined in a 2 mL reaction vessel with a screw cap and the volume was reduced in a vacuum concentrator to 200 µL. EtOH (800 µL) was added and the solution was transferred into a 3 mL borosilicate glass vial with conical bottom (Wheaton NextGen 3 mL V-vials). The 2 mL reaction vessel was rinsed twice with EtOH/H2O 80:20 (v/v, 2 x 200µL) and the wash solutions were transferred to the 3 mL glass vials yielding the tentative stock solution with a volume of 1200 µL.

As analysis of the tentative stock solution by radio-HPLC (4 μ L dissolved in 126 μ L of 0.05% aq TFA/MeCN 90:10; same conditions as for the isolation) revealed a second peak ($t_R = 14.3 \text{ min}$, *cf*. Figure 3.7B), amounting to ca. 13% of the total peak area, determination of the specific activity of [³H]**3.25** was not feasible. Therefore, the specific activity of [³H]**3.25** was estimated based on the specific activity (2.41 TBq/mmol, 65.03 Ci/mmol) of [³H]UR-MK300¹⁹ prepared on the same day from the same lot of N-succinimidyl [2,3-³H]propionate. The molarity of the tentative stock solution was calculated from the specific activity of [³H]UR-MK300 (2.41 TBq/mmol) and the activity concentration (determined with a LS 6500 liquid scintillation counter (Beckmann-Coulter, Munich, Germany)) of the tentative stock solution with respect to [³H]**3.25** (2.47 · 10⁻⁵ TBq/mL, 87% of the total activity concentration). The final total activity concentration was adjusted to 18.5 MBq/mL (0.5 mCi/mL) by the addition of EtOH/H₂O (80:20 v/v) containing 0.22 mM ascorbinic acid (625 μ L) resulting in the final stock solution with 6.69 μ mol/L [³H]**3.25** and 76.9 μ mol/L ascorbinic acid (radiochemical yield for [³H]**3.25**: 28.9 MBq, 64.3%).

Investigation of the chemical stability of 3.25

The chemical stability of the "cold form" of the radioligand (**3.25**) was investigated in a mixture of EtOH/H₂O (80:20, v/v) at room temperature. Initially, a solution of **3.25** (100 μ M) in EtOH/H₂O (80:20, v/v) was prepared. The incubation was started by dilution of this 100 μ M solution of **3.25** with a mixture of EtOH/H₂O (80/20, v/v) resulting in a 15 μ M solution of **3.25**. After 0 h, 6 h, 12 h, 24 h, 48 h, 72 h and 7 days a 8 μ L aliquot was added to 192 μ L of a mixture of CH₃CN/ 0.05% aq. TFA (10:90, v/v). A sample (80 μ L) was immediately analyzed by analytical HPLC (Gradient 2, detection wavelength of 273 nm).

3.3.3 Pharmacological Methods

Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of Sf9 insect cells, expressing the $hH_2R-G_{s\alpha S}$ fusion protein or coexpressing the $hH_1R + RGS4$, the $hH_3R + G_{i\alpha 2} + \beta_1\gamma_2$ or $hH_4R + G_{i\alpha 2} + \beta_1\gamma_2$ proteins was described elsewhere.⁴¹

Radioligand competition binding assays were performed as described previously with minor modifications, using the following radioligands: $[{}^{3}H]$ mepyramine (Hartmann Analytic, Braunschweig, Germany; hH₁R, specific activity = 20 Ci/mmol, K_{d} = 4.5 nM, c_{final} = 5 nM), $[{}^{3}H]$ UR-DE257¹¹ batch 04/2015 (hH₂R, specific activity = 32.89 Ci/mmol, K_{d} = 12.2 nM, c_{final} = 20 nM), $[{}^{3}H]$ tiotidine (Hartmann Analytic, Braunschweig, Germany; hH₂R, specific activity = 78.42 Ci/mmol, K_{d} = 12.75 nM, c_{final} = 10 nM), $[{}^{3}H]$ N^{α}-methylhistamine (Hartmann Analytic, Braunschweig, Germany; hH₃R, specific activity = 80 Ci/mmol, K_{d} = 3 nM, c_{final} = 3 nM), $[{}^{3}H]$ histamine (Hartmann Analytic, Braunschweig, Germany; hH₄R, specific activity = 25 Ci/mmol, K_{d} = 15.9 nM, c_{final} = 10 nM).and $[{}^{3}H]$ UR-PI294²⁰ (hH₄R, specific activity = 93.3 Ci/mmol, K_{d} = 5.1 nM, c_{final} = 5 nM).

On the day of the experiment, Sf9 membranes were thawed and sedimented by centrifugation at 13,000 rpm at 4 °C for 10 min. The membranes were resuspended in ice cold binding buffer (12.5 mM MgCl₂, 1mM EDTA and 75 mM Tris/HCl, pH 7.4; in the following referred to as BB) and adjusted to a protein concentration of 2-4 $\mu g/\mu L$. 80 μL BB containing 0.2% BSA and the respective radioligand, followed by 10 μ L of the investigated ligands at various concentrations (dissolved in H₂O), were added to every well of a 96-well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with 0.3% polyethylenimine, using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with 200 µL scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h. The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

Functional GTPyS assay on Sf9 insect cell membranes

GTP γ S assays were performed as described previously¹⁴ with minor modifications. [³⁵S]GTP γ S (specific activity = 1000 Ci/mmol) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to 0.5-1.5 µg/µL.

Agonist mode: 80 μ L of BB containing BSA (0.05% final), GDP (1 μ M final) and [³⁵S]GTP γ S (20 nCi final), followed by 10 μ L of the investigated ligands at various concentrations (dissolved in H₂O) were added to every well of a 96-well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTP γ S assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine (1 μ M final).

Radioligand binding assays with [³H]UR-SB69 on Sf9 insect cell membranes

The radioligand binding experiments were performed as described for competition binding assays on Sf9 insect cell membranes with minor adjustments.

Saturation binding: For the determination of total binding 10 μ L of H₂O, followed by 70 μ L BB containing 0.2% BSA, were added to every well of a 96-well plate (Primaria clear flat bottom microplates, Corning, New York, USA). For the determination of unspecific binding 10 μ L famotidine (300 fold excess compared to the radioligand concentration, 3-600 μ M) in H₂O was added. 10 μ L of [³H]**3.25** in concentrations of 1-200 nM in H₂O (for economic reasons the radioligand was mixed 1:4 with the cold form) were added to every well and the incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 90 min at room temperature in the dark.

Competition binding: $[^{3}H]$ **3.25** (specific activity = 65.03 Ci/mmol, K_{D} = 15 nM) was used in a final concentration of 20 nM. The incubation time was 90 min.

Association: 70 µL BB containing 0.2% BSA, followed by 10 µL of membrane suspension, were added to every well of a 96-well plate (Primaria clear flat bottom microplates, Corning, New York, USA). 10 µL [3 H]**3.25** (final concentration: 20 nM) in H₂O and either 10 µL H₂O (total binding) or 10 µL famotidine (300-fold excess, unspecific binding) in H₂O were added at different time points (0-180 min) at 25 °C. After the last addition the bound radioligand was separated from the free radioligand. The last time point (last addition of the radioligand) represented the shortest incubation time (0 min) and the first time point the longest incubation time (180 min).

Dissociation: 70 μ L BB containing 0.2% BSA, followed by 10 μ L of membrane suspension, were added to every well of a 96-well plate (Primaria clear flat bottom microplates, Corning, New York,

USA). 10 μ L [³H]**3.25** (final concentration: 20 nM) in H₂O and either 10 μ L H₂O (total binding) or 10 μ L famotidine (300-fold excess, unspecific binding) in H₂O were added at different time points (0-180 min). Every time point was preincubated for 60 min at 25 °C. After the incubation time was over 100 μ L famotidine (final concentration: 3 μ M) in BB was added. After the last addition the bound radioligand was separated from the free radioligand. The last time point (last addition of the radioligand) represented the shortest dissociation time (0 min) and the first time point the longest dissociation time (180 min).

Cell culture

The preparation of stably transfected HEK cells (HEK293T-hH₂R-qs5⁴² and HEK293T-hH₂R- β Arr2^{43,44}) was described elsewhere.

Cells were cultivated at 37 °C in a water saturated atmosphere containing 5% CO₂. Dulbecco's Modified Eagle Medium, containing 4.5 g/L glucose, 3.7 g/L NaHCO₃, 110 mg/L sodium pyruvate (DMEM, Sigma-Aldrich Munich, Germany) and supplemented with 0.584 g/L L-glutamine (L-glutamine solution, Sigma-Aldrich Munich, Germany), 1% (v/v) Penicillin-Streptomycin (P/S, 10,000 U/mL, Sigma-Aldrich Munich, Germany), 10% (v/v) fetal calf serum (FCS, Biochrom GmbH, Merck, Berlin, Germany) were used as a culture medium. Additionally, 100 µg/mL hygromycin B (A.G. Scientific, Inc., San Diego, CA) and 400 µg/mL G418 (Biochrom GmbH, Merck, Berlin, Germany) were added to the culture medium of HEK293T-hH₂R-qs5 cells, and 400 µg/mL zeocin (InvivoGen, San Diego, USA) and 600 µg/mL G418 were added to the culture medium of HEK293T-hH₂R- β Arr2 cells.

Radioligand saturation binding assays with $[^{3}H]$ UR-SB69 on intact HEK293T-hH₂R-qs5 and HEK293T-hH₂R- β Arr2 cells

HEK293T cells were seeded in a 175-cm² culture flask 5-7 days prior to the experiment. On the day of the experiment, cells were trypsinized and detached with fresh culture medium (5 mL). After centrifugation (250 *g*, 10 min) the cell pellet was resuspended in Leibovitz's L-15 culture medium (L-15 medium, Gibco/Life Technologies, Carlsbad, USA) and the concentration was adjusted to 0.25-0.5 \cdot 10⁶ cells/mL. 80 µL cell suspension was added to every well of a 96-well plate (Primaria clear flat bottom microplates, Corning, New York, USA). 10 µL of [³H]**3.25** in concentrations of 1-200 nM in H₂O (for economic reasons the radioligand was mixed 1:4 with the cold form) and either 10 µL H₂O (total binding) or 10 µL famotidine (300 fold excess, unspecific binding) in H₂O were added to every well. The plates were shaken for 90 min at room temperature in the dark. The separation of bound from free radioligand was performed as described for membranes.

3.3.4 Data analysis

Retention factors k were calculated according to $k = (t_R - t_0)/t_0$ (t_0 = dead time). Corrected counts per minute (ccpm) from the GTPyS assay (agonist mode) were plotted against the log(concentration of the test compound), and data were analyzed by a four parameter logistic equation (GraphPad Prism Software 5.0, GraphPad Software, San Diego, CA), followed by normalization (0% = water value (basal activity), 100% = "top" histamine equation) and analysis by four-parameter logistic equation (log(agonist) vs. response – variable slope, GraphPad Prism). Data of the GTPyS assay (antagonist mode) were analysed by a four parameter logistic equation (GraphPad Prism), followed by normalization (100% = "top" of the four-parameter logistic fit, 0% = unspecifically bound radioligand (ccpm) determined in the presence of famotidine at 100 μ M) and analysis by four-parameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism). p/C_{50} values were converted into pK_B values according to the Cheng-Prusoff equation⁴⁵. Specific binding data from saturation binding experiments were plotted against the total radio labeled ligand concentration (approximately equivalent to the "free" ligand concentration) and analyzed by a two-parameter equation describing hyperbolic binding (one site - specific binding, GraphPad Prism) and unspecific binding data was analyzed by linear regression. Specific binding data from association binding experiments were analyzed by a two parameter equation describing an exponential rise to a maximum (one-phase association, GraphPad Prism) to obtain the observed association constant k_{obs} . Specific binding data from dissociation binding experiments were analyzed by a three parameter equation (one phase decay, GraphPad Prism) to obtain the dissociation rate constant k_{off} . Kinetic dissociation constants $K_{d (kin)}$ were calculated from k_{on} and $k_{off} (k_{on} = (k_{obs} - k_{off})/[L]; K_{d (kin)} = k_{off}/k_{on})$. Specific binding data from association and dissociation binding experiments were normalized (100% = Y_{max} (association) or Y_0 (dissociation)). Total binding data from radioligand competition binding experiments were plotted against log(concentration competitor) and analyzed by a fourparameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism), followed by normalization (100% = "top" of the four-parameter logistic fit, 0% = unspecifically bound radioligand ligand determined in the presence of famotidine at 100 μ M). Normalized data from competition binding experiments were again analyzed by a four-parameter logistic equation and obtained p/C_{50} values were converted into pK_i values according to the Cheng-Prusoff equation⁴⁵.

3.4 SUMMARY AND CONCLUSION

The combination of the 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole structure derived from famotidine or the guanidino-4-(3-aminophenyl)thiazole structure derived from ICI127032 with the derivatized squaramide moiety of BMY25368 led to propionylated H_2R high affinity antagonists (**3.24-3.26** and **3.30**, pK_i values: 6.8-7.65). The linking of two guanidinothiazole pharmacophores by a carbamoyl guanidine linker resulted in the high affinity bivalent antagonists 3.34 and 3.35 (pK_i values: 7.3 and 7.14). The ligand 3.25 showed a high affinity to the hH₂R (pK_i value: 7.65) and selectivity over the other subtypes (no affinity to the hH₁R, hH₃R: pK_i value of 5.3 and hH₄R: pK_i value of 4.4). The radiolabeled form [³H]**3.25** (radiochemical purity of 87%) bound in a saturable manner to membrane preparations of Sf9 cells and intact HEK293T cells, both expressing recombinant hH₂Rs and the specific binding was best fitted by nonlinear regression to a one-site binding model. The determined K_d values (15-22 nM) were similar to the K_i value of 23 nM determined with [³H]UR-DE257 on membrane preparations of Sf9 cells. Although a part of [³H]**3.25** bound in (pseudo)irreversible manner (plateau at 23%), the kinetic K_d value of 26 nM was comparable to the equilibrium one and the radioligand was completely displacable by histamine, famotidine and ICI127032. Lamtidine, by contrast, could only displace 75% of [³H]**3.25**.

However, the results of the biological evaluation of $[{}^{3}H]$ **3.25** should be looked at critically, as the radiochemical purity of the radioligand was only moderate to begin with (87%) and stability studies showed that the radiochemical purity was further decreasing over a period of 15 month to 45%. Furthermore, only one major impurity, with unknown H₂R affinity and potency, was formed. This was all very surprising, as the squaramide radioligand $[{}^{3}H]$ UR-DE257 was stable in EtOH for at least 24 month.¹¹ As such one can well assume that the 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole structure is the part vulnerable for decomposition. Most propably the impurity was formed by oxidation of the sulfide linker to either the sulfoxide or sulfone. For unequivocal identification of the impurity, a solution of the radioligand containing the impurity should be spiked with either the sulfoxide or the sulfone compound and be analyzed by RP-HPLC. For this purpose both compounds have to be specifically synthesized. Depending on the structure of the impurity, this results may also question the stability of previously described H₂R radioligands like $[{}^{3}H]$ ranitidine, $[{}^{3}H]$ cimetidine and $[{}^{3}H]$ tiotidine which contain the same sulfide linker.

Nevertheless, [³H]**3.25** could be a valuable molecular tool provided that purity and stability under storage conditions are improvable. Alternatively, it should also be considered to further optimize the derivatives of ICI127032 (e.g. **3.30**) in order to obtain more stable ligands suitable for radiolabelling with high affinity.

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Chapter 4

Aminopotentidine Derivatives as Highly Potent H₂R Antagonists: Synthesis and Pharmacological Characterization of Amine Precursors and "Cold" Forms of Potential Radioligands

Note: Prior to the submission of this thesis, parts of this chapter (the synthesis of **4.1**, **4.2** and **4.3**) were published in cooperation with partners:

Baumeister, P.; Erdmann, D.; <u>Biselli, S.</u>; Kagermeier, N.; Elz, S.; Bernhardt, G.; Buschauer, A. [³H]UR-DE257: Development of a Tritium-Labeled Squaramide-Type Selective Histamine H₂ Receptor Antagonist. *ChemMedChem* **2015**, 10, 83-93.

4.1 INTRODUCTION

The piperidinomethylphenoxypropylamine (potentidine) moiety is a privileged structure for H_2R antagonism and part of many high affinity ligands. Some of the most important representatives are aminopotentidine and its derivatives (Figure 4.1).^{1,2} It was reported that aminopotentidine shows a high antagonistic activity (pK_b value: 7.28) and a second or even a third substituent at the aromatic ring of the residue is well tolerated (e.g. iodoaminopotentidine: pK_b value: 7.52 and diiodoaminopotentidine: pK_b value: 7.79).¹ Also compounds with only one substituent in 3-position (e.g. Compound I) or even compounds with an aromatic heterocycle as aromatic ring (e.g. Compound II) showed a similar or slightly lower activity.¹ Interestingly, the ¹²⁵Iodine labeled iodoaminopotentidine was a very high affinity ligand at the hH₂R (K_d value of 0.32 nM) and the pK_i values of aminopotentidine, iodoaminopotentidine and iodoazidopotentidine obtained with the afore mentioned radioligand were considerably higher than the corresponding pK_b values.^{1,3}

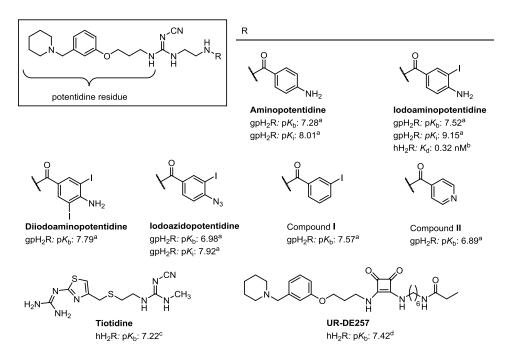


Figure 4.1. Structures of aminopotentidine, selected aminopotentidine derivatives, tiotidine and UR-DE257. ^aHirschfeld et al.^{1 b}Traiffort et al.^{4 c}Kelley et al.^{5 d}Baumeister et al.⁶

Although ¹²⁵iodine labeled ligands have, compared to tritium labeled ligands, the advantage of a higher specific activity, their preparation and usage require higher safety precautions and the ligands can only be used for 4-5 weeks after preparation.⁷ In our workgroup tritiated compounds are highly preferred due to their longer half-life. The published tritiated radioligands showed a 100-fold lower H_2R affinity (e.g. [³H]tiotidine and [³H]UR-DE257) compared to [¹²⁵I]iodoaminopotentidine.^{5,6}

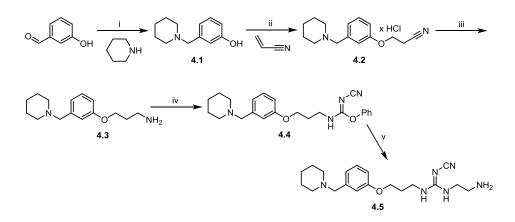
Here, the synthesis and pharmacological characterization of aminopotentidine and its analogs with different substituents (e.g. iodine, bromine, chlorine, trifluoromethyl) in the 3-position as precursors for potential tritiated ligands and synthesis of the "cold" form of potential radioligands are reported. The derivatisation of the terminal anilinic amino group of these

precursors was performed with different acylating reagents e.g. N-hydroxy succinimidyl propionate or propionic acid chloride, which are also available in tritiated form. Furthermore, a series of aminopotentidine derivatives containing a functionalized (propionylated, acetylated or methylated) aminomethyl substituent in 4-position of the aromatic ring was synthesized and characterized. In order to enable radiolabeling, the substituents of the amino-group were chosen with regard to commercially available labeling reagents such as tritiated N-hydroxy succinimidyl propionate, propionic acid chloride, acetyl chloride or methyl iodide.

4.2 RESULTS AND DISCUSSION

4.2.1 Chemistry

The intermediate 1-(2-aminoethyl)-2-cyano-3-(3-[3-(1-piperidinylmethyl)phenoxy]propyl)guanidine **4.5** was prepared in a five step synthesis as described before^{1,2,8} with minor modifications (Scheme 4.1). Starting from 3-hydroxybenzaldehyde a reductive amination with piperidine in the presence of formic acid (Leuckart-Wallach reaction) led to **4.1**. Addition of acrylonitrile to intermediate **4.1** led to **4.2** (variation of the Michael-addition reaction). The resulting cyanine **4.2** was reduced to the amine **4.3** by LiAlH₄. Coupling of diphenyl-*N*cyanocarbonimidate with **4.3** and subsequently with ethylene diamine resulted in the cyanoguanidine **4.5**.



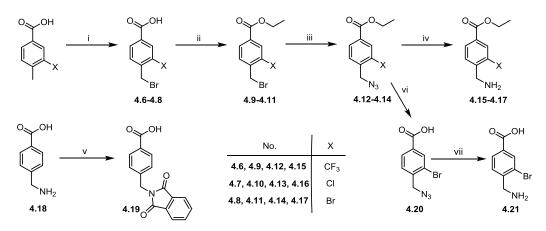
Scheme 4.1. Synthesis of 1-(2-Aminoethyl)-2-cyano-3-(3-[3-(1-piperidinylmethyl)phenoxy]propyl)guanidine **4.5**. Reagents and conditions: i) formic acid, 110° C, 3 h, 87%; ii) benzyltrimethylammonium hydroxide, reflux, 24 h, 55%; iii) LiAlH₄, anhydrous diethyl ether, RT, 4 h, 99.6%; iv) diphenyl-*N*-cyanocarbonimidate, 2-propanol, RT, 3 h, no purification; v) ethylene diamine, CH₃CN, 150° C, 15 min under microwave radiation, 60% over two steps.

The initial step of the synthesis of the aminopotentidine derivatives was the amide coupling of **4.5** and the respective benzoic acid derivatives, resulting in either the amine-precursors (**4.30**-**4.34**), amine-protected precursors (**4.39**, **4.45**) or directly in the final products (**4.42-4.44**, **4.49** and **4.50**).

The benzoic acid derivatives **4.15-4.17** were prepared in a four step synthesis (Scheme 4.2). Starting from 4-methyl benzoic acid derivatives, substituted with trifluoromethyl, chlorine, or bromine in 3-position, a side chain bromination using N-bromosuccinimide and AIBN were performed. The 4-bromomethyl benzoic acid derivatives **4.6-4.8** were transferred to the corresponding ethyl esters **4.9-4.11** using thionyl chloride and ethanol. A substitution reaction with sodium azide converted the ethyl esters **4.9-4.11** into the corresponding azides **4.12-4.14**. The 4-aminomethyl benzoic acid ethylesters **4.15-4.17** were synthesized from **4.12-4.14** by a Staudinger Reaction using triphenylphosphine.

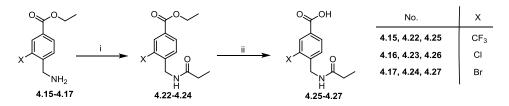
The phthalimide protected (4-Aminomethyl)benzoic acid (4.19) was synthesized from 4.18 and phthalic anhydride in the presence of acetic acid according to published protocols (Scheme 4.2).^{9,10} 4-(Aminomethyl)-3-bromobenzoic acid (4.21) was synthesized from 4.14 by first

saponification of the ester and second Staudinger reaction of the azide with triphenylphosphine to the amine (Scheme 4.2). The phthalimide protection of **4.21** under the same conditions as of **4.19** failed. This could be due to steric hindrance of the substituent in 3-position.



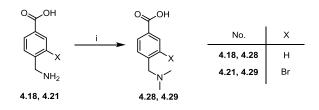
Scheme 4.2. Synthesis of benzoic acid derivatives **4.15-4.17** and **4.19-4.21**. Reagents and conditions: i) N-Bromosuccinimide, AIBN, CHCl₃, 75° C, 3-5 h, no purification; ii) thionyl chloride, EtOH, 65° C, 2-24 h, 43-52% over two steps; iii) NaN₃, anhydrous DMF, RT, 24 h, 49-91%; iv) PPh₃, THF/H₂O (1:2, v/v), RT, 24 h, 66-80%; v) phthalic anhydride, ACOH, reflux, 3.5 h, 84%; vi) NaOH, THF, RT, 24 h, 93%; vii) PPh₃, THF and H₂O, RT, 24 h, 27%.

The 4-(Propionamidomethyl)benzoic acids **4.25-4.27** were synthesized from **4.15-4.17** by amide coupling with succinimidyl propionate followed by saponification of the ethyl ester (Scheme 4.3).



Scheme 4.3. Synthesis of benzoic acid derivatives **4.25-4.27**. Reagents and conditions: i) TEA, succinimidyl propionate, CH₂Cl₂, RT, ON, no purification; ii) Aq. NaOH, THF, RT, ON, 49-84% over two steps.

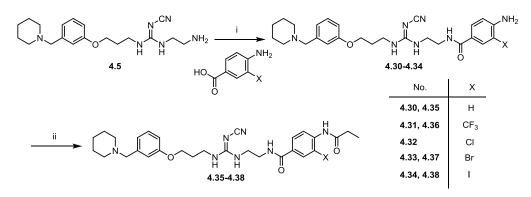
4-[(Dimethylamino)methyl]benzoic acid (**4.28**) and 3-bromo-4-[(dimethylamino)methyl]benzoic acid (**4.29**) were synthesized by Eschweiler-Clarke-Methylation of the corresponding amino acid **4.18** or **4.21** with formaldehyde in the presence of formic acid (Scheme 4.4).¹¹



Scheme 4.4. Synthesis of benzoic acid derivatives 4.28 and 4.29. Reagents and conditions: i) Aqueous formaldehyde, formic acid, reflux, ON, 94-100%.

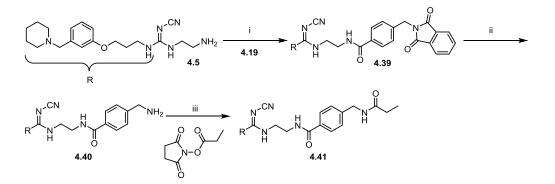
In scheme 4.5 the synthesis of the propionylated aminopotentidine derivatives **4.35-4.38** is depicted. Aminopotentidine¹ (**4.30**), iodoaminopotentidine^{1,3} (**4.34**) and its derivatives **4.31-4.33** were synthesized from **4.5** by amide coupling with the corresponding 4-amino benzoic acid using TBTU as coupling reagent and DIPEA as base. The anilinic amino group of **4.30-4.34** showed a reduced nucleophilicity which made the labeling by amide coupling a challenge. The classical

reaction with succinimidyl propionate in the presence of triethylamine (see synthesis of **4.22**-**4.24**, Scheme 4.3) failed. The propionylation by an excess of propionyl chloride in the presence of DMAP and TEA resulted in the desired products **4.35-4.38** in very low yields 11-23%.



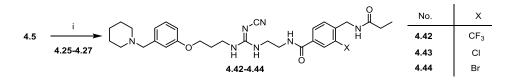
Scheme 4.5. Synthesis of the aminopotentidine derivatives **4.35-4.38**. Reagents and conditions: i) TBTU, DIPEA, CH₂Cl₂, RT, ON, 34-68%; ii) propionyl chloride, DMAP, TEA, CH₂Cl₂, RT, ON, 11-23%.

The aminopotentidine derivative **4.41** was synthesized in three steps starting by amide coupling of **4.5** and **4.19** using same conditions by analogy with **4.30-4.34** (Scheme 4.6). The resulting **4.39** was first phthalimide deprotected by hydrazine and then propionylated using succinimidyl propionate.



Scheme 4.6. Synthesis of the aminopotentidine derivative **4.41**. Reagents and conditions: i) TBTU, DIPEA, CH₂Cl₂, RT, ON, 74%; ii) Hydrazinium hydroxide, EtOH, RT, 4 h, 68%; iii) TEA, CH₂Cl₂, RT, ON, 81%.

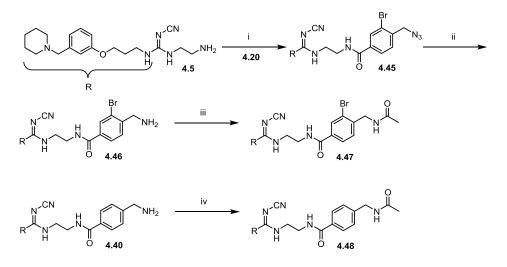
In case of the aminopotentidine derivatives **4.42-4.44** the final step was the amide coupling of the amine precursor **4.5** with the respective benzoic acids **4.25-4.27** using the same conditions as for the preparation of **4.30-4.34** (Scheme 4.7).



Scheme 4.7. Synthesis of the aminopotentidine derivatives 4.42-4.44. Reagents and conditions: i) TBTU, DIPEA, CH₂Cl₂, RT, ON, 6-44%.

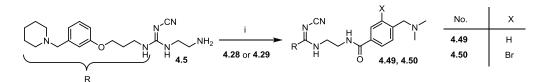
The synthesis of the acetylated aminopotentidine derivatives **4.47** and **4.48** is depicted in Scheme 4.8. The amide coupling of **4.5** and **4.20**, followed by a Staudinger Reaction and

acetylation of the resulting amine **4.46** with acetyl chloride resulted in the product **4.48**. The acetylation of **4.40** under similar conditions led to **4.48**.



Scheme 4.8. Synthesis of the aminopotentidine derivatives **4.47** and **4.48**. Reagents and conditions: i) HBTU, DIPEA, CH₂Cl₂, RT, ON, 44%; ii) PPh₃, THF/H₂O (1:2, v/v), RT, ON, 19%; iii) acetyl chloride, DIPEA, CH₂Cl₂, RT, ON, 20%; iv) acetyl chloride, DMAP, DIPEA, CH₂Cl₂, RT, 2 days, 25%.

The dimethylated aminopotentidine derivatives **4.49** and **4.50** were synthesized by coupling of **4.5** with the respective benzoic acid using HBTU as coupling reagent and DIPEA as base (Scheme 4.9).



Scheme 4.9. Synthesis of the aminopotentidine derivatives 4.49 and 4.50. Reagents and conditions: i) HBTU, DIPEA, CH₂Cl₂, RT, ON, 28-32%.

Adaptation of the synthetic route to radiosynthetic requirements

Radiosynthesis makes special demands on synthesis planning and reaction conditions. The labeling reaction is the key-step of the synthesis and should be the last step. In general, for radiosynthesis an excess of precursor compared to radioactive labeling reagent is used. This facilitates handling with respect of safety precautions and reduces the costs of the overall synthesis. Within the series of the "cold" forms of potential radioligands the propionylated ligands **4.37** and **4.38** showed the highest hH₂R affinity (see biological evaluation) and were therefore considered for radiolabeling.

However, the reaction conditions for the synthesis of **4.37** and **4.38** were not suitable for radiosynthesis due to the necessary high excess of the "cold" labeling reagent propionic chloride and the low yields. In order to adjust the reaction conditions to radiosynthesis standards, a test reaction was carried out in small scale and with an excess of **4.33** (6.98 µg, 10 eq). HRMS analysis of the reaction mixture showed no **4.37** was formed. This preliminary test showed that the

radiosynthesis of **4.37** (and **4.38**) will be a challenge. An alternative could be the synthesis of radiolabeled **4.50**. Comound **4.50** showed high hH_2R affinity in the range of UR-DE257 and tiotidine (see Biological evaluation). Dimethylation of **4.46** with methyliodide should be easily adaptable to radiosynthetic requirements. The main challenge will be to minimize formation of monomethylated and/or trimethylated by-product and the separation of **4.50** from the amine precursor and these by-products.

4.2.2 Biological Evaluation

H₂R affinity, selectivity compared to H₃R and antagonism

The aminopotentidine derivatives **4.30-4.38**, **4.40-4.44** and **4.46-4.50** were investigated in equilibrium competition binding experiments on membrane preparations from Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein using the antagonist [³H]UR-DE257⁶ as radioligand. The selectivity of several compounds for the hH_2R over the hH_3R was investigated by competition binding experiments using membranes of Sf9 insect cells co-expressing the hH_3R and $G_{\alpha i2}$ and $G_{\beta 1\gamma 2}$ proteins using [³H]N^{α}-methylhistamine as radioligand. Additionally, representative compounds were examined for hH_2R agonism in the GTPγS binding assay on membrane preparations from Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein. Ligands which exhibited no agonism were also investigated in antagonist mode versus histamine as agonist. Selected curves are depicted in Figure 4.1 and the results are summarized in Table 4.1.

Within the aminopotentidine derivatives iodoaminopotentidine (**4.34**) showed the highest hH_2R affinity. The p K_i value of 9.51 was in good agreement with the literature (gpH₂R: p K_i value of 9.15¹ and hH₂R: K_d value of 0.32 nM⁴). Also the 3-brominated aminopotentidine **4.33** showed a high hH₂R affinity with a p K_i value of 8.58. While the introduction of a bromine or iodine substituent in 3-position of the benzoic acid moiety (**4.33** or **4.34**) led to an increase in affinity compared to aminopotentidine (**4.30**, p K_i value of 7.7), a trifluoromethyl or chlorine substituent (**4.31** or **4.32**) retained the hH₂R affinity (p K_i value of 7.57-7.7). Likewise, the exchange of the anilinic amino group in 4-position with an aminomethyl group (**4.40**) was tolerated with almost no change in affinity (p K_i value of 7.6) compared to **4.30**. The 4-(propionamidomethyl)-3-bromobenzamide containing ligand **4.46** showed with a p K_i value of 7.71 a decreased affinity compared to the 3-brominated aminopotentidine **4.33**.

Within the target compounds ("cold" forms of potential radioligands) the propionylated iodoaminopotentidine **4.38** and the propionylated 3-bromo aminopotentidine **4.37** showed the highest hH₂R affinities with pK_i values of 8.18 and 8.5. The propionylation of the 3-halogenated aminopotentidines **4.35**, **4.36** and **4.38** resulted in a decrease of hH₂R affinity by one order of magnitude. By contrast, in case of **4.37**, the propionylated analog of **4.33**, the hH₂R affinity was retained (pK_i value of 8.5). The 4-(propionamidomethyl)benzamide containing ligands **4.41-4.44** showed only moderate affinity to hH₂R with pK_i values of 6.58-7.2. The acetylation (**4.48**: pK_i value of 6.4 and **4.47**: pK_i value of 7.09) as well as the propionylation (**4.41**: pK_i value of 6.58 and **4.46** resulted in a decrease of hH₂R affinity. The dimethylated analog of **4.40** (**4.49**) showed a low hH₂R affinity with a pK_i value of 5.61. In contrast, **4.50**, the dimethylated **4.46**, showed a high hH₂R affinity with a pK_i value of 7.54.

The ligands **4.30**, **4.31**, **4.33-4.38**, **4.40**, **4.41**, **4.43** and **4.44** showed a low hH_3R affinity with pK_i values of 4.5-5.22. Iodoaminopotentidine (**4.34**) showed with a 32000-fold higher affinity for the hH_2R compared to hH_3R the highest selectivity. Also the propionylated ligands **4.37** and **4.38** showed a lower but still excellent selectivity (6900- or 2500-fold).

All investigated ligands (4.30, 4.31, 4.33-4.35, 4.37, 4.38, 4.40, 4.41 and 4.44) were antagonists or inverse agonists in the GTP γ S binding assay. Except for the p K_b values of 4.40 and 4.44 (p K_b value:

7.5 and 6.5), the calculated pK_b values, obtained in the antagonistic mode, were considerably lower compared to the pK_i values. Especially bromoaminopotentidine **(4.33)** and iodoaminopotentidine **(4.34)**, which were high affinity hH₂R antagonists in binding experiments (pK_i values of 8.58 and 9.51), showed only low activities in the GTP_YS binding assay (pK_b values of 6.5 and 6.7). With a pK_b value of 7.6, the propionylated 3-bromo aminopotentidine **4.37** showed the highest activity among the investigated ligands.

Antagonism of aminopotentidine (**4.30**) and iodoaminopotentidine (**4.34**) (pK_b values of 6.0 and 6.7) measured with a GTPyS binding assay on membrane preparations from Sf9 insect cells expressing the hH₂R-G_{sαS} fusion protein were lower than described in literature (pK_b values of 6.63 and 7.46, obtained by steady-state GTPase assay).^{5,12} Discrepancies between pK_i values of aminopotentidine (**4.30**) and iodoaminopotentidine (**4.34**) from radioligand binding and pK_b values from functional experiments were also described for the gpH₂R by Hirschfeld et al (gpH₂R: pK_i values of 8.01 and 9.15 vs. pK_b values of 7.28 and 7.52)¹. Hirschfeld et al gave as possible explanation the different experimental setups which led to varying access to the H₂R (e.g. guinea pig striatal membranes vs. intact isolated guinea pig right atrium). When a very similar experimental setup was used (in our study membrane preparations in both binding and functional studies), the use of different competitors (histamine vs. radiolabeled antagonist) could lead to the different results¹³. Agonists and antagonists may stabilize different receptor conformations that exhibit different affinities for the investigated agonists/antagonists/inverse agonists.¹³ For the antagonistic radioligand [³H]tiotidine it was already shown that it binds only to a fraction of the functionally active H₂Rs.⁵

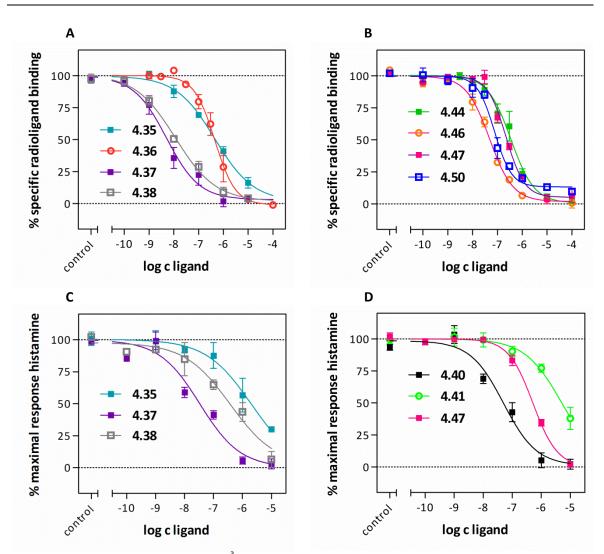


Figure 4.1. Displacement of the radioligand [³H]UR-DE257 (c = 20 nM, K_d = 12.2 nM) by (A) compounds **3.35-3.38** and (B) compounds **4.44**, **4.46**, **4.47**, **4.50** and antagonism of (C) compounds **4.35**, **4.37**, **4.38** and (D) compounds **4.40**, **4.41,4.47** on hH₂R determined in a GTPγS assay (antagonistic mode). Histamine (1 µM) was used for stimulation. Both determined on membrane preparations of Sf9 insect cells expressing the hH₂R-G_{sαS} fusion protein. Data represent mean values ± SEM of 2-4 experiments performed in triplicate.

Table 4.1. Affinities of the aminopotentidine derivatives **4.30-4.38**, **4.40-4.44** and **4.46-4.50** to $H_{2,3}R$, obtained from equilibrium competition binding studies and H_2R antagonism expressed in the calculated pK_b values determined by a GTP γ S assay.

	hH₂R				hH₃R		
	Binding ^a		GTPγS ^c		Binding ^d		Selectivity ^e
Compound	р <i>К</i> і	N	р <i>ЕС</i> ₅₀ (рК _В)	Ν	р <i>К</i> і	Ν	hH₂R:hH₃R
His	6.53 ± 0.04	3	5.80 ± 0.06	9	7.8 ± 0.1	3	19:1
4.30 (APT)	$7.7 \pm 0.1/$ gpH ₂ R: 8.01 ¹	3	$(6.0 \pm 0.4)/$ $(6.63)^{5}$	3	4.92 ± 0.08	3	1:600
4.31	7.7 ± 0.2 ^b	3	(6.6 ± 0.4)	4	5.03 ± 0.07	3	1:470
4.32	7.57 ± 0.01	3	n.d.	-	n.d.	-	n.d.
4.33	8.58 ± 0.05	2	(6.5 ± 0.4)	4	4.8 ± 0.1	3	1:6000
4.34 (IAPT)	9.51 ± 0.06/ gpH ₂ R: 9.15 ¹	3	$(6.7 \pm 0.1)/$ $(7.46)^{12}$	3	5.0 ± 0.1	4	1:32000
4.35	6.60 ± 0.08	3	(5.9 ± 0.2)	2	4.50 ± 0.09	4	1:130
4.36	6.7 ± 0.1	3	n.d.	-	4.7 ± 0.1	3	1:100
4.37	8.5 ± 0.3	3	(7.6 ± 0.1)	3	4.66 ± 0.07	4	1:6900
4.38	8.18 ± 0.07	3	(6.6 ± 0.4)	3	4.82 ± 0.09	4	1:2500
4.40	7.6 ± 0.2^{b}	2	(7.5 ± 0.2)	4	5.22 ± 0.09	2	1:240
4.41	6.58 ± 0.04	3	(5.4 ± 0.3)	3	4.75 ± 0.06	3	1:68
4.42	6.98 ± 0.08	3	n.d.	-	n.d.	-	n.d.
4.43	7.2 ± 0.1	3	n.d.	-	4.96 ± 0.03	2	1:170
4.44	6.9 ± 0.1	3	(6.5 ± 0.1)	4	5.04 ± 0.08	2	1:72
4.46	7.71 ± 0.07	3	n.d.	-	n.d.	-	n.d.
4.47	7.09 ± 0.09	3	n.d.	-	n.d.	-	n.d.
4.48	6.4 ± 0.1	4	n.d.	-	n.d.	-	n.d.
4.49	5.61 ± 0.07	3	n.d.	-	n.d.	-	n.d.
4.50	7.54 ± 0.09	3	n.d.	-	n.d.	-	n.d.

Competition binding assay on membrane preparations of Sf9 insect cells: ^aexpression of the H_2R-G_{sas} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, K_d = 12.2 nM or ^b[³H]tiotidine, c = 10 nM, K_d = 12.75 nM), ^dco-expression of the hH₃R and $G_{\alpha i2}$ and $G_{\beta 1\gamma 2}$ proteins (radioligand: [³H]N^{α}-methylhistamine, c = 3 nM, K_d = 3 nM). ^c[³⁵S]GTP γ S assay determined on membrane preparations of Sf9 insect cells expressing the hH₂R-G_{sas} fusion protein. The intrinsic activity (α) of histamine was set to 1.00, and α values of investigated compounds were referred to this value. The p K_B values were determined in the antagonist mode versus histamine (c = 1 µM) as agonist. ^eSelectivity represents the ratio of the corresponding K_i values. Data represent mean values ± SEM of N experiments performed in triplicate. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means ± SEM of N independent experiments, each performed in triplicate.

4.3 EXPERIMENTAL SECTION

4.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Fluka (Buchs, Swiss), Alfa Aesar (Karlsruhe, Germany), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 (0.040-0.063 mm). Microwave assisted synthesis was performed with an Initiator 2.0 (Biotage, Uppsala, Sweden) using microwave reaction vials (Biotage, Uppsala, Sweden) combined with caps and septa. Automated flash chromatography was performed with a 971-FP flash-purification system (Agilent Technologies, Santa Clara, CA). Pre-packed columns (SuperFlash SF10-4 g, SF12-8 g, SF 15-12 g und SF15-24 g, Agilent Technologies, Santa Clara, CA) were used throughout. Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution (0.8 g ninhydrin, 200 mL n-buthanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA). Nuclear Magnetic Resonance (¹H NMR, ¹³C NMR and ¹⁹F NMR) spectra were recorded on a Bruker Avance-300 (7.05 T, ¹H: 300 MHz, ¹³C: 75.5 MHz, ¹⁹F: 282 MHz), Avance-400 (9.40 T, ¹H: 400 MHz, ¹³C: 100.6 MHz), or Avance-600 (14.1 T; ¹H: 600 MHz, ¹³C: 150.9 MHz; cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in δ (ppm) relative to external standards. Multiplicities are specified with the following abbreviations: s (singlet), d (doublet), t (triplet), q (quartet), qui (quintet), m (multiplet), br s (broad signal), as well as combinations thereof. In certain cases 2D-NMR techniques (COSY, HSQC, HMBC and NOESY) were used to assign ¹H and ¹³C chemical shifts. Low-resolution mass spectrometry (MS) was performed on a Finnigan ThermoQuest TSQ 7000 instrument using an electrospray ionization (ESI) source or on a Finnigan SSQ 710A instrument (EI-MS, 70 eV). High-resolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass Q-TOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K-2001 detector. A YMC Triart C18 (150 x 20 mm, 5 µm, YMC Europe GmbH, Dinslacken, Germany) served as RP-column at a flow rate of 15 mL/min at room temperature. A detection wavelength of 220 nm and mixtures of CH₃CN and 0.1% ag. NH₃ as mobile phase were used throughout. CH₃CN was removed from the eluates under reduced pressure (final pressure: 80 mbar) at 45 °C prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a Vacuubrand RZ 6 rotary vane vacuum pump). Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A (250 x 4.6 mm, 5 μ m, t_0 = 2.9 min, Phenomenex, Aschaffenburg, Germany) served as RP-column for acidic runs (flow rate of 0.8 mL/min) and mixtures of 0.05% TFA in CH₃CN (A) and 0.05% aq. TFA (B) were used as mobile phase. A YMC Triart C18 (150 x 2 mm, 5 μ m, t_0 = 1.63 min; YMC, Japan) served as RP-column for basic runs (flow rate of 0.35 mL/min) and mixtures of 0.1% NH_3 in CH₃CN (A) and 0.1% aq. NH₃ (B) were used as mobile phase. Helium degassing, room temperature and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration: 100-500 μ M) were either prepared from stock solution (10 mM in DMSO) in a mixture of CH₃CN and H₂O corresponding to the initial eluent composition, or as a one to one mixture of the eluate (preparative HPLC) with Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 3 (basic conditions, YMC Triat): 0-30 min: A/B 10:90-90:10, 30-32 min: 90:10-95:5, 32-42 min: 95:5. Microanalysis was performed on a Vario micro cube (Elementar, Langenselbold, Germany).

4.3.2 Experimental protocols and analytical data

3-(Piperidin-1-ylmethyl)phenol (4.1)^{2,6}

Piperidine (4.18 g, 49.1 mmol, 2 eq) and formic acid (98%, 3,3 g, 71,3 mmol, 2,6 eq) were added under ice cooling to 3-hydroxybenzaldehyde (3.00 g, 24.6 mmol, 1 eq) and the reaction mixture was stirred at 110 °C for 3 h. After cooling to room temperature, the reaction mixture was poured in H₂O (15 mL). The aqueous solution was alkalized using ammonia solution (25%, w/w). The precipitated white solid product was filtered off and dried *in vacuo* (4.06 g, 87%). Mp: 134-138 °C (Lit.² mp. 136-137 °C). R_f = 0.2 (EtOAc/PE 85:15). ¹H-NMR (400 MHz, [D₆]DMSO, COSY): δ (ppm) 1.37-1.40 (m, 2H), 1.45-1.50 (m, 4H), 2.29 (br s, 4H), 3.31 (s, 2H), 6.59-6.62 (m, 1H), 6.67-6.68 (d, 1H, *J* 7.4 Hz), 6.70-6.71 (m, 1H), 7.05-7.09 (t, 1H, *J* 7.7 Hz), 9.23 (s, 1H). ¹³C-NMR (100 MHz, [D₆]DMSO, COSY): δ (ppm) 24.0, 25.6, 53.9, 62.9, 113.7, 115.4, 119.3, 128.9, 140.1, 157.2. MS (LC-MS, CI, NH₃, t_R = 11.7 min): m/z (%) 192.1 (100) [*M*+H]⁺. C₁₂H₁₇NO (191.13).

3-[3-(Piperidin-1-ylmethyl)phenoxy]propanenitrile hydrochloride (4.2)^{2,6}

4.1 (4.00 g, 20.9 mmol, 1 eq) was suspended in acrylonitrile (14 mL, 210 mmol, 10 eq) and a catalytic amount of benzyltrimethylammonium hydroxide (40% in MeOH, 0.2 mL) was added. The reaction mixture was stirred over night under reflux. The excess of acrylonitrile was removed under reduced pressure and the oily residue was dissolved in diethyl ether (25 mL). The organic layer was washed with sodium hydroxide solution (5%, w/w, 20 mL) and H₂O (20 mL). The combined aqueous layers were extracted with diethyl ether (20 mL). The organic layers were combined and dried over Na₂SO₄. The product was precipitated as HCl salt with HCl in 2-propanol (5-6 M). Removal of the solvent *in vacuo* afforded the product as white solid (3.25 g, 55%). Mp: 163.5-165.7 °C (Lit. ² Mp: 161-162 °C). R_f = 0.2 (EtOAc/PE 85:15). ¹H-NMR (400 MHz, [D₆]DMSO, HSQC, NOESY): δ (ppm) 1.28-1.39 (m, 1H), 1.66-1.88 (m, 5H), 2.76-2.85 (m, 2H), 3.03-3.06 (t, 2H, J 5.9 Hz), 3.22-3.25 (d, 2H, J 11.8 Hz), 4.20-4.24 (m, 4H), 7.02-7.05 (dd, 1H, ⁴J 2.0 Hz, ³J 8.3 Hz), 7.18-7.20 (d, 1H, J 7.7 Hz), 7.35-7.39 (m, 2H), 10.92 (br s, 1H). ¹³C-NMR (100 MHz, [D₆]DMSO, HSQC, NOESY): δ (ppm) 17.9, 21.4, 22.0, 51.5, 58.6, 62.8, 115.6, 117.3, 118.8, 124.1, 129.9, 131.4, 157.7. MS (LC-MS, CI, NH₃, t_{R} = 13.5 min): m/z (%) 245.2 (100) [M+H]⁺. C₁₅H₂₀N₂O · HCl (244.34 + 36.46).

3-[3-(Piperidin-1-ylmethyl)phenoxy]propan-1-amine (4.3)^{2,6}

4.2 (3.00 g, 10.71 mmol, 1 eq) was slowly added to a suspension of lithium aluminium hydride (609 mg, 16.90 mmol, 1.6 eq) in anhydrous diethyl ether (30 mL) and the reaction mixture was stirred for 4 h at room temperature. After the remaining lithium aluminium hydride was hydrolyzed with sodium hydroxide solution (5%, w/w, 35 mL), diethyl ether (50 mL) was added and the insoluble material was filtered off. The two layers were separated and the organic layer was dried over Na₂SO₄. Removal of the solvent *in vacuo* afforded the product as colorless oil (2.59 g, 99.6%). $R_{\rm f}$ = 0.1 (EtOAc/MeOH + 1-3 drops TEA 80:20). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.36 (br s, 2H), 1.39-1.45 (m, 2H), 1.54-1.59 (m, 4H), 1.88-1.95 (qui, 2H, *J* 6.5 Hz), 2.36 (m, 4H), 2.90 (t, 2H, *J* 6.7 Hz), 3.43 (s, 2H), 4.04 (t, 2H, *J* 6.2 Hz), 6.76-6.78 (m, 1H), 6.87-6.89 (m, 2H), 7.19 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 24.5, 26.1, 33.3, 39.4, 54.6, 63.9, 65.9, 113.0, 115.3, 121.6, 129.1, 140.5, 159.1. HRMS (ESI): $m/z [M+H]^+$ calcd. for C₁₅H₂₅N₂O⁺: 249.1961, found: 249.1961. C₁₅H₂₄N₂O (248.37).

Phenyl-N'-cyano-N-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)carbamimidate (4.4) ⁸

4.3 (600 mg, 2.42 mmol, 1 eq) was added to a solution of diphenyl-N-cyanocarbonimidate (633 mg, 2,66 mmol, 1,1 eq) in 2-propanol (90 mL) and the reaction mixture was stirred for 3 h at room temperature. The solvent was removed under reduced pressure and the product was recristallized in diethylether. **4.4** obtained as a white solid (877 mg, 93%) which still contained the by-product phenol and was directly used for the next step. $R_{\rm f} = 0.9$ (CH₂Cl₂ / 7 N NH₃ in MeOH 90:10). MS: (LC-MS, ESI, $t_{\rm R} = 5.2$ min): m/z (%) 393,1 (100) $[M+{\rm H}]^+$. C₂₃H₂₈N₄O₂ (392.22).

1-(2-Aminoethyl)-2-cyano-3-(3-[3-(1-piperidinylmethyl)phenoxy]propyl)guanidine (4.5)¹

Ethylendiamine (307 mg, 5.10 mmol, 20 eq) was added to a stirring solution of 4.4 (100 mg, 0.26 mmol, 1 eq) in CH₃CN (5 mL). The reaction mixture was stirred for 15 min at 150 °C under microwave radiation. The solvent was removed under reduced pressure and the residue was dissolved in CH₂Cl₂ (7 mL). The organic layer was washed three times with H₂O (7 mL) and three times with aqueous sodium hydroxide solution (5 M, 7 mL). The organic layer was dried over Na₂SO₄. Removal of the solvent *in vacuo* afforded the product as colorless oil (55 mg, 60%). $R_{\rm f}$ = 0.3 (CH₂Cl₂ / 7 N NH₃ in MeOH 90:10).¹H-NMR (400 MHz, CDCl₃, COSY, HSQC, HMBC): δ (ppm) 1.39-1.41 (m, 2H), 1.50-1.56 (m, 4H), 1.97-2.03 (qui, 2H, J 6.2 Hz), 2.33 (br s, 4H), 2.80 (t, 2H, J 5.2 Hz), 3.18 (t, 2H, J 5.2 Hz), 3.39 (m, 4H), 4.00 (t, 2H, J 5.8 Hz), 6.20 (br s, 0.8H), 6.74-6.76 (m, 1H), 6.85-6.88 (m, 2H), 7.17 (t, 1H, J 7.7 Hz). ¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.46-1.48 (m, 2H), 1.57-1.63 (m, 4H), 2.02-2.08 (m, 2H), 2.42 (br s, 4H), 2.72-2.75 (t, 2H, J 6.3 Hz), 3.23-3.26 (t, 2H, J 6.3 Hz), 3.41-3.44 (t, 2H, J 6.7 Hz), 3.47 (s, 2H), 4.05-4.08 (t, 2H, J 5.8 Hz), 6.86-6.90 (m, 2H), 6.95-6.96 (m, 1H), 7.20-7.25 (t, 1H, J 7.8 Hz). ¹³C-NMR (100 MHz, CDCl₃, COSY, HSQC, HMBC): δ (ppm) 24.4, 26.0, 28.9, 39.1, 41.8, 45.3, 54.6, 63.8, 65.4, 112.9, 115.1, 119.2, 122.0, 129.2, 140.5, 158.6, 161.3. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₉H₃₁N₆O⁺: 359.2554, found: 359.2552. C₁₉H₃₀N₆O (358.49).

General procedure for the synthesis of the ester protected benzoic acid derivatives 4.9-4.11

In order to remove residual water and the stabilisator from the CHCl₃, the solvent was filtered through extra dry AlO₃ (50 g for 100 mL). The 4-methyl benzoic acid derivative (1 eq) and AIBN (0.1 eq) were suspended in CHCl₃ (15-20 mL). N-bromosuccinimide (1.5 eq) was added portion wise and under stirring to the hot (75 °C) suspension. The reaction mixture was stirred for 3-5 h at 75 °C. Removal of the solvent under reduced pressure afforded the 4-(bromomethyl)benzoic acid derivatives **4.6-4.8** which were used for the next step without further purification. The crude **4.6**, **4.7** or **4.8** was suspended in EtOH (10 mL) and thionyl chloride (2-3 eq) was added drop wise. The reaction mixture was stirred for 2-24 h at 65 °C. The product was purified by automated flash chromatography. Varying amounts of the respective ethyl 4-(chloromethyl) benzoate derivative were formed as by-product and couldn't be separated by flash chromatography.

Ethyl 4-(bromomethyl)-3-(trifluoromethyl)benzoate (4.9)¹⁴

4.9 was prepared from 4-methyl 3-(trifluoromethyl) benzoic acid (500 mg, 2.45 mmol, 1 eq), AIBN (40 mg, 0.24 mmol, 0.1 eq), N-bromosuccinimide (654 mg, 3.67 mmol, 1.5 eq) in CHCl₃ (20 mL) and ester formation with thionyl chloride (583 mg, 4.90 mmol, 2 eq) in EtOH (10 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-97:3 in 20 min) and removal of the solvent *in vacuo* afforded the product as colourless oil (280 mg, 43%). $R_{\rm f} = 0.45$ (PE/EtOAc 95:5).¹H-NMR (300 MHz, CDCl₃, COSY, HSQC, HMBC): δ (ppm) 1.41 (t, 3H, J 7.14 Hz), 4.38-4.45 (q, 2H, J 7.14 Hz), 4.78 (s, 2H), 7.74-7.77 (m, 1H), 8.22-8.25 (m, 1H), 8.33 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃, COSY, HSQC, HMBC): δ (ppm) 14.3, 41.4, 61.7, 123.6 (q, 1C, $J_{C,F}$ 275.25 Hz), 127.35 (q, 1C, $J_{C,F}$ 5.85 Hz), 128.5 (q, 1C, $J_{C,F}$ 31.52 Hz), 130.8, 131.9, 133.2, 140.13 (q, 1C, $J_{C,F}$ 1.40 Hz), 164.9. ¹⁹F-NMR (282 MHz, CDCl₃): δ (ppm) -60.24. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₁H₁₁ClF₃O₂⁺: 267.0394, found: 267.0395. C₁₁H₁₀BrF₃O₂ (311.10).

Ethyl 4-(bromomethyl)-3-chlorobenzoate (4.10)¹⁵

4.10 was prepared from 3-chloro 4-methyl benzoic acid (380 mg, 2.23 mmol, 1 eq), AIBN (37 mg, 0.22 mmol, 0.1 eq), N-bromosuccinimide (595 mg, 3.34 mmol, 1.5 eq) in CHCl₃ (15 mL) and ester formation with thionyl chloride (691 mg, 6.69 mmol, 3 eq) in EtOH (10 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-97.5:2.5 in 25 min) and removal of the solvent *in vacuo* afforded the product as colourless oil (320 mg, 52%). $R_{\rm f} = 0.74$ (PE/EtOAc 6:1).¹H-NMR (300 MHz, CDCl₃): δ (ppm) 1.40 (t, 3H, *J* 7.14 Hz), 4.35-4.42 (q, 2H, *J* 7.14 Hz), 4.72 (s, 2H), 7.56 (d, 1H, *J* 8.14 Hz), 7.92-7.95 (m, 1H), 8.06-8.07 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.3, 43.0, 61.6, 128.2 130.6, 130.8, 132.1, 134.1, 139.5, 165.0. HRMS: (ABCI): $m/z [M+H]^+$, calcd. for C₁₀H₁₁BrClO₂⁺: 276.9625, found: 276.9626. HRMS: (ABCI): $m/z [M+H]^+$, calcd. for C₁₀H₁₁Cl₂O₂⁺: 233.0131, found: 233.0131. C₁₀H₁₀BrClO₂ (277.54).

Ethyl 4-(bromomethyl)-3-bromobenzoate (4.11)^{16,17}

4.11 was prepared from 3-bromo 4-methyl benzoic acid (2.00 g, 9.30 mmol, 1 eq), AIBN (153 mg, 0.93 mmol, 0.1 eq), N-bromosuccinimide (2.48 g, 13.95 mmol, 1.5 eq) in CHCl₃ (150 mL) and ester formation with thionyl chloride (2.21 g, 18.60 mmol, 2 eq) in EtOH (50 mL) according to the general procedure. The crude product was purified first by column chromatography (PE/EtOAc 100:0-90:10) and then by automated flash chromatography (PE/EtOAc 100:0-85:15 in 40 min). Removal of the solvent *in vacuo* afforded the product as a slightly yellow oil (1.39 g, 46%). $R_f = 0.60$ (PE/EtOAc 5:1).¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.40 (t, 3H, J 7.19 Hz), 4.36-4.42 (q, 2H, J 7.19 Hz), 4.72 (s, 2H), 7.57 (d, 1H, J 8.01 Hz), 7.98-8.00 (m, 1H), 8.25 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.3, 45.5, 61.5, 123.8, 128.9, 130.6, 132.1, 134.1, 141.1, 164.8. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₁₀H₁₁Br₂O₂⁺: 320.9120, found: 320.9114. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₁₀H₁₁Br₂O₂⁺: 276.9626, found: 276.9624. C₁₀H₁₀Br₂O₂ (322.00).

General procedure for the synthesis of the azide derivatives 4.12-4.14

Sodium azide (4 eq) was added to a solution of **4.9-4.11** (1 eq) in anhydrous DMF (5-10mL). The reaction mixture was stirred for 24 h at room temperature. H_2O (80 mL) was added and the product was extracted three times with CH_2Cl_2 (60 mL). The organic layers were combined and the solvent was removed under reduced pressure. The residue was dissolved in EtOAc (80 mL) and the organic layer was washed three times saturated $CaCl_2$ solution (50 mL), one time with brine (50 mL) and was then dried over Na_2SO_4 . Removal of the solvent afforded the desired product.

Ethyl 4-(azidomethyl)-3-(trifluoromethyl)benzoate (4.12)

4.12 was prepared from **4.9** (490 mg, 1.84 mmol, 1 eq) and sodium azide (1478 mg, 7.35 mmol, 4 eq) according to general procedure. Removal of the solvent *in vacuo* afforded the product as a slightly yellow oil (430 mg, 83%). $R_{\rm f}$ = 0.5 (PE/EtOAc 95:5).¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 1.34 (t, 3H, *J* 7.10 Hz), 4.32-4.39 (q, 2H, *J* 7.10 Hz), 4.76 (s, 2H), 7.85-7.88 (m, 1H), 8.20 (m, 1H), 8.26-8.29 (m, 1H). ¹³C-NMR (75 MHz, [D₆]DMSO): δ (ppm) 13.9, 49.86 (q, 1C, $J_{\rm C,F}$ 2.08 Hz), 61.4, 123.4 (q, 1C, $J_{\rm C,F}$ 274.04 Hz), 126.35 (q, 1C, $J_{\rm C,F}$ 5.81 Hz), 127.2 (q, 1C, $J_{\rm C,F}$ 30.85 Hz), 130.1, 131.7, 133.3, 138.70 (q, 1C, $J_{\rm C,F}$ 1.46 Hz), 164.1. ¹⁹F-NMR (282 MHz, [D₆]DMSO): δ (ppm) -58.39. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₁H₁₁F₃N₃O₂⁺: 274.0798, found: 274.0797. C₁₁H₁₀F₃N₃O₂ (273.22).

Ethyl 4-(azidomethyl)-3-chlorobenzoate (4.13)

4.13 was prepared from **4.10** (260 mg, 1.12 mmol, 1 eq) and sodium azide (290 mg, 4.46 mmol, 4 eq) according to general procedure. Removal of the solvent *in vacuo* afforded the product as colorless oil (130 mg, 49%). $R_{\rm f}$ = 0.44 (PE/EtOAc 95:5).¹H-NMR (300 MHz, CDCl₃): δ (ppm) 1.40 (t, 3H, J 7.12 Hz), 4.35-4.42 (q, 2H, J 7.13 Hz), 4.68 (s, 2H), 7.54-7.57 (d, 1H, J 8.00 Hz), 8.00-8.03 (m, 1H), 8.50-8.51 (m, 1H). ¹³C-NMR (75 MHz, CDCl₃): δ (ppm) 14.4, 50.5, 61.7, 98.87, 129.91, 129.95,

132.02, 140.9, 144.4, 164.7. HRMS: (ESI): *m*/*z* [*M*+H]⁺, calcd. for C₁₀H₁₁ClN₃O₂⁺: 240.0534, found: 240.0538. C₁₀H₁₀ClN₃O₂ (239.66).

Ethyl 4-(azidomethyl)-3-bromobenzoate (4.14)

4.14 was prepared from **4.11** (290 mg, 1.05 mmol, 1 eq) and sodium azide (272mg, 4.18 mmol, 4 eq) according to general procedure. Removal of the solvent *in vacuo* afforded the product as yellow oil (270 mg, 91%). $R_{\rm f}$ = 0.33 (PE/EtOAc 95:5).¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.40 (t, 3H, *J* 7.12 Hz), 4.36-4.42 (q, 2H, *J* 7.12 Hz), 4.55 (s, 2H), 7.48-7.50 (d, 1H, *J* 7.96 Hz), 8.00-8.02 (m, 1H), 8.26 (m, 1H). HRMS: (ESI): $m/z [M+H]^+$, calcd. for C₁₀H₁₁BrN₃O₂⁺: 284.0029, found: 284.0032. C₁₀H₁₀BrN₃O₂ (284.11).

General procedure for the synthesis of the amine derivatives 4.15-4.17

4.12, 4.13 or **4.14** (1 eq) and triphenylphosphine (1.12 eq) were dissolved in a mixture of THF and H_2O (5/1, v/v, 4 mL). The reaction mixture was stirred over night at room temperature. The THF was removed under reduced pressure and the residue was diluted with H_2O (1-7 mL). The pH value was adjusted to two by adding aqueous HCl solution (0.5 M). The aqueous layer was washed three times with EtOAc (10-20 mL). Removal of the H_2O afforded the desired product as HCl salt.

Ethyl 4-(aminomethyl)-3-(trifluoromethyl)benzoate (4.15)

4.15 was prepared from **4.12** (170 mg, 0.62 mmol, 1 eq) and triphenylphosphine (183 mg, 0.70 mmol, 1.12 eq) dissolved in a mixture of THF and H₂O (5/1, v/v, 4 mL) according to general procedure. Removal of the solvent *in vacuo* afforded the product as yellow solid (140 mg, 80%). $R_{\rm f} = 0.36$ (CH₂Cl₂ / 2 N NH₃ in MeOH 95:5).¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 1.34 (t, 3H, J 7.11 Hz), 4.24 (br s, 2H), 4.33-4.40 (q, 2H, J 7.12 Hz), 8.02-8.05 (d, 1H, J 8.17 Hz), 8.21 (m, 1H), 8.29-8.32 (m, 1H), 8.99 (br s, 3H). ¹³C-NMR (75 MHz, [D₆]DMSO): δ (ppm) 13.9, 38.3(q, 1C, $J_{\rm C-F}$ 3.40 Hz), 61.4, 123.4 (q, 1C, $J_{\rm C-F}$ 274.35 Hz), 126.06 (q, 1C, $J_{\rm C-F}$ 5.56 Hz), 127.3 (q, 1C, $J_{\rm C-F}$ 30.57 Hz), 130.1, 131.0, 133.0, 137.1 (q, 1C, $J_{\rm C-F}$ 1.41 Hz), 164.0. ¹⁹F-NMR (282 MHz, [D₆]DMSO): δ (ppm) - 58.24. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₁H₁₃F₃NO₂⁺: 248.0893, found: 248.0895. C₁₁H₁₂F₃NO₂ (247.22).

Ethyl 4-(aminomethyl)-3-chlorobenzoate (4.16)

4.16 was prepared from **4.13** (920 mg, 3.84 mmol, 1 eq) and triphenylphosphine (1128 mg, 4.30 mmol, 1.12 eq) dissolved in a mixture of THF and H₂O (5/1, v/v, 20 mL) according to general procedure. Removal of the solvent *in vacuo* afforded the product as white solid (630 mg, 66%). $R_{\rm f}$ = 0.5 (CH₂Cl₂ / 2 N NH₃ in MeOH 95:5). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 1.33 (t, 3H, J 7.08 Hz), 4.19 (br s, 2H), 4.31-4.37 (q, 2H, J 7.10 Hz), 7.81-7.83 (m, 1H), 7.96-7.98 (m, 2H), 8.85 (br s,

3H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 14.0, 39.7, 61.3, 127.7, 129.4, 130.6, 131.4, 133.0, 136.7, 164.1. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₀H₁₃ClNO₂⁺: 214.0629, found: 214.0631. C₁₀H₁₂ClNO₂ (213.66).

Ethyl 4-(aminomethyl)-3-bromobenzoate (4.17)

4.17 was prepared from **4.14** (730 mg, 2.51 mmol, 1 eq) and triphenylphosphine (755 mg, 2.88 mmol, 1.12 eq) dissolved in a mixture of THF and H₂O (5/1, v/v, 15 mL) according to general procedure. Removal of the solvent *in vacuo* afforded the product as white solid (550 mg, 73%). Mp: 212-213 °C. $R_{\rm f}$ = 0.5 (CH₂Cl₂ / 3.5 N NH₃ in MeOH 95:5). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 1.34 (t, 3H, *J* 7.12 Hz), 4.18 (br s, 2H), 4.31-4.37 (q, 2H, *J* 7.11 Hz), 7.77-7.79 (d, 1H, *J* 8.11 Hz), 8.01-8.03 (m, 1H), 8.15 (m, 1H), 8.78 (br s, 3H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 14.0, 41.8, 61.3, 123.1, 128.2, 130.3, 131.4, 132.6, 138.3, 164.0. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₀H₁₃BrNO₂⁺: 258.0124, found: 258.0127. C₁₀H₁₂BrNO₂ (258.12).

4-((1,3-Dioxoisoindolin-2-yl)methyl)benzoic acid (4.19)^{9,10}

(4-Aminomethyl)benzoic acid (**4.18**) (200 mg, 1.32 mmol, 1 eq) and phthalic anhydride (212 mg, 1.46 mmol, 1.1 eq) were suspended in acetic acid (1 mL) and the reaction mixture was stirred under reflux for 3.5 h. The solvent was removed under reduced pressure and the residue was suspended in H₂O (2 mL). The product was filtered off and dried *in vacuo*. **4.19** was obtained as white solid (312 mg, 84%). Mp: 262-266 °C (Lit. mp⁹: 267-270 °C). $R_f = 0.8$ (CH₂Cl₂ / MeOH + 2 drops TFA 95:5). ¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 4.84 (s, 2H), 7.42 (d, 2H, *J* 8.22 Hz), 7.85-7.94 (m, 6H), 12.95 (br s, 1H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 40.6, 123.3, 127.4, 129.6, 129.9, 131.6, 134.6, 141.5, 167.0, 167.7. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₁₆H₁₂NO₄⁺: 282.0761, found: 282.0764. C₁₆H₁₁NO₄ (281.27).

4-(Azidomethyl)-3-bromobenzoic acid (4.20)

4.14 (1.29 g, 4.54 mmol, 1 eq) was suspended in THF (50 mL). Aqueous NaOH solution (1 mol/L, 23 mL, 5 eq) was added and the reaction mixture was stirred over night at room temperature. The pH value was adjusted with aqueous HCl solution (0.5 mol/L) to two. The THF was removed under reduced pressure and the product was extracted with EtOAc (3 x 40 mL). The organic layers were combined, washed with brine (40 mL) and dried over Na₂SO₄. Removal of the solvent *in vacuo* afforded the product as beige solid (1.08 g, 93%). Mp: 124 °C. $R_f = 0.1$ (PE / EtOAc 95:5). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 4.64 (s, 2H), 7.64 (d, 1H, J 7.92 Hz), 7.95-7.98 (m, 1H), 8.12 (m, 1H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 53.8, 123.6, 129.3, 131.2, 133.1, 133.7, 140.0, 166.2. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₈H₇BrN₃O₂⁺: 255.9716, found: 255.9719. C₈H₆BrN₃O₂ (256.06).

4-(Aminomethyl)-3-bromobenzoic acid (4.21)¹⁸

4.20 (280 mg, 1.09 mmol, 1 eq) and triphenylphosphine (321 mg, 1.22 mmol, 1.12 eq) were dissolved in a mixture of THF and H₂O (5/1, v/v, 6 mL). The reaction mixture was stirred over night at room temperature. The yellow precipitate was filtered off and washed with CH₂Cl₂ (1 mL). Removal of residual solvent *in vacuo* afforded the product as yellow solid (67 mg, 27%). ¹H-NMR (400 MHz, [D₆]DMSO + 3 drops of TFA): δ (ppm) 4.20-4.22 (m, 2H), 7.66 (d, 1H, *J* 8.06 Hz), 8.00-8.03 (m, 1H), 8.15 (m, 1H), 8.44 (br s, 3H). ¹³C-NMR (100 MHz, [D₆]DMSO + 3 drops of TFA): δ (ppm) 42.6, 123.6, 129.1, 130.5, 133.2, 133.6, 138.3, 166.1. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₈H₉BrNO₂⁺: 229.9811, found: 229.9811. C₈H₈BrNO₂ (230.06).

General procedure for the synthesis of the benzoic acid derivatives 4.25-4.27

4.15, **4.16** or **4.17** (1 eq) and TEA (3 eq) were dissolved in CH_2Cl_2 (3 mL). Succinimidyl propionate (1.5 eq) was added and the reaction mixture was stirred over night at room temperature. The mixture was diluted with CH_2Cl_2 (5 mL) and the organic layer was washed three times with aqueous HCl solution (0.5 mol/L, 10 mL), two times with H_2O (10 mL) and with brine (10 mL). The organic layer was dried over Na_2SO_4 and the solvent was removed under reduced pressure to afford the intermediates **4.22-4.24**. These intermediates were ester deprotected subsequently. **4.22**, **4.23** or **4.24** was suspended in THF (3-5 mL). Aqueous NaOH solution (1 mol/L, 5 eq) was added and the reaction mixture was stirred over night at room temperature. The pH value was adjusted with aqueous HCl solution (0.5 mol/L) to 2-7. The THF was removed under reduced pressure and the product was extracted with EtOAc (3 x 5-20 mL). The organic layers were combined and dried over Na_2SO_4 . Removal of the solvent under reduced pressure afforded the desired product.

4-(Propionamidomethyl)-3-(trifluoromethyl)benzoic acid (4.25)

4.25 was prepared from **4.15** (80 mg, 0.28 mmol, 1 eq), TEA (86 mg, 0.84 mmol, 3 eq) and succinimidyl propionate (72 mg, 0.42 mmol, 1.5 eq) dissolved in CH_2Cl_2 (3 mL) according to general procedure. **4.22** was obtained as a yellow oily solid (90mg, 99.9%). Removal of the ester protecting group by aqueous NaOH solution (1 mol/L, 1.4 mL) afforded **4.25** as slightly pink solid (50 mg, 66%). $R_f = 0.2$ (PE / EtOAc +2 drops TFA 1:2). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 1.10 (t, 3H, *J* 7.59 Hz), 2.27 (q, 2H, *J* 7.58 Hz), 4.53-4.54 (m, 2H), 7.66 (d, 1H, *J* 8.28 Hz), 8.21-8.24 (m, 2H), 8.49 (t, 1H, *J* 5.70 Hz). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 9.7, 28.3, 38.5 (q, 1C, J_{C-F} 2.89 Hz), 123.9 (q, 1C, J_{C-F} 273.79 Hz), 126.1 (q. 1C, J_{C-F} 5.68 Hz), 126.3 (q, 1C, J_{C-F} 30.78 Hz), 128.9, 130.1, 133.1, 142.6 (q,1C, J_{C-F} 1.46 Hz), 165.9, 173.2. ¹⁹F-NMR (282 MHz, [D₆]DMSO): δ (ppm) - 59.5. HRMS: (ESI): m/z [M+H]⁺, calcd. for $C_{12}H_{13}F_3NO_3$ ⁺: 276.0842, found: 276.0843. $C_{12}H_{12}F_3NO_3$ (275.23).

3-Chloro-4-(propionamidomethyl)benzoic acid (4.26)

4.26 was prepared from **4.16** (200 mg, 0.80 mmol, 1 eq), TEA (243 mg, 2.40 mmol, 3 eq) and succinimidyl propionate (205 mg, 1.20 mmol, 1.5 eq) dissolved in CH_2CI_2 (8 mL) according to general procedure. **4.23** was obtained as a white oily solid (230 mg). Removal of the ester protecting group by aqueous NaOH solution (1 mol/L, 4.1 mL) afforded **4.26** as white solid (150 mg, 84%). $R_f = 0.2$ (PE / EtOAc +2 drops TFA 1:2). ¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 1.04 (t, 3H, *J* 7.58 Hz), 2.21 (q, 2H, *J* 7.58 Hz), 4.35-4.37 (m, 2H), 7.42 (d, 1H, *J* 7.85 Hz), 7.86-7.89 (m, 2H), 8.40 (t, 1H, *J* 5.72 Hz), 13.26 (s, 1H). ¹³C-NMR (75 MHz, [D₆]DMSO): δ (ppm) 9.8, 28.2, 39.9, 127.9, 128.6, 129.5, 131.0, 132.0, 141.4, 165.9, 173.2. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₁₁H₁₃ClNO₃ ⁺: 242.0579, found: 242.0581. C₁₁H₁₂ClNO₃ (241.67).

3-Bromo-4-(propionamidomethyl)benzoic acid (4.27)

4.27 was prepared from **4.17** (260 mg, 1.01 mmol, 1 eq), TEA (204 mg, 2.02 mmol, 2 eq) and succinimidyl propionate (258 mg, 1.51 mmol, 1.5 eq) dissolved in CH_2CI_2 (5 mL) according to general procedure. **4.24** was obtained as a white oily solid (220 mg, 70%). Removal of the ester protecting group by aqueous NaOH solution (1 mol/L, 3.2 mL) afforded **4.27** as white solid (90 mg, 49%). $R_f = 0.2$ (PE / EtOAc +2 drops TFA 1:2). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 1.05 (t, 3H, *J* 7.58 Hz), 2.22 (q, 2H, *J* 7.58 Hz), 4.33 (d, 2H, *J* 5.87 Hz), 7.39-7.41 (m, 1H), 7.90-7.93 (m, 1H), 8.06-8.07 (m, 1H), 8.39 (t, 1H, *J* 5.86 Hz), 13.23 (br s, 1H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 9.8, 28.3, 42.4, 122.0, 128.4, 128.5, 131.2, 132.8, 143.0, 165.8, 173.2. HRMS: (ESI): m/z [M+H]⁺, calcd. for $C_{11}H_{13}BrNO_3^+$: 288.0052, found: 288.0055. $C_{11}H_{12}BrNO_3$ (286.13).

General procedure for the methylation of 4.19 and 4.21 (synthesis of 4.28 and 4.29)

4.19 or **4.21** was dissolved in formic acid (1 mL) under heating. Aqueous formaldehyde solution (37%, 1 mL) was added and the reaction mixture was stirred over night under reflux. The mixture was cooled to room temperature and aqueous HCl solution (20%, 1mL) was added. Removal of the solvent *in vacuo* afforded the desired product.

4-[(Dimethylamino)methyl]benzoic acid hydrochloride(4.28)¹¹

4.28 was prepared from **4.19** (400 mg, 2.65 mmol, 1 eq) according to general procedure. The product was obtained as white solid (570 mg, 100%). ¹H-NMR (400 MHz, MeOD + 3 drops of TFA): δ (ppm) 2.88 (s, 6H), 4.42 (s, 2H), 7.65-7.67 (m, 2H), 8.11-8.13 (m, 2H). ¹³C-NMR (100 MHz, MeOD + 3 drops of TFA): δ (ppm) 41.8, 60.1, 130.2, 130.8, 132.3, 134.3, 167.4. HRMS: (ESI): $m/z [M+H]^+$, calcd. for C₁₀H₁₄NO₂ ⁺: 180.1019, found: 180.1022. C₁₀H₁₃NO₂ · HCl (179.22 + 36.46).

3-Bromo-4-[(dimethylamino)methyl]benzoic acid hydrochloride (4.29)

4.29 was prepared from **4.21** (50 mg, 0.22 mmol, 1 eq) according to general procedure. The product was obtained as yellow hygroscopic solid (60 mg, 94%). ¹H-NMR (400 MHz, MeOD + 3 drops of TFA): δ (ppm) 2.96 (s, 6H), 4.60 (s, 2H), 7.81 (d, 1H, *J* 7.99 Hz), 8.07-8.09 (m, 1H), 8.30-8.31 (m, 1H). ¹³C-NMR (100 MHz, MeOD + 3 drops of TFA): δ (ppm) 42.4, 59.8, 125.3, 129.1, 133.1, 133.8, 134.28, 134.30, 165.9. HRMS: (ESI): $m/z [M+H]^+$, calcd. for C₁₀H₁₃BrNO₂⁺: 258.0124, found: 258.0145. C₁₀H₁₂BrNO₂ · HCl (258.12 + 36.46).

General procedure for the synthesis of the aminopotentidine derivatives 4.30-4.34

4-Aminobenzoic acid, 4-amino-3-(trifluoromethyl)benzoic acid, 4-amino-3-chlorobenzoic acid, 4amino-3-bromobenzoic acid or 4-amino-3-iodobenzoic acid (1-1.5 eq) and DIPEA (3 eq) were dissolved in CH_2Cl_2 (5-100 mL). *O*-(Benzotriazol-1-yl)-*N*,*N*,*N'*,*N'*-tetramethyluroniumtetrafluoroborate (TBTU, 1.2-1.5 eq) was added and the mixture was stirred for 10-30 min. Subsequently, **4.5** (1eq) was added and the reaction mixture was stirred over night at room temperature. The organic layer was further diluted with CH_2Cl_2 (10-100 mL) and washed three times with H_2O (10-100 mL) and three times with aqueous NaOH solution (5%, w/w, 10-100 mL). The organic layer was dried over Na_2SO_4 and the solvent was removed *in vacuo*. The resulting crude product was either directly used in the next synthesis step or purified by preparative HPLC.

4-Amino-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Aminopotentidine, 4.30)^{1,3}

4.30 was prepared from 4-aminobenzoic acid (29 mg, 0.21 mmol, 1.5 eq), DIPEA (54 mg, 0.42 mmol, 3 eq), TBTU (67 mg, 0.21 mmol, 1.5 eq) and **4.5** (50 mg, 0.14 mmol, 1 eq) according to general procedure. Purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 33:67-70:30, $t_{\rm R}$ = 10.6 min) afforded the product as white solid (23 mg, 34%). Mp: 66-80 °C (Lit.^{1.3} Mp: 92-95 °C). $R_{\rm f}$ = 0.5 (CH₂Cl₂/3 M NH₃ in MeOH 7:2). RP-HPLC (gradient 2, 220 nm): 98% ($t_{\rm R}$ = 13.61 min, k = 3.7). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.47-1.49 (m, 2H), 1.63-1.71 (m, 4H), 2.02-2.09 (qui, 2H, *J* 6.1 Hz), 2.59 (br s, 4H), 3.38-3.55 (m, 6H), 3.63 (br s, 2H), 4.03-4.07 (m, 4H), 6.59-6.64 (d, 2H, *J* 8.7 Hz), 6.84-6.92 (m, 2H), 7.01 (br s, 1H), 7.07 (br s, 1H), 7.22 (t, 1H, *J* 7.9 Hz), 7.57-7.60 (d, 2H, *J* 8.6 Hz). ¹H-NMR (600 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 1.50 (br s, 2H), 1.62-1.66 (m, 4H), 2.02-2.06 (qui, 2H, *J* 6.2 Hz), 2.61 (br s, 4H), 3.37-3.39 (m, 2H), 3.42-3.47 (m, 4H), 3.65 (br s, 2H), 4.06 (t, 2H, *J* 5.8 Hz), 6.63-6.65 (d, 2H, *J* 8.6 Hz), 6.90-6.93 (m, 2H), 6.98 (br s, 1H), 7.25 (t, 1H, *J* 7.9 Hz), 7.56-7.59 (d, 2H, *J* 8.9 Hz). ¹³C-NMR (150 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 24.5, 25.8, 29.9, 40.4, 40.5, 42.7, 55.0, 63.9, 66.8, 114.7, 115.3, 117.5, 119.9, 122.7, 123.7, 130.0, 130.6, 137.6, 153.4, 160.4, 161.5, 171.1. HRMS: (ESI): $m/z [M+H]^+$, calcd. for C₂₆H₃₆N₇O₂⁺: 478.2925, found: 478.2927. C₂₆H₃₅N₇O₂ (477.61).

4-Amino-3-trifluoromethyl-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Trifluoromethylaminopotentidine, 4.31)

4.31 was prepared from 4-amino-3-trifluoromethylbenzoic acid (172 mg, 0.84 mmol, 1 eq), DIPEA (325 mg, 2.51 mmol, 3 eq), TBTU (323 mg, 1.00 mmol, 1.2 eq) and **4.5** (300 mg, 0.84 mmol, 1 eq) according to general procedure. The solvent was removed under reduced pressure and the residue was purified by column chromatography (CH₂Cl₂/3.5 M NH₃ in MeOH 97.5:2.5-95:5). Removal of the solvent *in vacuo* afforded the product as yellow solid (267 mg, 68%). 70 mg was further purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 30:70-60:40, t_R = 14.4 min). **4.31** was obtained as white solid (49 mg, 66%). Mp: 62-88 °C. R_f = 0.4 (CH₂Cl₂/7 M NH₃ in MeOH 97:3). RP-HPLC (gradient 2, 220 nm): 96% (t_R = 17.83 min, k = 5.2). ¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.43-1.44 (m, 2H), 1.54-1.60 (m, 4H), 2.00-2.06 (m, 2H), 2.39 (br s, 4H), 3.38-3.48 (m, 8H), 4.04 (t, 2H, *J* 5.87 Hz), 6.81-6.88 (m, 3H), 6.92 (m, 1H), 7.19 (t, 1H, *J* 7.88 Hz), 7.72-7.74 (m, 1H), 7.93-7.94 (m, 1H). ¹³C-NMR (100 MHz, CD₃OD): δ (ppm) 25.2, 26.5, 30.0, 40.4, 40.6, 42.6, 55.4, 64.7, 66.7, 112.7 (q, 1C, J_{C-F} 30.35 Hz), 114.7, 117.0, 117.4, 120.0, 122.1, 123.5, 126.4 (q, 1C, J_{C-F} 271.34 Hz), 127.6 (q, 1C, J_{C-F} 5.39 Hz), 130.3, 133.0, 139.8, 150.5 (q, 1C, J_{C-F} 1.63 Hz), 160.3, 161.6, 169.7. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₇H₃₅F₃N₇O₂⁺: 546.2799, found: 546.2802. C₂₇H₃₄F₃N₇O₂ (545.61).

4-Amino-3-chloro-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Chloroaminopotentidine, 4.32)

4.32 was prepared from 4-amino-3-chlorobenzoic acid (96 mg, 0.56 mmol, 1 eq), DIPEA (216 mg, 1.67 mmol, 3 eq), TBTU (215 mg, 0.67 mmol, 1.2 eq) and **4.5** (200 mg, 0.56 mmol, 1 eq) according to general procedure. The solvent was removed under reduced pressure and the residue was purified by automated flash chromatography (CH₂Cl₂/MeOH 100:0-75:25 in 53 min). Removal of the solvent *in vacuo* afforded the product as white solid (140 mg, 49%). 90 mg was further purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 30:70-70:30, $t_{\rm R}$ = 20.6 min). **4.32** was obtained as white solid (60 mg, 33%). Mp: 100-103 °C. $R_{\rm f}$ = 0.4 (CH₂Cl₂/1.7 M NH₃ in MeOH 90:10). RP-HPLC (gradient 1, 220 nm): 97.2% ($t_{\rm R}$ = 17.74 min, k = 5.1). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.35 (br s, 2H), 1.45-1.47 (m, 4H), 1.89-1.93 (m, 2H), 2.27 (br s, 4H), 3.23-3.35 (m, 8H, interfering with the water signal), 3.95 (t, 2H, *J* 6.13 Hz), 5.88 (s, 2H), 6.76-6.77 (m, 2H), 6.82-6.83 (m, 2H), 7.07-7.19 (m, 3H), 7.53-7.55 (m, 1H), 7.73 (m, 1H), 8.30 (m, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 24.0, 25.5, 28.6, 38.4, 38.8, 41.0, 53.9, 62.8, 65.0, 112.7, 114.1, 114.6, 116.1, 118.0, 120.9, 122.2, 127.2, 128.4, 129.0, 140.3, 147.4, 158.5, 159.5, 165.7. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₆H₃₅ClN₇O₂⁺: 512.2535, found: 512.2544. C₂₆H₃₄ClN₇O₂ (512.06).

4-Amino-3-bromo-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (Bromoaminopotentidine, 4.33)

4.33 was prepared from 4-amino-3-bromobenzoic acid (121 mg, 0.56 mmol, 1 eq), DIPEA (216 mg, 1.67 mmol, 3 eq), TBTU (215 mg, 0.67 mmol, 1.2 eq) and **4.5** (200 mg, 0.57 mmol, 1 eq)

according to general procedure. The crude product was obtained as sticky yellow solid (315 mg). 215 mg was purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 40:60-80:20, t_{R} = 13.1 min). **4.33** was obtained as white solid (115 mg, 54%). Mp: 94 °C. R_f = 0.5 (CH₂Cl₂/3 M NH₃ in MeOH 90:10). IR (KBr): 3325, 2935, 2165, 1585, 1500, 1300,1255, 1160, 755 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 98.7% ($t_{\rm R}$ = 16.76 min, k = 4.8). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.42-1.43 (m, 2H), 1.54-1.60 (m, 4H), 2.00-2.07 (qui, 2H, J 6.1 Hz), 2.41 (br s, 4H), 3.38-3.52 (m, 8H), 4.02 (t, 2H, J 5.7 Hz), 4.53 (s, 2H), 6.35 (br s, 1H), 6.64 (br s, 1H), 6.68 (d, 1H, J 8.4 Hz), 6.79-6.82 (dd, 1H, J 8.1 Hz, J 2.1 Hz), 6.88 (d, 1H, J 7.6 Hz), 6.92 (br s, 1H), 7.19 (t, 1H, J 7.8 Hz), 7.44 (br s, 0.9H), 7.51-7.54 (dd, 1H, J 8.4 Hz, J 2.0 Hz), 7.91 (d, 1H, J 2.0 Hz). ¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.42-1.44 (m, 2H), 1.53-1.59 (m, 4H), 1.99-2.05 (qui, 2H, J 6.4 Hz), 2.37 (br s, 4H), 3.35-3.46 (m, 8H), 4.03 (t, 2H, J 5.8 Hz), 6.78 (d, 1H, J 8.6 Hz), 6.81-6.87 (m 2H), 6.90-6.91 (m, 1H), 7.19 (t, 1H, J 7.9 Hz), 7.54-7.56 (dd, 1H, J 8.5 Hz, J 2.1 Hz), 7.89 (d, 1H, J 2.1 Hz). ¹³C-NMR (100 MHz, CD₃OD): δ (ppm) 25.1, 26.4, 30.0, 40.3, 40.5, 42.6, 55.4, 64.7, 66.6, 108.3, 114.6, 115.3, 116.9, 119.9, 123.4, 124.1, 128.8, 130.2, 133.3, 139.8, 150.3, 160.2, 161.5, 169.5. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂₆H₃₅BrN₇O₂⁺: 556.2030, found: 556.2032. C₂₆H₃₄BrN₇O₂ (556.51).

4-Amino-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-3iodobenzamide (Iodoaminopotentidine, 4.34)¹

4.34 was prepared from 4-amino-3-iodobenzoic acid (147 mg, 0.56 mmol, 1 eq), DIPEA (216 mg, 1.67 mmol, 3 eq), TBTU (215 mg, 0.67 mmol, 1.2 eq) and **4.5** (200 mg, 0.57 mmol, 1 eq) according to general procedure. The crude product was obtained as yellow oil (419 mg). 319 mg was purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 40:60-80:20, $t_{\rm R}$ = 14.0 min). **4.34** was obtained as white solid (166 mg, 65%). Mp: 94 °C (Lit.¹ Mp: 114-117 °C decomposition). $R_{\rm f}$ = 0.5 (CH₂Cl₂/7 M NH₃ in MeOH 90:10). IR (KBr) 3320, 2935, 2165, 1590, 1490, 1300, 1260, 1150 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 98.3% ($t_{\rm R}$ = 17.46 min, k = 5.0).¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.43-1.44 (m, 2H), 1.54-1.60 (m, 4H), 2.00-2.06 (qui, 2H, *J* 6.3 Hz), 2.39 (br s, 4H), 3.35-3.46 (m, 8H), 4.03 (t, 2H, *J* 5.8 Hz), 6.74 (d, 1H, *J* 8.5 Hz), 6.82-6.88 (m, 2H), 6.91 (m, 1H), 7.20 (t, 1H, *J* 7.9 Hz), 7.56-7.59 (dd, 1H, *J* 8.5 Hz, *J* 2.1 Hz), 8.11 (d, 1H, *J* 2.1 Hz). ¹³C-NMR (100 MHz, CD₃OD): δ (ppm) 25.2, 26.5, 30.1, 40.4, 40.6, 42.6, 55.4, 64.8, 66.7, 82.3, 114.2, 114.7, 117.0, 120.0, 123.5, 124.7, 129.8, 130.3, 139.8, 139.9, 153.0, 160.3, 161.6, 169.4. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂₆H₃₄IN₇O₂: C 51.75, H 5.68, N 16.25, found: C 51.25, H 5.66, N 16.19. C₂₆H₃₄IN₇O₂ (603.51).

General procedure for the propionylation of the aminopotentidine derivatives

The respective aminopotentidine derivative **4.30-4.34**, 4-(dimetylamino)pyridine (DMAP, 0.1-1.1 eq) and triethylamine (2-5 eq) were dissolved in CH_2Cl_2 (2-3 mL). The mixture was stirred for several minutes and propionyl chloride (3 eq) was added. The reaction mixture was stirred over night at room temperature. The organic layer was further diluted with CH_2Cl_2 (3 mL), washed three times with aqueous NaOH solution (5%, w/w, 5 mL) and then dried over Na₂SO₄. The product was purified by preparative HPLC.

N-(2-[2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino]ethyl)-4-(propionamido)benzamide (4.35)

4.35 was prepared from **4.30** (100 mg, 0.21 mmol, 1 eq), 4-(dimethylamino)pyridine (28 mg, 0.23 mmol, 1.1 eq), triethylamine (106 mg, 1.05 mmol, 5 eq) and propionyl chloride (58 mg, 0.63 mmol, 3 eq) according to general procedure. Due to incomplete conversion additional propionyl chloride (116 mg, 1.26 mmol, 6 eq) and triethylamine (212 mg, 2.1 mmol, 10 eq) were added. The reaction mixture was stirred for 17 h at room temperature. Purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 40:60-70:30, t_R = 11.3 min) afforded the product as white solid (26 mg, 23%). Mp: 171-176 °C. $R_f = 0.3$ (CH₂Cl₂/7 M NH₃ in MeOH 95:5). IR (KBr): 3295, 2935, 2160, 1665, 1590, 1525, 1440, 1375, 1345, 1310, 1260, 1205, 845, 770, 690, 580 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 97.9% ($t_{\rm R}$ = 15,71 min, k = 4.4). ¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.18-1.22 (t, 3H, J 7.6 Hz), 1.45-1.46 (m, 2H), 1.56-1.61 (m, 4H), 2.01-2.07 (qui, 2H, J 6.2 Hz), 2.38-2.44 (m, 6H), 3.40-3.44 (m, 4H), 3.48-3.51 (m, 4H), 4.04 (t, 2H, J 5.9 Hz), 6.84-6.90 (m, 2H), 6.93 (br s, 1H), 7.21 (t, 1H, J 7.8 Hz), 7.65 (d, 2H, J 8.7 Hz), 7.77 (d, 2H, J 8.7 Hz). ¹³C-NMR (100 MHz, CD₃OD): δ (ppm) 10.1, 25.1, 26.4, 30.0, 31.2, 40.4, 40.7, 42.5, 55.3, 64.6, 66.7, 114.8, 117.1, 120.3, 123.5, 129.3, 130.2, 130.3, 139.4, 139.5, 143.5, 160.3, 161.6, 170.3, 175.6. HRMS: (ESI): $m/z [M+H]^+$, calcd. for $C_{29}H_{40}N_7O_3^+$: 534.3187, found: 534.3189. $C_{29}H_{39}N_7O_3$ (533.68).

N-(2-[2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino]ethyl)-3-(trifluoromethyl)-4-(propionamido)benzamide (4.36)

4.36 was prepared from 4.31 (183 mg, 0.34 mmol, 1 eq), 4-(dimethylamino)pyridine (45 mg, 0.37 mmol, 1.1 eq), triethylamine (172 mg, 1.70 mmol, 5 eq) and propionyl chloride (94 mg, 1.02 mmol, 3 eq) according to general procedure. Due to incomplete conversion additional propionyl chloride (94 mg, 1.02 mmol, 3 eq) were added after stirring over night. The reaction mixture was stirred overnight at room temperature. Purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 35:65-55:45, t_R = 24.2 min) afforded the product as white solid (31 mg, 15%). Mp: 137-138 °C. R_f = 0.3 (CH₂Cl₂/7 M NH₃ in MeOH 97:3). RP-HPLC (gradient 2, 220 nm): 96.4% (t_{R} = 17.86 min, k = 5.2). ¹H-NMR (600 MHz, CD₃OD, COSY, HSQC, HMBC, NOESY): δ (ppm) 1.22 (t, 3H, J 7.57 Hz), 1.45 (br s, 2H), 1.56-1.60 (m, 4H), 2.02-2.06 (m, 2H), 2.41-2.48 (m, 6H), 3.41-3.45 (m, 6H), 3.51-3.53 (m, 2H), 4.04-4.06 (m, 2H), 6.84-6.88 (m, 2H), 6.94 (s, 1H), 7.20 (t, 1H, J 7.84 Hz), 7.72-7.73 (m, 1H), 8.03-8.05 (m, 1H), 8.18 (s, 1H). ¹³C-NMR (150 MHz, CD₃OD, COSY, HSQC, HMBC, NOESY): δ (ppm) 10.0, 25.1, 26.4, 30.0, 30.4, 40.3, 40.7, 42.3, 55.3, 64.7, 66.6, 114.7, 116.9, 119.8, 123.4, 124.8 (q, 1C, J_{CF} 272.45 Hz), 126.0 (q, 1C, J_{CF} 30.42 Hz), 126.8 (q, 1C, J_{C-F} 5.27 Hz), 130.2, 130.4, 132.5, 133.4, 139.4, 139.8, 160.2, 161.5, 168.5, 176.3. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₃₀H₃₉F₃N₇O₃⁺: 602.3061, found: 602.3065. Anal. calcd. for C₃₀H₃₈F₃N₇O₃ · H₂O: C 58.15, H 6.51, N 15.82, found: C 57.82, H 6.21, N 15.30. C₃₀H₃₈F₃N₇O₃ (601.68).

N-(2-[2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino]ethyl)-3-bromo-4-(propionamido)benzamide (4.37)

4.37 was prepared from **4.33** (100 mg, 0.18 mmol, 1 eq), 4-(dimethylamino)pyridine (2.2 mg, 0.02 mmol, 0.1 eq), triethylamine (36 mg, 0.36 mmol, 2 eq) and propionyl chloride (25 mg, 0.27 mmol, 1.5 eq) according to general procedure. Due to incomplete conversion additional propionyl chloride (25 mg, 0.27 mmol, 1.5 eq), 4-(dimethylamino)-pyridine (2.2 mg, 0.018 mmol, 0.1 eq) and triethylamine (36 mg, 0.36 mmol, 2 eq) were added after 3 h of stirring. The reaction mixture was stirred over night at room temperature. Purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 40:60-60:40, $t_{\rm R}$ = 17.6 min) afforded the product as white solid (12 mg, 11%). Mp: 155-159 °C decomposition. $R_{\rm f}$ = 0.6 (CH₂Cl₂/7 M NH₃ in MeOH 95:5). IR (KBr): 3310, 2935, 2165, 1600, 1540, 1510, 1465, 1275, 1195, 700, 565 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 95.1% ($t_{\rm R}$ = 17.34 min, k = 5.0). ¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.22 (t, 3H, J 7.6 Hz), 1.44 (m, 2H), 1.55-1.59 (qui, 4H, J 5.6 Hz), 2.01-2.05 (qui, 2H, J 6.2 Hz), 2.40 (br s, 4H), 2.47-2.51 (q, 2H, J 7.6 Hz), 3.40-3.42 (m, 4H), 3.45 (br s, 2H), 3.47-3.49 (m, 2H), 4.03-4.05 (t, 2H, J 5.9 Hz), 6.83-6.85 (dd, 1H, J 8.1 Hz, J 2.0 Hz), 6.87 (d, 1H, J 7.6 Hz), 6.92 (m, 1H), 7.19 (t, 1H, J 7.8 Hz), 7.75-7.77 (dd, 1H, J 8.5 Hz, J 2.0 Hz), 7.90 (d, 1H, J 8.5 Hz), 8.08 (d, 1H, J 2.0 Hz). ¹³C-NMR (100 MHz, CD₃OD): δ (ppm) 10.0, 25.1, 26.4, 30.0, 30.7, 40.3, 40.7, 42.3, 55.3, 64.7, 66.6, 114.7, 116.9, 117.5, 119.8, 123.4, 126.2, 128.0, 130.2, 133.1, 133.2, 139.7, 140.4, 160.2, 161.5, 168.6, 175.5. HRMS: (ESI): $m/z [M+H]^+$, calcd. for $C_{29}H_{39}BrN_7O_3^+$: 612.2292, found: 612.2295. C₂₉H₃₈BrN₇O₃ (612.57).

N-[2-(2-Cyano-3-(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)guanidino)ethyl]-3-iodo-4-(propionamido)benzamide (4.38)

4.38 was prepared from **4.34** (100 mg, 0.17 mmol, 1 eq), 4-(dimethylamino)pyridine (22 mg, 0.18 mmol, 1.1 eq), triethylamine (84 mg, 0.83 mmol, 5 eq) and propionyl chloride (46 mg, 0.5 mmol, 3 eq) according to general procedure. Due to incomplete conversion additional propionyl chloride (46 mg, 0.50 mmol, 3 eq) was added after 5 hours of stirring. The reaction mixture was stirred over night at room temperature. Purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/1% aq. NH₃ 40:60-80:20, t_R = 13.7 min) afforded the product as white solid (24 mg, 22%). Mp: 148-150 °C. R_f = 0.3 (CH₂Cl₂/7 M NH₃ in MeOH 95:5). IR (KBr): 3310, 2935, 2160, 1645, 1595, 1540, 1510, 1315, 1275, 1200, 1155, 1120, 785, 695, 570 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 97.2% ($t_{\rm R}$ = 17.50 min, k = 5.0). ¹H-NMR (400 MHz, CD₃OD): δ (ppm) 1.25 (t, 3H, J 7.6 Hz), 1.44-1.45 (m, 2H), 1.55-1.60 (m, 4H), 2.01-2.07 (qui, 2H, J 6.2 Hz), 2.40 (br s, 4H), 2.45-2.50 (q, 2H, J 7.6 Hz), 3.39-3.44 (m, 6H), 3.47-3.50 (m, 2H), 4.05 (t, 2H, J 5.9 Hz), 6.83-6.88 (m, 2H), 6.93 (m, 1H), 7.20 (t, 1H, J 7.8 Hz), 7.68 (d, 1H, J 8.4 Hz), 7.78-7.81 (dd, 1H, J 8.4 Hz, J 2.0 Hz), 8.33 (d, 1H, J 2.0 Hz). ¹³C-NMR (100 MHz, CD₃OD): δ (ppm) 10.2, 25.2, 26.5, 30.0, 30.8, 40.4, 40.7, 42.4, 55.4, 64.7, 66.7, 114.7, 117.0, 119.9, 123.5, 126.7, 128.9, 130.3, 132.3, 134.0, 139.7, 139.9, 143.5, 160.3, 161.6, 168.5, 175.5. HRMS: (ESI): $m/z [M+H]^{+}$, calcd. for $C_{29}H_{39}IN_7O_3^{+}$: 660.2154, found: 660.2160. C₂₉H₃₈IN₇O₃ (659.57).

N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-((1,3-dioxoisoindolin-2-yl)methyl)benzamide (4.39)

4.19 (390 mg, 1.39 mmol, 1 eq) and DIPEA (538 mg, 4.16 mmol, 3 eq) were dissolved in CH_2Cl_2 (120 mL) and TBTU (534 mg, 1.66, 1.2 eq) was added. The mixture was stirred for 10 min at room temperature. **4.5** (497 mg, 1.39 mmol, 1 eq) dissolved in CH_2Cl_2 (2 mL) was added and the reaction mixture was stirred over night at room temperature. The organic layer was washed two times with H_2O (100 mL), aqueous NaOH solution (5%, w/w, 100 mL) and brine (100 mL). The organic layer was dried over Na_2SO_4 . The solvent was removed under reduced pressure and the residue was purified by column chromatography ($CH_2Cl_2/3.5$ M NH₃ in MeOH 97.5:2.5-95:5). Removal of the solvent *in vacuo* afforded the product as light yellow solid (640 mg, 74%). Mp: 70-72 °C. $R_f = 0.3$ ($CH_2Cl_2/7$ M NH₃ in MeOH 90:10). ¹H-NMR (400 MHz, [D_6]DMSO): δ (ppm) 1.35-1.37 (m, 2H), 1.46-1.47 (m, 4H), 1.87-1.94 (m, 2H), 2.28 (br s, 4H), 3.26-3.36 (m, 8H), 3.94-3.97 (m, 2H), 4.82 (s, 2H), 6.77-6.84 (m, 3H), 7.08-7.20 (m, 3H), 7.38 (d, 2H, J 8.27 Hz), 7.79 (d, 2H, J 8.30 Hz), 7.86-7.92 (m, 2H), 8.57 (t, 1H, J 5.36 Hz). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 19.6, 24.3, 25.8, 28.9, 39.85, 39.94, 41.3, 54.5, 63.7, 65.7, 113.3, 115.3, 119.0, 122.1, 123.6, 127.8, 128.6, 129.3, 132.1, 133.2, 134.3, 139.8, 140.1, 158.6, 160.3, 168.0, 168.6. HRMS: (ESI): m/z [M+H]⁺, calcd. for $C_{35}H_{40}N_7O_4^+$: 622.3136, found: 622.3149. $C_{35}H_{39}N_7O_4$ (621.74).

4-(Aminomethyl)-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.40)

4.39 (360 mg, 0.58 mmol, 1 eq) and hydrazinium hydroxide (145 mg, 2.9 mmol, 5 eq) were dissolved in EtOH (20 mL). The reaction mixture was stirred for 4 h at room temperature and subsequently cooled down in the freezer for 30 min. The precipitated by-product 2,3dihydrophthalazine-1,4-dione was filtered off and washed with EtOH (5 mL). The organic layers were combined and the solvent was removed under reduced pressure. Purification by column chromatography (CH₂Cl₂/3.5 M NH₃ in MeOH 95:5 isocratic) afforded the product as white solid (193 mg, 68%). 70 mg of 4.40 were further purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 30:70-60:40, $t_{\rm R}$ = 14.4 min). The product was obtained as white solid (49 mg, 47%). Mp: 62-88 °C. R_f = 0.75 (CH₂Cl₂/6 M NH₃ in MeOH 80:20). RP-HPLC (gradient 2, 220 nm): 98.9% ($t_{\rm R}$ = 11.84 min, k = 3.1). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.35 (br s, 2H), 1.44-1.48 (m, 4H), 1.91 (qui, 2H, J 6.49 Hz), 2.28 (br s, 4 H), 3.25-3.29 (m, 4H, interfering with the water signal), 3.34-3.37 (m, 4H, interfering with the water signal), 3.76 (s, 2H), 3.95 (t, 2H, J 6.20 Hz), 6.77-6.78 (m, 1H), 6.83-6.84 (m, 2H), 7.08-7.12 (m, 2H), 7.18 (t, 1H, J 7.98 Hz), 7.39 (d, 2H, J 8.16 Hz), 7.77 (d, 2H, J 8.18 Hz), 8.52 (t, 1H, J 5.45 Hz). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 24.0, 25.6, 28.6, 38.4, 38.9, 40.8, 45.1, 53.9, 62.8, 65.0, 112.7, 114.6, 118.0, 120.9, 126.8, 127.1, 129.0, 132.3, 140.3, 147.0, 158.5, 159.5, 166.7. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂₇H₃₈N₇O₂⁺: 492.3081, found: 492.3085. C₂₇H₃₇N₇O₂ (491.64).

N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-(propionamidomethyl)benzamide (4.41)

4.40 (118 mg, 0.24 mmol, 1 eq), triethylamine (49 mg, 0.48 mmol, 2 eq) and succinimidyl propionate (62 mg, 0.36 mmol, 1.5 eq) were dissolved CH₂Cl₂ (5 mL). The reaction mixture was stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 35:65-70:30, $t_{\rm R}$ = 11.8 min) afforded the product as white solid (106 mg, 81%). Mp: 166-170 °C. $R_{\rm f}$ = 0.5 $(CH_2CI_2/7 \text{ M NH}_3 \text{ in MeOH 90:10})$. RP-HPLC (gradient 2, 220 nm): 96.6% (t_R = 14.87 min, k = 4.1). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.02 (t, 3H, J 7.59 Hz), 1.36 (br s, 2H), 1.46-1.47 (m, 4H), 1.91 (qui, 2H, J 6.47 Hz), 2.14 (q, 2H, J 7.60 Hz), 2.29 (br s, 4H), 3.25-3.29 (m, 2H, interfering with the water signal), 3.33-3.36 (m,4H, interfering with the water signal), 3.95 (t, 2H, J 6.15 Hz), 4.28 (d, 2H, J 5.94 Hz), 6.78-6.79 (m, 1H), 6.83-6.84 (m, 2H), 7.07-7.12 (m, 2H), 7.17-7.20 (m, 1H), 7.29 (d, 2H, J 8.07 Hz), 7.77 (d, 2H, J 8.22 Hz), 8.30 (t, 1H, J 5.88 Hz), 8.53 (t, 1H, J 5.46 Hz). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 10.0, 24.0, 25.5, 28.5, 28.6, 38.4, 38.9, 40.8, 41.7, 53.9, 62.7, 65.0, 112.7, 114.7, 118.0, 121.0, 126.9, 127.2, 129.1, 132.7, 140.3, 143.1, 158.5, 159.5, 166.6, 173.0. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₃₀H₄₂N₇O₃⁺: 548.3344, found: 548.3349. Anal. calcd. for C₃₀H₄₁N₇O₃: C 65.79, H 7.55, N 17.90, found: C 65.50, H 7.43, N 17.84. C₃₀H₄₁N₇O₃ (547.70).

General procedure for the synthesis of the aminopotentidine derivatives 4.42-4.44

4.25, **4.26** or **4.27** (1 eq) and DIPEA (3 eq) were dissolved in CH_2CI_2 (15-35 mL). O-(Benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium-tetrafluoroborate (TBTU, 1.2 eq) was added and the mixture was stirred for 10-15 min. Subsequently, **4.5** (1 eq) was added and the reaction mixture was stirred over night at room temperature. The organic layer was washed three times with H_2O (30 mL), two times with aqueous NaOH solution (5%, w/w, 15-30 mL) and 1-2 times with brine (15-30 mL). The organic layer was dried over Na_2SO_4 and the solvent was removed in vacuo. The resulting crude product was purified by preparative HPLC.

N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-(propionamidomethyl)-3-(trifluoromethyl)benzamide (4.42)

4.42 was prepared from **4.25** (80 mg, 0.29 mmol, 1 eq), DIPEA (109 mg, 0.84 mmol, 3 eq), TBTU (105 mg, 0.34 mmol, 1.2 eq) and **4.5** (104 mg, 0.29 mmol, 1 eq) according to general procedure. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 25:75-65:35, $t_R = 24.2$ min) afforded the product as white solid (10 mg, 6%). Mp: 94-96 °C. $R_f = 0.3$ (CH₂Cl₂/MeOH 90:10). RP-HPLC (gradient 2, 220 nm): 95.0% ($t_R = 17.77$ min, k = 5.1). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.03 (t, 3H, *J* 7.45 Hz), 1.37 (br s, 2H), 1.48 (br s, 4H), 1.88-1.92 (m, 2H), 2.20 (q, 2H, *J* 7.59 Hz), 2.33 (br s, 3H), 2.88 (br s, 1H), 3.24-3.44 (8H, interfering with the water signal), 3.95 (t, 2H, *J* 6.20 Hz), 4.45 (d, 2H, *J* 5.72 Hz), 6.80-6.85 (m, 3H), 7.09-7.10 (m, 2H), 7.18-7.21 (m, 1H), 7.55 (d, 1H, *J* 8.17 Hz), 8.07 (d, 1H, *J* 8.19 Hz), 8.14 (s, 1H), 8.40 (t, 1H; *J* 5.87 Hz), 8.80 (t, 1H, *J* 5.59 Hz). ¹³C-NMR (150

MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 9.8, 23.8, 25.4, 28.3, 28.5, 38.4, 38.5, 38.9, 40.6, 53.7, 62.6, 65.0, 112.9, 114.8, 118.0, 121.0, 124.1 (q, 1C, J_{C-F} 274.40 Hz), 124.6 (q, 1C, J_{C-F} 5.95 Hz), 125.0, 126.1 (q, 1C, J_{C-F} 30.83 Hz), 128.7, 129.1, 131.2, 133.2, 141.0, 158.5, 159.4, 165.1, 173.3. HRMS: (ESI): m/z [M+H]⁺, calcd. for $C_{31}H_{41}F_3N_7O_3^{+1}$: 616.3217, found: 616.3223. $C_{31}H_{40}F_3N_7O_3$ (615.70).

3-Chloro-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-(propionamidomethyl)benzamide (4.43)

4.43 was prepared from **4.26** (120 mg, 0.50 mmol, 1 eq), DIPEA (193 mg, 1.49 mmol, 3 eq), TBTU (191 mg, 0.60 mmol, 1.2 eq) and **4.5** (178 mg, 0.50 mmol, 1 eq) according to general procedure. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 25:75-70:30, $t_R = 20.9$ min) afforded the product as white solid (138 mg, 48%). $R_f = 0.7$ (CH₂Cl₂/2 M NH₃ in MeOH 90:10). RP-HPLC (gradient 2, 220 nm): 99.3% ($t_R = 16.61$ min, k = 4.7). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.02 (t, 3H, *J* 7.61 Hz), 1.36 (br s, 2H), 1.48 (br s, 4H), 1.91 (qui, 2H, *J* 6.55 Hz), 2.23 (q, 2H, *J* 7.56 Hz), 2.32 (br s, 4H), 3.24-3.39 (m, 8H, interfering with the water signal), 3.95 (t, 2H, *J* 6.12 Hz), 4.33 (d, 2H, *J* 5.82 Hz), 6.78-6.80 (m, 1H), 6.84-6.85 (m, 2H), 7.07-7.11 (m, 2H), 7.20 (t, 1H, *J* 7.75 Hz), 7.37 (d, 1H, *J* 7.99 Hz), 7.75-7.76 (m, 1H), 7.87 (m, 1H), 8.33 (t, 1H; *J* 5.94 Hz), 8.65 (t, 1H, *J* 5.53 Hz). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 9.9, 23.8, 25.4, 28.4, 28.6, 30.7, 38.4, 39.9 (1C under solvent peak (38.7-40.3)), 40.6, 53.7, 62.6, 65.0, 112.8, 114.8, 118.0, 121.1, 126.0, 127.7, 128.5, 129.1, 131.9, 134.6, 139.7, 140.2, 158.5, 159.4, 165.1, 173.2. HRMS: (ESI): $m/z [M+H]^+$, calcd. for C₃₀H₄₁ClN₇O₃⁺: 582.2954, found: 582.2960. C₃₀H₄₀ClN₇O₃ (582.15).

3-Bromo-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-(propionamidomethyl)benzamide (4.44)

4.44 was prepared from **4.27** (62 mg, 0.22 mmol, 1 eq), DIPEA (84 mg, 0.65 mmol, 3 eq), TBTU (84 mg, 0.26 mmol, 1.2 eq) and **4.5** (78 mg, 0.22 mmol, 1 eq) according to general procedure. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 35:65-70:30, $t_R = 16.2$ min) afforded the product as white solid (60 mg, 44%). Mp: 165 °C decomposition. $R_f = 0.7$ (CH₂Cl₂/3.5 M NH₃ in MeOH 90:10). RP-HPLC (gradient 2, 220 nm): 99.5% ($t_R = 16.51$ min, k = 4.7). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.03 (t, 3H, J 7.63 Hz), 1.35 (br s, 2H), 1.44-1.48 (m, 4H), 1.90 (qui, 2H, J 6.48 Hz), 2.20 (q, 2H, J 7.63 Hz), 2.27 (br s, 4H), 3.24-3.29 (m, 4H), 3.33-3.36 (m, 4H, interfering with the water signal), 3.95 (t, 2H, J 6.11 Hz), 4.30 (d, 2H, J 5.82 Hz), 6.76-6.78 (m, 1H), 6.82-6.83 (m, 2H), 7.07-7.11 (m, 2H), 7.17-7.20 (m, 1H), 7.34 (d, 1H, J 8.11 Hz), 7.79-7.81 (m, 1H), 8.04-8.05 (m, 1H), 8.34 (t, 1H; J 5.89 Hz), 8.65 (t, 1H, J 5.56 Hz). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 9.9, 24.0, 25.5, 28.4, 28.6, 38.4, 38.9, 40.6, 42.4, 53.9, 62.8, 65.0, 112.7, 114.6, 118.0, 120.9, 122.0, 126.5, 128.3, 129.0, 130.9, 134.7, 140.3, 141.2, 158.4, 159.4, 165.0, 173.2. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₃₀H₄₁BrN₇O₃⁺: 626.2449, found: 626.2454. C₃₀H₄₀BrN₇O₃ (626.60).

4-(Azidomethyl)-3-bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.45)

4.20 (400 mg, 1.56 mmol, 1.4 eq) and DIPEA (433 mg, 3.35 mmol, 3 eq) were dissolved in CH₂Cl₂ (100 mL). *O*-(Benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium-hexafluorophosphate (HBTU, 508 mg, 1.56 mmol, 1.4 eq) was added and the mixture was stirred for 30 min. Subsequently, **4.5** (400 mg, 1.12 mmol, 1 eq) was added and the reaction mixture was stirred over night at room temperature. The organic layer was washed two times with saturated aqueous NaHCO₃ solution (100 mL) and with brine (100 mL). Purification by column chromatography (CH₂Cl₂/MeOH 95:5-90:10) and removal of the solvent *in vacuo* afforded the product as yellow oil (290 mg, 44%). *R*_f = 0.2 (CH₂Cl₂/2 M NH₃ in MeOH 90:10). ¹H-NMR (400 MHz, CD₃OD): *δ* (ppm) 1.57-1.59 (m, 2H),1.72-1.78 (m, 4H), 2.01-2.07 (m, 2H), 2.94 (br s, 4H), 3.40-3.45 (m, 4H), 3.49-3.52 (m, 2H), 3.99 (s, 2H), 4.07 (t, 2H, J 5.78 Hz), 4.54 (s, 2H), 6.96-7.00 (m, 2H), 7.09-7.10 (m, 1H), 7.30 (t, 1H, J 7.94 Hz), 7.51 (d, 1H, J 7.94 Hz), 7.80-7.82 (m, 1H), 8.07-8.08 (m, 1H). ¹³C-NMR (100 MHz, CD₃OD): *δ* (ppm) 22.1, 23.4, 28.4, 39.2, 39.3, 40.9, 53.0, 53.7, 61.2, 65.7, 114.9, 116.7, 122.8, 123.3, 126.4, 128.3, 129.7, 130.0, 131.6, 132.8, 135.6, 138.7, 159.1, 160.1, 167.2. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₇H₃₅BrN₉O₂⁺: 596.2092, found: 596.2114. C₂₇H₃₄BrN₉O₂ (596.53).

4-(Aminomethyl)-3-bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.46)

4.45 (390 mg, 0.65 mmol, 1 eq) was dissolved in a mixture of THF and H_2O (1/2, v/v, 7.5 mL). Triphenylphosphine (427 mg, 1.63 mmol, 2.5 eq) dissolved in mixture of THF and H_2O (5/2, v/v, 7 mL) was added dropwise and the reaction mixture was stirred over night at room temperature. The solvent was partially removed under reduced pressure and the residue was purified by column chromatography (CH₂Cl₂/MeOH 90:10 - CH₂Cl₂/0.1% NH₃ in MeOH 90:10). The product was obtained as yellow oil (200 mg, 54%). Further purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/1% aq. NH₃ 10:90-91:9, t_{R} = 19.9 min) afforded the product as white fluffy solid (70 mg, 19%). Mp: 73-77 °C. R_f = 0.4 (CH₂Cl₂/0.2% NH₃ in MeOH 95:5). RP-HPLC (gradient 3, 220 nm): 98.4% ($t_{\rm R}$ = 28.19 min, k = 16.3). ¹H-NMR (600 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 1.46 (br s, 2H), 1.57-1.61 (m, 4H), 2.02-2.06 (m, 2H), 2.44 (br s, 4H), 3.41-3.43 (m, 4H), 3.48-3.50 (m, 4H), 3.90 (s, 2H), 4.05 (t, 2H, J 5.82 Hz), 6.85-6.89 (m, 2H), 6.94 (s, 1H), 7.21 (t, 1H, J 7.92 Hz), 7.53 (d, 1H, J 7.99 Hz), 7.78-7.80 (m, 1H), 8.04 (m, 1H). ¹³C-NMR (150 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 25.0, 26.3, 30.0, 40.4, 40.7, 42.3, 46.7, 55.3, 64.6, 66.6, 114.7, 117.0, 119.8, 123.5, 124.3, 127.7, 130.1, 130.3, 132.7, 135.8, 139.5, 146.1, 160.3, 161.5, 168.9. HRMS: (ESI): m/z $[M+H]^+$, calcd. for $C_{27}H_{37}BrN_7O_2^+$: 570.2187, found: 570.2183. $C_{27}H_{36}BrN_7O_2$ (570.54).

4-(Acetamidomethyl)-3-bromo-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.47)

4.46 (70 mg, 0.12 mmol, 1 eq) and DIPEA (32 mg, 0.25 mmol, 2 eq) were dissolved in CH_2CI_2 (5 mL). Acetyl chloride (14 mg, 0.18 mmol, 1.5 eq) dissolved in CH_2CI_2 (5 mL) was added drop wise

over a period of 2 h and the reaction mixture was stirred over night at room temperature. The organic layer was further diluted with CH₂Cl₂ (10 mL) and washed two times with aqueous NaOH solution (5%, w/w, 20 mL) and with brine (20 mL). Removal of the solvent in vacuo and purification by preparative HPLC (column: YMC Triart C_{18} , gradient: 0-30 min: MeCN/0.1% aq. NH₃ 20:80-91:9, $t_{\rm R}$ = 16.05 min) afforded the product as white fluffy solid (30 mg, 40%). The purity determined by analytical HPLC was under 95% and therefore the product was again purified by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 10:90-70:30, $t_{\rm R}$ = 23.7 min). The product was obtained as white fluffy solid (15 mg, 20%). Mp: 89-95 °C. $R_{\rm f}$ = 0.3 (CH₂Cl₂/0.2% NH₃ in MeOH 90:10). RP-HPLC (gradient 3, 220 nm): 99.0% (*t*_R = 28.01 min, *k* = 16.2). ¹H-NMR (600 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 1.45 (br s, 2H), 1.56-1.60 (m, 4H), 2.01-2.05 (m, 5H), 2.43 (br s, 4H), 3.40-3.42 (m, 4H), 3.47-3.49 (m, 4H), 4.04 (t, 2H, J 5.81 Hz), 4.44 (s, 2H), 6.84-6.88 (m, 2H), 6.93 (m, 1H), 7.20 (t, 1H, J 7.90 Hz), 7.39 (d, 1H, J 8.04 Hz), 7.74-7.76 (m, 1H), 8.03 (m, 1H). ¹³C-NMR (150 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 22.5, 25.0, 26.3, 30.0, 40.3, 40.7, 42.3, 44.5, 55.3, 64.6, 66.6, 114.8, 117.0, 119.8, 123.5, 124.1, 127.5, 130.1, 130.3, 132.8, 136.1, 142.4 (2C), 160.3, 161.5, 168.8, 173.4. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂₉H₃₉BrN₇O₃⁺: 612.2292, found: 612.2298. C₂₉H₃₈BrN₇O₃ (612.57).

4-(Acetamidomethyl)-N-(2-(2-cyano-3-(3-(3-(piperidin-1ylmethyl)phenoxy)propyl)guanidino)ethyl)benzamide (4.48)

4.40 (70 mg, 0.14 mmol, 1 eq), 4-(dimethylamino)-pyridine (17 mg, 0.14 mmol, 1 eq) and DIPEA (28 mg, 0.21 mmol, 1.5 eq) were dissolved in CH₂Cl₂ (5 mL). Acetyl chloride (12 mg, 0.16 mmol, 1.1 eq) dissolved in CH₂Cl₂ (2 mL) was added drop wise and the reaction mixture was stirred over night at room temperature. Additional acetyl chloride (12 mg, 0.16 mmol, 1.1 eq) and DIPEA (28 mg, 0.21 mmol, 1.5 eq) were added and mixture was again stirred over night at room temperature. The organic layer was further diluted with CH_2CI_2 (40 mL) and washed with H_2O (50 mL), aqueous NaOH solution (1 M, 50 mL) and brine (50 mL). The solvent was removed under reduced pressure and the residue was purified by column chromatography (CH₂Cl₂/MeOH 100:0-90:10). The product was obtained as colourless oil (40 mg, 53%). Further purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 20:80-91:9, t_R = 14.02 min) afforded the product as white fluffy solid (19 mg, 25%). Mp: 75-85 °C. R_f = 0.2 $(CH_2Cl_2/1.5 \text{ N NH}_3 \text{ in MeOH 90:10})$. RP-HPLC (gradient 3, 220 nm): 99.8% (t_R = 25.91 min, k = 14.9). ¹H-NMR (600 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 1.45 (br s, 2H), 1.56-1.60 (m, 4H), 2.00 (s, 3H), 2.02-2.06 (qui, 2H, J 6.22 Hz), 2.41 (br s, 4H), 3.41-3.43 (m, 4H), 3.46 (s, 2H), 3.49-3.51 (m, 2H), 4.05 (t, 2H, J 5.87 Hz), 4.40 (s, 2H), 6.84-6.89 (m, 2H), 6.94 (m, 1H), 7.21 (t, 1H, J 7.87 Hz), 7.35 (d, 2H, J 8.40 Hz), 7.77 (m, 2H). ¹³C-NMR (150 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 22.5, 25.1, 26.4, 30.0, 40.3, 40.7, 42.4, 43.8, 55.3, 64.7, 66.6, 114.7, 116.9, 119.9, 123.4, 128.58, 128.60, 130.2, 134.2, 139.8, 144.1, 160.3, 161.6, 170.5, 173.2. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for $C_{29}H_{40}N_7O_3^+$: 534.3187, found: 534.3191. $C_{29}H_{39}N_7O_3$ (533.68).

N-(2-(2-Cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-((dimethylamino)methyl)benzamide (4.49)

4.28 (33 mg, 0.15 mmol, 1.1 eq) and DIPEA (90 mg, 0.70 mmol, 5 eq) were dissolved in CH₂Cl₂ (10 mL). O-(Benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium-hexafluorophosphate (HBTU, 74 mg, 0.20 mmol, 1.4 eq) was added and the mixture was stirred for 30 min. Subsequently, 4.5 (50 mg, 0.14 mmol, 1 eq) was added and the reaction mixture was stirred over night at room temperature. The organic layer was diluted with CH₂Cl₂ (40 mL) and washed with H₂O (50 mL), aqueous NaOH solution (1 M, 50 mL) and brine (50 mL). Purification by automated flash chromatography (CH₂Cl₂/MeOH 100:0-80:20 in 20 min) and removal of the solvent in vacuo afforded the product as colourless oil (40 mg, 56%). Further purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 10:90-91:9, $t_{\rm R}$ = 21.50 min) afforded the product as white fluffy solid (20 mg, 28%). Mp: 62-68 °C. $R_{\rm f}$ = 0.2 (CH₂Cl₂/2 M NH₃ in MeOH 95:5). RP-HPLC (gradient 3, 220 nm): 99.0% ($t_{\rm R}$ = 30.29 min, k = 17.6). ¹H-NMR (600 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 1.45 (br s, 2H), 1.57-1.61 (m, 4H), 2.02-2.06 (qui, 2H, J 6.20 Hz), 2.24 (s, 6H), 2.43 (br s, 4H), 3.41-3.43 (m, 4H), 3.48-3.51 (m, 6H), 4.05 (t, 2H, J 5.87 Hz), 6.85-6.89 (m, 2H), 6.94-6.95 (m, 1H), 7.21 (t, 1H, J 7.85 Hz), 7.40 (d, 2H, J 8.22 Hz), 7.78-7.79 (m, 2H). ¹³C-NMR (150 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 25.1, 26.4, 30.0, 40.3, 40.6, 42.4, 45.3, 55.3, 64.5, 64.6, 66.6, 114.7, 117.0, 119.9, 123.4, 128.4, 130.3, 130.7, 134.5, 139.6, 142.9, 160.3, 161.6, 170.5. HRMS: (ESI): *m/z* [*M*+H]⁺, calcd. for C₂₉H₄₂N₇O₂⁺: 520.3395, found: 520.3394. $C_{29}H_{41}N_7O_2$ (519.69).

3-Bromo-*N*-(2-(2-cyano-3-(3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)guanidino)ethyl)-4-((dimethylamino)methyl)benzamide (4.50)

4.29 (60 mg, 0.20 mmol, 1.1 eq) and DIPEA (120 mg, 0.93 mmol, 5 eq) were dissolved in CH₂Cl₂ (10 mL). O-(Benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium-hexafluorophosphate (98 mg, 0.26 mmol, 1.4 eq) was added and the mixture was stirred for 30 min. Subsequently, 4.5 (66 mg, 0.19 mmol, 1 eq) was added and the reaction mixture was stirred over night at room temperature. The organic layer was diluted with CH_2CI_2 (40 mL) and washed with H_2O (50 mL), aqueous NaOH solution (1 M, 50 mL) and brine (50 mL). Purification by column chromatography (CH₂Cl₂ 100% - $CH_2CI_2/1$ M NH₃ in MeOH 90:10) and removal of the solvent *in vacuo* afforded the product as colourless oil (50 mg, 45%). Further purification by preparative HPLC (column: YMC Triart C₁₈, gradient: 0-30 min: MeCN/0.1% aq. NH₃ 10:90-91:9, $t_{\rm R}$ = 22.38 min) afforded the product as white fluffy solid (36 mg, 32%). Mp: 105-115 °C. R_f = 0.3 (CH₂Cl₂/1.5 M NH₃ in MeOH 90:10). RP-HPLC (gradient 3, 220 nm): 99.6% ($t_{\rm R}$ = 33.11 min, k = 19.3). ¹H-NMR (600 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 2.06 (br s, 2H), 1.56-1.60 (m, 4H), 2.02-2.06 (qui, 2H, J 6.21 Hz), 2.29 (s, 6H), 2.41 (br s, 4H), 3.41-3.45 (m, 6H), 3.49-3.51 (m, 2H), 3.61 (s, 2H), 4.05 (t, 2H, J 5.87 Hz), 6.84-6.89 (m, 2H), 6.93 (m, 1H), 7.20 (t, 1H, J 7.85 Hz), 7.53 (d, 1H, J 8.04 Hz), 7.76-7.78 (m, 1H), 8.05 (d, 1H, J 1.80 Hz). ¹³C-NMR (150 MHz, CD₃OD, COSY, HSQC, HMBC): δ (ppm) 25.1, 26.4, 30.0, 40.3, 40.7, 42.3, 45.6, 55.3, 63.6, 64.7, 66.6, 114.7, 116.9, 119.8, 123.4, 125.9, 127.3, 130.2, 132.4, 132.9, 136.1, 139.8, 142.4, 160.2, 161.5, 168.9. HRMS: (ESI): m/z [M+H]⁺, calcd. for C₂₉H₄₁BrN₇O₂⁺: 598.2500, found: 598.2502. C₂₉H₄₀BrN₇O₂ (598.59).

4.3.3 Pharmacological Methods

Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein or coexpressing the $hH_3R + G_{i\alpha 2} + \beta_1\gamma_2$ proteins was described elsewhere.¹⁹

Radioligand competition binding assays were performed as described previously with minor adjustments using the following radioligands: $[{}^{3}H]UR-DE257^{6}$ (hH₂R: specific activity = 32.89 Ci/mmol, K_{d} = 12.2 nM, c_{final} = 20 nM) or $[{}^{3}H]N^{\alpha}$ -methylhistamine (Hartmann Analytic, Braunschweig, Germany; hH₃R: specific activity = 80 Ci/mmol, K_{d} = 3 nM, c_{final} = 3 nM).

On the day of the experiment Sf9 membranes were thawed and sedimented by centrifugation at 13,000 rpm at 4 °C for 10 min. The membranes were resuspended in ice cold binding buffer (75 mM Tris/HCl, pH 7.4 containing 12.5 mM MgCl₂, 1mM EDTA and; in the following referred to as BB) and adjusted to a protein concentration of 2-4 μ g/ μ L. 80 μ L BB containing 0.2% BSA and the respective radioligand, followed by 10 μ L of the investigated ligands at various concentrations (dissolved in H₂O or 10 mM HCl, prepared less than 10 min prior), were added to every well of a 96-well plate (Primaria clear flat bottom microplates, Corning, New York, USA). Incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with 0.3% polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with 200 µL scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h. The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

Functional GTPyS assay on Sf9 insect cell membranes

GTP γ S assays were performed as described previously²⁰ with minor modifications. [³⁵S]GTP γ S (specific activity = 1000 Ci/mmol) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to 0.5-1.5 µg/µL.

Agonist mode: 80 μ L of BB containing BSA (0.05% final), GDP (1 μ M final) and [³⁵S]GTP γ S (20 nCi final), followed by 10 μ L of the investigated ligands at various concentrations (dissolved in H₂O) were added to every well of a 96-well plate (Primaria clear flat bottom microplates, Corning, New York, USA). Incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTP γ S assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine (1 μ M final).

4.3.4 Data analysis

Retention factors k were calculated according to $k = (t_R - t_0)/t_0$ ($t_0 = dead$ time; $t_R = retention$ time). Corrected counts per minute (ccpm) from the GTPyS assay (agonist mode) were plotted against the log(concentration of the test compound), and data were analyzed by a four parameter logistic equation (GraphPad Prism Software 5.0, GraphPad Software, San Diego, CA), followed by normalization (0% = water value (basal activity), 100% = "top" histamine equation) and analysis by four-parameter logistic equation (log(agonist) vs. response – variable slope, GraphPad Prism). Data of the GTPyS assay (antagonist mode) were analysed by a four parameter logistic equation (GraphPad Prism), followed by normalization (100% = "top" of the fourparameter logistic fit, 0% = unspecifically bound radioligand (ccpm) determined in the presence of famotidine at 100 μ M) and analysis by four-parameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism). p/C_{50} values were converted into pK_{B} values according to the Cheng-Prusoff equation²¹. Total binding data from radioligand competition binding experiments were plotted against log(concentration competitor) and analyzed by a fourparameter logistic equation (log(inhibitor) vs response - variable slope, GraphPad Prism), followed by normalization (100% = "top" of the four-parameter logistic fit, 0% = unspecifically bound radioligand ligand determined in the presence of famotidine at 100 μ M). Normalized data from competition binding experiments were again analyzed by a four-parameter logistic equation and obtained p IC_{50} values were converted into p K_i values according to the Cheng-Prusoff equation²¹.

4.4 References

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Chapter 5

Fluorescence Labeled H₂R Ligands with BMY25368 Core Structure: Synthesis, Characterization and Application in Flow Cytometry, Confocal Microscopy and High Content Imaging

Note: Prior to the submission of this thesis, parts of this chapter (the synthesis of **5.2**, **5.7** and **5.10**) were published in cooperation with partners:

Baumeister, P.; Erdmann, D.; <u>Biselli, S.</u>; Kagermeier, N.; Elz, S.; Bernhardt, G.; Buschauer, A. [³H]UR-DE257: Development of a Tritium-Labeled Squaramide-Type Selective Histamine H₂ Receptor Antagonist. *ChemMedChem* **2015**, 10, 83-93.

The synthesis of **5.3**, **5.12**, **5.13** and Py-5 were performed by Mengya Chen during her Master Thesis 2015.

5.1 INTRODUCTION

Fluorescence labeled GPCR ligands have become an attractive alternative to radiotracers for the investigation of ligand-receptor interactions. Besides advantages with respect to safety issues and waste disposal, fluorescent ligands are a prerequisite for the application of a plethora of optical techniques (confocal microscopy, FRET,¹ FRAP,² TIRF,³ high content imaging,⁴ fluorescence polarization⁵). In general, a fluorescent ligand consists of a pharmacophore, a linker and the fluorophore. A major challenge in the development of fluorescent ligands for aminergic GPCRs is to retain affinity, when a bulky fluorophore is attached to a relatively small ligand. It is important to consider that the attachment site, the type and length of the linker as well as the nature of the fluorophore (size, net charge and lipophilicity) might affect receptor affinity as well as selectivity and can lead to unfavorable physicochemical properties (high unspecific binding, aggregation, internalization, etc.).⁶ Nonetheless, various fluorescent ligands for aminergic GPCRs have been reported, for example for muscarinic, ⁷⁻⁹ α and β adrenergic, ¹⁰⁻¹² dopamine, ¹³ histamine H₁^{14,15} and H_3^{16-18} receptors. In the H_2R field, most of the reported fluorescent ligands are emitting at wavelengths below 550 nm (common fluorophores: fluorescein, acridine, 5-(dimethylamino)naphthalin-1-sulfonic acid amide (dansyl), N-methylanthranilic acid amide and 1cyanoisoindol-2-yl).^{19,20} The majority of the fluorescent ligands consist of piperidinomethylphenoxypropylamino (potentidine) pharmacophore, derived either from roxatidine or iodoaminopotentidine, which is linked to the fluorophore by an alkyl chain (Figure 5.1).

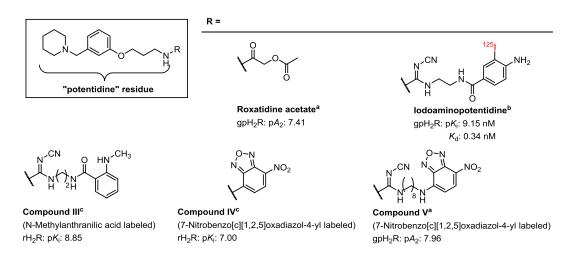


Figure 5.1. Structures of the standard H_2R antagonists roxatidine acetate and iodoaminopotentidine²¹ as well as fluorescently labeled derivatives^{19,20}. pA_2 values were ^adetermined on the isolated guinea pig right atrium¹⁹ and pK_i values and the pK_d value were ^bdetermined on membranes of guinea pig striatum^{21,22} and ^cdetermined on homogenates of COS-7 cells transiently expressing the H_2R^{20} .

Labeling with relatively small chromophores such as the N-methylanthranilic acid amide or a 7-nitrobenzo[c][1,2,5]oxadiazol-4-yl moiety resulted in fluorescent ligands (Compound III-V) with high H₂R affinity (pA₂ or pK_i values: >7.0) (Figure 5.1).^{19,20} However, the reported ligands were inapplicable for cell-based methods like confocal microscopy and flow cytometry due to the high cellular autofluorescence at the emission wavelength which resulted in low signal-to-noise ratios. In order to expand the range of applications and avoid the high cellular autofluorescence,

fluorescent ligands labeled with red-emitting fluorophores (emission wavelength > 600 nm) are required.

Recently, with the squaramide-type radioligand $[{}^{3}H]$ UR-DE257 our group developed a highaffinity and highly subtype selective hH₂R antagonist (K_{d} value: 31 nM)²³ (Figure 5.2) consisting of the pharmacophore of BMY 2536²⁴, which is linked to the tritium labeled propionic acid amide by a hexyl linker. Replacing the radiolabeled propionic acid amide by red-emitting fluorophores was the starting point for the development of high affinity hH₂R fluorescent ligands.²⁵ The most promising results were achieved by the pyridinium (Py-5) labeled ligand UR-DE229 and the cyanine labeled ligand (S0536) UR-DE56 (Figure 5.2).^{25,26} Both ligands were antagonists in the GTPase assay with pK_{b} values of 7.06-7.66 and showed a low unspecific binding in flow cytometric binding assays and confocal microscopy.

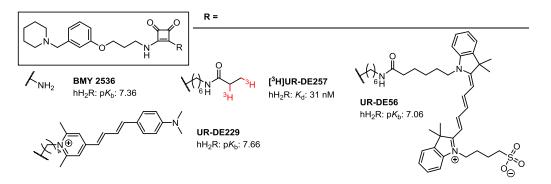


Figure 5.2. Chemical structures of the H₂R radioligand [³H]UR-DE257, the parent compound BMY 2536, as well as of the pyridinium labeled H₂R antagonist UR-DE229 and the cyanine labeled H₂R antagonist UR-DE56. Binding affinities (radioligand binding assay, K_d value)²³ and antagonism (steady-state GTPase assay, pK_b value)^{23,25} were determined on membranes of Sf9 insect cells expressing the hH₂R-G_{sas} fusion protein.

The present study is aiming at fluorescent high affinity H₂R antagonists with improved optical and physicochemical properties to gain access to a wide range of potential applications, in particular to confocal microscopy and to high throughput or/and high content imaging systems. Therefore, the fluorescent labeled antagonists UR-DE229 and UR-DE56 were investigated in different assay systems. Furthermore, a small library of fluorescent ligands were synthesized for the exploration of the impact of length of the alkyl linker and the variation of the net charge of the fluorophores by coupling the positively charged pyrilium dye (Py-5) or differently charged cyanine dyes (positive: S2197, neutral: S0536 or negative: S0586, succinimidyl esters) with various amine precursors (Figure 5.3).

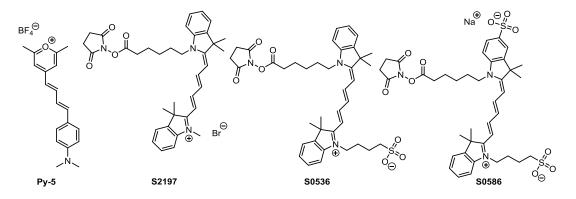
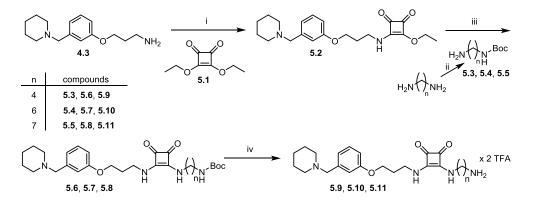


Figure 5.3. Chemical structures of the fluorescent dyes used for the preparation of the fluorescent H₂R ligands.

5.2 RESULTS AND DISCUSSION

5.2.1 Chemistry

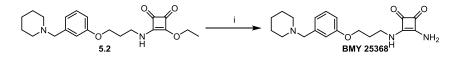
Starting from 3-[3-(piperidin-1-ylmethyl)phenoxy]propan-1-amine (4.3) the amine precursors were prepared in a three step synthesis (Scheme 5.1) as described before.²⁷ 4.3 was treated with 3,4-diethoxycyclobut-3-ene-1,2-dione (5.1) to form the corresponding squaric acid ester amide 5.2, before a second amidation with mono-boc protected alkanediamines (5.3-5.5) was performed. The intermediates 5.3-5.5 were accessible by reaction of di-*tert*-butyl dicarbonate (Boc₂O) with an excess of the alkanediamines. The resulting *tert*-butyl carbamates 5.6-5.8 were cleaved with either TFA or HCl and the products were purified by preparative HPLC to afford the amine precursors 5.9-5.11 as TFA-salts.



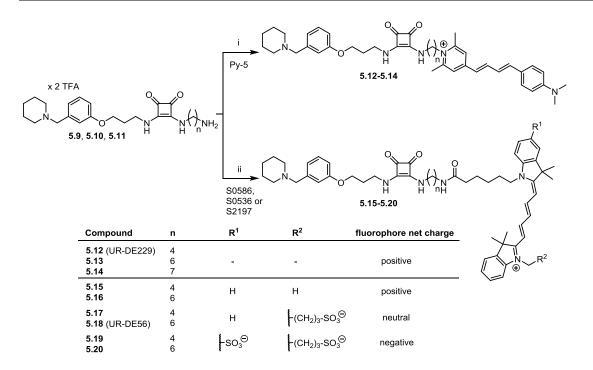
Scheme 5.1. Synthesis of the amine precursors **5.9**, **5.10**, **5.11**. Reagents and conditions: i) EtOH, RT, 2 h, 78%; ii) Boc₂O, CHCl₃, 0 °C to RT, ON, 44-92%; iii) EtOH, RT, ON, 42-83%; iv) TFA or HCl, CH₂Cl₂, RT, ON, 65-93%.

BMY 25368 was synthesized from **5.2** by treatment with ammonia in MeOH (Scheme 5.2) according to published protocols.^{24,28}

The pyridinium labeled fluorescent ligands **5.12-5.14** were synthesized by direct coupling of the pyrilium dye Py-5 (chameleon label) with the respective amine precursor **5.9-5.11** under basic conditions (Scheme 5.3).²⁹ The reaction progresses rapidly accompanied by a change in color from blue to red. The cyanine labeled fluorescent ligands **5.15-5.20** were derived from the amine precursors **5.9-5.11** by amide coupling with the succinimidyl ester of the fluorescent dyes (S2197, S0536, S0586).



Scheme 5.2. Synthesis of the parent compound BMY 25368. Reagents and conditions: i) EtOH, RT, ON, 74%.



Scheme 5.3. Synthesis of the fluorescent compounds 5.12-5.20. Reagents and conditions: i) DMF, TEA or DIPEA, RT, 90-120 min, 24-32%; ii) DMF, DIPEA, RT, 45-90 min, 18-44%.

5.2.2 Fluorescence properties of the labeled ligands

The fluorescence properties of representative compounds (**5.13**, **5.14**, **5.16**, **5.18**, **5.20**) were determined in PBS at pH 7.4 and PBS containing 1% (w/v) of BSA. The excitation and corrected emission spectra of the pyridinium labeled compound **5.14** and the cyanine labeled compounds (**5.16**, **5.18**, **5.20**) are depicted in Figure 5.4. The Stoke's shift was much more pronounced for the Py-5 labeled compounds (Figure 5.4, Table 5.1). These compounds (**5.12-5.14**) can be excited with the argon laser at 488 nm. The cyanine labeled compounds showed a considerably lower Stoke's shift, allowing excitation by the red diode laser at 635 nm.

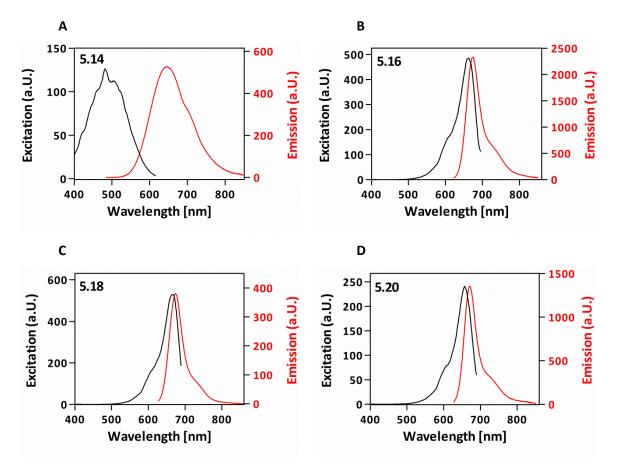


Figure 5.4. Excitation and corrected emission spectra of A: Py-5 (**5.14**), B: S0223 (**5.16**), C: S0436 (**5.18**) and D: S0387 (**5.20**) labeled compounds dissolved in PBS containing 1% BSA. Ligand concentration: 1.5-6 μ M. Excitation and emission spectra were recorded with an excitation slit of 10 nm and an emission slit of 10 nm. The excitation wavelength was chosen as close to the absorption maximum as possible (**5.14**) or at an inflection point (**5.16**, **5.18** and **5.20**). The emission wavelength was chosen close to the emission maximum (**5.14**) or at an inflection point (**5.16**, **5.18** and **5.20**).

The quantum yields (Table 5.1) were determined in PBS and in PBS containing 1% (w/v) of BSA. For the pyridinium labeled compounds **5.13** and **5.14** the quantum yields were very low in PBS alone (less than 4%), whereas the addition of BSA increased the quantum yield up to 24% (7-fold). In case of the investigated cyanine labeled compounds **5.16**, **5.18**, **5.20** the quantum yields were significantly higher in PBS (19-27%) and increased (38-45%) in the presence of BSA. This effect was described for several cyanine and pyridinium labeld ligands primarily resulting from intermolecular hydrophobic and electrostatic interactions of the fluorophores with the protein.^{30,31}

Compound	Dye ^a	PBS		PBS + 1	% BSA
		$\lambda_{ex}/\lambda_{em}$	ф (%)	$\lambda_{ex}/\lambda_{em}$	φ (%)
5.13	Py-5	457/706	3.2	481/646	22.9
5.14	Py-5	458/705	3.7	481/646	24.4
5.16	S0223	646/663	26.5	663/672	44.6
5.18	S0436	648/669	19.3	667/676	41.7
5.20	S0387	649/669	24.2	656/670	38.3

Table 5.1. Fluorescence properties of compounds **5.13**, **5.14**, **5.16**, **5.18** and **5.20** in PBS (pH 7.4) and PBS containing 1% BSA: excitation/emission maxima and fluorescent quantum yields ϕ (reference: cresyl violet perchlorate).

^aFluorescent dye used for the preparation of the fluorescent ligand.

5.2.3 Biological Evaluation

H₂R affinity, selectivity and antagonism

The fluorescent H₂R antagonists **5.12-5.20** as well as the parent compound BMY2536 and the amine precursors **5.9-5.11** were investigated in equilibrium competition binding experiments on membrane preparations from Sf9 insect cells expressing the $H_2R-G_{s\alpha S}$ fusion protein using the antagonist [³H]UR-DE257²³ as radioligand. The selectivity of the compounds for the H_2R compared to H_3R and, for selected ligands, to H_4R was investigated by competition binding experiments using Sf9 insect cell membranes co-expressing either the H_3R or the H_4R and $G_{\alpha i2}$ and $G_{\beta 1\gamma 2}$ proteins with [³H]histamine as radioligand. Representative radioligand binding curves are depicted in Figure 5.5 and the results are shown in Table 5.2.

Radioligand competition binding experiments revealed that most of the fluorescent labels were tolerated with no or a slight decrease in affinity (Table 5.2). Exceptions were the cyanine labeled ligands **5.19** and **5.20** in which the introduction of the S0387 fluorophore with a negative net charge resulted in a decrease in hH₂R affinity (**5.19**: pK_i value: 5.69, **5.20**: pK_i value: 5.88) compared to the corresponding amine-precursors (**5.9**: pK_i value: 6.52, **5.10**: pK_i value: 7.87). The pyridinium labeled ligands **5.12-5.14** showed, independent of linker length, high hH₂R affinities (pK_i value: 7.71-7.76) in the same range as the parent compound BMY 25368 (pK_i value: 7.80). In case of **5.12**, the hH₂R affinity even increased with the labeling (amine precursor **5.9**: pK_i value: 6.52 compared with **5.12**: pK_i value: 7.75). In the cyanine series, ligand **5.16**, labeled with fluorophore S0223 (positive net charge), and ligand **5.18**, labeled with fluorophore S0436 (neutral net charge), showed the highest hH₂R affinity (**5.16**: pK_i value: 7.67 and **5.18**: pK_i value: 7.11).

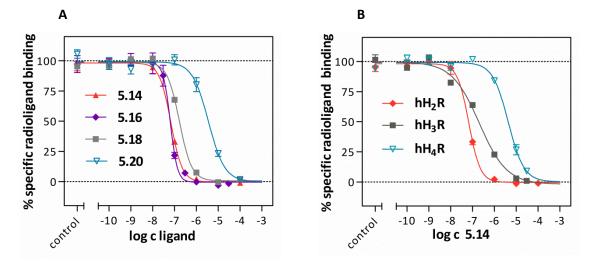


Figure 5.5. Displacement of the radioligand [³H]UR-DE257 (c = 20 nM, K_d = 12.2 nM) by the fluorescent ligands **5.14**, **5.16**, **5.18** and **5.20** determined on membrane preparations of Sf9 insect cells expressing the hH₂R-G_{sa5} fusion protein (A). Displacement of the respective radioligand from membrane preparations of Sf9 insect cells expressing the hH₂R-G_{sa5} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, K_d = 12.2 nM), co-expressing the hH₃R and G_{ai2} plus G_{β1γ2} proteins (radioligand: [³H]histamine c = 15 nM, K_d = 12.1 nM) or co-expressing the hH₄R and G_{ai2} plus G_{β1γ2} proteins (radioligand: [³H]histamine c = 10 nM, K_d = 15.9 nM) by the fluorescent ligand **5.14** (B). Data represent mean values ± SEM of 2-3 experiments performed in triplicate.

Interestingly, labeling with Py-5, S0223 and S0436 led to an increase in H_3R receptor affinity up to two orders of magnitude compared to the corresponding amine precursors (Table 5.2). The pyridinium labeled ligands **5.12-5.14** and the cyanine labeled ligands **5.16** and **5.18** showed slight selectivity for the H_2R over hH_3R (p K_i values: 6.4 - 7.18). In case of the cyanine labeled ligands **5.15** and **5.17**, the selectivity even changed in favour of the H_3R (p K_i value: 7.0 and 6.8).

Table 5.2. Affinities of the parent compound BMY2536, the amine precursors **5.9-5.11** and fluorescent ligands **5.12-5.20** to $hH_{2-4}R$, obtained from equilibrium competition binding studies on membrane preparations from Sf9 insect cells, expressing the respective histamine receptor subtype.

			hH₂R ^c		hH₃R ^d		hH₄R ^e	
Compound	Dyeª	n⁵	р <i>К</i> і	Ν	р <i>К</i> і	Ν	р <i>К</i> і	Ν
His	-	-	6.53 ± 0.04	3	7.8 ± 0.1	3	7.65 ± 0.03	3
BMY 25368	-	-	7.80 ± 0.01^{23}	3	4.66 ± 0.01	3	n.d.	-
5.9	-	4	6.52 ± 0.04	4	4.96 ± 0.07	3	n.d.	-
5.10	-	6	7.87 ± 0.02	3	5.08 ± 0.03	3	n.d.	-
5.11	-	7	7.86 ± 0.02	3	5.12 ± 0.05	3	n.d.	-
5.12	Py-5	4	7.75 ± 0.02	3	6.4 ± 0.1	3	n.d.	-
5.13	Py-5	6	7.71 ± 0.04	3	7.11 ± 0.08	4	5.64 ± 0.06	3
5.14	Py-5	7	7.763 ± 0.008	3	7.01 ± 0.04	3	5.57 ± 0.07	3
5.15	S0223	4	6.57 ± 0.02	3	7.0 ± 0.1	3	n.d.	-
5.16	S0223	6	7.67 ± 0.07	3	7.18 ± 0.03	2	n.d.	-
5.17	S0436	4	6.49 ± 0.04	3	6.8 ± 0.1	3	n.d.	-
5.18	S0436	6	7.105 ± 0.003	3	6.59 ± 0.03	2	n.d.	-
5.19	S0387	4	5.69 ± 0.08	3	5.45 ± 0.04	3	n.d.	-
5.20	S0387	6	5.88 ± 0.09	3	n.d.	-	n.d.	-

^aDye used for the preparation of the fluorescent ligand. ^bLength of the linker between the squaric acid amide and the fluorescent dye given as the number of carbon atoms. Competition binding assay on membrane preparations of Sf9 insect cells: ^cexpression of the hH₂R-G_{sαS} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, K_d = 12.2 nM), ^dco-expression of the hH₃R and G_{αi2} and G_{β1γ2} proteins (radioligand: [³H]histamine c = 15 nM, K_d = 12.1 nM) or ^eco-expression of the hH₄R and G_{αi2} plus G_{β1γ2} proteins (radioligand: [³H]histamine c = 10 nM, K_d = 15.9 nM). The incubation period was 60 min. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means ± SEM of N independent experiments, each performed in triplicate.

Compounds **5.13** and **5.14** were also investigated for hH_4R affinity. These ligands showed a high preference for the hH_2R over hH_4R . The impaired hH_2R selectivity with respect to the hH_3R limited the application of the fluorescent ligands to recombinant systems in which the hH_2R was expressed.

The amine precursors **5.9**, **5.10**, **5.11** as well as representative fluorescent ligands (**5.13**, **5.14**, **5.16**, **5.18**) were investigated for hH_2R antagonism in a GTP γ S assay on membrane preparations of Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein.³² The results are summarized in Table 5.3 and representative concentration-response curves derived from GTP γ S assays performed in antagonist mode are depicted in Figure 5.6A.

The parent compound BMY 25368, the amine precursors **5.9-5.11** as well as the investigated fluorescent ligands **5.13**, **5.14**, **5.16** and **5.18** were antagonists at hH₂R in a GTP_YS assay. In general, the pK_b values were in good agreement with the pK_i values from radioligand competition binding. Only the amine intermediates **5.9** and **5.10** as well as the cyanine labeled ligand **5.18** showed significally lower pK_b values compared to the pK_i values (difference up to one order of magnitude). Furthermore, representative fluorescent ligands (**5.13**, **5.14**, **5.16** and **5.19**) were investigated for hH₂R agonism in a β -arrestin2 recruitment assay (split luciferase complementation) on HEK293T-hH₂R- β Arr2 cells.³³ All investigated ligands were antagonists in the β -arrestin2 recruitment assay, indicating that no β -arrestin2 mediated internalization of the receptor-ligand-complex took place (Figure 5.6B).

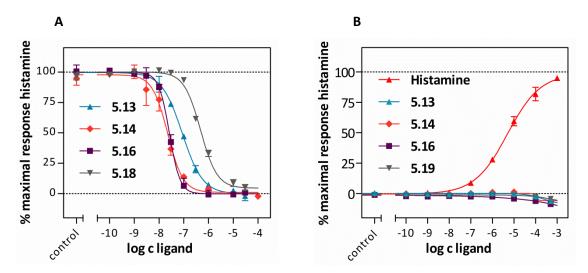


Figure 5.6. Antagonism of the fluorescent ligands **5.13**, **5.14**, **5.16** and **5.18** on H_2R determined in a GTPγS assay (antagonistic mode) on membrane preparations of Sf9 insect cells expressing the H_2R-G_{saS} fusion protein (A). In the antagonist mode histamine (1 µM) was used for stimulation. Concentration-response curves of histamine, the fluorescent ligands **5.13**, **5.14**, **5.16** and **5.19** on H_2R determined by a luciferase complementation assay measuring β-arrestin2 recruitment on HEK293T- H_2R - β Arr2 cells (B). Data represent mean values ± SEM of 2-3 experiments performed in duplicate (β-arrestin2) or triplicate (GTPγS).

			hH	l₂R (G	ΓΡγS) [°]
No.	Dye ^a	n ^b	р <i>К</i> ь (р <i>ЕС</i> ₅₀)	Ν	α
His	-	-	(5.80 ± 0.06)	9	1.0
BMY 25368	-	-	7.03 ± 0.02	2	-0.011 ± 0.007
5.9	-	4	5.8 ± 0.2	2	-0.040 ± 0.003
5.10	-	6	6.73 ± 0.08	3	-0.025 ± 0.005
5.11	-	7	7.1 ± 0.1	3	-0.011 ± 0.006
5.13	Py-5	6	7.21 ± 0.04	3	-0.016 ± 0.002
5.14	Ру-5	7	7.9 ± 0.1	3	-0.038 ± 0.006
5.16	S0223	6	7.73 ± 0.04	3	-0.026 ± 0.003
5.18	S0436	6	6.49 ± 0.03	3	-0.015 ± 0.005

Table 5.3. hH_2R antagonism and the calculated pK_b values of histamine, BMY 25368, the amine-precursors **5.9-5.11** and the fluorescent ligands **5.13**, **5.14**, **5.16** and **5.18** determined by a GTP γ S assay.

^aDye used for the preparation of the fluorescent ligand. ^bLength of the linker between the squaric acid amide and fluorescent dye given as the number of carbon atoms. ^c[³⁵S]GTPyS assay determined on membrane preparations of Sf9 insect cells expressing the hH₂R-G_{saS} fusion protein. The intrinsic activity (α) of histamine was set to 1.00, and α values of investigated compounds were referred to this value. The pK_B values of neutral antagonists were determined in the antagonist mode versus histamine (c = 1 µM) as agonist. Data represent mean values ± SEM of 2-3 experiments performed in triplicate.

Confocal microscopy

The hH₂R binding of the fluorescently labeled ligands **5.14**, **5.16** and **5.18** was also investigated by confocal microscopy (Figure 5.7). All investigated ligands were localized at the cell membrane of HEK293T-hH₂R-qs5 cells. The ligands **5.14** and **5.18** showed a low unspecific binding at a concentration of 100 nM in the presence of a 300-fold excess of famotidine, whereas **5.16** showed a higher unspecific binding.

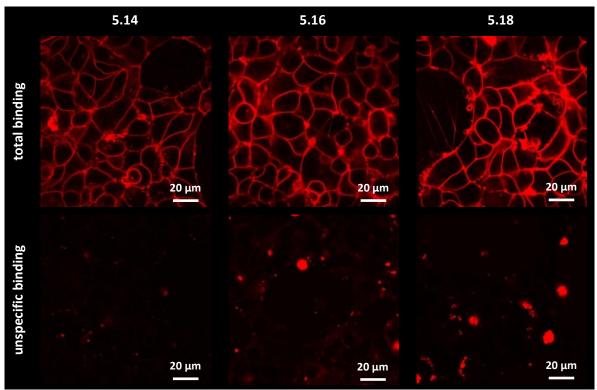


Figure 5.7. Localization of the fluorescent H₂R ligands **5.14** (100 nM), **5.16** (100 nM) and **5.18** (100 nM) at the membrane of HEK293T-hH₂R-qs5 cells determined by confocal microscopy after 20 min of incubation (25 °C). Unspecific binding was determined in the presence of famotidine (300-fold excess). All images were acquired with a Zeiss Axiovert 200 M microscope equipped with the LSM 510 Laser scanner. A 63x/1.40 oil immersion objective was used.

Flow cytometric hH₂R binding studies on HEK293T-hH₂R-qs5 cells

Fluorescent ligands **5.12-5.18**, which showed moderate to high hH₂R affinity (p K_i values > 6.0) were used for binding studies by flow cytometry on HEK293T-hH₂R-qs5 cells²⁶. Representative saturation binding curves are depicted in Figure 5.8 and the results are summarized in Table 5.4. All investigated ligands afforded K_d values which were in good agreement with the K_i values obtained from competition binding experiments with [³H]UR-DE257 on membranes of Sf9 insect cells. Within the pyridinium labeled ligands **5.12-5.14** the carbon linker length (n = 4-7) had no significant influence on affinity (K_d value: 14.9 – 27.9 nM), and unspecific binding was low (under 10% relative to total binding around the K_d).

The cyanine labeled ligand **5.16** (positive charge of the fluorophore) showed the highest affinity (K_d value: 13.9 nM) within the cyanine series, but a higher unspecific binding (20% relative to

total binding around the K_d value). The introduction of a sulfonic acid group into the cyanine moiety and the associated change of fluorophore charge to neutrality (compound **5.18**) resulted in a slight decrease in affinity (K_d value: 48.2 nM), but had a positive effect regarding unspecific binding (around 10% relative to total binding around the K_d value).

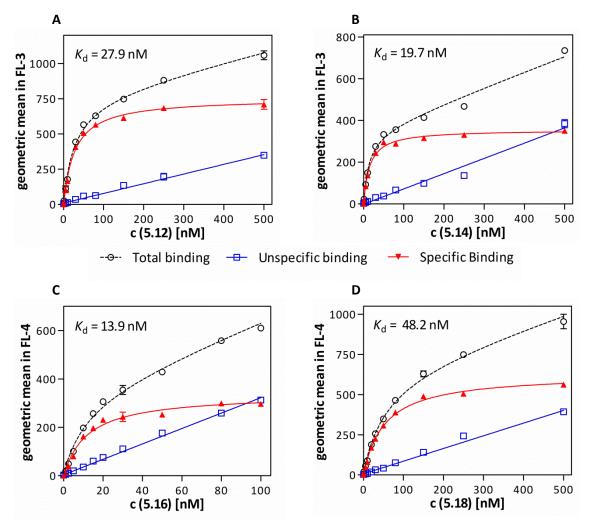


Figure 5.8. Representative flow cytometric saturation binding experiments on HEK293T-hH₂R-qs5 cells with the fluorescent ligands (A) **5.12**, (B) **5.14**, (C) **5.16** and (D) **5.18**. Unspecific binding was determined in the presence of famotidine (300-fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 90 min. Error bars of specific binding represent propagated errors calculated according to the Gaussian law. Error bars of total and unspecific binding represent the SEM. Experiments were performed in dublicate.

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Table 5.3. hH ₂ R saturation binding data of fluorescent ligands 5.12-5.18 determined by flow cytometry on HEK293-
hH ₂ R-qs5 cells in comparison with competition binding data determined on membrane preparations from Sf9 insect
cells, expressing the hH ₂ R.

Νο	Dyeª	n ^b	<i>K</i> _d (nM) ^c	<i>K</i> _i (nM) ^d
5.12	Ру-5	4	27.9 ± 2.0	18 ± 1
5.13	Ру-5	6	14.9 ± 2.3	20 ± 2
5.14	Ру-5	7	19.7 ± 4.4	23.6 ± 0.4
5.15	S0223	4	289 ± 29	270 ± 14
5.16	S0223	6	13.9 ± 2.2	22 ± 4
5.17	S0436	4	684 ± 95	328 ± 28
5.18	S0436	6	48.2 ± 2.3	78.5 ± 0.5

^aDye used for the preparation of the fluorescent ligand. ^bLength of the linker between the squaric acid amide and fluorescent dye given as the number of carbon atoms. ^cFlow cytometric saturation binding experiments on HEK293T-hH₂R-qs5 cells. Unspecific binding was determined in the presence of famotidine (300-fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 90 min. Data represent mean ± SEM from three independent experiments (each performed in duplicate). Specific binding data were analyzed by an equation describing one-site (monophasic) binding. ^dCompetition binding assay on membrane preparations of Sf9 insect cells hH₂R-G_{sαS} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, K_d = 12.2 nM), *cf.* results from Table 5.2 here shown as K_i values to facilitate comparison.

The association and dissociation kinetics of **5.14**, **5.16** and **5.18** were determined on HEK293ThH₂R-qs5 cells at 37 °C using flow cytometry (*cf.* Figure 5.9). The Py-5 labeled ligand **5.14** (c = 50 nM) showed a fast association i. e. the plateau was reached after approx. 20 min. The dissociation of **5.14** (c = 50 nM, 90 min pre-incubation) in the presence of famotidine was incomplete. After 150 min, the residual specific binding of the fluorescent ligand amounted to approximately 60% (curve plateau at 57%). The investigated cyanine dyes **5.16** and **5.18** (c = 15 nM and 25 nM) showed much slower association, i. e. the plateau was reached after 140 min. The association is strongly concentration dependent which resulted in the differences in association rate between **5.14** (c = 50 nM) and **5.16** (c = 15 nM) or **5.18** (c = 25 nM). A comparison of the association rate constants (k_{on}) which are concentration independent, revealed that Py-5 ligand **5.14** (k_{on} value: 0.0043 min⁻¹ nM⁻¹) associated two times faster than **5.16** (k_{on} value: 0.002 min⁻¹ nM⁻¹) (Table 5.5). The dissociation of the ligands **5.16** and **5.18** (90 min pre-incubation) was incomplete after 150 min, reaching a plateau at 22% (**5.16**) and 67% (**5.18**) of initially bound ligand. These data suggested a (pseudo)irreversible binding^{34,35} of **5.14**, **5.16** and **5.18**.

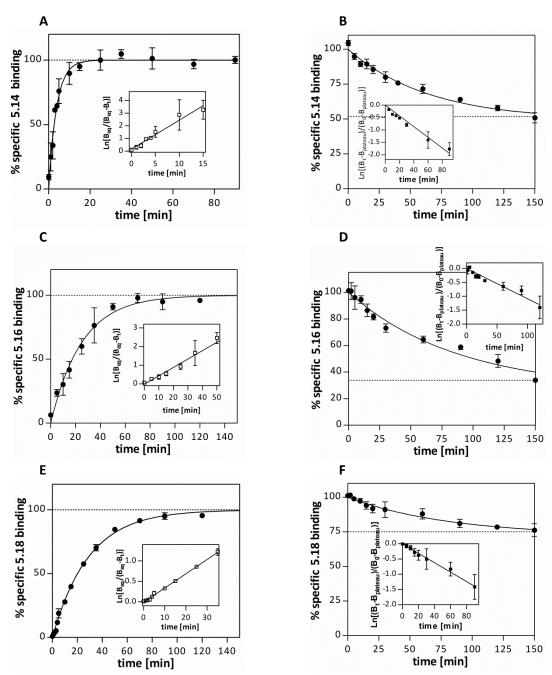


Figure 5.9. Association and dissociation kinetics of **5.14** (A, B), **5.16** (C, D) and **5.18** (E, F) determined at intact HEK293ThH₂R-qs5 cells at 37 °C by flow cytometry. (A) Association of **5.14** (c = 50 nM) to the hH₂R as a function of time (nonlinear regression: $k_{obs} = 0.24 \text{ min}^{-1}$). Inset: $\ln[B_{eq}/(B_{eq} - B_t)]$ versus time, $k_{obs} = \text{slope} = 0.23 \text{ min}^{-1}$. (B) Dissociation of **5.14** (preincubation: 90 min, c = 50 nM) in the presence of famotidine (c = 15 µM) from the hH₂R as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{off} = 0.027 \text{ min}^{-1}$, $t_{1/2} = 26 \text{ min}$, plateau = 57%), Inset: $\ln[(B_t - B_{plateau})/(B_0 - B_{plateau})]$ versus time, $\text{slope} \cdot (-1) = k_{off} = 0.023 \text{ min}^{-1}$. (C) Association of **5.16** (c = 15 nM) to the hH₂R as a function of time (nonlinear regression: $k_{obs} = 0,061 \text{ min}^{-1}$). Inset: $\ln[B_{eq}/(B_{eq} - B_t)]$ versus time, $k_{obs} = \text{slope} = 0.045 \text{ min}^{-1}$. (D) Dissociation of **5.16** (preincubation: 90 min, c = 15 nM) in the presence of famotidine (c = 4.5 µM) from the hH₂R as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{off} = 0.012 \text{ min}^{-1}$. (E) Association of **5.18** (c = 25 nM) to the hH₂R as a function of time (nonlinear regression: $k_{obs} = 0,038 \text{ min}^{-1}$). Inset: $\ln[B_{eq}/(B_{eq} - B_t)]$ versus time, $k_{obs} = \text{slope} = 0,035 \text{ min}^{-1}$. (D) Dissociation of **5.18** (preincubation: 90 min, c = 25 nM) in the presence of famotidine (c = 7.5 µM) from the hH₂R as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{obs} = \text{slope} - 0,035 \text{ min}^{-1}$. (D) Dissociation of **5.18** (preincubation: 90 min, c = 25 nM) in the presence of famotidine (c = 7.5 µM) from the hH₂R as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{off} =$ 0.015 min⁻¹, $t_{1/2} = 75 \text{ min}$, plateau = 67%), Inset: $\ln[(B_t-B_{plateau})/(B_0-B_{plateau})]$ versus time, s

Nevertheless, the equilibrium dissociation constants of **5.14** ($K_{d(kin)}$ value: 6.7 nM), **5.16** ($K_{d(kin)}$ value: 6.3 nM) and **5.18** ($K_{d(kin)}$ value: 16.7 nM), calculated from kinetics (nonlinear regression, $K_{d(kin)} = k_{off}/k_{on}$), were consistent with the K_d values obtained from saturation binding experiments (Table 5.5).

Pseudo-irreversible binding to the human, rat and mouse H_2R was also observed in case of the closely related radioligand [³H]UR-DE257 using either Sf9 insect cell membranes or HEK293T-hH₂R-CreLuc cells.²³ Squaramides, such as BMY25368 and the amine precursor **5.10**, were described as insurmountable H_2R antagonists.²³ Unlike standard the antagonist famotidine, both compounds caused a concentration-dependent depression of the maximal agonist response of the guinea pig right atrium.

Several GPCR ligands were reported to show a similar behavior in kinetic and functional experiments³⁴⁻³⁷ and several explanations were provided such as irreversible (covalent) binding to the receptor,³⁸ a slow rate of dissociation from the receptor,³⁴ a slow rate of interconversion between inactive and active receptor conformations,³⁹ stabilization of an inactive ligand-specific receptor conformation,^{40,41} binding to a site distinct from the agonist binding site⁴² or internalization of the ligand-receptor-complex³⁷.

Table 5.4. hH_2R binding characteristics of **5.14**, **5.16** and **5.18** determined by flow cytometry on HEK293T- hH_2R -qs5 cells at 37 °C.

		Binding Kinetics		Saturation Binding
	k off ^a	k on ^b	k _{off} /k _{on} ^c	K _{d (sat)} ^d
Νο	(min⁻¹)	(min ⁻¹ nM ⁻¹)	К _{d (kin.)} (nМ)	(nM)
5.14	0.027 ± 0.002	0.0043 ± 0.001	6.7 ± 1.7	19.7 ± 4.4
5.16	0.011 ± 0.003	0.0020 ± 0.0009	6.3 ± 2.7	13.9 ± 2.2
5.18	0.015 ± 0.013	0.00093 ± 0.00021	16.70 ± 3.3	48.2 ± 2.3

^aDissociation rate constant derived from nonlinear regression. ^bAssociation rate constant derived from nonlinear regression; calculated from k_{obs} , k_{off} and the fluorescent ligand concentration (cf. Figure 5.9 and experimental section). ^cKinetically determined dissociation constant. ^dEquilibrium dissociation constant determined by saturation binding. Data represent means ± SEM from two to three independent experiments (each performed in duplicate).

Regardless of the slow dissociation, the pyridinium labeled ligand **5.14** and the cyanine labeled ligand **5.18** were selected for equilibrium competition binding experiments with HEK293T-hH₂R-qs5 cells. The results are summarized in Table 5.6 and the competition binding curves are depicted in Figure 5.10. Both ligands were completely displaceable by histamine and standard H₂R antagonists like famotidine and ICI127032.

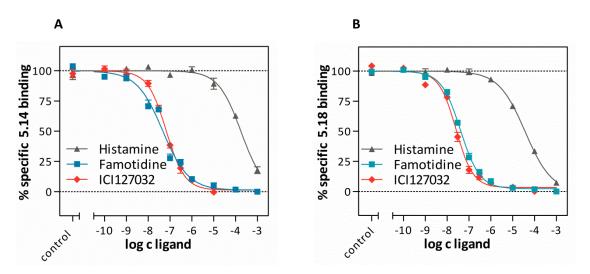


Figure 5.10. Displacement of (A) the fluorescent ligand **5.14** (c = 50 nM, K_d = 19.7nM) or [B) the fluorescent ligand **5.18** (c = 25 nM, K_d = 48.2 nM) by histamine, famotidine and ICI 127032 determined on HEK293T-hH₂R-qs5 cell by flow cytometry. Cells were incubated with the fluorescent ligands and the test compounds at RT in the dark for 90 min. Data represent means ± SEM from three independent experiments (each performed in duplicate).

The p K_i values of the standard H₂R antagonists famotidine and ICI127032 determined by flow cytometric equilibrium competition binding with the fluorescently labeled ligand 5.14 or 5.18 were in good agreement with data derived from radioligand equilibrium binding experiments with $[^{3}H]$ UR-DE257 at sf9 insect cell membranes. However, the pK_i value of the endogenous ligand histamine determined by flow cytometry with either 5.14 or 5.18 was considerably lower compared to radioligand binding at sf9 insect cell membranes. This could be explained by the different receptor-G-protein-complexes used. In the Sf9-system a hH₂R-G_{sq5} fusion protein was utilized, which was reported to have a high affinity (pK_{ih} value: 7.10) and a low affinity binding site (pk_{\parallel} value: 5.08) for histamine when [³H]tiotidine was used as a radioligand (shallow biphasic competition binding curve).⁴³ In case [³H]UR-DE257 was used as radioligand, the resulting doseresponse curve of histamine was best fitted by a four-parameter logistic fit (shallow monophasic competition binding curve). The corresponding pK_i value of 6.53 was between the ones of the high affinity and low affinity binding site, which indicates, together with the shallow curve slope, that the high affinity and low affinity binding site are also available but not resolved in this experimental set up. In the HEK293T-hH₂R-qs5 cells the hH₂R was stably co-expressed with the chimeric G_{α} protein qs5-HA. It was reported, that these cells showed no high affinity binding site for histamine in radioligand equilibrium competition binding experiments with [³H]tiotidine which resulted in monophasic competition binding curves and a pK_i value of 3.95.²⁶ This pK_i value was in good agreement with the results determined by flow cytometry using fluorescent ligand 5.14 or 5.18 as competitor and the same cell type (Table 5.5).

	Radioligand Binding	Flow Cytometry (5.14) ^c	Flow Cytometry (5.18) ^d
Νο	рК _i	рК _i	р К _і
Histamine	6.53 ± 0.04 ^a (3.93 ^b) ²⁶	4.30 ± 0.04	4.61 ± 0.05
Famotidine	7.25 ± 0.03^{a}	7.90 ± 0.02	7.58 ± 0.04
ICI127032	7.70 ± 0.07^{a}	7.73 ± 0.02	7.73 ± 0.03

Table 5.5. hH_2R binding (pK_i values) of histamine, famotidine and ICI127032 determined by radioligand competition binding ($[{}^{3}H]UR$ -DE257^a or $[{}^{3}H]$ tiotidine^b) and by displacement of fluorescent ligands **5.14**^c and **5.18**^d, respectively.

^{a,b}Determined by competition binding with ^a[³H]UR-DE257 (c = 20 nM) at membrane preparations of sf9 insect cells expressing $hH_2R-G_{S\alpha s}$ or with ^b[³H]tiotidine (c = 5 nM) at intact HEK293T-hH₂R-qs5 cells. ^{c,d}Determined by competition binding with ^c**5.14** (c = 50 nM) or ^d**5.18** (c = 25 nM) at intact HEK293T-hH₂R-qs5 cells. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means ± SEM of 3 independent experiments, each performed in ^{a,b}triplicate or ^{c,d}duplicate.

The fluorescent ligands **5.12**, **5.14-5.18** and **5.20** were also applied to high content imaging hH₂R binding assays enabling measurement on live and adherent HEK293T-hH₂R-qs5 cells in the 96-well plate format. The fluorescent ligands were incubated with the cells for 60 min at room temperature, washed with PBS, and images were obtained by an IN Cell Analyzer 2000 plate reader. Figure 5.11 shows representative images after incubation with the fluorescent ligands **5.14** (250 nM), **5.16** (75 nM), **5.18** (75 nM) and **5.20** (500 nM) for 60 min, followed by a washing step. All fluorescent ligands were localized at the cell membrane and there was a clear difference between total and unspecific binding.

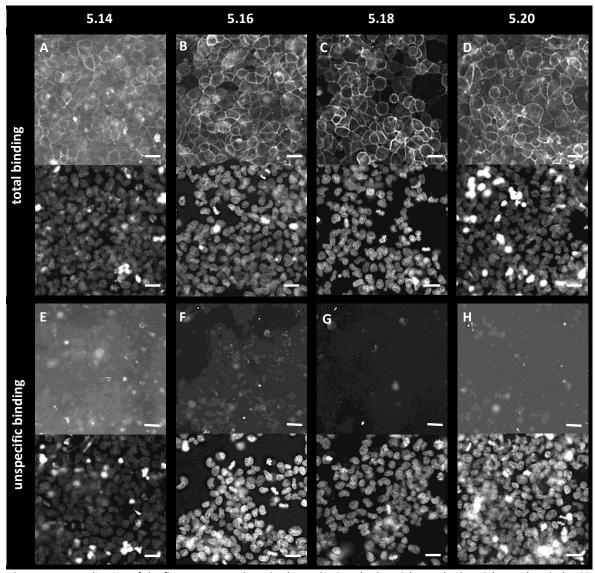


Figure 5.11. Localization of the fluorescent H_2R ligands A) **5.14** (250 nM), B) **5.16** (75 nM), C) **5.18** (75 nM) and D) **5.20** (500 nM) at the membrane of HEK293T-hH₂R-qs5 cells determined by high content imaging after 60 min of incubation (RT). Hoechst 33342 was used as a nuclear stain. Unspecific binding of **5.14** (E), **5.16** (F), **5.18** (G) and **5.20** (H) were determined in the presence of famotidine (300-fold excess). All images were acquired with a INCell Analyzer 2000. Scale bar: 20 μ m.

The acquired images were transformed into saturation binding curves (representative curves are shown in Figure 5.12). The K_d values determined by high content imaging at adherent HEK cells (Table 5.7) were generally in good agreement with the ones obtained by flow cytometry

(Table 5.4). The pyridinium labeled ligands **5.12** and **5.14** showed a slightly lower affinity for the hH_2R (K_d values: 74.6 nM and 90.9 nM) compared to flow cytometry and radioligand competition binding. In contrast the cyanine labeled ligands **5.15-5.18** and **5.20** showed a similar or slightly higher affinity for the hH_2R . Within the series the ligands **5.16** and **5.18** showed the highest affinity with K_d values of 16.2 nM and 17.9 nM.

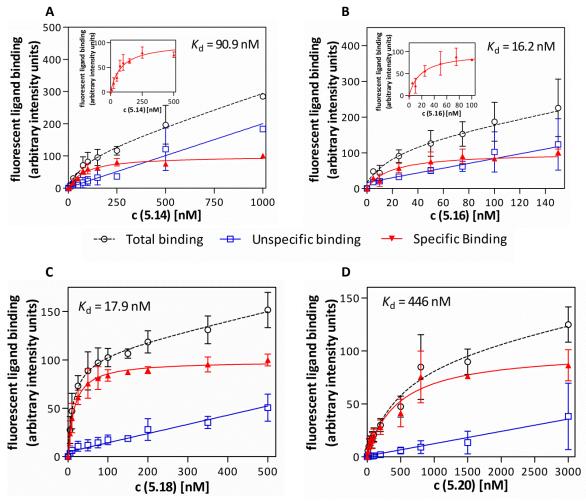


Figure 5.12. High content imaging: saturation binding experiments on adherent HEK293T-hH₂R-qs5 cells with fluorescent ligands (A) **5.14**, (B) **5.16**, (C) **5.18** and (D) **5.20**. Unspecific binding was determined in the presence of famotidine (300-fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min. Error bars represent mean \pm SEM from at least two independent experiments (each performed in duplicate).

			Adherent C	Cells
No	Dye ^a	n ^b	K _d (nM)	N
5.12	Ру-5	4	74.6 ± 0.5	2
5.14	Ру-5	7	90.9 ± 8.0	3
5.15	S0223	4	105 ± 21	5
5.16	S0223	6	16.2 ± 4.1	2
5.17	S0436	4	179 ± 44	3
5.18	S0436	6	17.9 ± 5.6	3
5.20	S0387	6	446 ± 41	2

Table 5.6. hH₂R saturation binding data of fluorescent ligands **5.12**, **5.14-5.18** and **5.20** determined by high content imaging on adherent HEK293-hH₂R-qs5 cells.

^aDye used for the preparation of the fluorescent ligand. ^bLength of the linker between the squaric acid amide and the fluorescent dye given as the number of carbon atoms. ^cSaturation binding experiments on adherent cells determined by high content imaging. Unspecific binding was determined in the presence of famotidine (300-fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min. Error bars represent mean ± SEM from N independent experiments (each performed in duplicate). Specific binding data were analyzed by an equation describing one-site (monophasic) binding.

Additionally, the association and dissociation kinetics of 5.18 were determined on adherent HEK293T-hH₂R-qs5 cells at 37 °C using high content imaging. The fluorescent ligand (50 nM) was added to the cells and the images of the same well were acquired at different time points (no washing step necessary). Figure 5.13 (A-D) shows the binding of 5.18 after increasing perids of incubation. The dissociation of 5.18 (50 nM) was determined in the presence of famotidine (300fold excess) after 60 min pre-incubation with the fluorescent ligand. Sample images at different time points are shown in Figure 5.13 (E-H). The resulting association and dissociation binding curves are depicted in Figure 5.14. Association and dissociation rate of 5.18 (k_{on} value: 0.0098 min^{-1} nM⁻¹ and k_{off} value: 0.091 min⁻¹, cf. Table 5.7) were ten times faster compared to flow cytometry. In high content imaging 5.18 (c = 50 nM) showed a fast association which was completed after approximately 10 min. The dissociation of 5.18 (c = 50 nM) in the presence of famotidine was incomplete. After 25 min, the residual specific binding of the fluorescent ligand reached a plateau at approximately 78%, suggesting (pseudo)irreversible binding, which was in good agreement with the results determined by flow cytometry. The images of the dissociation showed that the residual fluorescent ligand was preferentially located at the cell membrane and there was only low fluorescence in the cytoplasm. It is a matter of speculation why the association and dissociation kinetics were ten times faster compared to the kinetics determined by flow cytometry. Presumably, this was attributable to the different experimental conditions (adherent cells in monolayer vs cellsuspension, cell concentration: $3 \cdot 10^9$ cells/mL vs 0.5 - $1 \cdot 10^6$ cells/mL in assay, fluorescent ligand concentration: 50 nM vs 25 nM).

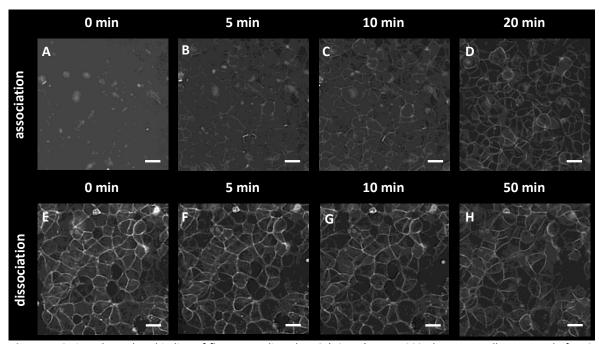


Figure 5.13.Time-dependent binding of fluorescent ligand **5.18** (50 nM) at HEK293T-hH₂R-qs5 cells measured after 0 min (A), 5 min (B), 10 min (C) and 20 min (D) and replacement of **5.18** (50 nM) with famotidine (300-fold excess) after pre-incubation for 60 min measured after 0 min (E), 5 min (F), 10 min (G) and 50 min (H). Cells were incubated with fluorescent ligand at 37 °C and images were acquired with high content imaging (INCell Analyzer 2000), scale bar: 20 μ m.

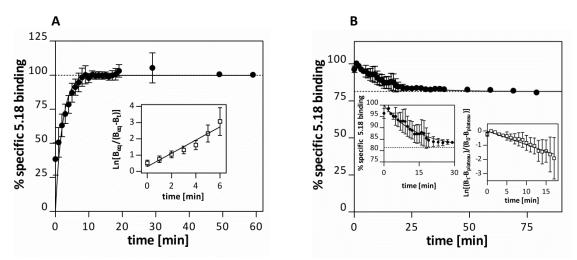


Figure 5.14. Association (A) and dissociation (B) kinetics of **5.18** at adherent HEK293T-hH₂R-qs5 cells at 37 °C determined by high content imaging. (A) Association of **5.18** (c = 50 nM) to the hH₂R as a function of time (nonlinear regression: $k_{obs} = 0.40 \text{ min}^{-1}$). Inset: $\ln[B_{eq}/(B_{eq} - B)]$ versus time, $k_{obs} = \text{slope} = 0.41 \text{ min}^{-1}$. (B) Dissociation of 18 (preincubation: 60 min, c = 50 nM) in the presence of famotidine (c = 15 µM) from the hH₂R as a function of time, showing incomplete monophasic decline (nonlinear regression: $k_{off} = 0.091 \text{ min}^{-1}$, $t_{1/2} = 13 \text{ min}$, plateau = 78%), inset left: magnification of the first 60 min, inset right: $\ln[(B_t-B_{plateau})/(B_0-B_{plateau})]$ versus time, slope-(-1) = $k_{off} = 0.099 \text{ min}^{-1}$. Data represent means ± SEM from two independent experiments (each performed in duplicate).

Nevertheless, the dissociation constant of **5.18** ($K_{d(kin)}$ value: 9.3 nM), calculated from kinetics (nonlinear regression, $K_d = k_{off}/k_{on}$), was consistent with the K_d value obtained from saturation binding experiments (Table 5.8).

Table 5.7. hH₂R binding characteristics of **5.18** determined by high content imaging on adherent HEK293T-hH₂R-qs5 cells (37 °C)

	k off ^a	k on ^b	k _{off} /k _{on} ^c	K _{d (sat)} ^d
No	(min⁻¹)	(min⁻¹ nM⁻¹)	К _{d (kin.)} (nМ)	(nM)
5.18 (IN Cell)	0.091 ± 0.084	0.0098 ± 0.0004	9.3 ± 0.4	17.9 ± 5.6

^aDissociation rate constant derived from nonlinear regression. ^bAssociation rate constant derived from nonlinear regression; calculated from k_{obs} , k_{off} and the fluorescent ligand concentration (cf. Figure 5. and experimental section). ^cKinetically determined dissociation constant. ^dEquilibrium dissociation constant determined by saturation binding. Data represent means ± SEM from two independent experiments (each performed in duplicate).

By high content imaging (96 well format) it was also possible to perform competition binding experiments with the cyanine labeled ligand **5.18** on HEK293T-hH₂R-qs5 cells. The results are summarized in Table 5.9 and the competition binding curves are depicted in Figure 5.15. Ligand **5.18** was completely displaceable by histamine and the standard H₂R antagonists lamtidine and ICI127032. The pK_i values of lamtidine and ICI127032 were in good agreement with the results from radioligand competition binding. The pK_i value of the endogenous ligand histamine determined by high content imaging with **5.18** was considerably lower compared to radioligand binding at sf9 insect cell membranes, but in good agreement with the results from flow cytometry, when using HEK293T-hH₂R-qs5 cells.

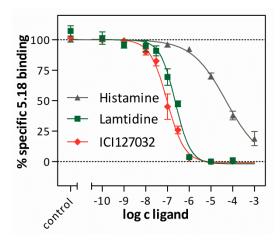


Figure 5.15. Displacement of the fluorescent ligand **5.18** (c = 50 nM, K_d = 17.9 nM) by histamine, lamtidine and ICI 127032 determined on adherent HEK293T-hH₂R-qs5 cells by high content imagig. Cells were incubated with the fluorescent ligands and the test compounds at RT in the dark for 60 min. Data represent means ± SEM from three independent experiments (each performed in duplicate or triplicate).

	Radioligand Binding ^a	Fluorescent Ligand 5.18 ^b
No	р <i>К</i> і	р <i>К</i> і
Histamine	6.53 ± 0.04	4.54 ± 0.26
Lamtidine	6.8 ± 0.2	7.23 ± 0.03
ICI127032	7.70 ± 0.07	7.59 ± 0.09

Table 5.8. hH₂R binding (pK_i values) of histamine, lamtidine and ICI127032 determined by radioligand binding ($[{}^{3}H]$ UR-DE257) and by fluorescent ligand binding (**5.18**).

^aDetermined by competition binding with [³H]UR-DE257 (c = 20 nM) at membrane preparations of sf9 insect cells expressing hH_2R-G_{Sas} . ^bDetermined by competition binding with **5.18** (c = 50 nM) at adherent HEK293T-hH₂R-qs5 cells. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data represent means ± SEM of 3 independent experiments, each performed in triplicate or duplicate.

The pyridinium labeled ligands **5.12** and **5.14** were also applied to binding studies on suspended HEK293T-hH₂R-qs5 cells using an imaging flow cytometer. This cytometer acquires multiple high-resolution images of every cell in the flow. The fluorescent ligands were incubated with the cells for 60 min at room temperature. The fluorescent ligands **5.12** and **5.14** were mainly localized at the cell membrane. Exemplary images of cells incubated with **5.12** (500 nM) are shown in Figure 5.16A. Only focused single cells were included in the data analysis and masks of the cell membrane and the whole cell were superimposed, enabling the determination of the fluorescent intensity in the region of interest.(Figure 5.16B and 5.16C).

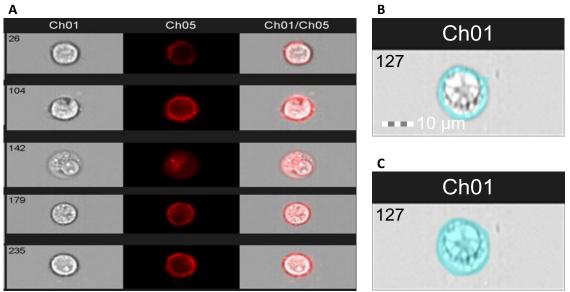
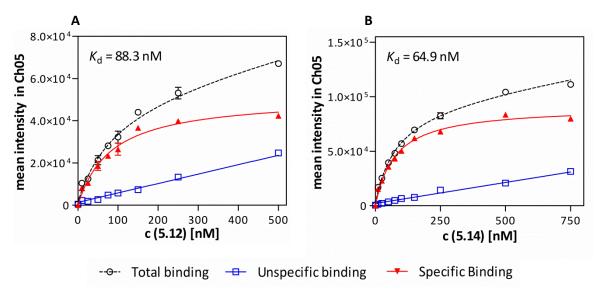


Figure 5.16. Typical images of focused single cells after 60 min of incubation with pyridinium labeled ligand **5.12** (c = 500 nM). Ch01: bright field, Ch05: emission at 702/85 nm, Ch01/Ch05: merged channels. Light blue coloring: (B) cell membrane mask and (C) cell mask.

The resulting saturation binding curves are shown in Figure 5.17 and the results are summarized in Table 5.10. The K_d values of **5.12** (88.3 nM) and **5.14** (64.9 nM) were by approx. a factor of 3 higher than the K_d values obtained by flow cytometry (Table 5.4). By contrast, the respective



dissociation constants were consistent with those (K_d values: 74.6 nM and 90.9 nM) determined by high content imaging (Table 6), irrespective of the data processing procedure.

Figure 5.17. Imaging flow cytometric saturation binding experiments on adherent HEK293T-hH₂R-qs5 cells with pyridinium labeled ligands (A) **5.12** and (B) **5.14**. For the evaluation of the cell images the cell membrane mask was used. Unspecific binding was determined in the presence of famotidine (300-fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min. Error bars represent mean \pm SEM from at least two independent experiments (each performed in duplicate).

	Imaging Flow Cytometry		
No	<i>K</i> _d (nM) whole cell mask ^a	<i>K</i> _d (nM) cell membrane mask ^b	
5.12	87.9 ± 2.0	88.3 ± 5.5	
5.14	62.1 ± 2.0	64.9 ± 4.6	

Table 5.9. hH_2R saturation binding data of fluorescent ligands **5.12** and **5.14** determined by imaging flow cytometry on HEK293T-hH₂R-qs5 cells.

For the evaluation of cell images either (A) the whole cell mask or (B) the cell membrane mask was used. Unspecific binding was determined in the presence of famotidine (300-fold excess). Cells were incubated with the fluorescent ligands at RT in the dark for 60 min. Error bars represent mean ± SEM from three independent experiments (each performed in duplicate). Specific binding data were analyzed by an equation describing one-site (monophasic) binding.

Additionally, histamine and ICI127032 were investigated by equilibrium competition binding experiments with fluorescent ligand **5.14** on HEK293T-hH₂R-qs5 cells using the imaging flow cytometer. The resulting pK_i values were in good agreement with the results from flow cytometry and high content image analysis (Table 5.11).

cells using different systems: flow cytometry (5.14), high content imaging on adherent cells (5.18) and imaging flow
cytometry (5.14).

Table 5.10. hH₂R binding (pK_i values) of histamine and ICI127032 determined fluorimetrically on HEK293T-hH₂R-qs5

	Flow Cytometry	IN Cell Analyzer	ImageStream X
	(5.14) ^a	(5.18) ^b	(5.14) ^c
Νο	р <i>К</i> і	р <i>К</i> і	р <i>К</i> i
Histamine	4.30 ± 0.04	4.54 ± 0.26	4.73 ± 0.08
ICI127032	7.73 ± 0.02	7.59 ± 0.09	7.81 ± 0.05

^aDetermined by displacement of **5.14** (c = 50 nM) in the presence of increasing concentrations of the respective H_2R ligand at intact cells by flow cytometry. ^bDetermined by displacement of **5.18** (c = 50 nM) in the presence of increasing concentrations of the respective H_2R ligand at adherent cells. ^cDetermined by competition binding of **5.14** (c = 70 nM) in the presence of increasing concentrations of the respective H_2R ligand at adherent cells. ^cDetermined by competition binding of **5.14** (c = 70 nM) in the presence of increasing concentrations of the respective H_2R ligand at intact cells by imaging flow cytometry, for the evaluation of cell images the cell membrane mask was used. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data represent means ± SEM of 3 independent experiments, each performed in triplicate or duplicate.

5.3 EXPERIMENTAL SECTION

5.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). The active esters of the cyanine dyes (S0586, S0536, S2197) were obtained from FEW Chemicals (Bitterfeld-Wolfen, Germany). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 (0.040-0.063 mm). Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution (0.8 g ninhydrin, 200 mL n-buthanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA). Nuclear Magnetic Resonance (¹H NMR and ¹³C NMR) spectra were recorded on a Bruker Avance-300 (7.05 T, ¹H: 300 MHz, ¹³C: 75.5 MHz), Avance-400 (9.40 T, ¹H: 400 MHz, ¹³C: 100.6 MHz), or Avance-600 (14.1 T; ¹H: 600 MHz, ¹³C: 150.9 MHz; cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in δ (ppm) relative to external standards. Multiplicities are specified with the following abbreviations: s (singlet), d (doublet), t (triplet), q (quartet), qui (quintet), m (multiplet), br s (broad signal), as well as combinations thereof. In certain cases 2D-NMR techniques (COSY, HSQC and HMBC) were used to assign ¹H and ¹³C chemical shifts. Low-resolution mass spectrometry (MS) was performed on a Finnigan ThermoQuest TSQ 7000 instrument using an electrospray ionization (ESI) source or on a Finnigan SSQ 710A instrument (EI-MS, 70 eV). Highresolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass Q-TOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K-2001 detector. A Nucleodur 100-5 C18 (250 x21 mm, 5 µm, Macherey-Nagel, Dueren, Germany) and a Kinetex XB-C18 100A (250 x 21.2 mm, 5 µm, Phenomenex, Aschaffenburg, Germany) served as RP-columns at a flow rate of 15 ml/min at room temperature. Mixtures of CH₃CN and 0.1% aq. TFA were used as mobile phase and a detection wavelength of 220 nm was used throughout. CH₃CN was removed from the eluates containing BMY 25368, 5.9 and 5.10 under reduced pressure (final pressure: 80 mbar) at 45 °C prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a vacuubrand RZ 6 rotary vane vacuum pump). In case of the fluorescently labeled ligands, the eluates were directly used for lyophilisation. Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A (250 x 4.6 mm, 5 µm, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of 0.05% TFA in CH₃CN (A) and 0.05% aq. TFA (B) were use as mobile phase. Helium degassing, room temperature, a flow rate of 0.8 mL/min and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration: 100-500 μ M) were either prepared from stock solution (5-10 mM in DMSO or DMSO/20 mM HCL 50:50) in a mixture of A and B corresponding to the initial eluent composition of the run, or as a one to one dilution of the eluate with Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5.

5.3.2 Experimental protocols and analytical data

The synthesis of the compound $4.3^{23,27}$ was described in chapter 4 and the tetrafluoroborate salt of Py-5^{29,44} was synthesized by Mengya Chen during her master thesis.

3-Ethoxy-4-(3-[3-(piperidin-1-ylmethyl)phenoxy]propylamino)cyclobut-3-ene-1,2-dione (5.2)²³

3,4-Diethoxycyclobut-3-ene-1,2-dione (226 mg, 1.33 mmol, 1.1 eq) was dissolved in EtOH (10 mL) and added to a solution of compound **4.3** (300 mg, 1.21 mmol, 1 eq) in EtOH (10 mL). The reaction mixture was stirred for 2 h at room temperature. After removal of the solvent under reduced pressure, the residue was dissolved in EtOAc (30 mL). The organic layer was washed with H₂O (3 x 20 mL) and dried over Na₂SO₄. The solvent was evaporated under reduced pressure and the crude product was purified by column chromatography (eluent: CH₂Cl₂ / 7 M NH₃ in MeOH 100:0-95:5). Removal of the solvent *in vacuo* afforded the product as yellow oil (350 mg, 78%). *R*_f = 0.5 (CH₂Cl₂ / 7 N NH₃ in MeOH 95:5). ¹H-NMR (400 MHz, CDCl₃, COSY, HSQC, HMBC): δ (ppm) 1.38-1.42 (m, 5H), 1.52-1.58 (m, 4H), 2.06-2.12 (m, 2H), 2.36 (br s, 4H), 3.42 (s, 2H), 3.66 (br s, 1.5H), 3.86 (br s, 0.6H), 4.00 (br s, 2H), 4.66-4.71 (q, 2H, *J* 7.3 Hz), 6.50 (br s, 0.2H), 6.72-6.75 (m, 1H), 6.86-6.89 (m, 2H), 7.16-7.20 (t, 1H, *J* 7.7 Hz), 7.27 (br, 0.7H); ¹³C-NMR (100 MHz, CDCl₃, COSY, HSQC, HMBC): δ (ppm) 15.8, 24.3 , 25.9, 30.1, 42.4, 54.5, 63.7, 64.8, 69.7, 112.9, 115.1, 122.0, 129.1, 140.3, 158.5, 172.6, 177.5, 182.7, 189.6. HRMS: (ESI) *m/z* [*M*+H]⁺ calcd. for C₂₁H₂₉N₂O₄⁺: 373.2122, found: 373.2146. C₂₁H₂₈N₂O₄ (372.20).

3-Amino-4-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclo-but-3-ene-1,2-dionehydrotrifluoracetate (BMY 25368)²⁴

Compound **5.2** (180 mg, 0.48 mmol, 1 eq) was dissolved in EtOH (20 mL) and a solution of NH₃ in MeOH (7 N, 10 mL) was added. The reaction mixture was stirred over night at room temperature. The solvent was evaporated under reduced pressure and the residue was purified by column chromatography (eluent: CH₂Cl₂ / 7 M NH₃ in MeOH 95:5-90:10). Removal of the solvent *in vacuo* afforded the product as light yellow solid (123 mg, 74.1%). Mp: 199 °C decomposition (Lit.²⁴ mp HCl-salt: 254-257 °C). IR (KBr): 3295, 3135, 2930, 1805, 1645, 1570, 1530, 1260, 695 cm⁻¹. 50 mg of the product were further purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-75:25, t_R = 8.8 min). Removal of the solvent from eluate by evaporation and lyophilisation afforded the product as a hygroscopic white solid (53 mg). R_f = 0.4 (CH₂Cl₂/7 N NH₃ in MeOH 7:1). RP-HPLC (gradient 2, 220 nm) (TFA-Salz): 99% (t_R = 11.38 min, k = 2.9). ¹H-NMR (400 MHz, [D₆]DMSO) (TFA-Salz): δ (ppm) 1.33-1.39 (m, 1H), 1.57-1.70 (m, 3H), 1.80-1.83 (m, 2H), 1.98-2.01 (m, 2H), 2.82-2.89 (m, 2H), 3.29-3.32 (m, 2H), 3.67 (br s, 2H, interfering with the water signal), 4.04-4.07 (t, 2H, *J* 6.0 Hz), 4.22-4.24 (d, 2H, *J* 4.9 Hz), 7.02-7.09

(m, 3H), 7.35-7.39 (t, 1H, *J* 7.9 Hz), 7.55 (br s, 1.8H), 7.68 (br s, 1H), 9.48 (br s, 1H). ¹H-NMR (400 MHz, CDCl₃, COSY, HSQC, HMBC) (TFA-Salz): δ (ppm) 1.46-1.56 (m, 1H), 1.70-1.83 (m, 3H), 1.91-1.95 (m, 2H), 2.07-2.17 (qui, 2H, *J* 6.2 Hz), 2.90-2.97 (m, 2H), 3.42-3.45 (m, 2H, *J* 12.6 Hz), 3.84 (br s, 2H), 4.11-4.14 (t, 2H, *J* 5.9 Hz), 4.23 (s, 2H), 7.01-7.04 (m, 3H), 7.34-7.38 (m, 1H). ¹³C-NMR (100 MHz, CDCl₃, COSY, HSQC, HMBC) (TFA-Salz): δ (ppm) 22.7, 24.1, 31.6, 42.5, 54.1, 61.7, 66.3, 117.1, 118.3, 124.4, 131.4, 131.7, 160.7, 170.6, 170.8, 184.1, 184.4. HRMS (ESI) (TFA-Salz): *m/z* [*M*+H]⁺, calcd. for C₁₉H₂₆N₃O₃⁺: 344.1969, found: 344.1973. C₁₉H₂₅N₃O₃ · C₂HF₃O₂ (343.19 + 114.02).

General procedure for the synthesis of the Mono-Boc-protected diamines

The corresponding alkane diamine (2 eq or 6 eq in case of butane-1,4-diamine) was dissolved in chloroform (30-35 mL). Di-*tert*-butyl dicarbonate (1 eq) was dissolved in chloroform (25-60 mL) and added dropwise over a period of 3 h under ice-cooling. The reaction mixture was allowed to warm up to room temperature while stirring overnight. Potentially precipitated alkane diamine was filtered off. The organic layer was washed with alkaline saturated NaCl solution (45 mL sat. aq. NaCl and 1 mL 1 M aq. NaOH), saturated NaCl solution (45 mL) and H₂O (45 mL). The organic layer was dried over sodium or magnesium sulphate and concentrated by evaporation under reduced pressure. The residue was purified by column chromatography on silica gel.

tert-Butyl N-(4-aminobutyl)carbamate (5.3)⁴⁵ Mengya Chen master thesis

Butane-1,4-diamine (2.42 g, 27.4 mmol, 6 eq) was treated with di-*tert*-butyl dicarbonate (1 g, 4.57 mmol, 1 eq) according to the general procedure. The product was purified by column chromatography (eluent: CH₂Cl₂/1% aq.NH₃ in MeOH 25:1-7:1). Removal of the solvent *in vacuo* afforded the product as beige hygroscopic solid (790 mg, 92%). $R_f = 0.2$ (CH₂Cl₂/NH₃in MeOH 10:1) ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.42 (s, 9H), 1.52 (m, 4H), 2.76 (br s, 2H), 3.11 (br s, 2H), 4.77 (br s, 0.8H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 27.30, 28.44, 29.22, 40.28, 41.06, 79.08, 156.14. MS (LC-MS, ESI): m/z (%) 189 (100) [M+H]⁺. C₉H₂₀N₂O₂ (188.15).

tert-Butyl N-(6-aminohexyl)carbamate (5.4)⁴⁵

Hexane-1,6-diamine (1.5 g, 12.9 mmol, 2 eq) was treated with di-*tert*-butyldicarbonate (1.4 g, 6.5 mmol, 1 eq) according to the general procedure. The product was purified by column chromatography (eluent: $CH_2Cl_2/2$ M NH₃ in MeOH 97.5:2.5-90:10). Removal of the solvent *in vacuo* afforded the product as colorless oil (620 mg, 44.3%). $R_f = 0.8$ ($CH_2Cl_2/7M$ NH₃ in MeOH 5:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.29-1.33 (m, 4H), 1.40-1.48 (m, 13H), 1.53 (br s, 2H), 2.67 (t, 2H, *J* 6.9 Hz), 3.07-3.11 (t, 2H, *J* 6.5 Hz), 4.55 (br s, 1H). MS (LC-MS, ESI, $t_R = 1.2$ min): m/z (%) 217.2 (100) [M+H]⁺, 161.1 (28) [M-C₄H₈]⁺. HRMS (ESI): m/z [M+H]⁺ calcd. for C₁₁H₂₅N₂O₂⁺: 217.1911, found: 217.1931. C₁₁H₂₄N₂O₂ (216.33).

tert-Butyl N-(7-aminoheptyl)carbamate (5.5)⁴⁵

Heptan-1,7-diamine (2.0 g, 15.4 mmol, 2 eq) was treated with di-*tert*-butyldicarbonate (1.7 g, 7.7 mmol, 1 eq) according to the general procedure. The product was purified by column chromatography (eluent: $CH_2Cl_2/2$ M NH₃ in MeOH 97.5:2.5-90:10) Removal of the solvent *in vacuo* afforded the product as colorless, oily solid (950 mg, 53.7%). $R_f = 0.2$ ($CH_2Cl_2/2M$ NH₃ in MeOH 5:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.25 (br s, 6H), 1.31 (br s, 2H), 1.37-1.42 (m, 13H), 2.61 (t, 2H, *J* 6.9 Hz), 3.00-3.05 (m, 2H), 4.67 (br s, 1H). MS (LC-MS, ESI, $t_R = 1.4$ min): m/z (%) 231.2 (100), $[M+H]^+$, 175.1 (23) $[M-C_4H_8]^+$. HRMS (ESI): m/z $[M+H]^+$ calcd. for $C_{12}H_{27}N_2O_2^+$: 231.1067, found: 231.2066. $C_{12}H_{26}N_2O_2$ (230.35).

General procedure for the synthesis of Boc-protected squaramides with alkylamine spacers

Compound **5.2** (1 eq) was dissolved in EtOH (15-25 mL) and added to the respective mono-Bocprotected diamine (1.1 eq) in EtOH (15-25 mL). The reaction mixture was stirred over night at room temperature. The solvent was evaporated under reduced pressure and the product was crystallized from EtOAc (2 mL), NH₃ in MeOH (7 M, 50 μ L) and n-hexane (3 mL). The precipitate was purified by column chromatography on silica gel.

tert-Butyl [4-([3,4-dioxo-2-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclobut-1-en-1-yl]amino)butyl]carbamate (5.6)

Compound **5.6** was synthesized from **5.2** (2.8 g, 7.63 mmol, 1 eq) and **3** (1.58 g, 8.39 mmol, 1.1 eq) according to the general procedure. The product was purified by column chromatography (eluent: $CH_2Cl_2/1\%$ aq. NH₃ in MeOH 40:1-20:1). Removal of the solvent *in vacuo* afforded the product as light yellow hygroscopic solid (1.66 g, 42%). $R_f = 0.2$ ($CH_2Cl_2/2M$ NH₃ in MeOH 5:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.41 (s, 9H), 1.49-1.73 (m, 6H), 1.88 (br s, 4H), 2.00-2.19 (m, 2H), 2.95 (br s, 4H), 3.09-3.15 (m, 2H), 3.65 (q, *J* 6.3 Hz, 2H), 3.80 (q, *J* 6.2 Hz, 2H), 3.94 (s, 2H), 4.15 (t, *J* 6.3 Hz, 2H), 4.87 (br s, 1H), 6.85-6.91 (m, 2H), 7.22-7.26 (m, 2H), 7.99 (br s, 1.8H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 22.6, 23.5, 26.9, 28.5, 30.4, 40.1, 41.1, 44.1, 53.4, 53.8, 62.2, 65.4, 79.1, 116.2, 116.7, 122.9, 129.9, 141.8, 156.2, 159.3, 168.0, 168.6, 182.4, 182.5. HRMS (ESI): $m/z [M+H]^+$ calcd. for $C_{28}H_{43}N_4O_5^+$: 515.3228, found: 515.3230. $C_{28}H_{42}N_4O_5$ (514.32).

tert-Butyl (6-[(3,4-dioxo-2-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclobut-1-en-1-yl)amino]hexyl)carbamate (5.7)²³

Compound **5.7** was synthesized from **5.2** (160 mg, 0.43 mmol, 1 eq) and **5.4** (102 mg, 0.47 mmol, 1.1 eq) according to the general procedure. The product was purified by column chromatography (eluent: $CH_2Cl_2/3.5 \text{ M NH}_3$ in MeOH 97.5:2.5-95:5). Removal of the solvent *in vacuo* afforded the product as light yellow solid (193 mg, 83%). Mp: 115.9-118.7 °C. $R_f = 0.8 (CH_2Cl_2/7 \text{ M NH}_3 \text{ in MeOH 90:10})$. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.30-1.47 (m, 17H), 1.59-1.62 (m, 6H), 2.08-2.14 (qui, 2H, *J* 6.3 Hz), 2.46 (br s, 4H), 3.03-3.08 (m, 2H), 3.48 (s, 2H), 3.60-3.65 (m, 2H), 3.79-

3.84 (m, 2H,), 4.02-4.05 (t, 2H, *J* 6.0 Hz), 5.13 (br s, 1H), 6.76-6.78 (dd, 1H, *J* 8.1 Hz, *J* 1.9 Hz), 6.84-6.89 (m, 2H), 7.16-7.20 (t, 1H, *J* 7.8 Hz), 7.36 (br s, 1.5 H). HRMS (ESI): $m/z [M+H]^+$ calcd. for $C_{30}H_{47}N_4O_5^+$: 543.3541, found: 543.3562. $C_{30}H_{46}N_4O_5$ (542.72).

tert-Butyl (7-[(3,4-dioxo-2-[(3-[3-(piperidin-1-ylmethyl)phenoxy]propyl)amino]cyclobut-1-en-1-yl)amino]heptyl)carbamate (5.8)

Compound **5.8** was synthesized from **5.2** (260 mg, 0.70 mmol, 1 eq) und **5.5** (177 mg, 0.77 mmol, 1.1 eq) according to the general procedure. The product was purified by column chromatography (eluent: $CH_2Cl_2/3.5$ M NH₃ in MeOH 97.5:2.5-95:5). Removal of the solvent *in vacuo* afforded the product as light yellow solid (204 mg, 48%). Mp: 109.8-112.9 °C. $R_f = 0.9$ ($CH_2Cl_2/7$ M NH₃ in MeOH 7:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.25-1.43 (m, 19H), 1.52-1.59 (m, 6H), 2.08-2.14 (qui, 2H, *J* 6.3 Hz), 2.37 (br s, 4H), 3.01-3.06 (m 2H), 3.40 (s, 2H), 3.60-3.64 (m, 2H), 3.82-3.84 (m, 2H), 4.01-4.04 (t, *J* 6.1 Hz, 2H), 5.15 (br s, 0.7H), 6.73-6.75 (m, 1H), 6.85-6.87 (m, 2H), 7.16 (t, 1H, *J* 7.8 Hz), 7.42 (br s, 0.7H), 7.50 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 24.4, 26.0, 26.5, 26.6, 28.6, 28.8, 30.0, 30.9, 31.0, 40.5, 41.7, 44.8, 54.7, 64.0, 64.9, 79.3, 113.4, 115.3, 122.1, 129.2, 140.1, 156.6, 158.8, 168.4, 182.6, 182.7. HRMS (ESI): $m/z [M+H]^+$ calcd. for $C_{31}H_{49}N_4O_5^+$: 557.3697, found: 557.3713. $C_{31}H_{48}N_4O_5$ (556.75).

3-((4-Aminobutyl)amino)-4-((3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)amino)cyclobut-3-ene-1,2-dione bis(hydrotrifluoracetate) (5.9)²³

Compound **5.6** (100 mg, 0.11 mmol, 1 eq) was stirred for 24 h at room temperature in a mixture of CH₂Cl₂ (25-30 mL) and TFA (1 mL). Removal of the solvent *in vacuo* afforded the product as a yellow hygroscopic solid (118 mg, 94%). For pharmacological characterization 57 mg of **9** were further purified by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 5:95-60:40, t_R = 12.8 min). The TFA-salt was obtained as white sticky solid (40 mg; 65%). R_f = 0.1 (CH₂Cl₂/1.8M NH₃ in MeOH 90:10). RP-HPLC (gradient 2, 220 nm): 98% (t_R = 10.66 min, k = 2.7). ¹H-NMR (600 MHz, [D₄]MeOH): δ (ppm) 1.47-1.54 (1H, m), 1.70-1.83 (7H, m), 1.92-1.94 (2H, m), 2.10 (2H, qui, *J* 6.3 Hz), 2.91-2.97 (4H, m), 3.42-3.44 (2H, m), 3.62 (2H, br s), 3.83 (2H, br s), 4.12 (2H, t, *J* 6.0 Hz), 4.22 (2H, s), 7.02-7.04 (2H, m), 7.08 (1H, br s), 7.36 (1H, t, *J* 7.9 Hz). ¹³C-NMR (150 MHz, [D₄]MeOH) (TFA-Salz): δ (ppm) 22.7, 24.1, 25.3, 29.1, 31.7, 40.2, 42.4, 44.3, 54.1, 61.7, 66.2, 118.4, 124.5, 131.4, 131.7, 160.8, 162.5 (q, *J* 40 Hz, TFA), 169.5, 169.7, 183.6 (2C). HRMS (ESI): m/z [*M*+H]⁺ calcd. for C₂₃H₃₅N₄O₃⁺: 415.2709, found: 415.2709. C₂₃H₃₄N₄O₃ · C₄H₂F₆O₄ (414.26 + 228.05).

General procedure for the Boc-deprotection of the amine precursors

The corresponding Boc-protected squaramide derivative (**5.7** or **5.8**) was stirred for 4-24 h at room temperature in a mixture of CH_2Cl_2 (25-30 mL) and HCl in 2-propanol (5-6 M, 13-15 mL). Removal of the solvent *in vacuo* afforded the product as HCl salt. Part of the product was further purified by preparative HPLC.

3-((6-Aminohexyl)amino)-4-((3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)amino)cyclobut-3-ene-1,2-dione bis(hydrotrifluoracetate) (5.10)²³

Compound **5.10** was synthesized from **5.7** (185 mg, 0.34 mmol, 1 eq) according to the general procedure. Removal of the solvent *in vacuo* afforded the product as yellow hygroscopic solid (188 mg). For pharmacological characterization 40 mg of **5.10** were purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-75:25, $t_R = 8.4$ min). The TFA-salt was obtained as highly hygroscopic sticky solid (45 mg; 93%). R_f (TFA-Salz) = 0.1 (CH₂Cl₂/7M NH₃ in MeOH 7:1). RP-HPLC (gradient 2, 220 nm) (TFA-Salz): 98% ($t_R = 12.17$ min, k = 3.2). ¹H-NMR (400 MHz, [D₄]MeOH) (TFA-Salz): δ (ppm) 1.42-1.44 (m, 4H), 1.48-1.57 (m, 1H), 1.62-1.68 (m, 4H), 1.70-1.84 (m, 3H), 1.91-1.96 (m, 2H), 2.07-2.14 (qui, 2H, *J* 6.3 Hz), 2.90-2.98 (m, 4H), 3.43-3.46 (m, 2H), 3.60 (br s, 2H), 3.83 (br s, 2H), 4.13 (t, 2H, *J* 5.9 Hz), 4.23 (s, 2H), 7.02-7.07 (m, 3H), 7.37 (t, 1H, *J* 7.9 Hz). ¹³C-NMR (100 MHz, [D₄]MeOH) (TFA-Salz): δ (ppm) 22.7, 24.1, 26.77, 26.85, 28.4, 31.6, 32.0, 40.6, 42.4, 45.0, 54.1, 61.7, 66.3, 117.2, 118.3, 124.5, 131.4, 131.7, 160.8, 162.2 (q, *J* 41 Hz, TFA), 169.7 (2C), 183.6, 183.9. HRMS (TFA-Salz): (ESI): *m/z* [*M*+H]⁺ calcd. for C₂₅H₃₉N₄O₃⁺: 443.3017, found: 443.3019. C₂₅H₃₈N₄O₃ · C₄H₂F₆O₄ (442.60 + 228.05).

3-((7-Aminoheptyl)amino)-4-((3-(3-(piperidin-1-ylmethyl)phenoxy)propyl)amino)cyclobut-3ene-1,2-dione bis(hydrotrifluoracetate) (5.11)

Compound 5.11 was synthesized from 5.8 (187 mg, 0.33 mmol, 1 eq) according to the general procedure. Removal of the solvent in vacuo afforded the product as yellow hygroscopic solid (197 mg). For pharmacological characterization 85 mg of 5.11 were further purified by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-75:25, t_{R} = 9.1 min). The TFA-salt was obtained as white hygroscopic solid (84 mg, 76%). $R_{\rm f}$ (TFA-Salz) = 0.5 (CH₂Cl₂/7M NH₃) in MeOH 5:1). RP-HPLC (gradient 2, 220 nm) (TFA-Salz): 98% (t_R = 12.83 min, k = 3.4). ¹H-NMR (400 MHz, $[D_6]$ DMSO) (TFA-Salz): δ (ppm) 1.28-1.39 (m, 7H), 1.51 (br s, 4H), 1.61-1.68 (m, 3H), 1.79-1.82 (m, 2H), 1.97-2.03 (qui, 2H, J 6.3 Hz), 2.72-2.89 (m, 4H), 3.29-3.32 (m, 2H), 3.48 (br s, 2H), 3.67-3.68 (br s, 2H), 4.05 (t, 2H, J 6.2 Hz), 4.23 (d, 2H, J 5.0 Hz), 7.00-7.05 (m, 2H), 7.09 (br s, 1H), 7.36 (t, 1H, J 7.9 Hz), 7.76-7.82 (br s, 5H), 9.68 (br s, 0.9H). ¹H-NMR (400 MHz, [D₄]MeOH, COSY, HSQC, HMBC) (TFA-Salz): δ (ppm) 1.39 (br s, 6H), 1.48-1.67 (m, 5H), 1.71-1.82 (m, 3H), 1.90-1.94 (m, 2H), 2.07-2.13 (qui, 2H, J 6.2 Hz), 2.89-2.98 (m, 4H), 3.42-3.45 (m, 2H), 3.59 (br s, 2H), 3.83 (br s, 2H), 4.12 (t, 2H, J 5.9 Hz), 4.23 (s, 2H), 7.01-7.07 (m, 3H), 7.35 (t, 1H, J 7.8 Hz). ¹³C-NMR (100 MHz, [D₄]MeOH, COSY, HSQC, HMBC) (TFA-Salz): δ (ppm) 22.7, 24.0, 27.0, 27.2, 28.4, 29.6, 31.6, 32.0, 40.6, 42.4, 45.1, 54.0, 61.7, 66.2, 117.3, 117.8 (q, J 291 Hz, TFA), 118.2, 124.5, 131.3, 131.7, 160.7, 162.2 (q, J 37 Hz, TFA), 169.5 (2C), 183.3, 183.5. HRMS (TFA-Salz): (ESI): m/z $[M+H]^{+}$ calcd. for $C_{26}H_{41}N_4O_3^{+}$: 457.3173, found.: 457.3178. $C_{26}H_{40}N_4O_3 \cdot C_4H_2F_6O_4$ (456.63 + 228.05).

General procedure for the synthesis of pyridinium-labeled fluorescent ligands

The respective amine-precursor **5.9**, **5.10** or **5.11** (1 eq) was dissolved in anhydrous DMF (200-400 μ L) and DIPEA (10 eq) or TEA (2-3 drops). The pyrylium dye Py-5 (3-8 eq) in anhydrous DMF (180-200 µl) was added portion wise (50 µL) every 20-30 min. Subsequent to the last addition, the reaction mixture was incubated for additional 30 min at RT in the dark. The reaction was stopped by addition of TFA (10 % in H_2O_{milliQ} , 50-200 µL). After dilution with a mixture of H_2O_{milliQ} with 5% MeCN and 0.1% TFA (2 mL) the respective fluorescently labeled ligand was isolated as TFA-salt by preparative HPLC. After lyophilisation of the eluate the product was obtained as a red highly hygroscopic solid.

Compound 5.9 labeled with Py-5 (5.12)²⁵ Mengya Chen master thesis

5.9 (1.95 mg, 4 µmol, 1 eq), Py-5 (4.41 mg, 12 µmol, 3 eq) and TEA (2-3 drops) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 5:95-90:10, $t_{\rm R}$ = 15.5 min) and removal of the solvent by lyophilisation afforded **5.12** as a highly hygroscopic, red solid (1.17 mg, 32%). RP-HPLC (gradient 2, 220 nm): 96.9% ($t_{\rm R}$ = 16.4 min, k = 4.7). HRMS (ESI): m/z [M]⁺ calcd. for C₄₂H₅₄N₅O₃⁺: 676.4221, found: 676.4227. C₄₂H₅₄N₅O₃⁺·C₄HF₆O₄ (676.42+227.04).

Compound 5.10 labeled with Py-5 (5.13) Mengya Chen master thesis

5.10 (2.21 mg, 5.00 µmol, 1 eq), Py-5 (11.02 mg, 30.00 µmol, 6 eq) and TEA (2-3 drops) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-90:10, $t_{\rm R}$ = 16.4 min) and removal of the solvent by lyophilisation afforded **5.13** as a highly hygroscopic, red solid (1.46 mg, 31%). RP-HPLC (gradient 2, 220 nm): 95.1% ($t_{\rm R}$ = 18.1 min, k = 5.2). ¹H-NMR (600 MHz, [D₄]MeOH): δ (ppm) 1.48-1.59 (m, 5H), 1.65-1.70 (m, 2H), 1.72-1.80 (m, 2H), 1.82-1.88 (m, 3H), 1.93-1.95 (m, 2H), 2.08-2.13 (qui, 2H, *J* 6.3 Hz), 2.80 (s, 6H), 2.92-2.97 (m, 2H), 3.03 (s, 6H), 3.42-3.45 (m, 2H), 3.62 (br s, 2H), 3.83 (br s, 2H), 4.13 (t, 2H, *J* 6.0 Hz), 4.23 (s, 2H), 4.36-4.38 (m, 2H), 6.58 (d, 1H, *J* 15.3 Hz), 6.77 (d, 2H, *J* 8.8 Hz), 6.93-6.97 (m, 1H), 7.00-7.07 (m, 4H), 7.37 (t, 1H, *J* 8.2 Hz), 7.44 (d, 2H, *J* 8.9 Hz), 7.63-7.68 (m, 1H), 7.71 (s, 2H). HRMS (ESI): m/z [M]⁺ calcd. for C₄₄H₅₈N₅O₃⁺: 704.4534, found: 704.4542. C₄₄H₅₈N₅O₃⁺: C₄HF₆O₄ (704.45+227.04).

Compound 5.11 labeled with Py-5 (5.14)

5.11 (5 mg, 7.30 µmol, 1 eq), Py-5 (21.4 mg, 58.42 µmol, 8 eq) and DIPEA (12.8 µL, 73.03 µmol, 10 eq) were applied according to general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 5:95-45:55, $t_{\rm R}$ = 28.9 min) and removal of the solvent by lyophilisation afforded **5.14** as a highly hygroscopic, red solid (1.59 mg, 24%). RP-HPLC (gradient 2, 220 nm): 96% ($t_{\rm R}$ = 18.4 min, k = 5.4). HRMS (ESI): m/z [M]⁺ calcd. for C₄₅H₆₀N₅O₃⁺: 718.4691, found: 718.4698. C₄₅H₆₀N₅O₃⁺: C₄HF₆O₄ (718.47+ 227.04).

General procedure for the synthesis of cyanine-labeled fluorescent ligands

A solution of the activated fluorescent dye (1 eq) in anhydrous DMF (30-100 μ l) was added to the respective amine-precursor **5.9** or **5.10** (2.0-3.3 eq) dissolved in anhydrous DMF (100 μ L) and DIPEA (8-10 eq). The reaction mixture was incubated for 45-90 min at RT in the dark. The reaction was stopped by addition of TFA (10 % in H₂O_{milliQ}, 45-60 μ L). After dilution with a mixture of H₂O_{milliQ} with 5% MeCN and 0.1% TFA (150-250 μ L) the respective fluorescently labeled ligand was isolated as TFA-salt by preparative HPLC. After lyophilisation of the eluate the product was obtained as a blue fluffy solid.

Compound 5.9 labeled with S0223 (5.15)

5.9 (8.88 mg, 13.12 µmol, 2.5 eq), S2197 (3.65 mg, 4.84 µmol, 1 eq) and DIPEA (7.7 µL, 44.20 µmol, 8 eq) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-70:30, $t_{\rm R}$ = 24.1 min) and removal of the solvent by lyophilisation afforded **5.15** as a hygroscopic, blue fluffy solid (1.13 mg, 18%). RP-HPLC (gradient 1, 220 nm): 96% ($t_{\rm R}$ = 25.3 min, k = 7.7). HRMS (ESI): m/z [M]⁺ calcd. for C₅₅H₇₁N₆O₄⁺: 879.5531, found: 879.5533. C₅₅H₇₁N₆O₄⁺ · C₄HF₆O₄⁻ (880.21 + 227.04).

Compound 5.10 labeled with S0223 (5.16)

5.10 (11.0 mg, 17.03 µmol, 3 eq), S2197 (3.75 mg, 5.68 µmol, 1 eq) and DIPEA (9.9 µL, 56.76 µmol, 10 eq) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-70:30, $t_{\rm R}$ = 24.9 min) and removal of the solvent by lyophilisation afforded **5.16** as a hygroscopic, blue fluffy solid (2.49 mg, 39%). RP-HPLC (gradient 2, 220 nm): 99% ($t_{\rm R}$ = 25.6 min, k = 7.8). HRMS (ESI): m/z [M]⁺ calcd. for C₅₇H₇₅N₆O₄⁺: 907.5844, found: 907.5839. C₅₇H₇₅N₆O₄⁺ · C₄HF₆O₄⁻ (908.26 + 227.04).

Compound 5.9 labeled with S0436 (5.17)²⁵

5.9 (7.78 mg, 12.11 µmol, 2.5 eq), S0536 (3.40 mg, 4.84 µmol, 1 eq) and DIPEA (6.8 µL, 38.75 µmol, 8 eq) were applied according to general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-65:35, $t_{\rm R}$ = 22.6 min) and removal of the solvent by lyophilisation afforded **5.17** as a hygroscopic, blue fluffy solid (1.49 mg, 25%). RP-HPLC (gradient 1, 220 nm): 97% ($t_{\rm R}$ = 23.8 min, k = 7.2). HRMS (ESI): m/z [M+H]⁺ calcd. for C₅₈H₇₇N₆O₇S⁺: 1001.5569, found: 1001.5573. C₅₈H₇₆N₆O₇S · C₄H₂F₆O₄ (1001.34 + 228.05).

Compound 5.10 labeled with S0436 (5.18)²⁵

5.10 (8.0 mg, 11.93 μ mol, 3.3 eq), S0536 (2.55 mg, 3.63 μ mol, 1 eq) and DIPEA (5.1 μ L, 29.01 μ mol, 8 eq) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 5:95-65:35, $t_{\rm R}$ = 27.0 min) and removal

of the solvent by lyophilisation afforded **5.18** as a hygroscopic, blue fluffy solid (1.53 mg, 34%). RP-HPLC (gradient 2, 220 nm): 97% (t_R = 23.1 min, k = 7.0). ¹H-NMR (600 MHz, [D₄]MeOH): δ (ppm) 1.33-1.36 (5H, m), 1.43-1.52 (5H, m), 1.58-1.61 (2H, m), 1.67-1.84 (18H, m), 1.90-1.94 (4H, m), 1.96-2.02 (2H, m), 2.04-2.08 (2H, m), 2.20 (2H, t, J 7.1 Hz), 2.87-2.95 (4H, m), 3.14 (2H, t, J 6.8 Hz), 3.40-3.42 (2H, m), 3.56 (2H, br s), 3.78 (2H, br s), 4.08-4.14 (6H, m), 4.20 (2H, s), 6.26 (1H, d, J 13.7 Hz), 6.32 (1H, d, J 13.7 Hz), 6.60 (1H, J 12.7 Hz), 6.98-7.01 (2H, m), 7.08 (1H, s), 7.23-7.34 (5H, m), 7.40 (2H, t, J 7.7 Hz), 7.47 (2H, t, J 7.0 Hz), 8.22 (2H, t, J 13.1 Hz). HRMS (ESI): m/z [M+H]⁺ calcd. for C₆₀H₈₁N₆O₇S⁺: 1029.5882, found: 1029.5883. C₆₀H₈₀N₆O₇S · C₄H₂F₆O₄ (1029.40 + 228.05).

Compound 5.9 labeled with S0387 (5.19).

5.9 (5.1 mg, 7.93 µmol, 2.5 eq), S0586 (2.6 mg, 3.17 µmol, 1 eq) and DIPEA (4.4 µL, 25.38 µmol, 8 eq) were applied according to general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 5:95-55:45, $t_{\rm R}$ = 22.6 min) and removal of the solvent by lyophilisation afforded **5.19** as a hygroscopic, blue fluffy solid (1.82 mg, 44%). RP-HPLC (gradient 1, 220 nm): 98% ($t_{\rm R}$ = 18.6 min, k = 5.4). HRMS (ESI): m/z [M+H]⁺ calcd. for $C_{58}H_{77}N_6O_{10}S_2^+$: 1081.5137, found: 1081.5140. $C_{58}H_{76}N_6O_{10}S_2 \cdot C_4H_2F_6O_4$ (1081.40 + 228.05).

Compound 5.10 labeled with S0387 (5.20)

5.10 (4.76 mg, 7.09 µmol, 2 eq), S0586 (2.85 mg, 3.54 µmol, 1 eq) and DIPEA (4.8 µL, 27.61 µmol, 8 eq) were applied according to the general procedure. Purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 5:95-55:45, $t_{\rm R}$ = 23.5 min) and removal of the solvent by lyophilisation afforded **5.20** as a hygroscopic, blue fluffy solid (0.91 mg, 19%). RP-HPLC (gradient 2, 220 nm): 99% ($t_{\rm R}$ = 17.6 min, k = 5.1). HRMS (ESI): m/z [M+H]⁺ calcd. for C₆₀H₈₁N₆O₁₀S₂⁺: 1109.5450, found: 1109.5447. C₆₀H₈₀N₆O₁₀S₂ · C₄H₂F₆O₄⁻ (1109.45 + 228.05).

Fluorescence spectroscopy and determination of quantum yields³⁰

The recording of fluorescence spectra and the determination of quantum yields were performed with a Cary Eclipse spectrofluorometer (Varian Inc., Mulgrave, Victoria, Australia). The photomultiplier voltage of the spectrofluorimeter was set to 400 V throughout. Depicted excitation and emission spectra were recorded with an excitation slit of 10 nm and an emission slit of 10 nm. Appropriate concentrations of the fluorescent ligands, with absorbances between 0.1 and 0.2 at the respective excitation wavelength, were determined with a Cary 100 UV/VIS (Varian Inc., Mulgrave, Victoria, Australia) photometer. Absorption spectra were recorded within a concentration range of 1.5-6 μ M. The excitation wavelength was chosen as close to the absorption maximum as possible (**5.13**, **5.14**) or at an inflection point (**5.20**, **5.18**, **5.16** and cresyl violet perchlorate). For the determination of quantum yields, cresyl violet perchlorate (Acros Organics, Geel, Belgium), with a reported quantum yield of 54% in EtOH⁴⁶, was used as a standard. All spectra were recorded in acryl cuvettes (10 x 10 mm, Ref. 67.755, Sarstedt, Nümbrecht, Germany).

Solutions of the fluorescent ligands in PBS and PBS containing 1% BSA (both pH 7.4) were prepared from 5 mM stock solutions in DMSO. Spectra of the cresyl violet standard were recorded in EtOH. The pure solvents with the same DMSO content were used as reference. The absorption spectra were immediately recorded at 22 °C. The emission spectra were recorded within 15-20 min after preparation of the solutions at a temperature of 22 °C using a 'medium scan rate'. The filter settings were 'auto' for the excitation and 'open' for the emission filter. Fluorescence spectra were recorded at two different slit adjustments (excitation/emission): 10/10 nm and 10/5 nm. The emission starting point was set 15 nm above the excitation wavelength. From every emission spectrum was multiplied with the corresponding lamp correction spectrum. These corrected net spectra were integrated up to 850 nm. From every raw absorption spectra. The absorbance at the excitation wavelength was obtained from the net absorption spectra. The quantum yield was calculated according to the following equation:

 $\Phi_{F(X)} = (A_s/A_x)(F_x/F_s)(n_x/n_s)^2 \Phi_{F(S)}$

 A_s is the absorbance and F_s the integral of the corrected net emission spectrum of the cresyl violet standard solution. A_x and F_x are the absorbance and the integral of the corrected net emission spectrum of the fluorescent ligand. The refractive indices of the solvents for the fluorescent ligands and the cresyl violet standard are denoted n_x and n_s (fluorescent ligands: $n_{PBS} = 1.33$; BSA content was neglected and cresyl violet: $n_{EtOH} = 1.36$). The reported quantum yield of cresyl violet perchlorate (in this case 54%) is referred to as $\phi_{F(X)}$.

5.4.3 Pharmacological Methods

Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein or coexpressing the $hH_3R + G_{i\alpha 2} + \beta_1\gamma_2$ or $hH_4R + G_{i\alpha 2} + \beta_1\gamma_2$ proteins was described elsewhere.⁴⁷

Radioligand competition binding assays were performed as described previously with minor adjustments using the following radioligands: $[{}^{3}H]UR-DE257^{23}$ (specific activity = 32.89 Ci/mmol, hH₂R: K_{d} = 12.2 nM, c_{final} = 20 nM) and $[{}^{3}H]$ histamine (Hartmann Analytic, Braunschweig, Germany; specific activity = 25 Ci/mmol; hH₃R: K_{d} = 12.1 nM, c_{final} = 15 nM, hH₄R: K_{d} = 15.9 nM, c_{final} = 10 nM).

On the day of the experiment Sf9 membranes were thawed and sedimented by centrifugation at 13,000 rpm at 4 °C for 10 min. The membranes were resuspended in ice cold binding buffer (12.5 mM MgCl₂, 1mM EDTA and 75 mM Tris/HCl, pH 7.4; in the following referred to as BB) and adjusted to a protein concentration of 2-4 μ g/ μ L. 80 μ L BB containing 0.2% BSA and the respective radioligand, followed by 10 μ L of the investigated ligands at various concentrations (dissolved in H₂O), were added to every well of a 96-well plate (in case of fluorescent ligands: Primaria clear flat bottom microplates, Corning, New York, USA; for other ligands: PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the

membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with 0.3% polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with 200 μ L scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h. The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

Functional GTPyS assay on Sf9 insect cell membranes

GTP γ S assays were performed as described previously⁴⁸ with minor modifications. [³⁵S]GTP γ S (specific activity = 1000 Ci/mmol) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to 0.5-1.5 µg/µL.

Agonist mode: 80 μ L of BB containing BSA (0.05% final), GDP (1 μ M final) and [³⁵S]GTP γ S (20 nCi final), followed by 10 μ L of the investigated ligands at various concentrations (dissolved in H₂O) were added to every well of a 96-well plate (in case of fluorescent ligands: Primaria microplates; for other ligands: PP microplates). Incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTP γ S assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine (1 μ M final).

Cell culture

The preparation of stably transfected HEK cells (HEK293T-hH₂R-qs5²⁶ and HEK293T-hH₂R- β Arr2^{33,49}) was described elsewhere.

Cells were cultivated at 37 °C in a water saturated atmosphere containing 5% CO₂. Dulbecco's Modified Eagle Medium, containing 4.5 g/L glucose, 3.7 g/L NaHCO₃, 110 mg/L sodium pyruvate (DMEM, Sigma-Aldrich Munich, Germany) and supplemented with 0.584 g/L L-glutamine (L-glutamine solution, Sigma-Aldrich Munich, Germany), 1% (v/v) Penicillin-Streptomycin (P/S, 10,000 U/mL, Sigma-Aldrich Munich, Germany), 10% (v/v) fetal calf serum (FCS, Biochrom GmbH, Merck, Berlin, Germany) were used as a culture medium. Additionally, 100 µg/mL hygromycin B (A.G. Scientific, Inc., San Diego, CA) and 400 µg/mL G418 (Biochrom GmbH, Merck, Berlin, Germany) were added to the culture medium of HEK293T-hH₂R-qs5 cells, and 400 µg/mL zeocin (InvivoGen, San Diego, USA) and 600 µg/mL G418 were added to the culture medium of HEK293T-hH₂R- β Arr2 cells.

Flow cytometric binding assays

All flow cytometric binding studies were performed with a FACS CaliburTM flow cytomter (Becton Dickinson, Heidelberg, Germany), equipped with an argon laser (488 nm) and a red diode laser (635 nm) according to general protocols^{26,50} with minor adjustments. The following instrument settings were used: FSC: E-1, SSC: 280 V, FI-3: 600 V and FI-4: 420-550 V. All samples were prepared in duplicate and recorded either in channel FI-3 (pyridinium dyes, excitation: 488 nm, emission filter: >670 nm) or in FI-4 (cyanine dyes, excitation: 635 nm, emission filter: 661 ± 18 nm). Sample measurement was complete after 30-45 s (this corresponds to approx. 20,000-90,000 gated events).

HEK293T-hH₂R-qs5 cells were seeded in a 175-cm² culture flask 5-7 days prior to the experiment. On the day of the experiment, cells were trypsinized and detached with fresh culture medium (5 mL). After centrifugation (250 g, 10 min) the cell pellet was resuspended in Leibovitz's L-15 culture medium (L-15 medium, Gibco/Life Technologies, Carlsbad, USA) and the concentration was adjusted to 0.5-1.0 \cdot 10⁶ cells/mL.

Saturation binding: 500 μ L of the cell suspension were either added to 5 μ L of DMSO/H₂O (30/70, v/v, total binding) or to 5 μ L of famotidine in DMSO/H₂O (30/70, v/v, unspecific binding, 300-fold excess to the fluorescent ligand). Incubation was started by the addition of 5 μ L of fluorescent ligand in DMSO/H₂O (30/70, v/v, 100-fold concentrated) in intervals of 1 min (measuring time per sample = one concentration) starting with the lowest concentration of total binding. After 90 min of shaking in the dark at 25 °C, the samples were transferred to 5 mL polystyrol FACS tubes (Sarstedt, Nümbrecht,Germany) and immediately measured.

Competition binding: To 500 μ L of cell suspension, 5 μ L of competitor in DMSO/H₂O (30/70, v/v); 100-fold concentrated) were added at increasing concentrations and 5 μ L of fluorescent ligand (concentration in the assay: 50 nM (**5.14**) or 25 nM (**5.18**)) in DMSO/H₂O (30/70, v/v, 100-fold concentrated) in intervals of 1 min (measuring time per sample). The incubation time was 90 min at 25 °C.

Association: 500 μ L of the cell suspension were either added to 5 μ L of DMSO/H₂O (30/70, v/v, total binding) or to 5 μ L of famotidine in DMSO/H₂O (30/70, v/v, unspecific binding, 300-fold excess to the fluorescent ligand). Incubation started by addition of 5 μ L of fluorescent ligand in DMSO/H₂O (30/70, v/v, 100-fold concentrated, final concentration: 50 nM (**5.14**), 15 nM (**5.16**) or 25 nM (**5.18**)). The incubation at 37 °C was stopped after different periods of time (0-120 min) by measuring the samples.

Dissociation: 500 μ L of the cell suspension were either added to 5 μ L DMSO/H₂O (30/70, v/v, total binding) or to 5 μ L famotidine in DMSO/H₂O (30/70, v/v, unspecific binding, 300-fold excess to the fluorescent ligand). 5 μ L of fluorescent ligand in DMSO/H₂O (30/70, v/v, 100-fold concentrated, final concentration of the fluorescent ligands: 50 nM (**5.14**), 15 nM (**5.16**) or 25 nM (**5.18**)) were added to every vessel and the samples were incubated at 25 °C for 90 min. The samples were centrifuged (250 *g*, 3.5 min,) and the supernatant, containing excess fluorescent ligand was aspirated. 500 μ L pf L-15 medium containing famotidine (300-fold excess to the final fluorescent ligand concentration, 15 μ M, 4.5 μ M or 7.5 μ M) were added to the cell pellet, before

the cells were resuspended. The incubation at 37 °C was stopped after different periods of time (0-150 min) by measuring the samples.

For data analysis the software FlowJo V10 (FlowJo, LLC, Ashland, USA) was used throughout. The geometrical mean values of Fl-3 or Fl-4 were obtained for a subpopulation of the gated cells which exhibited a high receptor density.

High content Imaging

Fluorescent ligand binding experiments with adherent $HEK293T-hH_2R-qs5$ cells (IN Cell Analyzer)

For high content imaging of adherent cells, a wide-field cell imaging system, the IN Cell Analyzer 2000 (GE Healthcare, Little Chalfont, UK), was used. An objective with a 20-fold magnification and a numerical aperture of 0.45, combined with a polychroic mirror (QUAD1), was used throughout. For imaging, two channels with different excitation/emission filters were applied: Cy5 channel (for cyanine dyes, excitation filter: 645/30 nm, emission filter: 705/72 nm, exposure time: 1,000 ms) or Cy3 channel (pyridinium dyes, excitation filter: 543/22 nm, emission filter: 605/64 nm, exposure time: 1,000 ms) and DAPI channel (H33342, excitation filter: 350/50 nm, emission filter: 455/50 nm, exposure time: 90 ms). 2.5-D images (imaging modality, the system uses the camera's CCD/sCMOS chip to integrate the signal over the specified Z section and then deconvolves the result for a pseudo 3-D projection) were obtained throughout. In some cases a brightfield channel (excitation filter: 473/10 nm, emission filter: 455/50 nm, exposure time: 50 ms, imaging modality: 2-D images, the system acquires a standard two-dimensional image) was additionally applied. In every channel, 4 images were obtained per well. Binding experiments were performed in duplicate and were repeated at least twice.

One day prior to the experiment, HEK293T-hH₂R-qs5 cells were trypsinized and detached with DMEM medium (high glucose without phenol red (Gibco/Life Technologies, Carlsbad, USA) containing 1% (v/v) Penicillin-Streptomycin (P/S, 10,000 U/mL, Gibco/Life Technologies, Carlsbad, USA) and 10% (v/v) FCS (in the following referred to as DMEM w/o medium). The cell suspension was adjusted to 0.6-0.75 \cdot 10⁶ cells/mL and 200 µl (120,000 - 150,000 cells/well) were seeded in every well of a µ-slide-96-well plate (Ibidi, Martinsried, Germany) using an automated reagent dispenser (Multidrop, Thermo Fisher Scientific, Waltham, USA). The cells were cultivated at 37 °C overnight in a water saturated atmosphere containing 5% CO₂.

On the day of the experiment, the medium was removed and cells (concentration approximately $3 \cdot 10^8$ cells/well) were covered with 80 µl fresh DMEM w/o medium additionally containing 0.1% BSA (Albumin bovine fraction V, SERVA, Heidelberg, Germany) and H33342 (1 µg/mL in H₂O, Sigma Aldrich, Munich, Germany).

Saturation binding: For determination of total binding, DMEM w/o medium (10 μ L) and for unspecific binding DMEM w/o medium (10 μ L) and famotidine as competitor (300-fold referring to the fluorescent ligand) were added. Incubation was started by addition of DMEM w/o medium (10 μ L) containing the respective concentrations of the fluorescent ligand (10-fold concentrated) to every well.

Competition binding: DMEM w/o medium (10 μ L) containing the investigated ligands at various concentrations (10-fold concentrated) was added. For determination of unspecific binding famotidine (100 μ M final) was used. Incubation time started by addition of DMEM w/o medium (10 μ L) containing the fluorescent ligand **18** (10-fold concentrated, 50 nM final) to every well.

After incubation at room temperature in the dark for 60 min, the medium was removed and the cells were washed with PBS (100 μ L) and covered with DMEM w/o medium (100 μ L) followed by immediate acquisition of images at 37 °C.

For kinetic studies the cells were seeded in a 96-well plate one day prior to the experiment as described before. For one association or dissociation experiment only four wells of the plate (two wells for total binding and unspecific binding respectively) were required. The remaining wells were used for other experiments like saturation or competition binding experiments. On the day of the experiment the medium was removed and cells were incubated with 100 μ l DMEM w/o medium containing 0.1% BSA and H33342 (1 μ g/mL, Sigma Aldrich, Munich, Germany) for 1 h. The medium was carefully aspirated, the cells were washed with PBS (100 μ L) and covered with 100 μ l DMEM w/o medium. Either 10 μ L (total binding) or 20 μ L (unspecific binding) medium were removed from the wells. DMEM w/o medium (10 μ L) containing famotidine (10-fold concentrated, 15 μ M final) was added to the two wells of unspecific binding.

Association: DMEM w/o medium (10 μ L) containing the fluorescent ligand **18** (10-fold concentrated, 50 nM final) was added to every well, and the plate was immediately transferred to the IN Cell Analyzer 2000. Images at different time points between 0 h and 1 h were acquired at 37 °C.

Dissociation: DMEM w/o medium (10 μ L) containing the fluorescent ligand **18** (10-fold concentrated, 50 nM final) was added to every well, and the plate was incubated in the dark for 1 h. The medium was aspirated and DMEM w/o medium (100 μ L) containing famotidine (15 μ M) was added. Immediately, images at different time points between 0 h and 1 h were acquired at 37 °C.

For data analysis, the software Developer Toolbox 1.9.2 (IN Cell Investigator, GE Healthcare, Little Chalfont, UK) was used. For counting the nuclei per image a segmentation algorithm was applied to the images of the DAPI-channel (nuclei staining) to define the targets (nuclei) and the measure "count" (output: number of targets contained within the region of interest) was applied to the images. For the quantification of bound ligand an object segmentation algorithm was applied to the images of Cy3/Cy5 channel to define the targets (areas where the fluorescent ligand is bound). In case of the pyridinium labeled ligands, the density measure "density level" (measures the gray level intensity within the targets; uncalibrated intensity unit for IN Cell images, the higher the value, the brighter the pixel) was applied to the targets. As statistical function "sum" (output: sum of the density levels of all identified targets within one image) was chosen and the result was divided through the nuclei count. In case of the cyanine labeled ligands the density measure "Density x Area" (mean density within the targets. As statistical function "sum" (output: sum of the Density x Area of all identified targets within one image) was chosen and the result was divided through the nuclei count.

Imaging Flow Cytometry (ImageStreamX)

Imaging flow cytometry was performed with an ImageStreamX Imaging Cytometer (Amnis/Merck Millipore, Darmstadt, Germany), equipped with an objective with 40-fold magnification and a numerical aperture of 0.75. For imaging, an argon laser (488 nm, excitation wavelength) and a diode laser (785 nm, side scatter) were used, instrument settings were 488 nm laser: 75-85 mW, 785 nm laser: 13 mW and velocity 60 mm/s. Images were obtained in channel CH5 (emission filter: 702/85 nm). Measurement was completed after counting 1,000 to 1,500 cells within the defined area limits (bright field, upper limit: 250 and lower limit: 100, measuring time 1.5 to 2.0 min per sample). Binding experiments were performed in duplicate and were repeated at least twice.

HEK293T-hH₂R-qs5 cells were detached with DMEM w/o medium. After centrifugation (300 g, 5 min) the cell pellet was resuspended in fresh DMEM w/o medium. The cell suspension was adjusted to $2 \cdot 10^6$ cells/mL.

Saturation binding: For determination of total binding DMEM w/o medium (20 μ L) containing 1% BSA (Albumin bovine fraction V, SERVA, Heidelberg, Germany) and for unspecific binding DMEM w/o medium (20 μ L) containing 1% BSA and famotidine as competitor (300-fold referring to the fluorescent ligand) were added to 1.5 ml reaction vessels (Sarstedt, Nümbrecht, Germany). The cell suspension (160 μ L) was transferred to the vessels and DMEM w/o medium (20 μ L) containing 1% BSA and the fluorescent ligand in different concentrations (10-fold concentrated) was added. Samples were prepared at intervals of 2 min (measuring time per sample) starting with the highest concentration. After incubation for 60 min at room temperature in the dark, samples were filtered through a 70 μ m nylon cell strainer (Falcon/Corning Inc., New York, USA) and images of the suspended cells were acquired with an ImageStreamX Imaging cytometer.

Competition binding: DMEM w/o medium (210 μ L) containing the investigated ligands in various concentrations (10-fold concentrated) was added to 1.5 ml reaction vessels. For determination of unspecific binding, famotidine (100 μ M final) was used. The cell suspension (160 μ L) was transferred to the vessels and DMEM w/o medium (10 μ L) containing the fluorescent ligand **14** (10-fold concentrated, 70 nM final) was added. All following steps were carried out as described for saturation binding.

For data analysis, the software IDEAS 6.0 (Amnis/Merck Millipore, Darmstadt, Germany) was used. From the gated cells only focused single cells were included in the data analysis. Furthermore, cells with high fluorescence intensity in the cytoplasm (dead or dying cells) were excluded. Then a mask for the cell membrane was created, allowing analysis of the fluorescent intensity in the area of the cell membrane. Additionally, the fluorescent intensity of the whole cell was analyzed.

Confocal Microscopy

Confocal microscopy was performed with a Zeiss Axiovert 200 M microscope equipped with the LSM 510 Laser scanner. A 63x/1.40 oil immersion objective was used.

Two days prior to the experiment, HEK293T-hH₂R-qs5 cells were trypsinized and seeded in a ibiTreat μ -slide 8-well chambered coverslip (Ibidi, Planegg, Germany) in DMEM (0.6 \cdot 10⁶ cells/mL, 250 μ L per well) containing 10% FCS and 1% P/S (10,000 U/mL, Sigma-Aldrich Munich, Germany). On the day of the experiment, the confluency of the cells was approximately 80-90%. The culture medium was replaced with L-15 containing 5% FCS and 1% P/S (120 μ L). For the determination of total binding, blank L-15 medium (40 μ L) and L-15 medium containing the respective fluorescent ligand (40 μ L, 5-fold concentrated, 100 nM final) were added. Unspecific binding was determined by analogy; with the exception that blank medium was replaced with L-15 medium containing famotidine (40 μ L, 5-fold concentrated, 30 μ M final). Images of total and unspecific binding were acquired after an incubation period of 20 min at room temperature. Table 5.12 shows the settings for the detection of the investigated fluorescent ligands.

Compounds	Excitation (laser intensity)	Filter	Pinhole (μm)
5.14	488 nm (10%)	LP 560	106
5.16	633 nm (10%)	LP 650	122
5.18	633 nm (10%)	LP 650	122

Beta-Arrestin2 recruitment assay

The β -Arrestin2 recruitment assays were performed as described previously for the H₁R using HEK293T-hH₂R- β Arr2 cells, stably expressing the hH₂R-ElucC and β Arr2-ElucNfusion constructs.⁴⁹

One day prior to the experiment, HEK293T-hH₂R- β Arr2 cells were trypsinized and detached with DMEM medium (high glucose without phenol red (Sigma Aldrich, Munich, Germany) containing 1% (v/v) P/S and 5% (v/v) FCS. The cell suspension was adjusted to $1.1 \cdot 10^6$ cells/mL and 90 µl (100,000 cells/well) were seeded in every well of a sterile, luciferase assay compatible, F-bottom 96-well plate (Cellstar[®], Greiner Bio-One, Kremsmünster, Österreich). The cells were cultivated at 37 °C overnight in a water saturated atmosphere containing 5% CO₂. The investigated ligands were added at increasing concentrations (10 µL), and the plate was incubated at 25 °C for 60 min under shaking. 50 µL of the medium were removed, and 50 µL of Bright-Glo reagent (Promega, Madison, USA) were added. Bioluminescence was immediately measured for 1 s per well using a GENios Pro microplate reader (Tecan, Salzburg, Austria).

5.3.4 Data analysis

Retention factors k were calculated according to $k = (t_R - t_0)/t_0$ (t_0 = dead time; t_R = retention time). Corrected counts per minute (ccpm) from the GTPyS assay (agonist mode) were plotted against the log(concentration of the test compound), and data were analyzed by a four parameter logistic equation (GraphPad Prism Software 5.0, GraphPad Software, San Diego, CA), followed by normalization (0% = water value (basal activity), 100% = "top" histamine equation) and analysis by four-parameter logistic equation (log(agonist) vs. response – variable slope, GraphPad Prism). Data of the GTPyS assay (antagonist mode) were analysed by a four parameter logistic equation (GraphPad Prism), followed by normalization (100% = "top" of the fourparameter logistic fit, 0% = unspecifically bound radioligand (ccpm) determined in the presence of famotidine at 100 μ M) and analysis by four-parameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism). pIC_{50} values were converted into pK_{B} values according to the Cheng-Prusoff equation⁵¹. The luminescence (RLU) from the βArrestin2 recruitment assay (agonist mode) were plotted against log(concentration of the test compound) and analyzed by a four parameter logistic equation (GraphPad Prism) followed by normalization (0% = water value (basal activity), 100% = "top" histamine equation) and analysis by fourparameter logistic equation (log(agonist) vs. response – variable slope, GraphPad Prism). Specific binding data from saturation binding experiments were plotted against the "free" fluorescently labeled ligand concentration and analyzed by a two-parameter equation describing hyperbolic binding (one site – specific binding, GraphPad Prism). Specific binding data from association binding experiments were analyzed by a two parameter equation describing an exponential rise to a maximum (one-phase association, GraphPad Prism) to obtain the observed association constant k_{obs} . Specific binding data from dissociation binding experiments were analyzed by a three parameter equation (one phase decay, GraphPad Prism) to obtain the dissociation rate constant k_{off} . Kinetic dissociation constants $K_{d (kin)}$ were calculated from k_{on} and k_{off} ($k_{on} = (k_{obs} - k_{obs})$ k_{off} /[L]; $K_{d (kin)} = k_{off}/k_{on}$). Specific binding data from association and dissociation binding experiments were normalized (100% = Y_{max} (association) or Y_0 (dissociation)). Total binding data from radioligand and fluorescent ligand competition binding experiments were plotted against log(concentration competitor) and analyzed by a four-parameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism), followed by normalization (100% = "top" of the four-parameter logistic fit, 0% = unspecifically bound radioligand/ fluorescent ligand determined in the presence of famotidine at 100 μ M). Normalized data from competition binding experiments was analyzed by a four-parameter logistic equation (log(inhibitor) vs response variable slope, GraphPad Prism) and obtained p/C_{50} values were converted into pK_i values according to the Cheng-Prusoff equation.⁵¹

5.4 SUMMARY AND CONCLUSION

The introduction of different fluorophores by derivatization of amino-functionalized precursors, structurally related to BMY 2536, led to fluorescently labeled H₂R antagonists. The highest affinities on the hH₂R (pK_i values > 7.0) in radioligand competition assays were obtained by the pyridinium labeled ligands 5.12-5.14 and the cyanine labeled ligands 5.16 (positively charged fluorophore, net charge: 2^+) and **5.18** (electroneutral fluorophore, net charge: 1^+). Interestingly, labeling with S0387, the cyanine dye with the negative net charge, led to a decrease in HH₂R affinity (pK_i values < 6.0). While the linker length (4-7 carbon atoms) had no significant influence on the hH₂R affinity within the pyridinium ligands, the cyanine ligands with the hexyl linker (5.16 and 5.18) showed an increased hH₂R affinity compared to the butyl linker derivatives. Even though the low selectivity towards the hH₃R limited the application to recombinant systems, the investigated fluorescent ligands proved to be useful tools for binding studies using different techniques (flow cytometry and high content imaging). The ligands 5.12-5.18 and 5.20 bound in a saturable manner to the hH_2R (flow cytometry) and the determined K_d values (best results: 5.12: 27.9 nM, 5.13: 14.9 nM, 5.14: 19.7 nM, 5.16: 13.9 nM and 5.18: 48.2 nM) were in good agreement with the corresponding K_i values. The $K_{d (kin)}$ values of **5.14**, **5.16** and **5.18**, calculated from kinetic experiments (nonlinear regression, flow cytometry) were consistent with the K_d values determined in saturation binding experiments. This indicates that the investigated fluorescent ligands follow in part the law of mass action even though they showed an incomplete dissociation (insurmountable antagonism). A similar behavior has been reported for several closely related ligands like the radioligand [³H]UR-DE257. Its amine precursor 5.10 as well as related squaramide type derivatives were reported as insurmountable antagonists, which caused a concentration-dependent depression of the maximal response to the agonist relative to investigated standard ligands (guinea pig right atrium)²³. Investigation of the association and dissociation kinetics of the cyanine labeled ligand 5.18 with high content imaging (INcCell Analyzer) revealed also incomplete dissociation and showed that the residual bound ligand is still located in the cell membrane. A possible explanation for the (pseudo-)irreversible binding of the fluorescent ligands 5.14, 5.16 and 5.18 is a slow rate of dissociation from the receptor as also suggested for the radioligand [³H]UR-DE257.²³ Nonetheless, the fluorescent ligands **5.14** and **5.18** can be used for the determination of binding affinities of unlabeled ligands in competition binding assays (shown for flow cytometry and two high content imaging systems: IN Cell Analyzer and imaging flow cytometry).

The high affinity fluorescent H₂R ligands **5.12-5.14**, **5.16** and **5.18** are an attractive nonradioactive alternative to the structurally related radioligand [³H]UR-DE257²³ with similar pharmacological properties. Additionally these fluorescent ligands can be versatile molecular tools giving access to a plethora of optical techniques such as confocal microscopy, FRET, FRAP, TIRF, high content imaging and fluorescence polarization.

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6. CARBAMOYLGUANIDINE- TYPE H₂R LIGANDS: EXPLORATION OF STABILITY AND SELECTIVITY COMPARED TO THE ACYLGUANIDINE- ANALOGUES

Note: the synthesis of the intermediates **6.13-6.18** and the cabamoylguanidines **6.47-6.52** as well as the investigation of the chemical stability of **6.49**, **6.50**, **6.52**, **UR-Bit22**, **UR-Bit23** and **UR-Bit29** were performed by Claudia Honisch during her Master Thesis 2015.

The radioligand binding experiments at the dopamine receptors were performed by Lisa Forster during her doctoral thesis (ongoing).

6.1 INTRODUCTION

 N^{G} -acylated hetarylpropylguanidines represent a class of potent histamine H_2R agonists.¹⁻³ The first generation N^{G} -acylguanidine ligands with an imidazole mojety lacked subtype selectivity, especially over the H_3R and H_4R (e.g. UR-AK24, Figure 6.1).³ The bioisosteric replacement of the imidazole with a amino(methyl)thiazole moiety resulted in ligands with improved selectivity for the H_2R and retained H_2R potency (Figure 6.1 and Figure 6.2).^{1,4} Investigation of optically active acylguanidine-type compounds (e.g. UR-AK24 and UR-PG267, Figure 6.1) in the GTPase assay revealed eudismic ratios from 1.1 to 3.2, indicating that stereochemistry plays only a minor role, if any.⁴

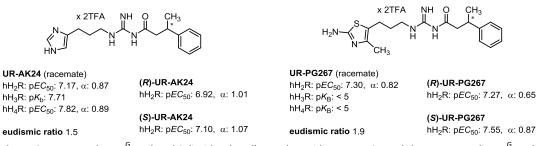


Figure 6.1. Monovalent N^G-acylated 3-(imidazol-4-yl)propylguanidine UR-AK24 and the corresponding N^G-acylated 3-(2amino-4-methylthiazol-5-yl)propylguanidine UR-PG267. Potencies were determined in a steady-state GTPase assay on membrane preparations of Sf9 insect cells expressing the respective receptor.

Interestingly, a broad variety of aliphatic and aromatic hydrocarbon residues was well tolerated in this class of H₂R agonists (Figure 6.2). An increase in H₂R potency was achieved by linking two acylguanidine pharmacophores together via alkyl spacer.² Bivalent ligands with an n-octyl linker (e.g. UR-AK381, Figure 6.2), showed the highest potency (p*EC*₅₀: 8.11, α : 0.53), although spacer length is too short to bridge the two orthosteric binding pockets of the individuals protomers of a putative receptor dimer.² The increased potency compared to that of the corresponding monovalent ligands seems to result from binding to an additional (allosteric) binding site of the same receptor.²

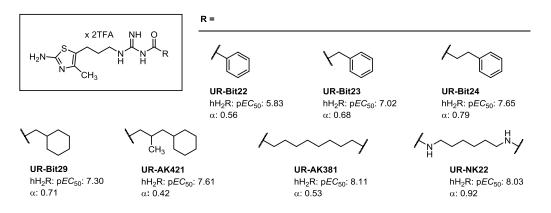


Figure 6.2. Representative H_2R agonists: monomeric and bivalent N^G -acylated 3-(2-amino-4-methylthiazol-5-yl)propylguanidines and the bivalent carbamoylguainidine UR-NK22.^{1,2,5} Potencies were determined in a steady-state GTPase assay on membrane preparations of Sf9 insect cells expressing the H_2R - $G_{s\alpha s}$ fusion protein.

As it was reported that N^G-acylguanidines undergo hydrolytic cleavage upon storage in aqueous solution, more stable analogues are needed.^{5,6} A bioisosteric approach replacing the N^G acylguanidine structure with a carbamoylguanidine has proven useful.^{5,7,8} This was successfully

applied in case of the bivalent N^{G} -acylated hetarylpropylguanidine H₂R agonists leading to highly potent ligands with improved long term stability (e.g. UR-NK22 (Figure 6.2), no decomposition after 7 days in PBS, PH 7.4 at RT).⁵

The aminothiazole moiety is a privileged structure for dopamine receptors. The most prominent example is the nonselective $D_{2-4}R$ agonist pramipexole (Figure 6.3). But also bulkier pramipexole derivatives such as CJ-1639 with high $D_{2-4}R$ affinity were reported.⁹ In preliminary binding studies on the $hD_{2long}R$ and $hD_{3}R$ variants, UR-NK22 and other bivalent amino(methyl)thiazole containing ligands showed in part moderate to high affinity towards these receptors (*ongoing Dissertation Lisa Forster*).

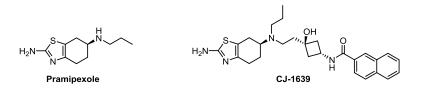


Figure 6.3. D₂₋₄R ligands pramipexole and CJ-1639.

For exploration of the structure-activity relationship (H_2R) and the structure-selectivity relationships (H_2R versus H_1R , H_3R and H_4R) of this class of compounds, in addition to bivalent ligands, a series of carbamoylguanidines with various aminothiazole-based substructures, i.e., the 3-(2-amino-4-methylthiazol-5-yl)propyl moiety, a conformationally constrained (amino-thiazolyl)phenyl and a 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl portion were synthesized (Figure 6.4) in this doctoral project. Additionally, two homobivalent ligands were prepared by replacement of the amino(methyl)thiazolepropyl moiety in UR-NK22 with either a (aminothiazolyl)phenyl or a pyrazolylpropyl moiety.

The synthesized monovalent and bivalent ligands were investigated in competition binding and functional assays (GTP γ S binding and β -arrestin2 recruitment assay). In addition, selected compounds were investigated for D_{2/3}R binding affinity on homogenates of HEK cells stably expressing the hD_{2long}R or hD₃R.

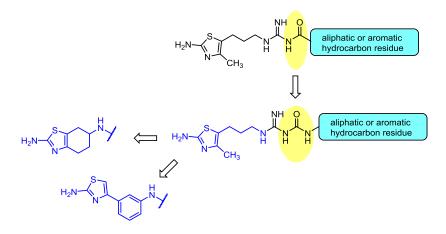
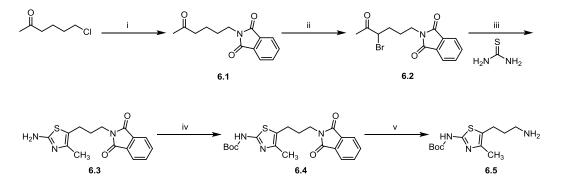


Figure 6.4. Bioisosteric replacement of the N^{G} -acylguanidine in H_2R agonists by a carbamoylguanidine and conformal restriction by replacement of the 3-(2-amino-4-methylthiazol-5-yl)propyl moiety by either an (aminothiazolyl)phenyl or an 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety.

6.2 RESULTS AND DISCUSSION

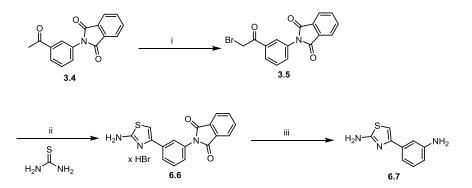
6.2.1 Chemistry

The synthesis of the Boc-protected amine building block 3-(2-amino-4-methylthiazol-5-yl) propylamine (**6.5**) is outlined in Scheme 6.1 according to published procedures.¹ Starting from 6-chlorohexan-2-one, a Gabriel reaction led to the phthalimide **6.1**. Subsequent bromination at room temperature resulted in the thermodynamically more stable intermediate **6.2**. After evaporation of the solvent, the residue was treated with thiourea to give the 2-amino-4-methylthiazole derivative **6.3** in a substitution/ring closure reaction. Protection of the free amino group by a *tert*-butoxycarbonyl function and subsequent hydrazinolysis led to the amine intermediate **3**-(2-amino-4-methylthiazol-5-yl) propylamine (**6.5**).



Scheme 6.1. Synthesis of the Boc-protected aminothiazole **6.5**. Reagents and conditions: i) phthalimide, K₂CO₃, DMF, 80 °C, 24 h, 70%; ii) Br₂, CH₂Cl₂, 1,6-dioxane, RT, 1 h, no purification; iii) DMF, 100 °C to RT, ON, 86%; iv) di-tert-butyldicarbonate, triethylamine, 4-(dimethylamino)-pyridine, CHCl₃, RT, ON, 20%; v) hydrazine-monohydrate, EtOH, RT, ON, 83%.

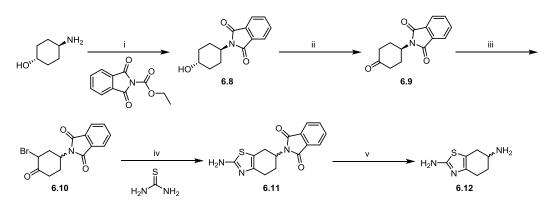
The synthesis of the second amine building block **6.7** was carried out by analogy with the synthesis of 2-guanidino-4-(3-phthalimidophenyl)thiazole hydrobromide (**3.6**) and is outlined in Scheme 6.2.¹⁰ The ketone **3.4** was treated with bromine and subsequently with thiourea in order to form the protected aminothiazole building block **6.6** in a good yield of 78% over two steps. Deprotection of the phthalimide group in a mixture of HCl and acetic acid afforded the conformationally constrained amine building block **6.7**.



Scheme 6.2. Synthesis of 2-amino-4-(3-aminophenyl)thiazole (**6.7**). Reagents and conditions: i) Br₂, HBr in AcOH, CHCl₃, RT, 1h, no purification; ii) EtOH, MeCN, reflux, 3 h, 78%; iii) HCl, AcOH, reflux, ON, 60%.

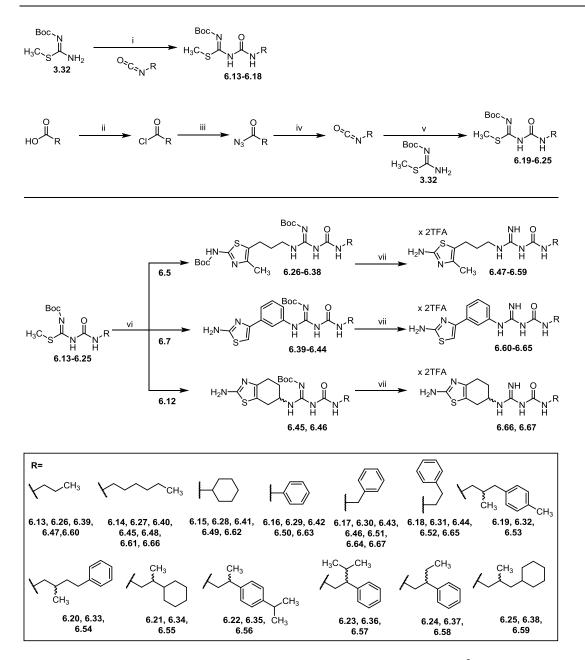
The synthetic route of the third amine building block was adopted from the synthesis of pramipexole (Scheme 6.3).¹¹ 4-Amino cyclohexanol was amino protected using

N-(ethoxycarbonyl)phthalimide; then the hydroxy group was oxidized by pyridinium chlorochromate. The resulting ketone **6.9** was brominated in alpha position and subsequently treated with thiourea to afford the phthalimide protected aminothiazole building block **6.11**. 4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (**6.12**) was obtained by refluxing **6.11** in a mixture of hydrochloric acid and acetic acid.



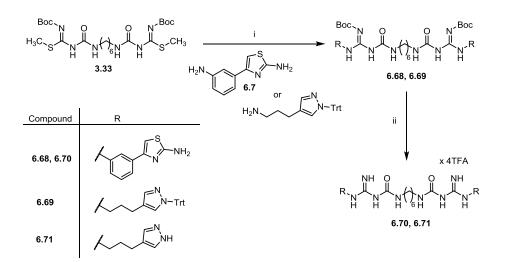
Scheme 6.3. Synthesis of the 4,5,6,7-tetrahydrobenzo[d]thiazole-2,6-diamine (**6.12**). Reagents and conditions: i) K_2CO_3 , H_2O , RT, 30 min, 98%; ii) pyridinium chlorochromate, anhydrous CH_2Cl_2 , RT, 3.5 h, 71%; iii) Br_2 , 1,6-dioxane, CH_2Cl_2 , RT, 1.5 h, no purification; iv) DMF, 100 °C, 2 h, 92%; v) HCl, AcOH, reflux, ON, 73%.

The N^G-carbamoylated guanidines were synthesized by guanidinylation of the amine building blocks 6.5, 6.7 and 6.12. S-methylcarbamoyl thiourea derivatives were used as guanidinylating reagents. These S-methylcarbamoyl thiourea derivatives (6.13-6.25) were synthesized by treatment of *N-tert*-butoxycarbonyl-S-methylisothiourea (3.32, synthesis in chapter 3) with the respective isocyanate (Scheme 6.4) in the presence of TEA at room temperature. The isocyanates were either commercially available (synthesis of 6.13-6.18) or were prepared by Curtius rearrangement from the carboxylic acid (see synthesis of 6.19-6.25). The branched carboxylic acids, which were previously synthesized in our group,^{1,12} were treated with oxalyl chloride and the resulting acid chloride was converted into acyl azide by treatment with sodium azide. Thermal decomposition of the acyl azide led to the isocyanate, which was used for the synthesis of the corresponding S-methylcarbamoyl thiourea derivative (Scheme 6.4). The N^Gcarbamoylated guanidines 6.47-6.67 were prepared by treating the amine building blocks 6.5, 6.7 or 6.12 with the respective S-methylcarbamoyl thiourea derivative 6.13-6.25 in the presence of $HgCl_2$ and base. The resulting protected carbamoylguanidine-type intermediates **6.26-6.46** were treated with TFA to obtain the N^G-carbamoylated guanidines **6.47-6.67** in a yield of 16-75%. after purification by preparative HPLC.



Scheme 6.4. Synthesis of the S-methylcarbamoyl thiourea derivatives **6.13-6.25** and the N^G-carbamoylated guanidines **6.47-6.67**. Reagents and conditions: i) TEA, CH_2Cl_2 , RT, ON, 57-85%; ii) Oxalylchloride, DMF, CH_2Cl_2 , 0 °C-RT, 25 min, no purification; iii) Sodium azide in H₂O, acetone, ice bath, 30 min, no purification; iv) CH_2Cl_2 , reflux, 30 min, no purification; v) TEA, CH_2Cl_2 , RT, ON, 38-77% over four steps; vi) HgCl₂, TEA, anhydrous CH_2Cl_2 , Ar-atmosphere, RT, ON, 35-97%; vii) TFA, CH_2Cl_2 , RT, ON, 16-75%

The bivalent carbamoylguanidine-type compounds **6.70** and **6.71** were synthesized using established protocols (Scheme 6.5).^{5,13} The amine precursor **6.7** or 3-(1-trityl-1*H*-pyrazol-4-yl)prop-1-yl-amine was treated with guanidinylating reagent **3.33** in the presence of HgCl₂ and base. Subsequent treatment of the protected intermediates **6.68** and **6.69** with TFA afforded the bivalent N^{G} -carbamoylated guanidines **6.70** and **6.71**.



Scheme 6.5. Synthesis of the bivalent N^G-carbamoylated guanidines (**6.70**, **6.71**). Reagents and conditions: i) HgCl₂, TEA, anhydrous CH₂Cl₂, Ar-atmosphere, RT, ON, 81-91%; ii) TFA, CH₂Cl₂, RT, ON, 52-67%.

6.2.2 Chemical stability of monovalent carbamoylguanidines compared to acylguanidines

N^G-Carbamoylguanidines are reported to show higher stability in basic solution compared to the respective N^G-acylguanidines.⁶ Previous investigations on stability (assay conditions: PBS pH 7.4) of bivalent aminothiazole-containing carbamoylguanidines compared to their acylguanidine counterparts showed that after 7 days 55% of the acylated guanidine decomposed while the carbamoylated guanidine remained intact.⁵

In order to compare the chemical stability of the monovalent thiazole containing carbamoylated guanidines with the corresponding acylated guanidines, and to investigate the influence of various hydrocarbon residues, the compounds **6.49**, **6.50** and **6.52** as well as the acylguanidines UR-Bit22, UR-Bit23 and UR-Bit29 were dissolved in PBS (pH 7.4), incubated at RT for 7 days and analysed by analytical HPLC (conditions see experimental section). Whereas the carbamoylated guanidines remained stable over this period of time (Figure 6.5), decomposition of the acylated guanidines was highly dependent on the nature of hydrocarbon residue (Figure 6.6). After 7 days approximately 62% of the acylguanidine UR-Bit23 was decomposed. However, only 51% of UR-Bit29 and only 33% of UR-Bit22 was decomposed after the same time. The formation of a decomposition product (t_R : 1.97 min) could be observed over a period 7 days (Figure 6.6).

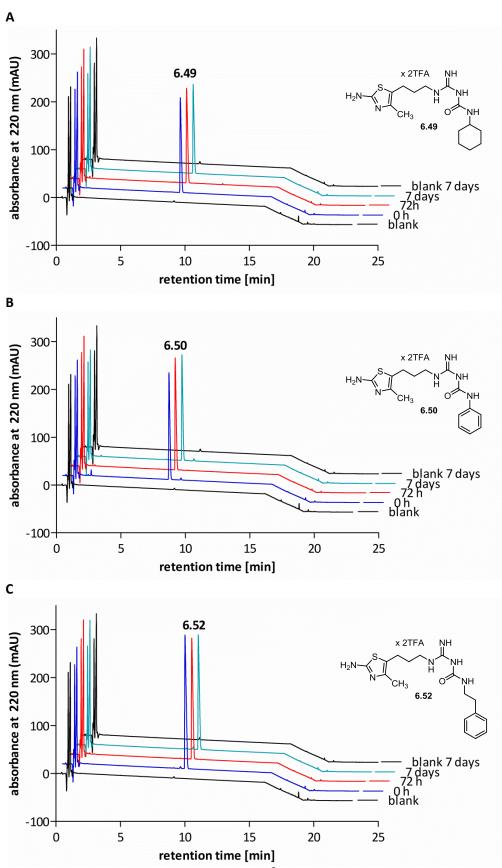


Figure 6.5. HPLC chromatograms (λ : 220 nm) of the N^G-carbamoylated guanidines after different time of incubation in PBS (pH 7.4): (A) **6.49**, t_R: 9.14min; (B) **6.50**, t_R: 8.23 min; and (C) **6.52**, t_R: 9.52 min.

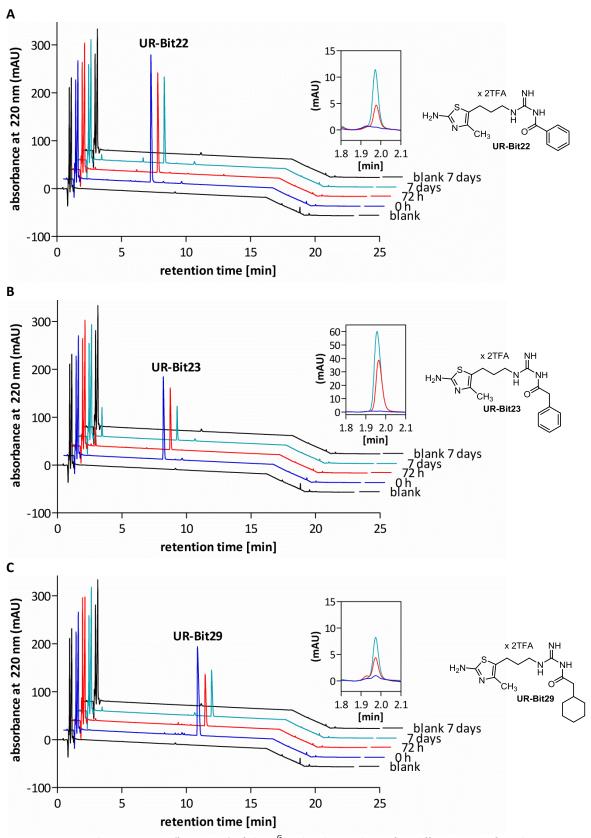


Figure 6.6. HPLC chromatograms (λ : 220 nm) of the N^G-acylated guanidines after different time of incubation in PBS (pH 7.4): (A) UR-Bit22 (t_R : 6.78min), Inset: reduced scaling, decomposition product (t_R : 1.97); (B) UR-Bit23, t_R : 7.71 min, Inset: reduced scaling, decomposition product (t_R : 1.96); and (C) UR-Bit29, t_R : 10.38 min, Inset: reduced scaling, decomposition product (t_R : 1.97).

6.2.3 Biological Evaluation

hH₂R affinities and subtype selectivities

The aminothiazole containing ligands **6.47-6.67**, the bivalent ligands **6.70** and **6.71**, histamine, UR-NK22, UR-Bit23 and pramipexole were investigated in equilibrium competition binding experiments on membrane preparations from Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein using the antagonist [³H]UR-DE257¹⁴ as radioligand. Selected displacement curves are shown in Figure 6.7 and the results are summarized in Table 6.1. The selectivity of representative compounds for the hH_2R compared to hH_1R , hH_3R and hH_4R was investigated by competition binding experiments using membrane preparations from Sf9 insect cells co-expressing either the $hH_1R-G_{s\alpha S}$ fusion protein and RGS4 ([³H]Mepyramine as radioligand) or the $hH_{3/4}R$ and $G_{\alpha i2}$ and $G_{\beta 1\gamma 2}$ proteins ([³H]histamine or [³H]UR-PI294 as radioligand).

The monovalent N^G-carbamoylated amino(methyl)thiazolyl propylguanidines 6.47-6.59 showed moderate to high hH₂R affinity (Table 6.1). In general, aliphatic as well as aromatic residues were well tolerated, only ligand **6.50** which contains a phenyl residue showed a low affinity with a pK_i value of 5.50. The highest hH₂R affinities showed the compounds **6.55** (pK_i value: 7.40) and **6.48** (pK_i value: 7.54) which contain a branched cyclic or a linear aliphatic residue, respectively. In a similar manner as the bivalent N^G-acylated amino(methyl)thiazolylpropylguanidines, bivalent N^Gcarbamoylated amino(methyl)thiazolyl propylguanidines (e.g. UR-NK22)⁵ showed a higher affinity to the hH_2R (pK_i value: 8.02) compared to the monovalent ligands (6.47-6.59). In comparison, ligand 6.51 showed a considerably higher hH_2R affinity (pK_i value: 7.16) than the corresponding acylguanidine UR-Bit23 (pK_i value: 6.3). Incorporation of an aminothiazolylphenyl (compounds 6.60-6.65) or a 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl (compounds 6.66 and 6.67) moiety resulted in a decrease in hH₂R affinity by one to two order(s) of magnitude. In comparison with the D_{2/3}R ligand pramipexole (pK_i value: 4.86), which also contains a 2-amino-4,5,6,7tetrahydrobenzothiazol-6-yl head group, the ligands 6.66 and 6.67 showed increased affinity for the hH₂R with p K_i values of 5.95 and 6.29. The replacement of the amino(methyl)thiazolyl propyl head group of the bivalent ligand UR-NK22 with either an aminothiazolylphenyl residue (6.70) or a pyrazolylpropyl residue (6.71) resulted in a strong decrease of hH_2R affinity (pK_i value: 5.98 and 6.75).

The monovalent N^G-carbamoylated amino(methyl)thiazolyl propylguanidines 6.47-6.59 showed a clear preference for the hH_2R over the other subtypes. In case of the aminothiazolylphenylguanidines 6.60-6.65 and the bivalent ligands (6.70 and 6.71) the affinity for the hH₄R was low (pK_i values 4-5) and, except for **6.63**, a slight preference for the hH₂R was obtained. While 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl containing ligand 6.66 showed a similar affinity to hH₁R, hH₂R, hH₃R with pK_i values of 5.71-5.95 and a lower affinity to the hH₄R, ligand 6.67 showed a preference for hH_2R over the other subtypes of around one order of magnitude.

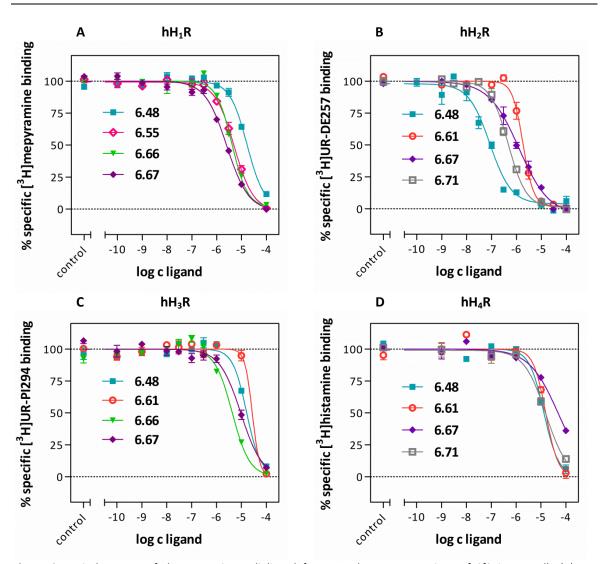


Figure 6.7. Displacement of the respective radioligand from membrane preparations of Sf9 insect cells (A) coexpressing the hH₁R-G_{sαS} fusion protein and RGS4 (radioligand: [³H]mepyramine, c = 5 nM, $K_d = 4.5$ nM), (B) expressing the hH₂R-G_{sαS} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, $K_d = 12.2$ nM), (C) co-expressing the hH₃R and G_{αi2} plus G_{β1γ2} proteins (radioligand: [³H]UR-PI294 c = 2 nM, $K_d = 1.1$ nM) or (D) co-expressing the hH₄R and G_{αi2} plus G_{β1γ2} proteins (radioligand: [³H]bistamine c = 10 nM, $K_d = 15.9$ nM) by exemplary monovalent carbamoylated guanidines. Data represent mean values ± SEM of 2-3 experiments performed in triplicate.

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Table 6.1. Affinities of histamine, pramipexole, UR-NK22, UR-Bit23, the monovalent carbamoylated guanidines **6.47**-**6.67** and the bivalent carbamoylated guanidines **6.70**-**6.71** to $hH_{1-4}R$, obtained from equilibrium competition binding studies on membrane preparations from Sf9 insect cells, expressing the respective histamine receptor subtype.

Compound	hH ₁ R ^a		hH₂R⁵		hH₃R℃		hH₄R ^e	
	р <i>К</i> і	Ν	р <i>К</i> і	Ν	р <i>К</i> і	Ν	р <i>К</i> і	Ν
Histamine	n.d.	-	6.53 ± 0.04	3	7.8 ± 0.1	3	7.65 ± 0.03	3
Pramipexole	n.d.	-	4.86 ± 0.07	5	n.d.	-	n.d.	-
UR-NK22	6.06 ± 0.05^5	2	8.07 ± 0.05^5	3	5.94 ± 0.16^{5}	4	5.69 ± 0.07 ⁵	3
UR-Bit23	n.d.	-	6.3 ± 0.4	2	n.d.	-	n.d.	-
6.47	4.54 ± 0.02	3	6.98 ± 0.11	3	4.35 ± 0.01^{d}	3	4.06 ± 0.06	2
6.48	5.11 ± 0.03	3	7.54 ± 0.07	4	5.25 ± 0.02^{d}	3	5.09 ± 0.02	2
6.49	4.61 ± 0.09	3	6.77 ± 0.27	3	n.d.	-	4.50 ± 0.06	2
6.50	n.d.	-	5.50 ± 0.08	3	n.d.	-	4.65 ± 0.05	2
6.51	5.21 ± 0.02	3	7.16 ± 0.05	3	n.d.	-	4.72 ± 0.09	2
6.52	n.d.	-	6.8 ± 0.2	4	n.d.	-	4.83 ± 0.06	2
6.53	5.8 ± 0.1	3	7.14 ± 0.08	2	5.49 ± 0.01^{d}	3	5.44 ± 0.02^{f}	3
6.54	5.87 ± 0.09	3	7.20 ± 0.04	3	5.04 ± 0.04	3	5.49 ± 0.08^{f}	3
6.55	5.63 ± 0.06	3	7.40 ± 0.01	2	5.00 ± 0.08	3	5.72 ± 0.05^{f}	3
6.56	5.31 ± 0.02	3	6.83 ± 0.08	3	5.10 ± 0.02	3	5.58 ± 0.01	3
6.57	5.23 ± 0.03	3	6.99 ± 0.05	3	4.93 ± 0.03	3	5.23 ± 0.03^{f}	3
6.58	5.10 ± 0.08	3	7.11 ± 0.03	3	4.78 ± 0.03	3	5.17 ± 0.02^{f}	3
6.59	5.42 ± 0.07	3	7.15 ± 0.02	4	5.13 ± 0.01	3	5.43 ± 0.01^{f}	3
6.60	n.d.	-	5.28 ± 0.06	4	n.d.	-	<4.0	2
6.61	n.d.	-	6.16 ± 0.07	4	4.96 ± 0.07^{d}	3	5.02 ± 0.06	2
6.62	n.d.	-	5.35 ± 0.02	3	n.d.	-	4.87 ± 0.03	2
6.63	n.d.	-	3.43 ± 0.07	3	n.d.	-	4.2 ± 0.1	2
6.64	n.d.	-	5.7 ± 0.1	3	n.d.	-	4.46 ± 0.01	2
6.65	n.d.	-	5.40 ± 0.05	3	n.d.	-	4.5 ± 0.1	2
6.66	5.71 ± 0.02	3	5.95 ± 0.06	4	5.82 ± 0.03^{d}	3	4.78 ± 0.05	2
6.67	5.92 ± 0.02	3	6.29 ± 0.08	3	5.48 ± 0.06^{d}	3	4.52 ± 0.01	2
6.70	n.d.	-	5.98 ± 0.06	3	n.d.	-	5.0 ± 0.2	2
6.71	n.d.	-	6.75 ± 0.04	3	n.d.	-	5.02 ± 0.07	3

Competition binding assay on membrane preparations of Sf9 insect cells ^aco-expression of the H_1R-G_{scas} fusion protein and RGS4 (radioligand: [³H]mepyramine, c = 5 nM, K_d = 4.5 nM), ^bhH₂R-G_{scas} fusion protein (radioligand: [³H]UR-DE257, c = 20 nM, K_d = 12.2 nM), ^cco-expression of the H_3R and G_{ai2} and $G_{\beta_{13}2}$ proteins (radioligand: [³H]histamine c = 15 nM, K_d = 12.1 nM or ^d[³H]UR-PI294 c = 2 nM, K_d = 1.1 nM) or ^eco-expression of the hH₄R and G_{ai2} plus $G_{\beta_{13}2}$ proteins (radioligand: [³H]histamine c = 10 nM, K_d = 15.9 nM or ^f[³H]UR-PI294 c = 5 nM, K_d = 5.1 nM). The incubation period was 60 min. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means ± SEM of N independent experiments, each performed in triplicate.

hH₂R agonism or antagonism in the GTPγS binding assay and βArrestin2 recruitment assay

The N^G-carbamoylated guanidines **6.47-6.71**, histamine, UR-NK22, UR-AK421, UR-Bit22, UR-Bit23, UR-Bit24, UR-Bit29 and pramipexole were examined for hH_2R agonism in the GTP γ S binding assay on membrane preparations from Sf9 insect cells expressing the hH_2R -G_{sas} fusion protein. Ligands which exhibited no agonism were also investigated in the antagonistic mode versus histamine as agonist. All identified agonists and selected antagonists were additionally investigated for agonism in the β Arrestin2 recruitment assay on HEK293T-hH₂R- β Arr2 cells, stably expressing the hH_2R -ElucC and β Arr2-ElucNfusion constructs.¹⁵ Prior studies suggested only minor differences between β Arrestin1 and β Arrestin2 recruitment¹⁵ and therefore the β Arrestin1 was not considered. Representative concentration response curves are depicted in Figure 6.8 (GTP γ S binding assay) and Figure 6.9 (β Arrestin2 recruitment assay). The results from these experiments are summarized in Table 6.2.

The monovalent N^G-carbamoylated amino(methyl)thiazolylpropylguanidines **6.47-6.59** were partial to full agonists in the GTP γ S binding assay and showed moderate to high hH₂R potencies (up to 80 fold the potency of histamine; pEC₅₀ values of 6.3-7.7) generally in good accordance to the acylguanidines (results from the GTPase assay)^{1,16} (Table 6.2). At high concentrations (\geq 10-100 μ M) the signal decreased again resulting in nearly bell-shaped concentration response curves (Figure 6.8). Compound **6.51** (α : 0.91, pEC₅₀: 7.5) showed the highest potency combined with full agonism, whereas the corresponding acylguanidine UR-Bit23 (α : 0.68, pEC₅₀: 6.59).was a partial agonist with only moderate potency.

Interestingly, the aminothiazolylphenyl containing monovalent ligands **6.60-6.65** and bivalent ligand **6.70** showed weak antagonism or inverse agonism at the hH_2R with pK_b values in good accordance to the pK_i values.

Incorporation of the less flexible 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl (compounds **6.66** and **6.67**) moiety resulted in partial agonism (**6.67**, α : 0.57, pEC₅₀: 6.7) or in weak partial agonism (**6.66**, α : 0.16, pEC₅₀: 5.57) due to bell-shaped concentration response curves. In the antagonist mode of the GTP_YS assay, **6.66** and **6.67** act as antagonists with pK_b values of 5.57 and 4.33. Also the structurally related dopamine receptor agonist pramipexole was a weak partial agonist at the hH₂R (α : 0.66, pEC₅₀: 5.07) with a sigmoidal curve. Although **6.67** was a racemic mixture, it was more potent than the enantiomerically pure pramipexole.

The pyrazole containing bivalent ligand **6.71** (α : 0.38, pEC₅₀: 7.0) was a partial agonist with low potency compared to the corresponding amino(methyl)thiazole containing full agonist UR-NK22 (α : 0.92, pEC₅₀: 8.03)⁵.

All full or partial agonists identified in the GTP γ S assay (**6.47-6.49**, **6.71**) showed a lower potency and efficacy in the β Arrestin2 recruitment assay. The agonistic N^G-carbamoylated guanidine-type ligands exhibited some functional bias towards G-protein activation. This is in agreement with the findings for acylguanidines and bivalent carbamoylguanidines.¹⁵ The bias was most pronounced in case of **6.53**, which was a full agonist in the GTP γ S assay (α : 0.91) and a weak partial agonist in the β Arrestin2 recruitment assay (α : 0.33). Nearly similar bias was determined for **6.53**, **6.57** and **6.58** which were partial agonists in the GTP γ S assay (α : 0.52-0.68) and very weak partial agonists in the β Arrestin2 recruitment assay (α : 0.10-0.16). Also in the β Arrestin2 recruitment assay of the ligands **6.53-6.59** the signal decreased again at high concentrations (\geq 30-100 µM) resulting in nearly bell-shaped concentration response curves. Interestingly, the 2-amino-4,5,6,7-tetrahydrobenzothiazole containing compounds **6.66** and **6.67**, which showed the most pronounced bell-shaped curves in the GTPyS assay, didn't show a decreased signal at high concentrations in the β Arrestin2 recruitment assay. Ligand **6.66** and two selected antagonists (**6.63** and **6.65**) identified in the GTPyS exhibited no β Arrestin2 recruitment, too.

Bell shaped (or u-shaped) concentration response curves were described at several GPCRs such as muscarinic receptors,¹⁷ β_2 -adrenergic receptors¹⁸ and serotonin receptors¹⁹. Some reasons for such a curve shape could be cytotoxicity,²⁰ binding to multiple binding sites,²¹ multiple targets, receptor oligomers,²² agonist-induced desensitization¹⁸ or even due to physical properties like self-association of ligands into colloidal particals²³.

In case of the GTP γ S assay plausible explanations would be binding to multiple binding sites or direct interaction with the $G_{s\alpha S}$ subunit of the fusion protein due to the use of membrane preparations from Sf9 insect cells expressing the H_2R - $G_{s\alpha S}$ fusion protein instead of live cells. While in the GTP γ S assay all agonists (**6.47-6.59**, **6.66** and **6.67**) showed more or less pronounced bell-shape curves, in the β Arrestin2 recruitment assay only the agonists **6.53-6.59** showed a similar behavior. Interestingly, the ligands **6.48**, **6.50**, **6.52**, **6.66** and **6.67** with a distinguished bell-shaped curve in the GTP γ S assay showed in the β Arrestin2 recruitment assay a sigmoidal concentration response curve indicating that a different mechanism (e.g. cytotoxicity) is leading to the curve shape in the β Arrestin2 recruitment assay. Moreover, it should be mentioned that all chiral compounds (**6.53-6.59**, **6.66** and **6.67**) were racemic mixtures. Therefore, the bell shaped curve could also be due to opposite effects of the enantiomers. However, investigation of enantiomeric pure acylguanidine-type compounds in the GTPase assay revealed only low eudismic ratios (1.1-3.2), indicating the stereochemistry of the acyl residue plays only a minor role.⁴ However, according to the current state of knowledge this is all just speculation.

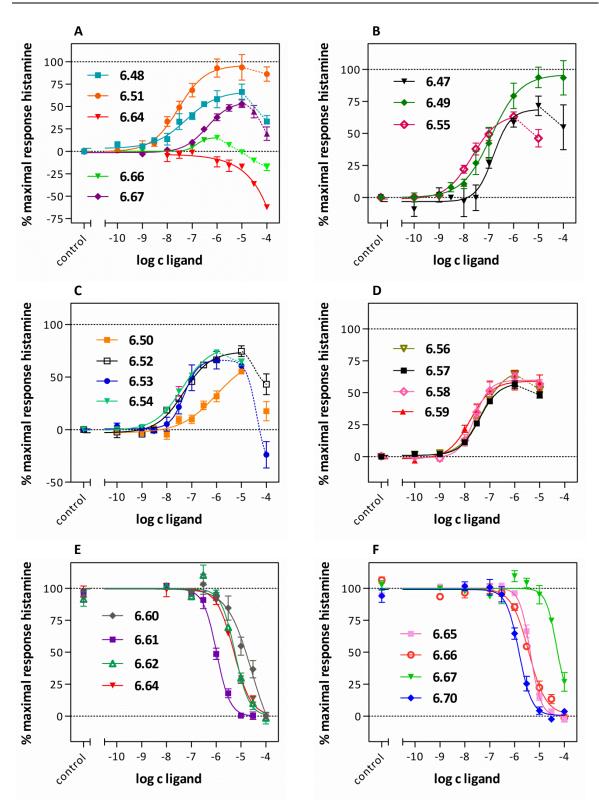


Figure 6.8. Concentration-response curves of representative monovalent carbamoylated guanidines on hH_2R determined by [³⁵S]GTP γ S binding assay on membrane preparations of Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein. (A) Ligands **6.48**, **6.51**, **6.64**, **6.66** and **6.67**. (B) Ligands **6.49** and **6.55**. (C) Ligands **6.50**, **6.52-6.54**. (D) Ligands **6.56-6.59**. (E) and (F) Exemplary ligands measured in the antagonist mode; histamine (1 μ M) was used for stimulation. Data represent mean values ± SEM of 2-5 experiments performed in triplicate.

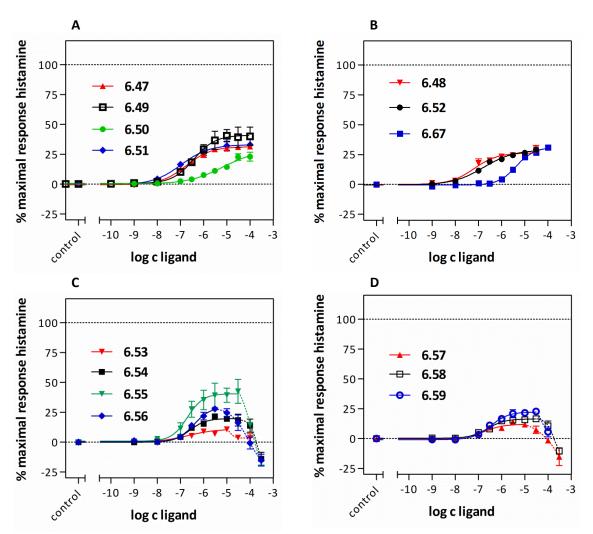


Figure 6.9. Concentration-response curves of representative monovalent carbamoylated guanidines on hH_2R determined by a luciferase complementation assay measuring β -arrestin2 recruitment on HEK293T-hH₂R- β Arr2 cells. (A) and (B): Ligands with sigmoidal dose-response-curves. (C) and (D): Ligands with bell-shaped dose-response-curves. Data represent mean values ± SEM of 3 experiments performed in duplicate.

Compound	hH₂R (GTPγS)ª		hH₂R (βArrestin2 Recruitment) ^b			
	р <i>ЕС</i> ₅₀ (р <i>К</i> ь)	Ν	α	р <i>ЕС</i> 50	Ν	α
Histamine	5.80 ± 0.06	9	1.0	5.38 ± 0.04	19	1.0
Pramipexole	5.07 ± 0.06	5	0.66 ± 0.08	4.4 ± 0.1	4	0.35 ± 0.03
UR-NK22	8.03 ± 0.02^5	2	0.92 ± 0.01^{5}	7.19 ¹⁵	-	0.30 ¹⁵
UR-AK421	7.61 ^{c 1}	-	0.42 ^{c1}	n.d.	-	n.d.
UR-Bit22	5.83 ^{c16}	-	0.56 ^{c16}	n.d.	-	n.d.
	6.59 ± 0.06/	3	0.68 ± 0.13/	n d		
UR-Bit23	7.02 ^{c1}	-	0.68 ^{c1}	n.d.	-	n.d.
UR-Bit24	7.65 ^{c1}	-	0.79 ^{c 1}	7.72 ¹⁵	-	0.14 ¹⁵
UR-Bit29	7.30 ^{c1}	-	0.71 ^{c1}	n.d.	-	n.d.
6.47	6.75 ± 0.06	3	0.79 ± 0.09	6.77 ± 0.04	3	0.32 ± 0.01
6.48	7.41 ± 0.07	3	0.60 ± 0.09	7.07 ± 0.02	3	0.28 ± 0.03
6.49	6.8 ± 0.2	3	0.96 ± 0.07	6.41 ± 0.03	3	0.43 ± 0.06
6.50	6.3 ± 0.2	2	0.59 ± 0.02	5.47 ± 0.06	3	0.25 ± 0.02
6.51	7.5 ± 0.1	4	0.91 ± 0.07	7.00 ± 0.08	3	0.33 ± 0.03
6.52	7.4 ± 0.2	3	0.72 ± 0.06	6.78 ± 0.03	3	0.29 ± 0.02
6.53	7.2 ± 0.1	5	0.68 ± 0.07	6.9 ± 0.1	3	0.10 ± 0.01
6.54	7.53 ± 0.06	3	0.69 ± 0.02	6.45 ± 0.06	3	0.20 ± 0.02
6.55	7.66 ± 0.08	3	0.65 ± 0.03	6.65 ± 0.09	3	0.40 ± 0.06
6.56	7.46 ±0.03	3	0.59 ± 0.02	6.55 ± 0.02	3	0.25 ± 0.03
6.57	7.51 ± 0.06	3	0.52 ± 0.03	6.8 ± 0.1	3	0.11 ± 0.01
6.58	7.5 ± 0.1	3	0.60 ± 0.04	6.57 ± 0.08	3	0.16 ± 0.02
6.59	7.76 ± 0.04	3	0.60 ±0.07	6.52 ± 0.07	3	0.22 ± 0.01
6.60	(4.91 ± 0.09)	3	-0.037	n.d.	-	n.d.
6.61	(6.14 ± 0.03)	3	-0.54	n.d.	-	n.d.
6.62	(5.40 ± 0.02)	3	-0.47	n.d.	-	n.d.
6.63	(4.8 ± 0.1)	3	-0.20 ± 0.02	-	-	0.018 ± 0.009
6.64	(5.44 ± 0.02)	3	-0.62	n.d.	-	n.d.
6.65	(5.52 ± 0.05)	3	-0.28 ± 0.04	-	-	0.043 ± 0.009
6.66	6.83 ± 0.06/	3	0.16 + 0.07			
6.66	(5.57 ± 0.04)	3	0.16 ± 0.07	-	-	0.045 ± 0.005

Table 6.2. hH₂R agonism or antagonism and the calculated p EC_{50} or p K_b values of histamine, pramipexole, UR-NK22, UR-AK421, UR-Bit22, UR-Bit23, UR-Bit24, UR-Bit29, the monovalent carbamoylated guanidines **6.47-6.67** and the bivalent carbamoylated guanidines **6.70-6.71** determined by a GTP γ S assay and β Arrestin2 recruitment assay.

Compound	hH₂R (GTPγS) ^ª			hH ₂ R (βArrestin2 Recruitment) ^b		
	р <i>ЕС</i> ₅₀ (р <i>К</i> _b)	Ν	α	р <i>ЕС</i> ₅₀ (р <i>К</i> _b)	Ν	α
6.67	6.7 ± 0.3/	3	0.53 ± 0.05	5.36 ± 0.05	3	0.31 ± 0.03
	(4.33 ± 0.09)	3		5.50 ± 0.05		
6.70	(5.98 ± 0.06)	3	-0.46	n.d.	-	n.d.
6.71	7.0 ± 0.8	2	0.38 ± 0.03	5.75 ± 0.07	3	0.26 ± 0.08

Table 6.2 continued.

^a[³⁵S]GTP_YS assay determined on membrane preparations of Sf9 insect cells expressing the hH₂R-G_{saS} fusion protein. ^bβArrestin2 recruitment assay determined on HEK293T-hH₂R-βArr2 cells, stably expressing the hH₂R-ElucC and βArr2-ElucNfusion constructs. The incubation period was 60 min. The intrinsic activity (α) of histamine was set to 1.00, and α values of investigated compounds were referred to this value. The pK_B values of neutral antagonists were determined in the antagonist mode versus histamine (c = 1 µM) as agonist. Data represent mean values ± SEM of N independent experiments performed in triplicate (GTP_YS assay) or duplicate (βArrestin2 recruitment assay). ^cSteady-state GTPase assay determined on membrane preparations of Sf9 insect cells expressing the hH₂R-G_{saS} fusion protein.

hD_{2long}R and hD₃R affinities of carbamoylated guanidines

Aminothiazole containing ligands such as pramipexole and its derivatives are described as high affintiy dopamine receptor ligands (preferred D_2R , D_3R and D_4R).²⁴ Therefore, haloperidol, selected N^G-carbamoylated guanidines containing an aminothiazole moiety (**6.53**, **6.55**, **6.58**, **6.59**, **6.61**, **6.66**, **6.67**) and the bivalent ligand **6.71** were investigated in equilibrium competition binding experiments on homogenates of HEK293T-CRE-Luc-h $D_{2long}R$ and/or HEK293T-CRE-Luc-h D_3R cells using [³H]N-methylspiperone as radioligand. The results are summarized in Table 6.3.

For the standard agonist pramipexole biphasic displacement curves were reported at $hD_{2long}R$ and $hD_{3}R$ with pK_{i} values for high (pK_{H} value) and low affinity binding (pK_{L} value).²⁵ The standard antagonist haloperidol and the investigated ligands **6.53**, **6.55**, **6.58**, **6.59**, **6.61**, **6.66**, **6.67** and **6.71** showed monophasic displacement curves. The pK_{i} values of haloperidol at the $hD_{2long}R$ and $hD_{3}R$ were in good accordance with literature results.²⁶

The amino(methyl)thiazolylpropyl (**6.53**, **6.55**, **6.58** and **6.59**), aminothiazolylphenyl (**6.61**) and 2amino-4,5,6,7-tetrahydrobenzothiazol-6-yl (**6.66** and **6.67**) containing ligand(s) showed a weak to moderate affinity for the $hD_{2long}R$ (p K_i value 5.6-6.6). The bivalent ligand (**6.71**) showed a moderate affinity with a p K_i value of 6.97. The amino(methyl)thiazolylpropyl containing ligands **6.53**, **6.55**, **6.58** and **6.59** clearly preferred the hH_2R over the $hD_{2long}R$. The 2-amino-4,5,6,7tetrahydrobenzothiazol-6-yl containing ligands (**6.66** and **6.67**) and the bivalent ligand **6.71** showed a comparable affinity to the hH_2R and the $hD_{2long}R$.

The ligands **6.53**, **6.55**, **6.58**, **6.59** and **6.71** showed a moderate affinity for the hD₃R with pK_i values between 6.9 and 7.6. The ligands **6.61**, **6.66** and **6.67** showed a low affinity for the hD₃R with pK_i values between 5.3 and 5.9. The N^G-carbamoylated amino(methyl)thiazolyl-propylguanidines **6.53**, **6.55**, **6.58** and **6.66** bound non-selectively to both the hH₂R and hD₃R receptors. The 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety containing ligand **6.67** showed a preference towards hH₂R and hD_{2long}R over the hD₃R. In contrast, the bivalent ligand **6.71** showed a preference towards hD₃R over the hD_{2long}R and hH₂R.

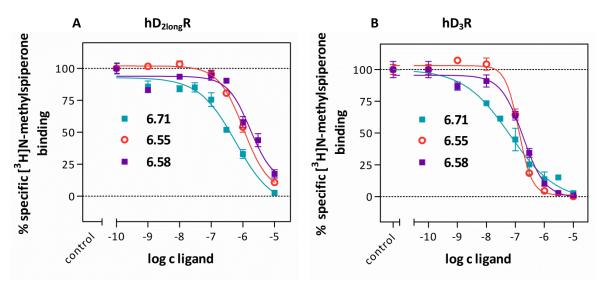


Figure 6.10. Displacement of $[^{3}H]N$ -methylspiperone from homogenates of (A) HEK293T-CRE-Luc-hD_{2long}R cells ($[^{3}H]N$ -Methylspiperone: K_{d} = 0.014 nM, c = 0.05 nM) or (B) HEK293T-CRE-Luc-hD₃R cells ($[^{3}H]N$ -methylspiperone: K_{d} = 0.026 nM, c = 0.05 nM) by selected monomeric carbamoylated guanidines. Data represent mean values ± SEM of 3 experiments performed in triplicate.

Table 6.3. Affinities of haloperidole, pramipexole and the monovalent carbamoylated guanidines **6.53**, **6.55**, **6.58**, **6.59**, **6.61**, **6.66**, **6.67** and **6.71** to the dopamine receptors hD_{2long}R and hD₃R, obtained from equilibrium competition binding studies.

	hD _{2long} R ^a		hD₃R ^b	
Compound	р <i>К</i> і	Ν	р <i>К</i> і	Ν
Haloperidol	9.60 ^{c 26} /	-	8.64 ^{c 26} /	-
паюренцої	9.3 ± 0.2	3	8.7 ± 0.1	3
Pramipexole	(p <i>K</i> _H : 7.40 / p <i>K</i> _L : 5.44) ^{d 25}	-	(p <i>K</i> _H : 9.06 / p <i>K</i> _L : 7.36) ^{d 25}	-
6.53	6.3 ± 0.1	3	7.07 ± 0.06	3
6.55	6.58 ± 0.03	3	7.36 ± 0.04	3
6.58	6.28 ± 0.08	3	7.19 ± 0.06	3
6.59	6.32 ± 0.08	3	6.88 ± 0.04	3
6.61	5.6 ± 0.2	3	5.89 ± 0.05	3
6.66	5.9 ± 0.1	3	5.9 ± 0.2	3
6.67	6.26 ±	1	5.3 ± 0.1	3
6.71	6.97 ± 0.06	3	7.6 ± 0.2	3

Determined by displacing [³H]N-methylspiperone (^ahD_{2long}R: $K_d = 0.0149$ nM, c = 0.05 nM or ^bhD₃R: $K_d = 0.0258$ nM, c = 0.05 nM) by increasing concentrations of the respective ligand at homogenates of ^aHEK293T-CRE-Luc-hD_{2long}R or ^bHEK293T-CRE-Luc-hD₃R cells. ^cdetermined on CHO cells stably expressing the hD_{2short}R or hD₃R. ^dHigh/low affinity binding determined on CHO cells stably expressing the hD_{2short}R or hD₃R. Data were analyzed by nonlinear regression and were best fitted to four-parameter sigmoidal concentration-response curves. Data shown are means ± SEM of N independent experiments, each performed in triplicate.

6.3 EXPERIMENTAL SECTION

6.3.1 General procedures

Chemicals and solvents were purchased from the following suppliers: Merck (Darmstadt, Germany), Acros Organics (Geel, Belgium), Fluka (Buchs, Swiss), Alfa Aesar (Karlsruhe, Germany), Sigma Aldrich (Munich, Germany) and TCI (Tokyo, Japan). All solvents were of analytical grade or distilled prior to use. Anhydrous solvents were stored over molecular sieve under protective gas. Deuterated solvents for NMR spectroscopy were purchased from Deutero (Kastellaun, Germany). For the preparation of buffers and HPLC eluents Millipore water was used throughout. Column chromatography was carried out using Merck silica gel 60 (0.040-0.063 mm). Automated flash chromatography was performed with a 971-FP flash-purification system (Agilent Technologies, Santa Clara, CA). Pre-packed columns (SuperFlash SF10-4 g, SF12-8 g, SF 15-12 g und SF15-24 g, Agilent Technologies, Santa Clara, CA) were used throughout. Reactions were monitored by thin layer chromatography (TLC) on Merck silica gel 60 F254 aluminium sheets, and compounds were detected with UV light at 254 nm and ninhydrin solution (0.8 g ninhydrin, 200 mL n-buthanol, 6 mL acetic acid). Melting points were determined with a B-540 apparatus (BÜCHI GmbH, Essen, Germany) and are uncorrected. IR spectra were measured on a NICOLET 380 FT-IR spectrophotometer (Thermo Electron Corporation, USA) or on a FTS 3000 MX spectrometer (Excalibur Series, Bio-Rad, Hercules, CA) equipped with an ATR unit (Specac Golden Gate Diamond Single Reflection ATR system). Nuclear Magnetic Resonance (¹H NMR and ¹³C NMR) spectra were recorded on a Bruker Avance-300 (7.05 T, ¹H: 300 MHz, ¹³C: 75.5 MHz), Avance-400 (9.40 T, ¹H: 400 MHz, ¹³C: 100.6 MHz), or Avance-600 (14.1 T; ¹H: 600 MHz, ¹³C: 150.9 MHz; cryogenic probe) NMR spectrometer (Bruker BioSpin, Karlsruhe, Germany). Chemical shifts are given in δ (ppm) relative to external standards. Multiplicities are specified with the following abbreviations: s (singlet), d (doublet), t (triplet), q (quartet), qui (quintet), m (multiplet), br s (broad signal), as well as combinations thereof. In certain cases 2D-NMR techniques (COSY, HSQC, HMBC and NOESY) were used to assign ¹H and ¹³C chemical shifts. High-resolution mass spectrometry (HRMS) was performed on an Agilent 6540 UHD Accurate-Mass Q-TOF LC/MS system (Agilent Technologies, Santa Clara, USA) using an ESI source. Preparative HPLC was performed with a system from Knauer (Berlin, Germany) consisting of two K-1800 pumps and a K-2001 detector. A Nucleodur 100-5 C18 (250 x 21 mm, 5 μm, Macherey-Nagel, Dueren, Germany), a Kinetex XB-C18 100A (250 x 21.2 mm, 5 µm, Phenomenex, Aschaffenburg, Germany) and a Interchim Puriflash PF15 C18 HQ (120 g, 15 µm, Interchim S. A., Montluçon, France) served as RPcolumns at a flow rate of either 15 mL/min (Kinetex and Phenomenex column) or 30 mL/min (Interchim column) at room temperature. A detection wavelength of 220 nm and mixtures of CH₃CN and 0.05-0.1% aq. TFA as mobile phase were used throughout. CH₃CN was removed from the eluates under reduced pressure (final pressure: 80 mbar) at 45 °C prior to lyophilisation (Christ alpha 2-4 LD lyophilisation apparatus equipped with a vacuubrand RZ 6 rotary vane vacuum pump). Analytical HPLC analysis was performed on a system from Meck Hitachi, composed of a D-6000 interface, a L-6200A pump, a AS2000A auto sampler and a L-4000 UV-VIS detector. A Kinetex XB-C18 100A (250 x 4.6 mm, 5 μ m, t_0 = 2.9 min, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of 0.05% TFA in CH₃CN (A) and 0.05% aq. TFA (B) were used as mobile phase. Helium degassing, room temperature, a flow rate of 0.8 mL/min and a detection wavelength of 220 nm were used throughout. Solutions for injection (concentration: 100-500 μ M) were either prepared from stock solution (10 mM in 20 mM aqueous HCl solution or 10 mM in a mixture of DMSO/ in 20 mM aqueous HCl solution 1:1) in a mixture of A and B corresponding to the initial eluent composition of the run, or as a one to one dilution of the eluate (preparative HPLC) with Millipore water. The following linear gradients were applied for analytical HPLC analysis: gradient 1: 0-30 min: A/B 5:95-80:20, 30-32 min: 80:20-95:5, 32-42 min: 95:5 or gradient 2: 0-30 min: A/B 10:90-80:20, 30-32 min: 80:20-95:5, 32-42 min: 3: 0-30 min: A/B 15:85-90:10, 30-35 min: 90:10. Microanalysis was performed on a Vario micro cube (Elementar, Langenselbold, Germany).

6.3.2 Experimental protocols and analytical data

The branched carboxylic acids (3-methyl 4-(4-methylphenyl)butanoic acid, 3-methyl 5-phenyl pentanoic acid, 3-cyclohexyl butanoic acid, 3-(4-prop-2-yl phenyl) butanoic acid, 4-methyl 3-phenyl pentanoic acid, 3-phenyl pentanoic acid, 4-cyclohexyl 3-methyl butanoic acid) were synthesized by Anja Kraus¹². The amine precursor 3-(1-trityl-1*H*-pyrazol-4-yl)prop-1-yl-amine was synthesized according to published protocols¹³.

2-(5-Oxohexyl)isoindoline-1,3-dione (6.1)^{1,16}

A solution of phthalimide (1.29 g, 8.74 mmol, 1 eq), 6-chlorohexan-2-one (2 g, 14.86 mmol, 1.7 eq) and K_2CO_3 (1.39 g, 10.05 mmol, 1.15 eq) in DMF (17 mL) were stirred at 80 °C for 24 h to obtain a white suspension. Ice cold water (60 mL) was added and the product was extracted with CHCl₃ (3 x 50 mL). The organic layers were combined and washed with brine (50 mL). The solvent was removed under reduced pressure. The crude product was dissolved in CHCl₃ (100 mL) and washed with water (2 x 50 mL) and brine (50 mL). The organic layer was dried over Na₂SO₄ and the solvent was removed under reduced pressure and the residue was purified by column chromatography (eluent: PE/EtOAc 90:10 - 70:30). Removal of the solvent *in vacuo* afforded the product as a white solid (1.5 g, 70%). Mp: 68.5 °C (Lit.¹⁶ mp: 73-75 °C). $R_f = 0.3$ (PE/EtOAc 7:3). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.51-1.67 (m, 4H), 2.07 (s, 3H), 2.43 (t, 2H, *J* 7.15 Hz), 3.63 (t, 2H, *J* 6.94 Hz), 7.63-7.67 (m, 2H), 7.75-7.80 (m, 2H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 20.8, 27.9, 29.9, 37.5, 42.9, 123.2, 132.1, 133.9, 168.4, 208.3. HRMS (ESI): $m/z [M+H]^+$ calcd. for C₁₄H₁₆NO₃⁺: 246.1125, found: 246.1128. C₁₄H₁₅NO₃ (245.28).

2-[3-(2-Amino-4-methylthiazol-5-yl)propyl]-1,3-dihydro-2H-isoindol-1,3-dione hydrobromide (6.3)^{1,16}

6.1 (5.00 g, 20.38 mmol, 1 eq) was dissolved in dioxane (100 mL). A solution of bromine (3.91 g, 24.46 mmol, 1.2 eq) in CH_2Cl_2 (60 mL) was added drop wise and the reaction mixture was stirred for 1 h at room temperature. Removal of the solvent *in vacuo* afforded **6.2** as a brown oil. The crude intermediate was dissolved in DMF (100 mL) and thiourea (1.55 g, 20.38 mmol, 1 eq) was added. The reaction mixture was stirred at 100 °C for 3 h and then over night at room temperature. The solvent was removed under reduced pressure and the residue was suspended

in EtOAc (100 mL). Subsequently, the precipitate was filtered off and washed with EtOAc (100 ml). Removal of residual solvent *in vacuo* afforded **6.3** as a beige solid (6.71 g, 86%). Mp: 211.0 °C (Lit.¹⁶ mp: 242 °C). $R_{\rm f}$ = 0.30 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, [D₆]DMSO) δ (ppm): 1.78-1.91 (m, 2H), 2.08-2.14 (m, 3H), 2.61-2.73 (m, 2H), 3.59-3.63 (m, 2H), 7.81-7.88 (m, 4H). HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₅H₁₆N₃O₂S⁺: 302.0958, found: 302.0963. C₁₅H₁₅N₃O₂S · HBr (301.09 + 80.91).

tert-Butyl 4-methyl-5-[3-(1,3-dioxo-1,3-dihydro-2*H*-isoindol-2-yl)propyl]thiazol-2-ylcarbamate (6.4)^{1,16}

6.3 (3.58 g, 9.40 mmol, 1 eq), di-*tert*-butyldicarbonate (2.46 g, 11.28 mmol, 1.2 eq), triethylamine (1.43 g, 14.09 mmol, 1.5 eq) and 4-(dimethylamino)-pyridine (105 mg, 0.94 mmol, 0.1 eq) were dissolved in CHCl₃ (40 ml). The mixture was stirred overnight at room temperature. The precipitate was filtered off. The product containing supernatant was washed with hydrochloric acid (0.25 M, 2 x 100 mL), brine (100 mL) and H₂O (100 mL). The organic layer was dried over Na₂SO₄ and the solvent was removed under reduced pressure. The residue was purified by automated flash chromatography (PE/EA 100:0 – 55:45 in 45 min). Removal of the solvent under reduced pressure afforded the product as yellow foam (750 mg, 20%). Mp: 81.5 °C (Lit.¹⁶ mp: 70-72 °C). *R*_f= 0.3 (PE/EA 60:40). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.51 (s, 9H), 1.95-2.02 (m, 2H), 2.22 (s, 3H), 2.71 (t, 2H, *J* 7.80 Hz), 3.74 (t, 2H, *J* 7.15 Hz), 7.69-7.72 (m, 2H), 7.83-7.85 (m, 2H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.4, 23.7, 28.3, 30.2, 37.5, 82.4, 123.1, 123.3, 132.1, 134.0, 141.5, 152.5, 157.6, 168.3. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₂₀H₂₄N₃O₄S⁺: 402.1482, found: 402.1489. C₂₀H₂₃N₃O₄S (401.48).

tert-Butyl 5-(3-aminopropyl)-4-methylthiazol-2-ylcarbamate (6.5)^{1,16}

To a suspension of **6.4** (710 mg, 1.77 mmol, 1 eq) in EtOH (7 ml) hydrazine-monohydrate (443 mg, 8.84 mmol, 5 eq) was added. The reaction mixture was stirred over night at room temperature. The precipitated phthalhydrazide (by-product) was filtered off. The product which was dissolved in the supernatant was purified by automated flash chromatography (CH₂Cl₂ / 1.75 N NH₃ in MeOH 100:0 – 90:10 in 35 min). Removal of the solvent *in vacuo* afforded the product as yellow oil (400 g, 83%). $R_{\rm f}$ = 0.56 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 80:20). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 1.46 (s, 9H), 1.54-1.61 (m, 2H), 2.11 (s, 3H), 2.55 (t, 2H, *J* 6.83 Hz), 2.63 (t, 2H, *J* 7.44 Hz). ¹³C-NMR (75 MHz, CDCl₃): δ (ppm) 14.5, 23.3, 28.3, 34.9, 41.2, 82.1, 123.8, 141.6, 152.8, 157.5. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₂H₂₂N₃O₂S⁺: 272.1427, found: 272.1433. C₁₂H₂₁N₃O₂S (271.38).

2-Amino-4-(3-phthalimidophenyl)thiazole hydrobromide (6.6)

To a solution of **3.4** (2 g, 7.54 mmol, 1 eq) in $CHCl_3$ (20 mL) was added HBr solution in acetic acid (45 % w/v, 1 mL) under stirring. Bromine (1.2 g, 7.54 mmol, 1 eq) in $CHCl_3$ (10 mL) was added drop wise. The reaction mixture was stirred for 1 h at room temperature. Removal of the solvent *in vacuo* afforded the **3.5** as a white solid, which was applied to the next step without further purification. The crude **3.5** was dissolved in hot CH_3CN (30 mL) and poured in a hot solution of

thiourea (574 mg, 7.54 mmol, 1 eq) in EtOH (30 mL). The reaction mixture was stirred under reflux for 3 h. Removal of the solvent *in vacuo* afforded a beige solid, which was recristallized in EtOAc (100 mL) and filtered through a Buchner funnel. **6.6** was afforded as a white solid (2.35 g, 78%). Mp: 269-279°C, decomposition. $R_f = 0.7$ (CH₂Cl₂/ 1.75 N NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 7.26 (s, 1H), 7.51-7.54 (m, 1H), 7.66 (t, 1H, *J* 7.90 Hz), 7.81-7.85 (m, 2H), 7.93-7.96 (m, 2H), 7.98-8.01 (m, 2H), 8.94 (br s, 3H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 104.3, 124.0, 125.3, 125.9, 128.7, 130.1, 130.6, 131.9, 133.0, 135.4, 139.7, 167.3, 170.6. HRMS: (ESI): m/z [M+H]⁺ calcd. for C₁₇H₁₂N₃O₂S⁺: 322.0645, found: 322.0650. C₁₇H₁₁N₃O₂S · HBr (321.35 + 80.91).

2-Amino-4-(3-aminophenyl)thiazole (6.7)²⁷

6.6 (1.69 g, 4.20 mmol, 1eq) was suspended in a mixture of concentrated hydrochloric acid (30 mL) and acetic acid (30 mL) and the reaction mixture was stirred under reflux overnight. The solvent was removed under reduced pressure and the residue was suspended in aqueous NaOH solution (0.03 M, 30 mL). The resulting precipitate (by-product: phthalic acid) was filtered through a Buchner funnel and washed with H₂O (20 mL). Aqueous layers were combined and part of the solvent was removed under reduced pressure. Aqueous NH₃ solution (25%, 5 mL) was added and the resulting precipitate was filtered off. The solid was washed with H₂O (40 mL) and the residual solvent was removed under reduced pressure. **6.7** was afforded as a yellow solid (560 mg, 60%). Mp: 177-178 °C. $R_f = 0.6$ (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). ¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 5.05 (br s, 2H), 6.43-6.47 (m, 1H), 6.77 (s, 1H), 6.91-7.03 (m, 5H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 101.1, 111.9, 113.5, 114.0, 129.3, 136.0, 149.1, 151.2, 168.3. HRMS: (ESI): m/z [M+H]⁺ calcd. for C₉H₁₀N₃S⁺: 192.0590, found: 192.0590. C₉H₉N₃S (191.25).

2-((1r, 4r)-4-hydroxycyclohexyl)isoindoline-1,3-dione (6.8)²⁸

K₂CO₃ (10.50 g, 75.97 mmol, 1.75 eq) was added to a solution of trans-4-aminocyclohexanol (5.00 g, 43.41 mmol, 1 eq) in H₂O (50 mL). Under stirring N-(ethoxycarbonyl)phthalimide (10.24 g, 49.92 mmol, 1.15 eq) was added and the reaction mixture was stirred for 30 min at room temperature. The resulting precipitate was filtered off and washed with H₂O (50 mL). Removal of the residual solvent *in vacuo* afforded the product as a beige solid (4.94 mg, 98%). %). Mp: 183-185 °C (Lit.²⁸ mp: 177-178 °C). R_f = 0.2 (PE/EtOAc 3:1). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 1.21-1.32 (m, 2H), 1.66-1.69 (m, 2H), 1.90-1.93 (m, 2H), 2.08-2.19 (m, 2H), 3.41-3.49 (m, 1H), 3.91-3.99 (m, 1H), 4.64 (d, 1H, *J* 4.26 Hz), 7.80-7.85 (m, 4H). ¹³C-NMR (100 MHz, [D₆]DMSO): δ (ppm) 27.8, 35.1, 49.9, 68.5, 123.4, 131.9, 134.8, 168.3. HRMS: (ESI): *m/z* [*M*+H]⁺ calcd. for C₁₄H₁₆NO₃⁺: 246.1125, found: 246.1126. C₁₄H₁₅NO₃ (245.28).

2-(4-oxocyclohexyl)isoindoline-1,3-dione (6.9)^{28,29}

6.8 (2.6 g, 10.61 mmol, 1 eq) was dissolved in anhydrous CH_2CI_2 (70 mL). Under Ar-atmosphere first two spoons of Celite[®] were added followed by pyridinium chlorochromate (5.58 g, 25.88

mmol, 2.44 eq). The reaction mixture was stirred for 3.5 h at room temperature. 2-Propanol (2 mL) was added and after stirring for additional 30 min the solvent was removed under reduced pressure. The residue was purified by automated flash chromatography (EtOAc 100:0 isocratic for 10 min). Removal of the solvent afforded the product as light beige solid (1.82 g, 71%). Mp: 144-145 °C. $R_{\rm f}$ = 0.4 (PE/EtOAc 1:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 2.02-2.10 (m, 2H), 2.48-2.54 (m, 4H), 2.65-2.79 (m, 2H), 4.57-4.68 (m, 1H), 7.69-7.75 (m, 2H), 7.80-7.87 (m, 2H). ¹³C-NMR (75 MHz, CDCl₃): δ (ppm) 28.6, 39.9, 48.3, 123.3, 131.8, 134.1, 168.1, 209.0. HRMS: (ESI): $m/z [M+H]^+$ calcd. for $C_{14}H_{14}NO_3^+$: 244.0968, found: 244.0973. $C_{14}H_{13}NO_3$ (243.26).

2-(2-Amino-4,5,6,7-tetrahydrobenzo[d]thiazol-6-yl)isoindoline-1,3-dione (6.11)^{11,29}

6.9 (1.93 g, 7.93 mmol, 1 eq) was dissolved in a mixture of dioxane (22 mL) and CH₂Cl₂ (14 mL). Bromine (1.40 g, 8.73 mmol, 1.1 eq) dissolved in CH₂Cl₂ (24 mL) was added drop wise and the reaction mixture was stirred for 1.5 h at room temperature. The solvent was removed under reduced pressure and the residue (crude **6.10**) was dissolved in DMF (60 mL). After the addition of thiourea (0.60 g, 7.93 mmol, 1 eq) the reaction mixture was stirred for 2 h at 100 °C. The solvent was removed under reduced pressure and the residue was suspended in EtOAc (50 mL). The precipitate was filtered off and the residual solvent was removed *in vacuo* to afford the product as a beige solid (2.77 g, 92%). Mp: >300 °C decomposition (Lit.¹¹ mp: 244-246 °C decomposition). $R_{\rm f}$ = 0.90 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). ¹H-NMR (400 MHz, [D₆]DMSO): δ (ppm) 1.99-2.05(m, 1H), 2.60-2.77 (m, 3H), 2.88-2.93 (m, 1H), 3.11-3.36 (m, 1H), 4.41-4.49 (m, 1H), 7.84.7.90 (m, 4H). HRMS: (ESI): m/z [M+H]⁺ calcd. for C₁₅H₁₄N₃O₂S⁺: 300.0801, found: 300.0817. C₁₅H₁₃N₃O₂S (299.35).

4,5,6,7-Tetrahydrobenzo[d]thiazole-2,6-diamine (6.12)²⁹

6.11 (200 mg, 0.53 mmol, 1 eq) was suspended in a mixture of hydrochloric acid (37%, w/v, 5 mL) and acetic acid (5mL). The reaction mixture was stirred over night at 100 °C. The solvent was removed under reduced pressure and the residue was dissolved in H₂O (5 mL). The pH value was adjusted to 6 by addition of NaOH solution (1 M, 2 mL). The aqueous layer was washed three times with CH₂Cl₂ in order to remove by-products. Then additional NaOH solution (5mL) was added until a pH value of 9 was reached. The aqueous layer was extracted with CH₂Cl₂ (10 mL) and six times with EtOAc (10 mL). In order to extract remaining product from the aqueous layer NaCl (1 spoon) was dissolved in the mixture and the layer was again extracted three times with EtOAc (100 mL). All organic layers were combined and dried over Na₂SO₄. Removal of the solvent *in vacuo* afforded the product as a yellow hygroscopic solid (65 mg, 73%). *R*_f= 0.13 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). ¹H-NMR (300 MHz, [D₆]DMSO): δ (ppm) 1.40-1.53 (m, 1H), 1.78-1.82 (m, 1H), 2.13-2.21 (m, 1H), 2.31-2.41 (m, 2H), 2.60-2.67 (m, 1H), 2.94-3.03 (m, 1H), 6.60 (br s, 2H). HRMS: (ESI): *m/z* [*M*+H]⁺ calcd. for C₇H₁₂N₃S⁺: 170.0746, found: 170.0747. C₇H₁₁N₃S (169.25).

General procedure for the synthesis of S-methylcarbamoyl thiourea derivatives (6.13-6.18) from isocyanates

The respective isocyanate (1 eq) and triethylamine (2.25 eq) were added to a solution of **3.32** (1.5 eq) in CH_2Cl_2 (17-20 mL). The reaction mixture was stirred overnight at room temperature. The organic layer was washed with three times with water (30 mL) and subsequently with brine (30 mL). The organic layer was dried over Na_2SO_4 and the solvent was removed under reduced pressure. The crude product was purified by flash chromatography.

S-Methyl-(N-(tertbutoxycarbonyl))-(N'-propylcarbamoyl)thiourea (6.13) *Claudia Honisch master thesis*

6.13 was prepared from propylisocyanate (149 mg, 1.75 mmol), **3.32** (500 mg, 2.63 mmol) and triethylamine (400 mg, 3.94 mmol) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min) and removal of the solvent *in vacuo* afforded the product as a white solid (0.31 g, 64%). Mp: 56.1-60.2 °C. $R_{\rm f}$ = 0.72 (PE/EtOAc 3:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 0.94 (t, 3H, *J* 7.4 Hz), 1.47 (s, 9H), 1.56 (m, 2H), 2.31 (s, 3H), 3.18 (q, 2H, *J* 6.9 Hz), 12.34 (s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 11.4, 14.3, 22.9, 28.0, 41.9, 82.6, 151.1, 161.9, 167.5. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₁H₂₂N₃O₃S⁺: 276.1376, found: 276.1379. C₁₁H₂₁N₃O₃S (275.37).

S-Methyl-(N-(tertbutoxycarbonyl))-(N'-hexylcarbamoyl)thiourea (6.14) *Claudia Honisch master thesis*

6.14 was prepared from hexylisocyanate (223 mg, 1.75 mmol), **3.32** (500 mg, 2.63 mmol) and triethylamine (400 mg, 3.94 mmol) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min) and removal of the solvent *in vacuo* afforded the product as a white solid (0.45 g, 81%). Mp: 56.4-59.2 °C. $R_{\rm f}$ =0.67 (PE/EtOAc 3:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 0.85-0.90 (m, 3H), 1.26-1.37 (m, 6H), 1.47 (s, 9H), 1.49-1.59 (m, 2H), 2.31 (s, 3H), 3.19-3.24 (m, 2H), 12.35 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.0, 14.3, 22.6, 26.6, 28.0, 29.7, 31.5, 40.2, 82.6, 151.1, 161.8, 167.7. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₄H₂₈N₃O₃S⁺: 318.1846, found: 318.1851. C₁₄H₂₇N₃O₃S (317.45).

S-Methyl-(N-(tertbutoxycarbonyl))-(N'-cyclohexylcarbamoyl)thiourea (6.15) *Claudia Honisch master thesis*

6.15 was prepared from cyclohexylisocyanate (219 mg, 1.75 mmol), **3.32** (500 mg, 2.63 mmol) and triethylamine (2.25 eq., 400 mg, 3.94 mmol) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min) and removal of the solvent *in vacuo* afforded the product as a white solid (0.47 g, 85%). Mp: 149.5-152.3 °C. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.12-1.26 (m, 3H), 1.31-1.42 (m, 2H), 1.47 (s, 9H), 1.59-1.66 (m, 1H), 1.71-1.76 (m, 2H), 1.93-1.97 (m, 2H), 2.34 (s, 3H), 3.53-3.63 (m, 1H), 12.43 (br , 1H). ¹³C-NMR (100 MHz, CDCl₃):

δ (ppm) 14.4, 24.9, 25.5, 28.0, 33.1, 49.1, 82.7, 151.0, 160.9, 167.6. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₄H₂₆N₃O₃S⁺: 316.1689, found: 316.1700. C₁₄H₂₅N₃O₃S (315.43).

S-Methyl-(N-(tert-butoxycarbonyl))-(N'-(phenylcarbamoyl)thiourea) (6.16) *Claudia Honisch master thesis*

6.16 was prepared from phenylisocyanate (210 mg, 1.75 mmol), **3.32** (500 mg, 2.63 mmol) and triethylamine (400 mg, 3.9 mmol) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-90:10 in 25 min) and removal of the solvent *in vacuo* afforded the product as a white solid (0.31 g, 57%). Mp: 126.3-129.6 °C. $R_{\rm f}$ = 0.66 (PE/EtOAc 3:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.50 (s, 9H), 2.40 (s, 3H), 7.08-7.12 (m, 1H), 7.31-7.35 (m, 2H), 7.42 (br s, 1H), 7.49-7.51 (m, 2H), 12.22 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.6, 28.0, 83.1, 119.4, 124.0, 129.1, 137.9, 151.0, 159.3, 169.2. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₄H₂₀N₃O₃S⁺: 310.1220, found: 310.1228. C₁₄H₁₉N₃O₃S (309.38)

S-Methyl-(N-(tertbutoxycarbonyl))-(N'-benzylcarbamoyl)thiourea (6.17) *Claudia Honisch master thesis*

6.17 was prepared from benzylisocyanate (234 mg, 1.75 mmol), **3.32** (500 mg, 2.63 mmol) and triethylamine (400 mg, 3.94 mmol) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min) and removal of the solvent *in vacuo* afforded the product as a white solid (0.45 g, 80%). Mp: 103.8-106.6 °C. $R_{\rm f}$ = 0.77 (PE/EtOAc 3:1). ¹H-NMR (400MHz, CDCl₃): δ (ppm) 1.49 (s, 9H), 2.34 (s, 3H), 4.43 (d, 2H, *J* 6.09 Hz), 6.09 (br s, 1H), 7.27-7.37 (m, 5H), 12.38 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.5, 28.0, 44.2, 83.0, 127.56, 127.64, 128.8, 138.2, 151.0, 155.3, 168.5. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₅H₂₂N₃O₃S⁺: 324.1376, found: 324.1384. C₁₅H₂₁N₃O₃S (323.41).

S-Methyl-(N-(tertbutoxycarbonyl))-(N'-(2-phenylethyl)carbamoyl)thiourea (6.18) *Claudia Honisch master thesis*

6.18 was prepared from phenethylisocyanate (258 mg, 1.75 mmol), **3.32** (500 mg, 2.63 mmol) and triethylamine (400 mg, 3.94 mmol) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 35 min) and removal of the solvent *in vacuo* afforded the product as a white solid (0.43 g, 73%). Mp: 101.8-103.7 °C. $R_{\rm f}$ = 0.73 (PE/EtOAc 3:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 1.49 (s, 9H), 2.29 (s, 3H), 2.86 (t, 2H, *J* 7.18 Hz) 3.47-3.52 (m, 2H), 7.17-7.34 (m, 5H), 12.32 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.4, 28.1, 35.9, 41.4, 82.8, 126.5, 128.6, 128.8, 138.8, 151.1, 161.7, 168.9. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₆H₂₄N₃O₃S⁺: 338.1533, found: 338.1546. C₁₆H₂₃N₃O₃S (337.44).

General procedure for the synthesis of S-methylcarbamoyl thiourea derivatives (6.19-6.25) from carbonic acids

The respective carbon acid (1 eq) was dissolved in CH_2CI_2 (2.5-4.5 mL) and cooled with an ice bath. DMF (25-45 µL) and oxalyl chloride (1.5 eq) were added under Ar-atmosphere. The reaction mixture was stirred for 10 min under cooling. The ice bath was removed and stirring was continued for another 15 min at room temperature. The solvent was carefully removed under reduced pressure (water bath temperature under 30 °C). The residue was dissolved in anhydrous acetone (2.5-4.5 mL) and added drop wise under cooling to an ice cold solution of sodium azide (2.4 eq) in H_2O (1-3 mL). The reaction mixture was stirred for 30 min under cooling. Brine (5-10 mL) was added and the acyl azide was extracted three times with CH₂Cl₂ (10 mL). The organic layers were combined and dried over Na₂SO₄. Molecular sieve was added and the solvent was partially removed under reduced pressure. The resulting yellow solution was stirred for 30 min under reflux conditions to afford the isocyanate. The solution was cooled to room temperature and 3.32 (1 eq) and triethylamine (5 eq) were added. The reaction mixture was stirred over night at room temperature. The molecular sieve was filtered off and the organic layer was washed three times with H_2O (10 mL) and three times with brine (10 mL). The organic layers were combined and dried over Na₂SO₄. The crude product was purified by either automated flash chromatography or column chromatography.

S-Methyl-(N-(tertbutoxycarbonyl))-(N'-[2-methyl 3-(4-methylphenyl)propyl]carbamoyl)thiourea (6.19)

6.19 was prepared from 3-methyl 4-(4-methylphenyl)butanoic acid (100 mg, 0.52 mmol, 1 eq), oxalyl chloride (99 mg, 0.75 mmol, 1.5 eq), sodium azide (85 mg, 1.30 mmol, 2.4 eq), **3.32** (100 mg, 0.52 mmol, 1 eq) and triethylamine (263 mg, 2.60 mmol, 5 eq) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 20 min) and removal of the solvent *in vacuo* afforded the product as colourless oil (140 mg, 71%). R_f = 0.80 (PE/EtOAc 3:1). ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 0.92 (d, 3H, *J* 6.5 Hz), 1.48 (s, 9H), 1.92-2.04 (m, 1H), 2.31-2.33 (m, 6H), 2.39-2.44 (m, 1H), 2.63-2.68 (m, 1H), 3.08-3.22 (m, 2H), 7.04-7.10 (m, 4H), 12.36 (br s, 1H). ¹³C-NMR (75 MHz, CDCl₃): δ (ppm) 14.3, 17.7, 21.0, 28.0, 35.7, 40.8, 46.0, 82.6, 128.9, 129.0, 135.5, 137.1, 151.2, 162.0, 167.2. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₉H₃₀N₃O₃S⁺: 380.2002, found: 380.2009. C₁₉H₂₉N₃O₃S (379.52).

S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-methyl 4-phenylbutyl)carbamoyl]thiourea (6.20)

6.20 was prepared from 3-methyl 5-phenyl pentanoic acid (150 mg, 0.78 mmol, 1 eq), oxalyl chloride (149 mg, 1.17 mmol, 1.5 eq), sodium azide (122 mg, 1.87 mmol, 2.4 eq), **3.32** (148 mg, 0.78 mmol, 1 eq) and triethylamine (395 mg, 3.90 mmol, 5 eq) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent *in vacuo* afforded the product as colourless oil (220 mg, 74%). $R_{\rm f}$ = 0.66 (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 1.01 (d, 3H, *J* 6.57 Hz), 1.43-1.51 (m, 10H), 1.62-1.78 (m, 2H), 2.35 (br s, 3H), 2.55-2.77 (m, 2H), 3.05-3.25(m, 2H), 7.15-7.31 (m, 5H), 12.40 (br s, 1H). ¹³C-NMR (100 MHz,

CDCl₃): δ (ppm) 14.4, 17.7, 28.2, 33.3, 33.4, 36.3, 46.1, 82.7, 125.9, 128.46, 128.49, 142.5, 151.3, 162.2, 167.4. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₉H₃₀N₃O₃S⁺: 380.2002, found: 380.2025. C₁₉H₂₉N₃O₃S (379.52).

S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-cyclohexyl propyl)carbamoyl]thiourea (6.21)

6.21 was prepared from 3-cyclohexyl butanoic acid (150 mg, 0.88 mmol, 1 eq), oxalyl chloride (168 mg, 1.32 mmol, 1.5 eq), sodium azide (137 mg, 2.11 mmol, 2.4 eq), **3.32** (167 mg, 0.88 mmol, 1 eq) and triethylamine (446 mg, 4.41 mmol, 5 eq) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent *in vacuo* afforded the product as colourless oil (210 mg, 67%). $R_{\rm f}$ = 0.81 (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 0.89 (d, 3H, *J* 6.91 Hz), 0.97-1.30 (m, 6H), 1.48 (s, 9H), 1.62-1.76 (m, 6H), 2.37 (br s, 3H), 2.99-3.09 (m, 1H), 3.22-3.30 (m, 1H), 12.43 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.4, 14.5, 26.67, 26.74, 26.8, 28.1, 28.6, 30.9, 38.7, 40.6, 44.1, 82.6, 151.3, 162.2, 167.2. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₇H₃₂N₃O₃S⁺: 358.2159, found: 358.2199. C₁₇H₃₁N₃O₃S (357.51).

S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-(4-prop-2-yl phenyl) propyl)carbamoyl]thiourea (6.22)

6.22 was prepared from 3-(4-prop-2-yl phenyl) butanoic acid (150 mg, 0.73 mmol, 1 eq), oxalyl chloride (138 mg, 1.09 mmol, 1.5 eq), sodium azide (114 mg, 1.75 mmol, 2.4 eq), **3.32** (138 mg, 0.73 mmol, 1 eq) and triethylamine (368 mg, 3.64 mmol, 5 eq) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent *in vacuo* afforded the product as colourless oil (220 mg, 77%). $R_{\rm f}$ = 0.74 (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 1.24-1.30 (m, 9H), 1.48 (s, 9H), 2.29 (br s, 3H), 2.85-2.99 (m, 2H), 3.24-3.33 (m, 1H), 3.44-3.53 (m, 1H), 7.12-7.21 (m, 4H), 12.37 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.4, 19.4, 24.1, 28.15, 28.23, 33.8, 39.6, 47.1, 82.7, 126.8, 127.2, 141.5, 147.3, 151.2, 162.1, 167.5. HRMS (ESI) m/z (M+H)⁺ calcd. for C₂₀H₃₂N₃O₃S⁺: 394.2159, found: 394.2159. C₂₀H₃₁N₃O₃S (393.55).

S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(3-methyl 2-phenyl butyl)carbamoyl]thiourea (6.23)

6.23 was prepared from 4-methyl 3-phenyl pentanoic acid (150 mg, 0.78 mmol, 1 eq), oxalyl chloride (149 mg, 1.17 mmol, 1.5 eq), sodium azide (122 mg, 1.87 mmol, 2.4 eq), **3.32** (148 mg, 0.78 mmol, 1 eq) and triethylamine (395 mg, 3.90 mmol, 5 eq) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent *in vacuo* afforded the product as colourless oil (130 mg, 44%). $R_{\rm f}$ = 0.81 (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 0.75 (d, 3H, *J* 6.71 Hz), 1.02 (d, 3H, *J* 6.71 Hz), 1.48 (s, 9H), 1.85-1.97 (m, 1H), 2.24 (s, 3H), 2.50-2.58 (m, 1H), 3.25-3.37 (m, 1H), 3.79-3.88 (m, 1H), 7.10-7.35 (m, 5H), 12.36 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 14.4, 20.5, 20.9, 28.0, 31.5, 43.1, 52.6, 82.8, 126.7, 128.48, 128.54, 141.6, 151.0, 161.4, 167.7. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₉H₃₀N₃O₃S⁺: 380.2002, found: 380.2034. C₁₉H₂₉N₃O₃S (379.52).

S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(2-phenyl butyl)carbamoyl]thiourea (6.24)

6.24 was prepared from 3-phenyl pentanoic acid (150 mg, 0.84 mmol, 1 eq), oxalyl chloride (160 mg, 1.26 mmol, 1.5 eq), sodium azide (131 mg, 2.02 mmol, 2.4 eq), **3.32** (160 mg, 0.84 mmol, 1 eq) and triethylamine (426 mg, 4.21 mmol, 5 eq) according to general procedure. Purification by automated flash chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent *in vacuo* afforded the product as colourless oil (190 mg, 62%). $R_{\rm f}$ = 0.71 (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 0.82 (t, 3H, *J* 7.37 Hz), 1.48-1.85 (m, 11H), 2.37 (br s, 3H), 2.66-2.76 (m, 1H), 3.25-3.33 (m, 1H), 3.56-3.65 (m, 1H), 7.15-7.36 (m, 5H), 12.53 (br s, 1H). ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 12.0, 14.4, 26.8, 28.2, 45.7, 47.8, 82.7, 126.8, 128.0, 128.8, 142.6, 151.3, 162.0, 167.5. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₈H₂₈N₃O₃S⁺: 366.1846, found: 366.1895. C₁₈H₂₇N₃O₃S (365.49).

S-Methyl-(N-(tertbutoxycarbonyl))-[N'-(3-cyclohexyl 2-methyl propyl)carbamoyl]thiourea (6.25)

6.25 was prepared from 4-cyclohexyl 3-methyl butanoic acid (150 mg, 0.76 mmol, 1 eq), oxalyl chloride (144 mg, 1.13 mmol, 1.5 eq), sodium azide (118 mg, 1.82 mmol, 2.4 eq), **3.32** (144 mg, 0.76 mmol, 1 eq) and triethylamine (383 mg, 3.78 mmol, 5 eq) according to general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) and removal of the solvent *in vacuo* afforded the product as colourless oil (110 mg, 38%). $R_{\rm f}$ = 0.72 (PE/EtOAc 3:1). ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 0.79-0.95 (m, 5H), 0.99-1.06 (m, 1H), 1.14-1.34 (m, 5H), 1.49-1.83 (m, 16H), 2.68 (br s, 2H), 2.97-3.24 (m, 2H), 13.08 (br s, 1H). ¹³C-NMR (75 MHz, CDCl₃): δ (ppm) 14.6, 17.9, 26.3, 26.4, 26.7, 28.0, 30.3, 32.9, 34.2, 34.8, 42.4, 46.6, 83.6, 150.9, 162.0, 168.2. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₈H₃₄N₃O₃S⁺: 372.2315, found: 372.2321. C₁₈H₃₃N₃O₃S (371.54).

General procedure for the synthesis of the N^G-carbamoylated guanidines

The respective S-methylcarbamoyl thiourea derivative **6.13-6.25** or **3.33** (1 eq) and the amine **6.5**, **6.7**, **6.12** (1-1.5 eq) or 3-(1-trityl-1*H*-pyrazol-4-yl)prop-1-yl-amine (3 eq) were dissolved in anhydrous CH_2Cl_2 (5-25 mL). Triethylamine (2.25-5 eq) and $HgCl_2$ (2-2.2 eq) were added under Aratmosphere. The reaction mixture was stirred over night at room temperature. The resulting suspension was filtered through Celite[®] in order to remove the mercury salt and the crude product was purified by either automated flash chromatography or column chromatography. Removal of the solvent *in vacuo* afforded the Boc-protected products **6.26-6.46**, **6.68** and **6.69**. Subsequently, the deprotection was performed by stirring the intermediate in a mixture of CH_2Cl_2 and TFA over night at room temperature. The solvent was removed *in vacuo* and the product was purified by preparative HPLC.

4-Methyl-5-(3-(3-(propylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.47) *Claudia Honisch master thesis*

6.47 was prepared from **6.13** (150 mg, 0.54 mmol, 1 eq), **6.5** (148 mg, 0.54 mmol, 1 eq), HgCl₂ (293 mg, 1.08 mmol, 2 eq) and triethylamine (164 mg, 1.62 mmol, 3 eq) dissolved in CH_2Cl_2 (25

mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min) afforded the Boc-protected intermediate 6.26 as a glassy colourless solid (250 mg, 93%). 230 mg (0.46 mmol) of 6.26 were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05% aq. TFA 15:85-37:63, $t_{\rm R}$ = 12.6 min) afforded the product as white fluffy solid (140 mg, 58%). Mp: 72.4-73.2 °C. R_f= 0.62 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). IR (KBr): 602.13 660.12 697.18 726.78 761.84 797.25 844.40 1144.09 1197.67 1242.49 1269.25 1391.01 1439.56 1467.73 1545.20 1697.45 2882.67 2972.53 3122.82 3288.70 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 99.23% ($t_{\rm R}$ =10.92 min, k = 2.8). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.85 (t, 3H, J 7.36 Hz), 1.42-1.48 (m, 2H), 1.72 (qui, 2H, J 7.23 Hz), 2.07 (s, 3H), 2.59 (t, 2H, J 7.47 Hz), 3.04-3.07 (m, 2H), 3.22-3.26 (m, 2H, interfering with the water signal), 7.49 (br s, 1H), 8.49 (br s, 2H), 8.87 (br s, 2H), 8.99 (br s, 1H), 10.22 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.1, 11.6, 22.0, 22.2, 28.9, 1C under solvent peak (38.7-40.3), 40.9, 116.3, 132.6, 153.7, 153.8, 167.6. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₂H₂₃N₆OS⁺: 299.1649, found: 299.1651. $C_{12}H_{22}N_6OS \cdot C_4H_2F_6O_4$ (298.16 + 228.05).

4-Methyl-5-(3-(3-(hexylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.48) *Claudia Honisch master thesis*

6.48 was prepared from 6.14 (150 mg, 0.47 mmol, 1 eq), 6.5 (128 mg, 0.47 mmol, 1.5 eq), HgCl₂ (257 mg, 0.95 mmol, 2 eq) and triethylamine (144 mg, 1.42 mmol, 2.25 eq) dissolved in CH₂Cl₂ (25 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min) afforded the Boc-protected intermediate 6.27 as a glassy colourless solid (90 mg, 35%). 85 mg (0.16 mmol) of 6.27 were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (10 mL) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05% aq. TFA 24:76-46:54, $t_{\rm R}$ = 13.0 min) afforded the product as white hygroscopic solid (50 mg, 56%). R_f= 0.56 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). RP-HPLC (220 nm, gradient 2): 99.1% (t_R =16.70 min, k = 4.8). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.84-0.86 (m, 3H), 1.24-1.28 (m, 6H), 1.39-1.43 (m, 2H), 1.71 (qui, 2H, J 7.27 Hz), 2.05 (s, 3H), 2.58 (t, 2H, J 7.44 Hz), 3.06-3.09 (m, 2H), 3.22-3.25 (m, 2H, interfering with the water signal), 6.57 (br s, 1H), 7.49 (br s, 1H), 8.50 (br s, 2H), 8.73 (br s, 2H), 9.00 (br s, 1H), 10.39 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.8, 13.8, 21.98, 22.03, 25.8, 28.8, 29.0, 30.8, 2Cs under solvent peak (38.7-40.3), 116.3, 133.5, 153.7, 153.8, 167.4. HRMS (ESI) m/z (M+H)⁺ calcd. for $C_{15}H_{29}N_6OS^{+}$: 341.2118, found: 341.2127. $C_{15}H_{28}N_6OS \cdot C_4H_2F_6O_4$ (340.49 + 228.05).

4-Methyl-5-(3-(3-(cyclohexylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.49) *Claudia Honisch master thesis*

6.49 was prepared from **6.15** (150 mg, 0.47 mmol, 1 eq), **6.5** (129 mg, 0.47 mmol, 1.5 eq), HgCl₂ (258 mg, 0.95 mmol, 2 eq) and triethylamine (144 mg, 1.43 mmol, 2.25 eq) dissolved in CH_2Cl_2 (25 mL) according to the general procedure. Purification by automated flash chromatography

(PE/EtOAc 100:0-85:15 in 30 min) afforded the Boc-protected intermediate **6.28** as a glassy colourless solid (0.14 g, 55%). 140 mg (0.26 mmol) of **6.28** were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05% aq. TFA 19:81-42:58, $t_R = 13.9$ min) afforded the product as white fluffy solid (99 mg, 67%). Mp: 69.6-71.8°C. R_f = 0.82 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). FT-ATR: 723, 798, 839, 895, 1129, 1178, 1315, 1435, 1543, 1654, 2858, 2937, 3086, 3280 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 99.10% ($t_R = 14.55$ min, k = 4.0). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 1.11-1.32 (m, 5H), 1.51-1.54 (m, 1H), 1.63-1.77 (m, 6H), 2.08 (s, 3H), 2.59 (t, 2H, *J* 7.51 Hz), 3.23 (q, 2H, *J* 6.25 Hz), 3.44-3.46 (m, 1H), 7.47 (br s, 1H), 8.47 (br s, 2H), 8.96 (br s, 1H), 9.12 (br s, 2H), 10.06 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.3, 21.9, 24.1, 24.9, 28.7, 32.0, 1C under solvent peak (38.7-40.3), 48.3, 116.3, 131.4, 152.8, 153.7, 167.9. HRMS (ESI) m/z (*M*+H)⁺ calcd. for C₁₅H₂₇N₆OS⁺: 339.1962, found: 339.1970. C₁₅H₂₆N₆OS · C₄H₂F₆O₄ (338.47 + 228.05).

4-Methyl-5-(3-(3-(phenylcarbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.50) *Claudia Honisch master thesis*

6.50 was prepared from 6.16 (100 mg, 0.32 mmol, 1 eq), 6.5 (88 mg, 0.32 mmol, 1.5 eq), HgCl₂ (175 mg, 0.65 mmol, 2 eq) and triethylamine (98 mg, 0.97 mmol, 2.25 eq) dissolved in CH₂Cl₂ (25 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min) afforded the Boc-protected intermediate 6.29 as a glassy colourless solid (110 mg, 64%). 100 mg (0.19 mmol) of 6.29 were dissolved in a mixture of TFA (5 mL) and CH₂Cl₂ (1 mL) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05% aq. TFA 19:81-42:58, $t_{\rm R}$ = 12.4 min) afforded the product as white hygroscopic solid (63 mg, 60%). R_f= 0.80 (CH₂Cl₂/ 1.75 N NH₃ in MeOH 9:1). FT-ATR: 693, 719, 753, 798, 835, 1126, 1178, 1316, 1446, 1498, 1551, 1640, 1595, 2363, 3094, 3276 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 98.98% ($t_{\rm R}$ = 14.75 min, k = 4.09). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 1.75 (qui, 2H, J 7.21 Hz), 2.09 (s, 3H), 2.61 (t, 2H, J 7.50 Hz), 3.28 (q, 2H, J 6.47 Hz), 7.08-7.12 (m, 1H), 7.32-7.36 (m, 2H), 7.43-7.45 (m, 2H), 8.56 (br s, 2H), 8.93 (br s, 1H), 9.05 (br s, 2H), 10.47 (br s, 1H). ¹³C-NMR (150 MHz, $[D_6]$ DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.4, 21.9, 28.7, 1C under solvent peak (38.7-40.3), 116.3, 119.6, 123.9, 129.0, 133.8, 137.5, 153.4 (2C), 167.8. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₅H₂₁N₆OS⁺: 333.1492, found: 333.1496. C₁₅H₂₀N₆OS · C₄H₂F₆O₄ (332.43 + 228.05).

4-Methyl-5-(3-(3-((phenylmethyl)carbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.51) *Claudia Honisch master thesis*

6.51 was prepared from **6.17** (150 mg, 0.464 mmol, 1 eq), **6.5** (126 mg, 0.464 mmol, 1.5 eq), HgCl₂ (252 mg, 0.928 mmol, 2 eq) and triethylamine (141 mg, 1.392 mmol, 2.25 eq) dissolved in CH₂Cl₂ (25 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min) afforded the Boc-protected intermediate **6.30**

as a glassy colourless solid (170 mg, 67%). 90 mg (0.16 mmol) of **6.30** were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred for 6 h at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05% aq. TFA 19:81-42:58, $t_{\rm R}$ = 12.2 min) afforded the product as white hygroscopic solid (49 mg, 52%). $R_{\rm f}$ = 0.73 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). FT-ATR: 723.1, 797.7, 834.9, 1129.4, 1177.8, 1252.4, 1431.3, 1543.1, 1640.0, 3090.0, 3276.3 cm⁻¹. RP-HPLC (gradient 2, 220 nm): 97.4% ($t_{\rm R}$ = 14.89 min, k =4.1). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 1.72 (qui, 2H, *J* 7.22 Hz), 2.07 (s, 3H), 2.58 (t, 2H, *J* 7.49 Hz), 3.23 (q, 2H, *J* 6.66 Hz), 4.30 (d, 2H, *J* 5.83 Hz), 7.25-7.28 (m, 3H), 7.32-7.35 (m, 2H), 7.98 (br s, 1H), 8.52 (br s, 2H), 9.01 (br s, 1H), 9.10 (br s, 2H), 10.35 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.3, 21.9, 28.7, 40.0, 42.7, 116.3, 127.1, 127.2, 128.4, 131.6, 138.6, 153.7, 153.8, 167.8. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₆H₂₃N₆OS⁺: 347.1649, found: 347.1653. C₁₆H₂₂N₆OS · C₄H₂F₆O₄ (346.45 + 228.05).

4-Methyl-5-(3-(3-((2-phenylethyl)carbamoyl)guanidino)propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.52) *Claudia Honisch master thesis*

6.52 was prepared from **6.18** (150 mg, 0.444 mmol, 1 eq), **6.5** (120 mg, 0.444 mmol, 1 eq), HgCl₂ (241 mg, 0.888 mmol, 2 eq) and triethylamine (135 mg, 1.332 mmol, 3 eq) dissolved in CH₂Cl₂ (25 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-85:15 in 30 min) afforded the Boc-protected intermediate **6.31** as a glassy colourless solid (190 mg, 76%). 190 mg (0.34 mmol) of **6.31** were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred for 6 h at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Interchim, gradient: 0-30 min: MeCN/0.05% aq. TFA 19:81-46:54, t_R = 13.2 min) afforded the product as white hygroscopic solid (11 mg, 53%). R_f = 0.89 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). RP-HPLC (220 nm, gradient 2): 98.6% (t_R = 15.0 min, k = 4.2) ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 1.71 (qui, 2H, *J* 7.19 Hz), 2.08 (s, 3H), 2.58-2.60 (m, 2H), 2.75 (t, 2H, *J* 7.24 Hz), 3.24 (br s, 2H), 3.32-3.33 (m, 2H), 7.19-7.22 (m, 3H), 7.28-7.30 (m, 2H), 7.60 (br s, 1H), 8.52 (br s, 2H), 9.04 (br , 3H), 10.29 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.5, 21.9, 28.8, 35.0, 40.0, 40.7, 116.3, 126.2, 128.4, 128.6, 131.7, 138.9, 153.6, 153.7, 167.7. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₇H₂₅N₆OS⁺: 361.1805, found: 361.1812. C₁₇H₂₄N₆OS · C₄H₂F₆O₄ (360.48 + 228.05).

4-Methyl-5-(3-[3-([2-methyl 3-(4-methylphenyl)propyl]carbamoyl)guanidino]propyl)-2aminothiazole bis(hydrotrifluoroacetate) (6.53)

6.53 was prepared from **6.19** (84 mg, 0.22 mmol, 1 eq), **6.5** (60 mg, 0.22 mmol, 1 eq), HgCl₂ (120 mg, 0.44 mmol, 2 eq) and triethylamine (67 mg, 0.66 mmol, 3 eq) dissolved in CH_2Cl_2 (5 mL) according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 - 1:1) afforded the Boc-protected intermediate **6.32** as a yellow oil (90 mg, 68%). 90 mg of **6.32** were dissolved in a mixture of TFA (1 mL) and CH_2Cl_2 (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 0:100-65:35, $t_R = 16.9$ min) afforded the product

as white hygroscopic solid (30.1 mg, 32%). $R_{\rm f}$ = 0.73 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 4:1). RP-HPLC (gradient 2, 220 nm): 97.9% ($t_{\rm R}$ = 20.21 min, k = 5.97). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.78 (d, 3H, *J* 6.69 Hz), 1.72 (qui, 2H, *J* 7.20 Hz), 1.82-1.88 (m, 1H), 2.07 (s, 3H), 2.25 (s, 3H),2.28-2.31 (m, 1H), 2.55-2.60 (m, 3H), 2.89-2.93 (m, 1H), 3.05-3.09 (m, 1H),3.23 (q, 2H, *J* 6.50 Hz), 7.03-7.07 (m, 4H), 7.53 (br s, 1H), 8.51 (br s, 2H), 8.99 (br s, 1H), 9.14 (br s, 2H), 10.49 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.3, 17.0, 20.6, 21.9, 28.8, 34.9, 2Cs under solvent peak (38.7-40.3), 44.6, 116.3, 128.76, 128.79, 131.6, 134.7, 137.0, 153.77, 153.78, 167.9. HRMS (ESI) m/z (M+H)⁺ calcd. for C₂₀H₃₁N₆OS⁺: 403.2275, found: 403.2277. C₂₀H₃₀N₆OS · C₄H₂F₆O₄ (402.56 + 228.05).

4-Methyl-5-(3-[3-([2-methyl 4-phenylbutyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.54)

6.54 was prepared from 6.20 (90 mg, 0.24 mmol, 1 eq), 6.5 (64 mg, 0.24 mmol, 1 eq), HgCl₂ (129 mg, 0.47 mmol, 2 eq) and triethylamine (72 mg, 0.71 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.33 as a colourless oil (90 mg, 63%). 90 mg (0.15 mmol) of 6.33 were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-70:30, $t_{\rm R}$ = 18.22 min) afforded the product as white hygroscopic solid (30.0 mg, 32%). Mp: 61-68 °C. $R_{\rm f}$ = 0.35 (CH₂Cl₂ / 1% NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 99.8% (t_R = 18.49 min, k = 5.38). ¹H-NMR (600 MHz, $[D_6]$ DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.89 (d, 3H, J 6.45 Hz), 1.33-1.39 (m, 1H), 1.56-1.63 (m, 2H), 1.71 (qui, 2H, J 7.21 Hz), 2.05 (s, 3H), 2.51-2.65 (m, 4H), 2.96-3.01 (m, 1H), 3.05-3.09 (m, 1H), 3.21-3.25 (m, 2H, interfering with the water signal), 7.14-7.19 (m, 3H), 7.24-7.26 (m, 2H), 7.51(br s, 1H), 8.49 (br s, 2H), 8.82 (br s, 2H), 8.98(br s, 1H), 10.32 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.7, 17.2, 22.0, 28.9, 32.4, 32.5, 35.6, 1C under solvent peak (38.7-40.3), 44.8, 116.3, 125.6, 128.20, 128.23, 132.9, 142.2, 153.7, 153.8, 167.5. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₇H₂₅N₆OS⁺: 403.2275, found: 403.2277. C₂₀H₃₀N₆OS · C₄H₂F₆O₄ (402.56 + 228.05).

4-Methyl-5-(3-[3-([2-cyclohexylpropyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.55)

6.55 was prepared from **6.21** (90 mg, 0.25 mmol, 1 eq), **6.5** (68 mg, 0.25 mmol, 1 eq), HgCl₂ (137 mg, 0.50 mmol, 2 eq) and triethylamine (76 mg, 0.76 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate **6.34** as a colourless oil (130 mg, 89%). 130 mg (0.22 mmol) of **6.34** were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-70:30, t_R = 18.8 min) afforded the product as white hygroscopic solid (80.0 mg, 59%). Mp: 51-57 °C. R_f = 0.34 (CH₂Cl₂/ 1.75 N NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 99.0% (t_R = 18.61 min, k = 5.42). ¹H-NMR (600

MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.78 (d, 3H, *J* 6.8 Hz), 0.90-1.22 (m, 6H), 1.43-1.47 (m, 1H), 1.55-1.61 (m, 3H), 1.67-1.74 (m, 4H), 2.07 (s, 3H), 2.58-2.60 (m, 2H), 2.90-2.94 (m, 1H), 3.10-3.14 (m, 1H), 3.22-3.25 (m, 2H), 7.44 (br s, 1H), 8.49 (br s, 2H), 8.98-9.07 (m, 3H), 10.34 (br s, 1H),13.55 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.4, 14.0, 22.0, 26.06, 26.16, 26.23, 27.9, 28.8, 30.3, 37.8, 2Cs under solvent peak (38.7-40.3), 42.9, 116.3, 131.8, 153.70, 153.74, 167.8. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₈H₃₃N₆OS⁺: 381.2431, found: 381.2436. C₁₈H₃₂N₆OS · C₄H₂F₆O₄ (380.56 + 228.05).

4-Methyl-5-(3-[3-([2-(4-prop-2-ylphenyl)propyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.56)

6.56 was prepared from 6.22 (90 mg, 0.23 mmol, 1 eq), 6.5 (62 mg, 0.23 mmol, 1 eq), HgCl₂ (124 mg, 0.46 mmol, 2 eq) and triethylamine (69 mg, 0.69 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.35 as a colourless oil (120 mg, 85%). 120 mg (0.19 mmol) of 6.35 were dissolved in a mixture of TFA (1 mL) and CH_2CI_2 (5 mL) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-70:30, $t_{\rm R}$ = 19.1 min) afforded the product as white hygroscopic solid (80.0 mg, 64%). Mp: 59-64 °C. R_f = 0.34 (CH₂Cl₂ / 1.5 N NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 96.3% ($t_{\rm R}$ = 19.33 min, k = 5.67). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 1.16-1.18 (m, 9H), 1.72 (qui, 2H, J 7.14 Hz), 2.08 (s, 3H), 2.58-2.61 (m, 2H), 2.82-2.89 (m, 2H), 3.22-3.25 (m, 4H), 7.14-7.18 (m, 4H), 7.38 (br s, 1H), 8.51 (br s, 2H), 9.01 (br s, 1H), 9.23 (br s, 2H), 10.36 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.2, 19.1, 21.9, 23.9, 28.7, 33.0, 38.6, 1C under solvent peak (38.7-40.3), 46.0, 116.3, 126.3, 127.0, 131.3, 141.5, 146.4, 153.68, 153.71, 168.0. HRMS (ESI) m/z $(M+H)^+$ calcd. for $C_{21}H_{33}N_6OS^+$: 417.2431, found: 417.2435. $C_{21}H_{32}N_6OS + C_4H_2F_6O_4$ (416.59 + 228.05).

4-Methyl-5-(3-[3-([3-methyl 2-phenylbutyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.57)

6.57 was prepared from **6.23** (90 mg, 0.24 mmol, 1 eq), **6.5** (64 mg, 0.24 mmol, 1 eq), HgCl₂ (129 mg, 0.47 mmol, 2 eq) and triethylamine (72 mg, 0.71 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate **6.36** as a colourless oil (110 mg, 77%). 110 mg (0.18 mmol) of **6.36** were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. The solvent was removed *in vacuo* and the residue was purified by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-70:30, $t_R = 17.9$ min). The resulting product (80 mg) showed an insufficient purity and was purified a second time by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 5:95-65:35, $t_R = 20.2$ min). Lyophilisation afforded the product as white hygroscopic solid (40.0 mg, 17%). Mp: 59-64 °C. $R_f = 0.34$ (CH₂Cl₂/ 1.5 N NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 98.6% ($t_R = 18.31$ min, k = 5.31). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.66 (d,

3H, *J* 6.73 Hz), 0.91 (d, 3H, *J* 6.72 Hz), 1.69 (qui, 2H, *J* 7.17 Hz), 1.83-1.88 (m, 1H), 2.06 (s, 3H), 2.52-2.57 (m, 3H), 3.19-3.20 (m, 2H), 3.31-3.36 (m, 1H), 3.56-3.60 (m, 1H), 7.11-7.16 (m, 3H), 7.19-7.22 (m, 1H), 7.28-7.30 (m, 2H), 8.45 (br s, 2H), 8.93 (br s, 1H), 9.16 (br s, 2H), 10.08 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.3, 20.0, 20.7, 21.9, 28.7, 30.4, 1C under solvent peak (38.7-40.3), 42.0, 51.5, 116.3, 126.4, 128.1, 128.4, 131.3, 141.4, 153.51, 153.54, 167.9. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₂₀H₃₁N₆OS⁺: 403.2275, found: 403.2282. C₂₀H₃₀N₆OS · C₄H₂F₆O₄ (402.56 + 228.05).

4-Methyl-5-(3-[3-([2-phenylbutyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.58)

6.58 was prepared from 6.24 (90 mg, 0.25 mmol, 1 eq), 6.5 (67 mg, 0.25 mmol, 1 eq), HgCl₂ (134 mg, 0.50 mmol, 2 eq) and triethylamine (75 mg, 0.74 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate 6.37 as a colourless oil (110 mg, 76%). 110 mg (0.19 mmol) of 6.37 were dissolved in a mixture of TFA (1 mL) and CH_2CI_2 (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-70:30, $t_{\rm R}$ = 16.1 min) afforded the product as white hygroscopic solid (80.0 mg, 69%). Mp: 55-60 °C. $R_{\rm f}$ = 0.34 (CH₂Cl₂ / 1.5 N NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 97.5% (t_R = 16.71 min, k = 4.76). ¹H-NMR (600 MHz, $[D_6]$ DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.71 (t, 3H, J 7.32 Hz), 1.46-1.54 (m, 1H), 1.64-1.73 (m, 3H), 2.06 (s, 3H), 2.56-2.58 (m, 2H), 2.62-2.67 (m, 1H), 3.20-3.29 (m, 3H), 3.34-3.38 (m, 1H), 7.18-7.21 (m, 3H), 7.28-7.31 (m, 3H), 8.50 (br s, 2H), 8.98 (br s, 1H), 9.08 (br s, 2H), 10.37 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.4, 11.7, 22.0, 25.8, 28.8, 1C under solvent peak (38.7-40.3), 44.6, 46.7, 116.3, 126.4, 127.7, 128.4, 132.0, 142.5, 153.67, 153.69, 167.8. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₉H₂₉N₆OS⁺: 389.2118, found: 389.2118. $C_{19}H_{28}N_6OS \cdot C_4H_2F_6O_4$ (388.53 + 228.05).

4-Methyl-5-(3-[3-([3-cyclohexyl 2-methylpropyl]carbamoyl)guanidino]propyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.59)

6.59 was prepared from **6.25** (80 mg, 0.22 mmol, 1 eq), **6.5** (59 mg, 0.22 mmol, 1 eq), HgCl₂ (117 mg, 0.43 mmol, 2 eq) and triethylamine (66 mg, 0.65 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by column chromatography (PE/EtOAc 3:1 isocratic) afforded the Boc-protected intermediate **6.38** as a colourless oil (140 mg, 97%). 140 mg (0.23 mmol) of **6.38** were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-70:30, *t*_R = 19.9 min) afforded the product as white hygroscopic solid (110.0 mg, 75%). Mp: 61-66 °C. *R*_f= 0.34 (CH₂Cl₂ / 1 N NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 99.7% (*t*_R = 20.10 min, *k* = 5.93). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 0.73-0.88 (m, 5H), 0.93-0.97 (m, 1H), 1.07-1.23 (m, 4H), 1.26-1.31 (m, 1H), 1.59-1.75 (m, 8H), 2.08 (s,3H), 2.58-2.61 (m, 2H), 2.84-2.89 (m, 1H), 3.02-3.06 (m, 1H), 3.23-3.26 (m, 2H), 7.47 (br s,1H), 8.51 (br s, 2H), 9.00-9.06 (m, 3H), 10.42 (br s, 2H), 9.00-9.06 (m, 2H), 7.47 (br s, 1H), 8.51 (br s, 2H), 9.00-9.06 (m, 2H), 10.42 (br s, 2H), 9.00-9.06 (m, 2H),

1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC, NOESY): δ (ppm) 11.4, 17.7, 22.0, 25.7, 25.8, 26.2, 28.8, 29.6, 32.5, 33.5, 34.2, 1C under solvent peak (38.7-40.3), 41.7, 45.3, 116.3, 131.9, 153.77, 153.78, 167.8. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₉H₃₅N₆OS ⁺: 395.2588, found: 395.2590. C₁₉H₃₄N₆OS · C₄H₂F₆O₄ (394.58 + 228.05).

4-(3-(3-(Propylcarbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.60)

6.60 was prepared from **6.13** (80 mg, 0.29 mmol, 1 eq), **6.7** (56 mg, 0.29 mmol, 1 eq), HgCl₂ (158 mg, 0.58 mmol, 2 eq) and triethylamine (88 mg, 0.87 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-50:50 in 25 min) afforded the Boc-protected intermediate **6.39** as a yellow oil (70 mg, 57%). 70 mg (0.17 mmol) of **6.39** were dissolved in a mixture of TFA (2 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-45:55, t_R = 10.3 min) afforded the product as white hygroscopic solid (40.0 mg, 44%). R_f = 0.10 (PE/EtOAc 2:1). RP-HPLC (gradient 1, 220 nm): 99.2% (t_R = 13.87 min, k = 3.78). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 0.86 (t, 3H, *J* 7.50 Hz), 1.43-1.49 (m, 2H), 3.06-3.10 (m, 2H), 7.15 (s, 1H), 7.23-7.24 (m, 1H), 7.49 (t, 1H, *J* 7.74 Hz), 7.60 (t, 1H, *J* 5.56 Hz), 7.72-7.73 (m, 1H), 7.78-7.80 (m, 1H), 8.62 (br s, 1H), 8.97 (br s, 1H), 10.15 (s, 1H), 10.74 (s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 11.2, 22.2, 41.0, 103.0, 123.0, 124.81, 124.84, 130.1, 134.0, 135.5, 147.0, 153.5, 153.6, 168.7. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₄H₁₉N₆OS⁺: 319.1336, found: 319.1346. C₁₄H₁₈N₆OS · C₄H₂F₆O₄ (318.40 + 228.05).

4-(3-(3-(Hexylcarbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.61)

6.61 was prepared from 6.14 (90 mg, 0.28 mmol, 1 eq), 6.7 (54 mg, 0.28 mmol, 1 eq), HgCl₂ (154 mg, 0.57 mmol, 2 eq) and triethylamine (86 mg, 0.85 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-55:45 in 25 min) afforded the Boc-protected intermediate 6.40 as a yellow solid (110 mg, 84%). 110 mg (0.24 mmol) of 6.40 were dissolved in a mixture of TFA (2 mL) and CH_2Cl_2 (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 20:80-70:30, $t_{\rm R}$ = 13.4 min) afforded the product as white hygroscopic solid (60.0 mg, 43%). R_f= 0.10 (PE/EtOAc 2:1). RP-HPLC (gradient 1, 220 nm): 99.2% ($t_{\rm R}$ = 19.25 min, k = 5.64). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 0.84-0.87 (m, 3H), 1.22-1.30 (m, 6H), 1.41-1.44 (m, 2H), 3.09-3.13 (m, 2H), 7.15 (s, 1H), 7.23-7.24 (m, 1H), 7.49 (t, 1H, J 7.93 Hz), 7.59 (t, 1H, J 5.50 Hz), 7.73 (m, 1H), 7.79 (d, 1H ,7.89 Hz), 8.65 (br s, 1H), 8.97 (br s, 1H), 10.26 (s, 1H), 10.80 (s, 1H). ¹³C-NMR (150 MHz, $[D_6]$ DMSO, COSY, HSQC, HMBC): δ (ppm) 13.9, 22.0, 25.9, 28.8, 30.9, 1C under solvent peak (38.7-40.3), 102.9, 122.9, 124.76, 124.80, 130.1, 134.0, 135.5, 147.0, 153.4, 153.6, 168.7. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₇H₂₅N₆OS⁺: 361.1805, found: 361.1813. C₁₇H₂₄N₆OS · $C_4H_2F_6O_4$ (360.48 + 228.05).

4-(3-(3-(Cyclohexylcarbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.62)

6.62 was prepared from 6.15 (90 mg, 0.29 mmol, 1 eq), 6.7 (55 mg, 0.29 mmol, 1 eq), HgCl₂ (155 mg, 0.57 mmol, 2 eq) and triethylamine (87 mg, 0.86 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-50:50 in 25 min) afforded the Boc-protected intermediate 6.41 as a yellow oil (90 mg, 69%). 90 mg (0.20 mmol) of 6.41 were dissolved in a mixture of TFA (2 mL) and CH_2Cl_2 (5 mL) and stirred over night at room temperature. Removal of the solvent in vacuo and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 20:80-50:50, $t_{\rm R}$ = 11.8 min) afforded the product as white hygroscopic solid (70 mg, 61%). Mp: 80-84 °C. R_f= 0.05 (PE/EtOAc 2:1). RP-HPLC (gradient 2, 220 nm): 99.5% ($t_{\rm R}$ = 16.34 min, k = 4.6). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.14-1.32 (m, 5H), 1.51-1.54 (m, 1H), 1.63-1.66 (m, 2H), 1.77-1.80 (m, 2H), 3.45-3.51 (m, 1H), 7.15 (s, 1H), 7.23-7.25 (m, 1H), 7.50 (t, 1H, J 7.91 Hz), 7.59 (d, 1H, J 7.58 Hz), 7.73 (m, 1H), 7.79-7.81 (m, 1H), 8.65 (br s, 1H), 8.96 (br s, 1H), 9.93 (s, 1H), 10.74 (s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 24.6, 25.4, 32.5, 48.9, 103.5, 123.6, 125.3, 125.5, 130.6, 134.4, 136.0, 147.4, 153.1, 154.1, 169.2. HRMS (ESI) m/z $(M+H)^{+}$ calcd. for $C_{17}H_{23}N_6OS^{+}$: 359.1649, found: 359.1649. $C_{17}H_{22}N_6OS + C_4H_2F_6O_4$ (358.46 + 228.05).

4-(3-(3-(Phenylcarbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.63)

6.63 was prepared from **6.16** (81 mg, 0.26 mmol, 1 eq), **6.7** (50 mg, 0.26 mmol, 1 eq), HgCl₂ (142 mg, 0.52 mmol, 2 eq) and triethylamine (79 mg, 0.78 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-50:50 in 30 min) afforded the Boc-protected intermediate **6.42** as a yellow oil (30 mg, 25%). 30 mg (0.07 mmol) of **6.42** were dissolved in a mixture of TFA (2 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-70:30, *t*_R = 13.0 min) afforded the product as white hygroscopic solid (27 mg, 71%). *R*_f= 0.4 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). RP-HPLC (gradient 1, 220 nm): 99.4% (*t*_R = 16.70 min, *k* = 4.76). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 7.10-7.14 (m, 2H), 7.26-7.28 (m, 1H), 7.34-7.37 (m, 2H), 7.45-7.46 (m, 2H), 7.51 (t, 1H, J 7.84 Hz), 7.77-7.78 (m, 1H), 7.81-7.83 (m, 1H), 8.76-8.92 (m, 3H), 9.92 (s, 1H), 10.10 (br s, 1H), 10.70 (s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 102.9, 119.5, 122.9, 124.0, 124.6, 124.8, 129.1, 130.1, 134.0, 136.1, 137.3, 148.0, 151.1, 153.3, 168.5. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₇H₁₇N₆OS⁺: 353.1179, found: 353.1180. C₁₇H₁₆N₆OS · C₄H₂F₆O₄ (352.41 + 228.05).

4-(3-(3-((Phenylmethyl)carbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.64)

6.64 was prepared from **6.17** (85 mg, 0.26 mmol, 1 eq), **6.7** (50 mg, 0.26 mmol, 1 eq), HgCl₂ (142 mg, 0.52 mmol, 2 eq) and triethylamine (79 mg, 0.78 mmol, 3 eq) dissolved in CH₂Cl₂ (8 mL)

according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-60:40 in 30 min) afforded the Boc-protected intermediate **6.43** as a yellow oil (110 mg, 90%). 100 mg (0.21 mmol) of **6.43** were dissolved in a mixture of TFA (1 mL) and CH₂Cl₂ (5 mL) and stirred for 12 h at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-70:30, t_R = 13.1 min) afforded the product as white hygroscopic solid (90 mg, 70%). R_f = 0.1 (PE/EtOAc 2:1). RP-HPLC (gradient 1, 220 nm): 99.1% (t_R = 16.57 min, k = 4.71). ¹H-NMR (600 MHz, MeOD, COSY, HSQC, HMBC): δ (ppm) 4.41 (s, 2H), 7.25-7.28 (m, 1H), 7.32-7.35 (m, 4H), 7.40-7.41 (m, 1H), 7.60 (t, 1H, *J* 7.81 Hz), 7.71-7.72 (m, 1H), 7.75-7.77 (m, 1H). ¹³C-NMR (150 MHz, MeOD, COSY, HSQC, HMBC): δ (ppm) 44.6, 104.7, 124.8, 127.1, 127.9, 128.5, 128.6, 129.7, 132.3, 133.3, 135.4, 139.4, 142.1, 155.3, 155.9, 172.6. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₈H₁₉N₆OS⁺: 367.1336, found: 367.1341. C₁₈H₁₈N₆OS · C₄H₂F₆O₄ (366.44 + 228.05).

4-(3-((2-Phenylethyl)carbamoyl)guanidino)phenyl)-2-aminothiazole bis(hydrotrifluoroacetate) (6.65)

6.65 was prepared from **6.18** (88 mg, 0.26 mmol, 1 eq), **6.7** (50 mg, 0.26 mmol, 1 eq), HgCl₂ (142 mg, 0.52 mmol, 2 eq) and triethylamine (79 mg, 0.78 mmol, 3 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-60:40 in 30 min) afforded the Boc-protected intermediate **6.44** as a yellow solid (80 mg, 63%). 80 mg (0.17 mmol) of **6.44** were dissolved in a mixture of TFA (2 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-70:30, *t*_R = 13.9 min) afforded the product as white hygroscopic solid (50 mg, 50%). *R*_f= 0.3 (CH₂Cl₂ / 1.75 N NH₃ in MeOH 9:1). RP-HPLC (gradient 1, 220 nm): 99.8% (*t*_R = 17.32 min, *k* = 4.97). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 2.77 (t, 2H, *J* 7.16 Hz), 3.36-3.39 (m, 2H), 7.13 (s, 1H), 7.79-7.80 (m, 1H), 8.62 (br s, 1H), 8.96 (br s, 1H), 10.06 (s, 1H), 10.67 (s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 34.9, 40.7, 102.9, 122.9, 124.7, 124.8, 126.3, 128.4, 128.7, 130.1, 133.9, 135.9, 138.8, 147.6, 153.40, 153.44, 168.5. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₉H₂₁N₆OS⁺: 381.1492, found: 381.1493. C₁₉H₂₀N₆OS · C₄H₂F₆O₄ (380.47 + 228.05).

6-(3-(Hexylcarbamoyl)guanidino)-2-amino-4,5,6,7-tetrahydrobenzo[d]thiazole bis(hydrotrifluoroacetate) (6.66)

6.66 was prepared from **6.14** (140 mg, 0.44 mmol, 1 eq), **6.12** (90 mg, 0.53 mmol, 1.2 eq), HgCl₂ (239 mg, 0.88 mmol, 2 eq) and triethylamine (134 mg, 1.32 mmol, 3 eq) dissolved in CH₂Cl₂ (11 mL) according to the general procedure. Purification by automated flash chromatography (CH₂Cl₂ /MeOH 100:0-85:15 in 30 min) afforded the Boc-protected intermediate **6.45** as a yellow solid (110 mg, 57%). 110 mg (0.25 mmol) of **6.45** were dissolved in a mixture of TFA (2 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-60:40, $t_{\rm R}$ = 12.0 min) afforded the product as white hygroscopic solid (46 mg, 33%). Mp: 121-

122 °C. $R_{\rm f}$ = 0.2 (CH₂Cl₂ / 1.7 N NH₃ in MeOH 9:1). RP-HPLC (gradient 1, 220 nm): 95.3% ($t_{\rm R}$ = 17.10 min, k = 4.90). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 0.84-0.86 (m, 3H), 1.24-1.29(m, 6H), 1.39-1.42 (m, 2H), 1.83-1.97 (m, 2H), 2.51-2.60 (m, 3H), 2.86-2.89 (m, 1H), 3.06-3.09 (m, 2H), 4.04-4.05 (m, 1H), 7.50 (s, 1H), 8.67 (br s, 4H), 9.07 (s,1H), 10.17 (s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 14.3, 21.7, 22.5, 26.3, 26.8, 28.4, 29.3, 31.3, 1C under solvent peak (38.7-40.3), 46.9, 112.0, 136.5, 153.7, 154.2, 168.9. HRMS (ESI) m/z (M+H)⁺ calcd. for C₁₅H₂₇N₆OS ⁺: 339.1962, found: 339.1962. C₁₅H₂₆N₆OS · C₄H₂F₆O₄ (338.47 + 228.05).

6-(3-((Phenylmethyl)carbamoyl)guanidino)-2-amino-4,5,6,7-tetrahydrobenzo[d]thiazole bis(hydrotrifluoroacetate) (6.67)

6.67 was prepared from **6.17** (192 mg, 0.59 mmol, 1 eq), **6.12** (110 mg, 0.65 mmol, 1.1 eq), HgCl₂ (320 mg, 1.18 mmol, 2 eq) and triethylamine (180 mg, 1.77 mmol, 3 eq) dissolved in CH₂Cl₂ (16 mL) according to the general procedure. Purification by automated flash chromatography (CH₂Cl₂/MeOH 100:0-85:15 in 30 min) afforded the Boc-protected intermediate **6.46** as a yellow solid (80 mg, 34%). 80 mg (0.18 mmol) of **6.46** were dissolved in a mixture of TFA (2 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-60:40, *t*_R = 11.2 min) afforded the product as white solid (16 mg, 16%). Mp: 140-145 °C. *R*_f= 0.2 (CH₂Cl₂/ 1.7 N NH₃ in MeOH 9:1). RP-HPLC (gradient 1, 220 nm): 97.0% (*t*_R = 14.40 min, *k* = 3.97). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.83-1.89 (m, 1H), 1.95-1.97 (m, 1H), 2.53-2.60 (m, 3H), 2.86-2.89 (m, 1H), 4.03-4.05 (m, 1H), 4.30 (d, 2H, *J* 5.70 Hz), 7.24-7.28 (m, 3H), 7.32-7.34 (m, 2H), 8.02 (br s, 1H), 8.74 (br s, 4H), 9.07 (br s, 1H), 10.36 (br s, 1H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 21.2, 26.3, 27.9, 42.7, 46.4, 111.5, 127.1, 127.2, 128.4, 138.6, 140.2, 153.2, 153.8, 168.5. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₁₆H₂₁N₆OS⁺: 345.1492, found: 345.1485. C₁₆H₂₀N₆OS · C₄H₂F₆O₄ (344.44 + 228.05).

1-(Amino[3-(2-aminothiazol-4-yl)phenylamino]methylene)-3-(6-[3-(amino[3-(2-aminothiazol-4-yl)phenylamino]methylene)ureido]hexyl)urea tetra(hydrotrifluoracetate) (6.70)

6.70 was prepared from **3.33** (130 mg, 0.24 mmol, 1 eq), **6.7** (136 mg, 0.71 mmol, 3 eq), HgCl₂ (143 mg, 0.53 mmol, 2.2 eq) and DIPEA (155 mg, 1.20 mmol, 5 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 80:20-20:80 in 30 min) afforded the Boc-protected intermediate **6.68** as a yellow foam (180 mg, 91%). 160 mg (0.19 mmol) of **6.68** were dissolved in a mixture of TFA (2 mL) and CH₂Cl₂ (5 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Kinetex, gradient: 0-30 min: MeCN/0.1% aq. TFA 15:85-45:55, *t*_R = 13.4 min) afforded the product as white solid (140 mg, 67%). Mp: 128-133 °C. *R*_f= 0.1 (PE/EtOAc 2:1). RP-HPLC (gradient 1, 220 nm): 97.3% (*t*_R = 14.99 min, *k* = 4.17). ¹H-NMR (600 MHz, MeOD, COSY, HSQC, HMBC): δ (ppm) 1.38-1.41 (m, 2H), 1.56-1.58 (m, 2H), 3.24 (t, 2H, *J* 7.03 Hz), 7.35-7.41 (m, 1H), 7.60 (t,1H, *J* 7.95 Hz), 7.71 (m, 1H), 7.75-7.76 (m, 1H). ¹³C-NMR (150 MHz, MeOD, COSY, HSQC, HMBC): δ (ppm) 27.4, 30.3, 40.7, 104.6, 124.7, 127.0, 127.9, 132.1, 133.2, 135.5, 142.0, 155.3, 155.9, 172.6. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₂₈H₃₅N₁₂O₂S₂⁺: 635.2442, found: 635.2443. C₂₈H₃₄A₁₂O₂S₂ · C₄H₂F₆O₄ (634.78 + 228.05).

6.71 was prepared from **3.33** (129 mg, 0.24 mmol, 1 eq), 3-(1-trityl-1*H*-pyrazol-4-yl)prop-1-yl-amine (260 mg, 0.71 mmol, 3 eq), HgCl₂ (143 mg, 0.53 mmol, 2.2 eq) and DIPEA (155 mg, 1.20 mmol, 5 eq) dissolved in CH₂Cl₂ (5 mL) according to the general procedure. Purification by automated flash chromatography (PE/EtOAc 100:0-70:30 in 20 min) afforded the Boc-protected intermediate **6.69** as a white foam (230 mg, 81%). 190 mg (0.16 mmol) of **6.69** were dissolved in a mixture of TFA (2 mL) and CH₂Cl₂ (10 mL) and stirred over night at room temperature. Removal of the solvent *in vacuo* and purification by preparative HPLC (column: Nucleodur, gradient: 0-30 min: MeCN/0.1% aq. TFA 10:90-55:45, *t*_R = 13.8 min) afforded the product as white solid (80 mg, 52%). Mp: 84-88 °C. *R*_f= 0.1 (CH₂Cl₂ / 1.7 N NH₃ in MeOH 9:1). RP-HPLC (gradient 2, 220 nm): 97.3% (*t*_R = 12.84 min, *k* = 3.43). ¹H-NMR (600 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 1.26 (br s, 4H), 1.41-1.43 (m, 4H), 1.76 (qui, 4H, *J* 7.25 Hz), 2.46 (t, 4H, *J* 7.60 Hz), 3.06-3.09 (m, 4H), 3.21-3.24 (m, 4H), 7.49 (s, 6H), 8.47 (br s, 4H), 9.00 (br s, 2H), 10.34 (br s, 2H). ¹³C-NMR (150 MHz, [D₆]DMSO, COSY, HSQC, HMBC): δ (ppm) 20.5, 25.9, 28.9, 29.2, 1C under solvent peak (38.7-40.3), 40.4, 118.7, 132.2, 153.78, 153.82. HRMS (ESI) *m/z* (*M*+H)⁺ calcd. for C₂₂H₃₉N₁₂O₂⁺: 503.3313, found: 503.3309. C₂₂H₃₈N₁₂O₂ · C₄H₂F₆O₄ (502.63 + 228.05).

Investigation of the chemical stability Claudia Honisch Master Thesis

The chemical stability of the compounds **6.49**, **6.50** and **6.52** as well as the acylguanidines UR-Bit22, UR-Bit23 and UR-Bit29 was investigated in PBS (pH = 7.4) at room temperature. The incubation was started by dilution of a 10 mM solution of the compound in aqueous HCl (20 mM) with PBS (pH 7.4), resulting in a 100 μ M solution. After 0 h, 72 h and 7 days a 20 μ L aliquot was added to 20 μ L of a mixture of aqueous TFA solution (1%)/H₂O_{millipore}/MeCN (6:3:1, v:v:v). A sample (20 μ L) was immediately analyzed by analytical HPLC. A system from Agilent Technologies, composed of a 1290 Infinity binary pump equipped with a degasser, a 1290 Infinity autosampler, a 1290 Infinity thermostated column compartment and a 1260 Infinity diode array was used. A Kinetex XB-C18 100A (100 x 3 mm, 2.6 μ m, Phenomenex, Aschaffenburg, Germany) served as RP-column. Mixtures of CH₃CN (A) and 0.05% aq. TFA (B) were used as mobile phase. Helium degassing, 25 °C, a flow rate of 0.5 mL/min and a detection wavelength of 220 nm were used throughout. The following linear gradient was applied for the HPLC analysis: 0-15 min: A/B 5:95-35:65, 15-18 min: 35:65-95:5, 18-23 min: 95:5 (isocratic).

6.3.3 Pharmacological Methods

Radioligand competition binding assay on Sf9 insect cell membranes

Preparation of the membranes of Sf9 insect cells expressing the $hH_2R-G_{s\alpha S}$ fusion protein or coexpressing the hH_1R + RGS4, the hH_3R + $G_{i\alpha 2}$ + $\beta_1\gamma_2$ or hH_4R + $G_{i\alpha 2}$ + $\beta_1\gamma_2$ proteins was described elsewhere.³⁰

Radioligand competition binding assays were performed as described previously with minor adjustments using the following radioligands: $[{}^{3}H]$ mepyramine (Hartmann Analytic, Braunschweig, Germany; specific activity = 20 Ci/mmol; hH₁R: K_{d} = 4.5 nM, c_{final} = 5 nM), $[{}^{3}H]$ UR-DE257¹⁴ (hH₂R: specific activity = 32.89 Ci/mmol, K_{d} = 12.2 nM, c_{final} = 20 nM), $[{}^{3}H]$ UR-PI294³¹ (specific activity = 93.3 Ci/mmol; hH₃R: K_{d} = 1.1 nM, c_{final} = 2 nM; hH₄R: K_{d} = 5.1 nM, c_{final} = 5 nM) and $[{}^{3}H]$ histamine (Hartmann Analytic, Braunschweig, Germany; specific activity = 25 Ci/mmol; hH₃R: K_{d} = 12.1 nM, c_{final} = 15 nM; hH₄R: K_{d} = 15.9 nM, c_{final} = 10 nM).

On the day of the experiment Sf9 membranes were thawed and sedimented by centrifugation at 13,000 rpm at 4 °C for 10 min. The membranes were resuspended in ice cold binding buffer (12.5 mM MgCl₂, 1mM EDTA and 75 mM Tris/HCl, pH 7.4; in the following referred to as BB) and adjusted to a protein concentration of 2-4 μ g/ μ L. 80 μ L BB containing 0.2% BSA and the respective radioligand, followed by 10 µL of the investigated ligands at various concentrations (dissolved in H₂O), were added to every well of a 96-well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with 0.3% polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with 200 µL of scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h. The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

Functional GTPyS assay on Sf9 insect cell membranes

GTP γ S assays were performed as described previously⁵ with minor modifications. [³⁵S]GTP γ S (specific activity = 1000 Ci/mmol) was purchased from Hartmann Analytic (Braunschweig, Germany). Sf9 membranes were prepared in the same manner as for radioligand competition binding and the protein concentration was adjusted to 0.5-1.5 µg/µL.

Agonist mode: 80 μ L of BB containing BSA (0.05% final), GDP (1 μ M final) and [³⁵S]GTP γ S (20 nCi final), followed by 10 μ L of the investigated ligands at various concentrations (dissolved in H₂O) were added to every well of a 96-well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Incubation was started by addition of the membrane suspension (10 μ L). The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters

(Whatman GF/C, Maidstone, UK) using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany).

Antagonist mode of the GTP γ S assay was performed in the same way as the agonist mode, but in the presence of the agonist histamine (1 μ M final).

Cell culture

The preparation of stably transfected HEK293T-hH₂R- β Arr2^{15,32} cells was described elsewhere.

Cells were cultivated at 37 °C in a water saturated atmosphere containing 5% CO₂. Dulbecco's Modified Eagle Medium, containing 4.5 g/L glucose, 3.7 g/L NaHCO₃, 110 mg/L sodium pyruvate (DMEM, Sigma-Aldrich Munich, Germany) and supplemented with 0.584 g/L L-glutamine (L-glutamine solution, Sigma-Aldrich Munich, Germany), 1% (v/v) penicillin-streptomycin (P/S, 10,000 U/mL, Sigma-Aldrich Munich, Germany), 10% (v/v) fetal calf serum (FCS, Biochrom GmbH, Merck, Berlin, Germany) were used as a culture medium. Additionally, 400 μ g/mL zeocin (InvivoGen, San Diego,USA) and 600 μ g/mL G418 were added to the culture medium of HEK293T-hH₂R- β Arr2 cells.

β-Arrestin2 recruitment assay

The β -Arrestin2 recruitment assays were performed as described previously for the H₁R using HEK293T-hH₂R- β Arr2 cells, stably expressing the hH₂R-ElucC and β Arr2-ElucNfusion constructs³².

One day prior to the experiment, HEK293T-hH₂R- β Arr2 cells were trypsinized and detached with DMEM medium (high glucose without phenol red (Sigma Aldrich, Munich, Germany) containing 1% (v/v) P/S and 5% (v/v) FCS. The cell suspension was adjusted to $1.1 \cdot 10^6$ cells/mL and 90 µl (100,000 cells/well) were seeded in every well of a sterile, luciferase assay compatible, F-bottom 96-well plate (Cellstar®, Greiner Bio-One, Kremsmünster, Österreich). The cells were cultivated at 37 °C overnight in a water saturated atmosphere containing 5% CO₂. The investigated ligands were added at increasing concentrations (10 µL), and the plate was incubated at 25 °C for 60 min under shaking. 50 µL of the medium were removed, and 50 µL of Bright-Glo reagent (Promega, Madison, USA) were added. Bioluminescence was immediately measured for 1 s per well using a GENios Pro microplate reader (Tecan, Salzburg, Austria).

Radioligand competition binding assay on homogenates from HEK293T cells expressing the $hD_{2long}R$ or $hD_{3}R$

The preparation of the cells (HEK293T-CRE-Luc- $hD_{2long}R$ and HEK293T-CRE-Luc- $hD_{3}R$ cells), their cultivation, the preparation of the homogenates and radioligand competition binding assays were performed by Lisa Foster as a part of her ongoing dissertation.

Radioligand competition binding assays were performed as described for Sf9 cell membranes with some adjustments using the radioligand [³H]N-methylspiperone (specific activity of 77

Ci/mmol, Novandi Chemistry AB, Södertälje, Sweden; $hD_{2long}R$: $K_d = 0.014$ nM, $c_{final} = 0.05$ nM; hD_3R : $K_d = 0.026$ nM, $c_{final} = 0.05$ nM).

Cell lines expressing the human $D_{2long}R$ and D_3R were grown in 150 mm dishes to 80-90% confluency. Cells were rinsed with ice-cold PBS⁻ and scraped from the dishes using a cell scraper in the presence of harvest buffer (10 mM Tris·HCl, 0.5 mM EDTA, 5.5 mM KCl, 140 mM NaCl; pH 7.4) supplemented with protease inhibitors. After centrifugation (500g, 5 min), the $D_{2long}R$ expressing cells were resuspended in homogenate buffer (50 mM Tris·HCl, 5 mM EDTA, 1.5 mM CaCl₂, 5 mM MgCl₂, 5 mM KCl, 120 mM NaCl; pH 7.4) and the D_3R expressing cells were resuspended in Tris·HCl, 5 mM MgSO₄; pH 7.4) and stored at -80°C. After thawing, the cells were resuspended in homogenate buffer or Tris-MgSO₄ buffer respectively and homogenized using an Ultraturrax (5 times for 5 s on ice). The homogenate was centrifuged (50 000 *g*, 6 °C, 15 min), the pellet resuspended in binding buffer (50 mM Tris·HCl, pH, 7.4, containing 1mM EDTA, 5 mM MgCl₂, 100 µg/mL bacitracin;) and homogenized using a syringe and needle. The homogenate was stored in small aliquots at -80°C.

For radioligand binding, to 80 μ L of the homogenate suspension, 10 μ L of radioligand and 10 μ L of the ligand at various concentrations (dissolved in binding buffer), were added to every well of a 96-well plate (PP microplates 96 well, Greiner Bio-One, Frickenhausen, Germany). Unspecific binding was determined using (+)-butaclamol at a final concentration of 2 μ M, instead of the ligand. The plates were shaken for 60 min at room temperature in the dark. Subsequently, bound radioligand was separated from free radioligand by filtration through glass microfiber filters (Whatman GF/C, Maidstone, UK), treated with 0.3% polyethylenimine (PEI), using a 96-well Brandel harvester (Brandel Inc., Unterföhring, Germany). The punched out filter pieces were transferred into clear, flexible 96-well PET microplate (round bottom, 1450-401, Perkin Elmer, Rodgau, Germany). Each well was supplemented with 200 μ L of scintillation cocktail (Rotiscint Eco plus, Roth, Karlsruhe, Germany) and incubated in the dark for at least 4 h. The radioactivity was measured with a MicroBeta2 1450 scintillation counter (Perkin Elmer, Rodgau, Germany).

6.3.4 Data analysis

Retention factors k were calculated according to $k = (t_{R} - t_0)/t_0$ (t_0 = dead time).

Total binding data from radioligand competition binding experiments were plotted against log(concentration competitor) and analyzed by a four-parameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism Software 5.0, San Diego, CA), followed by normalization (100% = "top" of the four-parameter logistic fit, 0% = unspecifically bound radioligand/ fluorescent ligand determined in the presence of famotidine at 100 μ M). Normalized data from competition binding experiments was again analyzed by a four-parameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism) and obtained p/C₅₀ values were converted into pK_i values according to the Cheng-Prusoff equation³³.

Data of the GTP γ S assay (agonist mode) were processed by plotting the corrected counts per minute (ccpm) against log(concentration). In most cases data analysis by bell-shaped fit was ambiguous due to lack of data points at high concentrations (>100 μ M). The data points at high concentrations (>100 μ M). The data points at high concentrations (>100 μ M), where the signal decreased again, were excluded in the analysis by a four parameter logistic equation (GraphPad Software). The concentration response curve was normalized (0% = water value (basal activity), 100% = "top" histamine equation) and again analyzed by a four-parameter logistic equation (log(agonist) vs. response – variable slope, GraphPad Prism).

Data from the GTP_YS assay (antagonist mode) were processed by plotting the ccpm against log(concentration) and analysis by a four parameter logistic equation (GraphPad Prism), followed by normalization (100% = "top" of the four-parameter logistic fit, 0% = unspecifically bound [³⁵S]GTP_YS (ccpm) determined in the presence of famotidine at 100 μ M) and analysis by four-parameter logistic equation (log(inhibitor) vs response – variable slope, GraphPad Prism). p/C₅₀ values were converted into pK_B values according to the Cheng-Prusoff equation³³.

Data of the beta-Arrestin2 recruitment assay (agonist mode) were processed by plotting the luminescence (RLU) against log(concentration). In all cases data analysis by bell-shaped fit was ambiguous due to lack of data points at high concentrations (>300 μ M). The data points at high concentrations (>10-100 μ M), where the signal decreased again, were excluded in the analysis by a four parameter logistic equation (GraphPad Prism). The concentration response curve was normalized (0% = water value (basal activity), 100% = "top" histamine equation) and again analyzed by a four-parameter logistic equation (log(agonist) vs. response – variable slope, GraphPad Prism).

6.4 SUMMARY AND CONCLUSION

The bioisosteric replacement of the acylguanidine moiety in the monomeric N^G-acylated amino(methyl)thiazolylpropylguanidines by a carbamoylguanidine moiety resulted in ligands with retained or even improved hH₂R potency with pEC₅₀ values of 6.3-7.76 (ligands 6.47-6.59). A variety of aliphatic and aromatic residues were well tolerated. Compounds containing an amino(methyl)thiazolyl propyl moiety achieved up to 80 fold the potency of histamine and partial to full agonistic activities with more or less pronounced bell-shaped concentration-response curves. Additionally, the ligands 6.53, 6.55, 6.58 and 6.59 were investigated for dopamine receptor affinity (hD_{2long}R and hD₃R). These ligands showed affinity to the hD_{2long}R (pK_i : 6.28-6.58) and hD_3R (pK_i: 6.88-7.36). There was a clear preference for the hH_2R over the $hD_{2long}R$. But only ligand 6.59 showed a clear preference towards the hH_2R over the hD_3R and the ligands 6.53, 6.55 and 6.58 bound non-selective to both receptors. Incorporation of the less flexible 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety, derived from pramipexole, resulted in partial (6.67, Emax: 0.53, pEC₅₀: 6.7) or in weak partial agonism (6.66, Emax: 0.16, pEC₅₀: 6.83) and bell-shaped concentration-response curves. Interestingly, these two ligands showed a decreased hD_{2long}R and hD₃R affinity compared to pramipexole. Incorporation of an aminothiazolylphenyl moiety (compounds 6.60-6.65) resulted in weak antagonism at the hH₂R (pK_h : 4.8-6.14). Replacement of the amino(methyl)thiazolepropyl moiety of UR-NK22 with either a (aminothiazolyl)phenyl (6.70) or a pyrazolylpropyl molety (6.71) resulted in strongly decreased hH_2R affinity and antagonistic or partial agonistic activity.

All full or partial agonists identified in the GTP γ S assay showed a lower potency and efficacy in the β Arrestin2 recruitment assay. The ligands **6.53-6.59** also showed in the β Arrestin2 recruitment assay more or less pronounced bell-shaped concentration-response curves. While in case of the GTP γ S assay plausible explanations would be binding through multiple binding sites or direct interaction with the G_{sas} subunit of the fusion protein, in the β Arrestin2 recruitment assay also cytotoxicity could be the reason.

High affinity ligands like **6.48**, **6.51** and **6.55** were highly selective for H_2R over the other subtypes (Figure 6.11). The investigation of the binding affinity of aminothiazole containing ligands like **6.55** to the $hD_{2long}R$ and the hD_3R showed that achieving selectivity over these dopamine receptors might be a challenge. Notwithstanding, the monomeric N^G-carbamoylated amino(methyl)thiazolyl propylguanidines represent a good alternative to monomeric acylguanidines and dimeric ligands with high stability against hydrolytic cleavage. However, according to the current state of knowledge this is just speculation.

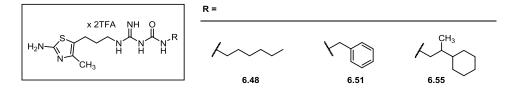


Figure 6.11. N^G-Carbamoylated 3-(2-amino-4-methylthiazol-5-yl)propylguanidines **6.48**, **6.51** and **6.55** as high potency H₂R agonists.

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Chapter 7

SUMMARY

The H₂R, an aminergic GPCR, is one of four receptor subtypes (H₁R, H₂R, H₃R, H₄R) which mediate the action of the biogenic amine histamine. Activation of H₂R results e. g. in gastric acid secretion,^{1,2} positive inotropic and chronotropic effects³. In humans, the H₂R is located on parietal cells in the stomach,² in the brain,^{4,5} on neutrophiles and eosinophiles⁶ as well as on smooth muscle cells⁷. However, the (patho-) physical role of the H₂R, especially in the brain, is still far from being understood. Therefore this work aimed at the development of selective high affinity molecular tools for the H₂R, including agonists, antagonists and radiolabeled as well as fluorescent H₂R ligands.

The number of high affinity tritiated radioligands for the H₂R is very limited. Guanidinothiazole containing ligands such as famotidine or ICI127032 represent a class of surmountable H₂R antagonists.^{8,9} The combination of the 2-guanidino-4-[(2-aminoethyl)thiomethyl]thiazole structure derived from famotidine or the guanidino-4-(3-aminophenyl)thiazole structure derived from ICI127032 with a derivatized squaramide or a cyanoguanidine moiety ("urea equivalent") led to propionylated high affinity H₂R antagonists. N-[8-(2-[2-(2-guanidinothiazol-4ylmethylthio)ethylamino]-3,4-dioxocyclobut-1-ene-1-ylamino)octyl]propionamide (3.25) showed a p K_i value of 7.65 at the hH₂R and selectivity over the other subtypes (no affinity at the hH₁R, hH₃R: pK_i value of 5.3 and hH₄R: pK_i value of 4.4). The radiolabled form [³H]**3.25** bound in a saturable manner (K_d values of 15-22 nM) to membrane preparations of Sf9 cells and intact Hek cells both recombinantly expressing hH₂Rs. Although a part of [³H]**3.25** bound in (pseudo)irreversible manner (plateau at 23%), the kinetic K_d value of 26 nM was comparable to that determined at equilibrium, and the radioligand ³H]**3.25** was completely displacable by histamine, famotidine and ICI127032. However, the radioligand showed a radiochemical purity of only 87% and low stability in stock solution (radiochemical purity: <45% after 15 month). Nevertheless, [³H]**3.25** can be a valuable molecular tool provided that purity and stability under storage conditions are improvable.

Aminopotentidine and it's derivatives are reported as high affinity H_2R antagonists. Iodination in 3 position of the 4-aminobenzoic acid amide moiety results in an enormous gain in affinity (iodoaminopotentidine).^{4,5} Aminopotentidine and its analogs with different substituents (e.g. iodine, bromine, chlorine, trifluoromethyl) in position 3, were prepared and propionylated. Within the series of propionylated derivatives the brominated ligand (4.37) and the iodinated ligand (4.38) showed the highest hH₂R affinities (pK_i values of 8.5 and 8.18) along with excellent selectivities over the hH₃R (6900- or 2500-fold). In general, for radiosynthesis an excess of precursor compared to radioactive labeling reagent is used. However, an adjustment of the reaction conditions for radiosynthesis might be challenging due to the necessary high excess of the "cold" labeling reagent propionic chloride and the low yields in the "cold" reaction. A test reaction under radiosynthesis conditions failed. To overcome this problem, a series of aminopotentidine derivatives containing a functionalized (propionylated, acetylated or methylated) aminomethyl substituent in 4-position of the aromatic ring was prepared. The dimethylated 3-bromo substituted ligand 4.50 showed the highest affinity within the series with a p K_i value of 7.54. The synthesis of radiolabeled **4.50** is accessible by dimethylation of **4.46** with [³H]methyl iodide.

Fluorescent ligands have become an attractive alternative to radioligands for the investigation of ligand-receptor interactions. Besides advantages with respect to safety issues and waste disposal,

fluorescent ligands are a prerequisite for the application of a plethora of optical techniques (confocal microscopy, FRET,¹⁰ FRAP,¹¹ TIRF,¹² high content imaging,¹³ fluorescence polarization¹⁴). In order to expand the range of applications and avoid the high cellular autofluorescence, fluorescent ligands labeled with red-emitting fluorophores (emission wavelength > 600 nm) are required. Recently, a series of fluorescent ligands with a piperidinomethylphenoxypropylamino (potentidine) pharmacophore was reported.¹⁵ The most promising ligands within this series were the squaramide-type ligands UR-DE229 and UR-DE56 which contained pyridinium or a cyanine fluorophore. Aiming at fluorescent high affinity H_2R antagonists with improved optical and physicochemical properties to gain access to a wide range of potential applications, the fluorescent labeled antagonists UR-DE229 (5.12) and UR-DE56 (5.18) were prepared and investigated in different assay systems (radioligand competition binding assay, GTPyS binding assay, flow cytometric binding assay and high content imaging). Furthermore, a small library of fluorescent ligands was synthesized for the exploration of the impact of length of the alkyl linker and the net charge of the fluorophores by coupling the positively charged pyrilium dye (Py-5) or differently charged cyanine dyes with amine precursors by linkers, differing in length (number of atoms). The highest affinities to the hH₂R (pK_i values > 7.0) in radioligand competition assays were obtained in case of the pyridinium labeled ligands 5.12-5.14 and the cyanine labeled ligands **5.16** (positively charged fluorophore, net charge: 2^+) and **5.18** (electroneutral fluorophore, net charge: 1^+). While the linker length (4-7 carbon atoms) had no significant influence on the H_2R affinity in case of the pyridinium ligands, the cyanine ligands with a hexyl linker (5.16 and 5.18) showed an increased hH₂R affinity compared analogs, containing a butyl linker. Despite the low selectivity towards the hH₃R the investigated fluorescent ligands proved to be useful tools for binding studies using different techniques (flow cytometry, high content imaging and confocal microscopy), when genetically engineered cells, expressing the H_2R were used. Investigated ligands bound in a saturable manner to the hH₂R (flow cytometry and high content imaging) and the determined K_d values were in good agreement with the corresponding K_i values from radiolligand binding experiments. K_{d (kin)} values calculated from kinetic experiments (flow cytometry or high content imaging) were consistent with the K_d values determined in saturation binding experiments even though they showed an incomplete dissociation (insurmountable antagonism). High content imaging and confocal microscopy showed that residual bound ligand was still located in the cell membrane. Nonetheless, the fluorescent ligands 5.14 and 5.18 also proved to be useful for the determination of binding affinities of unlabeled ligands in competition binding assays (flow cytometry and high content imaging).

 N^{G} -acylated amino(methyl)thiazolepropylguanidines are reported as a class of potent and selective histamine H₂R agonists, but the acylguanidine group is prone to hydrolytic cleavage upon storage in aqueous solution^{16,17}. A bioisosteric approach, replacing the N^G-acylguanidine structure with a N^G-carbamoylguanidine, led in many cases to more stable compounds.¹⁷⁻¹⁹ For exploration of the structure-activity (H_2R) and the structure-selectivity relationships of this class of compounds, in addition to dimeric ligands, a series of carbamoylguanidines with various aminothiazole-based substructures, i.e., the 3-(2-amino-4-methylthiazol-5-yl)propyl moiety, a conformationally constrained (aminothiazolyl)phenyl and а 2-amino-4,5,6,7tetrahydrobenzothiazol-6-yl portion were prepared. Compounds containing the conformationally constrained aminothiazolylphenyl moiety were antagonists with only a weak affinity to the hH₂R and compounds containing the less flexible 2-amino-4,5,6,7-tetrahydrobenzothiazol-6-yl moiety, derived from pramipexole, showed partial (6.67, α : 0.53, pEC₅₀ value: 6.7) or weak partial agonism (6.66, α : 0.16, pEC₅₀ value: 6.83) and bell-shaped concentration-response curves. Amino(methyl)thiazolyl propyl containing compounds achieved up to 80-fold the potency of histamine (pEC₅₀ values of 6.3-7.76) and partial to full agonistic activities with more or less pronounced bell-shaped concentration-response curves. A variety of aliphatic and aromatic residues was well tolerated and the agonistic N^G-carbamoylated guanidine-type ligands exhibited some functional bias towards G-protein activation. Additionally, representative ligands were investigated for dopamine receptor affinity (hD_{2long}R and hD₃R). These ligands showed affinity to the hD_{2long}R (pK_i value: 5.6-6.97) and hD₃R (pK_i value: 5.3-7.6).

In conclusion, this work afforded new radiolabeled and fluorescence labeled molecular tools for the hH_2R and showed that N^G -carbamoylated amino(methyl)thiazolepropylguanidines are G-Protein biased, high affinity hH_2R agonists with good longterm stability.

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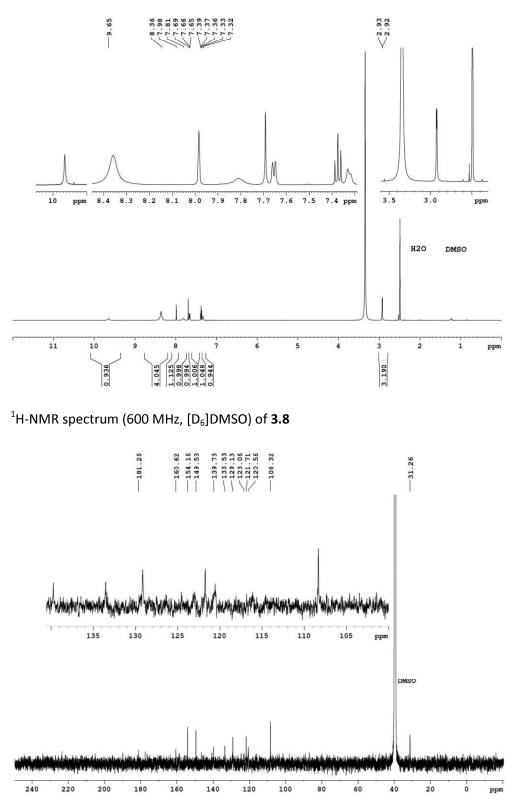
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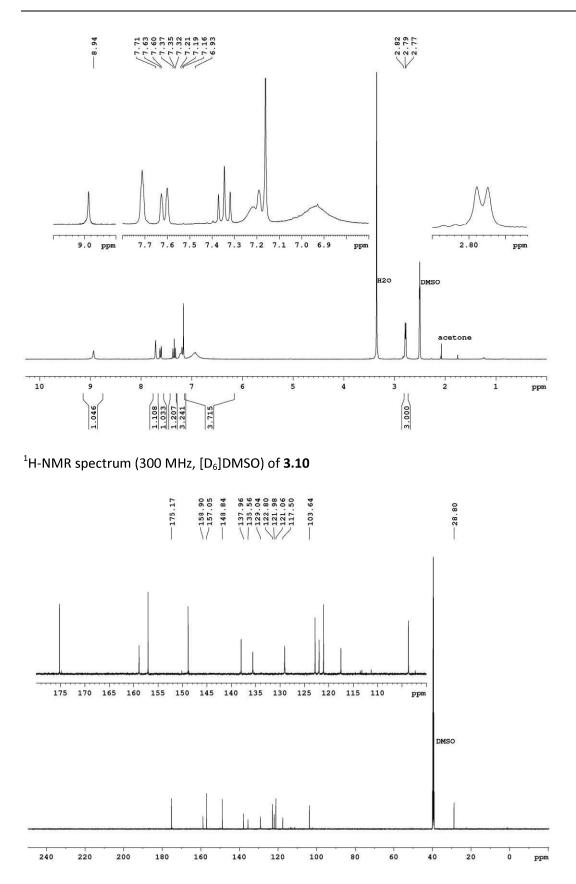
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APPENDIX

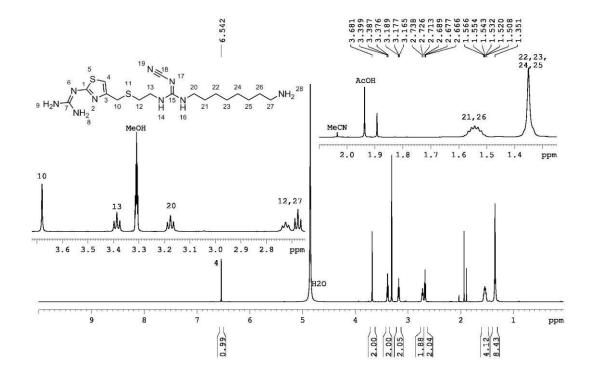




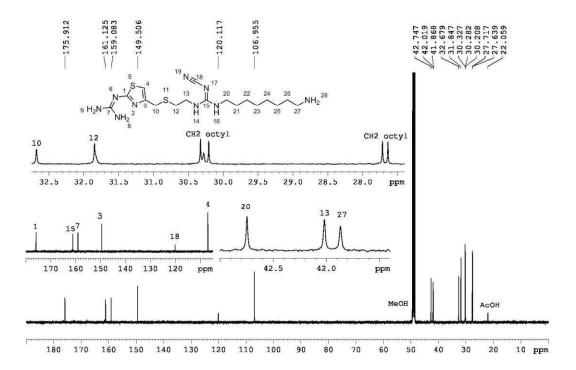
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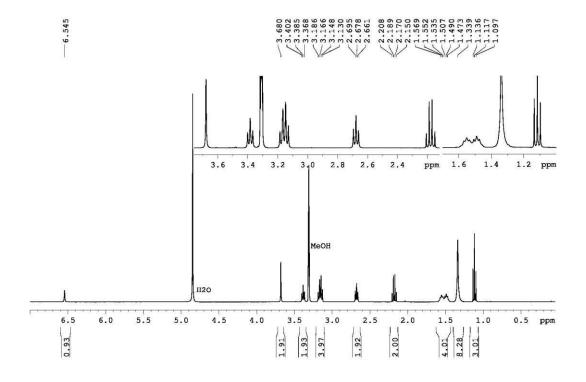
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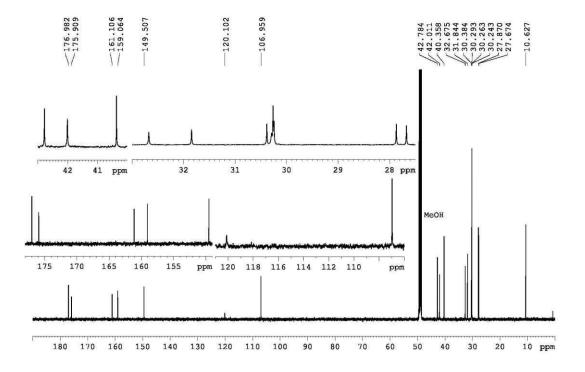
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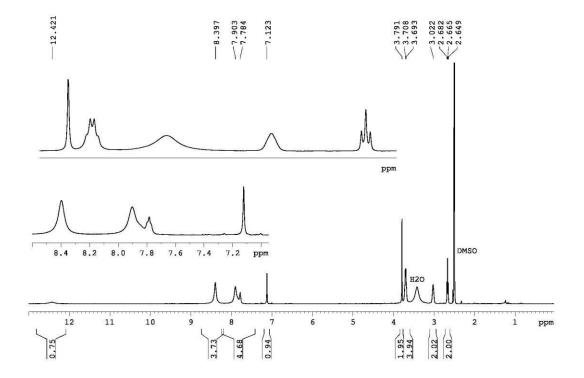
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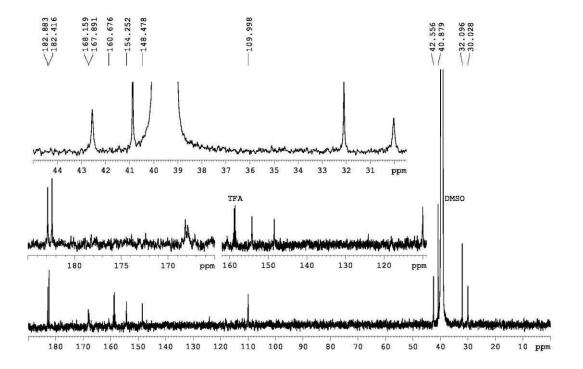


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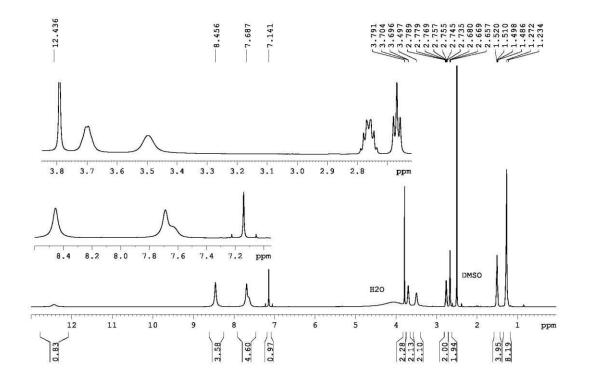


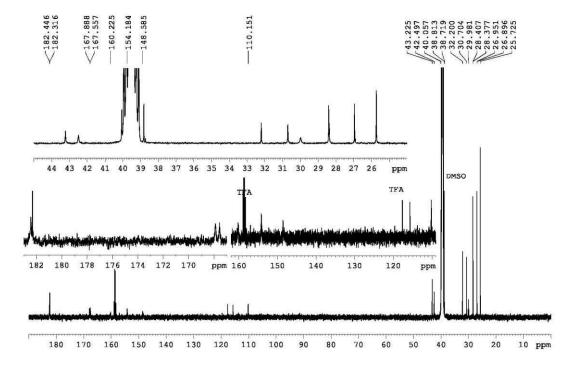
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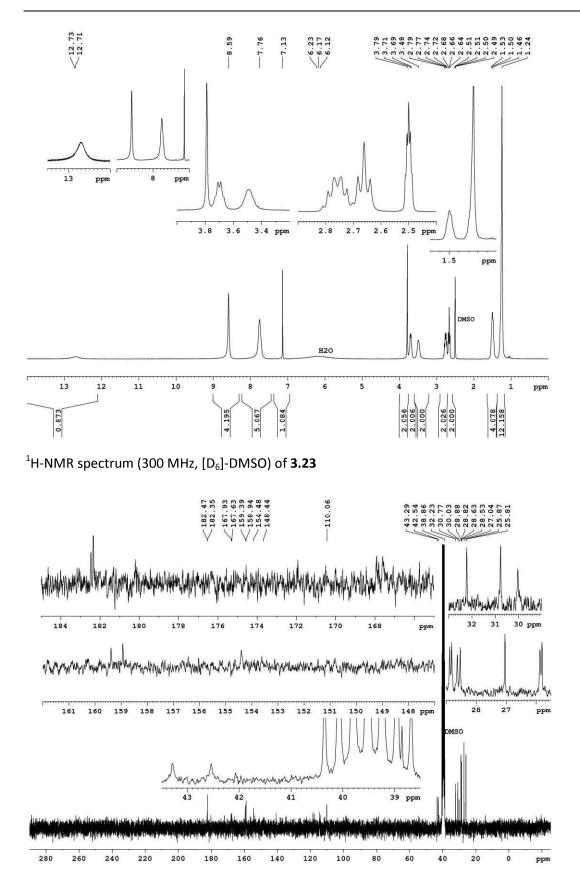


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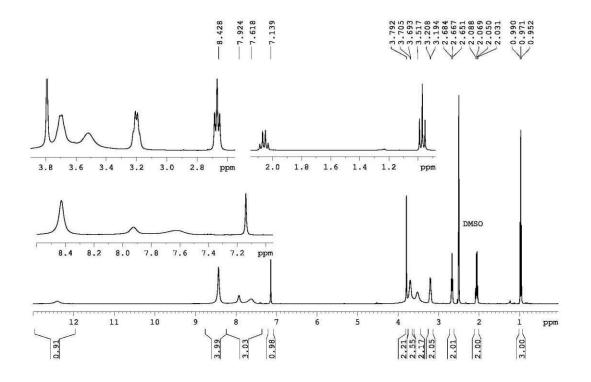


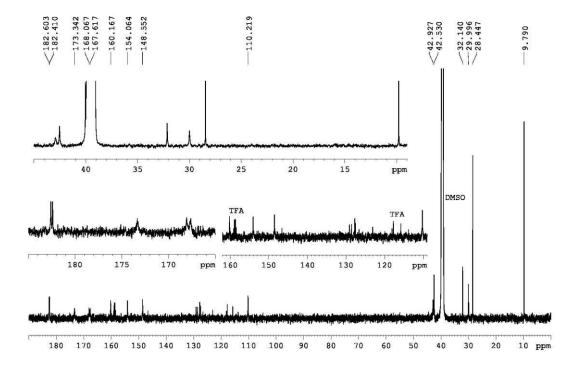


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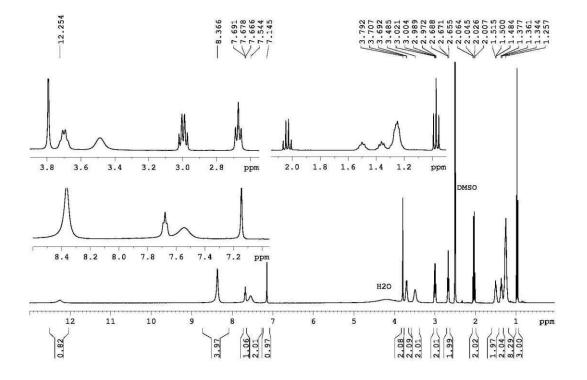


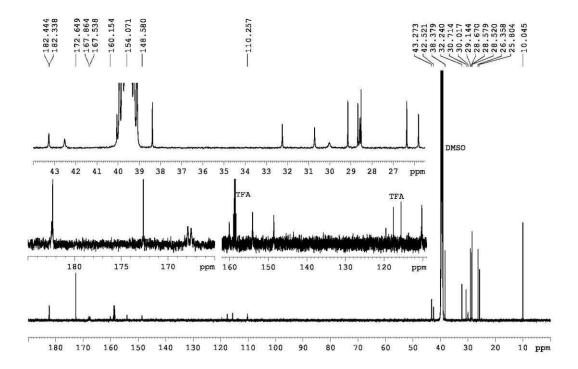
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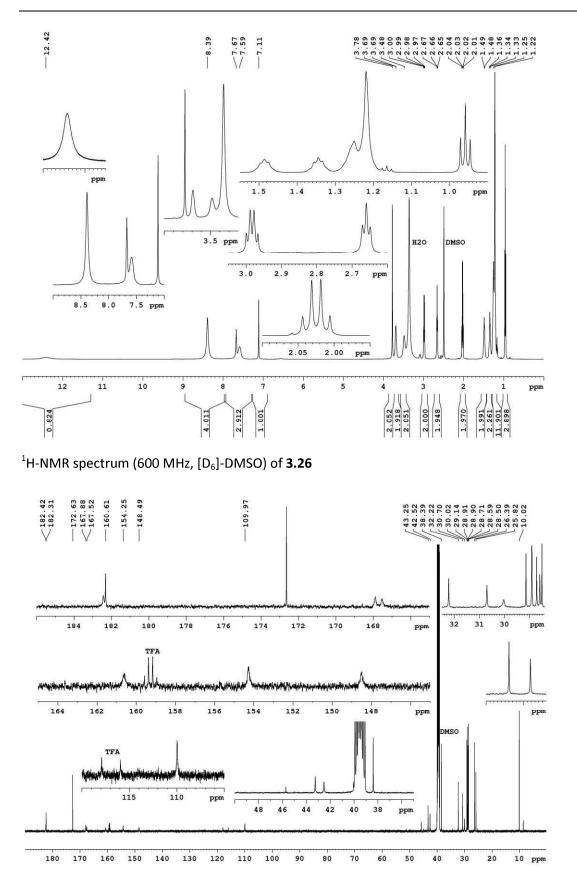


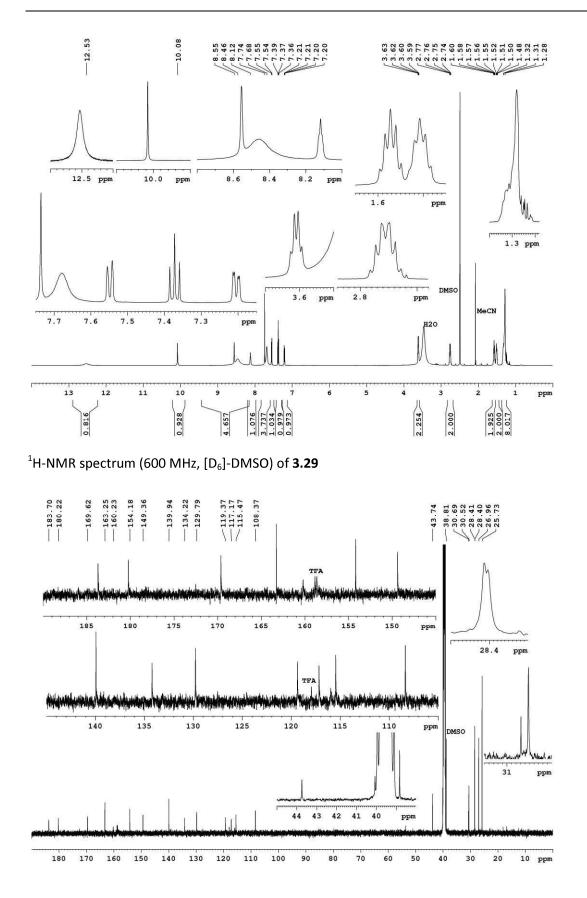
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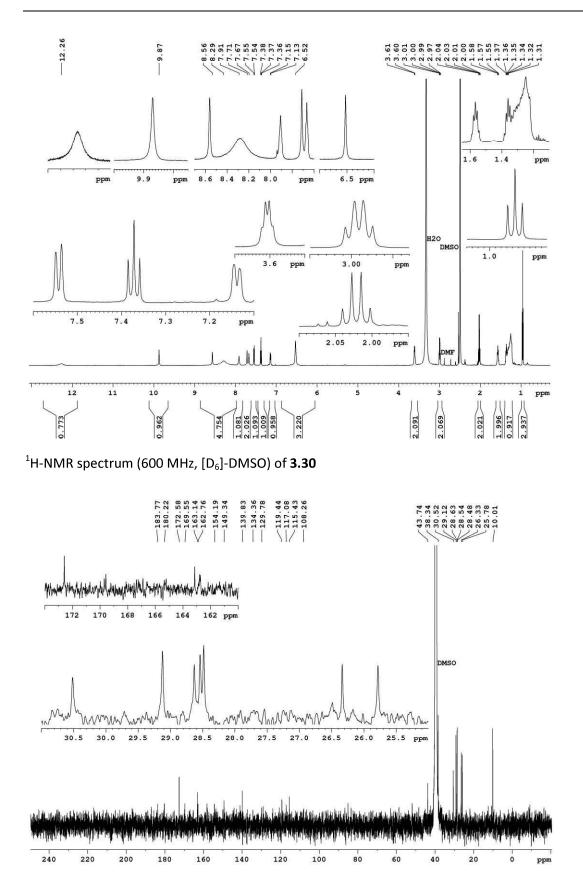


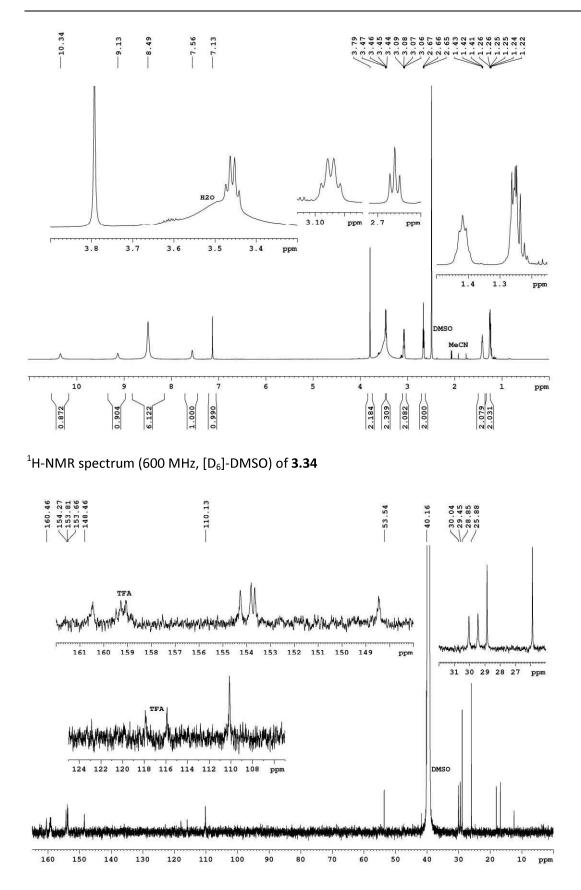


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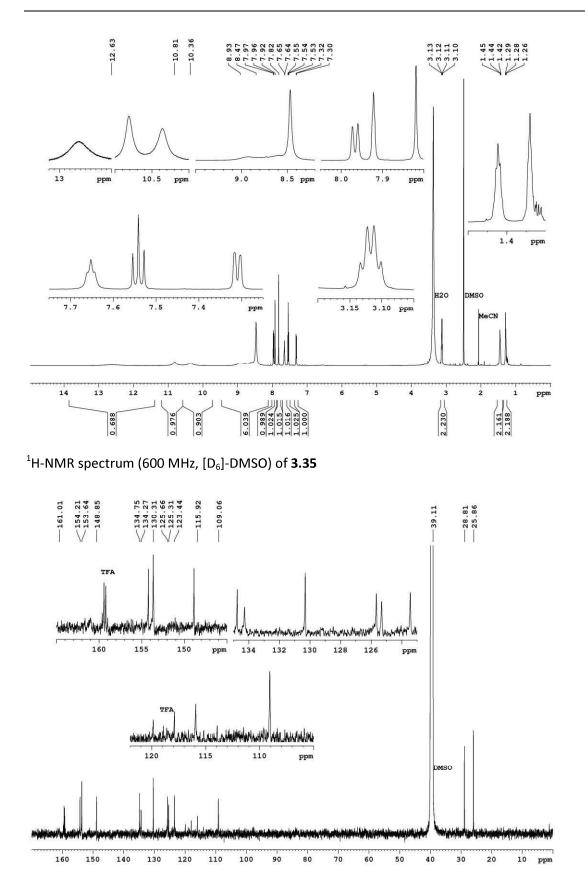




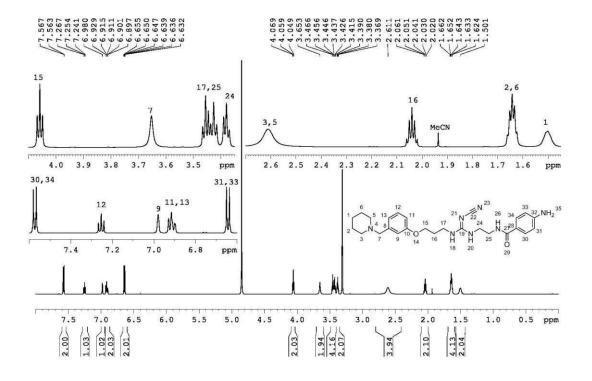




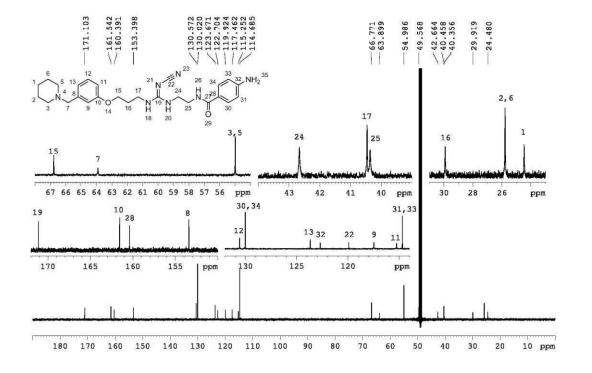
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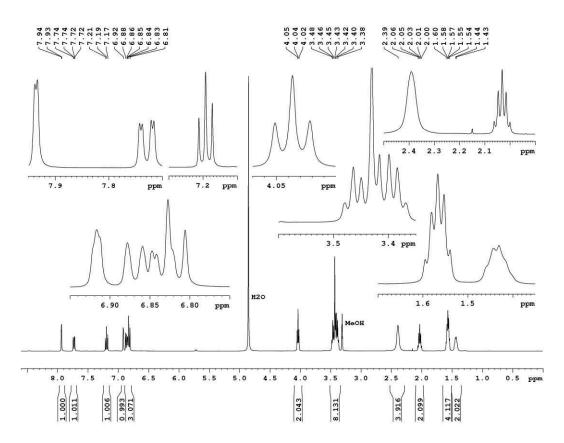
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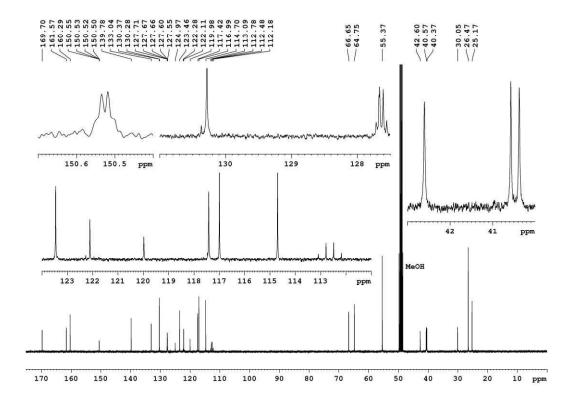
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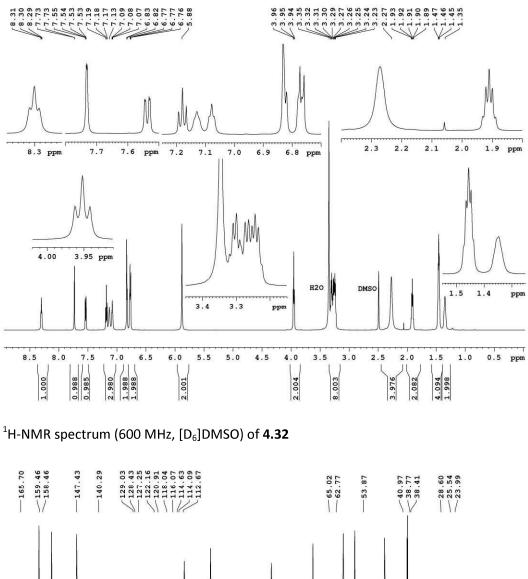
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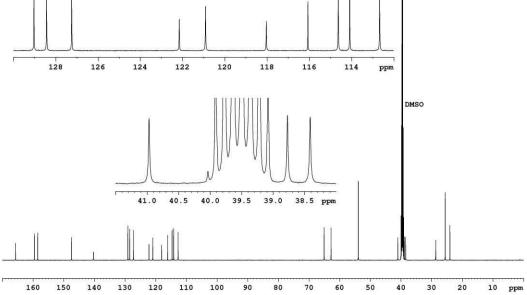


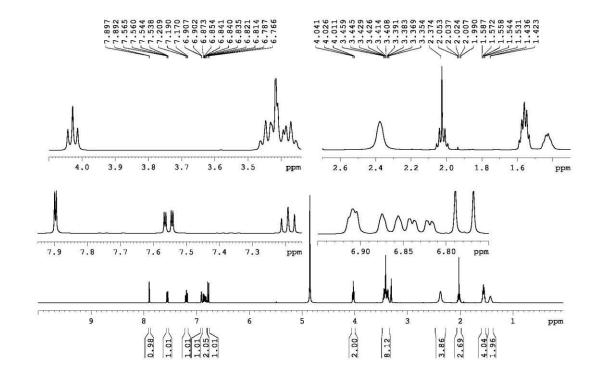
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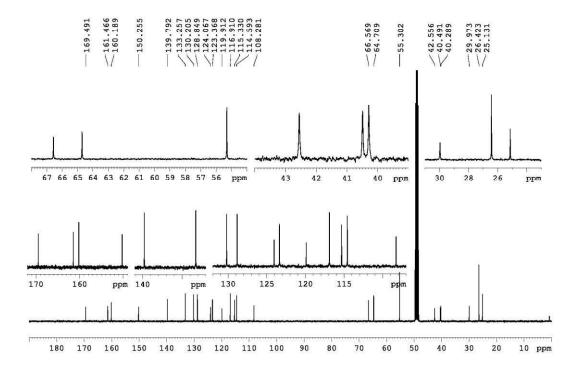


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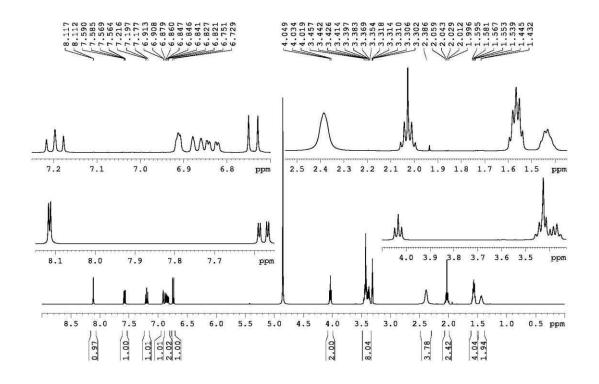


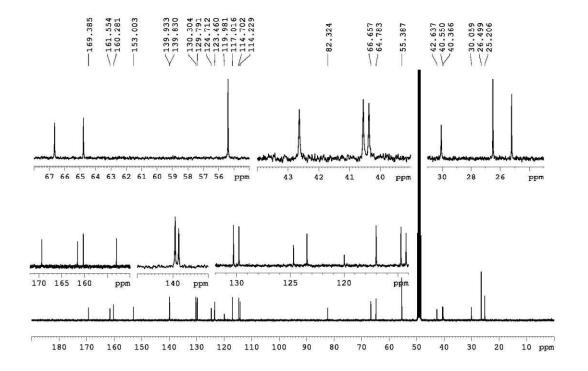




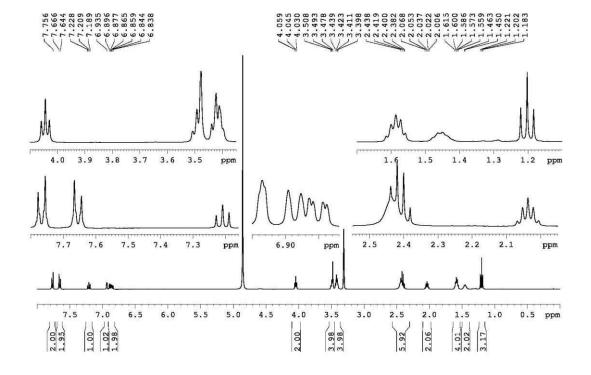


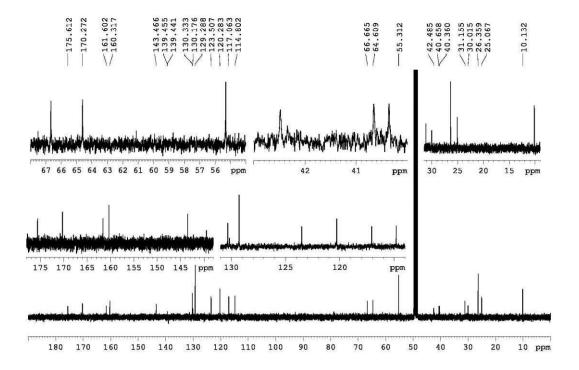
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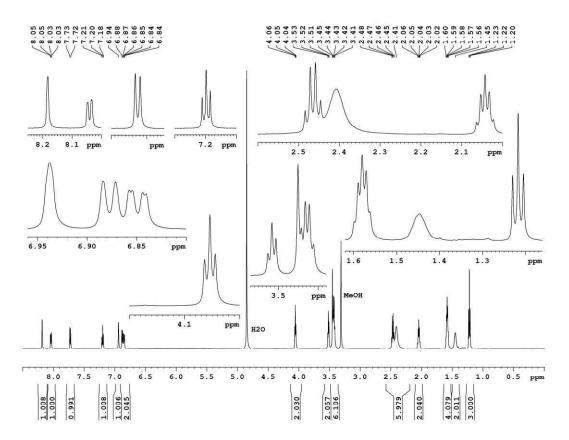


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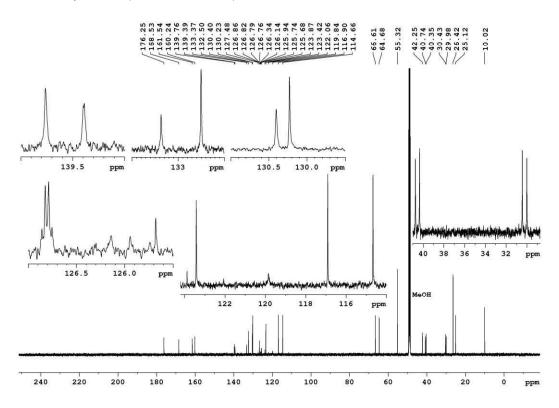


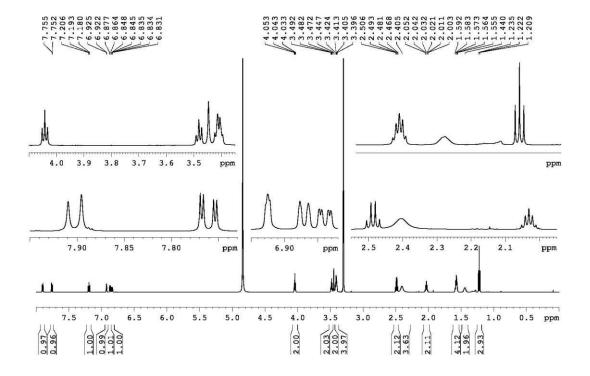


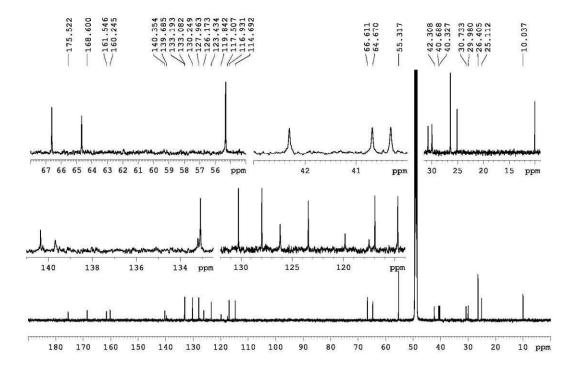
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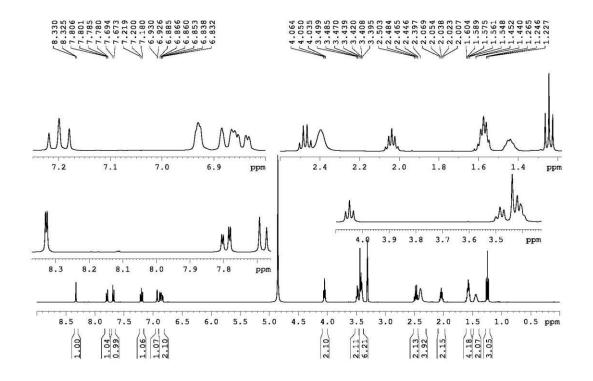
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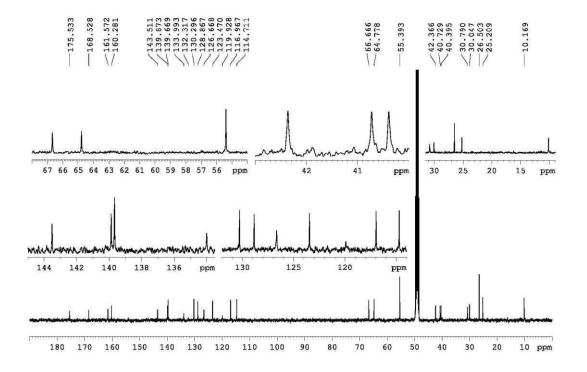




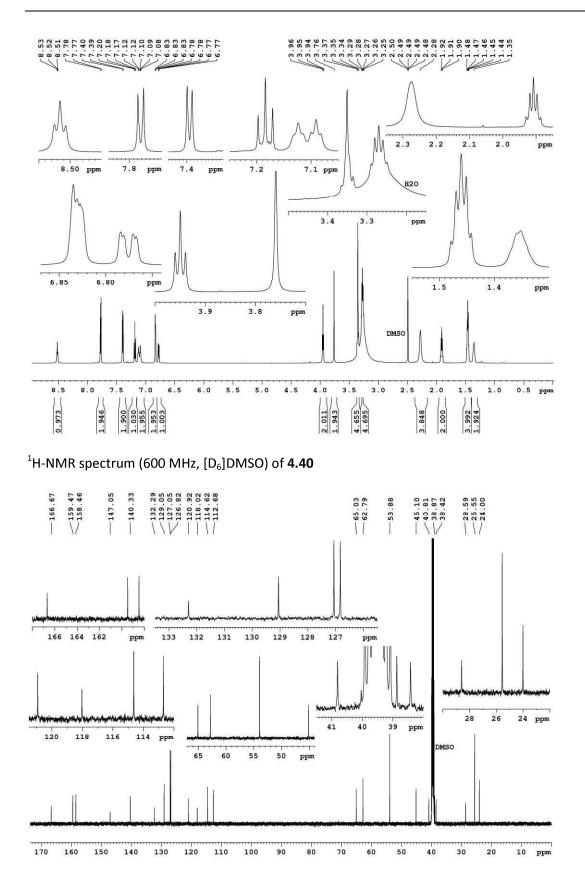


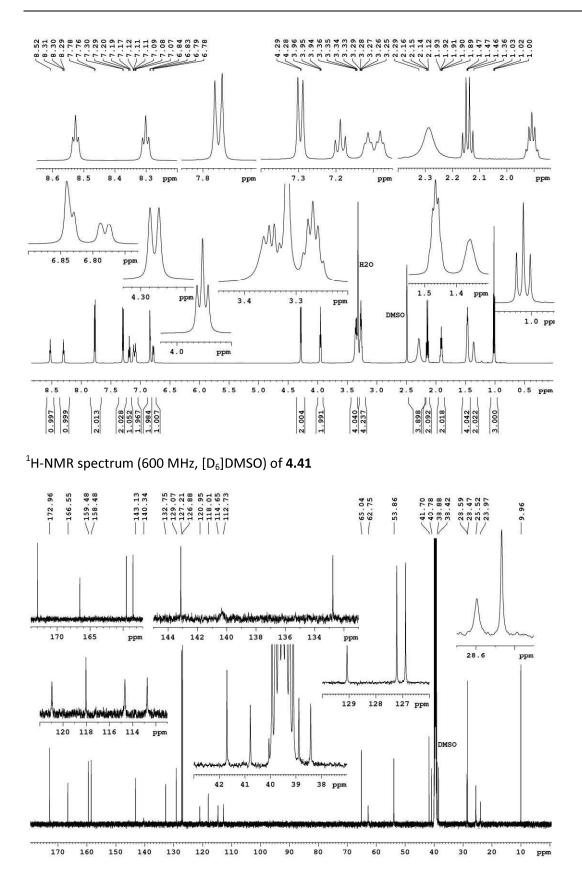
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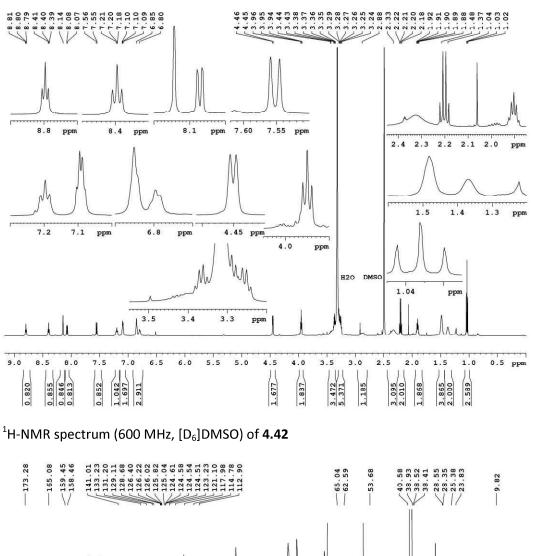


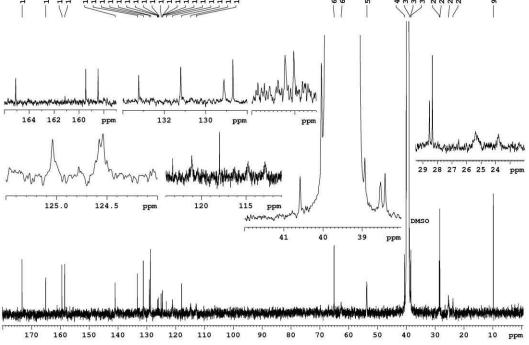


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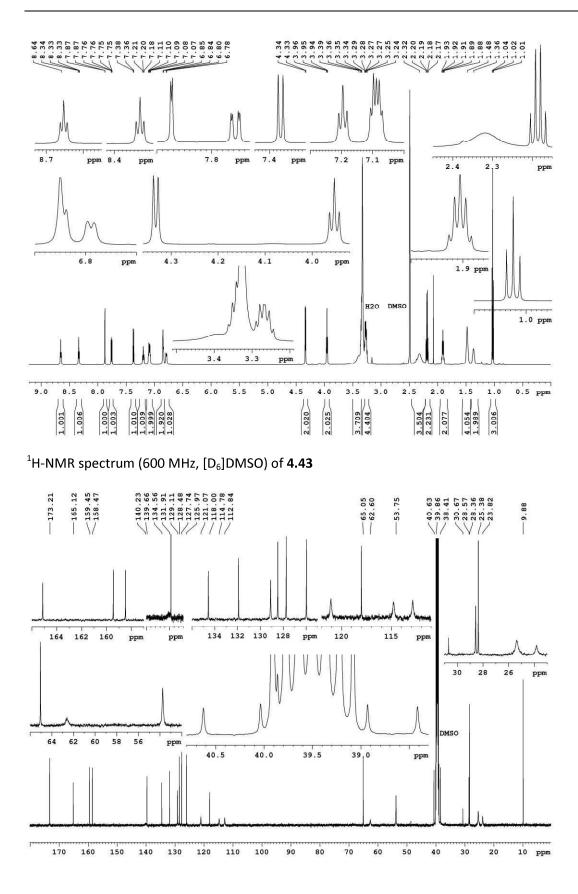


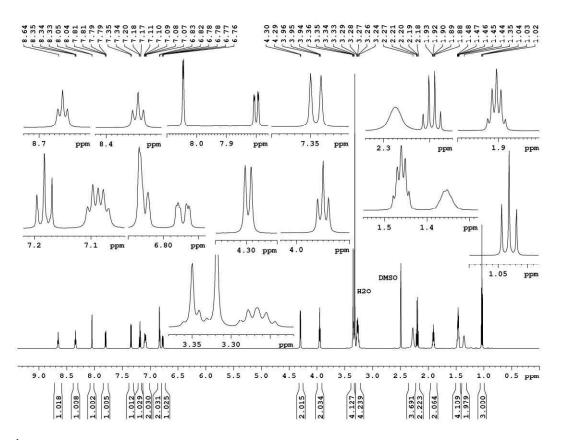




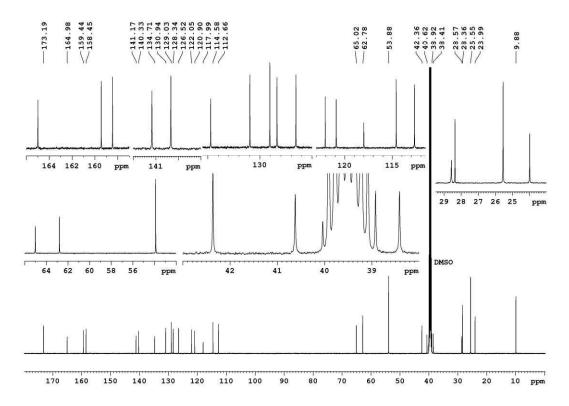


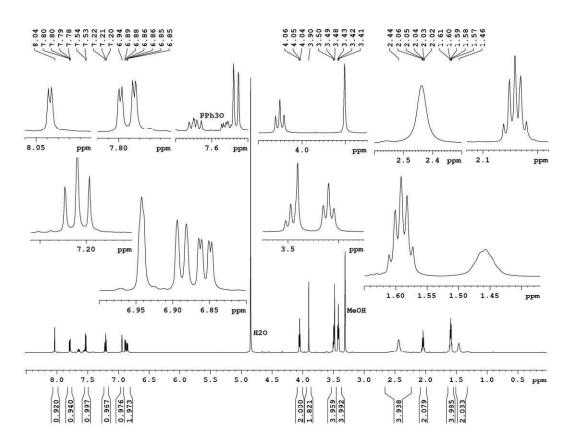
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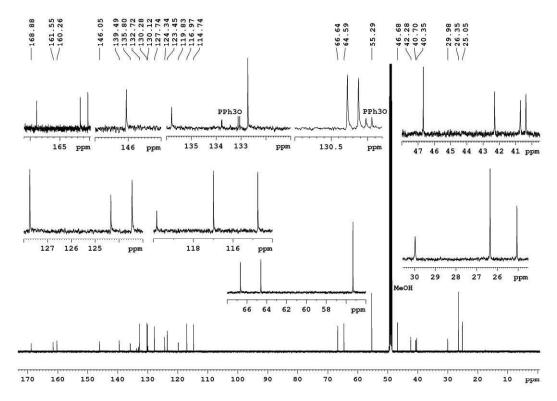


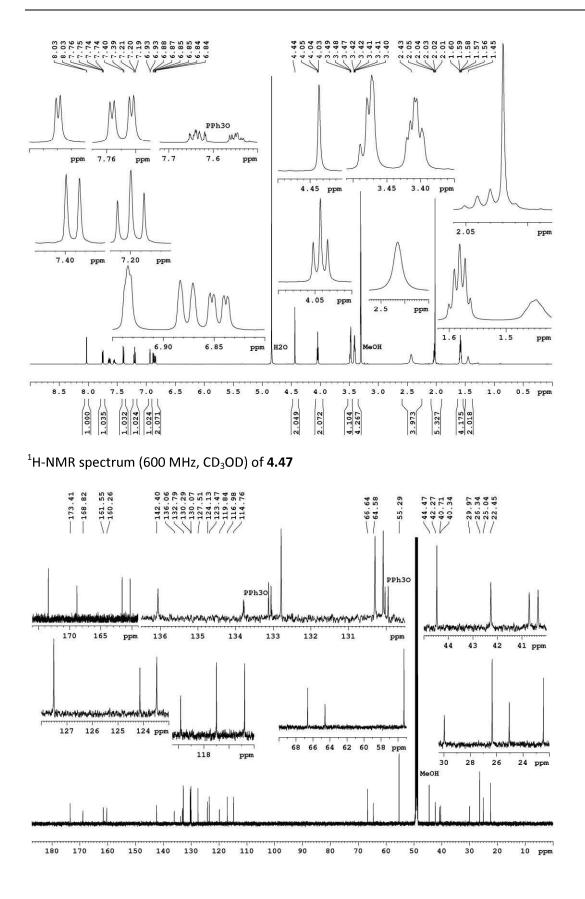


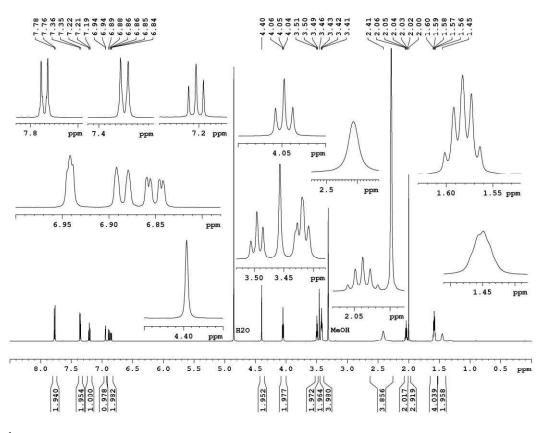
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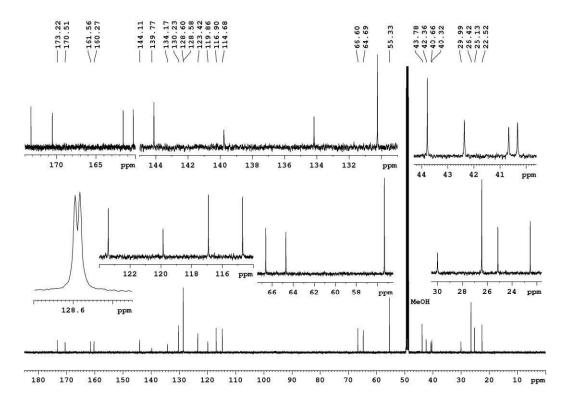


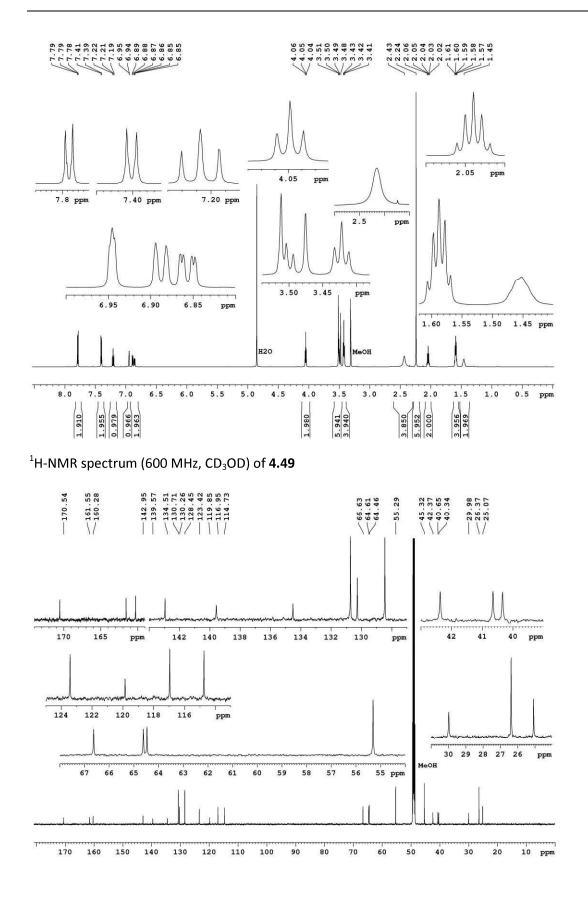




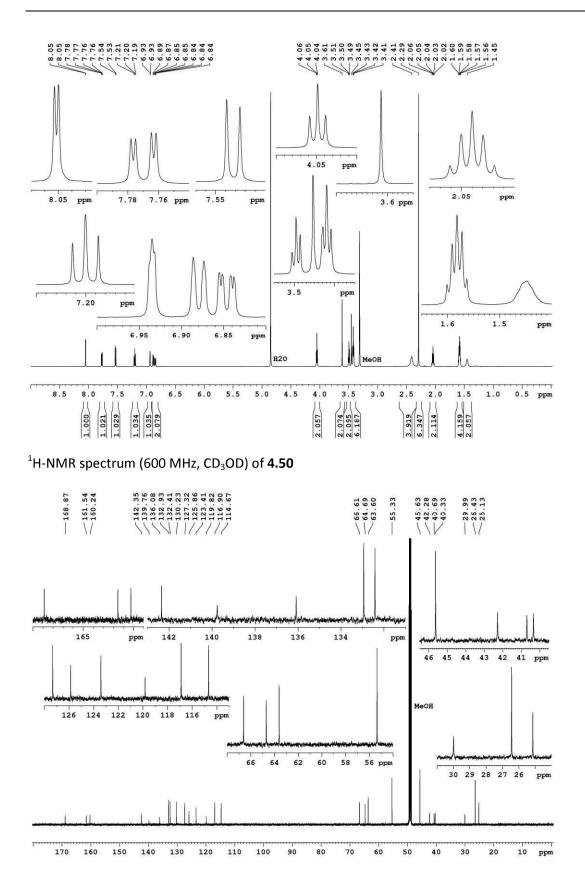




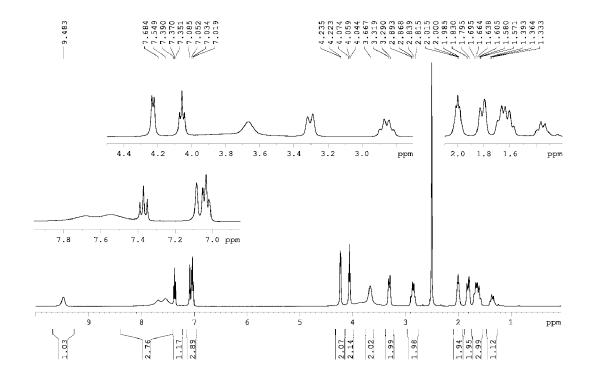




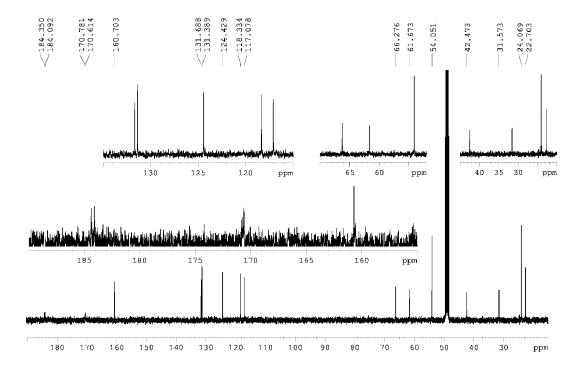
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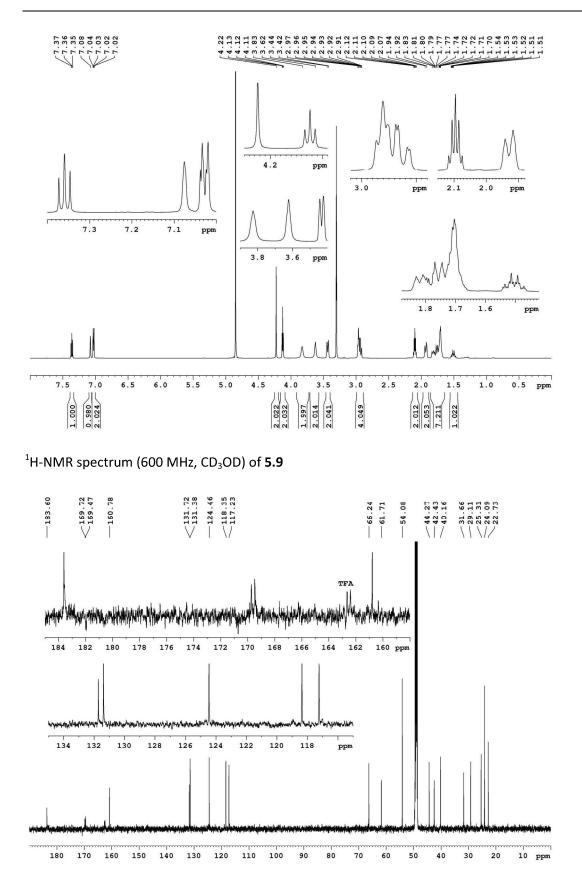
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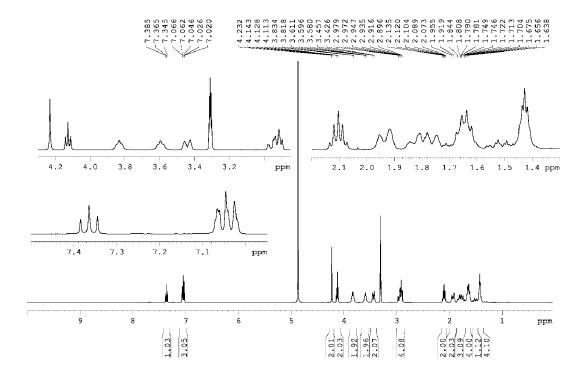
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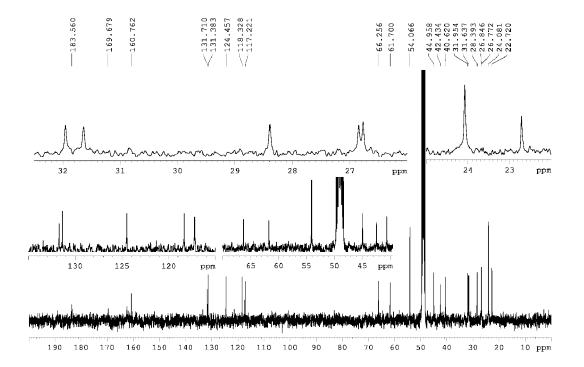
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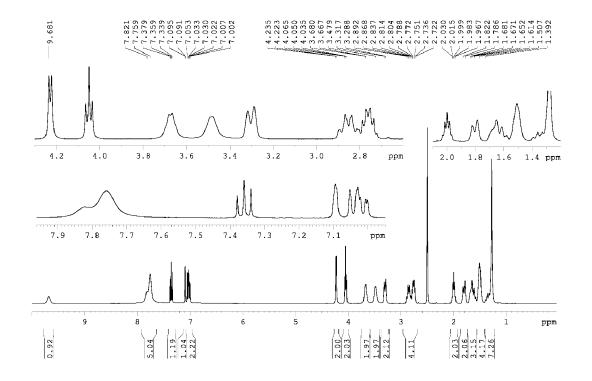
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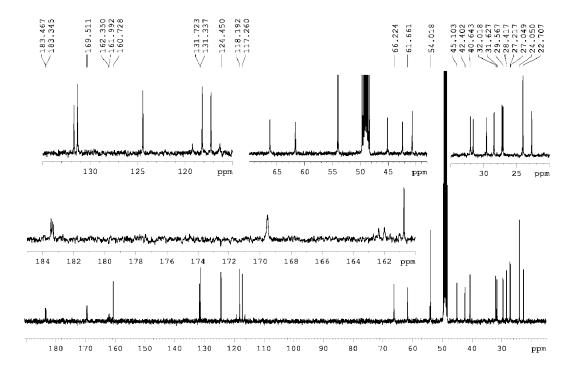


¹H-NMR spectrum (400 MHz, CD_3OD) of **5.10**



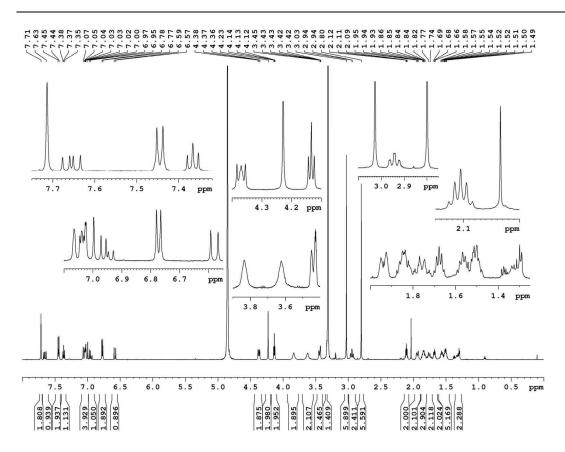
 $^{\rm 13}\text{C-NMR}$ spectrum (100 MHz, CD₃OD) of **5.10**



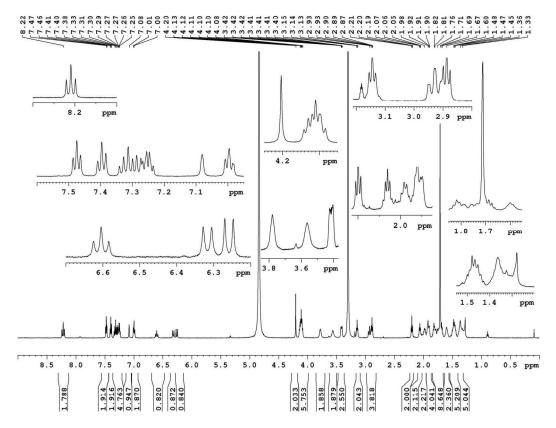


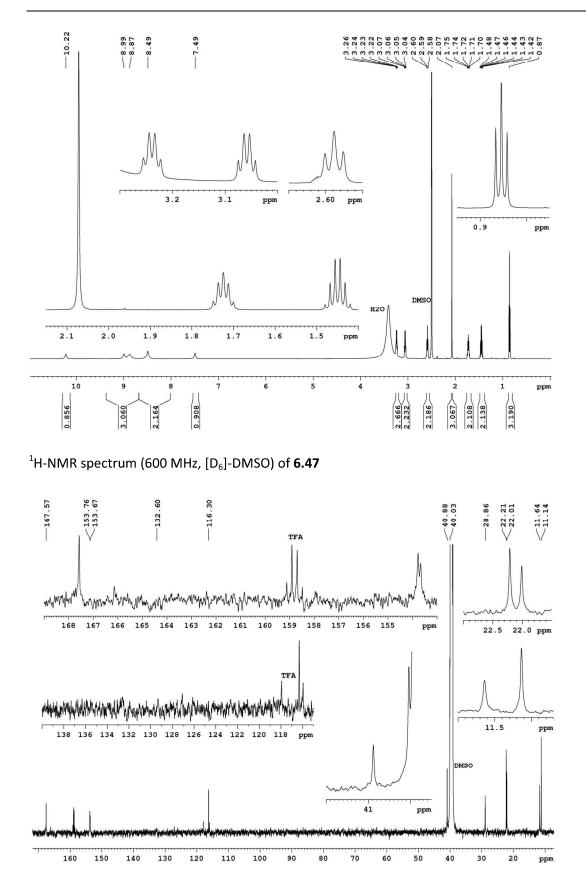
 $^{\rm 13}\text{C-NMR}$ spectrum (100 MHz, CD₃OD) of **5.11**

Appendix

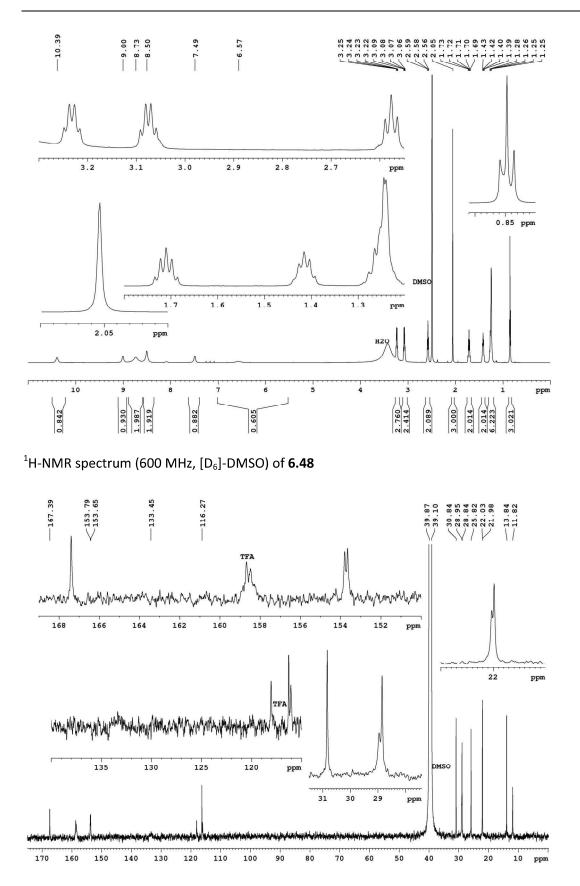


 1 H-NMR spectrum (600 MHz, CD₃OD) of **5.13**

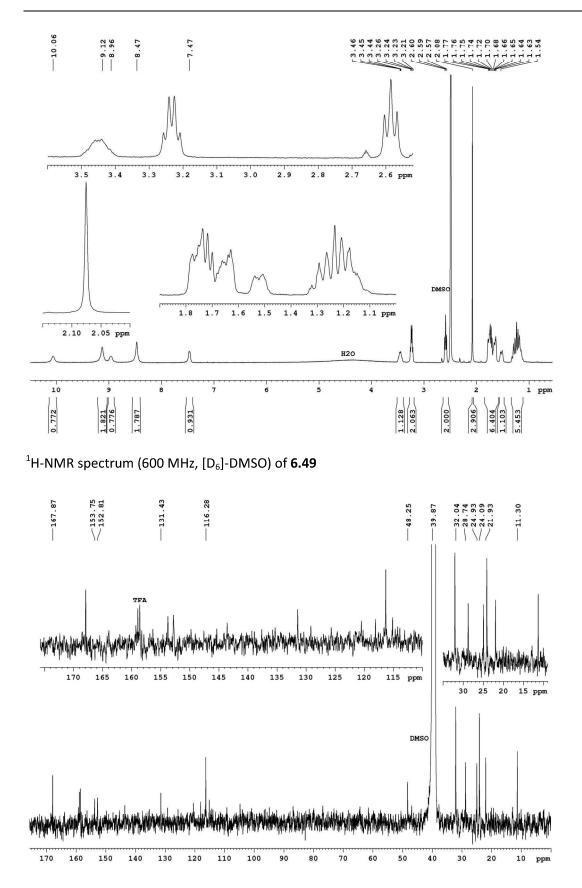


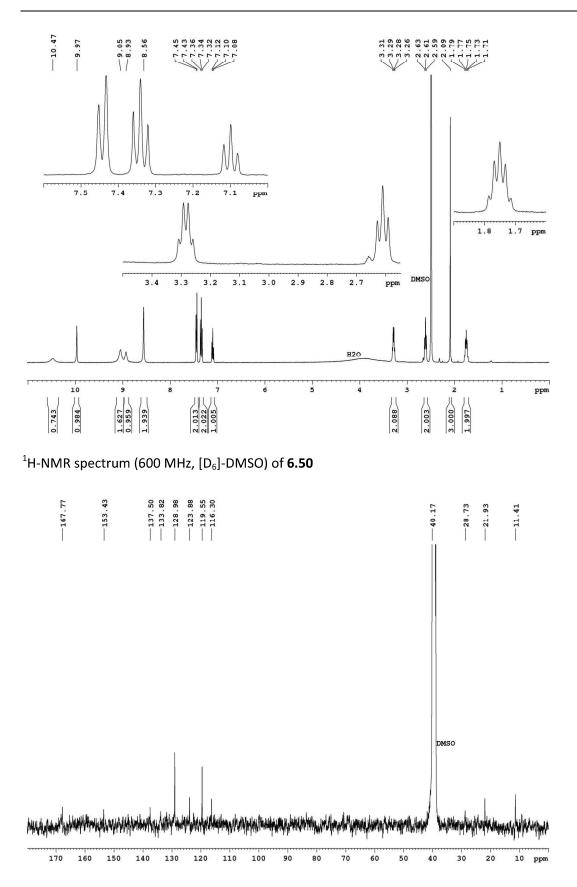


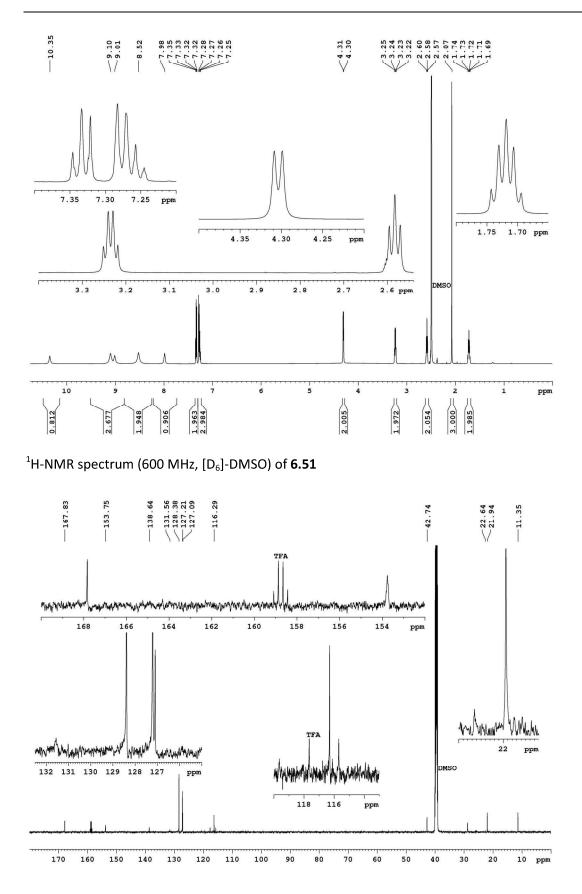
¹³C-NMR spectrum (150 MHz, [D₆]-DMSO) of **6.47**

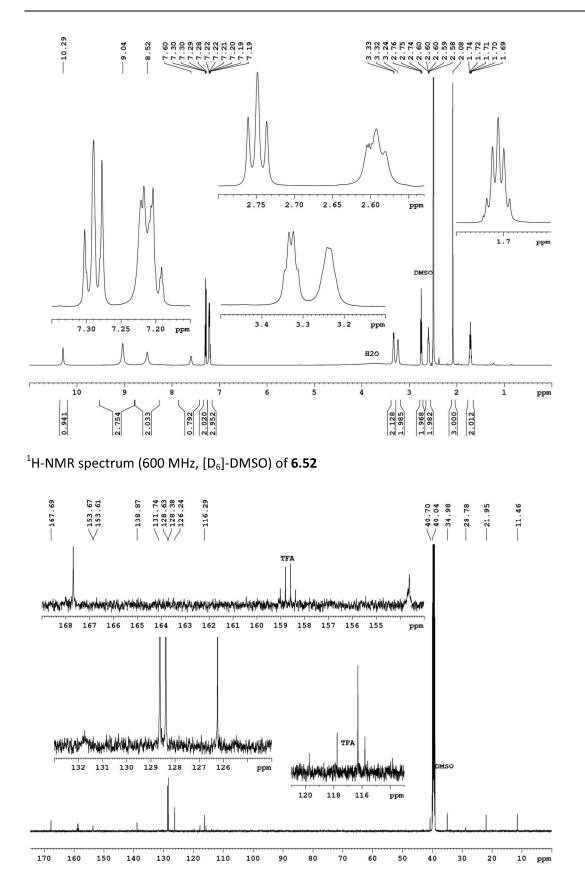


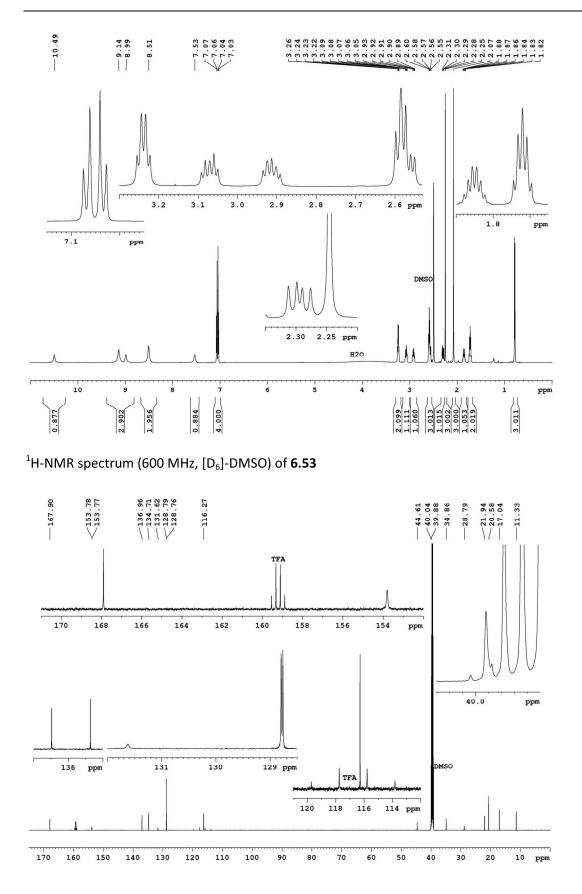
¹³C-NMR spectrum (150 MHz, [D₆]-DMSO) of **6.48**

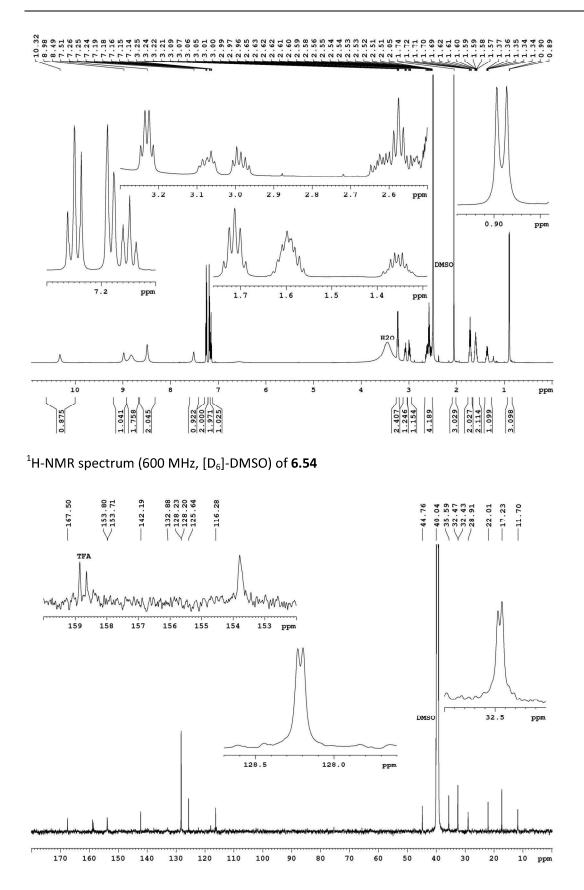




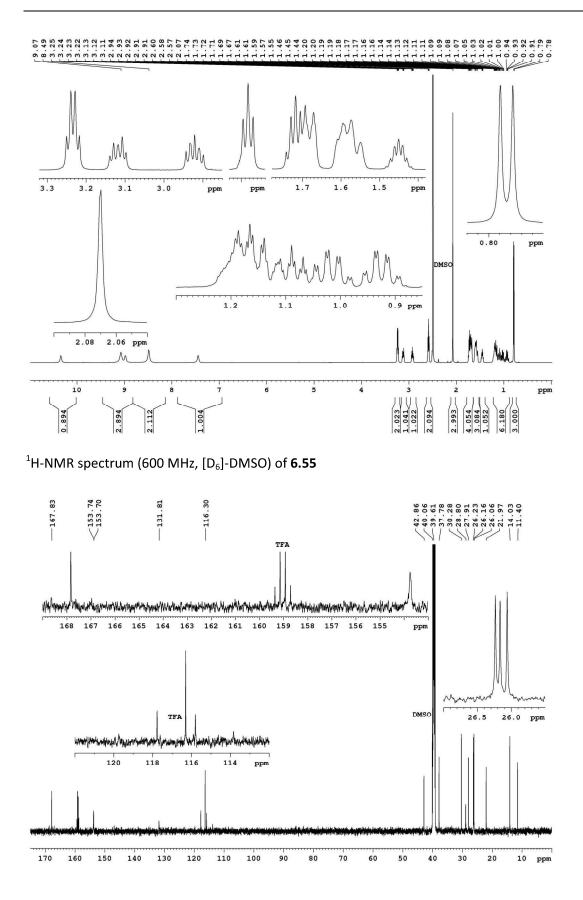


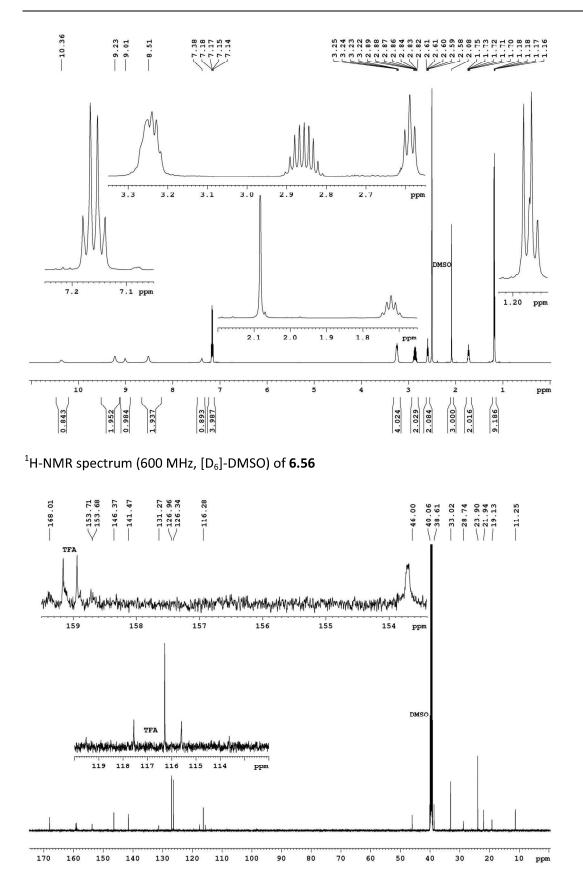




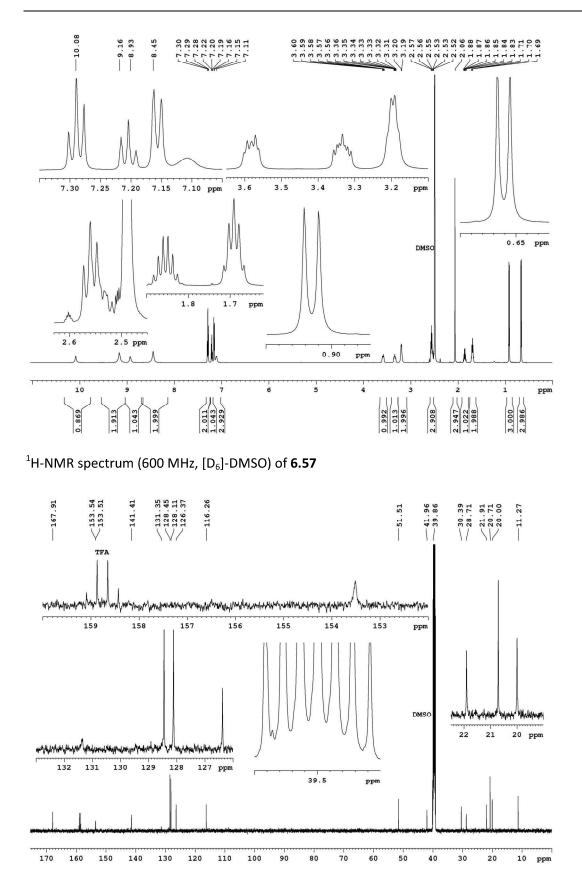


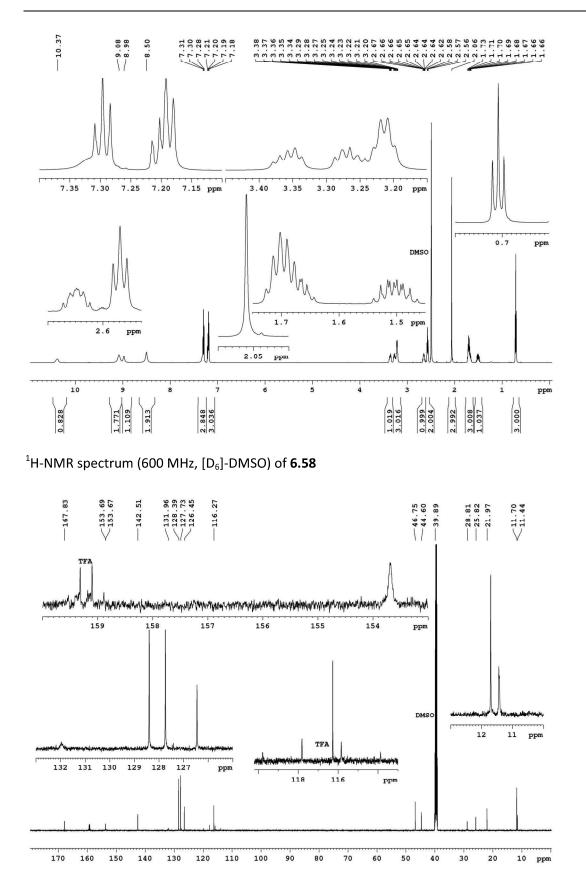
¹³C-NMR spectrum (150 MHz, [D₆]-DMSO) of **6.54**

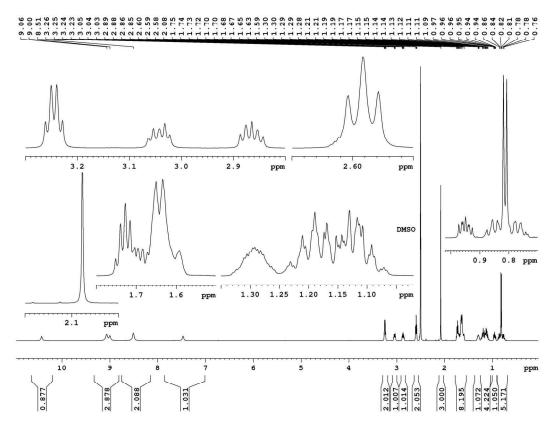


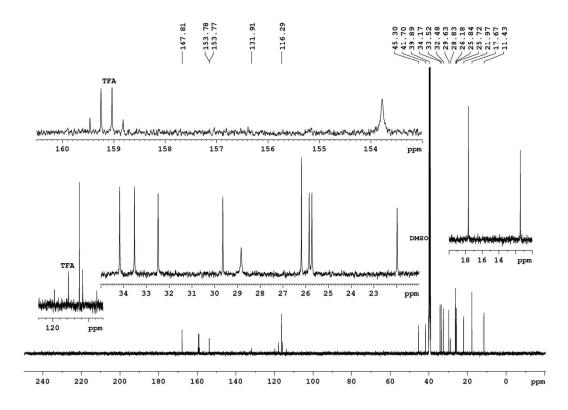


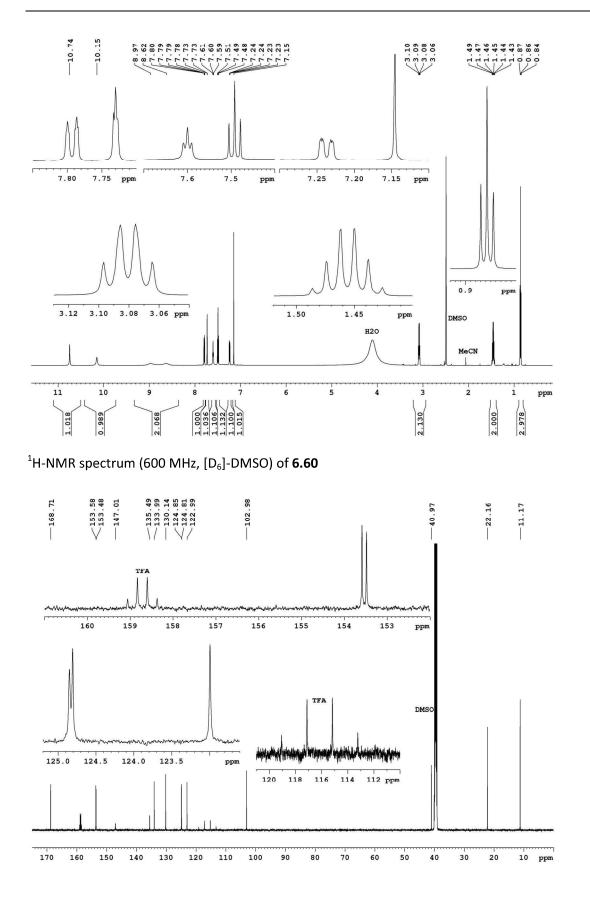
¹³C-NMR spectrum (150 MHz, [D₆]-DMSO) of **6.56**



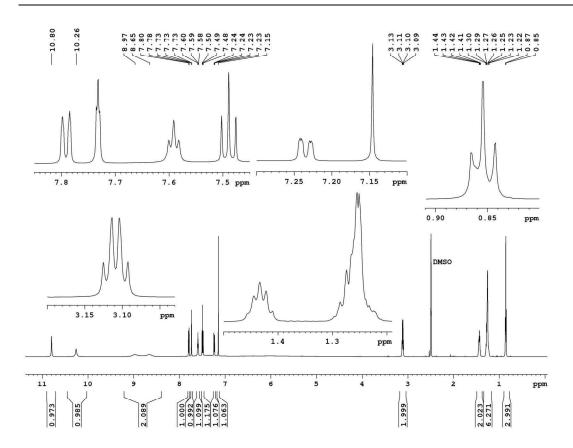




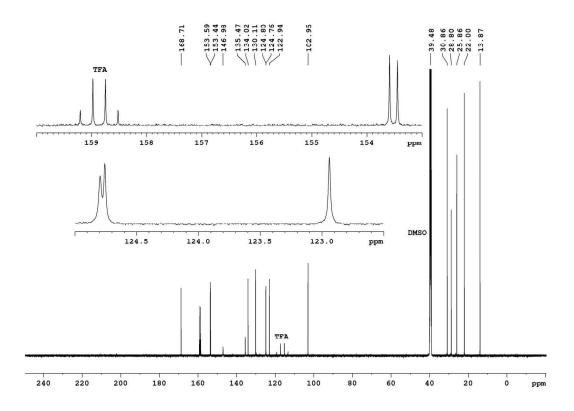


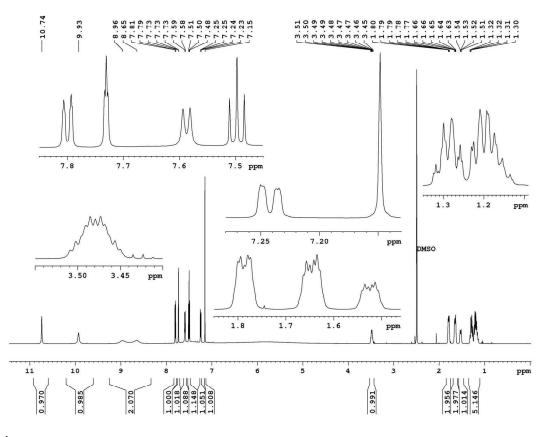


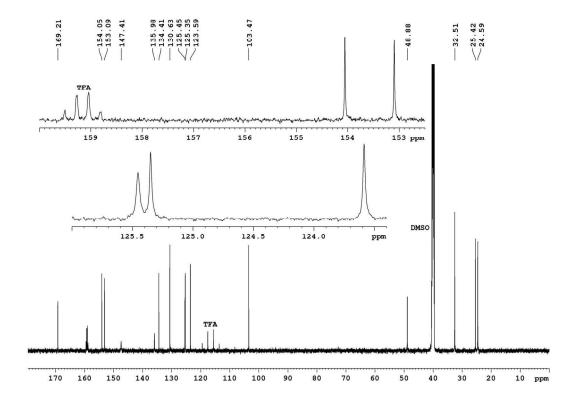
¹³C-NMR spectrum (150 MHz, [D₆]-DMSO) of **6.60**

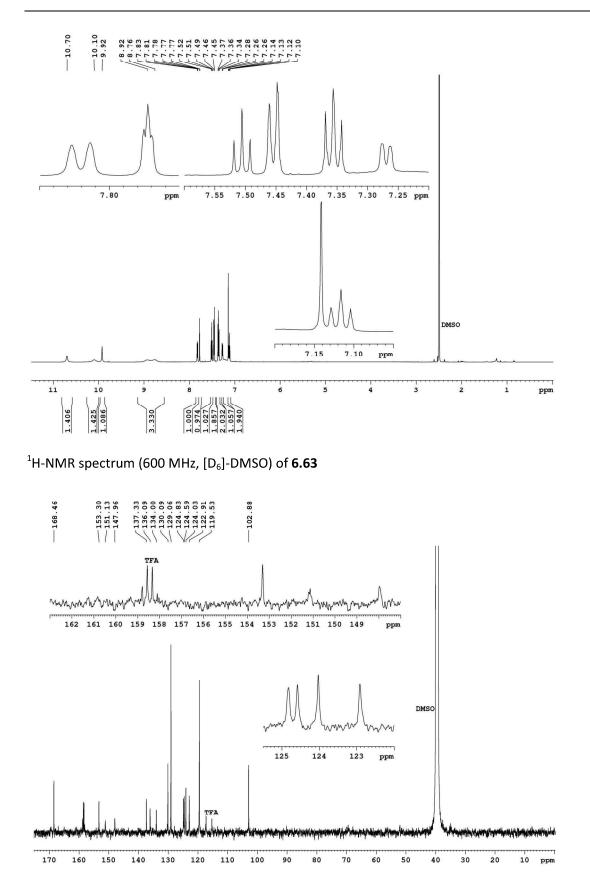


¹H-NMR spectrum (600 MHz, [D₆]-DMSO) of **6.61**

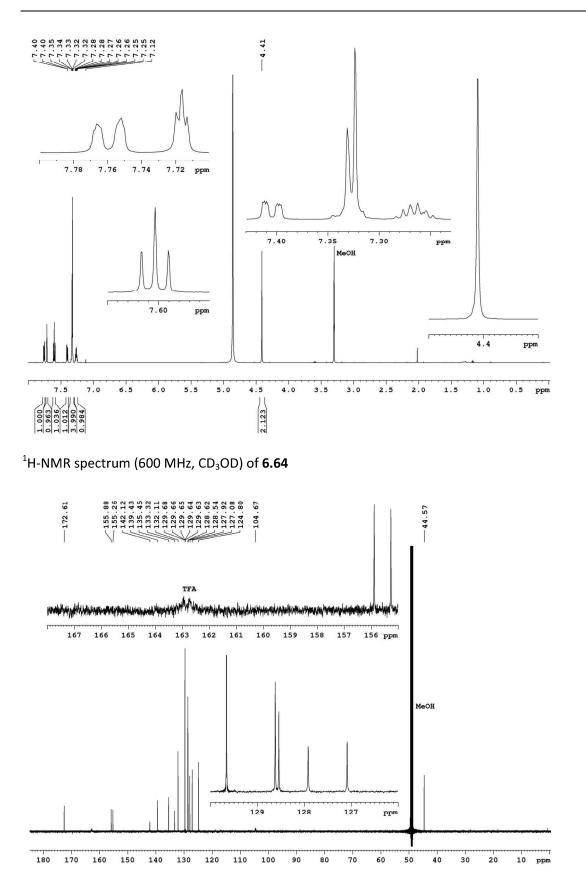


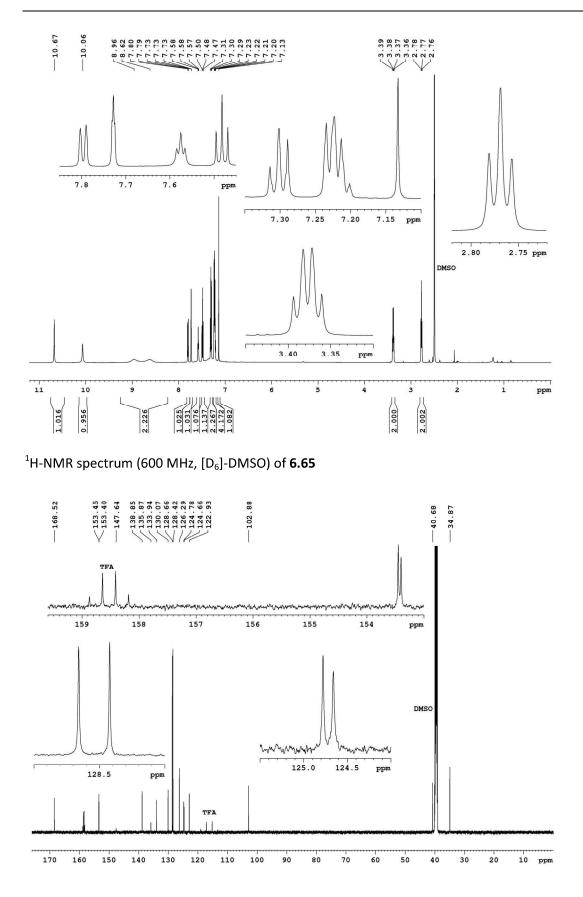


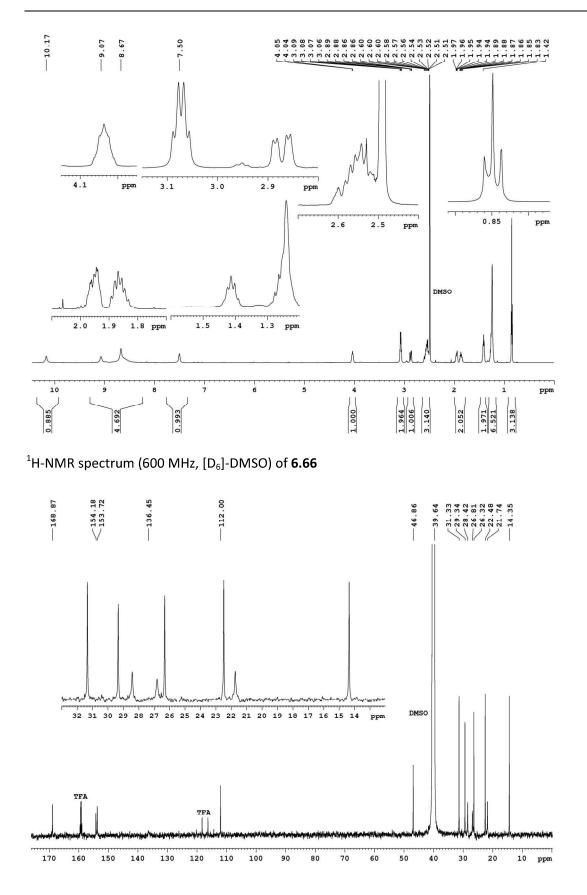




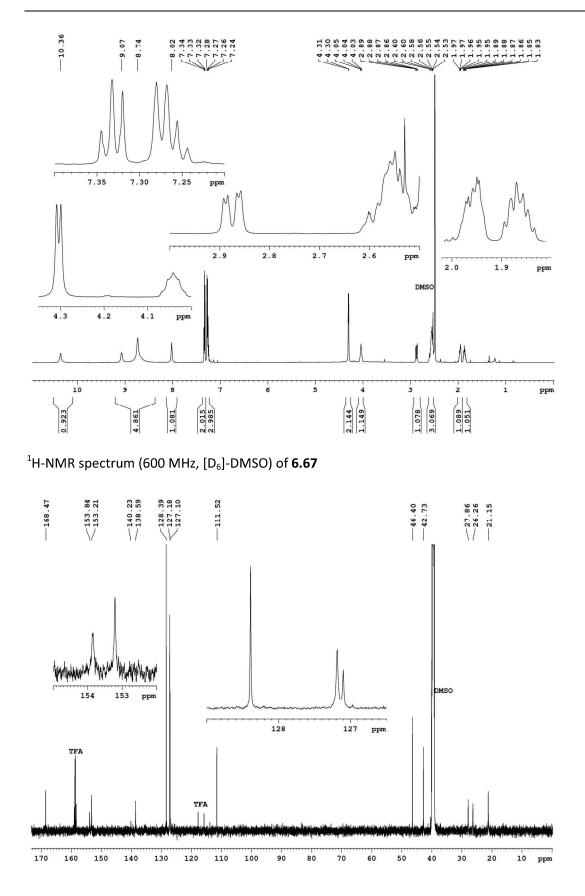
¹³C-NMR spectrum (150 MHz, [D₆]-DMSO) of **6.63**

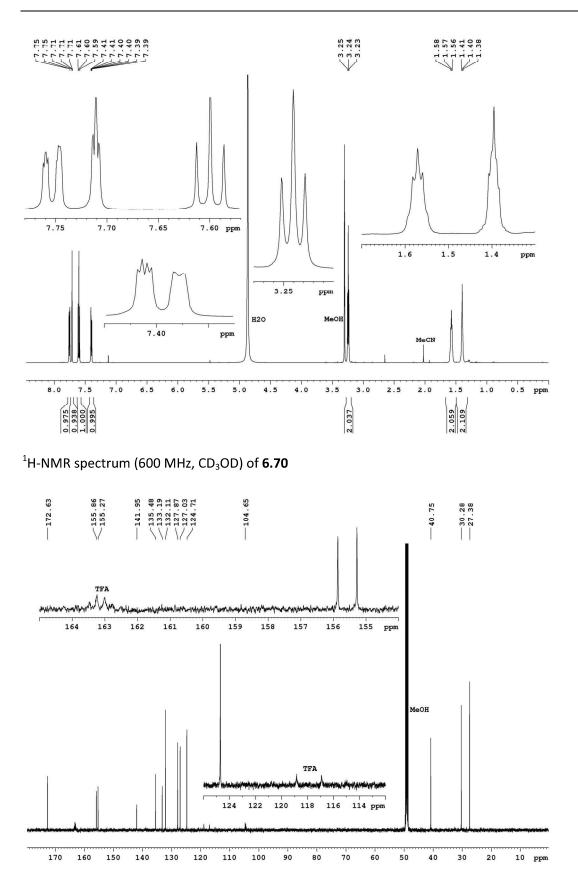




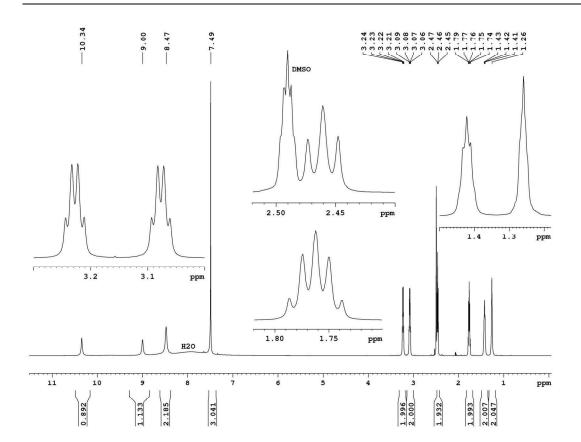


¹³C-NMR spectrum (150 MHz, [D₆]-DMSO) of **6.66**

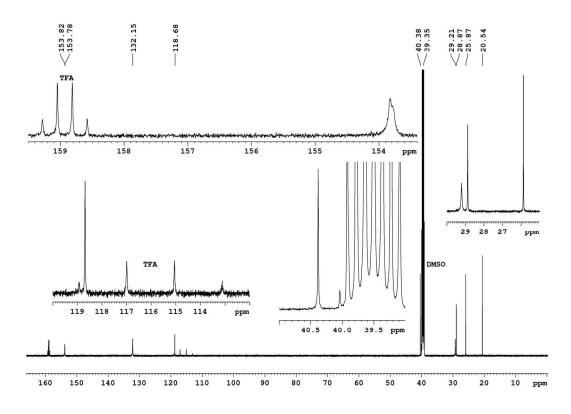




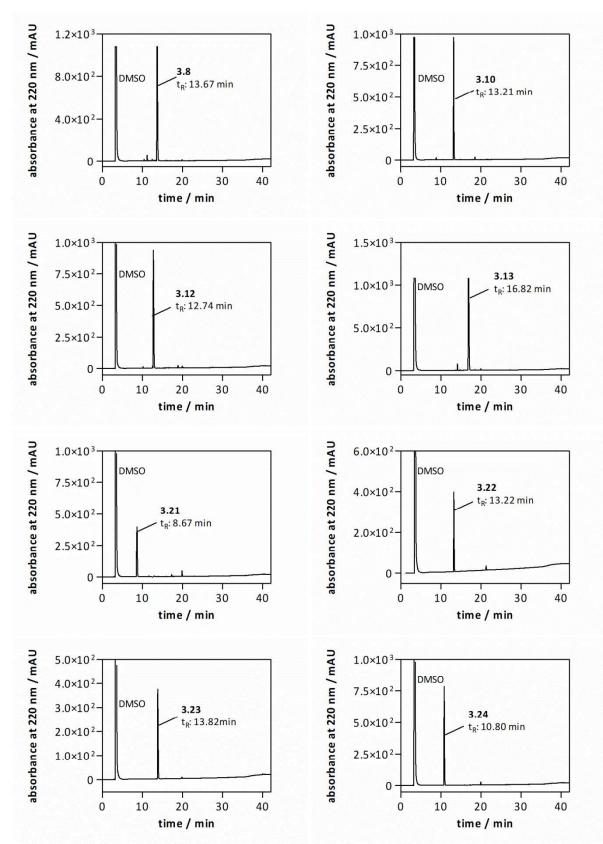
 $^{13}\text{C-NMR}$ spectrum (150 MHz, CD₃OD) of **6.70**



¹H-NMR spectrum (600 MHz, [D₆]-DMSO) of **6.71**

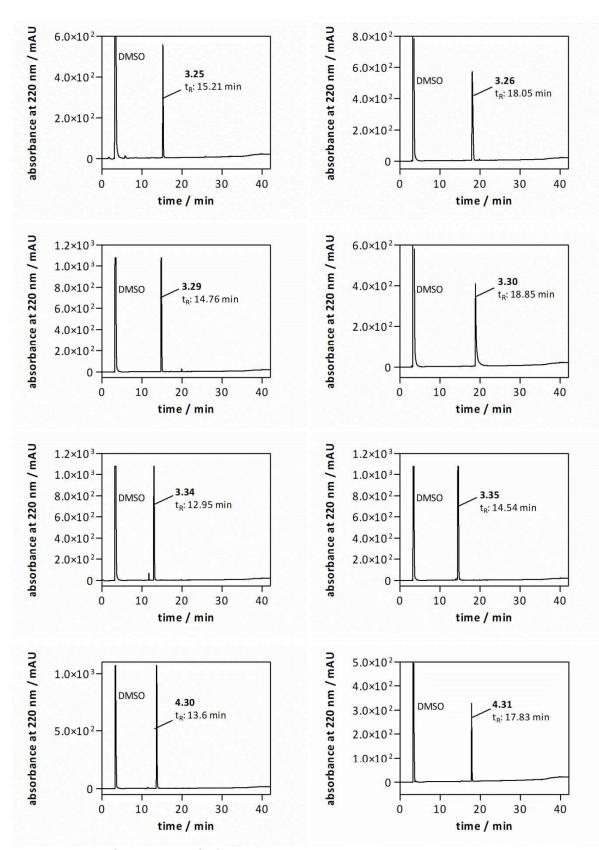


 $^{13}\text{C-NMR}$ spectrum (150 MHz, [D₆]-DMSO) of **6.71**

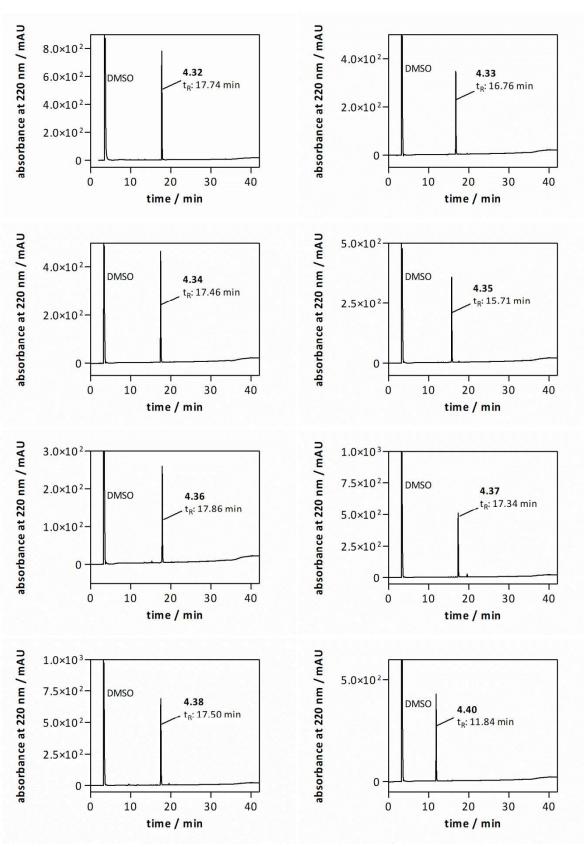


RP-HPLC CHROMATOGRAMS

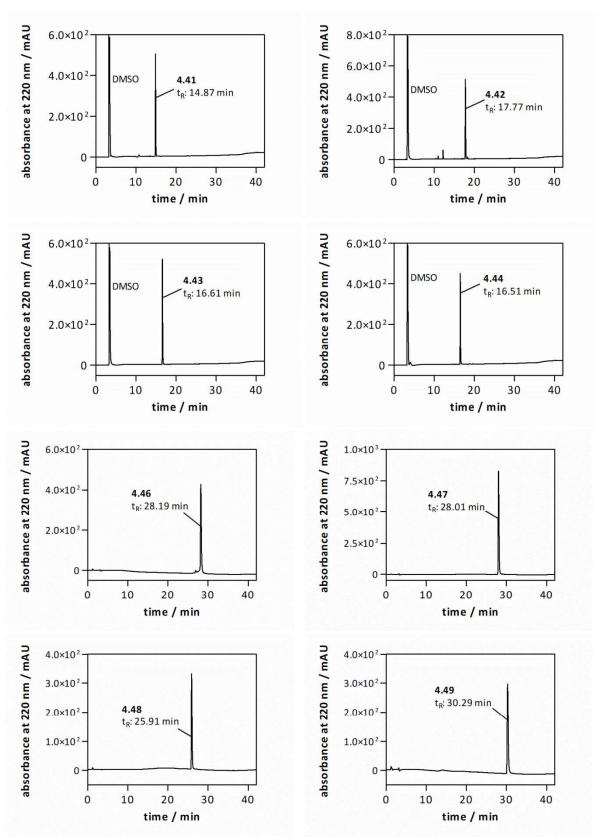
RP-HPLC analysis (purity control) of 3.8, 3.10, 3.12, 3.13, 3.21-3.24



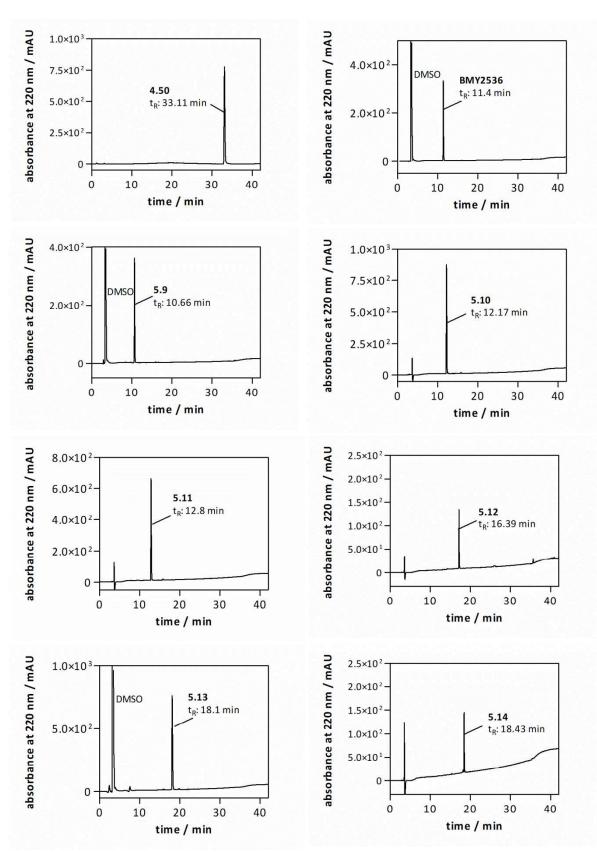
RP-HPLC analysis (purity control) of 3.25, 3.26, 3.29, 3.30, 3.34, 3.35, 4.30, 4.31



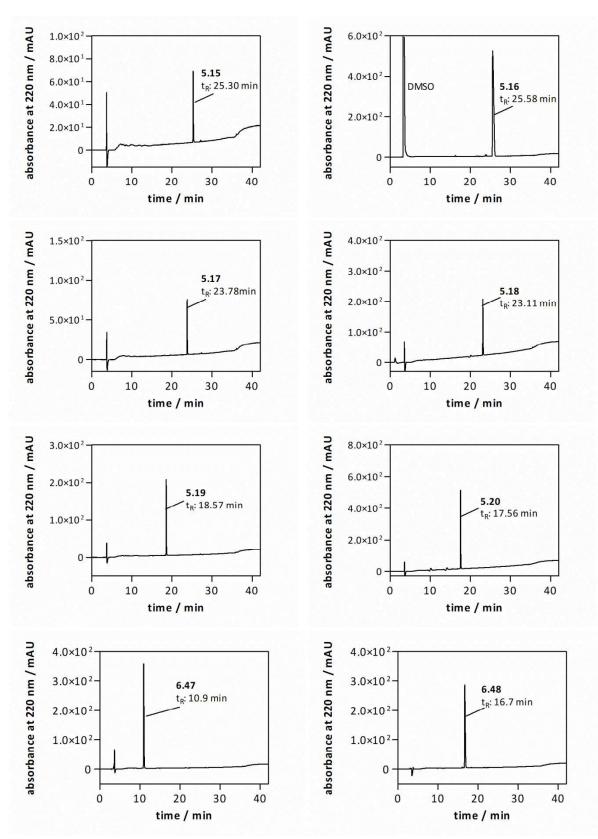
RP-HPLC analysis (purity control) of 4.32-4.38, 4.40



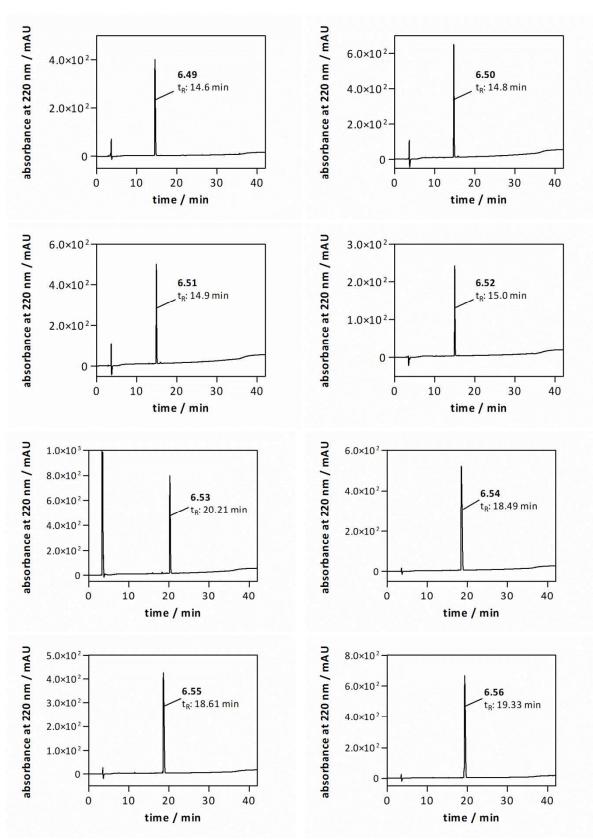
RP-HPLC analysis (purity control) of 4.41-4.44, 4.46-4.49



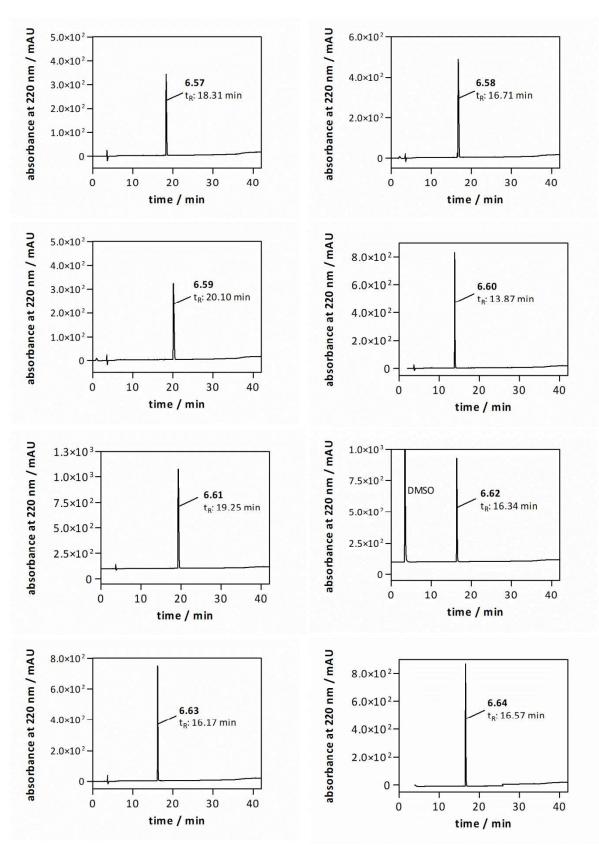
RP-HPLC analysis (purity control) of 4.50, BMY2536, 5.9-5.14



RP-HPLC analysis (purity control) of 5.15-5.20, 6.47, 6.48

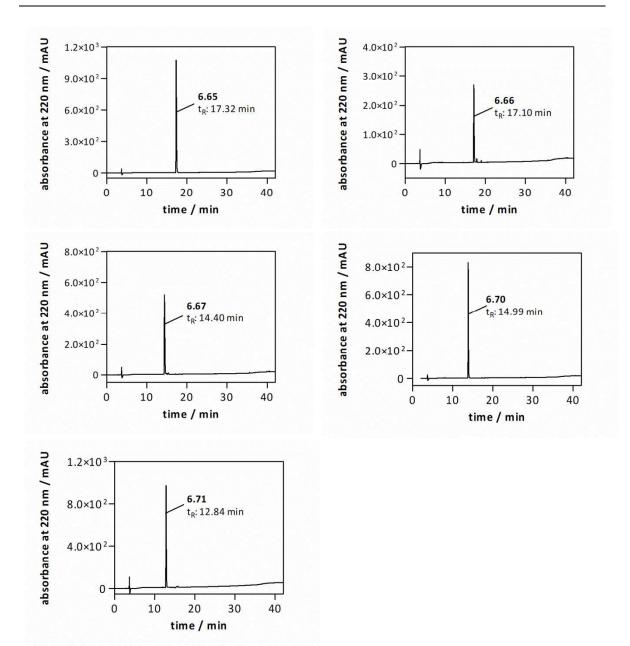


RP-HPLC analysis (purity control) of 6.49-6.56



RP-HPLC analysis (purity control) of 6.57-6.64





RP-HPLC analysis (purity control) of 6.65-6.67, 6.70, 6.71

ABBREVIATIONS

A.C.	adapulul gualaça	
AC AcOH	adenylyl cyclase acetic acid	
AIBN		
AML	azobisisobutyronitrile acute myeloid leukemia	
aq.	aqueous	
Ar	aryl	
B _{max}	maximum number of binding sites	
Boc	tert-butoxycarbonyl	
Bq	becquerel	
by brs	broad signal	
BRET	-	
BSA	bioluminescence resonance energy transfer bovine serum albumin	
c	concentration	
cAMP	cyclic adenosine monophosphate	
COSY	correlated spectroscopy	
CNS	central nervous system	
cpm	counts per million	
d	doublet	
δ	chemical shift	
DAG	diacylglycerol	
DCM	dichloromethane	
DIPEA	<i>N,N</i> -diisopropylethylamine	
DMAP	4-dimethylaminopyridine	
DMEM	dulbecco´s modified eagle medium,	
DMF	dimethylformamide	
DMSO	dimethylsulfoxide	
D _n R	, Dopamine receptor subtypes, n = 1,2,3,4,5; human isoform: hD _n R	
DPM	desintegrations per minute	
EC ₅₀	agonist concentration which induces 50% of the maximum response	
EDTA	ethylenediaminetetraacetic acid	
E _{max}	intrinsic activity	
eq.	equivalent(s)	
ESI	electrospray ionization	
EtOAc	ethyl acetate	
EtOH	ethanol	
FCS	fetal calf serum	
FP	fluorescence polarization	
FRAP	fluorescence recovery after photobleaching	
FRET	Förster resonance energy transfer	
GDP	guanosine diphosphate	
GEF	guanosine nucleotide exchange factor	
GPCR	G-Protein coupled receptor	
GRK	G-Protein coupled receptor kinase	
GTP	guanosine triphosphate short splice variant of the human denamine D, recentor	
hD_2R_{short}	short splice variant of the human dopamine D_2 receptor	

H _n R	Histamine receptor subtypes, n = 1,2,3,4; isoforms: dog (cH_nR), guinea pig (gp H_nR),	
Π _n r	human (hH _n R), mouse (mH _n R), rat (rH _n R)	
h		
	hour(s)	
HBTU	2-(1 <i>H</i> -benzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate	
HEK293T	human embryonic kidney cells	
НМВС	heteronuclear multiple bond correlation	
HPLC	high-performance liquid chromatography	
HRMS	high resolution mass spectrometry	
HSQC	heteronuclear single quantum coherence	
IC ₅₀	inhibitor concentration which suppresses 50% of an agonist induced effect or displaces 50% of a labeled ligand from the binding site	
IP ₃	inositol trisphosphate	
IR	infrared spectroscopy	
k	retention factor	
K _b	dissociation constant derived from a functional assay	
K _d	dissociation constant derived from a saturation binding assay	
Ki	dissociation constant derived from a competition binding assay	
k_{obs}	observed association rate constant	
$k_{\rm off}$	dissociation rate constant	
k _{on}	association rate constant	
lit.	literature	
m	multiplet	
Μ	molar (mol/L)	
MeCN	acetonitrile	
MeOH	methanol	
mol	mole(s)	
mp	melting point	
N ^G	guanidine nitrogen	
NMR	nuclear magnetic resonance	
NOESY	nuclear Overhauser effect spectroscopy	
ON	over night	
PBS	phosphate buffered saline	
PE	petroleum ether	
PLC	phospholipase C	
q	quartet	
qui	quintet	
R _f	relative to front	
RP	reversed phase	
RT	room temperature	
S	(1) singlet or (2) second(s)	
SEM	standard error of the mean	
t	(1) triplet or (2) time	
	2-(1H-Benzotriazole-1-yl)-1,1,3,3-tetramethylaminium tetrafluoroborate	
TEA TFA	triethyl amine trifluoroacetic acid	
THF	tetrahydrofuran	
	Ceranyarotaran	

TIRF	total internal reflection fluorescence
TLC	thin layer chromatography
ТМ	transmembrane
t _R	retention time
Tris	tris(hydroxymethyl)aminomethane
UV	ultraviolet

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Sabrina Biselli